



## PHYSICS

### BOOKS - SUNIL BATRA 41 YEARS IITJEE PHYSICS (HINGLISH)

#### MODERN PHYSICS

Jee Main And Advanced

1. To produce characteristic  $X$  - rays using a Tungsten target in an x - ray generator , the accelerating should be greater than ..... Volts and the energy of the characteristic is ..... eV .

(The binding energy of the innermost electron in Tungsten is - 40keV).



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2. The radioactive decay rate of a radioactive element is found to be  $10^3$  disintegration // second at a certain time . If the half life of the element is one second , the decay rate after one second ..... And after three second is .....



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3. The maximum kinetic energy of electrons emitted in the photoelectric effect is linearly dependent on the ..... Of the incident radiation .



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4. In the final Uranium radioactive series the initial nucleus is  ${}_{92}^{238}\text{U}$  and the final nucleus is  ${}_{82}^{206}\text{Pb}$  . When Uranium nucleus decays to lead , the number of  $\alpha$  - particle is ..... And the number of  $\beta$  - particles emitted is .....



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5. When the number of electron striking the anode of an X - ray tube is increase , the ..... Of the emitted X - ray increases , while when the speeds of the electrons the anode are increased , the cut - off wavelength of the emitted X - ray .....



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6. when Boron nuclus  $\left( {}_{-3}^{10}B \right)$  is bombarded by neudrons , a- particle are emitted . The resulting nucleus is of the elenent ..... And has the mass different ..... are called isotopes .



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7. Atoms having the same ..... but different ..... Are colled isotopes .



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8. The binding energy per nucleon number for deuteron ( ${}_1^2\text{H}$ ) and helium ( ${}_2^4\text{He}$ ) are  $1.1\text{MeV}$  and  $7.0\text{MeV}$  respectively. The energy released when two deuterons fuse to form a helium nucleus ( ${}_2^4\text{He}$ ) is .....



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9. In the forward bias arrangement of a  $p - n$  junction rectifier, the  $p$  and  $n$  is connected to the ..... Terminal of the battery and the direction of the current is from ..... to ..... In the rectifier.



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10. ..... biasing of  $p - n$  junction. Offers high resistance to current flow across the junction. The biasing is obtained by connecting the  $p - side$  to the ..... Terminal of the battery



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11. The wavelength of the characteristic X - ray  $k_{\alpha}$  line emitted by a hydrogens like element is  $0.32\lambda$  . The wavelength of the  $K_{\beta}$  line emitted by the same element will be .....



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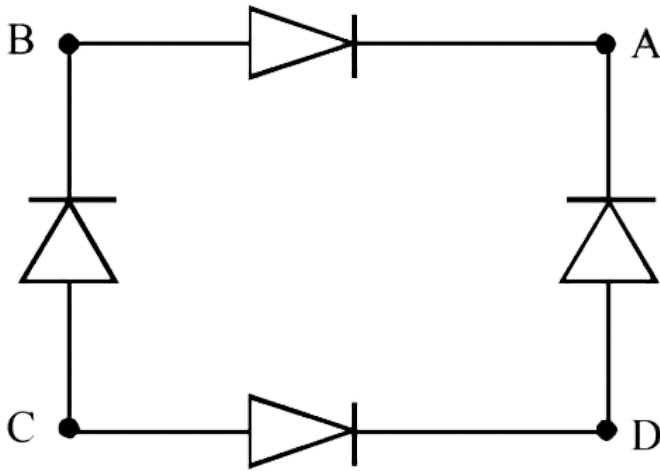
12. The Bohr radius of the fifth electron of phosphorous atom ( $a \rightarrow \text{micrometer} = 15$ ) acting as a dopant in silicon ( $\text{relative dielectric constant} = 12$ ) is ..... Å



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13. For the given circuit shown in fig to act as full wave rectifier , the a, c, input should be connected across ..... and ..... And the d. c. output would

suppear across ..... And .....



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14. In an X-ray tube , electrons accelerated through a potential different of 15000 volts strike a copper target . The speed of the emitted X - ray inside the tube is .....  $m/s$

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15. In the Bohr model to the local energy of the electron in a quantum state  $n$  is .....



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16. In the nuclear process ,  ${}_{-}(6)C^{11} \rightarrow {}_2B^{11} + {}^{11}\beta^{+} + X$ ,  $X$  stands for.....



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17. In a ..... Biased  $p - \alpha$  junction , the net flow of holes is from the  $n$  region to the  $p$  region .



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18. A potential difference , the  $20kv$  is applied across an X- ray tube . The minimum wavelength of x- ray generated is .....Å .



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19. The wavelength of  $k_{\alpha}$  X- rays produced by an X - rays tube is  $0.76\text{\AA}$  .

The atomic number of the anode material of the tube is .....



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20. Fill in the blanks with appropriate items :

Consider the following reaction,  ${}^2_1\text{H} + {}^2_1\text{H} = {}^4_2\text{He} + Q$ .

Mass of the deuterium atom =  $2.0141u$  , Mass of the helium atom =  $4.0024u$

This is a nuclear \_\_\_\_\_ reaction in which the energy  $Q$  is released is \_\_\_\_\_ MeV.



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21. The kinetic energy of photoelectrons emitted by a photosensitive surface depends on the intensity of the incident radiation



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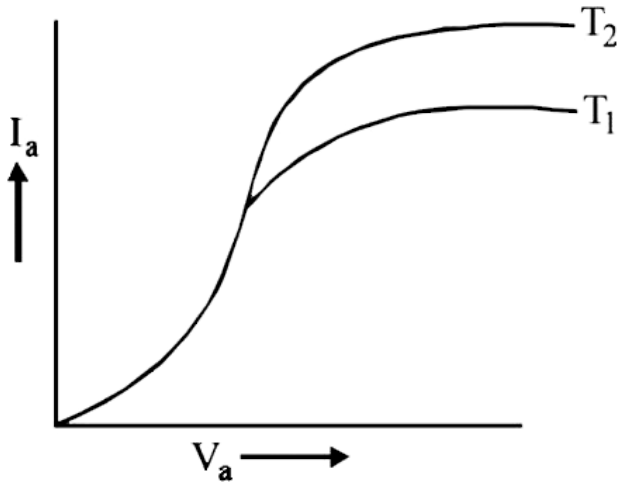


22. In a photoelectric emission process the maximum energy of the photoelectrons increase with increasing intensity of the incident light .



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23. For a diode the variation of its anode current  $I_a$  with the anode voltage  $V_a$  at two different cathode temperatures  $T_1$  and  $T_2$



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24. The order of magnitude of the density of nuclear matter is  $10^4 \text{ kg m}^{-3}$

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25. The plate resistance of a triode is  $3 \times 10^3$  ohms and its mutual of the triode is

A.  $5 \times 10^{-5}$

B. 4.5

C. 45

D.  $(0.2 \times 10^3)$

**Answer: B**

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26. The half life of radioactive Radon is  $3.8 \text{ days}$ . The time at the end of which  $\frac{1}{20} \text{th}$  of the radon sample will remain undecayed is  
(given  $\log e = 0.4343$ )

A.  $3.8days$

B.  $16.5days$

C.  $33days$

D.  $76days$

**Answer: B**



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27. An alpha particle of energy  $5MeV$  is scattered through  $180^\circ$  by a heavy nucleus. The distance of approach is of the order of

A.  $1\text{\AA}$

B.  $10^{-10}cm$

C.  $10^{-12}cm$

D.  $10^{-15}cm$

**Answer: C**



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28. Beta rays emitted by a radioactive material are

- A. electromagnetic radiations.
- B. the electrons orbiting around the nucleus .
- C. charged particle emitted by the nucleus .
- D. neutral particles

**Answer: C**



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29. If element with principal quantum number  $n > 4$  were not allowed in nature , the number of possible elements would be

- A. 60
- B. 32

C. 4

D. 64

**Answer: A**



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**30.** Consider the spectral line resulting from the transition  $n = 2 \rightarrow n = 1$  in the atoms and ions given. The shortest wavelength is produced by

A. Hydrogen atom

B. Deuterium atoms

C. Singly Ionized Helium

D. Doubly Ionised Lithium

**Answer: D**



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31. The equation

$$4\left[{}^1_1\text{H}^+ \rightarrow {}^4_2\text{He}^{2+} + 2e^- + 26\text{MeV}\right] \text{ represents}$$

A. beta - decay`

B.  $\lambda$  - decay

C. *flasion*

D. *fission*

Answer: C



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32. Fast neutrons can easily be slowed down by

A. the use of shielding

B. passing then through water

C. elastic collisions with heavy ouclet

D. applying a strong electric field

**Answer: B**



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**33.** Consider a particle ,  $\beta$  particle and  $\gamma$  - rays , each having an energy of  $0.5\text{MeV}$  . In increase order of panetrating poewr , the radiation are .

A.  $\alpha, \beta, \gamma$

B.  $\alpha, \gamma, \beta$

C.  $\beta, \gamma, \alpha$

D.  $\gamma, \beta, \alpha,$

**Answer: A**



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34. As energy of  $24.6\text{eV}$  is required to remove one of the required to remove both the electrons from a nuture brfore alon is

A. 38.2

B. 94.2

C. 51.8

D. 79.0

**Answer: D**



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35. A radiacative moterial docays by simulataneous emission of two particle from the with respective half - lives 1620 and 810 year . The time , in year , after which one - fourth of the material remins is

A. 1080

B. 2430



C. 3240

D. 4860

**Answer: A**



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**36.** The probability of electrons to be found in the conduction band of an intrinsic semiconductor at a finite temperature

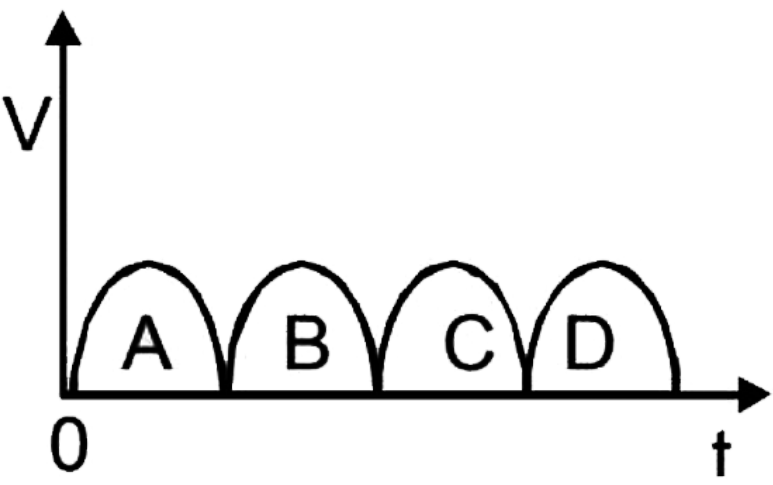
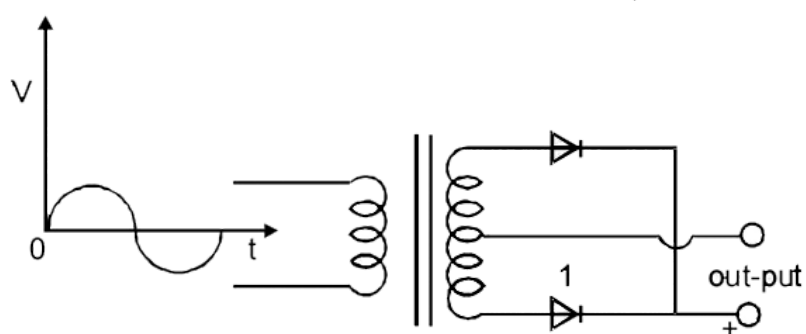
- A. increase exponentially with increases band gap
- B. decreases exponentially with increases band gap
- C. decreases with increases temperature
- D. is independent of the temperature and the band gap

**Answer: B**



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37. A full - wave rectifier circuit along the out - put is shown in figure . The contribution (s) from the diode 1 is are



- A. C
- B. A, C

C.  $B, D$

D.  $A, B, C, D$

**Answer: B**



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**38.** As per Bohr model , the minimum energy (in eV) required to remove electron from the ground state of doubly ionized  $Li$  alon ( $Z = 3$ ) is

A. 1.51

B. 13.6

C. 40.8

D. 122.4

**Answer: D**



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**39.** Which of the following statement is not true ?

- A. The resistance of intrinsic semiconductors decreases with increase of temperature .
- B. Doping pure Si with trivalent impurities give p- type semiconductors
- C. The majority in n- type semiconductor diode are holes
- D. A p- n junction can act as a semiconductor diode

**Answer: C**



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**40.** The maximum kinetic energy of photoelectrons emitted from a surface when photons of energy  $6eV$  fall on it is  $4eV$  . The stopping potential , in volt is

- A. 2
- B. 4

C. 6

D. 10

**Answer: B**



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41. In hydrogen spectrum the wavelength of  $H_\alpha$  line is  $656\text{nm}$ , where in the spectrum of a distant galaxy  $H_\alpha$  line wavelength is  $706\text{nm}$ . Estimated speed of the galaxy with respect to earth is,

A.  $2 \times 10^8 \text{ m/s}$

B.  $2 \times 10^7 \text{ m/s}$

C.  $2 \times 10^6 \text{ m/s}$

D.  $2 \times 10^5 \text{ m/s}$

**Answer: B**



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42. A particle of mass  $M$  at rest decays into two particles of masses  $m_1$  and  $m_2$  having non zero velocity. The ratio of the de Broglie wavelengths of the masses  $\lambda_1 / \lambda_2$  is

A.  $m_1 / m_2$

B.  $m_2 / m_1$

C. 1.0

D.  $\sqrt{m_1} / \sqrt{m_2}$

Answer: C



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43. Which of the following is a correct statement ?

