

CHEMISTRY

BOOKS - SURA CHEMISTRY (TAMIL ENGLISH)

ORGANIC NITROGEN COMPOUNDS

Choose The Correct Answer

1. Which of the following reagent can be used to convert nitrobenzene to aniline

A. Sn/HCl

B. ZnHg/NaOH

C. $LiAlH_4$

D. All of these

Answer: A::C



2. The method by which aniline cannot be prepared is

A. degradation of benzamide with Br_2 /NaOH.

B. potassium salt of phthalimide treated with chlorobenzene

followed by hydrolysis with aqueous NaOH solution.

- C. hydrolysis of phenylcyanide with acidic solution
- D. reduction of nitrobenzene by Sn/HCl

Answer: B

3. Which one of the following will not undergo Hofmann bromamide reaction

A. $CH_3CONHCH_3$

 $\mathsf{B.}\,CH_3CH_2CONH_2$

 $\mathsf{C.}\,CH_3CONH_2$

D. $C_6H_5CONH_2$

Answer: A

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4. Assertion : Acetamide on reaction with KOH and bromine gives acetic acid.

Reason : Bromine catalyses hydrolysis of acetamide.

A. if both assertion and reason are true and reason is the

correct explanation of assertion.

B. if both assertion and reason are true but reason is not the

correct explanation of assertion.

C. assertion is true but reason is false

D. both assertion and reason are false.

Answer: D



A. bromomethane

B. α -bromo sodium acetate

C. methanamine

D. acetamide

Answer: C



6. Which one of the following nitro compounds does not react with nitrous acid

A.
$$CH_3-CH_2-CH_2-NO_2$$

 $\mathsf{B.} \left(CH_3 \right)_2 CH - CH_2 NO_2$

 $\mathsf{C.} (CH_3)_3 CNO_2$



Answer: C



Answer: C

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8. The product formed by the reaction an aldehyde with a primary amine

A. carboxylic acid

B. aromatic acid

C. schiff's base

D. ketone

Answer: C

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9. Which of the following reaction is not correct.

A. $CH_3CH_2NH_2 \stackrel{NHO_2}{\longrightarrow} CH_3CH_2OH + N_2$





D. None of these

Answer: B

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10. When aniline reacts with acetic anhydride the product formed is

A. o-aminoacetophenone

B. m-aminoacetophenone

C. p-aminoacetophenone

D. acetanilide

Answer: D



11. The order of basic strength for methyl substituted amines in equeous solution is

A. $N(CH_3)_3 > N(CH_3)_2H > N(CH_3)H_2 > NH_3$ B. $N(CH_3)H_2 > N(CH_3)_2H > N(CH_3)_3 > NH_3$ C. $NH_3 > N(CH_3)H_2 > N(CH_3)_2H > N(CH_3)_3$ D. $N(CH_3)_2H > N(CH_3)H_2 > N(CH_3)_3 > NH_3$

Answer: D



12.
$$C_6H_5NO_2 \xrightarrow{Fe/HCl} A \xrightarrow{NaNO_2/HCl} B \xrightarrow{H_2O} C$$
 'C' is

A. $C_6H_5 - OH$

 $\mathsf{B.}\, C_6H_5-CH_2OH$

 $C. C_6H_5 - CHO$

D. $C_6H_5NH_2$

Answer: A



- 13. Nitrobenzene on reaction with fuming nitric acid at
- $80-100^{\,\circ}\,C$ forms which one of the following products?
 - A. 1,4 dinitrobenzene
 - B. 2,4,6 tirnitrobenzene
 - C. 1,2 dinitrobenzene
 - D. 1,3 dinitrobenzene

Answer: D

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14. $C_5H_{13}N$ reacts with HNO_2 to given an optically active compound - The compound is

A. pentan - 1 - amine

B. pentan - 2- amine

C. N,N - dimethylpropan - 2 - amine

D. N - methylbutan - 2 - amine

Answer: B

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15. Secondary nitro alkanes react with nitrous acid to form

A. red solution

B. blue solution

C. green solution

D. yellow solution

Answer: B

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16. Which of the following amines does not undergo acetylation?

A. t- butylamine

B. ethylamine

C. diethylamine

D. triethylamine

Answer: D



17. Which one of the following is most basic?

A. 2,4 - dichloroaniline

B. 2,4 - dimethyl aniline

C. 2,4 - dinitroaniline

D. 2,4, - dibromoaniline

Answer: B

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18. Ammonium salt of bezoic acid is heated strongly with P_2O_5 and the product so formed is reduced and then treated with $NaNO_2/HCl$ at low temperature. The final compound formed is

A. Benzene diazonium chloride

B. Benzyl alcohal

C. Phenol

D. Nitrosobenzene

Answer: B



19. Among the following, the reaction that proceeds through an

electrophilic substitution is

В. 📄





Answer: B

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20. Nitroation of nitrobenzene results in

A. o-dinitrobenzene

B. 1,2,3 trinitrobenzene

C. m-dinitrobenzene

D. p-dinitrobenzene

Answer: C

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21. Which among the following reaction will give a secondary amine?

