

## CHEMISTRY

# BOOKS - GURUKUL BOOKS & PACKAGING CHEMISTRY (HINGLISH)

# JULY 2016

### Section I

**1.** Schottky defects are observed in which solid among the following?

A. Brass

B. Cesium chloride

C. Zinc sulphide

D. Stainless steel

#### Answer:

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2. No machine has an efficiency unity', is stated in

A. First law of thermodynamics

B. Second law of thermodynamics

C. Third law of thermodynamics

D. Hess's law of constant heat summation



**3.** Which among the following reaction is an example of a zero order reaction?

A. 
$$H_{2(g)} + I_{2(g)} o 2HI_{(g)}$$
  
B.  $2H_2O_{2(l)} o 2H_2O_{(l)} + O_{2(g)}$ 

#### **Answer:**



4. Potential of saturated colomel electrode is

A. 0.242 V

B. 1.1 V

C. 0.337 V

D. 0.28 V

Answer:

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**5.** which of the following compounds is used as a semipermeable membrane?

A. Potassium ferrocyanide

B. potassium argentocyanide

C. Sodium meta aluinate

D. Copper ferrocyanide

#### Answer:

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**6.** which among the following minerals does not contain aluminium?

A. Cryolite

**B. Siderite** 

C. China clay

D. Corundum



7. The group 15 element having inner electronic configuration

as of argon is

A. Phosphorous (z = 15)

B. Antimony (z = 51)

C. Arsenic (z = 33)

D. Nitogen (z = 7)

#### **Answer:**

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8. Write chemical reactions involved in Van Arkel method for

refining Titanium.





**9.** Explain the relationship between Gibb's standard energy change of the reaction and equilibrium constant.

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**10.** A face centred cube (FCC) consists of how many atoms? Explain.

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11. Describe isolation method in determination of rate law and

order of reaction.



**12.** Explain the following methods to protect metals from corrosion:

(a) Galvanization (b) Passivation

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13. Write the Nernst equation and explain the terms involved.

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14. what happens when dilute suphuric acid is treated with:

(a) Fe (b)  $CaF_2$  ?



**15.** Define :

(a) Osmosis (b) Freezing point

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**16.** The rate constant of a first order reaction are 0.58  $s^{-1}$  at 313 K and  $0.045s^{-1}$  at 293 K. what is the energy of activation for the reaction?

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17. Calculate the standard enthalpy of the reaction,

$$2C_{(\,{
m graphite})}\,+\,3H_{2\,(\,g\,)}\,
ightarrow\,C_{2}H_{6\,(\,g\,)}\,\Delta H^{\,\circ}\,=\,?$$

from the following  $\Delta H^{\,\circ}\,$  values:

$$C_2 H_{6\,(\,g\,)} \,+\, rac{7}{2} O_{2\,(\,g\,)} \, o\, 2 C O_{2\,(\,g\,)} \,+\, 3 H_2 O_{\,(\,1\,)} \,\Delta H^{\,\circ} \,=\, -\, 1560 k J$$

(b) 
$$H_{2(g)} + \frac{1}{2}O_{2(g)} \rightarrow H_2O_{(1)}\Delta H^\circ = -285.8kJ$$
  
(c)  $C_{(\text{graphite})} + O_{2(g)} \rightarrow CO_{2(g)}\Delta H^\circ = -393.5kJ$ 

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**18.** 3.795 g of sulphur is dissolved in 100g of  $CS_2$ . This solution boils at 319.81 K. What is the molecular formula of sulphur in solution? The boiling point of  $CS_2$  is 319.45 kJ (Given that  $K_b$  for  $CS_2 = 2.42 K k g mol^{-1}$  and atomic mass of S = 32)

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**19.** Write the reaction involved in large scale preparation of niric acid.

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**20.** State third law of thermodynamics. Write applications of standard molar entropy.

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21. The Conductivity of 0.02M Ag  $NO_3$  at  $25^\circ C$  is  $2.428 \times 10^{-3} \Omega^{-1} cm^{-1}$ . What is its molar conductivity?