A. Beta rays are same as cathode rays

B. Gamma rays are high energy neutrons

C. Alpha particle are singly ionised helium atoms

D. Proton and neutrons have exactly the same mass

**Answer: A**



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**44.** Order of magnitude of density of uranium nucleus is , [m =  $1.67 \times 10^{-27}$  kg]

A.  $10^{20} \text{ kg/m}^3$

B.  $10^{17} \text{ kg/m}^3$

C.  $10^{14} \text{ kg/m}^3$

D.  $10^{11} \text{ kg/m}^3$

**Answer: B**



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45. Ne nucleus , the after absorbing energy , decays into two  $\alpha$  -  $partic \leq$  and an unknown nucleus . The unknown nucleus is

A. nitrogen

B. carbon

C. boron

D. oxygen

**Answer: B**

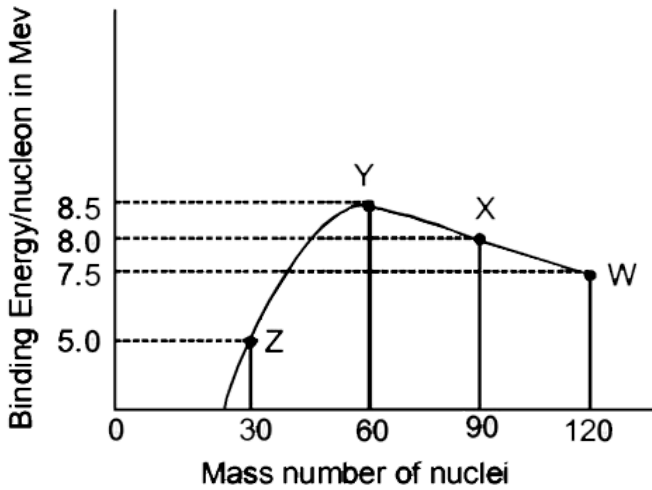


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46. Binding energy per vs mass curve for nucleus is shown in the figure  $W, X, Y$  and  $Z$  are four nuclei indicated on the curve . The process that



would release energy is



- A.  $Y \rightarrow 2Z$
- B.  $W \rightarrow X + Z$
- C.  $W \rightarrow 2Y$
- D.  $X \rightarrow Y + Z$

**Answer: C**



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47. Imagine an atom made up of a proton and a hypothetical particle of double the mass of the electron but having the same charge as the electron. Apply the Bohr atom model and consider all possible transitions of this hypothetical particle that will be emitted level . The longest wavelength photon that will be emitted has longest wavelength  $\lambda$  (given in terms of the Rydberg constant  $R$  for the hydrogen atom) equal to

A.  $9(5R)$

B.  $36/(5R)$

C.  $18/(5R)$

D.  $4/(5R)$

**Answer: C**



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**48.** The electron in a hydrogen atom makes a transition from an excited state to the ground state. Which of the following statements is true?

- A. Its kinetic energy increase and its potential and total energies decreases
- B. Its kinetic energy decrease , potential energy increase and its total energies remain the same
- C. Its kinetic and total energy decrease and its potential energy increases
- D. Its kinetic potential and total energies decreases

**Answer: A**



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**49.** Two radioactive  $X_1$  and  $X_2$  have decay constants  $10\lambda$  and  $\lambda$  respectively . If initially they have the same number of nuclei, then the ratio of the number of nuclei of  $X_1$  to that of  $X_2$  will be  $1/e$  after a time .

A.  $\frac{1}{10\lambda}$

B.  $\frac{1}{11\lambda}$

C.  $\frac{11}{10\lambda}$

D.  $\frac{1}{9\lambda}$

**Answer: D**



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**50.** Electrons with energy  $80\text{keV}$  are incident on the tungsten target of an X - rays tube , k- shell electrons of tungsten have  $72.5\text{keV}$  energy X- rays emitted by the tube contain only

A. a continuous X - rays spectrum (Bremsstrahlung) with a maximum wavelength of  $0.155\text{\AA}$

B. a continuous X - rays spectrum (Bremsstrahlung) with all wavelength

C. the characteristic X - rays spectrum of tungsten`

D. a continuous X - rays spectrum (Bremasstrahlung) with a maximum wavelength of  $0.155\text{\AA}$  and the characteristic X - rays spectrum of tungsten.

**Answer: D**



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**51.** The electron emitted in beta radiation originates from

- A. inner orbits of atoms
- B. free electrons existing in nuclei
- C. decay of a neutron in a nucleus
- D. photon escaping from the nucleus

**Answer: C**



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52. The transition from the state  $n = 4 \rightarrow n = 3$  in a hydrogen - like atom results in ultraviolet radiation. Infrared radiation will be obtained in the transition

A.  $2 \rightarrow 1$

B.  $3 \rightarrow 2$

C.  $4 \rightarrow 2$

D.  $5 \rightarrow 4$

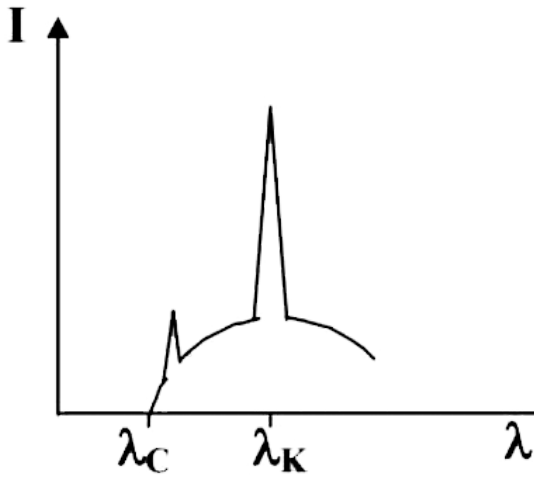
**Answer: D**



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53. The intensity of X- ray from a Coolidge tube is plotted against wavelength  $\lambda$  as shown in the figure . The minimum wavelength found  $\lambda_c$  and the wavelength of the  $K_\alpha$  line is  $\lambda_\lambda$ , As the accelerating voltage is

increase



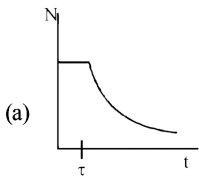
- A.  $\lambda_\lambda - \lambda_c$  increase
- B.  $\lambda_\lambda - \lambda_c$  decrease
- C.  $\lambda_\lambda$  increase
- D.  $\lambda_\lambda$  decrease

**Answer: A**

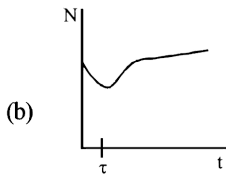


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54. A radioactive sample consists of two distinct species having equal number initially. The mean life time of one species is  $\tau$  and that of the other is  $5\tau$ . The decay products in both cases are stable. A plot is made of the total number of radioactive nuclei as a function of time, which of the following figure best represents the form of this plot?



A.



B.

C. 

**Answer: D**



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55. The potential difference applying to an X-ray tube is  $5kV$  and the current through it is  $3.2mA$ . Then the number of electrons striking the target per second is

A.  $2 \times 10^{16}$

B.  $5 \times 10^6$

C.  $1 \times 10^{17}$

D.  $4 \times 10^{15}$

**Answer: A**



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56. A Hydrogen atom and  $Li^{++}$  ion are both in the second excited state .

If  $l_H$  and  $l_{Li}$  are their respective energies, then

A.  $l_H > l_{Li}$  and  $|E_H| > |E_{Li}|$

B.  $l_H = l_{Li}$  and  $|E_H| < |E_{Li}|$

C.  $l_H = l_{LI}$  and  $|E_H| > |E_{Li}|$

D.  $l_H > l_{LI}$  and  $|E_H| > |E_{Li}|$

**Answer: B**



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57. The half - life of  $^{215}\text{At}$  is  $100\mu, s$ . The time taken for the radioactivity of a sample of  $^{215}\text{At}$  to decay to  $1/16^{th}$  of its initial value is

A.  $400\mu s$

B.  $63\mu s$

C.  $40\mu s$

D.  $300\mu s$

**Answer: A**



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58. Which of the following process represents a  $\gamma$  - decay?

A.  ${}^A_Z X + \gamma \rightarrow {}^A_Z X + a + b$

B.  ${}^A_Z X + {}^1_0 n \rightarrow {}^{A-3}_{Z-2} X + c$

C.  ${}^A_Z X \rightarrow {}^A_Z X + f$

D.  ${}^A_Z X + e_{-1} \rightarrow {}^A_Z X + g$

Answer: C



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59. The electric potential between a proton and an electron is given by

$$V = V_0 \ln\left(\frac{r}{r_0}\right), \text{ where } r_0 \text{ is a constant. Assuming Bohr's model to be}$$

applicable, write variation of  $r_n$  with  $n$ ,  $n$  being the principal quantum number?

A.  $r_n \propto n$

B.  $r_n \propto 1/n$

C.  $r_n \propto n^2$

D.  $r_n \propto 1/n^2$

**Answer: A**



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60. If the atom  $_{100}\text{Fm}^{257}$  follows the Bohr model the radius of  $_{100}\text{Fm}^{257}$  is a time the Bohr radius, then find  $n$ .

A. 100

B. 200

C. 4

D.  $1/4$

**Answer: D**



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61. For uranium nucleus how does its mass vary with volume?

A.  $m \propto V$

B.  $m \propto 1/V$

C.  $m \propto \sqrt{V}$

D.  $m \propto V^{1/2}$

Answer: A



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62. A nucleus with mass number 220 initially at rest emits an  $\alpha$ -particle.

If the  $Q$  value of the reaction is  $5.5 \text{ MeV}$ , calculate the kinetic energy of the  $\alpha$ -particle.

A.  $4.4 \text{ MeV}$

B.  $5.4 \text{ MeV}$

C.  $5.6\text{MeV}$

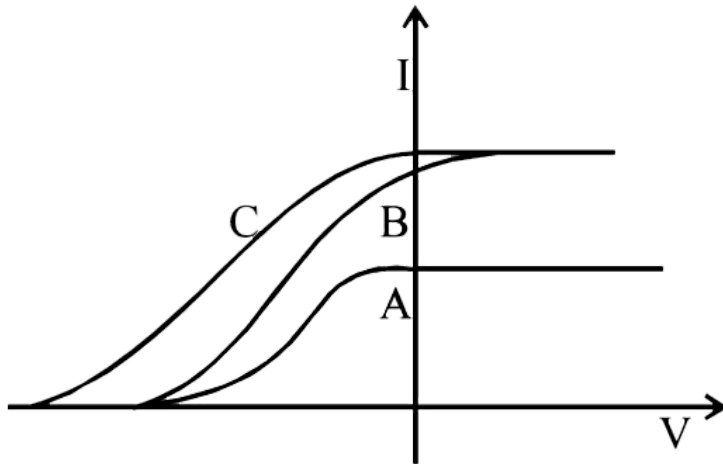
D.  $6.5\text{MeV}$

**Answer: B**



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63. In a photoelectric experiment anode potential is plotted against plate current.



A.  $A$  and  $B$  will have different intensities while  $B$  and  $C$  will have different frequencies

B.  $B$  and  $C$  will have different intensities while  $A$  and  $C$  will have different frequencies

C.  $A$  and  $B$  will have different intensities while  $A$  and  $C$  will have different frequencies

D.  $B$  and  $B$  will have equal intensities while  $A$  and  $B$  will have different frequencies

**Answer: D**



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**64.** A 280 days old radioactive substance shown an activity of  $6000 \text{ dps}$  100 days later its activity between 3000 days what was its initial activity ?

A.  $20000 \text{ dps}$

B.  $24000 \text{ dps}$

C.  $12000 \text{ dps}$

D.  $6000dps$

**Answer: B**



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65. A proton has kinetic energy  $E = 100keV$  which is equal to that of a photo is  $\lambda_1$  . The ratio of  $k_2 / \lambda_1$  is proportional to

A.  $E^2$

B.  $E^{1/2}$

C.  $E^{-1}$

D.  $E^{-1/2}$

**Answer: D**



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66.  $k_\alpha$  wavelength emitted by an atom of atomic number  $Z = 11$  is  $\lambda$  find the atomic number for an atom that emits  $k_\alpha$  radiation with wavelength 43.

- A.  $Z = 6$
- B.  $Z = 4$
- C.  $Z = 11$
- D.  $Z = 44$

**Answer: A**



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67. A photon collides with a stationary hydrogen atom in ground state inelastically. Energy of the order of micro second another photon collides with same hydrogen atom inelastically with an energy of  $15\text{eV}$  what will be observed by the detector?

- A. One photon of energy  $10.2\text{eV}$  and an electron of energy  $1.4\text{eV}$
- B. 2 photon of energy of  $1.4\text{eV}$
- C. 2 photon of energy of  $10.2\text{eV}$
- D. One photon of energy  $10.2\text{eV}$  and another photon of  $1.4\text{eV}$

**Answer: A**



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**68.** A beam of electron is as *YDSE* experiment . The slit width is  $d$  when the velocity of electron is increase , then

- A. no interference is observed
- B. fringer width increases
- C. fringer width decreases
- D. fringer width remain same

**Answer: C**

69. If a star can convert all the  ${}^4\text{He}$  nuclei completely into oxygen nuclei, the energy released per oxygen nucleus is . He nucleus is 4.0026 amu and mass of oxygen nucleus is 15.9994 amu]

A.  $7.6\text{ MeV}$

B.  $56.12\text{ MeV}$

C.  $10.24\text{ MeV}$

D.  $23.9\text{ MeV}$

**Answer: C**

70.  ${}_{87}^{223}\text{Ra}$  is a radioactive substance having half life of 4 days Find the probability that a nucleus undergoes after two half lives

A. 1

B.  $\frac{1}{2}$

C.  $\frac{3}{4}$

D.  $\frac{1}{4}$

**Answer: C**



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**71.** In the option given below , let  $E$  denote the rest mass energy of a nucleus and  $n$  a neutron .The correct option is

A.  $E({}_{92}^{236}\text{U}) > E({}_{92}^{236}\text{I}) + E({}_{92}^{236}\text{Y}) + 2E(n)$

B.  $E({}_{92}^{236}\text{U}) < E({}_{92}^{137}\text{I}) + E({}_{39}^{97}\text{Y}) + 2E(n)$

C.  $E({}_{92}^{236}\text{U}) < E({}_{56}^{140}\text{Ba}) + E({}_{36}^{94}\text{Kr}) + 2E(n)$

D.  $E({}_{92}^{236}\text{U}) = E({}_{56}^{140}\text{Ba}) + E({}_{36}^{94}\text{Kr}) + 2E(n)$

**Answer: A**

72. The largest wavelength in the ultraviolet region of the hydrogen spectrum is 122 nm. The smallest wavelength in the infrared region of the hydrogen spectrum (to the nearest integer) is

A.  $802\text{nm}$

B.  $823\text{nm}$

C.  $1882\text{nm}$

D.  $1648\text{nm}$

**Answer: B**

73. Electrons with de-Broglie wavelength  $\lambda$  fall on the target in an X-rays tube. The cut off wavelength of the emitted X-rays is

A.  $\lambda_0 = (2mc\lambda^2)/h$

B.  $\lambda_0 = \frac{2h}{mc}$

C.  $\lambda_0 = \frac{2m^2c^2\lambda^3}{h^2}$

D.  $\lambda_0 = \lambda$

**Answer: A**



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**74.** Which one of the following statements is *WRONG* in the context of X-rays generated from an X-ray tube?

