

CHEMISTRY

BOOKS - NTA MOCK TESTS

NTA JEE MOCK TEST 76

Chemistry

1. Consier the following ions

1. Ni^{2+}

2. Co^{2+}

3. $Cr^{2\,+}$

4. Fe^{3+}

Atomic number : Cr = 24, Fe = 26, Co = 27, Ni = 28.

The correct sequence of increasing order of the number of unpaired electrons in these ions is

- A. 1, 2, 3, 4
- B. 4, 2, 3, 1
- C. 1, 3, 2, 4

D. 3, 4, 2, 1

Answer: A



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- $Cr(OH)_4^- + OH^-
 ightarrow CrO_4^{2-} + H_2O + ne$?

2. What is the value of n in the following equation:

- - B. 6
 - C. 5

A. 3

D. 2

Answer: A

3. For given first order reaction, the reactant reduced to 1/4th its initial value in 10 min. The rate constant of the reaction is

A.
$$0.1386~\mathrm{min}^{-1}$$

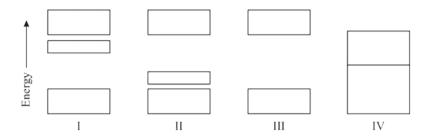
B. $0.0693 \, \mathrm{min}^{-1}$

C. $0.1386 \text{ mol } L^{-1} min^{-1}$

D. $0.0693 \text{ mol } L^{-1} min^{-1}$

Answer: A





A. I

B. II

C. III

D. IV

Answer: A



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5. The reaction of solid XeF_2 with AsF_5 in $1\!:\!1$ ratio affords

A. XeF_4 and AsF_3

 $B. XeF_6$ and AsF_3

C. $\left[XeF
ight]^{+}\left[AsF_{6}
ight]^{-}$

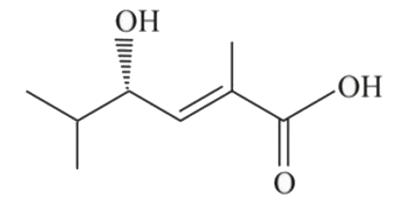
D. $\left[Xe_{2}F_{3}
ight]^{+}\left[AsF_{6}
ight]^{-}$

Answer: C



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6. For the compound.



the stereochemical notations are

A. 2Z, 4R

B. 2Z, 4S

C. 2E, 4R

D. 2E, 4S

Answer: D



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7. The set representing the correct order of ionic radius is

A.
$$Li^+ > Be^{2+} > Na^+ > Mg^{2+}$$

B.
$$Na^+ > Li^+ > Mg^{2+} > Be^{2+}$$

C.
$$Li^+ > Na^+ > Mq^{2+} > Be^{2+}$$

D.
$$Mg^{2+} > Be^{2+} > Li^+ > Na^+$$

Answer: B





0.09 g of aluminium is deposited. What is the value of X?

- A. 10A
- B. 20 A
- C. 30 A
- D. 40 A

Answer: A



- 9. Arrange the following anilies in decreasing order of basicity
- 1. $C_6H_5NH_2$
- $o-CH_{3C_6H_4NH_2}$
- $3. m CH_3C_6H_4NH_2$
- 4. $p CH_3C_6H_4NH_2$

A.
$$4 > 1 > 2 > 3$$

$${\rm B.}\,2 > 4 > 3 > 1$$

$$\mathsf{C.}\,1 > 2 > 3 > 4$$

D.
$$4 > 3 > 2 > 1$$

Answer: B



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10. Heat of neutralisation of strong acid and strong base under 1atm and $25\,^{\circ}\,C$ is -13.7kcal. If standard Gibbs energy change for dissociation of water to H^+ and OH^- is -19.14kcal, the change in standard entropy for dissociation of water is:

$$\mathsf{C.}-18.25$$

Answer: B



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11. What is (Z) is the following sequence of reaction?