 $\begin{array}{l} \text{A. } CH_3CONH_2 \xrightarrow{Br_2/KOH} \\ \text{B. } CH_3CONH_2 \xrightarrow{LiAlH_4} \\ \text{C. } CH_3CN \xrightarrow{Na/C_2H_5OH} \\ \text{D. } CH_3NC \xrightarrow{LiAlH_4} \end{array}$

Answer: D

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22. Electrolytic reduction of nitrobenzene in strongly acidic medium gives

A. aniline

B. p-aminophenol

C. m-nitroanline

D. azoxybenzene

Answer: B



23. $(CH_3)_2 - CH - NO$ when boiled with HCl/H_2O gives compound 'X'. Identify 'X'.

A. CH_3COOH

$$\mathsf{B}.\,(CH_3)_2C=O$$

 $C. CH_3 CHO$

 $\mathsf{D.}\, CH_3NO_2$

Answer: B



24. Ethylamine on heatiing with CS_2 in the presence of $HgCl_2$

forms

A. C_2H_5NCS

 $\mathsf{B.}\, C_2 H_5 CN$

 $\mathsf{C.}\,C_2H_5NS$

D. $(C_2H_5)_2S$



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25. Which of the following cannot be prepared by Sandmeyer's

reaction?

A. Iodobenzene

B. Fluorobenzene

C. Both A and B

D. Neither A nor B

Answer: C

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26. Identify the product of the following reaction.

 $CH_3MgBr + Cl - CN \rightarrow ?$

A. ethanenitrile

B. acetamide

C. propanenitrile

D. acetaldoxime

Answer: A

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27. Positive carbylamine test is not shown by?

A. methyl amine

B. aniline

C. N-ethyl aniline

D. triethyl amine

Answer: D

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28. The gas responsible for Bhopal Gas Tragedy is

A. Phosgene

B. Methyl isocyanide

C. Methyl isocyanate

D. Methyl isothlocyanide

Answer: C



29. Identify the amide which will not undergo Hofmann bromamide reaction.

A. $CH_3CONHCH_3$

 $\mathsf{B.}\,CH_3CH_2CONH_2$

C. CH_3CONH_2

D. Both B and C

Answer: A

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30. Which among the following will from significant amount of meta isomer on nitration?

A.	

В. 📄





Answer: B

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31. Which among the following exists as zwitter ion?

A. sulphanilic acid

B. salicylic acid

C. acetanilide

D. All of these



32. Bromo ethane reacts with silver nitrite to give

A. $C_2H_5NO_2$

- $\mathsf{B.}\,C_2H_5-O-NO$
- $\mathsf{C.}\, C_2H_5Ag+NaBr$

 $\mathsf{D.}\, C_2 H_5 NC$

Answer: A



33. In nitro alkanes $-NO_2$ group is converted to $-NH_2$ group

by the reaction with

A. Sn/HCl

B. Zn dust

C. Zn/NH_2

D. Zn/NaOH

Answer: A

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34. When nitromethane is reduced with Zn dust $+NH_4Cl$ in neutral medium, we get

A. CH_3NH_2

 $\mathsf{B.}\, C_2H_5NH_2$

 $\mathsf{C.}\,CH_3NHOH$

 $\mathsf{D.}\, C_2 H_5 COOH$

Answer: C



35. The compound that is most reactive towards electrophilic nitration is

A. Toluene

B. Benzene

C. Benzoic acid

D. Nitrobenzene



36. Nitromethane condenses with acetaldehyde to give

A. nitro propane

B. 1-nitro 2-proanol

C. 2-nitro-1-propanol

D. 3-nitro propanol

Answer: B



37. Which of the following compounds has the smell of bitter

almonds?

A. aniline

B. nitro methane

C. benzene sulphoni acid

D. Nitrobenzene

Answer: D

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38. Nitrobenzene on electrolytic reduction in con.sulphuric acid,

the intermediate formed is

A. $C_6H_5NH-NHC_6H_5$

 $\mathsf{B.}\, C_6H_5-NHOH$

C. $C_6H_5-N=N-C_6H_5$

 $\mathsf{D.}\, C_6H_5HSO_4$

Answer: B



39. Electrophile used in the nitration of benzene is

A. hydronium ion

B. sulphonic acid

C. nitronium ion

D. bromide ion

Answer: C

40. The reduction of $CH_3 - CH_2 - C \equiv N$ with sodium and

alcohol results in the formation of



- $\mathsf{B.}\,CH_3-CH_2-CH_2-OH+N_2$
- $\mathsf{C.}\,CH_3-CH_2-CH_2-NH_2$
- $\mathsf{D}.\,CH_3-CH_2-NH_2$

Answer: C



41. The basic character of amines is due to the

- A. tetrahedral structure
- B. presence of nitrogen atom
- C. lone pair of electrons on nitrogen atom
- D. high electronegativity of nitrogen

Answer: C

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42. The organic compound that undergoes carbylamine reaction

is

A. $(C_2H_5)_2NH$

 $\mathsf{B.}\, C_2H_5NH_2$

 $\mathsf{C}.\,(C_2H_5)_3N$

D. $\left(C_2H_5
ight)_4N^+L^-$

Answer: B

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43. Primary amine acts as

A. Electrophile
B. Lewis base
C. Lewis acid

D. Free redical

Answer: B



44. Oxidation of aniline with acidified potassium dichromate gives

A. p-benzo quinone

B. benzoic acid

C. benzaldehyde

D. benzyl alcohol

Answer: A

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45. Which one of the following is a secondary amine?