A. Wavelength of characteristic X-rays decreases when the atomic number of the target increases

B. Cut-off wavelength of the continuous X-rays depends on the atomic number of the target

- C. Intensity of the characteristic X -rays depend on the electrical power given to the X- rays tube
- D. cut - off wavelength of the continuous X- rays depends on the energy of the electrons in the X-rays tube

**Answer: B**



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75. A radioactive sample  $S_1$  having an activity  $5\mu Ci$  has twice the number of inucle as another sample  $S_2$  which has as activity of  $10\mu Ci$  . The half lives of  $S_1$  and  $S_2$  can be

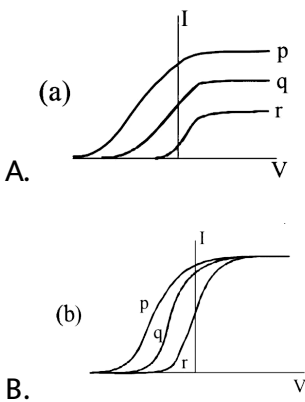
- A. 20years and 5years, respectively
- B. 20years and 10years, respectively
- C. 10yearseatch
- D. 5yearseatch

Answer: A

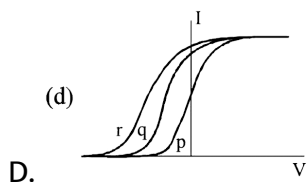
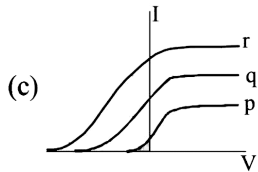


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76. photoelectric affect experiments are performer using ther different medal plates  $p$ ,  $q$  and  $r$  having work function  $\phi_p = 2.0$  eV,  $\phi_q = 2.5$  eV and  $\phi_r = 3.0$  eV respectively. A light beam contains wave lengths  $\lambda \leq 550$  nm,  $450$  nm and  $350$  nm with equal intensities illuminate each of the plate. The correct  $I$  -  $V$  graph for the experiment is [Take  $hc = 1240$  eV nm]







**Answer: A**



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77. The wavelength of the spectral line in the Balmer series of hydrogen atom is  $6561 \text{ \AA}$ . The wavelength of the second spectral line in the Balmer series of singly-ionized beryllium atom is

A.  $1215 \text{ \AA}^2$

B.  $1640 \text{ \AA}^2$

C.  $2430 \text{ \AA}^2$

D.  $4687 \text{ \AA}^2$

**Answer: A**



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**78.** A pulse of light of duration  $100ns$  is absorbed completely by a small object initially at rest. Power of the pulse is  $30mW$  and the speed of light is  $3 \times 10^8 ms^{-1}$ . The final momentum of the object is

A.  $0.3 \times 10^{-17} kgms^{-1}$

B.  $3.0 \times 10^{-17} kgms^{-1}$

C.  $1.0 \times 10^{-17} kgms^{-1}$

D.  $9.0 \times 10^{-17} kgms^{-1}$

**Answer: B**



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79. If  $\lambda_{c\mu}$  is the wavelength of  $K_{\alpha}$  X-rays  $\in$   $eof \cap p$  or  $(a \rightarrow micvंबर29)$  and  $\lambda_{(Mo)}$  is the wave  $\leq n > hofthek(a)$   $X$  - rays  $\in$   $eofmolyndevm(a \rightarrow micvंबर42)$  then ratio  $\lambda_{(Ca)}$   $^{(1 \lambda_{(Mo)})}$  is close to

A. 1.99

B. 2.14

C. 0.50

D. 0.48

**Answer: B**



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80. A metal is illuminated by light of two different wavelength  $248nm$  and  $310nm$ . The maximum speeds of the photoelectron corresponding in these wavelength are  $u_1$  and  $u_2$

respectively. If the ratio  $u_1 : u_2 = 2 : 1$  and  $hc = 1240 \text{ eV nm}$ , the work function of the material is nearly

A.  $3.7 \text{ eV}$

B.  $3.2 \text{ eV}$

C.  $2.8 \text{ eV}$

D.  $2.5 \text{ eV}$

**Answer: A**



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**81.** The electrostatic energy of  $Z$  protons uniformly distributed throughout a spherical nucleus of radius  $R$  is given by

$$E = \frac{3Z(Z-1)e^2}{54\pi\epsilon_0 R}$$

The measured masses of the neutron

${}^1_1\text{H}$ ,  ${}^{15}_7\text{N}$  and  ${}^{15}_8\text{O}$  are  $1.008665u$ ,  $1.007825u$ ,  $15.000109u$  and  $15.0030$

respectively. Given that the ratio of both the  ${}^{12}_7\text{N}$  and  ${}^{15}_8\text{O}$

nuclei are same,  $1u = 931.5 \text{ MeV } c^{-2}$  (at the speed of light) and

$$e^-(2)/(4 \pi \epsilon_0) = 1.44 \text{ MeV}$$

$$f_m A s \sum \in g t \hat{t} h e d \Leftrightarrow e r e n c e b e t w e e n t h e b \in d \in g e 4 \neq r g i e s o f$$

${}_{7}^{15}\text{N}$  and  ${}_{8}^{15}\text{O}$  is purely due to the electric energy, The radius of the nucleus of the nuclei is

A.  $2.85 \text{ fm}$

B.  $3.03 \text{ fm}$

C.  $3.42 \text{ fm}$

D.  $3.80 \text{ fm}$

**Answer: C**



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**82.** An accident in a nuclear laboratory resulting in deposition of a certain amount of radioactive material of half life 18 days inside the laboratory. Tests revealed that the radiation was 64 times more than the permissible level required for safe operation of the laboratory. What is

the minimum number of days after which the laboratory can be considered safe for use?

- A. 64
- B. 90
- C. 108
- D. 120

**Answer: C**



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**83.** The shortest wavelength of X- rays emitted from an X- rays tube depends on

- A. the current in the tube
- B. the voltage applied to the tube
- C. the nature of the gas in tube

D. the atomic number of the target material

**Answer: B::D**



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**84.** The threshold wavelength for photoelectric from a material is  $5200\text{\AA}$   
photoelectric will be emitted when this material is  
illuminated with monochromatic radiation from a

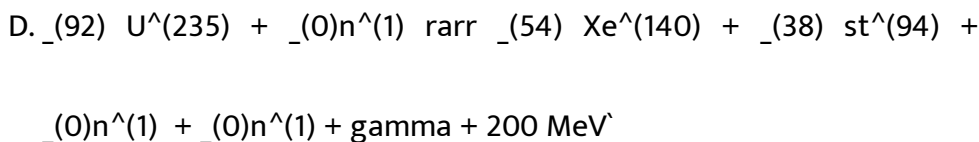
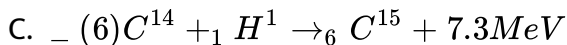
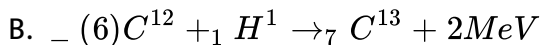
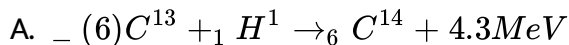
- A. 50 watt infrared lamp
- B. 1 - watt infra- red lamp
- C. 50 watt ultraviolet lamp
- D. 1 - watt ultraviolet lamp

**Answer: C::D**



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85. from the following equation pick out the possible nuclear fission reactions



Answer: B::C



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86. In Bohr's model of the hydrogen atom

A. the radius of the orbit is proportional to  $n^2$

B. the total energy of the electron in  $n$ th orbit is proportional to  $n$



- C. the angular momentum of the electron in an orbit is an integral multiple of  $\frac{h}{2\pi}$
- D. the magnitude of potential energy of the electron in any orbit is greater than in K.E.

**Answer: A::C::D**



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**87. Selected the correct statement from the following**

- A. A diode can be used as a rectifier
- B. A triode cannot be used as a rectifier
- C. the current in a diode is always proportional to the applied voltage
- D. The linear portion of the  $I - V$  characteristic of a triode is used for amplification without distortion

**Answer: A**



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88. for a give a plate voltage , the plate current in a triode valve is maximum when the potential of

- A. the grid is positive and plate is negative
- B. the grid is zero and plate is positive
- C. the grid is negative and plate is positive
- D. the grid is positive and plate is positive

**Answer: D**



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89. The X- rays beam coming from an X- rays tube will be

- A. monochromatic
- B. having all wavelength smaller than a certain maximum wavelength

- C. having all wavelength largest than a certain maximum wavelength
- D. having all wavelength lying between a minimum and a maximum wavelength

**Answer: C**



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**90.** The mass number of a nucleus is

- A. always less than its atomic number
- B. always more than its atomic number
- C. sometimes equal to its atomic number
- D. sometimes more than and sometimes equal to its atomic number

**Answer: C::D**



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91. photoelectric effect supports quantization nature of light because

- A. there is a minimum frequency of light below which no photoelectrons are emitted
- B. the maximum kinetic energy of photoelectrons depends only on the frequency of light and not on its intensity
- C. even on the metal surface is faintly illuminated the photoelectrons leave the surface immediately
- D. electric charge of the photoelectron is quantized

**Answer: A::B::C**



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92. Dividing a negative beta decay

- A. as atomic electron is ejection

- B. as electron which is already present within the nucleuse is ejection
- C. a neclues in the nucleuse decay emitting an electron
- D. a part of the necule the binding energy of the nucleuse is converted into an electron

**Answer: C**



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**93. During a nuclear fasion reaction**

- A. a heavy nucleuse break into two fragments by itself
- B. a light nucleus bomberded by theremal neutrous breaks up
- C. a heavy nucleus bomberded by thermal neudrons break up
- D. two light nucleus combered by thermal and posibly other product

**Answer: D**



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94. The potential different applied to an X-rays tube is increase .As a result , in the emitted radiation

- A. the intersity increase
- B. the minimum wavelength increasres
- C. the intencity remain unchanged
- D. the minimum wavelength decrease

**Answer: C::D**



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95. A freshly prepared radioactive source of half life 2 hr emited radiation of intencity which is 64 times the permissibe level . The minimum time after which it would be possible to work safely with this source is

A.  $6hr$

B.  $12hr$

C.  $24hr$

D.  $128hr$

**Answer: B**



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**96.** impurity atoms with which pure silicon should be doped to make a p - type semiconductor are those of

A. phosphorus

B. boron

C. antimony

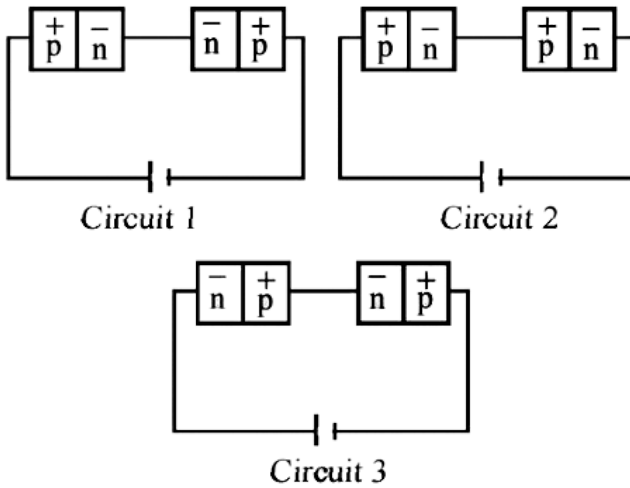
D. aluminium

**Answer: B::D**



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97. Two identical p-n junctions may be connected in series in which a battery in three ways, fig. The potential drops across the two p-n junction are equal in



- A. circuit 1 and circuit2
- B. circuit 2 and circuit3
- C. circuit 3 and circuit1
- D. circuit 1 only

**Answer: B**



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98. The decay constant of a radioacation sample is  $\lambda$  . The half life and mean life of the sample are respectively given by

- A.  $1/\lambda$  and  $(\ln 2)/\lambda$
- B.  $(\ln 2)/\lambda$  and  $1/\lambda$
- C.  $\lambda(\ln 2)$  and  $1/\lambda$
- D.  $\lambda/(\ln 2)$  and  $1/\lambda$

**Answer: B**



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99. When a monochromatic point source of light is at a distance of 0.2 m from a photoelectron cell the cut off voltage and the saturation current are respectively  $0.6V$  and  $18.0mA$  if the same is placed 0.6m away from the photoelectric cell , then

A. the stopping potential will be  $0.2V_o <$

B. the stopping potential will be  $0.6V_o <$

C. the stopping potential will be  $6.0V_o <$

D. the stopping potential will be  $2.0V_o <$

**Answer: B::D**



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**100.** In an n- p-n transistor circuit, the collector current is  $10mA$  if 90 % of the electrons reach the collector.

A. the emitter current will be  $9mA$

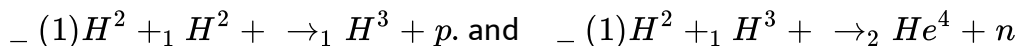
B. the base current will be  $1mA$

C. the emitter current will be  $11mA$

D. the base current will be  $-1mA$

**Answer: B::C**

101. A star initially has  $10^{40}$  deuterons it product energy via the process



If the deuteron supply of the average power radiated by the state is  $10^{16} \text{ W}$ , the deuteron supply of the state is exhausted in a time of the order of .

The masses of the nuclei are as follows:

$$M({}_1^2\text{H}) = 2.014 \text{ a}\mu,$$

$$M(p) = 1.007 \text{ amu}, M(n) = 1.008 \text{ amu}, M({}_2^4\text{He}) = 4.001 \text{ amu}.$$

A.  $10^6 \text{ s.}$

B.  $10^8 \text{ s.}$

C.  $10^{12} \text{ s.}$

D.  $10^{16} \text{ s.}$

**Answer: C**

102. photons of energy  $4.25\text{eV}$  strike the surface of metal A, the ejection photoelectric have maximum kinetic energy  $T_A\text{eV}$  if the de Broglie wavelength of these photoelectron is  $\lambda_B = 2\lambda_A$ , then

A. The work function of A is  $2.25\text{eV}$

B. The work function of B is  $4.20\text{eV}$

C.  $T_A = 2.00\text{eV}$

D.  $T_B = 2.75\text{eV}$

Answer: A::B::C



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103. which of the following statement (s) is (are) correct ?

