



## **BIOLOGY**

# BOOKS - SURA BIOLOGY (TAMIL ENGLISH)

# **SELF-EVALUATION 7 MARKS**

Answer All The Questions Physics

**1.**  $_{88}Ra^{226}$  experiences three lpha-decay . Find the number of neutrons in the daughter element.



3. (a) Why does sound travel faster on a rainy

day than on a dry day?

(b) Write the use of radioactive isotopes used

in industries?

**4.** (a) What do you understand by the term 'ultrasonic vibration'?

(b) State three uses of ultrasonic vibrations.

(c) Name three animals which can hear ultrasonic vibrations.



5. Compare the properties of alpha, beta, and

gamma radiations.

6. State the universal law of gravitation and

derive its mathematical expression .



7. What is meant by mass defect?

(b) 100 W bulb draws 680 mA current. How

much time will be required to pass 30 C of

charge through the bulb?

**8.** A bullet of mass 50 g moving with a speed of 300  $ms^{-1}$  is brought to rest in 1 s . Find the impulse and the force .

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9. An object of height 3 cm is placed at 10 cm

from a concave lens of focal length 15 cm. find

the size of the image.

**10.** (a) Find the final temperature of a copper rod. Whose area of cross section changes from  $10m^2$  to  $11m^2$  due to heating. The copper rod is initially kept at 90K. (Coefficient of superficial expansion is 0.0021/K). (b) The far position of a myopic person is 90 cm is front of the eye. what is the nature and power of the lens required to correct the problem?



**11.** State Soddy and Fajan's displacement law.



difference.





**15.** Explain the experiment of measuring the real and apparent expansion of a liquid with a neat diagram.



**16.** (a) Classify the types of force based on their application.

(b) Define coefficient of linear expansion. Write

its equation.

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its equation.

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17. State Joule's law of heating.

**18.** How will you determine the velocity of sound by echo method?

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**19.** Distinguish between the resistivity and conductivity of a conductor .

**20.** A person who is sitting at a distance of 400 m from a source of sound is listening to a sound of 600 Hz. Find the timee period between successive compressions from the source?

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21. Explain the principle & structure of atom

bomb.



**22.** (a) Derive the Magnification of a compound.

(b) Two block of masses 8 kg and 2 kg respectively lie ono a smooth horizontal surface in contact with one other. They are pushed by a horizontally applied force of 15 N. calculate the force exerted on the 2 kg mass.



**23.** (a) Define the unit of current.

(b) How do we see distant and closer objects?



24. Derive the ideal gas equation by combining

the empirical gas laws.



25. Explain about domestic electric circuits.

(circuit diagram not required)

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**26.** (a) A cricket ball of mass 100g moving with a speed of  $20ms^{-1}$  is brought to rest by a player. Find the change in momentum of ball. (b) Draw a ray diagram of formation of images by a convex lens. 27. (a) State boyle's law.

(b) State Newton's second law.

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#### 28. State and prove the law of conservation of

linear momentum.

**29.** (a) State Rayleigh's law of scattering.

(b) Calculate the energy consumed by 120 w toaster in 20 min.



#### **30.** List any five properties of light.



**31.** (a) Define dispersion of light.

(b) Give the application of torque.

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**32.** (a) What are the medical applications of echo?

(b) How can you calculate the speed of sound

using echo?

33. With an illustration, explain the method of

calculation for areal expansion of an object.



34. Write any five electrical components used

in electrical circuit and draw its symbol.

**35.** (a) A body of mass 2 kg moving with uniform velocity of 40  $ms^{-1}$  collides with another body at rest if two bodies move together with a velocity o 20  $ms^{-1}$ . Find the mass of the other body. (b) Distinguish between linear, areal (or)

superficial expansion.

**36.** (a) Air temperature in the Rajasthan desert can reach  $46^{\circ}C$ . What is the velocity of sound in air at that temperature ?  $(V_0 = 331ms^{-1})$ . (b) A sound wave has a frequency of 200 Hz and a speed of  $400ms^{-1}$  in a medium. Find the wavelength of the sound wave.

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37. Explain the construction and working of a

'Compound Microscope'.





38. What happens to the resistance, as the

conductor is made thicker ?

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Answer All The Questions Chemistry

1. (a) What is a polytomic molecule?

(b) When an aqueous solution of potassium chloride is added to an aqueous solution of

silver nitrate, a white precipitate is formed.

Give the chemical equation of this reaction.



**2.** Classify the following compounds based on the patterns of carbon chain and give their structural formula : (i) Propane (ii) Benzene (iii) Cyclobutane (iv) Furan

**3.** What is rust ? Give the equation for formation of rust .

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**4.** (a) What happens during a chemical change?

(b) Calculate the number of molecules in 11g of

 $CO_2$ .