A. aniline

B. diphenyl amine

C. sec.butylamine

D. tert.butylamine

Answer: B

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46.
$$C_6H_5NH_2 \xrightarrow{NaNO_2/HCl} X$$
 Identify X.

A. C_6H_5Cl

 $\mathsf{B.}\, C_6H_5NHON$

 ${\rm C.}\, C_6H_5N_2Cl$

 $\mathrm{D.}\, C_6 H_5 OH$

Answer: C

47. Which of the following will not undergo diazotisation?

A. m-toludine

B. aniline

C. p-amino phenol

D. benzyl amine

Answer: D



48. Aniline differs from ethylamine by the reaction with

A. metallic sodium

B. an alkyl halide

C. chloroform and caustic potash

D. nitrous acid

Answer: D

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49. When aqueous solution of benzene diazonium chloride is

boiled the product formed is

A. benzyl alcohol

B. benzene $+N_2$

C. phenol

D. phenyl hydroxylamine

Answer: C




50. Which among the following dissolves in NaOH?

A. CH_3NO_2

 $\mathsf{B.} \left(CH_3 \right)_3 CNO_2$

 $\mathsf{C.}\, C_6H_5NO_2$

D. C_6H_5ONO

Answer: A

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51. On hydrolysis nitro ethane gives

A. methyl amine and hydroxyl amine

B. acetic acid and methyl amine

C. acetic acid and hydroxyl amine

D. formic acid and hydroxyl amine

Answer: C



52. When nitro methane condenses with acetone it gives

A. 1-nitro-2-propanol

B. 1-nitro-2-methoxy propane

C. 1-nitro-2-methoxy-ethanol

D. 1-nitro-2-methoxy-propanol

Answer: D



53. Which gives an yellow oil with nitrous acid?

A. $1^\circ\,$ amine

B. 2° amine

C. 3° amine

D. quarternary salt

Answer: B

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54. Alkyl cyanide on reduction with $LiAlH_4$ ether gives,

A. amine

B. tertiary amine

C. primary amine

D. 2° amine

Answer: C

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55. An organic compound 'A' with molecular formula C_3H_9N neigher give nitrogen nor yellow oily liquid with HNO_2 . Then A is

A. $CH_3CH_2 - CH_2 - NH_2$

B. $CH_3 - CH_3 - NH - CH_3$

C. 📄

D. All of these

Answer: C

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56. The test for aniline is

A. Fehling's test

B. Ferric chloride test

C. Bromination test

D. Riemer Tiemann test

Answer: C



57. Chloropicrin (CCl_3NO_2) is used as

A. dyes

B. pharmaceuticals

C. explosives

D. soil sterilizing agents

Answer: D

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58. $C_6H_5CHO + Methylamine \rightarrow X$

 $Y + H_2 SO_4
ightarrow p - {
m amino \ benzene \ sulphonic \ acid \ } .$

Then X and Y are

A. Schiff's Base, $C_6H_5NH_2$

 $\mathsf{B.}\, C_6H_6,\, C_6H_5CHO$

C. benzal -N-methyl amine, Toluene

Answer: A



59. Which among the following is a tertiary nitro compound?





 $\mathsf{C.}\,CH_3CH_2CH_2-NO_2$



Answer: A

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60. $CH_3NO_2+6[H] \stackrel{X\,?}{\longrightarrow} CH_3NH_2+2H_2O$. 'X' is

A. Sn/HCl

B. Zn/NH_4Cl

C. Glucose + NaOH

D. $Na_3A_5O_3/NaOH$

Answer: A

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61. $CH_3CH_2NO_2 \xrightarrow{SnorZn/HCl} X + 2H_2O.$ Identify 'X'.

A. CH_3NH_2

B. CH_3CONH_2

 $\mathsf{C.}\,CH_3CH_2NH_2$

D. CH_3CH_2CONH

Answer: C



62. Tertiary amine is less bsic than secondary amine because of

A. delocalisation of π electrons

B. resonance effect

C. inductive effect

D. steric effect

Answer: D

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63. An organic compound C_7H_9N anwers carbylamine reaction but produces primary alcohol with nitrous acid. The compound may be

A. p-toludine

B. o-toluidine

C. benzylamine

D. m-toludine

Answer: C



64. Which of the following nitro-compounds behave as an scid in

the presence of strong alkaki?

A. Primary

B. Secondary

C. Tertiary

D. Both A and B

Answer: A

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65. Which of the following will not undergo Hoffman's bromamide reaction?