- A. The rest mass of a stable nucleus is less than the sum of the rest masses of its separated nucleons
- B. The rest mass of a stable nucleus is greater than the sum of the rest masses of its separated nucleons
- C. In nuclear fission, energy is released by fusing two nuclei of medium mass (approximately  $100a\mu$ )
- D. In nuclear fission, energy is released by fragmentation of a very heavy nucleus

**Answer: A::D**



**Watch Video Solution**

**104.** Holes are charge carriers in

- A. intrinsic semiconductors
- B. ionic solids

C. p- type semiconductor

D. intrinsic

**Answer: B::C**



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**105.** A transistor is used in the common emitter mode as an amplifier ,

Then

A. the base - emitter junction is forward - biased

B. the base - emitter junction is reverse- biased

C. the input signal is connected in series with the voltage applied to  
the base- emitter junction

D. the input signal is connected in series with the voltage applied to  
the base- collector junction

**Answer: A::C**

**106.** Let  $m_p$  be the mass of a proton,  $M_1$  the mass of a  $_{10}^{20}\text{Ne}$  nucleus and  $M_2$  the mass of a  $_{20}^{40}\text{Ca}$  nucleus. Then

A.  $M_2 = 2M_1$

B.  $M_2 > 2M_1$

C.  $M_2 < 2M_1$

D.  $M_1 < 10(m_n + m_p)$

**Answer: C::D**

**107.** The electron in a hydrogen atom makes a transition  $n_1 \rightarrow n_2$  where  $n_1$  and  $n_2$  are the principal quantum numbers of the two states. Assume the Bohr model to be valid. The time period of the electron in the initial

state is eight times that in the state . The possible values of  $n_1$  and  $n_2$  are

A.  $n_1 = 4, n_2 = 2$

B.  $n_1 = 8, n_2 = 2$

C.  $n_1 = 8, n_2 = 1$

D.  $n_1 = 6, n_2 = 3$

**Answer: A::D**



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**108.** The half-life of  $^{131}_{51}\text{I}$  is days. Given a sample of  $^{131}_{51}\text{I}$  at time  $t = 0$ , we get that

A. no nucleus will decay  $t = 4\text{days}$

B. no nucleus will decay  $t = 8\text{days}$

C. no nucleus will decay  $t = 16\text{days}$



D. a given nucleus may decay at any time after  $t = 0$

**Answer: D**



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**109.** In a p- n junction diode not connected to any circuit,

- A. the potential is the same everywhere
- B. the p- type side is at a higher potential than the n - type side
- C. there is an electric field at the junction directed from the p- type side to the n - type side
- D. there is an electric field at the junction directed from the n - type side to the p- type side

**Answer: C**



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110. X- rays are produced in an X- rays tube operating at a given accelerating voltage . The wavelength of the continuous X- rays has values from

A.  $0 \rightarrow \infty$

B.  $\lambda_{\min} \rightarrow \infty$  where  $\lambda_{\min} > 0$

C.  $0 \rightarrow \lambda_{\max}$  where  $\lambda_{\max} < \infty$

D.  $\lambda_{\min} \rightarrow \lambda_{\max}$  where  $0 < \lambda_{\min} < \lambda_{\max} < \infty$

**Answer: B**



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111. The work function of a substance is  $4.0\text{eV}$  The longest wavelength of light that can cause photoelectron emission from this substance is approximately

A.  $540\text{nm}$

B.  $400nm$

C.  $310nm$

D.  $220nm$

**Answer: C**



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**112.** The half - life period of a radioactive element X is same as the mean - life time of the another radicoactive electront Y initial both of then the same number of atom . Then

A. X and Y have the same decay rate initial ly

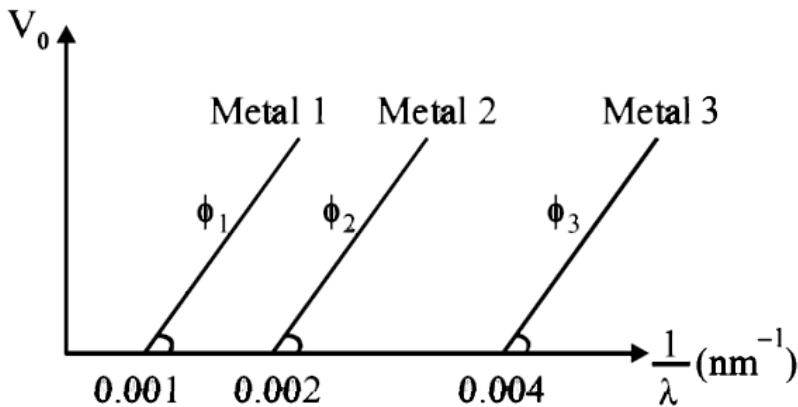
B. X and Y dacay at the same decay rate always

C. Y will dacay at a faster rate then X

D. X will dacay at a faster rate then Y

**Answer: C**

113. The graph between the stopping potential ( $V_0$ ) and  $\left(\frac{1}{\lambda}\right)$  is shown in the figure  $\phi_1$ ,  $\phi_2$  and  $\phi_3$  are work function, which of the following is //are correct



A.  $\phi_1 : \phi_2 : \phi_3 = 1 : 2 : 4$

B.  $\phi_1 : \phi_2 : \phi_3 = 4 : 2 : 1$

C.  $\tan \theta \propto \frac{hc}{\epsilon}$

D. ultraviolet light can be used to emit photoelectrons from metal 2 and

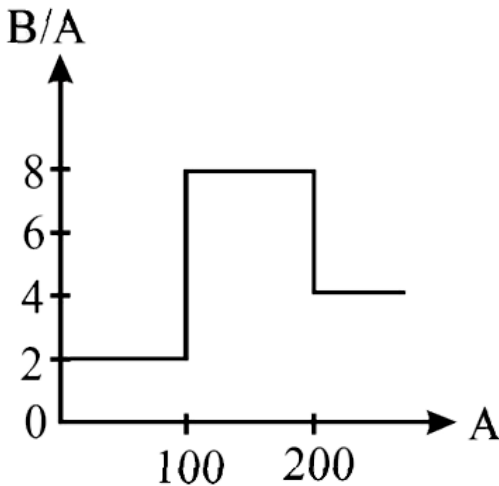
3only

Answer: A::C



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114. Assume that the nucleus binding energy per nucleus ( $B/A$ ) versus mass number ( $A$ ) is as shown in the figure. Use this plot to choose the energy the correct (s) given below



A. fusion of two nucleus with mass number typing in the range of  $1 < A < 50$  will release energy

- B. fusion of two nucleus with mass number typing in the range of  $51 < A < 100$  will release energy
- C. fusion of a nucleus typing in the mass of  $100 < A < 200$  will release energy when broken into two equal fragments
- D. fusion of a nucleus typing in the mass range of  $200 < A < 260$  will release energy when broken into two equal fragments

**Answer: B::D**



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**115.** The radius of the orbit of an electron in a Hydrogen - like atom is  $4.5s_0$  where  $s_0$  is the bohr radius its orbital angular momentum is  $\frac{3b}{2\pi}$  it is given that  $b$  is plank constant and  $R$  is rabdery constant .The possible wavelength ( $\lambda$ ) , when the atom de- exciter , is (are)

A.  $\frac{9}{32R}$

B.  $\frac{9}{16R}$

C.  $\frac{9}{5R}$

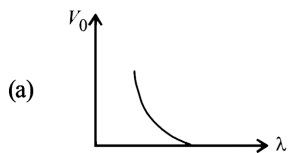
D.  $\frac{4}{3R}$

**Answer: A::C**

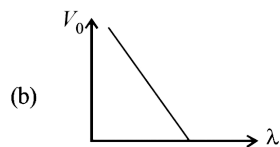


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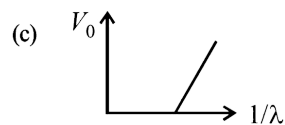
**116.** For photo - electric effect with incident photo wavelength  $\lambda$  the stopping is  $V_0$  identify the correct variation(s) of  $V_0$  with  $\lambda$  and  $1/\lambda$



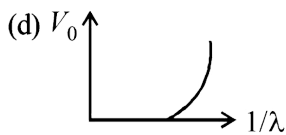
A.



B.



C.



D.

**Answer: A::C**



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117. A fission reaction is given by  ${}_{92}^{236}\text{U} \rightarrow {}_{54}^{140}\text{Xe} + {}_{38}^{94}\text{St} + x + y$ , where  $x$  and  $y$  are two particle Consider  ${}_{92}^{236}\text{U}$  to be at rest, the kinetic energies of the products are denoted by  $k_{xe}, K_{st}, K_s(2\text{MeV})$  and respectively. Let the binding energy per nucleus of  ${}_{92}^{236}\text{U}, {}_{54}^{140}\text{Xe}$  and  ${}_{38}^{94}\text{St}$  be  $7.5\text{MeV}, 8.4\text{MeV}$  and  $8.5\text{MeV}$ , respectively Considering different conservation laws, the correct option (s) is (are)

A.  $x = n, y = n, K_{St} = 129\text{MeV}, K_{xe} = 86\text{MeV}$

B.  $x = p, y = e^-, K_{St} = 129\text{MeV}, K_{xe} = 86\text{MeV}$

C.  $x = p, y = n, K_{St} = 129\text{MeV}, K_{xe} = 86\text{MeV}$



D.  $x = n, y = n, K_{St} = 86MeV, K_{xe} = 129MeV$

**Answer: A**



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**118.** High excited state for hydrogen - like atom (also called rydberg quantum number  $n$ , where  $n \geq 1$ ) which of the following statement is (are) true?

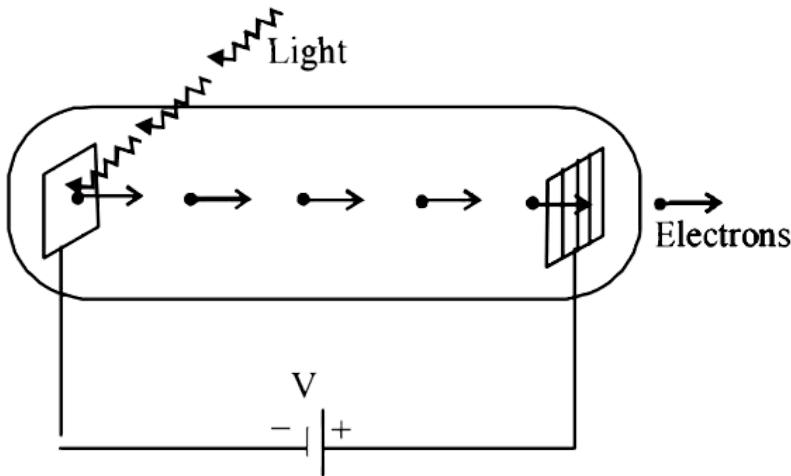
- A. Relative change in the radii of two consecutive orbits does not depend on  $Z$
- B. Relative change in the radii of two consecutive orbits varies as  $1/n$
- C. Relative change in the energy of two consecutive orbits varies as  $1/n^3$
- D. Relative change in the angular momenta of two consecutive orbits varies as  $1/n$

Answer: A::B::D



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119. Light of wavelength  $\lambda_{ph}$  falls on a plate a vacuum tube as shown in the figure. The work function of the conducting material kept at a distance  $d$  from the cathode A potential difference  $V$  is maximum between the electrodes if the minimum de Broglie wavelength of the electrons passing through the anode is  $\lambda_e$  which of the following statement (s) is (are) true?



A.  $\lambda_e$  decrease with increase in  $\phi$  and  $\lambda_{ph}$

B.  $\lambda_e$  is approximately halved, if  $d$  is doubled

C. for large potential difference ( $V \gg \phi/e$ ).  $\lambda_e$  is approximately halved if  $V$  is made four times

D.  $\lambda_e$  increase at the same rate as  $\lambda_{ph} f$  or  $\lambda_{ph} < hc/\phi$

Answer: C



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120. A single electron orbits around a stationary nucleus of charge  $+Ze$

when  $Z$  is a constant and  $e$  is the magnitude of the electronic charge if

47.2 eV excites the  $\leq$  electron to the second Bohr orbit  $\rightarrow$  the third Bohr orbit  $\rightarrow$  the fourth Bohr orbit (iii) The wave  $\leq n > h$  of the  $\leq$  electron  $\neq$  the potential  $\neq$  the potential  $\neq$  the angular momentum of the  $\neq$  the hydro  $\geq na \rightarrow m$

$= 13.6 \text{ eV}$  Bohr radius  $= 5.3 \times 10^{-11} \text{ m}$  velocity of light  $= 3 \times 10^8$

$\text{m/sec}$  Planck's constant  $= 6.6 \times 10^{-34} \text{ J} \cdot \text{sec}$



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**121.** Hydrogen atom in ground state is excited by means of monochromatic radiation of wavelength  $975\text{\AA}$ . How many different lines are possible in the resulting spectrum? Calculate the largest energy for hydrogen atom as  $13.6\text{eV}$



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**122.** How many electrons and mass number in a nucleus of atomic number 11 and mass 24?

(i) number of electrons = (ii) number of protons = (iii) number of neutrons =



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**123.** The energy needed to detach the electron of a hydrogen-like ion in ground state is a system (a) what is the wavelength of the radiation emitted when the electron jumps from the first excited state to the ground state? (b) What is the radius of the orbit for this atom?



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- 124.** A double ionised lithium atom is hydrogen like with atomic number 3
- (i) Find the wavelength of the radiation to excite the electron in  $Li^{++}$  from the first to the third bohr orbit (Ionisation energy of the hydrogen atom equals 13.6 eV)
- (ii) How many spectral lines are observed in the emission spectrum of the above excited system ?



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**125.** A triode has plate characteristics in the form of parallel lines in the region of our interest. At a given grid voltage  $V_g$  (in volts) by the algebraic relation

$$I_p = 0.125V_p - 7.5$$

for grid voltage  $V_g = -3$  volts,  $I_p = 5$  milliamperes, determine the plate resistance ( $r_p$ ), transconductance ( $g_m$ ) and the amplification factor ( $\mu$ ) for the triode.



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**126.** A particle of charge equal to that of an electron -  $e$ , and mass in a nucleus the charge  $+3e$  (Take the mass of the nucleus to be infinite Assuming that the bohr model of the atom is applicable to this system

- (i) Derive an expression for the radius of the bohr orbit
- (ii) find the value of  $n$  which the radius is approximated the same as that of the bohr orbit for the hydrogen atom
- (iii) find the wavelength of the radiation emitted when the  $\mu^-$  meson jump from the third orbit of the first orbit



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**127.** A gas of identical hydrogen-like atoms has some atoms in the lowest in lower (ground) energy level  $A$  and some atoms in a particular upper (excited) energy level  $B$  and there are no atoms in any other energy level. The atoms of the gas make transition to higher energy level by absorbing monochromatic light of photon energy  $2.7eV$ .

Subsequently, the atom emit radiation of only six different photon

energies. Some of the emitted photons have energy  $2.7\text{eV}$  some have energy more, and some have less than  $2.7\text{eV}$ .

a Find the principal quantum number of the initially excited level  $B$

b Find the ionization energy for the gas atoms.

c Find the maximum and the minimum energies of the emitted photons.