5. Write notes on various factors affecting

solubility.



6. Explain the salient features of periods in the

modern periodic table.

7. (a) What are Bio degradable and Non-

biodegradable detergents?

(b) Give appropriate reasons for alloying.



**8.** (a) What is a homoatomic molecule? Give two examples.

(b) State two conditions necessary for rusting

of iron.



**9.** explain solid solution ,Liquid solution and Gaseous solution .



**10.** Explain hydraulic washing with a neat diagram.



11. Explain the salient features of periods in

the modern periodic table.

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**12.** (a) The aquatic animals live more in cold regionn why?

(b) How to identify saturated annd

unsaturated compounds?

**13.** Give a test to identify the presence of alcohol.



14. How will you determine the atomicity of

gases using Avogadro's hypothesis?

15. Explain the mechanism of cleansing action

of soaps and detergents.

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**16.** On analysing an impure sample of sodium chloride, the percentage of chloride was found to be 45.5 what is the percentage of pure sodium chloride in the sample?



17. (a) Wriite any three uses of copper.

(b) Ionisation energy decreases down the group in periodic table. Give reason.



**18.** Explain smelting process.

**19.** (a) How do detergents cause water pollution? Suggest remedial measures to prevent this pollution.

(b) Differentiate combination and

decomposition reactions.

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20. In what way hygroscopic substances differ

from deliquescent dubstances .

**21.** Calculate the gram molecular mass of the following (i)  $CH_4$ 

(ii)  $CO_2$ 

(iii)  $NH_3$ .

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# **22.** A hot saturated solution of copper sulphate forms crystals as it cools . Why ?

**23.** (a) Explain Bassermerisation.

(b) What s molar volume of a gas?



25. Explain the factors influcencing the rate of

a reaction





28. Explain the classification based on the

direction of the reaction.

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**29.** (a) Define heterocylic compounds. Give an example.

(b) 60 grams of NaOH is dissolved in 120 grams of water at  $25^{\circ}C$  to form a saturated solution. Find mass percentage of solute and solvent.



30. (a) Explain the types of double displacement reactions with examples.(b) What are the advantages of detergents over soaps?

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31. Derive the relationship between relative

molecular mass and Vapour density.
**32.** (a) Powdered  $CaCO_3$  reacts much faster with HCl than with solid marble chips. Account for the following.

(b) Define atomic mass unit?



**33.** Atomic radii decreases as we move from left to right of periodic table. Justify your answer.





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**35.** How is ethanol manufactured from

sugarcane?

36. Write down the steps involved in a

metallurgical process.



**37.** (a) What happens when  $MgSO_4.7H_2O$  is

heated? Write the appropriate equation

(b) Define solublity.



38. Methods of preventing corrosion



# Answer All The Questions Biology

**1.** (a) What is collateral vascular bundle?(b) Why did Mendel select pea plant for his



2. Why is the management of forest and wildlife resource considered as a challenging task?

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**3.** Explain the role of Okazaki fragments.

4. (a) How is the circulatory system disegned

in leech to comensate the heart structure?

(b) What are the advantages of using biogas ?

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5. Trace the pathway followed by water molecules from the time it enters a plant root to the time it escapes into the atmosphere from a leaf.





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7. (a) Differentiate the following Aerobic and

Anaerobic respiration.

(b) What are allosomes?



8. How does rainwater harvesting structures

recharge ground water?

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**9.** How does the light dependent reaction differ from the light independent reaction? What are the end product and rectants in each? Where does each reaction occur within the chloroplast?

10. (a) What is a dental formula?(b) What are the various routes by which transmission of human immuno deficiency virus takes place?

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**11.** (a) Write a note on euploidy.

(b) Enumerate the functions of blood.

12. Illustrate the structure and functions of

brain.

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**13.** (a) Differentiate the following monocot root and Dicot root.

(b) What will happen if you cut planaria into

small fragments?

**14.** (a) Draw the digestive system of leech and label the parts.

(b) Write a note an smoking hazard and

effects of Tobacco.



**15.** Write the physiological effect of Auxin.

**16.** The sex of the new born child is a matter of chance and neither of the parents may be considered responsible for it.What would be the possible fusion of gametes to determine the sex of the child?



17. (a) Draw diagrams to represent the types of

concentric vascular bundles.

(b) List the function of cerebrospinal fluid.





18. (a) Draw the dorsal view of brain of rabbit

& label the parts.

(b) What is colostrum? How is milk production

hormonally regulated?

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**19.** Write a note on DNA replication.

**20.** (a) What are heart sounds? How are they produced?

(b) What is the aim of wild life conservation?



## 21. (a) Draw the ultrastructure of mitochondria

and label the parts.

(b) How does locomotion take place in leech?

22. Explain the male reproductive system of

rabbit with a labelled diagram.



