



# CHEMISTRY

# PHYSICAL, INORGANIC, AND ORGANIC CHEMISTRY

# FUNDAMENTAL CONCEPT

**Organic Chemistry Fundamental Concept** 

1. Among the following the dissociation constant is highest for

A.  $C_6H_5OH$ 

 $\mathsf{B.}\, C_6H_5CH_2OH$ 

- $\mathsf{C.}\,CH_3C=CH$
- D.  $CH_3NH_3^+Cl^-$



2. Which of the following is more basic than aniline? .

A. p – Nitroaniline

B. Benzyl amine

C. Diphenyl amine

D. Triphenyl amine

Answer: 2

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3. Which one of the following is most reactive towards electrophilic

reagent?











4. The order of stability of the following tautomeric compounds is

$$\begin{array}{ccc} & & & OH & & O \\ & & & & | & & O \\ (i). \ CH_2 = & CH - CH_2 - C - CH_3 \Leftrightarrow \\ & & & O & & O \\ (ii). \ CH_3 - & C - CH_2 - C - CH_3 \Leftrightarrow \\ & & OH & O \\ (iii). \ CH_3 - & C = CH - C - CH_3 \end{array}$$

A. III > II > I

 $\mathsf{B}.\,II>I>III$ 

 $\mathsf{C}.\,II > III > I$ 

 $\mathsf{D}.\, I > II > III$ 

Answer: 1

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5. The correct order of C-N bond lengths for the followin compounds

is :



# Answer: A



7. Correct order of basic nature is

A.  $NH_3 > CH_2NH_2 > C_6H_5NH_2$ 

 ${\rm B.} \ C_{6}H_{5}NH_{2} > NH_{3} > CH_{3}NH_{2}$ 

 $\mathsf{C.}\,CH_3NH_2>NH_3>C_6H_5NH_2$ 

D. None of these

#### Answer: 3

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**8.** In the anion  $HCOO^-$  , the carbon-oxygen bonds are found to be of equal length. This is due to :

A. The anion is obtained by the removal of a proton from the acid molecule.

B. Electronic orbitals of carbon atoms are hybridised.

C. The C = O bond is weaker than C - O bond.

D. The anion  $HCOO^-$  has two equally stable resonating structures.

9. Which of the following orders of acid strength is correct?

A. RCOOH > CH = CH > HOH > ROH

 $\mathsf{B.} RCOOH > ROH > HOH > CH = CH$ 

 $\mathsf{C.} RCOOH > HOH > ROH > CH = CH$ 

 $\mathsf{D}. \textit{RCOOH} > \textit{HOH} > \textit{CH} = \textit{CH} > \textit{ROH}$ 

Answer: 3

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**10.** 
$$C_2H_5OH \xrightarrow{X} C_2H_5Cl$$

X can be except :

A.  $SOCl_2$ 

B.  $PCl_3$ 

 $C. PCl_5$ 

 $\mathsf{D.}\, NaCl$ 

Answer: D

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11. The conversion of Cl-CH=CH-Cl to  $CHCl_2-CHCl_2$  can be

carried out with ,

A. HCl(excess)

 $\mathsf{B.}\,Cl_2\,/\,hv$ 

 $\mathsf{C.}\,Cl_2\,/\,CCl_4$ 

 $\mathsf{D.}\,Cl_2\,/\,NaOH$ 

Answer: 3

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12. In which of the following rate of  $S_N 1$  is very fast ?

$$Cl$$
A.  $CH_3 - \overset{Cl}{CH} - CH_3$ 
B.  $Ph - \overset{l}{CH} - CH_3$ 
C.  $CH_3CH_2 - O - CH_2 - Cl$ 
D.  $CH_3 - CH = CH - Cl$ 

#### Answer: 3

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13. Which is the correct order for basicity?