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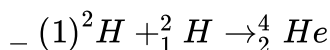
**128.** Electrons in hydrogen like atom ( $Z = 3$ ) make transition from the fifth to the fourth orbit and from the third orbit. The resulting radiation is incident normally on a metal plate and ejects photoelectrons. The stopping potential for the photoelectrons ejected by the longer wavelength

(Rydberg constant  $= 1.094 \times 10^7 \text{m}^{-1}$ )



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**129.** It is proposed to use nuclear fusion



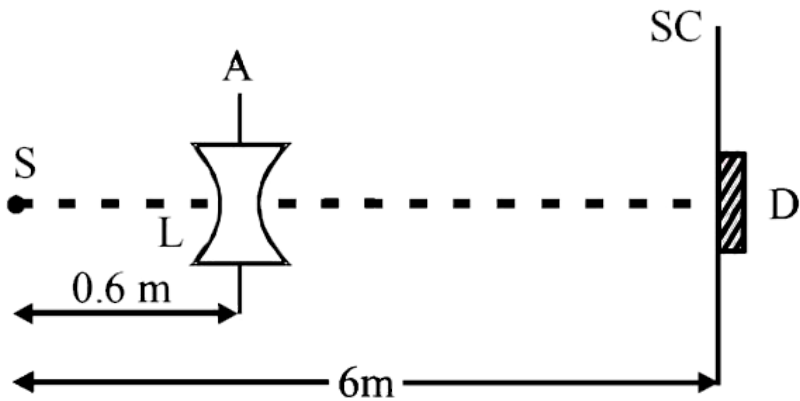
in a nucleus of  $200\text{MW}$  rating if the energy from the above reaction is used with a 25 per cent efficiency in the reactor, how many grams of deuterium fuel will be needed per day (The masses of  ${}_1^2\text{H}$  and  ${}_4^2\text{He}$  are 2.0141 atomic mass unit and 4.0028 atomic mass unit respectively)



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**130.** A monochromatic point source radiating wavelength  $6000\text{\AA}$  with power  $2\text{ W}$  has an aperture A of diameter  $0.1\text{ m}$  and a large screen SC are placed as shown in fig, A photoemissive detector D of surface area  $0.5\text{ cm}^2$  is placed at the centre of the screen. The efficiency of the detector for the photoelectron generation per incident photon is 0.9



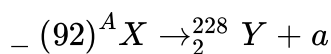


- (a) Calculate the photon flux at the centre of the screen and the photocurrent in the detector.
- (b) If the concave lens  $L$  of focal length  $0.6\text{ m}$  is inserted in the aperture as shown, find the new values of photon flux and photocurrent. Assume a uniform average transmission of  $80\%$  from the lens.
- (c) If the work function of the photoemissive surface is  $1\text{ eV}$ , calculate the values of the stopping potential in the two cases (without and with the lens in the aperture).



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**131.** A nucleus X, initially at rest, undergoes alpha decay according to the equation,



(a) Find the value of  $A$  and  $Z$  in the above process.

(b) The alpha particle produced in the above process is found to move in a circular track of radius  $0.11\text{m}$  in a uniform magnetic field of  $3\text{ Tesla}$  find the energy (in MeV) released during the process and the binding energy of the parent nucleus X

Given that :  $m({}_2^4\text{He}) = 4.0026\text{u}$ ,  $m({}_1^1\text{H}) = 1.0078\text{u}$ .

$m({}_2^4\text{He}) = 4.0026\text{u}$ ,  $m({}_1^1\text{H}) = 1.0078\text{u}$



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**132.** Light from a discharge tube containing hydrogen atoms falls on the surface of a piece of sodium. The kinetic energy of the fastest photoelectrons emitted from sodium is  $0.73\text{eV}$ . The work function for sodium is  $1.82\text{eV}$  find

(a) the energy of the photons causing the photoelectric emission,

(b) the quantum number of the two level involved in the emission of these photons,

(c) the change in the angular momentum of the electron in the hydrogen atom in the above transition and,

(d) the recoil speed of emitted atom assuming it to be at rest before the transition.

(Ionization potential of hydrogen is  $13.6\text{eV}$ )



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**133.** A small quantity of solution containing  $Ne^{24}$  radio nuclide (half life =  $15\text{hour}$ ) of activity  $1.0\text{ microcurie}$  is injected into the blood of a person. A sample of the blood of volume  $1\text{cm}^3$  taken after  $5\text{ hour}$  shows an activity of the blood in the body of the person. Assume that radioactive solution mixed uniformly in the blood of the person ( $1\text{curie} = 3.7 \times 10^{10}$  disintegrations per second)



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**134.** A hydrogen like atom (atomic number  $Z$ ) is in a higher excited state of quantum  $n$ , The excited atom can make a two photon of energy  $10.2$  and  $17.0\text{eV}$  respectively, Alternately the atom from the same excited state by successively emitting two photons of energies  $4.25\text{eV}$  and  $5.95\text{eV}$  respectively

Determine the value of  $n$  and  $Z$  (Ionization energy of H-atom  $= 13.6\text{eV}$ )



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**135.** An electron in a hydrogen-like atom, is in an excited state it has a total energy of  $-3.45\text{eV}$  Calculate (i) the kinetic energy and (ii) the de Broglie wavelength of the electron.



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**136.** At a given instant there are  $25\%$  undecayed radio-active nuclei in a sample. After  $10$  seconds the number of undecayed nuclei reduces to  $12.5\%$  Calculate (i) mean-life of the nucleus, and (ii) the time in which

the number of undecayed nuclei will further to 6.25 % of the reduced number .



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**137.** Assume that the de Broglie wave associated with an electron can can from a standing wave between the atome arrange in a one dimensional array with nodes at each of the atomic sites it is found that one such standing wave if the distance  $d$  between the aloms of the arry is  $2M139waveisaga \in f$  or  $mad$  if  $dis \in creased \rightarrow 2.5\text{\AA}$  . A similar standing in the distance if  $d$  find the energy of the electrons velts and tghe least value of  $d$  for which the standing wave type described above can from .



Watch Video Solution

**138.** The elecron curium  $_{96}^{248}\text{Cm}$  On has a mean life of  $10^{13}$  second ist pirmary dacay mode with a probilly of 8 % and the letter with a probilly of 92 % Each fission released  $200\text{MeV}$  of energy . The masses involved in

a - decay are as follows

$$_{94}^{248}\text{Cm} = 248.072220u, \text{ }_{94}^{244}\text{Pu} = 244.064100u \text{ and } {}_2^4\text{He} = 4.002603u$$

calculate the power output from a sample of

$$10^{20}\text{Cm} \rightarrow m(1u = 931\text{MeV}/e^2)$$



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**139.** Nuclei of a radioactive  $A$  are being produced at a constant rate  $\alpha$ .

The element has a decay constant  $\lambda$ . At time  $t = 0$  there are  $N_0$  nuclei of the element.

(a) calculate the number  $N$  of nuclei of  $A$  at same  $t$

(b) if  $\alpha = 2\lambda N_0$  calculate the number of nuclei of  $A$  after  $t = \frac{1}{\lambda} \ln 2$  if  $\alpha = \lambda N_0$  and the limit value of  $N$  as  $t \rightarrow \infty$



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**140.** photoelectrons are emitted when  $40\text{nm}$  radiation is incident on a surface of work function  $1.9\text{eV}$  These photoelectron pass through a region containing a  $\alpha$ -particle. A maximum energy electron combines with an

$\alpha$  - particle to form a  $He^+$  ion emitting a single photon in this process  
 $He^+$  ions thus formed are in their fourth excited state find the energies  
 in eV of the photons typing is the 2 to  $4eV$  range, that are likely to be  
 emitted during and after the combination  $[Take h = 4014 \times 10^{-15} eVs]$



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**141.** A hydrogen-like atom of atomic number  $Z$  is in an excited state of  
 quantum number  $2n$  it can emit a maximum energy photon of energy  
 $40.8 eV$  is emitted  $f \in dn, Z$  and the ground state  $e \neq rgy (\in eV) f$  or this a  
 – exclamation, Ground state  $e \neq rgy$  of hydro  $\geq na \rightarrow mis - 13.6 eV$



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**142.** when a beam of  $10.6 eV$  photons of intensity  $2.0 W/m^2$  falls on a  
 platinum surface of area  $1.0 \times 10^4 m^2$  and work function  $5.6 eV$ ,  $0.53 \%$   
 of the incident photons eject photoelectrons find the number of  
 photoelectrons emitted per second and their minimum energies (in  
 eV) Take  $1 eV = 1.6 \times 10^{-19} J$





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143. In a nuclear reaction  $^{235}\text{U}$  undergoes fission liberating  $200 \text{ MeV}$  energy. There are  $10\%$  efficiency and produces  $1000 \text{ MW}$  power. If the reactor is to function for 10 years. Find the total mass of required.



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144. A nucleus at rest undergoes a decay emitting an  $\alpha$  particle of de Broglie wavelength  $\lambda = 5.76 \times 10^{-15} \text{ m}$  if the mass of particle is  $4.002 \text{ amu}$ , determine the total kinetic energy in the final state. Hence, obtain the mass of the parent nucleus in amu ( $1 \text{ amu} = 931.470 \text{ MeV}/c^2$ ).



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145. A radioactive nucleus  $X$  decays to a nucleus  $Y$  with a decay constant  $\lambda_x = 0.1 \text{ s}^{-1}$ ,  $Y$  further decays to a stable nucleus  $Z$  with a



decay constant  $\lambda_y = 1/30\text{s}^{-1}$  initially, there are only X nuclei and their number is  $N_0 = 10^{20}$  set up the rate equations for the population of  $X \rightarrow Y$  and  $Z$  The population of Y nucleus as a function of time is given by

$$N_y(t) = N_0 \lambda_x / (\lambda_x - \lambda_y) \left( \exp(-\lambda_y t) - \exp(-\lambda_x t) \right)$$
  
 is maximum and determine the population of X and Y at that instant.



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**146.** A hydrogen-like atom (described by the Bohr model) is observed to emit six wavelengths, originating from all possible transitions between  $-0.85\text{eV}$  and  $-0.544\text{eV}$  (including both these values)

(a) Find the atomic number of the atom

(b) Calculate the smallest wavelength emitted in these transitions.

(Take  $hc = 1240\text{eV} \cdot \text{nm}$ , ground state energy of hydrogen atom  $= 13.6\text{eV}$ )



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**147.** Two metallic plate  $A$  and  $B$  , each of area  $5 \times 10^{-4} m^2$ , are placed parallel to each at a separation of  $1cm$  plate  $B$  carries a positive charge of  $33.7 \times 10^{-12} C$  A monocharomatic beam of light , with photoes of energy  $5eV$  each , starts falling on plate  $A$  at  $t = 0$  so that  $10^{16}$  photons fall on it per sqare mater for every  $10^6$  incident photons fall on it per square meter per second Assume that one photoelectron is emitted for every  $10^6$  incident photons . Also assume that all the emitted photoelectron are collected by plate  $B$  and the work function of plate  $A$  remain constant at the value  $2eV$  Determine

- (a) the number of photoelectrons emitted up to  $i = 10s$ ,
- (b) the magnitude of the electron field between the plate  $A$  and  $B$  at  $i = 10s$ , and
- (c ) the kinetic energy of the most energotic photoelectrons emitted at  $i = 10s$  whenit reaches plate  $B$

Negilect the time taken by the photoelectrons to reach plate

$B$  Take  $\epsilon_0 = 8.85 \times 10^{-12} C^2 N - m^2)$



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**148.** frequency of a photon emitted due to transition of electron of a certain element from  $L \rightarrow K$  shell is found to be  $4.2 \times 10^{18} \text{ Hz}$  using moseley's law, find the atomic number of the element , given that the Rydberg's constant  $R = 1.1 \times 10^7 \text{ m}^{-1}$



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**149.** A radioactive sample emit  $n\beta$  - particle is 2 sec , in next  $5eV$  sec it emit 0.75 n beta`- particle , what is the mean life of the sample?



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**150.** In a photon electric experiment set upm , photons of energy  $5eV$  falls on the cathode having work function  $3eV$  (a) if the seturation current is  $i = 4\mu A$  for intensity  $10^{-5} \text{ W/m}^2$ , then plot a graph between anode potential and current (b) Also draw a graph for intensity of incident radiation  $2 \times 10^{-5} \text{ W/m}^2$



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151. A radioactive sample of  $^{238}\text{U}$  decays to Pb through a process for which the half-life is  $4.5 \times 10^9$  year. Find the ratio of number of nuclei of Pb to  $^{238}\text{U}$  after a time of  $1.5 \times 10^9$  year. Given  $(2)^{19} = 1.26$



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152. The photons from the Balmer series in Hydrogen spectrum having wavelength between  $450\text{nm}$  to  $700\text{nm}$  are incident on a metal surface of work function  $2\text{eV}$ . Find the maximum kinetic energy of ejected electron. (Given  $hc = 1242\text{ eV nm}$ )



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153. The potential energy of a particle of mass  $m$  is given by

$$V(x) = \lambda_1 x^2 \quad \text{and} \quad \lambda_2 x^3$$

are the de-Broglie wavelength and the number of de-Broglie waves associated with the particle, when  $x = \lambda_1$  and  $x = \lambda_2$

and  $\lambda_1$  and  $\lambda_2$  respectively, if the  $\rightarrow$  tale  $\neq$  rgy of partic  $\leq$  is  $E_-(0)$  find

$$\lambda_1 / \lambda_2$$



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**154.** Highly energetic electron are bombarded in a target of an element containing 30 neutrons The ratio of nucleus to that of Helium nucleus is  $(14)^{1/3}$ . Find (a) atomic number of the nucleus (b) the frequency of  $k_\alpha$  line of the X-rays produced ( $R = 1.1 \times 10^7 m^{-1}$  and  $c = 3 \times 10^8 m/s$ )



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**155.** In hydrogen - like atom ( $z = 1$ ) with line of Lyman series has wavelength  $\lambda$  the de - broglie's wavelength of electron in the level from which it originated is also  $\lambda$  Find the value of  $n$  ?



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156. In a mixture of  $H - He^+$  gas ( $He^+$  is singly ionized He atom),  $H$  atom and  $He^+$  ions are excited to their respective first excited state subsequently  $H$  atoms transfer their total excitation energy to  $He^+$  ions (by collisions) Assume that the Bohr model of atom is exactly valid.

The quantum number  $n$  of the state finally populated in  $He^+$  ions is -

A. 2

B. 3

C. 4

D. 5

**Answer: C**



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157. In a mixture of  $H - He^+$  gas ( $He^+$  is singly ionized He atom),  $H$  atom and  $He^+$  ions are excited to their respective first excited state subsequently  $H$  atoms transfer their total excitation energy to  $He^+$  ions

(by collisions) Assume that the bohr model of atom is exactly veld.

