



# CHEMISTRY

# PHYSICAL, INORGANIC, AND ORGANIC CHEMISTRY

# NITROGEN CONTAINING COMPOUNDS

**Organic Chemistry Nitrogen Containing Compounds** 

1. The positive carbylamine test is not given by

A. anilinie

- B. 2, 4 dimethyl aniline
- C. N -methyl -O -methyline
- D. p- methyl benzylamine

# Answer: C



2. Which of the following compound gives secondary amine on reduction.

A. Alkyl Nitrite

B. Carbylamine

C. Alkylcyanide

D. secodary nitro compound

### Answer: 2

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3. Hinsber's reagent is used to distinguish between

A.  $1^\circ$  and  $2^\circ$  amine

B.  $2^\circ$  and  $3^\circ$  amine

C.  $1^\circ$  and  $3^\circ$  amine

D. All of these

# Answer: 4



Ethyl amine  $\xrightarrow{HNO_2} (A) \xrightarrow{PCl_5} (B) \xrightarrow{KCN} (C) (C)$  is,

A. Propanenitrile

B. Trethylamine

C. diethylamine

D. propylamine

Answer: 1

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5. In which of the following reaction migration of alkyl group from '' C''

to ''  $N^\prime$  ' terminus is not occuring

A. Pinacol-Pinacolone rearrangement

B. Curitius rearrangement

C. Lossen Rearrangement

D. Beckmann Rearrangement

Answer: 1

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**6.** The R in the given reaction sequence is -

$$\mathsf{RNH}_2 \xrightarrow{\mathsf{CHCl}_2} \mathsf{A} \xrightarrow{\mathsf{H}_2/\mathsf{Pd}} \mathsf{B} \xrightarrow{1.\mathsf{CHJ} \text{success}} \mathsf{C}$$















# Answer: 2

A.



8. The Cimetidine has serveral nitrogen atom in its structure. Identify the

most basic atom of given cimetidine derivative.



A. I
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B. II

C. III

D. V





Answer: C



**10.** Which of the following is capable of forming a zwitter ion ?

- A.  $C_6H_5-OH$
- $\mathsf{B.}\, C_6 H_4 \big(N H_2\big)_2$
- C.  $CH_2OH$
- $\mathsf{D}.\,H_2N-CH_2-COOH$



**11.** p – chloroaniline and anilinium hydrogen chloride can be distinguished by

A. Sandmayer reaction

B. Carbylamine reaction

C. Hinsberg's reaction

D.  $AgNO_3$ 

# Answer: D

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**12.** In the diazotisation of anline with sodium nitrite and hydrochloride acid, an excess of hydrochloric acid is used primarily to

A. Suppress the concentration of free aniline for coupling

- B. Suppress hydrolysis of phenol
- C. Ensure a stoichiometric amount of nitrous acid
- D. Neutralize the base liberated

#### Answer: 1

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**13.** Starting from propanoic acid, the following reaction were carried acid  $\xrightarrow{SOCl_2} X \xrightarrow{NH_3} Y \xrightarrow{Br_2 + KOH} Z$  What is the compound?

A. 
$$CH_3 - CH_2 - Br$$

$$\mathsf{B}. CH_3 - CH_2 - NH_2$$

C.

D. 
$$CH_3 - CH_2 - CH_2 - NH_2$$

14. In the reaction,

$$C_{6}H_{5}NH_{2} \xrightarrow[0-5^{\circ}C]{NaNO_{2}+HCl} (A) \xrightarrow[KCN]{CuCN} (B) \xrightarrow[H^{+}/H_{2}O]{C} (C)$$

the product (C) is

A.  $C_6H_5CH_2NH_2$ 

 $\mathsf{B.}\, C_6H_5COOH$ 

 $\mathsf{C.}\, C_6H_5OH$ 

D. none of these



**15.** In the given reaction product P is :



- A. 1 Butane
- B.2 Butene
- C.2 Methylpropene
- D. Ethane





17. Identify  $A, C_6 H_{11} N$ , for which the given information is availabel .

 $(i) {
m Baeyer \ reagent}$  - No test  $(ii) Ph - SO_2 Cl o {
m acidic \ compound \ soluble}$ 





**18.** An optically inactive amine (A) is methylated with excess of  $CH_3I$  and there after reacted with AgOH to form alkene (B). The alkene B after ozonolysis gives HCHO and  $CH_3CH_2CH_2CHO$ . A and B in the reaction are respectively.

