

CHEMISTRY

BOOKS - VGS PUBLICATION-BRILLIANT

THE P-BLOCK ELEMENTS - GROUP 13

Very Short Answer Questions

1. Discuss the pattern of variation in the oxidation states of Boron of Thallium .



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2. How do you explain higher stability of Tl Cl_3 ?



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3. Why does BF_3 behave as a Lewis acid ?



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4. Is boric acid a protic acid ? Explain



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5. What happens when boric acid is heated ?



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6. Describe the shapes of BF_3 and BH_4^- .

Assign the hybridization of boron in these species.



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7. Explain why atomic radius of Ga is less than that of Al.



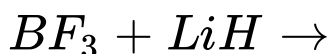
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8. Explain inert pair effect.



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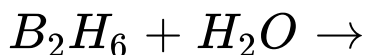
9. Write balanced equations for





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10. Write balanced equations for



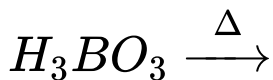
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11. Write balanced equations for



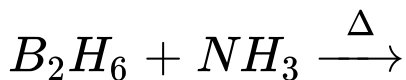
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12. Write balanced equations for



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13. Write balanced equations for



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14. Why is boric acid polymeric ?



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15. What is the hybridization of B in diborane and borazine ?



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16. Write the electronic configuration of group-13 elements .



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17. Give the formula of borazine . What is its common name ?



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18. Give the formulae of

(a) Borax , (b) Colemanite



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19. Give two uses of aluminium .



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20. What happens when Li AlH_4 and BCl_3 mixture in dry ether is warmed



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21. What happens when Borax is heated with H_2SO_4 ?



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22. Sketch the structure of Orthoboric acid .



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23. Write the structure of $AlCl_3$ as a dimer.



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24. Metal borides (having ^{10}B) are used as protective shield - why ?



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Short Answer Questions

1. Write reactions to justify amphoteric nature of aluminium .



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2. What are electron deficient compounds ? Is BCl_3 an electron deficient species ? Explain.



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3. Suggest reasons why the B-F bonds lengths in BF_3 (130 pm) and BF_4^- (143 pm) differ.



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4. B-Cl bond has a bond moment. Explain why BCl_3 molecule has zero dipole moment.



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5. Explain the structure of boric acid.



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6. What happens when Borax is heated strongly ?



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7. What happens when Boric acid is added to water ?



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8. What happens when Aluminium is heated with dilute NaOH ?



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9. What happens when BF_3 is treated with ammonia ?



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10. What happens when Hydrated alumina is treated with aq.NaOH solution ?



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11. Give reasons

Conc. HNO_3 can be transported in aluminium container.



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12. Give reasons

A mixture of dil. NaOH and aluminium pieces
is used to open drain.



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13. Give reasons

Aluminium alloys are used to make aircraft
body



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14. Give reasons

Aluminium utensils should not be kept in water overnight.



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15. Give reasons

Aluminium wire is used to make transmission cables .



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16. Explain why the electronegativity of Ga, In and Tl will not vary very much.



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17. Explain borax bead test with a suitable example



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18. Explain the structure of diborane.



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19. Explain the reactions of aluminium with acids.



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20. Write a short note on the anomalous behaviour of boron in the group - 13.



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21. Aluminium reacts with dil. HNO_3 but not with conc. HNO_3 -explain.



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22. Give two methods of preparation of diborane.



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23. How does diborane react with H_2O ?



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24. How does diborane react with CO ?



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25. How does diborane react with $N(CH_3)_3$?



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26. Al_2O_3 is amphoteric -explain with suitable reactions.



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27. $Na_2B_4O_7$ + Conc.
 $H_2SO_4 \rightarrow A \xrightarrow[\text{(ii) Ignite}]{\text{(i) } C_2H_5OH} B$ (Green edged
flame)

Identify A and B .



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Long Answer Questions

1. How are borax and boric acid prepared ?

Explain the action of heat on them.



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2. How is diborane prepared ? Explain its structure .



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3. Write any two methods of preparation of diborane. How does it react with (i) Carbon monoxide and (ii) Ammonia ?



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