The wavelength of light emitted in the visible region by  $He^+$  ions after collisions with  $H$  atoms is -

A.  $6.5 \times 10^{-7} m$

B.  $3.6 \times 10^{-7} m$

C.  $4.8 \times 10^{-7} m$

D.  $4.0 \times 10^{-7} m$

**Answer: C**



**Watch Video Solution**

**158.** In a mixture of  $H - He^+$  gas ( $He^+$  is singly ionized He atom),  $H$  atom and  $He^+$  ions are excited to their respective first excited state subsequently  $H$  atoms transfer their total excitation energy to  $He^+$  ions (by collisions) Assume that the bohr model of atom is exactly veld.

The ratio of the kinetic energy of the  $n = 2$  electron for the  $H$  atom to the of  $He^+$  ion is -

A. 14

B.  $1/2$

C. 1

D. 2

**Answer: A**



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**159.** Scientists are working hard to develop nuclear fusion reactor. Nuclei of heavy hydrogen,  ${}_1^2\text{H}$ , known as deuterium and denoted by  $D$ , can be thought of as a candidate for fusion reactor. The  $D - D$  reaction is  ${}_1^2\text{H} + {}_1^2\text{H} \rightarrow {}_2^3\text{He} + n + e^- + \text{energy}$  in the core of fusion reactor. A gas of heavy hydrogen of  ${}_1^2\text{H}$  nuclei and electrons is known as plasma. The nuclei move randomly in the reactor to take place. Usually, the temperature in the reactor core is too high and too natural will not be used to confine the plasma for a time  $10^{-9}$  before the particles fly away from the case if  $n$  is the density (number per volume) of determines,



the product  $nt_0$  is called Lawson number in one of the criteria, a reactor is termed successful if Lawson number is greater than  $5 \times 10^{14} \text{ s/cm}^2$

it may be helpful to use the following Boltzmann constant

$$\lambda = 8.6 \times 10^{-5} \text{ eV/k}, \frac{e^2}{4\pi\epsilon_0} = 1.44 \times 10^{-9} \text{ eVm}$$

in the case of nucleus fusion reactor, the gas becomes plasma because of

- A. strong nucleus force acting between the deuterons
- B. coulomb force acting between the deuterons
- C. coulomb force acting between deuteron - electron pairs
- D. the high temperature maintained inside the reactor

**Answer: D**



**Watch Video Solution**

**160.** Scientists are working hard to develop nuclear fusion reactor. Nuclei of heavy hydrogen,  ${}_1^2\text{H}$ , known as deuterium and denoted by  $D$ , can be thought of as a candidate for fusion reactor. The  $D - D$  reaction is  ${}_1^2\text{H} + {}_1^2\text{H} \rightarrow {}_2^3\text{He} + n + e^- + \gamma$  in the core of fusion reactor and

gas of heavy hydrogen of  $(1)^2H$  nuclei and electrons is known as plasma. The nuclei move randomly in the reactor to take place. Usually, the temperature in the reactor core is too high and too natural will can be used to confine the plasma for a time  $t_0$  before the particles fly away from the case if  $n$  is the density (number volume) of determines, the product  $nt_0$  is called Lawson number in one of the criteria, a reactor is termed successful if Lawson number is greater than  $5 \times 10^{14} s/cm^2$

it may be helpful to use the following Boltzmann constant

$$\lambda = 8.6 \times 10^{-5} eV/k, \frac{e^2}{4\pi\epsilon_0} = 1.44 \times 10^{-9} eVm$$

Assume that two deuteron nuclei in the core of fusion reactor at temperature energy  $T$  are moving toward each other, each with kinetic energy  $1.5kT$ , when the separation between them is large enough to neglect Coulomb potential energy. Also neglect any interaction from other particles in the core. The minimum temperature  $T$  required for them to reach a separation of  $4 \times 10^{-15} m$  is in the range

A.  $1.0 \times 10^9 K < T < 2.0 \times 10^9 K$

B.  $2.0 \times 10^9 K < T < 3.0 \times 10^9 K$

C.  $3.0 \times 10^9 K < T < 4.0 \times 10^9 K$

$$D. 4.0 \times 10^9 K < T < 5.0 \times 10^9 K$$

**Answer: A**



**Watch Video Solution**

**161.** Scientists are working hard to develop nuclear fusion reactor. Nuclei of heavy hydrogen,  ${}_1^2H$ , known as deuterium and denoted by  $D$ , can be thought of as a candidate for fusion reactor. The  $D - D$  reaction is  ${}_1^2H + {}_1^2H \rightarrow {}_2^4He + n + e + \text{energy}$  in the core of fusion reactor a gas of heavy hydrogen of  ${}_1^2H$  nuclei and electrons is known as plasma. The nuclei move randomly in the reactor to take place. Usually, the temperature in the reactor core are too high and too natural will can be used to confine the plasma for a time  $10^{-8}$  before the particles fly away from the case if  $n$  is the density (number per volume) of deuterium, the product  $nt_0$  is called Lawson number in one of the criteria, a reactor is termed successful if Lawson number is greater than  $5 \times 10^{14} s/cm^2$  it may be helpful to use the following Boltzmann constant

$$\lambda = 8.6 \times 10^{-5} eV/k, \frac{e^2}{4\pi\epsilon_0} = 1.44 \times 10^{-9} eVm$$

Result of calculations for four different designs of a fusion reactor using  $D - D$  reaction are given below which of these is most promising based on Lawson criterion ?

A. deuteron density  $= 2.0 \times 10^{12} \text{ cm}^{-3}$ ,

confinement time  $= 5.0 \times 10^{-3} \text{ s}$

B. deuteron density  $= 8.0 \times 10^{14} \text{ cm}^{-3}$ ,

confinement time  $= 9.0 \times 10^{-1} \text{ s}$

C. deuteron density  $= 4.0 \times 10^{23} \text{ cm}^{-3}$ ,

confinement time  $= 1.0 \times 10^{-11} \text{ s}$

D. deuteron density  $= 1.0 \times 10^{24} \text{ cm}^{-3}$

confinement time  $= 4.0 \times 10^{-12} \text{ s}$

**Answer: B**



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**162.** When a particle is restricted to move along x- axis between  $x = 0$  and  $x = a$  where  $a$  is of nanometer dimension, its energy can take only certain specific values. The allowed energies of the particles only in such a restricted region, correspond to the formation of standing wave with nodes at its end  $x = 0$  and  $x = a$ . The wavelength of this standing wave is related to the linear momentum  $p$  of the particle according to the de Broglie relation. The energy of the particle of mass  $m$  is related to its linear momentum as

$E = \frac{p^2}{2m}$ . Thus, the energy of the particle can be denoted by a quantum number  $n$  taking value  $1, 2, 3, \dots$  ( $n = 1$ , called the ground state)

corresponding to the number of loops in the standing wave use the model described above to answer the following three questions for a particle moving in the line

$x = 0 \rightarrow x = a$  Take  $h = 6.6 \times 10^{-34} \text{ Js}$  and  $e = 1.6 \times 10^{-19} \text{ C}$

The allowed energy for the particle for a particular value of  $n$  is proportional to

A.  $a^{-2}$

B.  $a^{-3/2}$

C.  $a^{-1}$

D.  $a^2$

**Answer: A**



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**163.** When a particle is restricted to move along x- axis between  $x = 0$  and  $x = a$  where  $a$  is of nanometer dimension, its energy can take only certain specific values. The allowed energies of the particles only in such a restricted region, correspond to the formation of standing wave with nodes at its end  $x = 0$  and  $x = a$ . The wavelength of this standing wave is related to the linear momentum  $p$  of the particle according to the de Broglie relation. The energy of the particle of mass  $m$  is related to its linear momentum as

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corresponding to the number of loops in the standing wave use the model described above to answer the following there question for a particle moving in the line

$$x = 0 \rightarrow x = a \text{ Take } h = 6.6 \times 10^{-34} Js \text{ and } e = 1.6 \times 10^{-19} C$$

If the mass of the particle is  $m = 1.0 \times 10^{-30} kg$  and  $a = 6.6 nm$  the energy of the particle in its ground state is closest to

A.  $0.8 meV$

B.  $8 meV$

C.  $80 meV$

D.  $800 meV$

**Answer: B**



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**164.** When a particle is restricted to move along x- axis between  $x = 0$  and  $x = a$  where  $a$  is of nanometer dimension, its energy can take only certain specific values. The allowed energies of the particles only

in such a restricted region, correspond to the formation of standing wave with nodes at its end  $x = 0$  and  $x = a$ . The wavelength of this standing wave is related to the linear momentum  $p$  of the particle according to the de Broglie relation. The energy of the particle of mass  $m$  is related to its linear momentum as

$E = \frac{p^2}{2m}$ . thus, the energy of the particle can be denoted by a quantum number  $n$  taking value  $1, 2, 3, \dots$  ( $n = 1$ , called the ground state) corresponding to the number of loops in the standing wave use the model described above to answer the following question for a particle moving in the line

$x = 0 \rightarrow x = a$  Take  $h = 6.6 \times 10^{-34} \text{ Js}$  and  $e = 1.6 \times 10^{-19} \text{ C}$

The speed of the particle, that can take discrete values, is proportional to

A.  $n^{-3/2}$

B.  $n^{-1}$

C.  $n^{1/2}$

D.  $n^0$

**Answer: D**



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**165.** The key feature of Bohr's spectrum of hydrogen atom is the quantization of angular momentum when an electron is revolving around a proton we will extend this to a general rotational motion to find quantized rotational energy of a diatomic molecule assuming it to be rigid. The rule applied is Bohr's quantization condition. A diatomic molecule has moment of inertia  $I$  by Bohr's quantization condition its rotational energy in the  $n^{\text{th}}$  level ( $n = 0$  is not allowed) is

A.  $\frac{1}{n^2} \left( \frac{h^2}{8\pi^2 I} \right)$

B.  $\frac{1}{n} \left( \frac{h^2}{8\pi^2 I} \right)$

C.  $n \left( \frac{h^2}{8\pi^2 I} \right)$

D.  $n^2 \left( \frac{h^2}{8\pi^2 I} \right)$

**Answer: D**

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**166.** The key feature of Bohr's spectrum of hydrogen atom is the quantization of angular momentum when an electron is revolving around a proton we will extend this to a general rotational motion to find quantized rotational energy of a diatomic molecule assuming it to be rigid. The rate to energy applied is Bohr's quantization condition it is found that the excitation from ground to the first excited state of rotation for the  $CO$  molecule is close to  $\frac{4}{\pi} \times 10^{11} Hz$  then the moment of inertia of  $CO$  molecule about its center of mass is close to

$$(Take h = 2\pi \times 10^{-34} Js)$$

A.  $2.76 \times 10^{-46} kgm^2$

B.  $1.87 \times 10^{-46} kgm^2$

C.  $4.67 \times 10^{-47} kgm^2$

D.  $1.17 \times 10^{-47} kgm^2$

**Answer: B**



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167. The key feature of Bohr's spectrum of hydrogen atom is the quantization of angular momentum when an electron is revolving around a proton we will extend this to a general rotational motion to find quantized rotational energy of a diatomic molecule assuming it to be rigid. The rule applied is Bohr's quantization condition

In a  $CO$  molecule, the distance between  $C$  ( $mass = 12 \text{ a.m.u}$ ) and  $O$  ( $mass = 16 \text{ a.m.u}$ ) where  $1 \text{ a.m.u} = (5/3) \times 10^{-27} \text{ kg}$ , is close to

A.  $2.4 \times 10^{-10} \text{ m}$

B.  $1.9 \times 10^{-10} \text{ m}$

C.  $1.3 \times 10^{-10} \text{ m}$

D.  $4.4 \times 10^{-11} \text{ m}$

**Answer: C**



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**168.** The  $\beta$  - *decay* process , discoverwd around 1900 , is basically the decay of a neutron ( $n$ ) in the laboratory , a proton ( $p$ ) and an electron ( $e^-$ ) are observed as the decay that the kinetic energy of the electron should be a constant . But experimentally , it was observed that the electron kinetic energy has continuous spectrum Considering a three-body decay process , i.e.

$n \rightarrow p + e^- + \bar{\nu}_e$ , around 1930 , pauli explained the observed ( $\bar{\nu}_e$ ) to be massless and possessing negligible energy , and the neutrino to be at rest , momentum and energy conservation principle are applied from this calculate , the maximum kinetic energy of the electron is  $0.8 \times 10^6 eV$

The kinetic energy carried by the proton is only the recoil energy

If the neutron had a mass of  $3eV/c^2$  (where  $c$  is the speed of light ) instead of zero mass , what should be the range of the kinetic energy  $K$  of the electron ?

A.  $0 \leq k \leq 0.8 \times 10^6 eV$

B.  $3.0eV \leq k \leq 0.8 \times 10^6 eV$

C.  $3.0eV \leq k \leq 0.8 \times 10^6 eV$

D.  $0 \leq k \leq 0.8 \times 10^6 \text{ eV}$

**Answer: D**



**Watch Video Solution**

**169.** The  $\beta - \text{decay}$  process , discoverwd around 1900 , is basically the decay of a neutron ( $n$ ) in the laboratory , a proton ( $p$ ) and an electron ( $e^-$ ) are observed as the decay that the kinetic energy of the electron should be a constant . But experimentally , it was observed that the electron kinetic energy has continuous spectrum Considering a three-body decay process , i.e.

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The kinetic energy carried by the proton is only the recoil energy

What is the maximum energy of the anti-neutrino ?

A. zero

B. Much less than  $0.8 \times 10^6 \text{ eV}$

C. Nearly  $0.8 \times 10^6 \text{ eV}$

D. Much larger than  $0.8 \times 10^6 \text{ eV}$

**Answer: C**



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### 170. STATEMENT - 1

If the accelerating potential in an X - rays tube is increased, the wavelength of the characteristic X- rays do not change .

### STATEMENT -2

When an electron beam strikes the target in an X- rays tube, part of the kinetic energy is converted into X - rays energy .