A. Ethanamide

B. Propanamide

C. Methanamide

D. Phenyl methanamide



67. Aniline react with benzoyl chloride in the presence of sodium

hydroxide and gives benzanilide. This reaction is known as

A. Gattermann reaction

B. Sandmeyer's reaction

C. Schotten -Baumann reaction

D. Gomberg -Bachmann reaction

Answer: C



68. The product obtained when nitrobenzene is treated with Zn/NaOH is

A. aniline

B. azoxybenzene

C. azobenzene

D. hydrazobenzene

Answer: D



69. Which one of the following is the most basic?

A. Ammonia

B. Methylamine

C. Dimethylamine

D. Aniline

Answer: C



70. Methyl isocyanide on reduction using $LiAlH_4$ gives :

A. Methyl amine

B. Ethyl amine

C. Dimethyl amine

D. Trimethyl amine

Answer: C

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71. The reagent that cannot be used for the conversion of $C_6H_5NO_2
ightarrow C_6H_5NH_2.$

A. Sn/HCl

B. $LiAlH_4$

 $\mathsf{C}.\,H_2/Ni$

D. Zn/NaOH

Answer: D



'C' is

A. C_6H_5-OH

 $\mathsf{B.}\, C_6H_5-CH_2OH$

 $\mathsf{C.}\,C_6H_5-CHO$

$\mathsf{D.}\, C_6H_5NH_2$

Answer: A



73. Assertion (A) : Methyl cyanide has higher boiling point than methyl isocyanide.

Reason (R) : Dipole moment of methyl cyanide is higher than that of methyl iso cyanide.

A. (A) and (R) are true and (R) is the correct explaination of

(A)

B. Both (A) and (R) are true but (R) does not explain (A)

C. (A) is true but (R) is false

D. Both (A) and (R) are false

Answer: A View Text Solution

74. Choose the incorrect statement among the following

A. Primary amine show intermolecular bond

B. Isopropyl amine is a secondary amine

C. Amines are basic in nature

D. Tertiary amines do not show intermolecular hydrogen

bonds

Answer: B

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1. Write down the possible isomers of the $C_4H_9NO_2$ give their

IUPAC names.

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2. There are two isomers with the formula CH_3NO_2 . How will

you distinguish between them?

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3. What happens when

2 - Nitropropane boiled with HCl.



4. What happens when

Nitrobenzene undergo electrolytic reduction in strongly acidic

medium.

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5. What happens when

Oxidation of tert - butylamine with $KMnO_4$.



6. What happens when

Oxidation of acetoneoxime with trifluoroperoxy acetic acid.

7. How will you convert nitrobenzene into

1,3,5 - trinitrobenzene
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8. How will you convert nitrobenzene into
o and p- nitrophenol
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9. How will you convert nitrobenzene into

m- nitro aniline

10. How will you convert nitrobenzene into

azoxybenzene



13. How will you convert nitrobenzene into

aniline

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14. Identify compounds A,B and C in the following sequence of reactions.

$$C_6H_5NO_2 \xrightarrow{Fe/HCl} A \xrightarrow{HNO_2}_{273K} B \xrightarrow{C_6H_5OH} C$$

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15. Identify compounds A,B and C in the following sequence of

reactions.

$$C_6H_5N_2Cl \stackrel{CuCN}{\longrightarrow} A \stackrel{H_2O/H^+}{\longrightarrow} B \stackrel{NH_3}{\longrightarrow} C$$



16. Identify compounds A,B and C in the following sequence of

reactions.

$$CH_3CH_2I \stackrel{NaCN}{\longrightarrow} A \stackrel{OH^-}{\longrightarrow} B \stackrel{NaOH + Br_2}{\longrightarrow} C$$

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17. Identify compounds A,B and C in the following sequence of

reactions.

$$CH_3NH_2 \stackrel{CH_3Br}{\longrightarrow} A \stackrel{CH_3COCl}{\longrightarrow} B \stackrel{B_2H_6}{\longrightarrow} C$$

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18. Identify compounds A,B and C in the following sequence of

reactions.

$$C_6H_5NH_2 \xrightarrow{(OH_3CO)_2O} A \xrightarrow{HNO_3} B \xrightarrow{H_2O/H^+} C$$



21. Write short notes on Hofmann's bromide reaction .





25. Write short notes on Carbylamine reaction .

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26. Write short notes on the following			
Maustard oil reaction			
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27. Write short notes on the following Coupling reaction			
O Watch Video Solution			

28. Write short notes on the following

Diazotisation



30. How will you distinguish between primary secondary and

tertiary alphatic amines.



31. Why aniline does not undergo Friedel - Crafts reaction ?



32. According for the following

Diazonium salts of aromatic amines are more stable than those

of aliphatic amines.



33. According for the following

 pK_b of aniline is more than that of methylamine.



34. According for the following

Gabriel phthalimide synthesis is prefered for synthesising

•	•
primary	amines.

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35. Ethylamine is soluble in water whereas aniline is not soluble
in water. Watch Video Solution

36. Amines are more basic than



37. Can the amino group, in the aniline molecule, become meta-

directing in an electrophilic substitution reaction ?

38. Arrange the following.

In increasing order of solubility in water, $C_6H_5NH_2, (C_2H_5)_2NH, C_2H_5NH_2$

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39. Arrange the following.

In increasing order of basic strength

aniline, p - toludine, P - nitroaniline



40. Arrange the following.

In increasing order of basic strength

$C_{6}H_{5}NH_{2}, C_{6}H_{5}NHCH_{3}, C_{6}H_{5}NH_{2}, P-Cl-C_{6}H_{4}-NH_{2}$

C	Watch V	ídeo	Solution

41. Arrange the following.

In increasing order of basic strength in gas phase $(C_2H_5)CH_2, (C_2H_5)NH, (C_2H_5)_5N$ and NH_3 .

