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## CHEMISTRY

## BOOKS - NTA MOCK TESTS

## NTA JEE MOCK TEST 76

## Chemistry

1. Consier the following ions
2. $N i^{2+}$
3. $\mathrm{Co}^{2+}$
4. $C r^{2+}$
5. $F e^{3+}$

Atomic number : $C r=24, F e=26, C o=27, N i=28$.
The correct sequence of increasing order of the number of unpaired
electrons in these ions is
A. $1,2,3,4$
B. $4,2,3,1$
C. 1, 3, 2, 4
D. $3,4,2,1$

## Answer: A

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2. What is the value of n in the following equation:

$$
\mathrm{Cr}(\mathrm{OH})_{4}^{-}+\mathrm{OH}^{-} \rightarrow \mathrm{CrO}_{4}^{2-}+\mathrm{H}_{2} \mathrm{O}+n e ?
$$

A. 3
B. 6
C. 5
D. 2
3. For given first order reaction, the reactant reduced to $1 / 4$ th its initial value in 10 min . The rate constant of the reaction is
A. $0.1386 \mathrm{~min}^{-1}$
B. $0.0693 \mathrm{~min}^{-1}$
C. $0.1386 \mathrm{~mol} \mathrm{~L}^{-1} \mathrm{~min}^{-1}$
D. $0.0693 \mathrm{~mol} \mathrm{~L}^{-1} \mathrm{~min}^{-1}$

## Answer: A

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4. The band structure in n -type semiconductor is

I

II

III

IV
A. I
B. II
C. III
D. IV

## Answer: A

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5. The reaction of solid $X e F_{2}$ with $A s F_{5}$ in 1:1 ratio affords
B. $X e F_{6}$ and $A s F_{3}$
C. $[X e F]^{+}\left[A s F_{6}\right]^{-}$
D. $\left[X e_{2} F_{3}\right]^{+}\left[A s F_{6}\right]^{-}$

## Answer: C

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6. For the compound.

the stereochemical notations are
A. $2 \mathrm{Z}, 4 \mathrm{R}$
B. 2Z, 4S
C. $2 \mathrm{E}, 4 \mathrm{R}$
D. $2 \mathrm{E}, 4 \mathrm{~S}$

## Answer: D

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7. The set representing the correct order of ionic radius is
A. $\mathrm{Li}^{+}>\mathrm{Be}^{2+}>\mathrm{Na}^{+}>\mathrm{Mg}^{2+}$
B. $\mathrm{Na}^{+}>\mathrm{Li}^{+}>\mathrm{Mg}^{2+}>\mathrm{Be}^{2+}$
C. $\mathrm{Li}^{+}>\mathrm{Na}^{+}>\mathrm{Mg}^{2+}>\mathrm{Be}^{2+}$
D. $\mathrm{Mg}^{2+}>\mathrm{Be}^{2+}>\mathrm{Li}^{+}>\mathrm{Na}^{+}$

## Answer: B

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8. When X amperes of current is passed through molten $\mathrm{AlCl}_{3}$ for 96.5 s . 0.09 g of aluminium is deposited. What is the value of X ?
A. 10A
B. 20 A
C. 30 A
D. 40 A

## Answer: A

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9. Arrange the following anilies in decreasing order of basicity
10. $\mathrm{C}_{6} \mathrm{H}_{5} \mathrm{NH}_{2}$
$o-\mathrm{CH}_{3 \mathrm{C}_{6} \mathrm{H}_{4} \mathrm{NH}_{2}}$
11. $\mathrm{m}-\mathrm{CH}_{3} \mathrm{C}_{6} \mathrm{H}_{4} \mathrm{NH}_{2}$
12. $p-\mathrm{CH}_{3} \mathrm{C}_{6} \mathrm{H}_{4} \mathrm{NH}_{2}$
A. $4>1>2>3$
B. $2>4>3>1$
C. $1>2>3>4$
D. $4>3>2>1$

## Answer: B

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10. Heat of neutralisation of strong acid and strong base under 1 atm and $25^{\circ} \mathrm{C}$ is -13.7 kcal . If standard Gibbs energy change for dissociation of water to $\mathrm{H}^{+}$and $\mathrm{OH}^{-}$is -19.14 kcal , the change in standard entropy for dissociation of water is:
A. 18.25
B. 110.2
C. -18.25
D. None of these

