

India's Number 1 Education App

CHEMISTRY

BOOKS - NTA MOCK TESTS

NTA JEE MOCK TEST 22

Chemistry

1. Incorrect order of properties given in

parenthesis are

A. Cl > S > F > O (Electron affinity)

$B. O > F > B > C(IE_2)$

$\mathsf{C}.\,N > O > P > S(IE_1)$

 $\mathsf{D}.\, O > C > B > N(ext{Electron affintiy})$

Answer: A

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2. Incorrect statement about Mn_2O_7 is :

A. It is a covalent oxide

B. Mn - O - Mn linkage is present

C. It is purple in colour

D. Mn is tetrahedrally surrounded by O

atoms

Answer: C

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3. Which of the following species is not expected to be a ligand?

A. $O_2^{\,-}$

B. NO^+

C. NH_4^+

$\mathsf{D.}\, CO$

Answer: C

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4. Which of the following reaction does not

occur spontaneously?

A.
$$Fe^{+3} + Cd \rightarrow Cd^{+2} + Fe^{+2}$$

B. $Fe^{+3} + I^- \rightarrow Fe^{+2} + I_2$
C. $Fe^{+2} + MnO_4^- \xrightarrow{H^+} Fe^{+3} + Mn^{+2}$
D. $Fe^{+3} + Ce^{+3} \rightarrow Fe^{+2} + Ce^{+4}$

Answer: D

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5. Total number of ionisation isomers of $[Co(en)_2(Br)(Cl)]NO_3$ are (including the given compound)

A. 3

B.4

C. 5

D. 9

Answer: A



6. If the pK_a of a weak and acid HA is 4.80 and the pK_b of a weak base BOH is 4.78. Then, the pH of an aqeuous solution of the corresponding salt, BA will be A. 8.23 B. 9.41 C. 7.01 D. 5.91

Answer: C



7. 500 g impure sample of $CaCO_3$ on heating gives 70 g of CaO. Percentage impurities in sample is

- A. 25~%
- B. 50 %
- C. 75 %
- D. 80~%

Answer: C



8. Number of photons having wavelength 632.8 nm, emitted by 5 mW laser source in 1 second is

A. $1.6 imes10^{19}$

B. $1.6 imes 10^{16}$

C. $1.6 imes 10^{25}$

D. $1.6 imes 10^{13}$

Answer: B

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9. Which of the following increases in magnitude as the atomic number of alkali metals increases?

A. Electronegativity

B. First ionization energy

C. ionic radius

D. Melting point

Answer: C

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10. The following reaction descomposes $N_2O_5 \rightarrow 2NO_2 + \frac{1}{2}O_2$. At a 25 degree centigrade the rate constant of the reaction is $5 \times 10^{-3} \sec^{-1}$. The initial pressure of N_2O_5 is 0.2 atm. If total pressure of gaseous mixture becomes 0.35 atm, then calculate the time of decomposition of N_2O_5 .

- A. 238.6 seconds
- B. 138.6 seconds
- C. 69.6 seconds

D. 89.3 seconds

Answer: B

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11. The P-T graph, as given below, was observed for a process on an ideal gas. Which of the

statement is true ?



A.
$$w=\,+\,ve,\Delta H=\,+\,ve$$

B.
$$w=-ve, \Delta H=-ve$$

C.
$$w=-ve, \Delta H=+ve$$

D.
$$w=\,+\,ve,\,\Delta H=\,-\,ve$$

Answer: C



12. A metal crystallizes into two cubic systems face centred cubic (fcc) and simple cubic (SC), whose unit cell lengths are 4.5Å and 1.501Å respectively. Calculate the ratio of densities of face centred cubic and Simple cubic.

A. 0.15

B. 1.44

C. 2.25

D. 3.25

Answer: A

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13. The resistance of a conductivity cell contaning 0.001MKCl solution at 298K is 1500 Ω . What is the cell constant if conductivity of 0.001MKCl solution at 298K is $0.146 \times 10^{-3}Scm^{-1}$.

A. 0.200 cm^{-1}

B. 0.219 cm $^{-1}$

C. $0.195 cm^{-1}$

D. $0.190 cm^{-1}$

Answer: B

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14. The compound which acts as antioxidant in

packed food is

A. Benzoic acid

B. Butylated hydroxy Toluene (BHT)

C. Para chlorometaxylenol

D. Sodium bicarbonate

Answer: B

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15.



Compound C in above reaction sequence is





Answer: D

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16. For the reaction



The correct statement regarding above reaction is

A. The reaction completes in more than

one step

B. Dichloro carbene is an electrophile

which attack on benzene ring



C. C. intermediate is formed during intermediate is formed

during reaction

D. A & B both are correct

Answer: D



17. The number of primary, secondary and tertiary carbons in the following, structure are respectively



B. 3, 6, 3

C. 3, 6, 2

D. 3, 2, 1

Answer: B

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18.
$$CH_3 - \underset{\substack{||}{O}}{C} - CH = CH - CH_3 \xrightarrow{HBr}{O}$$

find the major product

A.
$$CH_3 - \displaystyle \underset{||}{C} - CH_2 - \displaystyle \underset{|}{C} CH - CH_3$$



D. none of these

Answer: A



19. How many resonating structures are possible for the following structure (including





A. 8

- B. 9
- C. 10
- D. 11

Answer: B





 D_2O/DO^- , the maximum possible difference in molecular weight of compound (A) and (B) is

A. 2

B. 3

C. 4

D. 8

Answer: C





21.

Products

In the above reaction, number of possible

alkenes (products) are



22. The number of paramagentic species

among the following is

 $O_2, CO, N_2, C_2, CsO_2, BaO_2, CO_2$

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23. The relative humidity of air is 80% at $27^{\circ}C$. If the aqueous tension at the same temperature is 27 mm of Hg. The partial pressure of water vapour in the air will be (in mm of Hg).

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24. Graph between $\log\left(\frac{x}{m}\right)vs\log P$ is provided for adsorption of NH_3 gas on metal surface. Calculate weight of NH_3 gas (in gm) absorbed by 24 g of metal surface at 2 atm

pressure. (Take $\log 2 = 0.3$)





25. Melamine is used in formation of melamine

formaldehye resin. How many nitrogen atoms

are present in melamine unit.



