

BIOLOGY

BOOKS - KUMAR PRAKASHAN KENDRA BIOLOGY (GUJRATI ENGLISH)

BIOMOLECULES

Section A Exam Oriented Questions Answers
From Darpan

1. How the living organisms are formed based on chemical organization?



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2. How the analysis of chemical components can be done?



3. Briefly describe the types of groups participating in the structure of amino acids.



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4. Give the information, regarding structure of lipids.



5. What are heterocyclic organic compounds? Explain with examples.



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6. What are primary and secondary metabolites ? State its importance and components with examples.



7. What are macro and micro biomolecules ? Explain with examples.



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8. Give information of structural organization of proteins. Mention their general functions.



9. Give information regarding structural organization and functions of polysaccharides ?



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10. Give information regarding nucleic acids.



11. Describe primary, secondary and tertiary structure of proteins. OR Describe structure of proteins.



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12. Which types of linkage, formed between different monomers, to form polymer? Explain the various types of linkage with Examples.



13. What are metabolic processes ? State its importance.



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14. What is metabolic basis for life? Discuss.



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15. "The living state is a non-equilibrium steady state." Explain.





16. Give information regarding nucleic acids.



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17. How does chemical processes occur ?

Which kind of changes are observed in it?



18. How the chemical transformation occurs through enzymes at higher rate ?



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19. Describe the mechanism of enzyme action



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20. Discuss the factors affecting enzyme activities.



21. How is naming of enzymes is done?

Describe types of enzymes with examples.



22. What are co-factors? Describe its types.



Section B Difference Scientific Reasons Give Differences 2 Marks

1. Purine and Pyrimidine



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2. Nucleoside and Nucleotide



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3. DNA and RNA



Section B Difference Scientific Reasons Give Scientific Reasons 2 Marks

1. Life is not possible without protein.



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2. Though molecular weight of lipid is less, it is included in Biomacromolecules.





Section C Definition Explanation Terms Full Name Definition Explanation 1 Mark

1. Give information regarding nucleic acids.



2. Illustrate a glycosidic, peptide and a phospho-diester bond.



3. Illustrate a glycosidic, peptide and a phospho-diester bond.



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4. Illustrate a glycosidic, peptide and a phospho-diester bond.



Section C Definition Explanation Terms Full Name Full Name 1 Mark

1. ATP:



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2. DNA and RNA



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3. DNA and RNA



Section D Textual Exercise

1. What are macromolecules ? Give Example.



2. Illustrate a glycosidic, peptide and a phospho-diester bond.



3. What is meant by tertiary structure of proteins?



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4. Find and write down structures of 10 interesting small molecular weight biomolecules. Find if there is any industry which manufactures the compounds by isolation. Find out who are the buyers.



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5. Proteins have primary structure. If you are given a method to know which amino acid is at either of the two termini (ends) of a protein, can you connect this information to purity or homogeneity of a protein?



6. Find out and make a list of proteins used as therapeutic agents. Find other applications of

proteins (e.g. Cosmetics etc.)

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7. Explain the composition of triglyceride.



8. Can you describe what happens when milk is converted into curd or yoghurt, from your understanding of proteins?



9. Can you attempt building models of biomolecules using commercially available atomic models (Ball and Stick models).



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10. Attempt titrating an amino acid against a weak base and discover the number of dissociating (ionizable) functional groups in the amino acid.



11. Draw the structure of the amino acid, alanine.



12. What are gums made of ? Is Fevicol different?



13. Find out a qualitative test for proteins, fats and oils, amino acids and test any fruit juice, saliva, sweat and urine for them.



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14. Find out how much cellulose is made by all the plants in the biosphere and compare it with how much of paper is manufactured by man and hence what is the consumption of plant material by man annually. What a loss of vegetation!



15. Describe the important properties of enzymes.



Section E Solution Of Ncert Exemplar Multiple Choice Questions Mcqs

1. It is said that elemental composition of living organisms and that of inanimate objects

(like earth's crust) are similar in the sense that all the major elements are present in both.

Then what would be the difference between these two groups? Choose a correct answer from among the following.

- (A) Living organisms have more gold in them than inanimate objects.
- (B) Living organisms have more water in their body than inanimate objects.
- (C) Living organisms have more carbon, oxygen and hydrogen per unit mass than inanimate objects.

- (D) Living organisms have more calcium in them than inanimate objects.
 - A. Living organisms have more gold in them than inanimate objects.
 - B. Living organisms have more water in their body than inanimate objects.
 - C. Living organisms have more carbon, oxygen and hydrogen per unit mass than inanimate objects.