 $HC\equiv CH\stackrel{\left(i
ight)2NaNH_{2}\left(ii
ight)2CH_{3}I}{\longrightarrow}\left(X
ight)\stackrel{HgSO_{4}\,,H_{2}SO_{4}}{\longrightarrow}\left(Y
ight)\stackrel{\left(i
ight)NaOH\,+\,Br_{3}\left(ii
ight)H_{3O^{+}}}{\longrightarrow}\left(X
ight)$

A.
$$CH_3CH_2CH_2CHO$$

C. CH_3CH_2COOH

B. $CH_3CH_2COCH_3$

D. $CH_3CH_2CH_2COOH$

Answer: C



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12. When dilute H_2SO_4 and H_2O_2 are added to a solution of chromate ions, an intense blue colour is produced, which is stable in ether. Theis is

due to the formation of A. $Cr_2O_7^{2\,-}$ B. Cr_2O_3 $C. CrO_5$ D. CrO_3 **Answer: C Watch Video Solution 13.** The vapour pressure of a pure liquid 'A' is 70 torr at $27^{\circ}\,C$. It forms an ideal solution with another liquid B. The mole fraction of B is 0.2 and total pressure of the solution is 84 torr at $27^{\circ}\,C$. The vapour pressure of pure liquid B at $27^{\circ}\,C$ is : A. 140 torr B. 56 torr C. 14 torr

_	70	L
D.	70	torr

Answer: A



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14. Gabriel phthalimide synthesis can be used for the preparation of amine from

A.
$$CH_3CH_2Br$$

B.
$$(CH_3)_3CBr$$

$$\mathsf{C.}\,p-CH_3OC_6H_4Br$$

D.
$$p-CH_3C_6H_4Br$$

Answer: A



15. The red colour of oxyhaemoglobin is mainly due to

A. d - d transition

B. metal to ligand charge transfer transition

C. ligand to metal charge transfer transition

D. intraligand $\pi-\pi^*$ transition

Answer: A



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16. The root mean square of gas molecules at 25 K and 1.5 bar is $100~{\rm m~s^{-1}}$. If the temperature is raised to 100 K and the pressure to 6.0 bar, the root mean square speed becomes

A. $200ms^{-1}$

B. $100ms^{-1}$

C. $400ms^{-1}$

_			_ 1
D.	160	00ms	_ I

Answer: A



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17. The sequence of an mRNA molecule produced from a DNA template strand with the composition

5'-AGCTACACT-3' is

A. 5' - AUGUAGCU - 3

B. 5'- UCGAUGUGA-3'

C. 5'- AGTGTAGCT - 3'

D. 5' - TCGATGTGA - 3'

Answer: B



18. Phosgene, $COCl_2$, a poisonous gas decomposes according to the equation

$$COCl_2(g) \Leftrightarrow CO(g) + Cl_2(g)$$

If $K_c=0.083$ at $900\,^\circ C$, What is the value of K_p ?

- A. 0.125
- B. 8.0
- C.6.1
- D.0.16

Answer: B



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19. Acrilan fibre used for cloth, carpets and blankets, it the polymer of

- A. acrylonitrile
- B. ethylacrylate

C. styrene

D. monochlorotrifluoro ethane

Answer: A



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20. The general trend in the properties of elements of carbon family shows that with increase in atomic number

A. the tendency towards catenation increases

B. the tendency to show $+\,2$ oxidation state increases

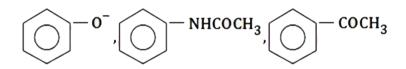
C. metallic character decreases

D. the tendency to form complexes with covalency higher than four decreases

Answer: B



21. How many of these compounds are more reactive towards electrophilic substitution reaction than toluene. Phenol, Anlline, Anisole, Benzaldehyde, Chorobenzene,





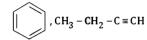
22. How many interhalogen compounds are not possible?

ICl, IBr, Brl, ClBr₃, ClF₃, BrCl₅, BrI₅, Icl₃, IF₇



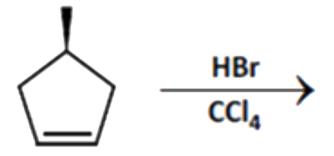
23. How many of these compounds can react with ammonical silver nitrate solution to give whilte ppt.

 $CH_2 = CH_2$, CH = CH, $CH_3 - C = CH$, $R - CH = CH_2$, $CH_3 - C = C - CH_3$,





24. Find the total number of feasible products (including stereoisomers) in the following reaction. (No carboncation rearrangements is observed)





25. The pH of a solution is 5.0 To this solution sufficient acid is added to decreases the pH to 2.0. The no. of times the concentration of $H^{\,+}$

increased is

