



## CHEMISTRY

### BOOKS - TARGET CHEMISTRY (HINGLISH)

#### ALKANES

#### Classical Thinking

1. Hydrocarbons are the simplest organic compounds containing the elements \_\_\_\_\_.

- A. Carbon and oxygen
- B. carbon and hydrogen
- C. hydrogen and nitrogen

D. Hydrogen and oxygen

**Answer: B**

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2. The alkanes  $C_1$  to  $C_4$  are the constituents of \_\_\_\_\_.

A. LPG (Liquefied petroleum gas )

B. CNG (compressed natural gas )

C. LNG ( Liquefied natural gas )

D. all of these

**Answer: D**

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3. \_\_\_\_\_ is one of the components of mangoes .

A. Hexane

B. Cyclopentane

C. Octane

D. CyClohexane

**Answer: D**



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4. The hydrocarbons containing rings of carbon atoms and having properties similar to that of aliphatic are called \_\_\_\_\_ Hydrocarbons

- A. Alicyclic
- B. Aromatic
- C. unsaturated
- D. none of these

**Answer: A**



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5. Alkanes have open chain structure and carbon atoms are linked together by \_\_\_\_\_.

- A. single covalent bond triple covalent bond
- B. triple covalent bond
- C. single ionic bond

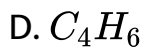
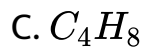
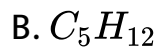
D. sigma and pi bond

**Answer: A**



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6. The molecular formula of butane is \_\_\_\_\_.



**Answer: A**



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7. Structural formula of an organic compound represents \_\_\_\_\_.

- A. Actual attachment of various atoms in a molecular
- B. Only number of various atoms in a molecule
- C. simplest ratio of number of atoms in a molecule
- D. the number of isomers of compound

**Answer: A**

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8. In methane carbon atom forms \_\_\_\_\_ C-H sigma bonds

- A. four

B. two

C. three

D. one

**Answer: A**



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9. The molecular geometry of each carbon atom in an alkane is \_\_\_\_\_.

A. octahedral

B. square planar

C. trigonal planar

D. tetrahedral

**Answer: D**



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**10.** Which of the following is CORRECT regarding propane ?

A. It contains two carbon - carbon single bonds and seven

C-H bonds.

B. It contains two carbon - carbon single bonds and eight

C-H bonds.

C. It contains two carbon - carbon single bonds and six C-

H bonds.

D. It contains two carbon - carbon single bonds and six C-

H bonds.



**Answer: B**



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11.  $\begin{array}{c} | \\ -CH \\ | \end{array}$  group in alkanes is called \_\_\_\_\_. Group .

A. carbene

B. methylene

C. isomethine

D. methine

**Answer: D**



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12. The straight chain are called \_\_\_\_\_ alkanes

A. branched

B. chain

C. normal

D. ring

**Answer: C**



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13. The minimum number of carbon atoms necessary for alkanes to form a branched structure is \_\_\_\_\_.

A. 4

B. 6

C. 3

D. 9

**Answer: A**



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**14.** In isoalkanes , one of the carbon atom is attached to \_\_\_\_  
other carbon atom (s) .

A. one

B. two

C. three

D. four

**Answer: C**



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15. The alkanes in which there are two single carbon branches on the second carbon atom of the parent chain are called \_\_\_ alkanes .

A. iso

B. neo

C. closed

D. ring

**Answer: B**



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16. \_\_\_\_\_ carbon atom is the one which is directly attached to two other carbon atoms .

A. Secondary

B. Tertiary

C. Primary

D. Quaternary

**Answer: A**



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17. Isomerism is the phenomenon in which two or more organic compounds have \_\_\_\_\_.

A. Same molecular formula but different structural formula

B. Same structural formula but different molecular formula

C. same general formula , but different structural formula

D. same empirical formula , same structure

**Answer: A**

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**18.** which of the following is the CORECT statement ?

A. Isomers have same physical and chemical properties .

B. isomers have same molecular formula

C. Isomical have same physical but different chemical properties

D. Isomers have different physical but same chemical properties .

**Answer: B**



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**19.** Which among the following is the NCORRECT pairing of alkanes with its corresponding number of isomers?

A. Butane -2- isomers

B. Pentane -3 isomers

C. Hexane -5 isomers

D. Heptane -8 isomers

**Answer: D**



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**20.** Isomers which can be interconverted through rotation around a single bond are

A. conformers

B. metamers

C. enantiomers

D. tautomers

**Answer: A**



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21. The energy required to rotate ethane molecule about the carbon bond is called \_\_\_ energy .

- A. potential
- B. Kinetic
- C. torsional
- D. Ionisation potential

**Answer: C**



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22. Which of the following statement is CORRECT ?

