

CHEMISTRY

BOOKS - CHHAYA CHEMISTRY (BENGALI ENGLISH)

PREVIOUS YEARS QUESTION PAPER 2019

Wbchse 2019 Part A

1. KBr undergoes 80% dissociation in its 0.5(M) aqueous solution. Calculate osmotic pressure of the solution at $27^{\circ}\,C$ temperature.





2. What is azeotrophic mixture? Can it be considered as ideal solution?



3. What is peptisation? Give one example.



4. Mention two differences between bleaching action of SO_2 and Cl_2 .

5. Which of the halogen hydracids forms bi-salts ? Write reasons.



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6. Which type of isomerism is known by the two complexes:

 $\begin{bmatrix} Co(NH_3)_5SO_4 \end{bmatrix}Br$ and $\begin{bmatrix} Co(NH_3)_5Br \end{bmatrix}So_4$? How would you ditinguish between two complexes ?



7. Write the name of monomer units of dacron and mention one use of it.



8. Which type of stiochiometric defects is showns by AgBr crystal?



9. Chromium (atomic mass = 52) metal has body - centred cubic structure. The radius of chromium atom is 124.3 pm. Calculate the density of chromium metal.



10. What do you mean by packing fraction of cubic unit cell?



11. Calculate packing fraction of a face-centred cubic unit cell .



12. What is the vapour pressure of pure water at $100^{\circ}C$ temperature ?



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13. Write cell reaction and calculate electrical work obtained from the following galvanic cell at standard condition . $Mg |Mg^{2+}(aq)| |Ag^+(aq)| Ag$ [Given :

$$E^0_{Mg^{2+}\,|Mg}=\ -\ 2.36vo<\ \ {
m and}\ \ E^0_{Ag^{+}\,|Ag}=\ 0.80vo<\ .$$



14. Which type of ores is concentrated by froth floatation process?



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15. Which of the following are coloured in aqueous solution ? Give reasons : $V^{3\,+}$, $Cu^{\,+}$, $Sc^{3\,+}$, $Fe^{3\,+}$



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16. Which of the following oxides are amphoteric?

 $Mn_2O_7, CrO_3, Cr_2O_3, V_2O_5$



17. What happens when MnO_4^{2-} ion is kept in acid medium ? Write balanced equation.



18. Write R/S configuration of and and



19. Write structural formula of A for the reaction





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20. Distinguish between propanone and pentan-3-one by a suitable chemical test.



21. Write the structural formulae of A,B,C,D in the reaction





22. Write equation of the following conversion . Salicylic and aspirin.



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23. Write structural formulae of products obtained in the reaction





24. Write structural formulae of products obtained in the reaction





25. Mention one use of Lucas' reagent.



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26. Convert aniline to fluorobenzene.



27. Write structural formula of the compound A to D:





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28. An organic compound A(C_3H_8O) on treatment with Cu dust at 573K gives B. B does not reduce Fehling's solution but gives a yellow precipitate of compound C with $I_2/NaOH$. Deduce the structures of A , B and C and their IUPAC names.



29. The following compound is an example of peptide.



- (r) Write Zwitterion structures of the compound.
- (s) How many amino acids will be obtained on hydrolysis of the compound?



30. Write equation for the reaction of glucose with periodic acid.



31. Draw the graph of half-life period $(t_{1/2})$ versus initial concentration of reactant $(A)_0$ for a zero order reaction. Given reasons in favour of your answer.



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32. The rate of a reaction at 400 K is 10 times than the rate of the reaction at 200 K. Calculate activation energy of the reaction.



33. The unit of rate constant of a chemical reaction is L^2 . mol^{-2} . s^{-1} . Calculate order of the reaction.



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34. The rates of a first order reaction after 10mins and 20 mins from the commencement of the reaction are $0.04mol.\ L^{-1}.\ s^{-1}$ and $0.03mol.\ L^{-1}.\ s^{-1}$ respectively. Calculate half-life period of the reaction.



35. What is the shape of XeF_4 molecule? Write the chemical equation of the reaction of XeF_4 and KI?



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36. What is Nessler's reagent? Where is it used?



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37. IF_7 can be prepared but BrF_7 cannot be prepared . Why ?



38. Write structural formula of the compounds A to F.

$$C_6H_5CONH_2 \stackrel{Br_2\,/\,KOH}{-\!\!\!\!-\!\!\!\!-\!\!\!\!-\!\!\!\!-} A \stackrel{Br_2\,/\,H_2O}{-\!\!\!\!\!-\!\!\!\!-\!\!\!\!-\!\!\!\!-} B$$



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39. Write structural formula of the compounds A to F.