## Answer: B

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11. What is $(Z)$ is the following sequence of reaction?
$H C \equiv C H \xrightarrow{(i) 2 \mathrm{NaNH}_{2}(i i) 2 \mathrm{CH}_{3} \mathrm{I}}(X) \xrightarrow{\mathrm{HgSO}_{4}, \mathrm{H}_{2} \mathrm{SO}_{4}}(Y) \xrightarrow{(i) \mathrm{NaOH}+\mathrm{Br}_{3}(i i) \mathrm{H}_{3 O^{+}}}($
A. $\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CHO}$
B. $\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{COCH}_{3}$
C. $\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{COOH}$
D. $\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{COOH}$

## Answer: C

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12. When dilute $\mathrm{H}_{2} \mathrm{SO}_{4}$ and $\mathrm{H}_{2} \mathrm{O}_{2}$ are added to a solution of chromate ions, an intense blue colour is produced, which is stable in ether. Theis is
due to the formation of
A. $\mathrm{Cr}_{2} \mathrm{O}_{7}^{2-}$
B. $\mathrm{Cr}_{2} \mathrm{O}_{3}$
C. $\mathrm{CrO}_{5}$
D. $\mathrm{CrO}_{3}$

## Answer: C

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13. The vapour pressure of a pure liquid 'A' is 70 torr at $27^{\circ} \mathrm{C}$. It forms an ideal solution with another liquid $B$. The mole fraction of $B$ is 0.2 and total pressure of the solution is 84 torr at $27^{\circ} \mathrm{C}$. The vapour pressure of pure liquid B at $27^{\circ} \mathrm{C}$ is :
A. 140 torr
B. 56 torr
C. 14 torr
D. 70 torr

## Answer: A

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14. Gabriel phthalimide synthesis can be used for the preparation of amine from
A. $\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{Br}$
B. $\left(\mathrm{CH}_{3}\right)_{3} \mathrm{CBr}$
C. $p-\mathrm{CH}_{3} O C_{6} \mathrm{H}_{4} \mathrm{Br}$
D. $p-\mathrm{CH}_{3} \mathrm{C}_{6} \mathrm{H}_{4} \mathrm{Br}$

## Answer: A

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15. The red colour of oxyhaemoglobin is mainly due to
A. d-d transition
B. metal to ligand charge transfer transition
C. ligand to metal charge transfer transition
D. intraligand $\pi-\pi^{*}$ transition

## Answer: A

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16. The root mean square of gas molecules at 25 K and 1.5 bar is $100 \mathrm{~m} \mathrm{~s}^{-1}$. If the temperature is raised to 100 K and the pressure to 6.0 bar, the root mean square speed becomes
A. $200 \mathrm{~ms}^{-1}$
B. $100 \mathrm{~ms}^{-1}$
C. $400 \mathrm{~ms}^{-1}$
D. $1600 \mathrm{~ms}^{-1}$

## Answer: A

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17. The sequence of an mRNA molecule produced from a DNA template strand with the composition

5'-AGCTACACT-3' is
A. 5' - AUGUAGCU - 3
B. 5'- UCGAUGUGA-3'
C. 5'- AGTGTAGCT - 3'
D. 5' - TCGATGTGA - $\mathbf{3 '}^{\prime}$

## Answer: B

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18. Phosgene, $\mathrm{COCl}_{2}$, a poisonous gas decomposes according to the equation

$$
\mathrm{COCl}_{2}(g) \Leftrightarrow C O(g)+C l_{2}(g)
$$

If $K_{c}=0.083$ at $900^{\circ} C$, what is the value of $K_{p}$ ?
A. 0.125
B. 8.0
C. 6.1
D. 0.16

## Answer: B

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19. Acrilan fibre used for cloth, carpets and blankets, it the polymer of
A. acrylonitrile
B. ethylacrylate
C. styrene
D. monochlorotrifluoro ethane

## Answer: A

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20. The general trend in the properties of elements of carbon family shows that with increase in atomic number
A. the tendency towards catenation increases
B. the tendency to show +2 oxidation state increases
C. metallic character decreases
D. the tendency to form complexes with covalency higher than four decreases

## Answer: B

21. How many of these compounds are more reactive towards electrophilic substitution reaction than toluene. Phenol, Anlline, Anisole, Benzaldehyde, Chorobenzene,


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22. How many interhalogen compounds are not possible ?
$I C l, I B r, B r l, C l B r_{3}, C l F_{3}, B r C l_{5}, B r I_{5}, I c l_{3}, I F_{7}$

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23. How many of these compounds can react with ammonical silver nitrate solution to give whilte ppt.



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24. Find the total number of feasible products (including stereoisomers) in the following reaction. (No carboncation rearrangements is observed)


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25. The pH of a solution is 5.0 To this solution sufficient acid is added to decreases the pH to 2.0. The no. of times the concentration of $H^{+}$