A. At any instant, 99% of ethane molecules have staggered and only about 1% have eclipsed conformation.

B. At any instant, 96% of ethane molecules have staggered and only about 4% have eclipsed conformation.

C. At any instant, 90% of ethane molecules have staggered and only about 10% have eclipsed conformation.

D. At any instant, 95% of ethane molecules have staggered and only about 5% have eclipsed conformation.

**Answer: A**



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**23.** In eclipsed conformation , the torsional angle of ethane is

\_\_\_\_\_.

A.  $60^\circ$

B.  $180^\circ$

C.  $0^\circ$

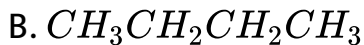
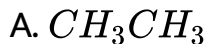
D.  $120^\circ$

**Answer: C**



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24. Which of the following is propane ?

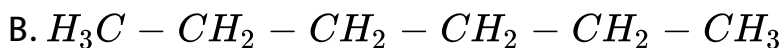


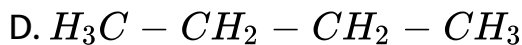
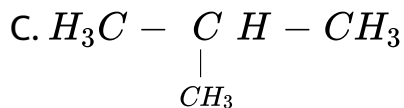
Answer: C



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25. Which of the following is n-hexane ?

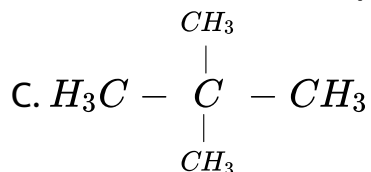
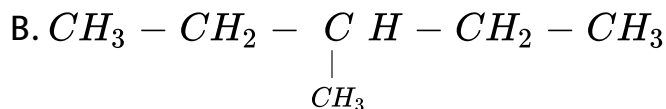
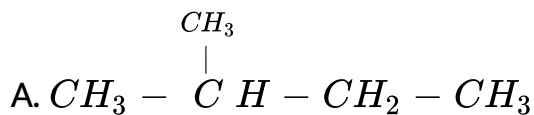




**Answer: B**

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**26.** Which of the following skeletal formulae represents 2-Methylpentane ?



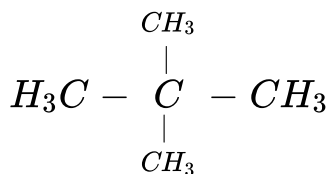


Answer: D



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27. IUPAC name of the following compound is \_\_\_.



- A. 1,2 Dimethylpropane
- B. 2,2-Dimethylpropane
- C. 2,3,-Dimethylpropane
- D. 2,2-Dimethylbutane

**Answer: B**



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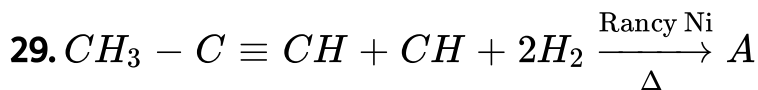
**28.** The IUPAC name of isobutane is \_\_\_\_\_.

- A. 2-Methylbutane
- B. Butane
- C. 2- Methylpropane
- D. 2,2-Dimethylpropane

**Answer: C**



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product 'A' in the given reaction is \_\_\_\_\_.

A. propene

B. Propane

C. ethane

D. butane

**Answer: B**



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30. The reaction in which there is removal of carbon dioxide from carboxylic acid is called \_\_\_\_\_.



- A. dehydrogenation
- B. dehalogenation
- C. dehydrohalogenation
- D. decarboxylation

**Answer: D**



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**31.** Alkyl halides on \_\_\_\_\_ with the Zn-Cu couple and alcohol ,  
give corresponding alkanes .

- A. Oxidation
- B. reduction
- C. hydrolysis

D. halogenation

**Answer: B**



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32. Alkyl halides on reduction with the Zn-Cu couple and alcohol, give corresponding alkanes. The order of reactivity of alkyl halides in the reaction is \_\_\_\_\_.