A. 
$$CH_3 - CH_2 - CH_2 - NH_2$$
 &  $CH_3 - CH_2 - CH = CH_2$ 

Β.

$$CH_3-CH_2-CH_2-CH_2-CH_2$$
 &  $CH_3-CH_2-CH=CH_2$ 

C.

$$CH_3 - CH_2 - CH_2 - CH_2 - NH_2 \quad \& \quad CH_3 - CH_2 - CH_2 - CH_2 - CH_2$$

D.

$$CH_3 - CH_2 - CH_2 - CH_2 - CH_3 \quad \& \quad CH_3 - CH_2 - CH_$$

Answer: 3

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19. The product P formed in the given reaction sequence is .











# Answer: C



**20.** Identify the correct statement.



A is more basic than B

C.



# Answer: 3



**21.** Which statement is incorrect.

A. Fluoro benzene can be synthesized in the laboratory from aniline by

diazotisation followed by heating with  $HBF_4$ 

B. Quarternary Ammonium hydroxides having  $\beta$  – hydrogen atom

give hlfmann elimination on pyrolysis.



D. Benzyl amine on reaction with  $NaNO_2/HCl$  followed by  $\beta$  –

Naphthol in slight basic medium forms a coloured dyo.

Answer: 4

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**22.** What will be the major product when 2 - Aminopropane is treated with nitrous acid ?

A. 1 - Propanol

B.2 - Propanol

C. Propene

D. Cyclopropane

Answer: 2

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**23.** Which of the following product (s) will be obtained when isopropylamine is treated with sodium nitrite and hydrochloric acid ?

A. 
$$CH_3 - CH - CH_3$$
  
 $\downarrow_{OH}$   
B.  $CH_2 - CH = CH_2$   
C.  $CH_3 - CH = CH_2$ 

D. All of these

#### Answer: 4

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**24.** Which of the following compounds will give  $N_2(g)$  on treatment with

 $HNO_2$ 

A.  $C_2H_5-NH_2$ 

 $\mathsf{B.}\,C_6H_5-NH-CH_3$ 

 $\mathsf{C}.\, NH_2 - CO - NH_2$ 

D. 
$$C_6H_5-CO-NH_2$$

Answer: 1



**25.** Compound  $(x)(m. f = C_7 H_8 N)$ , on reaction with  $NaNO_2$  and conc. HCl at  $O^{\circ}C$  followed by  $\beta$  – naphthol gives orange coloured dye. Compound (x) is :



Β.

A.



C.

D. All of these

# Answer: B



26. Pyridine is less basic than triethylamine because .

A. Pyridine has aromatic character

B. Nitrogen in pyridine is  $sp^2$  hybridised

C. Pyridine is a cyclic system

D. In pridine, lone pair of nitrogen is delocalised

# Answer: 2

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$$\begin{array}{l} \mathsf{A.} \begin{array}{c} \frac{Cl_2/Fe}{(a)} & \frac{HNO_3 + H_2SO_4, \Delta}{(b)} & \frac{NH_4HS}{(c)} \\ \\ \mathsf{B.} & \frac{HNO_3 + H_2SO_4. \Delta}{(a)} & \frac{NH_4HS}{(b)} & \frac{NaNO_2/HCl/\Delta \operatorname{boil}}{(c)} \\ \\ \mathsf{C.} & \frac{HNO_3 + H_2SO_4}{(a)} & \frac{(i) Cl_2/Fe}{(ii) NaOH/\Delta (b)} & \frac{Sn/HCl}{(c)} \end{array}$$

D. All are correct.



28. Arrange following Amines for rate of reaction with  $CHCl_3 + KOH$  ?



A. S > P > R > Q

 $\operatorname{B}.Q>R>P>S$ 

 $\operatorname{C}.S > P > Q > R$ 

 $\mathsf{D}.\,S > Q > R > P$ 

#### Answer: 1

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**29.** Which of the following is correct order of basic strength for the given compounds ?







(4) i> i> i> i> w> III

A. I > ii > iii > iv

 $\mathsf{B}.\,iv>iii>ii>i$ 

 $\mathsf{C}.\,iv>ii>iii>i$ 

D. I > ii > iv > iii

#### Answer: A

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**30.** Which nitrogen atom is most basic ?