A. Statement - 1 is true , Statement -2 is true Statement -2 is a correct explanation for Statement - 1

- B. Statement - 1 is true , Statement -2 is true Statement -2 is a NOT a correct explanation for Statement - 1
- C. Statement - 1 is true , Statement -2 is false
- D. Statement - 1 is false , Statement -2 is true

**Answer: B**

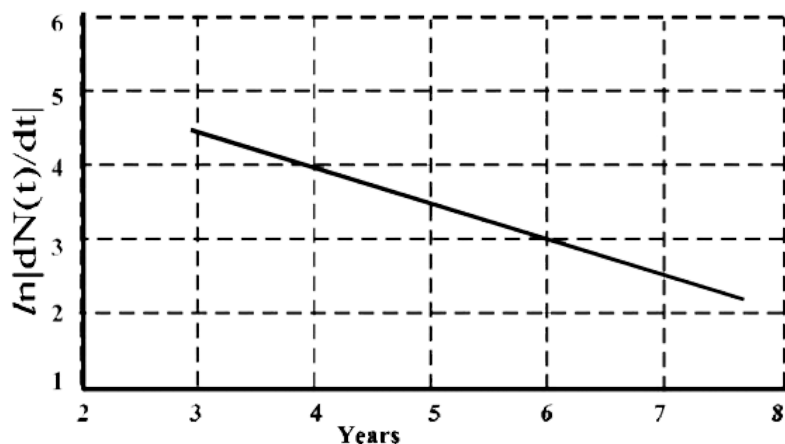
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**171.** An  $\alpha$ - particle and a proton are accelerated from rest by a potential difference of 100V. After this, their de-Broglie wavelengths are  $\lambda_a$  and  $\lambda_p$  respectively. The ratio  $\frac{\lambda_p}{\lambda_a}$ , to the nearest integer, is.

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**172.** To determine the half life of a radioactive element , a student plot a graph of in  $\left| \frac{dN(t)}{dt} \right|$  versus  $t$  , Here  $\left| \frac{dN(t)}{dt} \right|$  is the rate of radiation decay at time  $t$  , if the number of radioactive nuclei of this element decreases by

a factor of  $p$  after 4.16 year the value of  $p$  is



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**173.** The activity of a freshly prepared radioactive sample is  $10^{10}$  disintegrations per second, whose mean life is  $10^9$  s. The mass of an atom of this radioisotope is  $10^{-25}$  kg. The mass (in mg) of the radioactive is



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**174.** A silver of radius  $1\text{ cm}$  and work function  $4.7\text{ eV}$  is suspended from an insulating thread in free space. It is under continuous illumination of



200nm wavelength light AS photoelectron are emitted the sphere gas charged and acquired a potential . The maximum number of photoelectron emitted from the sphere is  $A \times 10^e$  (where  $1 < A < 10$ )

The value of  $z$  is



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**175.** A proton is first from very loward a nucleus with charge  $Q = 120e$ , where  $e$  is the nucles The de Brogle wavelength (in unit of fm) of the proton at its start is (tke the proton mass ,

$$m_p = (5/3) \times 10^{-27} \text{ kg} \quad h/s = 4.2 \times 10^{-15} \text{ Js/C},$$

$$\frac{1}{4\pi s_0} = 9 \times 10^9 \text{ m/F}, \quad 1 \text{ fm} = 10^{-15} \text{ m}$$



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**176.** The work function of Silver and sodium are 4.6 and 2.3eV, respectively . The ratio of the slope of the stopping potential versus frequency plot for silver to that of sodium is



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**177.** A freshly prepared of a radioisotope of half - life  $1386\text{s}$  has activity  $10^3$  disintegrations per second Given that  $\ln 2 = 0.693$  the fraction of the initial number of nuclei (expressed in nearest integer percentage ) that will decay in the first  $80\text{s}$  after preparation of the sample is

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**178.** A nuclear power supplying electrical power to a village uses a radioactive material of half life  $T$  year as the fuel . The amount of fuel at the beginning is such that the total power requirement of the village is  $12.5\%$  of the electrical power available from the plate at that time if the plate is able to meet the total power needs of the village for a maximum period of  $nT$  year , then the value of  $n$  is

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**179.** Consider a hydrogen atom with its electron in the  $n^{th}$  orbital. An electromagnetic radiation of wavelength  $90nm$  is used to ionize the atom. If the kinetic energy of the ejected electron is  $10.4eV$ , then the value of  $n$  is ( $hc = 1242eVnm$ )



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**180.** For a radioactive material, its activity  $A$  and rate of change of its activity  $R$  are defined as  $A = -\frac{dN}{dt}$  and  $R = \frac{dA}{dt}$  where  $N(t)$  is the number of nuclei at time  $t$ . Two radioactive sources  $P$  (mean life  $\tau$ ) and  $Q$  (mean life  $2\tau$ ) have the same activity at  $t = 2\tau$ .  $R_P$  and  $R_Q$  respectively, if  $\frac{R_P}{R_Q} = \frac{n}{e}$



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**181.** An electron is in an excited state of  $Li^{2+}$  ion with angular momentum  $3\hbar/2\pi$ . The de Broglie wavelength of the electron in this state is  $p\pi a_0$  (where  $a_0$  is the Bohr radius). The value of  $p$  is



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**182.** The isotope  ${}_{5}^{12}\text{B}$  having a mass  $12.014\text{u}$  undergoes  $\beta - \text{decay}$   $\rightarrow {}_{6}^{12}\text{C}$  has an excited state of the nucleus  $({}_{6}^{12}\text{C}^*)$  at  $4.041\text{MeV}$  above its ground state if  ${}_{5}^{12}\text{B}$  decays  $\rightarrow {}_{6}^{12}\text{C}^*$ , the maximum kinetic energy of the  $\beta - \text{particle}$  is in unit of  $\text{MeV}$  is  $(1\text{u} = 931.5\text{MeV}/c^2)$  where  $c$  is the speed of light in vacuum).



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**183.** A hydrogen atom in its ground state is hit by light of wavelength  $970\text{\AA}$ . The kinetic energy of the electron is  $1.237 \times 10^{-6}\text{eV}$  and the ground state energy of hydrogen atom is  $-13.6\text{eV}$ . The number of lines present in the emission spectrum is



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**184.** If  $13.6\text{eV}$  energy is required to ionized the hydrogen atom, then the energy required to remove an electron from  $n = 2$  is

A.  $10.2\text{eV}$

B.  $0\text{eV}$

C.  $3.4\text{eV}$

D.  $6.8\text{eV}$

**Answer: C**



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**185.** At absolute zero , Si acts as

A. non - metal

B. metal

C. insulator

D. none of these

**Answer: C**



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**186.** At a specific instant emission of radioactive compound is deflected in a magnetic field. The compound cannot emit

- A. I,ii,iii
- B. I,ii,iii,iv
- C. iv
- D. ii,iii

**Answer: A**



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**187.** Sodium and copper have work functions  $2.3eV$  and  $4.5eV$  respectively. Then the ratio of the wavelength is nearest

A. 1:2

B. 4:1

C. 2:1

D. 1:4

**Answer: C**



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**188.** Formation of covalent bonds in compound exhibits

A. wave nature of electron

B. particle nature of electron

C. bohr wave and particle nature of electron

D. none of these

**Answer: A**



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**189.** If  $N_0$  is the original mass of the substance of half - life period

$t_{1/2} = 5\text{year}$  then the amount of substance left after 15 year is

A.  $N_0/8$

B.  $N_0/16$

C.  $N_0/2$

D.  $N_0/4$

**Answer: A**



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**190.** By increasing the temperature , the specific of a conductor and a semiconductor

A. increases for both

B. decreases for both



C. increases, decreases

D. decreases , increases

**Answer: C**



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**191.** The energy band gap is maximum in

A. metals

B. superconductors

C. insulators

D. semiconductors

**Answer: C**



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**192.** The part of a transistor which is most heavily doped to produce large number of majority carriers is

- A. emitter
- B. base
- C. collector
- D. can be any of the above three

**Answer: A**



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**193.** which of the following are not electromagnetic waves ?

- A. cosmic rays
- B.  $\Gamma$  rays
- C.  $\beta$  - rays
- D. X- rays

**Answer: C**



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**194.** A strip of copper and another of germanium are cooled from room temperature to  $80K$  The resistance of

- A. each of these decreases
- B. copper strip increase and that of germanium decreases
- C. copper strip increase and that of germanium increase
- D. each of these increase

**Answer: C**



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**195.** Which of the following radiation has the least wavelength ?

A.  $\lambda$  – rays

B.  $\beta$  – rays

C.  $\alpha$  – rays

D.  $X$  – rays

**Answer: A**



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**196.** which a  $U^{238}$  nucleus original at rest , decay by emitting an alpha particle having a speed  $u$  , the recoil speed of the residual nucleus is

A.  $\frac{4u}{238}$

B.  $-\frac{4u}{234}$

C.  $\frac{4u}{234}$

D.  $-\frac{4u}{238}$

**Answer: C**



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**197.** The difference in the variation of resistance with temperature in a metal and a semiconductor arises essentially due to the difference in the

- A. crystal structure
- B. variation of the number of charge carriers with temperature
- C. type of bonding
- D. variation of scattering mechanism with temperature

**Answer: B**



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**198.** A radioactive sample at any instant has its disintegration rate 5000 disintegrations per minute. After 5 minutes, the rate is 1250 disintegrations per minute. Then, the decay constant (per minute)

A.  $0.4 \in 2$

B.  $0.2 \in 2$

C.  $0.1 \in 2$

D.  $0.8 \in 2$

**Answer: A**



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**199.** A nucleus with  $Z = 92$  emits the following in a sequence

$$\alpha, \beta^-, \beta^- \alpha, \alpha, \alpha, \alpha, \alpha, \beta^-, \beta^-, \alpha, \beta^+, \beta^+, \alpha$$

Then  $Z$  of the resulting nucleus is

A. 76

B. 78

C. 82

D. 74

**Answer: B**



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**200.** Two identical photocathodes receive light of frequency  $f_1$  and  $f_2$  if the velocities of the photo electrons (of mass  $m$ ) coming out are respectively  $v_1$  and  $v_2$  then

A.  $v_1^2 - v_2^2 = \frac{2h}{m} (f_1 - f_2)$

B.  $v_1 + v_2 = \left[ \frac{2h}{m} (f_1 + f_2) \right]^{1/2}$

C.  $v_1^2 + v_2^2 = \frac{2h}{m} (f_1 + f_2)$

D.  $v_1 + v_2 = \left[ \frac{2h}{m} (f_1 - f_2) \right]^{1/2}$

**Answer: A**



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**201.** Which of the following cannot be emitted by radioactive substances during their decay ?

- A. protons
- B. Neutrons
- C. Helium nuclei
- D. Electrons

**Answer: A**



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**202.** In the nuclear fusion reaction



given that the repulsive potential energy between the two nuclei is  $-7.7 \times 10^{-14} \text{ J}$ , the temperature at which the gases must be heated for the reaction is nearly

[Boltzmann's constant  $k = 1.38 \times 10^{-23} \text{ J/K}$ ]



A.  $10^7 K$

B.  $10^5 K$

C.  $10^3 K$

D.  $10^9 K$

**Answer: B::D**



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**203.** Which of the following atoms has the lowest ionization potential ?

A.  $_{(7)}^{14}N$

B.  $_{(55)}^{133}Cs$

C.  $_{(18)}^{40}Ar$

D.  $_{(8)}^{16}O$

**Answer: B**



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**204.** The wavelength involved in the spectrum of deuterium ( ${}_1^2D$ )

are slightly different from that of hydrogen spectrum because

- A. the size of the two nuclei are different
- B. the nuclear forces are different in the two cases
- C. the masses of the two nuclei are different
- D. the attraction between the electron and the nucleus is different in the two cases

**Answer: C**



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**205.** In the middle of the depletion layer of a reverse - biased  $p - n$  junction , the

- A. electron field is zero

- B. potential is maximum
- C. electron field is maximum
- D. potential is zero

**Answer: A**



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**206.** If the biniding energy of the electron of the electron in a hydrogen atom is  $13.6eV$  the energy required to remove the electron from the first excited state of  $Li^{++}$  is

- A.  $30.6eV$
- B.  $13.6eV$
- C.  $3.4eV$
- D.  $122.4eV$

**Answer: A**

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**207.** A radiation of energy  $E$  falls normally on a perfectly reflecting surface. The momentum transferred to the surface is

A.  $Ec$

B.  $2E/c$

C.  $E/c$

D.  $E/c^2$

**Answer: B**

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**208.** According to Einstein's photoelectric equation, the plot of the kinetic energy of the incident radiation given as straight line whose slope

- A. depends both on the intensity of the radiation and the metal used
- B. depend on the intensity of the radiation
- C. depends on the nature of the metal used
- D. is the for the all metal s and independent of the intensity of the radiation

**Answer: D**



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**209.** The work function of a substance is  $4.0\text{eV}$  The longest wavelength of light that can cause photoelectron emission from this substance is approximately

- A.  $310\text{nm}$
- B.  $400\text{nm}$
- C.  $540\text{nm}$
- D.  $220\text{nm}$

**Answer: A**



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**210.** A nucleus desintegrated into two nucleus which have their velocities in the ratio of 2 : 1 . The ratio of their nuiclear sizes will be

A.  $3^{\frac{1}{2}} : 1$

B.  $1 : 2^{1/3}$

C.  $2^{1/3} : 1$

D.  $1 : 3^{\frac{1}{2}}$

**Answer: B**



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**211.** The binding energies per nucleon for deuteron ( ${}_1H^2$ ) and helium ( ${}_2He^4$ ) are  $1.1MeV$  and  $7.0MeV$  respectively. The energy released when

two deuterons fuse to form a helium nucleus ( ${}_2\text{He}^4$ ) is.....