A.  $\text{R-Cl} > \text{R-Br} > \text{R-I}$

B.  $\text{R-Br} > \text{R-Cl} > \text{R-I}$

C.  $\text{R-Cl} > \text{R-I} > \text{R-Br}$

D.  $\text{R-I} > \text{R-Br} > \text{R-Cl}$

**Answer: D**



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33. In Wurtz reaction the reagent used is

A. barium

B. aluminium

C. sodium

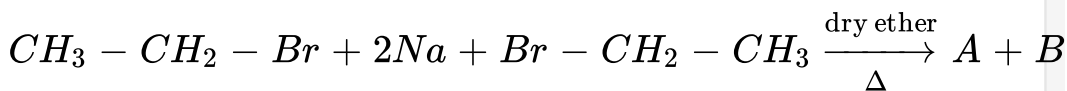
D. iron

Answer: C



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34. In the reaction



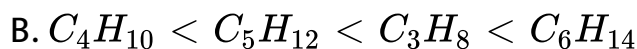
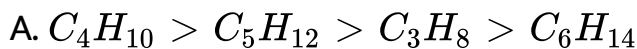
A and B are \_\_\_\_\_ respectively

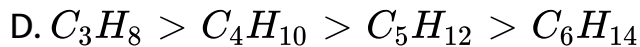
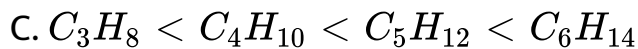
- A. n-butane and sodium bromide
- B. n-pentane and sodium bromide
- C. n-propane and sodium bromide
- D. ethane and sodium bromide

**Answer: A**

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**35.** Which of the following CORRECTLY represents the increasing order of melting point of the compounds ?





**Answer: C**

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**36.** The hydrocarbon which is a liquid at room temperature is:

A. butane

B. Propane

C. decane

D. ethane

**Answer: C**



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**37.** Alkanes mainly exhibit \_\_\_\_\_ reaction .

A. Substitution

B. elimination

C. addition

D. rearrangement

**Answer: A**



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**38.** Which of the following reaction of alkanes is explosive ?

- A. Iodination
- B. Bromination
- C. Fluorination
- D. Chlorination

**Answer: C**



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**39.** Methane reacts with excess of chlorine in diffused sunlight to give the final product as

- A. Carbon tetrachloride

B. chloroform

C. methylene dichloride

D. methyl chloride

**Answer: A**



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**40.** Reactivity of hydrogen atoms attached to different carbon atoms in alkane has the order:

A.  $2^\circ > 3^\circ > 1^\circ$

B.  $1^\circ > 2^\circ > 3^\circ$

C.  $3^\circ > 2^\circ > 1^\circ$

D.  $3^\circ < 1^\circ < 2^\circ$



**Answer: C**



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**41.** Nitration of alkanes involves replacement of one hydrogen atom by \_\_\_\_ group .

A. Amine

B. azo

C. nitro

D. cyanide

**Answer: C**



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42. Methane does NOT undergo \_\_\_\_\_.

A. chlorination

B. Bromination

C. nitration

D. combustion

**Answer: C**



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43. The nitration of higher alkanes involves substitution as well as \_\_\_\_\_.

A. C-H bond fissions

B. C-C bond fusions

C. C-H bond fusions

D. all of these

**Answer: A**



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**44.** The process in which higher hydrocarbons are broken down into lower hydrocarbons by controlled pyrolysis is called \_\_\_.

A. hydroysis

B. cracking

C. combustion

D. reduction

**Answer: B**



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**45.** When alkanes having more than five carbon atoms get cyclised to benzene and its homologues then the reaction is called \_\_\_\_\_.

A. hydrogenation

B. aromatization

C. reforming

D. both (B ) and (C )

**Answer: D**



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46. The dehydrogenation of n- Hexane in the presence of a catalyst gives \_\_\_\_\_.

- A. hexane and hydrogen
- B. pentene and hydrogen
- C. benzene and hydrogen
- D. hexyne and hydrogen

**Answer: C**



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47. The combustion of alkanes is a /an \_\_\_\_\_ reaction .

- A. exothermic
- B. endothermic
- C. aromatization
- D. all of these

**Answer: A**



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**48.** Alkanes ranging from \_\_\_\_\_ are used as lubricants ,

- A.  $C_5 - C_{10}$
- B.  $C_{10} - C_{15}$
- C.  $C_{17} - C_{20}$
- D.  $C_{21} - C_{30}$

**Answer: C**



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**49.** Incomplete combustion of alkanes in limited supply of air produces \_\_\_\_\_.

A. carbon black

B. lamp black

C. paraffin

D. alkenes

**Answer: A**



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50. \_\_\_\_\_ is used for the manufacture of printing ink .

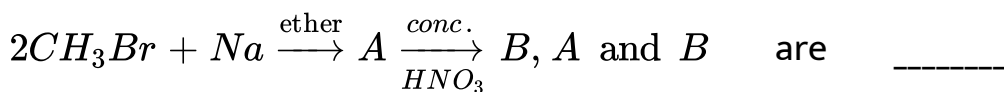
- A. Carbon black
- B. paraffin wax
- C. Lamp black
- D. Liquid alkanes

**Answer: A**



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51. In the reaction



respectively .