40. Write structural formula of the compounds A to F.

$$(CH_3)_3C-CHO \xrightarrow{ ext{conc.NaOH}} E+F$$



41. Convert the Benzaldehyde to cinnamic acid.

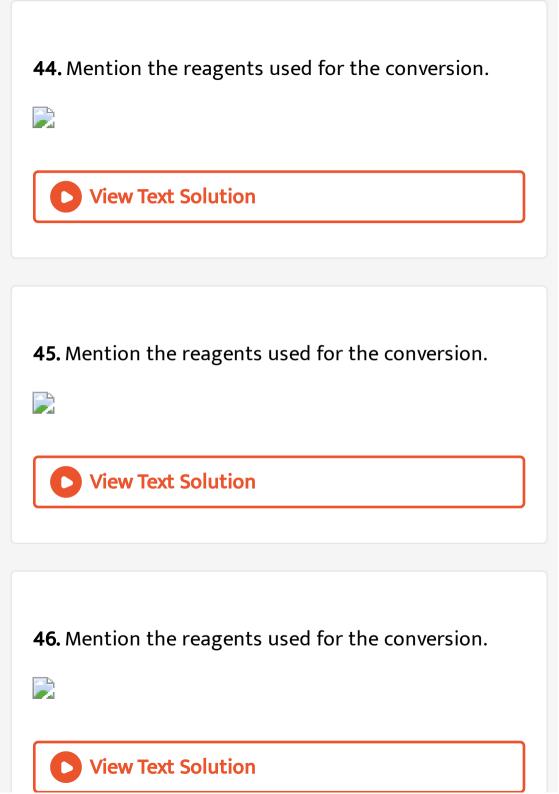


42. Convert the Acetic acid to Acetaldehyde.



43. What is the reason for the reducing property of formic acid? Give example of its reducing property.





47. Mention the reagents used for the conversion.





48. Mention the reagents used for the conversion.





49. Mention the reagents used for the conversion.





Wbchse 2019 Part B

1. Which of the following is strongly attracted by magnetic field /

A. Paramagnetic

B. Diamagnetic

C. Ferromagnetic

D. Antiferromagnetic

Answer: C

- **2.** Charge carried by 1 mole PO_4^{3-} ions is
 - A. 96500 C
 - B. 32167 C
 - C. 298500 C
 - D. 193000 C

Answer: A



| 3. Which of the following is lyophilic colloid? |
|--|
| A. Milk |
| B. Blood |
| C. Gold sol |
| D. Gum |
| Answer: A |
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| |

4. Which of the following is strongest base?

| A. NH_3 |
|--|
| B. PH_3 |
| C. AsH_3 |
| D. BiH_3 |
| |
| Answer: C |
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| |
| 5. Which of the following transition elements show $+7$ oxidation state ? |
| A. Cr |
| |

C. Fe

D. Ni

Answer: B



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6. In which of the following metal ions hybridisation state in sp^3d^2 /

A.
$$\left[Ni(CN)_4
ight]^2$$

B.
$$\left[Fe(CN)_6
ight]^{-4}$$

C. $\left[Co(NH_3)_6
ight]^{3+}$

D. $\left[FeF_6
ight]^{3}$

Answer: C



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7. Which of the following is most reactive in $S_N 2$ reaction?

A. CH_3Cl

B. CH_3CH_2Cl

 $\mathsf{C}.\left(CH_{3}\right)_{2}CHCl$

D. $(CH_3)_3$ CCl

Answer: B



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8. Which of the following cannot be identified by iodoform test?

A. CH_3CH_2Cl

 $B. \, CH_3CH(OH)CH_3$

C. $CH_3CH_2CH_2OH$

 $\mathsf{D.}\, CH_3CHO$

Answer: C

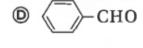


9. Which of the following compounds does not respond to aldol condensation reaction?

A.
$$CH_3CHO$$

B.
$$CH_3COCH_3$$

C.
$$CH_3CH_2CHO$$



D.

Answer: C

| 10. Which | of the f | following | bases | is | not | present | in | RNA |
|------------------|----------|-----------|-------|----|-----|---------|----|-----|
| | | | | | | | | |

- A. Adenine
- B. Guanine
- C. Thymine
- D. Uracil

Answer: D



| 11. Which of the following bases is not condesation |
|--|
| polymer ? |
| A. Bakelite |
| B. Nylon |
| C. Melamine-formaldehyde resin |
| D. Teflon |
| Answer: D |
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| |

12. Which of the following is antihistamine drug?

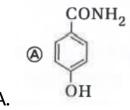
- A. Bromopheniramine
- B. Omiprazole
- C. Chloramphenicol
- D. Norethydrine

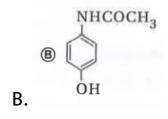
Answer: D

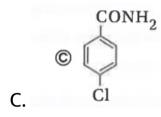


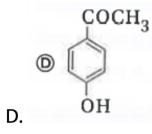
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13. Which of the following is paracetamol?









Answer: D



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14. Arrange $Li^+, Na^+, \ {
m and} \ Rb^+$ ions in the ascending order of their molar ionic conductance (





15. What is the unit of electro-chemical equivalent?



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16. By which property of colloid, we can distinguish between colloidal solution and true solution of same colour?



17. Why is magnetic moment of ferric compounds more than ferrous compounds?



18. Write general electronic configuration of actinoids



?

19. Why is BHA added in butter?

