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India's Number 1 Education App

## CHEMISTRY

## BOOKS - NTA MOCK TESTS

## NTA JEE MOCK TEST 22

Chemistry

1. Incorrect order of properties given in parenthesis are
A. $C l>S>F>O \quad$ (Electron affinity)
B. $O>F>B>C\left(I E_{2}\right)$
C. $N>O>P>S\left(I E_{1}\right)$
D. $O>C>B>N($ Electron affintiy $)$

Answer: A

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2. Incorrect statement about $\mathrm{Mn}_{2} \mathrm{O}_{7}$ is :
A. It is a covalent oxide

## B. $M n-O-M n$ linkage is present

C. It is purple in colour
D. Mn is tetrahedrally surrounded by O
atoms

Answer: C

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3. Which of the following species is not expected to be a ligand?
A. $O_{2}^{-}$
B. $N O^{+}$
C. $\mathrm{NH}_{4}^{+}$
D. $C O$

Answer: C

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4. Which of the following reaction does not occur spontaneously?
A. $F e^{+3}+C d \rightarrow C d^{+2}+F e^{+2}$

$$
\begin{aligned}
& \text { B. } F e^{+3}+I^{-} \rightarrow F e^{+2}+I_{2} \\
& \text { C. } F e^{+2}+\mathrm{MnO}_{4}^{-} \xrightarrow{H^{+}} F e^{+3}+M n^{+2} \\
& \text { D. } \mathrm{Fe}^{+3}+\mathrm{Ce}^{+3} \rightarrow \mathrm{Fe}^{+2}+C e^{+4}
\end{aligned}
$$

## Answer: D

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5. Total number of ionisation isomers of
$\left[\mathrm{Co}(e n)_{2}(\mathrm{Br})(\mathrm{Cl})\right] \mathrm{NO}_{3}$ are (including the given compound)
A. 3
B. 4
C. 5
D. 9

Answer: A

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6. If the $p K_{a}$ of a weak and acid HA is 4.80 and
the $p K_{b}$ of a weak base $B O H$ is 4.78. Then, the
pH of an aqeuous solution of the

## corresponding salt, BA will be

A. 8.23
B. 9.41
C. 7.01
D. 5.91

Answer: C

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7.500 g impure sample of $\mathrm{CaCO}_{3}$ on heating
gives 70 g of CaO . Percentage impurities in
sample is
A. $25 \%$
B. $50 \%$
C. $75 \%$
D. $80 \%$

Answer: C

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8. Number of photons having wavelength
632.8 nm , emitted by 5 mW laser source in 1
second is
A. $1.6 \times 10^{19}$
B. $1.6 \times 10^{16}$
C. $1.6 \times 10^{25}$
D. $1.6 \times 10^{13}$

Answer: B

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9. Which of the following increases in magnitude as the atomic number of alkali metals increases?
A. Electronegativity
B. First ionization energy
C. ionic radius
D. Melting point

Answer: C

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10. The following reaction descomposes
$\mathrm{N}_{2} \mathrm{O}_{5} \rightarrow 2 \mathrm{NO}_{2}+\frac{1}{2} \mathrm{O}_{2}$. At a 25 degree centigrade the rate constant of the reaction is
$5 \times 10^{-3} \mathrm{sec}^{-1}$. The initial pressure of $N_{2} O_{5}$
is 0.2 atm. If total pressure of gaseous mixture becomes 0.35 atm, then calculate the time of decomposition of $\mathrm{N}_{2} \mathrm{O}_{5}$.
A. 238.6 seconds
B. 138.6 seconds
C. 69.6 seconds
D. 89.3 seconds

Answer: B

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11. The P-T graph, as given below, was observed
for a process on an ideal gas. Which of the
statement is true ?