A. methane , nitromethane

B. ethane , nitroethane

C. nitromethane , methane

D. nitroethane , ethane

**Answer: B**



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**52.** Which of the following alkanes can be represented by only one structural formula ?

A. propane

B. Butane

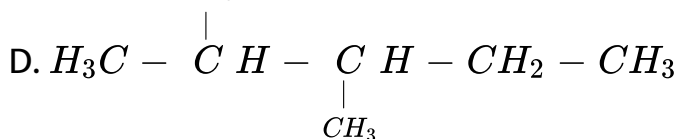
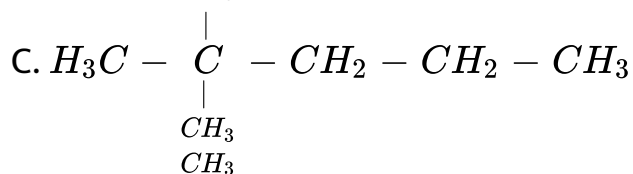
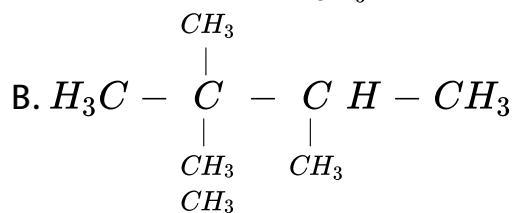
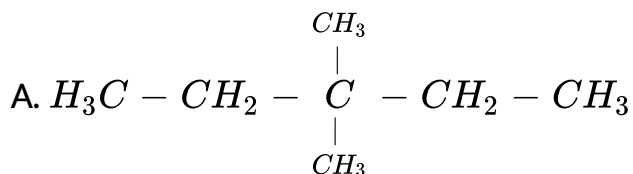
C. pentane

## D. Hexane

Answer: A

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### 53. 2,3-dimethylpentane



**Answer: D**



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54.  $C_2H_5I + 2[H] \xrightarrow{Zn - Cu / \text{alcohol}} X + HI$ . the product X is  
\_\_\_\_\_.

- A. ethane
- B. propane
- C. hexane
- D. butane

**Answer: A**



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55. \_\_\_\_\_ is formed as a byproduct in the chlorination of methane .

A. HBr

B. HI

C. HCl

D. HF

**Answer: C**



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56. In methane molecules , central carbon atom has \_\_\_\_\_ electrons in its valence shell .

A. 8

B. 6

C. 4

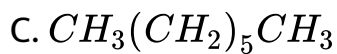
D. 10

**Answer: A**



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57. The condensed formula for n- Nonane is \_\_\_\_\_.



**Answer: A**



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58. \_\_\_\_\_ are saturated aliphatic hydrocarbons .

A. Alkenes

B. Alkanes

C. Alkynes

D. Alkyl groups

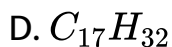
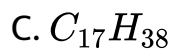
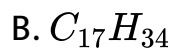
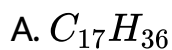
**Answer: B**



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## Critical Thinking

1. The molecular formula of the alkane with 17 carbon atoms is \_\_\_\_\_.



**Answer: A**



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2. The value of 'n' in a saturated hydrocarbon having the molecular formula  $C_nH_{10}$  is \_\_\_\_\_.

A. 2

B. 4

C. 5

D. 6

**Answer: B**



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3. In alkanes , the bond angle is \_\_\_\_\_.

A.  $109^\circ 28'$



B.  $109^\circ$

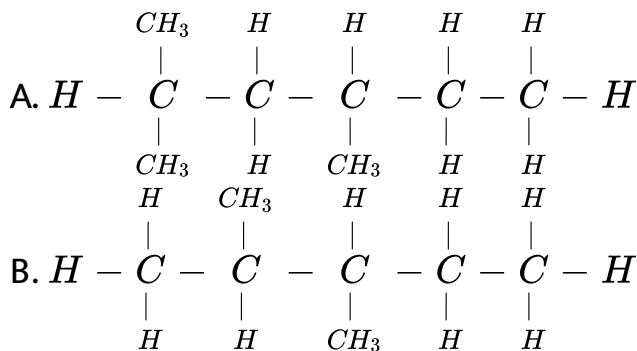
C.  $120^\circ$

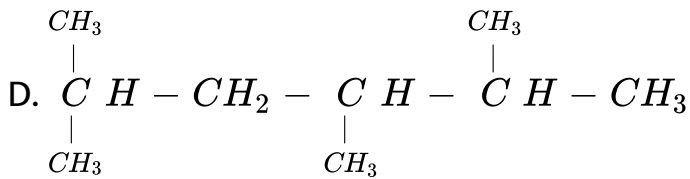
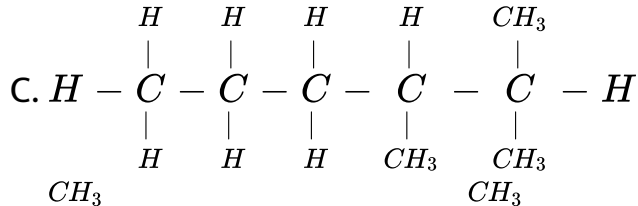
D.  $180^\circ$

Answer: A

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4. Which of the following CORRECTLY represents the structure of  $(CH_3)_2CHCH_2CH(CH_3)CH_2CH_3$ ?