A. 
$$F^{-} < I^{-}$$
  
B.  $N\overset{ extbf{\theta}}{H_{2}} < \overset{ extbf{\theta}}{O}H$   
C.  $CH_{2} = \overset{ extbf{\theta}}{C}H < CH \equiv \overset{ extbf{\theta}}{C}$   
D.  $CH_{3}NH_{2} < CH_{3}\overset{ extbf{\theta}}{N}H$ 



14. Pyridine is less basic than triethylamine because .

A. Pyridine has aromatic character

B. Nitrogen in pyridine is  $sp^2$  hybridised

C. Pyridine is a cyclic system

D. In pyridine, lone pair of nitrogen is delocalised

#### Answer: 2

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15. The decreasing order of the stability of the ions :

$$CH_3 = \overset{+}{\underset{(I)}{C}}H - CH_3 \qquad CH_3 - \overset{+}{\underset{(II)}{C}}H - OCH_3 \qquad CH_3 - \overset{+}{\underset{(III)}{C}}H - COCH_3^{-}$$

A. I > II > III

 $\mathsf{B}.\,III>II>I$ 

C. II > III > I

D. II > I > III

Answer: 4

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**16.** Which order is correct for both nucleophilicity and basicity ?

- A.  $Cl^{\,-}\,< I^{\,-}$
- $\mathsf{B.}\,H_2O < H_2S$
- C.  $MeO^- > F^-$
- D.  $EtO^- < OH^-$

#### Answer: C

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17. Arrange pH of the given compounds in decreasing order :

A. Phenol

B. Ethyl alcohol

C. Formic acid

D. Benzoic acid

Answer: 2

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18. Maximum enol content is in :







**19.** Consider the  $S_N$  1 solvolysis of the following hallides in aqueous

formic acid :



Which one of the following is correct sequence of the halides given above in the decreasing order of their reactivity ?

A. III > IV > II > I

 $\mathsf{B}.\,II > IV > I > III$ 

 $\mathsf{C}.\, I > II > III > IV$ 

 $\mathsf{D}.\,III > I > II > IV$ 

Answer: 1

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20. In the following groups:

(I) -OAc, (II) -OMe, (III)  $-OSO_2Me$ , (IV)  $OSO_2CF_3$  the order of

leaving group ability is-

A. i > ii > iii > iv

 $\mathsf{B}.\,iv>iii>i>ii$ 

 $\mathsf{C}.\,iii>ii>i>iv$ 

 $\mathsf{D}.\,ii>iii>iv>i$ 



**22.** Which of the following is the least stable carbanion?



23. Arrange stability of the given carbocation in decreasing order :



A. I > II > III > IV

 $\mathsf{B}.\,III>II>I>IV$ 

 $\mathsf{C}.\,IV > I > II > III$ 

 $\mathsf{D}.\,II > III > I > IV$ 

#### Answer: 2

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**24.** Which of the following compound give solvolysis reaction with slowest rate ?

A. 
$$P - \stackrel{\oplus}{N} \! R_3 - C_6 H_4 - C H_2 - B r$$

$$\mathsf{B}.\,p-CCl_3-C_6H_4-CH_2-Br$$

C. 
$$P-N\equiv C-C_{6}H_{4}-CH_{2}-Br$$

$$\mathsf{D}.\,C_6H_5-CH_2-Br$$

#### Answer: C

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25. Which of the following alkoxides is the most reactive nucleophile?

A.  $CH_3O^-$ 

- B.  $C_6H_5O^-$
- $C.(CH_3)_2CHO^-$
- D.  $(CH_3)_3 CO^-$

#### Answer: A

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# 26. Which is the correct matched for the following reactions



#### Answer: D



# 27. Which is correctly matched





**28.** The correct increasing order of the reactivity of halides for  $S_N$  1 reaction is :

A.

 $CH_3 - CH_2 - X \;\; < \;\; (CH_3)_2 CH - X \;\; < \;\; CH_2 = CH - CH_2 - .$ 

Β.

 $(CH_3)_2 CH - X \ < \ CH_3 - CH_2 - X \ < \ CH_2 \ < \ PhCH_2 - X$ 

C.

 $PhCH_2 - X < (CH_3)_2CH - X < CH_2 - CH_2 - X < CH_2$ 

D.