**22.** Sodium acetate reacts with Ethanoyl chloride to form:

A. Acetic acid

B. Acetone

C. Acetic anhydride

D. Sodium formate

#### Answer:

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23. Natalite is a mixture of:

A. diethly ether and methanol

B. diethyl ether and ethanol

C. dimethyl ether and methanol

D. dimethyl ether and ethanol



24. What is the effective atomic number of Fe (z = 26) in  $Fe(CN)_6$ 

A. 12

B. 30

C. 26

D. 36

#### **Answer:**



25. Maltose is a :



26. Which one of the following oxidation state of Manganese is

unstable?

 $\mathsf{A.}+2$ 

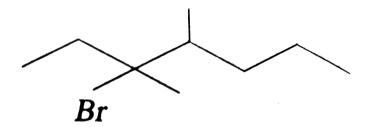
 $\mathsf{B.}+4$ 

C.+5

D.+7



27. IUPAC name of the following compounds is:



A. 3-Bromo-3,4 dimethylheptane

B. 3,4-dimethyl-3-bromoheptane

C. 5-Bromo-4,5-dimethylheptane

D. 4,5-dimethyl-5-bromoheptane



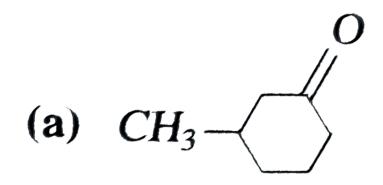
**28.** which of the following compounds is NOT prepared by the action of alcoholic  $NH_3$  an alkyl halide

A.  $CH_3NH_2$ B.  $CH_3-CH_2-NH_2$ C.  $CH_3-CH_2-CH_2-NH_2$ D.  $(CH_3)_3C-NH_2$ 

#### Answer:

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**29.** Write IUPAC names of the following compounds:





30. What are the sources of Vitamin C and Vitamin K?



31. Write four points of distinction between Lanthanoids and

Actinoids

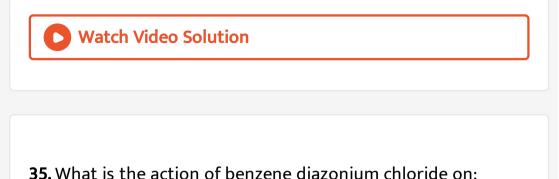
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<b>32.</b> How is Benzonitrile converted to Benzophenone?
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**33.** Write the formulae of the raw materials used for preparation of:

(a) Buna-S (b) Dextron

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34. Write a note on Sandmeyer's reaction



- (a) Phenol in alkaline medium
- (b) Aniline



**36.** Explain any two chemical methods of food preservation.



37. What is the action of the following reagents on glucose?

(a) bromine water (b) dilute  $HNO_3$ 

hydroxyl amine

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38. Define ligand. Write four postulates of Werner's theory.



**39.** What is metamerism?

Write the structure and IUPAC name of methyl-n-propyl ether.

What is action of hot HI on it?

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40. (a) How are the following polymers prepared?

- (1) Orlon
- (2) Teflon
- (b) Classify the following drugs into Analgesics and Antibiotics:
- (1) Ofloxacin
- (2) Morphine
- Ampicillin
- (4) Chloramphenical
- (c) Identify 'A' and 'B' and rewrite the reactions:

(1) 
$$(1) \xrightarrow{CH_3} NH_2 \xrightarrow{NaNO_2} (A) \xrightarrow{Cu_2Br_2} (B) + N_2 \uparrow (D)$$

- (2) [Math Processing Error]
- (3)

$$C_2H_5 - N^{+o}(CH_3)_3I^{\Theta} \xrightarrow{Arac{g_2\emptyset}{H_2}O} A \xrightarrow{\Delta} B + (CH_3)_3N + H_2O$$

**41.** How are the following conversions carried out ?

2-methylbutan-1ol-2methylbutanoic acid

Pheylethene into Benzoic acid

(3) Benzoic acid into metanitrobenzoic acid

What is the action of Benzene sulphonyl chloride on primary,

secondary and tertiary amines?

Write two uses of formaldehyde.