D. Living organisms have more calcium in them than inanimate objects.

Answer: C



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2. Many elements are found in living organisms either free or in the form of compounds. One of the following is not found in living organisms.

(A) Silicon

(B) Magnesium
(C) Iron
(D) Sodium
A. Silicon B. Magnesium
C. Iron
D. Sodium
Answer: A
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3. Amino acids have both an amino group and a carboxyl group in their structure. Which amongst the following is an amino acid?

- A. Formic acid
- B. Glycerol
- C. Glycolic acid
- D. Glycine

Answer: D



- **4.** An amino acid under certain conditions have both positive and negative charges simultaneously in the same molecule. Such a form of amino acid is called
- (A) Acidic form
- (B) Basic form
- (C) Aromatic form
- (D) Zwitterionic form
 - A. acidic form
 - B. basic form
 - C. aromatic form

D. zwitterionic form

Answer: D



- **5.** Which of the following sugars have the same number of carbon as present in glucose ?
- (A) Fructose
- (B) Erythrose

- (C) Ribulose(D) Ribose
 - A. Fructose
 - B. Erythrose
 - C. Ribulose
 - D. Ribose

Answer: A



- **6.** An acid soluble compound formed by phosphorylation of nucleoside is called
- (A) Nitrogen base
- (B) Adenine
- (C) Sugar phosphate
- (D) Nucleotide
 - A. nitrogen base
 - B. adenine
 - C. sugar phosphate
 - D. nucleotide

Answer: D



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- 7. When we homogenise any tissue in an acid, the acid soluble pool represents
- (A) Cytoplasm
- (B) Cell membrane
- (C) Nucleus
- (D) Mitochondria

A. cytoplasm

- B. cell membrane
- C. nucleus
- D. mitochondria

Answer: A



- 8. The most abundant chemical in living organisms could be
- (A) Protein
- (B) Water

(C) Sugar (D) Nucleic acid A. protein B. water C. sugar D. nucleic acid **Answer: B Watch Video Solution** **9.** A homopolymer has only one type of building block called monomer repeated 'n' number of times. A heteropolymer has more than one type of monomer. Proteins are heteropolymers usually made of

- A. 20 types of monomers
- B. 40 types of monomers
- C. 30 types of monomers
- D. only one type of monomer

Answer: A

10. Proteins perform many physiological functions. For example, some functions as enzymes. One of the following represents an additional function that some proteins discharge

- A. Antibiotics
- B. Pigment conferring colour to skin
- C. Pigments making colours of flowers

D. Hormones

Answer: D



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11. Glycogen is a homopolymer made of

A. glucose units

B. galactose units

C. ribose units

D. amino acids

Answer: A



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12. The number of 'ends' in a glycogen molecule would be

A. Equal to the number of branches plus one

- B. Equal to the number of branch points
- C. One

D. Two, one on the left side and another on the right side

Answer: A



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13. The primary structure of a protein molecule has

A. two ends

B. one end

C. three ends

D. no ends

Answer: A



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14. Which of the following reactions is not enzyme mediated in biological system?

A. Dissolving CO_2 in water

B. Unwinding the two strands of DNA

- C. Hydrolysis of sucrose
- D. Formation of peptide bond

Answer:



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Section E Solution Of Ncert Exemplar Very Short Answer Type Questions Vsq

1. Medicines are either man made (i.e., synthetic) or obtained from living organisms

like plants, bacteria, animals, etc. and hence, the latter are called natural products. Sometimes, natural products are chemically altered by man to reduce toxicity or side effects. Write against each of the following whether they were initially obtained as a natural product or as a synthetic chemical.

(a) Penicillin (b) Sulphonamide (c) Vitamin-C(d) Growth hormone



- 2. Select an appropriate chemical bond among ester bond, glycosidic bond, peptide bond and hydrogen bond and write against each of the following.
 - (a) Polysaccharide
- (b) Protein
- (c) Fat
- (d) Water



3. Write the name of any one amino acid, sugar, nucleotide and fatty acid.



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4. Reaction given below is catalysed by oxidoreductase between two substrates A and A' complete the reaction

A reduced + A' oxidised $\,
ightarrow$



5. How are prosthetic groups different from co-factors ?



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6. Glycine and alanine are different with respect to one substituent on the α -carbon. What are the other common substituent groups ?



7. Starch, cellulose, glycogen, chitin are polysaccharides found among the following.

Choose the one appropriate and write against each.