 $CH_2 = CH - CH_2 - X \ < \ Ph - CH_2 - X \ < \ (CH_3)_2 CH - X$ 



**29.** Reaction intermediate of EicB reaction is :

A. Carbocation

**B.** Carbanion

C. Benzyne

D. Free radical

Answer: B

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30. Select false statement from the following ?

A. Formation of dechlorocarbene from  $CHCl_3$  is an elmination

reaction.

B. Carbocations and free radicals are planar chemical species.

C. In the rearrangement of carbocation ,  $1^\circ\,-\,$  carbocation may

convert into  $2^\circ$  - carbocation.

D.  $CCl_3$  group is o, p - directing becausing it exhibit

hyperconjugation with benzene ring.

Answer: 4

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# 31. Which of the following is not electrophile ?

A.  $CN^{\,-}$ 

 $\mathsf{B.}\,H^{\,+}$ 

C.  $Br^+$ 

D.  $AlCl_3$ 

Answer: 1

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32. Which statement is incorrect :

A. Halogens are deactivating groups but have ortho-para directing

nature.

B. Nitrobenzene, gives metanitrotoluence on reaction with

 $CH_3Cl/AlCl_3$ .

- C. Nitro group is meta directing.
- D. Aniline does not give Friedal Craft reaction directly.

#### Answer: 2

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**33.** The correct leaving group ability order is :

A. 
$$\overset{ extbf{\theta}}{O}H > H_2O$$





 $\overset{ \rm e}{{\rm D.}} \overset{ \rm e}{CI} > \overset{ \rm e}{I}$ 

#### Answer: 3

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$$\textbf{34.} CH_3 - CH_2 - O - CH_2 - Cl \xrightarrow[]{CH_3MgBr}_{\text{other}}$$

Which statement is incorrect for the above reaction :

A.  $CH_3 - CH_2 - O - CH_2 - Cl$  react faster than

 $CH_2CH_2-O-CH_3$  when treaterd with  $CH_3MgBr$ .

B. Reasonance stabilised carbonium ion is formed in the above

reaction.

C. Product of the reaction is diethyl ether

D. Reaction is proceed through  $S_N 2$  mechanism.

# Answer: 2



 $\overset{_{\boldsymbol{\Theta}}}{\mathrm{D.}}\overset{_{\boldsymbol{\Theta}}}{NH_4}$ 



36. Select the correct option :

 $S_1$ : Catechol is less acidic than resorcinol.

 $S_2$ : Ortho - nitrophenol is less acidic than para - nitrophenol.

 $S_3$ : Ortho- cresol is less acidic than its meta and para isomers.

 $S_4$ : Generally ortho substituted phenols are less acidic than their coressponding meta & para isomers.

A. All are correct

B. Only  $S_1, S_2 \& S_3$  are correct.

C. Only  $S_1$  and  $S_2$  are correct

D. Only  $S_1$  and  $S_3$  are correct.

#### Answer: 2



37. The reaction is :

 $CH_3CHBr-CH_2Br+2KOH( ext{alc.}) \stackrel{\Delta}{\longrightarrow} CH_3-C \equiv CH+2KBr+2H_3CH$ 

A. Dehalogenation

- B. Dehydrohalogenation
- C. Decarboxylation
- D. Dehydration

# Answer: 2

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# 38. For the following

$$(i)I^{\,-}(ii)Cl^{\,-}(iii)Br^{\,-}$$

the increasing order of nucleophilicity would be:

A. 
$$I^- < Cl^- < Br^-$$
  
B.  $Br^- < Cl^- < I^-$   
C.  $I^- < Br^- < Cl^-$ 

D.  $Cl^- < Br^- < I^-$ 

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 ${\bf 39.}$  Orbital interation ( partial overlapping ) between the sigma bonds of

a substitutent group and a neithobouring pi orbital is known as

A. Hyperconjugation effect

**B.** Inductive effect

C. Steric effect

D. Dipole-dipole interactions

Answer: 1

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40. In which delocalisation of positive charge is possible







