



CHEMISTRY

PHYSICAL, INORGANIC, AND ORGANIC CHEMISTRY

ALKYL HALIDE, ALCOHOL, PHENOL, ETHER

ORGANIC CHEMISTRY(Alkyl Halide, Alcohol, Phenol, Ether)



dehydration with conc. H_2SO_4 predominantly forms



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2. A dextro- rotatory optically active alkyl halide undergoes hydrolysis by $S_N 2$ mechanism. The resulting alcohol is : A. Dextorotatory

B. Laevorotatory

C. Optically inactive due to racemization

D. May be dexture or laevorotatory

Answer: 2

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3. The reaction,



is an example of ,

- A. Wurtz reaction
- B. Wittig reaction
- C. Ullmann reaction
- D. Williamson reaction



4. Neopentyl bromide undergoes dehydro halogenation to give alkene even though it has no eta – hydrogen. This is due to :

A. E2 mechanism

B. by direct dehydrohalogenation

C. rearrangement of carbocation by E1

mechanism

D. E1cB mechanism

Answer: 3

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5. From Williamson's synthesis preparation of which

of the following is possible ?

A. Only symmetrical ethers

B. Only asymmetrical ethers

C. Both (1) and (2)

D. None of these

Answer: 3

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6. Ethyl alcohol is heated with conc. H_2SO_4 . The

product formed is :

A.
$$CH_3 = \overset{o}{\overset{|}{\underset{\scriptstyle |}{\underset{\scriptstyle |}{\underset{\scriptstyle |}{\underset{\scriptstyle |}{\atop\scriptstyle |}}}}} - OC_2H_6$$

B. $C_2 H_6$

C. $C_2 H_4$

D. C_2H_2

Answer: 3

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7. $CH_2 = CH_2 \xrightarrow{HBr} X \xrightarrow{Ag.KOH} Y \xrightarrow{Na_2CO_3} Z$. In the

above reaction sequence, Z is :

A. C_2H_5OH

B. C_2H_5I

C. CH_3CHO

D. CHI_3

Answer: D

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8. Methyl alcohol when reacted with carbon monoxide using cobalt or rhodium as catalyst , compound, 'A' is formed. On heating 'A' with HI in the presence of red phosphorus as catalyst 'B' is formed. Identify 'B'.

A. CH_3COOH

B. CH_3CHO

$\mathsf{C.}\,CH_3CH_2I$

D. CH_3CH_3

Answer: D



9. Which of the following is added to chloroform to slow down its aerial oxidationin presence of light ?

A. Carbonyl chloride

B. Ethyl alcohol

C. Sodium hydroxide

D. Nitric acid

Answer: B

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10. The catalyst used in the preparation of an alkyl chloride by the action of dry HCl on an alcohol is

A. Anhydrous $AlCl_3$

B. $FeCl_3$

C. Anhydrous $ZnCl_2$

 $\mathsf{D.}\, Cu$



11. Which of the following does not give a precipitate

with alcoholic $AgNO_3$?

A. Benzyl chloride

B. Chlorobenzene

C. Allyl chloride

D. t - butyl chloride.



12. Which of the following is the strongest nucleophile ?

A. $Br^{\,-}$

B. OH^{-}

C. $I^{\,-}$

D. $C_6H_5O^{\,-}$

Answer: 3

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13. 1- Phenyl ethanol can be prepared by the reaction of C_6H_5CHO with :

A. $CH_3Mgl,\,\,$ ether followed by hydrolysis

 $\mathsf{B.}\, C_2 H_5 Br + H_2 O$

C. $CH_3CH_2Br + {\sf Alc.}KOH$

D. $CH_2Br + Al_4C_3$

Answer: A



14. When alkyl halide is heated with dry Ag_2O . It

produces :

A. Ester

B. Ether

C. Ketone

D. Alcohol

Answer: 2

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15. When ethyl alcohol is oxidised with PCC, it gives :

A. CH_3COOH

 $\mathsf{B.}\,CH_3CH_2-O-CH_2CH_3$

$\mathsf{C}.\,CH_3CHO$

D. CH_3COCH_3

Answer: 3

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16. p - nitrophenol and o - nitrophenol are

separated by

A. Crystalization

B. Fractional crystalization

C. Distillation

D. Steam distillation

Answer: D

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17. Phosgene is a poisonous gas obtained in chloroform bottles, substance used to make it non-poisonous is :

A. Formic acis

B. Ethanol

C. Dichloro methane

D. CH_3COOH



Answer: A



19. $CHCl_3 + OH^- \rightarrow HCOO^-$, intermediate of

this reaction can be

- A. : CCl_3^-
- $\mathsf{B.}: CCl_2$
- C. Both (1) and (2)
- D. can not be predicted