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42. Arrange the following.

In increasing order of boiling point $C_6H_5OH, (CH_3)_2NH, C_2H_5NH_2$

43. Arrange the following.

In decreasing order of the pK_b values $C_2H_5NH_2, C_6H_5NHCH_3, (C_2H_5)_2NH$ and CH_3NH_2 Watch Video Solution

44. Arrange the following.

Increasing order of basic strength

 $C_6H_5NH_2, C_6H_5N(CH_3), (C_2H_5)_2NH \text{ and } CH_3NH_2$

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45. Arrange the following.

In decreasing order of basic strength





46. How will you prepare propan - 1- amine from

Butane nitrile

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47. How will you prepare propan - 1- amine from

Propanamide

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48. How will you prepare propan - 1- amine from

1 - nitropropane





N-nitrosodiethylamine

52. Identify A,B and C






56. A Dibromo derivative (A) on treatment with KCN followed by acid hydrolysis and heating gives a monobasic acid (B) along with liberation of CO_2 . (B) on heating with liquid ammonia followed by treating with Br_2/KOH gives (C) which on treating with $NaNO_3$ and HCl at low temprerature followed by oxidation gives a monobasic acid (D) having molecular mass 74. Identify A to D.

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57. Identify A to E in the following frequency of reactions.



2. Write all possible isomers for the following compounds.

 $C_3H_7 - NO_2$

3. Find out the product of the following reactions.

 $CH_3CH(Cl)COOH \stackrel{(i) NaNO_2}{\longrightarrow} ?[X]$

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4. Find out the product of the following reactions.

 $CH_3CH_2Br + NaNO_2 \stackrel{\mathrm{alcohal}/ riangle}{\longrightarrow} [Y].$



5. Predict the major product that would be obtained on nitration of the following compounds.



6. Predict the major product that would be obtained on nitration of the following compounds.



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7. Predict the major product that would be obtained on nitration

of the following compounds.



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8. Draw the stucture of the following compounds.

Neopentylamine



9. Draw the stucture of the following compounds.

Tert - butylamine

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10. Draw the stucture of the following compounds.

lpha - amino propionaldehyde

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11. Draw the stucture of the following compounds.

Tribenzylamine



12. Draw the stucture of the following compounds.

N- ethyl - N - methylhexen - 3 - amines



15. Give the correct IUPAC names for the following amines.



1. Amines have.....odour.

A. carbolic

B. mustard oil

C. fishy

D. garlic

Answer: C

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2. The major product obtained when aniline is treated with acetic anhydride in the the presence of pyridine is......

- A. p bromo acetanilide
- B. p bromo aniline
- C. acetanilide
- D. 2,4,6 tribromo aniline

Answer: A



3. Amine that cannot be prepared by Gabriel Phthalimide synthesis is.....

A. benzyl amine

B. methyl amine

C. ethyl amine

D. phenyl amine

Answer: D

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4. Primary amine reacts with $CHCl_3/KOH$ the product formed

is.....and the reactions is called.....reaction.

A. cyanide, carbyl amine

B. iso cyanide, carbylamine

C. isocyanide, mustard oil

D. cyanide, mustard oil

Answer: B

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5. Aniline treated with H_2SO_4 forms.....

A. S-diphenyl urea

B. acetanilide

C. m-amino benzene sulphonic acid

D. sulphanilic acid

Answer: D

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6. The product formed when an aldehyde reacts with a primary

amine is.....

A. carboxylic acid

B. Carbylamine

C. schiff's base

D. Azodyes

Answer: C

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7. The compound which gives an oily nitrosomine on reaction

with nitrous acid at low temperature is......

A. C_2H_5NH

 $\mathsf{B.}\,(CH_3)_2NH$

 $\mathsf{C}.\,(CH_3)_3N$

D. $(CH_3)_2 CHNH_2$

Answer: B



8. The isomerism exhibited by 1-nitro butane and 1-nitro-2-methyl

propane is.....

A. chain

B. tautomerism

C. position

D. functional

Answer: B



9. p -amino phenol is the product of readucing nitrobezene

in.....

A. acid medium

B. basic medium

C. electrolytic reduction

D. neutral medium

Answer: C

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10. Sandmeyer's reaction occurs in the presence of......and.....and....

A. cuprous halide and haloacid

B. cupric halide and halogen

C. copper-zinc couple and H_2

D. None of these

Answer: A

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11.
$$CH_{3}CONH_{2} \stackrel{Br_{2}/KOH}{\longrightarrow} CH_{3}Nh_{2} + CO_{2}$$

This is.....reaction.

A. Hoffman bromamide

B. HVZ

C. Gabriel reaction

D. Gomberg Bachmann

Answer: A

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12. The nitro group can be reduced to primary amino group by.....

A. $Sn/\operatorname{con.}HCl$

B. Zn dust

C. Zn/NH_4Cl

D. Zn/NaOH

Answer: A

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13. Conversion of benzene diazonium chloride to chlorobezene is

called.....

A. Sandmeyer's reaction

B. Stephen's reaction

C. Gomberg reaction

D. Schotten - Baumann reaction

Answer: A

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14. The intermidiate formed on electrolytic reduction of nitrobenzene in conc. H_2SO_4 is.....