**23.** (a) List the difference between RBC and WBC.

(b) Define triple fusion.



**24.** (a) Write a note on Triticale.

(b) Differentiate between Type-1 and Type-2 diabetes mellitus.



### 25. With a neat labelled diagram describe the

parts of a typical angiosperm ovule .



26. Describe mutation breeding with an example.
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27. (a) What is photosynthesis and where in a

cell does it occur?

(b) How can menstrual hygiene be maintained

during menstrual days?

**28.** (a) How was IR-8 variety produced?

(b) List the source of e-wastes.



29. With a neat labelled diagram explain the

structure of a neuron.

**30.** (a) Name the gaseous plant hormone.Describe its three different actions in plants.(b) Which hormone is known as stress hormone in plants ? Why?

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**31.** Distinguish between

undifferentiated cells and differenetiated cells.

32. What are the phases of menstrual cycle?

Indicate the changes in the ovary and uterus.



**33.** What are the hormones secreted by posterior lobe of the pituitary gland? Mention the tissues on which they exert their effect



34. (a) What precautions can be taken for

preventing heart disease?

(b) Distinguish between somatic gene therapy

and germ line gene therapy.

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**35.** Enumerate the importance of forest.



**37.** (a) State the applications of DNA finger printing technique.

(b) Describe the structure of spinal cord.







#### Answer All The Questions Science

1. Two bulbs are having the ratings as 60 W,

220 V and 40 W, 220 V respectively. Which one

has a greater resistance?



**2.** Calculate the coefficient of cubical expansion of a zinc bar whose volue is increased  $0.25m^3$  from  $0.3m^3$  due to change in its temperature of 50K.



**3.** A source producing a sound of frequency 500 Hz is moving towards is listener with a velocity of 30  $ms^{-1}$ . The speed of the sound is



#### listener?



**5.** For a person with hypermeteropia, the near point has moved to 1.5 m. calculate the focal

length of the correction lens in order to make

his eyes normal.



**6.** Find the mass of potassium chloride would be needed to form a saturated solution in 60 g of water at 303 K? Given that solubility of the KCl is 37/100 g at this temperature.

7. Calculate the pH of  $1 \times 10^{-4}$  molar solution of NaOH. Watch Video Solution

**8.** Identily A. B, C, and D from the following nuclear reactions.

(i)  ${}_{13}Al^{27} + A 
ightarrow {}_{15}P^{30} + B$ 

(ii)  $_{12}mg^{24} + B 
ightarrow {}_{11}Na^{24} + C$ 

(iii)  ${}_{92}U^{238} + B o {}_{93}Np^{239} + D$ 

**9.** Calculate the velocity of a moving body of mass 5 kg whose linear momentum is 2.5  $kgms^{-1}$ .



10. Find the mass of 2.5 mole of oxygen atom .

**11.** When an object is placed at 25 cm from a concave lens, a virtual image is produced at a distance of 10 cm. Calculate the magnification produced by the lens.

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12. Calculate the current and the resistance of

100W, 200V electric bulb in an electric circuit.

**13.** Find the speed of sound in air at  $23^{\circ}C$ . (consider the speed of sound in air at  $0^{\circ}C$  is  $331.3ms^{-1}$ ).

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**14.** A person with myopia can see objects plaecd at a distance of 4 m. if he wants to see objects at a distance of 20 m, what should be the focal length and power of the concave lens he must wear?

**15.** The potential difference between two conductor is 110 V. How much work in moving 5 C charge from one conductor to the other ?

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16. A radon specimen emits radiation of  $3.7 imes 10^3 GBq$  pe second. Convert this disintegration in terms of curie. (one curie  $= 3.7 imes 10^{10}$  disintegration per second)





**17.** At what speed should a source of sound move away from a stationary observer so that observer finds the apparent frequency equal to half of the original frequency?



**18.** A 110 V light bulb takes 0.9 A current and operates 12h/day. Determine the energy consumed by the bulb for 30 days.



distance from the hinges is 90 cm, with a force of 40 N. calculate the moment of the force about the hinges.





21. If 50 g was the loss in mass as a result of a

fissionable reaction, how much energy will

have been produced ?

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**22.** 0.3 mole of aluminium (Atomic mass of Al=27).

23. Calculate the pH of a solution in which the

concentration of the hydrogen ions is

 $1.0 imes 10^{-8} mol$  litre<sup>-1</sup>.



**24.** The solubility of sodium nitrate at  $50^{\circ}C$ and  $30^{\circ}C$  is 114g and 96g respectively. Find the amount of salt that will be thrown out when a saturated solution of sodium nitrate containing 50 g of water is cooled from  $50\,^\circ C$ 

to  $30^{\circ}C$ ?



**25.** If boiling point of water is  $95^{\circ}F$ . What will

be the reading in kelvin scale?