Answer: C





20. Ethers can not be prepared by :

A. Heating alkyl halide with dry silver oxide

B. Heating alkyl halide with ammonical KOH

C. Heating alkyl halide with pottasium alkoxide

D. Heating an excess of alcohol with alumina

under pressure





A.
$$CH_3 - \displaystyle \underset{|_{CH_2}}{CH_2} CH - O - CH_2 - CH_3$$

- B. $CH_3 O \displaystyle \underset{CH_3}{CH_3} CH_2 CH_3$
- $\mathsf{C}.\,CH_3-\,CH-CH_2CH_2OH$ CH_3
- D. $CH_3 CH_2 CH_2 CH_2OH$ CH_2

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22. Which of thye following will be most reactive for E_1 reaction ? .



B.
$$C_{6}H_{5}-\overset{Br}{C}-Br$$

C.
$$CH_2=CH-\stackrel{\scriptscriptstyle Br}{c}H-CH_3$$

n

D.
$$CH_2 = CH - \displaystyle \mathop{C}\limits_{|_{CH_3}}^{_{Br}} - CH_2$$



23. The order of reactivity of the following alcohols

towards conc. HCI is



III. 📄 IV. 📄

A. I > II > III > IV

 $\mathsf{B}.\, I > III > II > IV$

 $\mathsf{C}.\,IV>III>II>I$

D. IV > III > II > I

A 4 4 4 4 4 4 4 4 4

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24. In which of the following reaction, regioselectivity can be observed ?



25. The most probable product in the following reaction is :









Answer: C





Then what wil be the major product of the following

reaction



product



D. 2 & 3 both





X and Y are respectively :





The alkene formed as a mojor product in the above elimination reaction is:

A. (1)

 $\mathsf{B.}\,CH_2=CH_2$







$$29. CH_{3} - \overset{O}{C} - CCl_{3} \xrightarrow{CH_{3}MgBr} \xrightarrow{H_{3}O^{+}} \text{ product is}$$

$$A. CH_{3} - \overset{O}{C} - CH_{3}$$

$$B. CH_{3} - \overset{OH}{C} - CH_{3}$$

$$C. CH_{3} - \overset{OH}{C} - CCl_{3}$$

D.
$$CH_3 - \displaystyle \mathop{C}\limits_{\substack{|\ CH_3}}^{OH} - CH_3$$

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30.
$$CH_3CH_2Br \xrightarrow{Mg/Et_2O} A \xrightarrow{CO_2} \xrightarrow{H_2O^+} B$$
, Porduct B

is

- A. $CH_3 CH_2 CH_2OH$
- $\mathsf{B.}\,CH_3-CH_2-CHO$
- $C. CH_3 CH_2 COOH$

 $\mathsf{D}.\,CH_3-CH_2-CH_3$



D. All are incorrect.

Answer: C



32. Which is the product of the following elimination







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34.
$$C_3H_6Cl_2 \xrightarrow[2.H_2O^{\oplus}3.\Delta]{} CH_2 - CHCOOH$$
: Hence

reactant is :

A.
$$Cl-CH_2-CH_2-CH_2-Cl$$



Answer: 4

D.

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35.

(A), (B) and (C) respectively are



 $B_{\bullet} \overset{\scriptscriptstyle{(2)}}{\longrightarrow} \overset{\scriptstyle{(2)}}{\longrightarrow} \overset{\scriptstyle{(2)}}{\to} \overset{\scriptstyle{(2)}}{$

C. (3) O-CH-CH-CH, O-CH-CH-CH, O-CH-CH-CH, O

D. (4) Broker Operation



36. Which statement is incorrect.

A. Phenols decolourise Br_2 water due to

electrophilic substitution.



C. Tertiary butyl bromine is more reactive towards

hindered both E1 and E2 elimination among its

isomers

D. More is Number of $\beta - H$, more is the ease of

E2 reaction.

Answer: 4



37. Observe the following sequence of reaction and

find the structure of A.



A. $CH_3CH_2CH_2OH$

B. $CH_3CH(OH)CH_3$

$\mathsf{C.}\, CH_3 CH_2 OH$

$\mathsf{D.}\, CH_3 CH_2 CH_2 CH_2 OH$

Answer: 2





Product

Main product in above reaction is :









Answer: B



39. Which one of the following is not E2 reaction ?

$$\begin{array}{l} \mathsf{A.} \ CH_{3} - \underset{l}{CH} - CH_{2} - CH_{3} \xrightarrow{alc. KOH, \Delta} \\ & \stackrel{l}{\underset{Cl}{\overset{CH_{3}}{\rightarrow}}} \\ \mathsf{B.} \ CH_{3} - \overset{l}{\underset{C}{CH}} H - \underset{l}{CH} - CH_{3} \xrightarrow{(CH_{3})_{3}C\overset{\oplus}{O}K^{\oplus}} \end{array}$$

$$\mathsf{D}.\,CH_3 - \mathop{CH_3}\limits_{\stackrel{|}{CH_3}} - CH_2 - Br \xrightarrow[]{NH_3(l)}{\Delta}$$