**Answer: A**

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5. which of the following CORRECTLY represents bond -line formula of - n- butane ?

A. 

B. 

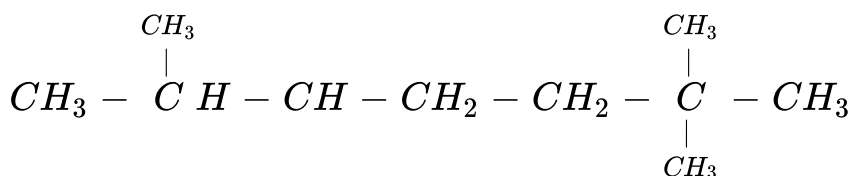
C. 

D. 

Answer: C

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6. The number of primary , secondary , tertiary and quaternary carbon atoms in the following alkane are \_\_\_\_\_ respectively



- A. 5,3,1 and 1
- B. 3,3,2 and 0
- C. 6,1,0 and 0
- D. 6,1,2 and 1

**Answer: A**



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7. In case of alkanes , the number of isomers \_\_\_\_\_ as the length of carbon chain increases.

A. decreases

B. increases

C. first increases then decreases

D. remains the same

**Answer: B**



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8. n-Butane and isobutane are examples of \_\_\_\_\_.

- A. chain isomers
- B. geometrical isomers
- C. position isomers
- D. tautomers

**Answer: A**



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9. Which of the following CORRECTLY represents chain isomers considering that all the valencies of carbon atoms are satisfied with hydrogen ?

A. 

B. 

C. 

D. 

**Answer: A**



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**10. Which of the following is NOT an isomer of hexane ?**

A. 2-Methylpentane

B. 2,2-Dimethylbutane

C. 2,2 -Dimethylpentane

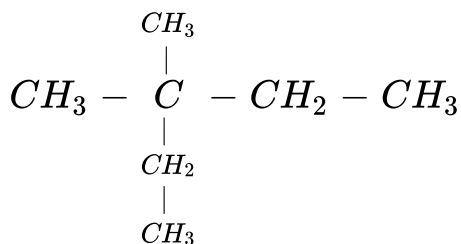
D. 3- Methylpentane

Answer: C



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11. IUPAC name of the following compound is \_\_\_\_.



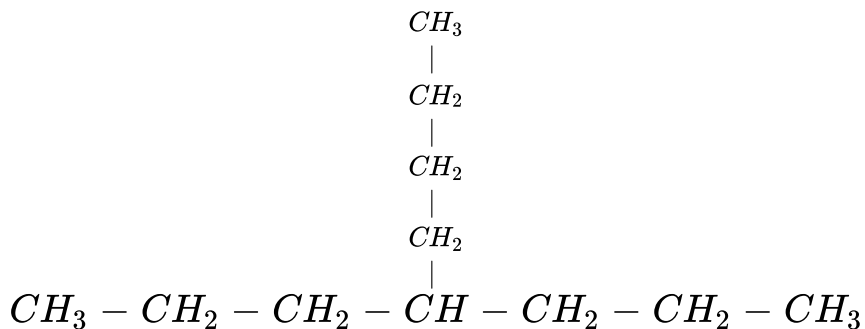
- A. 3- Ethyl-3-methylpentane
- B. 2-Ethyl-2-methylpentane
- C. 3,3-Dimethylpentane
- D. 3,3- Dimethylbutane

Answer: C



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12. Following compound represents \_\_\_\_\_.



- A. 4-propylheptane
- B. 4-Ethylcatane
- C. 4-propylcatne
- D. 4-propydecane

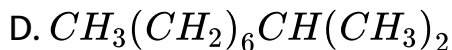
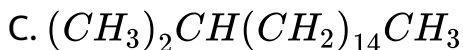
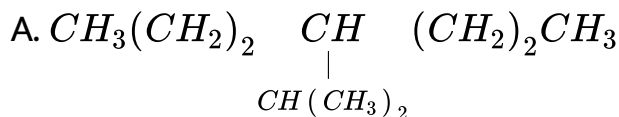
**Answer: C**



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13. In which of the following alkanes , parent chain contains 7 carbon atoms ?