A. $C_6H_5NH - NHC_6H_5$

 $\mathsf{B.}\, C_6H_5-NHOH$

 $\mathsf{C}.\, C_6H_5-N=N-C_6H_5$

D. All of these



15. The compound that does not show tautomerism is.....

A. nitrobenzene

B. nitro methane

C. nitro ethane

D. 2-nitro propane

Answer: A



16. Nitro-acinitro tautomerism is exhibited by.....

A. nitro methane

B. nitrobenzene

C. chloropicrin

D. o-toluidine

Answer: A

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17. CCl_3NO_2 is used as.....

A. soil sterilizing agent

B. organic synthesis

C. good solvent

D. antioxidant



18. The reaction between benzene diazonium chloride and benzene in the presence of NaOH is.....

A. Perkin's reaction

B. Gattermann's reaction

C. Sandmeyer reaction

D. Gomberg-Bachmann reaction

Answer: D

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19. Use of chloropicrin is as.....

A. explosive

B. dye

C. anaesthetic

D. sterilizing agent

Answer: D

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20. Oil of mirbane is.....

A. nitrobenzene

B. benzaldehyde

C. methyl salicylate

D. aspirin

Answer: A



21. Oxidation of aniline with acidified potassium dichromate gives

A. p-benzo quinone

B. benzoic acid

C. benzaldehyde

D. benzyl alcohol

Answer: A



22. The tertiary nitro compound is.....

A. 2-nitropropane

B. 1-nitropropane

C. 1-nitro-2,2-dimethyl propane

D. 2-nitro-2-methyl propane

Answer: D

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23. The intermediate formed in the nitration of benzene is.....

A. Arenium ion

B. Carbanion

C. Oxinium ion

D. Nitrite ion

Answer: A

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24. The basic sterength of amine is in the order of.....

A.
$$NH_3 > CH_3NH_2 > (CH_3)_2NH_3$$

B. $(CH_3)_2 NH > CH_3 NH_2 > NH_3$

 $\mathsf{C.}\,CH_3NH_2>(CH_3)_2NH>HN_3$

D.
$$NH_3 > (CH_3)_2 NH > CH_3 NH_2$$

Answer: B

1. Assertion (A) : Aniline reacts with Br_2/H_2O to give a white precipitate of 2,4,6-tribromoaniline.

Reason (R): Aniline is resonance stabilized.

A. (A) and (R) are true and (R) is the correct explaination of

(A)

B. Both (A) and (R) are true but (R) does not explain (A)

C. (A) is true but (R) is false

D. Both (A) and (R) are false

Answer: B



2. Assertion (A) : Aniline couples with $NaNO_2/HCl$ to give β -Naphthol which is a blue colour precipitate.

Reason (R) : The colour of the compound is due to conjugation of the ring.

A. (A) and (R) are true and (R) is the correct explaination of

(A)

B. Both (A) and (R) are true but (R) does not explain (A)

C. (A) is true but (R) is false

D. Both (A) and (R) are false

Answer: D



3. Assertion (A) : Methyl cyanide has higher boiling point than methyl isocyanide.

Reason (R) : Dipole moment of methyl cyanide is higher than that of methyl iso cyanide.

A. (A) and (R) are true and (R) is the correct explaination of

(A)

B. Both (A) and (R) are true but (R) does not explain (A)

C. (A) is true but (R) is false

D. Both (A) and (R) are false

Answer: A



4. Assertion (A) : Aniline is less basic than NH_3 .

Reason (R) : Greater the value of pK_b , stronger is the base.

A. (A) and (R) are true and (R) is the correct explaination of

(A)

B. Both (A) and (R) are true but (R) does not explain (A)

C. (A) is true but (R) is false

D. Both (A) and (R) are false

Answer: C



5. Assertion (A) : Ethyl amine gives positive carbylamine and mustard oil reaction.

Reason (R) : Ethyl amine is a 1° amine.

A. (A) and (R) are true and (R) is the correct explaination of

(A)

B. Both (A) and (R) are true but (R) does not explain (A)

C. (A) is true but (R) is false

D. Both (A) and (R) are false

Answer: A

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6. Assertion (A) : Aniline decolourises bromine water with the formation of white precipitate which is TNT.

Reason (R): Electrophilic substitution reaction proceed under

milder conditions in aniline because of the activated benzene ring.

A. (A) and (R) are true and (R) is the correct explaination of

(A)

B. Both (A) and (R) are true but (R) does not explain (A)

C. (A) is true but (R)is false

D. Both (A) and (R) are false

Answer: C

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Incorrect Statements

1. Pick out incorrect statement with respect to methyl nitrite

A. They are also called Pseudo acid

B. High electrical conductivity

C. Decolourises $FeCl_3$ solution

D. Dissolves in NaOH instantly

Answer: C

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2. Which among the following is the incorrect statement regarding the acidic nature of nitroalkanes?

A. 1° and 2° nitroalkanes show acidic character

B. nitroalkane dissolve in NaOH to give salt

C. aci form is more acidic than intro form

D. acidity decreases due to the - I effect of alkayl groups

Answer: D View Text Solution

3. Choose the incorrect statement among the following

A. Primary amine show intermolecular bond

B. Isopropyl amine is a secondary amine

C. Amines are basic in nature

D. Tertiary amines do not show intermolecular hydrogen

bonds

Answer: B

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1. Give the IUPAC name of the following compounds.