40. Most reactive alkyl halide towards E2 mechanism

is :

A.
$$\left(CH_{3}
ight) _{3}C-CH_{2}Br$$

B.
$$(2)(CH_3)_2CH-CH < CH_3$$

$$\mathsf{C.}\left(CH_{3}\right)_{3}C-CH_{2}CH_{2}Br$$



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41. The product which is not formed in the following

reaction :

$$Ph - egin{array}{ccc} CH_3 & & CH_3 & & \ ert H_1 & ert H_2 & ert H_1 & ert H_2 & ert H_1 & ert H_2 & ert H_1 &$$

A. HCOOH

B.
$$Ph - \mathop{C}\limits_{\substack{|| \ o}} - CH_3$$

$$\overset{CH_{3}}{\mid}{\mathsf{C}}.\,OHC-\overset{CH_{3}}{CH}-CH_{2}OH$$

D.
$$OHC - \overset{CH_3}{\overset{|}{C}H} H - CHO$$



42. Which of the following reagent is correct for preparation of but -2 – enoic acid from but -2 - en - 1 - ol?

A. $KMnO_4$ / H^+

B. $Cu/300^{\,\circ}C$

C. Pyridinium chloro chromate (PCC)

D. $(i)MnO_2$. (ii) Ammonical $AgNO_3,\,(iii)H^+$

Answer: 4

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43. An SN^2 reaction of an asymmetric carbon of a compound always gives :

A. an enantiomer of the substrate

B. a product with opposite rotation

C. a mixture of diastereomers

D. A single stereoisomer

Answer: 4





(x) and (y) are :

A. Structural isomers

B. Enantiomers

C. Different compounds

D. Identical compounds

Answer: 4





The above transformation proceeds through

A. Eelctrophilic - addition

B. Benzyne intermediate

C. Nucleophilic substitution

D. Electrophilic substitution

Answer: 3

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46. Racemic mixture is obtain in which substrate when it is treated with CH_3OH

A.
$$CH_{3} - \overset{H}{\overset{|}{C}}_{C} - CH_{2} - Cl$$

 $\overset{H}{\overset{H}{}_{H}}_{H}$
B. $CH_{3} - \overset{H}{\overset{C}{CH_{3}}}_{H} - CH_{2} - Cl$













Answer: C



Possible number of products and fractions on fractional distillation are respectively:

B. 4,2

C. 6,3

D. 8,4















50. In the given reaction A will be :







CeH3 C-CH2OH



С.

A.

(4) C₆H₅C-CH₂OCH₃ CH₃C-CH₂OCH.

Answer: B

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51. Reaction between neopentyl bromide and ethanol

gives 2 - ethoxy -2 - methylbutane as the

principal product . Which one of the following is not true about this reaction ?

A. This involves a 1, 2- hydride shift

B. This involves a 1, 2 - methyl shift.

C. This occurs through a S_N 1 mechanism

D. This is accompained with formation of alkenes

as minor product



52. In the given reaction



 $\left[X
ight]$ will be :





C. Mixture of (1) and (2)



D.

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$${f 53.}\,pH - \mathop{C}\limits^{OH}_{l} - \mathop{C}\limits^{NH_3}_{l}_{Me} - Me \; {NaNO_2\,,HCl\over\Delta} \;?$$

Major product is

 $egin{aligned} \mathsf{A}. \ Ph &- C(OH) - C(NO_3)Me_2 \ & igstarrow Me \ & egin{aligned} & OAg \ & OAg$

$$C. Ph - CO - CMe_3$$

D.
$$Ph - \displaystyle \mathop{C}\limits_{\substack{\mid \ Me}}^{Me} - \displaystyle \mathop{C}\limits_{\substack{\mid \ Me}} - \displaystyle \mathop{C}\limits_{\substack{\mid \ O}} - Me$$

Answer: D



54. The major product of following sequence of reactions is







D. (4) C-CH4 + (CH4)(C-CH4 - N) CH4

Β.



55. Select false statement from the following ?

A. (1) The step
$$F^-CF_2 \rightarrow CF_2 \rightarrow CF_2 = CCI_2$$

B. E-1 reaction can be ragio selective .

C. $S_N 2\&E2$ reactions are stereo specific.

configuration takes place.

Answer: 4

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56. What is the product in following reaction ?









Answer: 1

D.