Answer: A



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14. The IUPAC name of neohexane is \_\_\_\_\_

A. 2-Methylpentane

B. 2,3 -Dimethylhexane

C. 2,2-Dimethylbutane

D. 2,3 -Dimethylbutane

**Answer: C**

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15. The IUPAC name of following compound is \_\_\_\_\_.



A. 3,4 -Diethylhexane

B. 3,3-Diethylpropane

C. 3,4 -Dimethylhexane

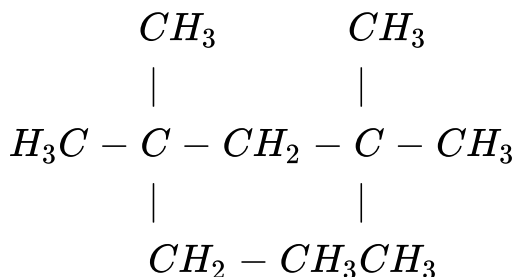
D. 3 – (1- Ethylpropyl) pentane

Answer: A



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16. Give the IUPAC name of following compound .



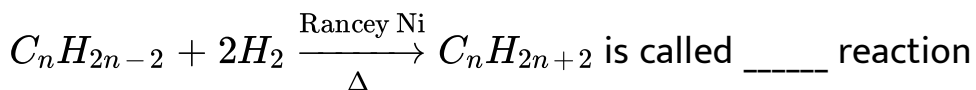
- A. 2,2,4,4-Tetramethylhexane
- B. 2,4-Tetramethylhexane
- C. 2,2,4-Tetramethylhexane
- D. 4,4-Tetramethylhexane

**Answer: A**



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**17. The reaction**



A. Sabatier and sanderson

B. Kolbe's

C. Friedel Craft 's

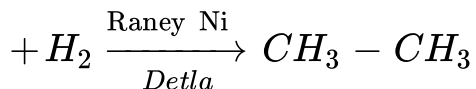
D. Cannizzaro 's

**Answer: A**



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18. Identify the reactant in the given reaction \_\_\_\_\_



- A. ethane
- B. ethene
- C. propyne
- D. propene

**Answer:**



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19. Which of the following salt on treatment with soda lime gives propane ?

- A. Sodium acetate
- B. Sodium ethanoate
- C. Sodium propanoate
- D. Sodium butanoate

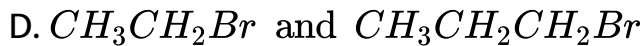
**Answer: D**



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**20.** To prepare a pure sample of n-hexane, using sodium metal as one reactant, the other reactant or reactants will be:

- A.  $CH_3CH_2$  and  $CH_3CH_2CH_2CH_2Br$
- B.  $CH_3Br$  and  $CH_3CH_2CH_2CH_2CH_2Br$



**Answer: C**

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21. which of the following has maximum boiling point ?

A. 

B. 

C. 

D. 

**Answer: A**





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22. The most volatile alkanes is \_\_\_\_\_

- A. n-pentane
- B. isopentane
- C. neopentne
- D. none of these

**Answer: C**



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23. Which catalyst is used in the bromination of alkanes ?

- A.  $Cr_2O_3$



B. Raney Ni

C.  $AlBr_3$

D.  $CaO$

**Answer: C**



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**24.** Arrange the following steps according to the order followed in the free radical mechanism for halogenation of alkanes ?

I. Chain initiation step

II. Chain termination step

III. Chain propagation step

A. III,I,II

B. I,II,III

C. I,III,II

D. III,II,I

**Answer: C**



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**25.** The nature of products formed in thermal decomposition of alkanes depends on \_\_\_\_.

A. Structure of alkanes

B. pressure

C. Pressure or absence of catalyst

D. all of these

**Answer: D**



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**26.** The complete oxidation of alkanes is known as \_\_\_\_\_.

A. combustion

B. halogenation

C. hydrogenation

D. pyrolysis

**Answer: A**



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27. Paraffin wax is a/ an \_\_\_\_\_.

- A. ester
- B. alcohol
- C. unsaturated hydrocarbon
- D. saturated hydrocarbon

**Answer: D**



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28. Constituents of paraffin wax contain solid alkanes ranging from \_\_\_\_\_.

- A.  $C_{11}$  to  $C_{20}$

B.  $C_{21}$  to  $C_{30}$

C.  $C_5$  to  $C_{15}$

D. all of these

**Answer: B**



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**29. which of the following is CORRECT ?**

A. The infinite number of intermediate conformations between staggered and eclipse conformations are called gauche conformation .