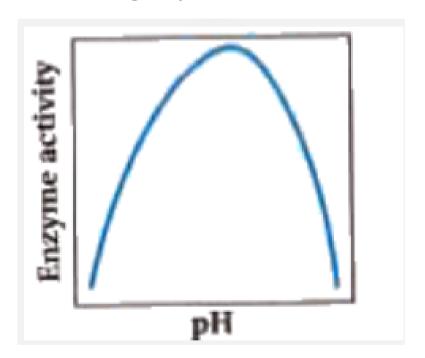
- (a) Cotton fibre
- (b) Exoskeleton of cockroach
- (c) Liver
 - (d) Peeled potato



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Section E Solution Of Ncert Exemplar Short **Answer Type Questions**

1. Enzymes are proteins. Proteins are long chains of amino acids linked to each other by peptide bonds. Amino acids have many functional groups in their structure.



These functional groups are many of them at least, ionisable. As they are weak acids and

bases in chemical nature, this ionisation is influenced by pH of the solution. For many enzymes activity is influenced by surrounding pH. This is depicted in the curve below, explain briefly.



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2. Is rubber a primary metabolite or a secondary metabolite? Write four sentences about rubber.



3. Describe primary, secondary and tertiary structure of proteins. OR Describe structure of proteins.



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4. Nucleic acids exhibit secondary structure, justify with example.



5. Comment on the statement 'living state is a non-equilibrium steady state to be able to perform work'.



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Section E Solution Of Ncert Exemplar Long Answer Type Questions

1. Formation of Enzyme substrate complex (ES) is the first step in the catalysed reactions.

Describe the other steps till the formation of product.



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2. What are different classes of enzymes? Explain any two with the type of reactions they catalyse.



3. Nucleic acids exhibit secondary structure. Describe through Watson-Crick model.



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4. What is the difference between a nucleotide and nucleoside? Give two examples of each with their structure.



5. Describe various forms of lipid with a few examples.



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Questions From Module Important Mcq For Neet

1. In which of the following part, organic substances present ?

A. Acid soluble

- B. Retenant
- C. Filterate
- D. Both (A) and (C)

Answer: D



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2. Separate, appropriate for analysis of Ca and

Mg.

Statement X : Water is evaporated from living organs.

Statement $Y: O_2$ is removed from dry parts.

Statement Z : Various inorganic elements are present in the ash.

A. X, Y true, Z wrong

B. X, Z true, Y wrong

C. X, Z wrong, Y true

D. X, Y wrong, Z true

Answer: B



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3.	Phospholipid	named	lecithin	is	found	in							
component of													
(A) Plasmalemma													
(B	(B) Cell wall												
(C) Middle lamella													
(D) Nuclear membrane													
	A. Plasmalemma B. Cell wall												
	C. Middle lamella												
	D. Nuclear membrane												

Answer: A



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4. Isolate in appropriate manner for secondary metabolites. a.Plant b.Fungi c.Bacteria d.Above

A. Plant

B. Fungi

C. Bacteria

D. Above all

Answer: D



- **5.** State the name of polymer of fructose.
- (A) Starch
- (B) Glycogen
- (C) Inulin
- (D) Chitin
 - A. Starch
 - B. Glycogen

- C. Inulin
- D. Chitin

Answer: C



- **6.** Each step of DNA is.....angle from next step.
- (A) 20°
- (B) 34°

(C) 10°

(D) 36°

A. 20°

B. 34°

C. 10°

D. 36°

Answer: D



7. Science involved in soure of energy development, form and time.....

A. Energy science

B. Exchange science

C. Redox science

D. Bio energy science

Answer: D



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- 8. What is transition stage?
 - A. Formation of new structure of product
 - B. Formation of new structure of substrate
 - C. Formation of new structure of enzyme
 - D. Formation of new intermediate molecule

Answer: B



9.	Give	the	name	of	cofactor	for	carboxy				
peptidase.											
(A) Zn											
(B) Cu											
(C) Mn											
(D) Mg											
	A. Zn										
	B. Cu										
	C. Mr	า									
	D. Mg	3									

Answer:



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Question Paper Answer The Following Questions Briefly Each Of 1 Mark

1. What are macro and micro biomolecules ? Explain with examples.



2. What is the structure of starch?



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3. State the location of phosphodiester in polynucleotide.



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4. How many molecules of organic acids formed in the presence of carbonic anhydrase





5. What is competitive inhibitor?



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Question Paper Give Answer As Per In Structures

Each Of 2 Marks

1. Write short note: Lecithin



2. Write the steps of inducer molecule.



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Question Paper Give Answer As Per In Structions Each Of 3 Marks

1. Describe concentration of substrate.



2. Give information of structural organization of proteins. Mention their general functions.



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Question Paper Describe In Detail

1. Explain chemical analysis.