A. $w=+v e, \Delta H=+v e$
B. $w=-v e, \Delta H=-v e$
C. $w=-v e, \Delta H=+v e$
D. $w=+v e, \Delta H=-v e$

Answer: C

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12. A metal crystallizes into two cubic systems face centred cubic (fcc) and simple cubic (SC), whose unit cell lengths are $4.5 \AA$ and $1.501 \AA$ respectively. Calculate the ratio of densities of face centred cubic and Simple cubic.
A. 0.15
B. 1.44
C. 2.25

## D. 3.25

## Answer: A

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13. The resistance of a conductivity cell contaning 0.001 MKCl solution at 298 K is 1500』. What is the cell constant if conductivity of $0.001 M K C l$ solution at 298 K is $0.146 \times 10^{-3} \mathrm{Scm}^{-1}$.
A. $0.200 \mathrm{~cm}^{-1}$
B. $0.219 \mathrm{~cm}^{-1}$
C. $0.195 \mathrm{~cm}^{-1}$
D. $0.190 \mathrm{~cm}^{-1}$

Answer: B

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14. The compound which acts as antioxidant in packed food is
A. Benzoic acid

# B. Butylated hydroxy Toluene (BHT) 

C. Para chlorometaxylenol
D. Sodium bicarbonate

Answer: B

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## EtOOC <br> $\mathrm{OCH}_{3} \mathrm{OCH}_{3}$

15. 

$$
\xrightarrow[(i i) \Delta]{(i) \mathrm{H}_{2} \mathrm{O}^{+}} A \xrightarrow[p H 5 / 6]{\xrightarrow{\mathrm{H}_{2} \mathrm{NOH}}} B \xrightarrow[\substack{(i i) \mathrm{H}_{3} \mathrm{O}^{+}}]{\left(\text {i) Conc. } \mathrm{H}_{2} \mathrm{SO}_{4}\right.} C
$$

Compound C in above reaction sequence is


## C. <br> 



## Answer: D

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16. For the reaction


The correct statement regarding above reaction is
A. The reaction completes in more than
one step
B. Dichloro carbene is an electrophile
which attack on benzene ring
during reaction
D. A \& B both are correct
17. The number of primary, secondary and tertiary carbons in the following, structure are respectively

A. $6,3,3$
B. $3,6,3$
C. $3,6,2$
D. 3, 2, 1

Answer: B

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18. $\mathrm{CH}_{3}-\underset{\substack{\| \\ O}}{\mathrm{C}}-\mathrm{CH}=\mathrm{CH}-\mathrm{CH}_{3} \xrightarrow{\mathrm{HBr}}$
find the major product

$$
\text { A. } \mathrm{CH}_{3}-\underset{\mid}{\mathrm{C}}-\mathrm{CH}_{2}-\underset{\mid}{\mathrm{C}} \mathrm{C}-\mathrm{CH}_{3}
$$

$$
\begin{aligned}
& \text { B. } \mathrm{CH}_{3}-\underset{\mid}{\mathrm{O}}-\underset{\mathrm{Br}}{\mathrm{O}}-\mathrm{CH}-\mathrm{CH}_{2}-\mathrm{CH}_{3} \\
& \text { C. } \mathrm{CH}_{3}-\underset{{ }_{\mathrm{O}}^{\mathrm{O}}}{\mathrm{C}}-\mathrm{CH}=\mathrm{CH}-\mathrm{CH}_{3} \\
& \text { D. none of these }
\end{aligned}
$$

## Answer: A

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19. How many resonating structures are possible for the following structure (including
the given structure)?

A. 8
B. 9
C. 10
D. 11

Answer: B

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After prolonged treatement of (A) by
$D_{2} \mathrm{O} / \mathrm{DO}^{-}$, the maximum possible difference
in molecular weight of compound (A) and (B)
is
A. 2
B. 3
C. 4
D. 8

## Answer: C

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21. 



## Products

In the above reaction, number of possible alkenes (products) are

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22. The number of paramagentic species among the following is
$\mathrm{O}_{2}, \mathrm{CO}, \mathrm{N}_{2}, \mathrm{C}_{2}, \mathrm{CsO}_{2}, \mathrm{BaO}_{2}, \mathrm{CO}_{2}$

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23. The relative humidity of air is $80 \%$ at
$27^{\circ} \mathrm{C}$. If the aqueous tension at the same temperature is 27 mm of Hg . The partial pressure of water vapour in the air will be (in $m m$ of Hg ).

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24. Graph between $\log \left(\frac{x}{m}\right) v s \log P$ is provided for adsorption of $\mathrm{NH}_{3}$ gas on metal surface. Calculate weight of $\mathrm{NH}_{3}$ gas (in gm)
absorbed by 24 g of metal surface at 2 atm pressure. (Take $\log 2=0.3$ )


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25. Melamine is used in formation of melamine
formaldehye resin. How many nitrogen atoms are present in melamine unit.

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