B. In neoalkanes , isopropyl group is present at one end of the normal chain of carbon atoms

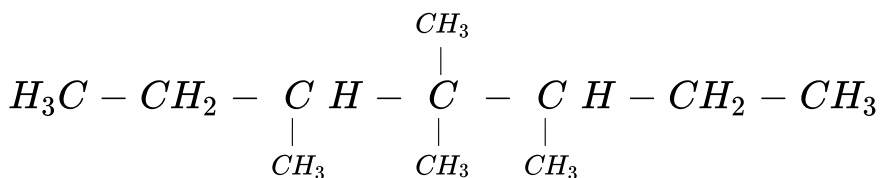
C. Methane is a non-polar molecules .

D. Raney nickel is an alloy cotaining equal amount of nickel and platinum .

**Answer: C**

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**30.** The given compound is



A. 3, 4, 4, 5 – Tetramethylheptnae

B. 4,4,5- Trimethylheptane

C. 3,3,4,5 -Tetramethylheptane

D. 3,4,5 -Trimethylheptane

**Answer: A**



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**31.** Normal hydrocarbon contains \_\_\_\_\_.

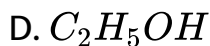
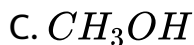
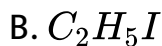
- A. Only primary carbon atoms with no branching
- B. either only primary or both primary and secondary carbon atoms with no branching
- C. either only primary or both primary and secondary carbon atoms with branching
- D. only primary carbon atom with branching

**Answer: B**



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**32.** Both methane and ethane can be prepared in one step from \_\_\_\_\_.



**Answer: A**



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33. A mixture of propene and methane is obtained by the cracking of

A. But -1- ene

B. But -2-ene

C. n-butane

D. propyne

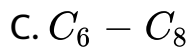
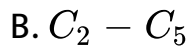
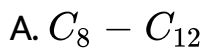
**Answer: C**



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**Competitive Thinking**

1. Gasoline has composition



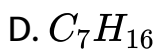
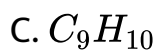
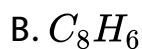
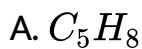
D. none of these

**Answer: C**



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2. which represents an alkane ?



**Answer: D**



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**3.** The correct statement the comparison of staggered and eclipsed conformations of ethane is:

A. The eclipsed conformation of ethane is more stable than staggered conformation even though the eclipsed conformation has torsional strain

B. The staggered conformation of ethane is more stable than eclipsed conformation because staggered conformation has no torsional strain

C. The staggered conformation of ethane is less stable than eclipsed conformation of ethane is less stable than eclipsed conformation because staggered conformation has torsional strain

D. The eclipsed conformation of ethane is more stable than staggered conformation , because eclipsed conformation has no torsional strain

**Answer: B**

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4. With respect to the conformers of ethane, which of the following statements is true ?

- A. Bond angle remains same but bond length changes .
- B. Bond angle changes but bond length remains same .
- C. Both bond angle and bond length changes
- D. Both bond angle and bond length remains same .

**Answer: D**



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5. Which one of the following contain isopropyl group

- A. 2,2,3,3-Tetramethylpentane
- B. 2-Methylpentane
- C. 2,2,3-Trimethylpentane
- D. 3,3- Dimethylpentane

**Answer: B**



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6. Propane is obtained from propene , by which of the following methods ?

- A. Wurtz reaction
- B. Dehydrogenation
- C. Frankland reaction
- D. Catalytic hydrogenation

**Answer: D**



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7. Dry distillation of sodium propanoate with soda lime gives

\_\_\_\_\_.

A. Propane

B. propene

C. ethane

D. ethene

**Answer: C**



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8. In Wurtz reaction the reagent used is

A. Na

B. Na/Liquid  $NH_3$

C. Na/ dry ether

D. Na/dry Alcohol

**Answer: C**



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9. Which one of the following cannot be prepared by Wurtz reaction ?

A.  $CH_4$

B.  $C_2H_6$

C.  $C_3H_8$

D.  $C_4H_{10}$



**Answer: A**



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**10.** When ethylbromide and propyl bromide are allowed to react with sodium in ether, they form

- A. one alkane
- B. two alkanes
- C. four alkanes
- D. three alkanes

**Answer: D**



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11. Methane can be prepared by:

- A. Wurtz's reaction
- B. decarboxylation
- C. hydrogenation reaction
- D. all of these

**Answer: B**



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12. Which of the following compounds has the highest boiling point ?

- A. n-Hexane

B. n-pentane

C. 2,2- Dimethylpropane

D. 2-Methylbutane

**Answer: A**



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**13.** Photochemical chlorination is initiated by a process of

A. pyrolysis

B. substituton

C. cracking

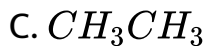
D. homolysis

Answer: D



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14. Which of the following is NOT formed by the reaction of  $Cl_2$  on  $CH_4$  in sunlight ?