3. Define tautomerism.



4. Which among the following nitro compound will not exhibit

tautomerism and why?

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5. Explain the conversion of benzene to nitrobenzene.

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6. Two isomers having the molecular formula $C_2H_5NO_2$ on hydrolysis in acidic medium gives (A) and (B). Identify (A) and (B) and explain the reactions involved.



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8. Complete the following :





9. Complete the following :



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10. Nitrobenzene does not undergo Friedel - Crafts reaction -

Give reason.

View Text Solution 11. Write about the functional isomerism of nitro methane? Watch Video Solution

12. Give the reduction of nitromethane in (a) acid medium, (b)

neutral medium.



13. Explain the reduction of nitro methane in different medium.


17. Give the structural formula of (i) 2-amino-2-methyl propane.

(ii) 2-(N, N-dimethyl) amino butane

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18. What happens when acetonitrile is hydrolysed?			
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19. Write the name and structure of four isomeric amines having			
the molecular formula $C_3H_9N.$			



20. How will you distinguish between ethylamine and diethylamine?



23. How will you distinguish between aniline and ethylamine.

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24. Explain why aniline is less basic than methylamine?

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25. How is benzene diazonium chloride prepared?
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26. Explain why amines are more basic than amides.
View Text Solution

27. When benzamide is treated with bromine and alkali gives compound A. Also when benzamide is reduced by $LiAlH_4$,

compound B is formed. Find A and B. Write the equations.

View Text Solution 28. $C_6H_5CH_2NH_2 \xrightarrow{HNO_2} A \xrightarrow{(O)} B \xrightarrow{Zn/Hg} C$ Identify A,B and C. **View Text Solution 29.** An aromatic primary amine A with molecular formula C_6H_7N undergoes diazotisation to give B.B when treated with hypophosphorous acid gives C. Identify A, B and C.



30. An aromatic simplest nitro compound A on reduction using Sn and HCl gives B. B undergoes carbylamine reaction. Identify A and B. Give any one use of compound A.



31. An organic compound A of molecular formula C_2H_5ON treated with bromine and KOH gives B of molecular formula CH_5N . Identify A and B. Write the equation involved.



33. $C_6HG_5NH_2 \xrightarrow{HNO_2} A \xrightarrow{KCN} C_{u_2(CN)_2} B \xrightarrow{H_2O} C$

Identify A, B and C.

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34. An aromatic hydrocarbon A on nitration gives B which is known as oil of mirbane. B on warming with con. H_2SO_4 gives compound C. Identify A,B and C.



$$\textbf{36.} \ C_6H_5CONH_2 \xrightarrow[KOH]{Br_2} X \xrightarrow[HOl]{HO2} Y \xrightarrow[Cu^+]{NaNO_2} Z$$

Identify X,Y and Z.

37.
$$C_2H_3N \xrightarrow{LiAl/H_4} B \xrightarrow{HNO_2}$$

Identify A,B and C.

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38. Identify the compounds A,B,C in the following reactions:

$$CH_3CONH_2 \stackrel{Br_2}{\longrightarrow} A \stackrel{HNO_2}{\longrightarrow} B \stackrel{\operatorname{Red} \mathrm{P}}{\underset{l_2}{\longrightarrow}} C$$

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39. Identify the compounds A,B,C in the following reactions:

 $C_2H_5Cl \stackrel{AgCN}{\longrightarrow} A \stackrel{Sn/HCl}{\longrightarrow} B \stackrel{HNO_2}{\longrightarrow} C$

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40. Bring aout the conversion of Chlorobenzene to aniline.
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41. Find 'X' and 'Y'

 $CH_3Br \stackrel{NaN_3}{\longrightarrow}$ 'X' $\stackrel{LiAlH_4}{\longrightarrow}$ 'Y'?

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44. Aniline reacts with bromine water to give a white precipitate.

Give the chemical formula of the white precipitate formed in the above reaction.



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45. Aniline reacts with bromine water to give a white precipitate.

Write the chemical reaction involved.

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46. Write a note on Libermann's nitroso test.
View Text Solution
47. Can aniline be prepared by Gabriel phthalimide reaction?
Justify your answer.
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48. Use the following reagents in the correct order and bring about the conversion of benzene to aniline.

Reagents

 $HCl. NH_3$, heat, alk $KMnO_4, CHCl_3/AlCl_3, Br_2/KOH$

that can be used.



49. Write a note on Sandmeyer reaction.

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50. Write short notes on Gomberg reaction.

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51. How is anisole prepared from benzene diazonium chloride?

52. How c	does benzened	liazonium chl	loride react wi	ith

KI

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53. How does benzenediazonium chloride react with

KCN



54. How does benzenediazonium chloride react with

$NaNO_2/Cu^+$



55. How is flurobenzene prepared from benzenediazonium chloride? Name the reaction.



56. Write a reaction to show the conversion of acetic acid to benzoic acid.

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57. Complete the following reaction



59. Write a note on Thrope nitrile conddensation.



60. Give the IUPAC name of follwing compounds



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63. How is prepared from N-alkyl formamide?



66. Write a note on addition reactions of alkyl isocynides.

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67. What is 'spirit of nitre'? Give its use.



Identify the product Y and give its IUPAC name.

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71. How will you convert diethylamine into N,N- diethylacetamide
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72. How will you convert diethylamine into

N-nitrosodiethylamine





1. Write a note on acidic nature of nitroalkanes.

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2. How are nitro alkanes prepared for the following alkyl bromides
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3. How are nitro alkanes prepared for the following

methane

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4. How will convert nitrobenzene to benzoic acid?



the equations.



7. Explain Levine and hauser acetylation.





- 8. Outline the preparation of
- (a) para nitroaniline from aniline
- (b) tri bromo aniline from aniline



9. How can the following conversion be effected?

Nitrobenzene to anisole.

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10. How can the following conversion be effected?

Chloro benzene to phenyl hydrazine

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11. How can the following conversion be effected?

Aniline to bezoic acid.

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12. How can the following conversion be effected?

Benzene diazonium chloride to ehtyl benzene.



13. Aniline reacts with Br_2/H_2O to give a tribromo derivative.

How would you convert aniline to get a monobromo derivate?

14. How are the following compounds obtained from benzene

diazonium chloride ?

(i) phenol, (ii) ester, (iii) p-hydroxy azo benzene.

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15. How are the following compounds obtained from benzene

diazonium chloride ?

(i) phenol, (ii) ester, (iii) p-hydroxy azo benzene.

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16. How are the following compounds obtained from benzene

diazonium chloride ?

(i) phenol, (ii) ester, (iii) p-hydroxy azo benzene.





19. Starting from the following reagents, how will your prepare ethanenitrile.

 CH_3Br

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20. Starting from the following reagents, how will your prepare ethanenitrile.

 CH_3CONH_2



21. Starting from the following reagents, how will your prepare ethanenitrile.

 $CH_3CH = NOH$



22. An organic compound 'A' on reduction gives compound 'B' which on reaction with trichloromethane and caustic potash forms 'C'. Compound 'C' on catalytic reduction given N-methyl benzenamine. Identify A,B and C write the reaction involved.



2. Draw a flow chart to show classification of nitro compound

giving examples for each type.



3. Account for the following :

 $(CH_3)_2 NH$ is stronger base than NH_3 .



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6. Account for the following :

On sulphonation of aniline, p-amino benzene sulphonic acid is

formed.

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7. What happens when ethylamine is treated with

 $CHCl_3/NaOH$

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8. What happens when ethylamine is treated with

 CS_2

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9. What happens when ethylamine is treated with

 $C_6H_5CHO.$

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 10. Account for

 Reduction of CH_3CN gives $CH_3CH_2NH_2$ while

 CH_3NC gives $(CH_3)_2NH.$



11. Account for

 $(CH_3)_2 NH$ requires two moler proportion of $CH_2 I$ to give the same crystalline product formed by $(CH_3)_2 N$ with one mole of $CH_3 I$.

12. Account for

Nitration of aniline with $conc.HNO_3$ may end up with same

meta nitro product.

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13. Account for

p-toluidine is a stronger base than p-nitroaniline.

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14. Account for the following :

Nitroethane is soluble NaOH.

15. Account for the following :

Nitroethane reacts with nitrous acid.

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16. Account for the following :

2-methyl-2-nitro propane has neither of the properties.



17. How can the following conversion be effected?

Nitrobenzene \rightarrow Nitrosobenzene.



18. How can the following conversion be effected?

Nitrobenzene \rightarrow Azoxybenzene.





21. How do primary, secondary and tertiary amines react with nitrous acid?

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22. Write a note on
Carbylamine reaction.
View Text Solution
23. Write a note on
Mustard oil reaction.
View Text Solution

24. Write a note on

Acetylation of benzylamine.



Diazotisation reaction.



27. Outline the mechanism of

Nitration of aniline.


1. How can the following conversion be effected?

Nitrobenzene to anisole.



2. How can the following conversion be effected?

Chloro benzene to phenyl hydrazine



3. How can the following conversion be effected?

Aniline to bezoic acid.



4. How can the following conversion be effected?

Benzene diazonium chloride to ehtyl benzene.

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Match Mideo Colution

5. How will you distinguish between primary secondary and tertiary alphatic amines.

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Long Answers
1. Write short notes on Carbylamine reaction .



2. Write short notes on the following

Maustard oil reaction

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3. Write a note on

Acetylation of benzylamine.

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4. Write a note on

Formation of Schiff's base.

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5. Write a note on

Diazotisation reaction.



6. Aniline does not undergo Friedel - Crafts reaction. Explain.

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7. Account for the following

Diazonium salts of aromatic amines are more stable than those

of aliphatic amines.



8. Account for the following

 pK_b of aniline is more than that of methylamine.

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9. Write short notes on the following
Gabriel phthalimide synthesis
Vatch Video Solution
10. Ethylamine is soluble in water whereas aniline is not . Give reason.

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11. Account for the following

Amines are more basic than amides.



12. Account for the following

Although amino group is o - and p - directing in aromatic

electrophilic substitution reactions, aniline on nitration gives a

substantial amount of m - nitroaniline.

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