Answer: D



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15. Which of the following reaction of alkanes is explosive ?

A. Fluorine

B. Chlorine

C. Bromine

D. Iodine

**Answer: A**



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16. Product obtained by nitration of propane is \_\_\_\_\_.

A. nitropropane

B. nitromethane

C. nitroethane

D. all of these

**Answer: D**

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17. Liquid hydrocarbon can be converted to a mixture of gaswous hydrocarbon by

A. Cracking

B. hydrolysis

C. Oxidation

D. distillation under reduced pressure

**Answer: A**



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**18.** Conversion of hexane into benzene involves the \_\_\_\_\_ reaction.

A. hydration

B. hydrolysis

C. hydrogenation

D. dehydrogenation

**Answer: D**



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19. Carbon black is carbon in a very finely divided state. It is used to make paints and printers ink. It is obtained by the thermal decomposition of

- A. acetylene
- B. benzene
- C. carbon tetrachloride
- D. methane

**Answer: D**



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20. Which of the following has maximum boiling point ?



- A. Isooctane
- B. n-Octane
- C. 2,2,3,3-Tetramethylbutane
- D. n-Butane

**Answer: B**



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**21. Which statement is NOT TRUE concerning alkanes ?**

- A. Large number of alkanes are soluble in water .
- B. All alkanes have a lower density than water
- C. At room temperature some alkanes are liquids, some are solids and some are gases

D. All alkanes burn

**Answer: A**



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22. A mixture of ethyl iodide and n-propyl iodide is subjected to Wurtz reaction. The hydrocarbon that will not be formed is

A. n-butane

B. n-propane

C. n-pentane

D. n-hexane

**Answer: B**



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23. Anhydrous sodium acetate on heating with soda lime gives \_\_\_\_\_.

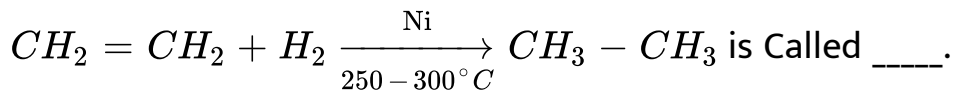
- A. acetic acid
- B. methane
- C. Calcium acetate
- D. ethane

**Answer: B**



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24. The reaction



- A. Wurtz's reaction
- B. Kolbe's reaction
- C. Sabatier and sanderson's reaction
- D. Carbylamine reaction

**Answer: C**

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25. The fourth higher homologue of ethane is \_\_\_\_\_.

- A. butane

B. pentane

C. hexane

D. heptane

**Answer: C**

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## Answers To Evaluation Test

1. The order of appearance of the following with rising temperature during the refining of crude oil is \_\_\_\_\_.

A. CNG , gasoline , diesel

B. Diesel , gasoline ,CNG

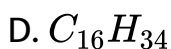
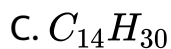
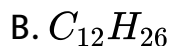
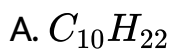
C. Gasoline , diesel ,CNG

D. Gasoline ,CNG ,diesel

**Answer: A**

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2. Which of the following CORRECTLY represents molecular formula of an alkane whose molecules weight was found to be 170 ?

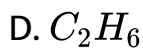
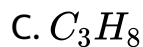
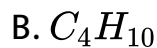


**Answer: B**



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3. What is the molecular formula of the alkane, the 5.6 litres of which weighed 11 g at STP ?



**Answer: C**



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4. In neohexane ,\_\_\_\_\_ secondary hydrogen atoms are present

A. zero

B. one

C. two

D. three

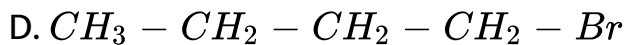
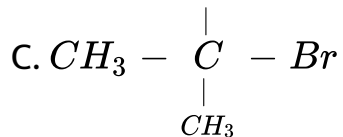
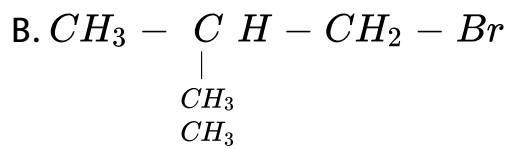
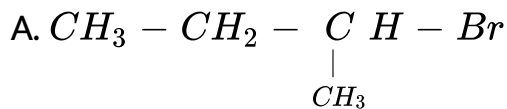
**Answer: C**



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5. What is the chief product obtained when n-butane is treated with  $Br_2$  in the presence of light at  $130^\circ C$ ?





**Answer: A**



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