A.  $23.6\text{MeV}$

B.  $26.9\text{MeV}$

C.  $13.9\text{MeV}$

D.  $19.2\text{MeV}$

**Answer: A**



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**212.** An alpha particle of energy  $5\text{MeV}$  is scattered through  $180^\circ$  by a found uramiam nucleus . The distance of closest approach is of the order of

A.  $10^{-12}\text{cm}$

B.  $10^{-10}\text{cm}$

C.  $1\text{\AA}$

D.  $10^{-15} \text{ cm}$

**Answer: A**



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**213.** When npn transistor is used as an amplifier

- A. electron move from collector to base
- B. boles move from emitted to base
- C. electron move from base to collector
- D. holes move from base to emitter

**Answer: C**



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**214.** For a transistor amplifier in common emitter configuration for load impedance of  $1k\Omega$ . ( $h_{fe} = 50$  and  $h_{oe} = 25 \times 10^{-6}$ ) the current gain is

- A.  $-24.8$
- B.  $-15.7$
- C.  $-5.2$
- D.  $-48.78$

**Answer: D**



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**215.** A piece of copper and another of germanium are cooled from room temperature to  $77K$ , the resistance of

- A. copper increases and germanium decreases
- B. each of them decreases
- C. each of them increases

D. copper decrease and germanium increase

**Answer: D**



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**216.** The manifestation of band structure in solids is due to

- A. Bohr's correspondence principle
- B. pauli's exclusion principle
- C. Heisenberg's encertainly principle
- D. Boltzmann's law

**Answer: B**



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**217.** When  $p - n$  junction diode is forward biased them

- A. both the depletion regain and harrier height are reduced
- B. the depletion regain is widened and harrier height is reduced
- C. the depletion regain , is reduced and harrier heighis increases
- D. Both the depletion regain and barrier are increases

**Answer: A**



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**218.** If radius of the  $_{13}^{27}\text{Al}$  necleus is estimated to be 3.6 fermi then the radius of  $_{52}^{125}\text{Te}$  nucleus be nearly

- A. 8 *fermi*
- B. 6 *fermi*
- C. 5 *fermi*
- D. 6 *fermi*

**Answer: B**

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219. Starting with a sample of pure  $^{66}\text{Cu}$ ,  $\frac{3}{4}$  of it decays into  $\text{Zn}$  in 15 minutes. The corresponding half-life is

A. 15 min *utes*

B. 10 min *utes*

C.  $7\frac{1}{2}$  min *utes*

D. 5 min *utes*

**Answer: D**

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220. A photocell is illuminated by a small bright source placed 1 m away when the same source of light is placed  $\frac{1}{2}$  m away. The number of electrons emitted by the photocathode would

A. increase by a factor of 4

B. decrease by a factor of 4

C. increase by a factor of 2

D. decrease by a factor of 4

**Answer: A**



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**221.** The electron conductivity of a semiconductor increases  $2480\text{nm}$  is incident on it . The band gap in (eV) for the semiconductor is

A.  $25\text{eV}$

B.  $1.1\text{eV}$

C.  $0.7\text{eV}$

D.  $0.5\text{eV}$

**Answer: D**

**222.** The intensity of gamma radiation from a given source is 1

On passing through  $36mm$  of lead , it is reduced to  $\frac{1}{8}$  . The thickness of lead which will reduce the intensity to  $\frac{1}{2}$  will be

A.  $9mm$

B.  $6mm$

C.  $12mm$

D.  $18mm$

**Answer: C**

**223.** In a common base amplifier , the phase difference between the input signal and output voltage is

A.  $\pi$

B.  $\frac{\pi}{4}$

C.  $\frac{\pi}{2}$

D. 0`

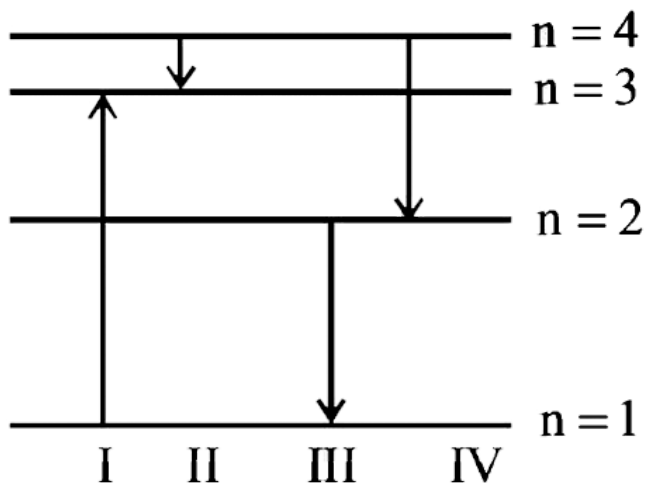
**Answer: D**



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**224.** The diagram shown the energy levels for an electron in a certain atom. Which transition shown the emission of a photon with the most

energy ?



A. iv

B. iii

C. ii

D. i

**Answer: B**



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225. If the kinetic energy of a electron , it's debroglie wavelength changes by the factor

A. 2

B.  $\frac{1}{2}$

C.  $\sqrt{2}$

D.  $\frac{1}{\sqrt{2}}$

**Answer: D**



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226. A necear transformation is denoted by  $X(n, a)_3^7Li$  Which of the following is the neclues of electron X?

A.  $_{-}(5)^{10}Be$

B.  $_{+}(12)C_6$

C.  $_{-}(4)^{11}Be$

D.  $-(5)^9 B$

**Answer: A**



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**227.** In a full wave rectifier circuit operating from  $50Hz$  mains frequency , the fundamental frequency in the ripple would be

A.  $25Hz$

B.  $50Hz$

C.  $70.7Hz$

D.  $100Hz$

**Answer: D**



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**228.** In a common base mode of a transition , the collector current is  $5.488mA$  for an emitter current of  $5.60mA$  . The value of the base current amplification factor ( $\beta$ ) will be

A. 49

B. 50

C. 51

D. 48

**Answer: A**



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**229.** The threshold frequency for a metallic surface corresponds to an energy of  $6.2eV$  and the stopping potential for a radiation incident on this surface is  $5V$  . The incident radiation lies in

A. ultra - violet region

B. infra- red regaion

C. visible region

D. X- ray rasion

**Answer: A**



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**230.** An alpha nucleus of energy  $\frac{1}{2}m\nu^2$  bombards a heavy nucleus of charge  $Ze$  . Then the distance of closed approach for the alpha nucleus will be prppportional to

A.  $\nu^2$

B.  $\frac{1}{m}$

C.  $\frac{1}{\nu^2}$

D.  $\frac{1}{Ze}$

**Answer: C**

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**231.** The time taken by a photoelectron to come out after the photon strikes is approximately

A.  $10^{-4} s$

B.  $10^{-10} s$

C.  $10^{-16} s$

D.  $10^{-1} s$

**Answer: B**

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**232.** When  ${}_3\text{Li}^7$  nuclei are bombarded by protons, and the resultant nuclei are  ${}_4\text{Be}^8$ , the emitted particle will be

A.  $\alpha$  particle

B.  $\beta$  particle

C.  $\gamma$  particle

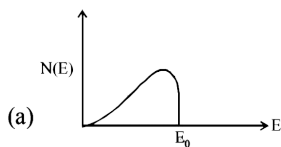
D. neutrons

**Answer: C**

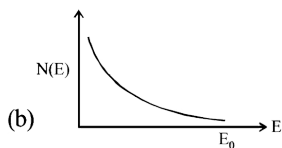


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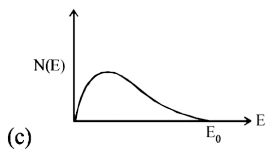
**233.** The energy spectrum of  $\beta$  - *particle* [number  $N(E)$  as a function of  $\beta$  - *e*  $\neq$  *rgyE*] emitter from a radioactive source is



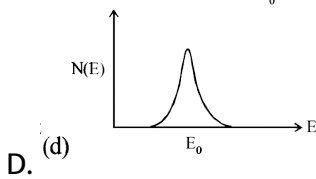
**A.**



**B.**



**C.**



**Answer: C**



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**234.** A solid which is not transparent to visible light and whose conductivity increase with temperature is formed by

- A. Ionic bonding
- B. Covalent bonding
- C. vander Waals bonding
- D. Metallic bonding

**Answer: B**



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**235.** If the ratio of the concentration of electron to that of holes in a semiconductor is  $\frac{7}{5}$  and the ratio of current is  $\frac{7}{4}$  then what is the ratio of their drift velocities ?

A.  $\frac{5}{8}$

B.  $\frac{4}{5}$

C.  $\frac{5}{4}$

D.  $\frac{4}{7}$

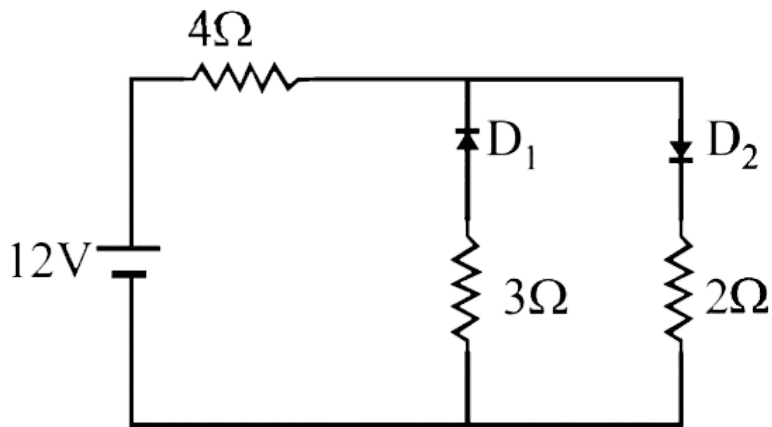
**Answer: C**



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**236.** The circuit has two opposotively connected ideal diodes in parallel what is the current flowing in the circuit ?





A.  $1.71A$

B.  $2.00A$

C.  $2.31A$

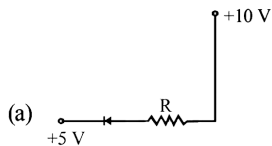
D.  $1.33A$

**Answer: B**

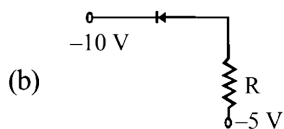


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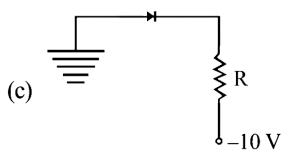
**237.** If the following which one of the diodes reverse biased?



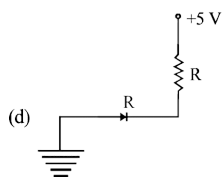
A.



B.



C.



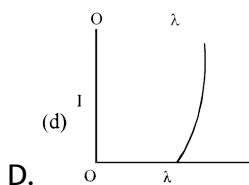
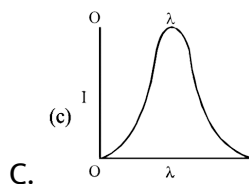
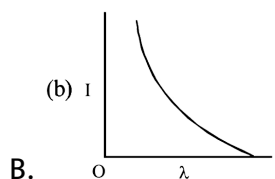
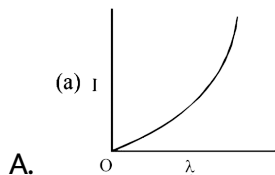
D.

**Answer: D**



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**238.** The anode voltage of a photocell is kept fixed . The wavelength  $\lambda$  of the light falling on the cathode varies as follows

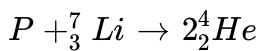


**Answer: B**



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**239.** If the binding energy per nucleon in  ${}^7_3\text{Li}$  and  ${}^4_2\text{He}$  are  $5.60\text{ MeV}$  and  $7.06\text{ MeV}$  respectively then in the reaction



energy of proton must be

A.  $28.24\text{MeV}$

B.  $17.28\text{MeV}$

C.  $1.46\text{MeV}$

D.  $39.2\text{MeV}$

**Answer: B**



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**240.** The half is the correct unit used to report the measurement of

A. the ability of a beam of gamma ray photons to produce ions in a target

B. the energy effect by radiation to a target

C. the biological effect of radiation

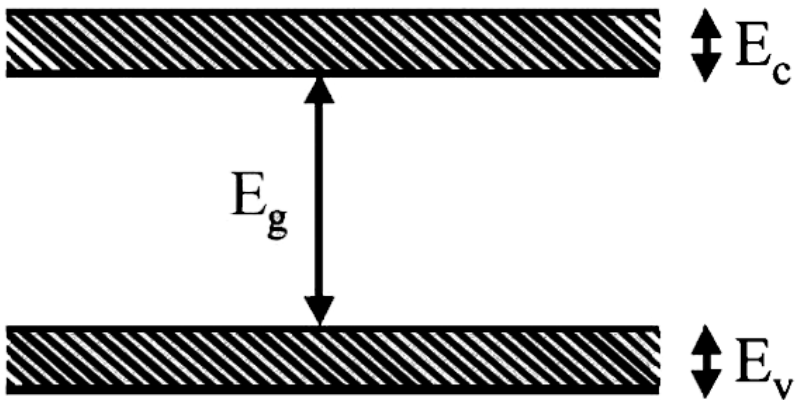
D. the rate of decay of a radioactive source

Answer: C



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241. If the lattice constant of this semiconductor is decreases, then which of the following is correct ?



A. All  $E_c$ ,  $E_g$ ,  $E_v$ ,  $\in$  crease

B.  $E_c$  and  $E_v \in$  crease but  $E_g$ , decrease

C.  $E_c$  and  $E_v$  decrease but  $E_g$ ,  $\in$  crease

D. All  $E_c$ ,  $E_g$ ,  $E_v$ , decrease

**Answer: C**



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**242.** The rms value of the electric field of the light from the sun is  $720\text{ N/C}$ . The energy density of the electromagnetic wave is

A.  $4.58 \times 10^{-6} \text{ J/m}^3$

B.  $6.37 \times 10^{-9} \text{ J/m}^3$

C.  $81.35 \times 10^{-12} \text{ J/m}^3$

D.  $3.3 \times 10^{-3} \text{ J/m}^3$

**Answer: A**



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**243.** If  $M_O$  is the mass of an oxygen isotope  ${}^{17}_8\text{O}$ ,  $M_p$  and  $M_N$  are the masses of a proton and a neutron respectively, the nuclear binding energy of the isotope is

- A.  $(M_O - 17M_N)c^2$
- B.  $(M_O - 8M_p)c^2$
- C.  $(M_O - 8M_p - 9M_N)c^2$
- D.  $M_Oc^2$

**Answer: C**



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**244.** In gamma ray emission from a nucleus

- A. only the proton number changes
- B. both the neutron and the proton number change
- C. there is no change in the proton number and the neutron number

D. only the neutron number changes

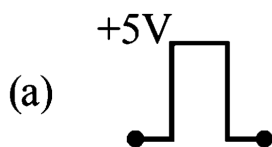
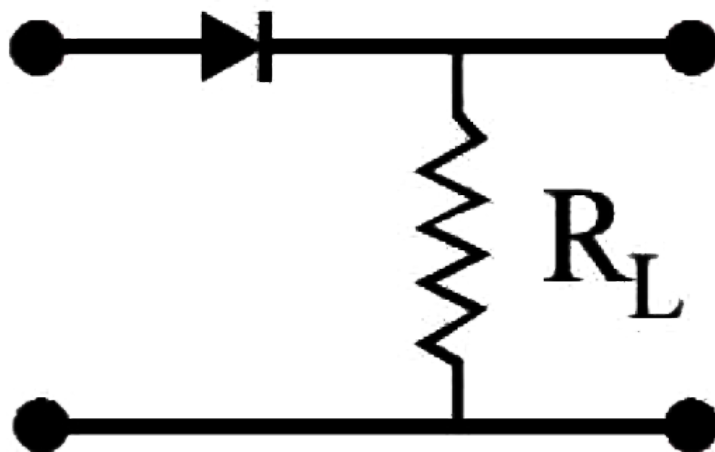
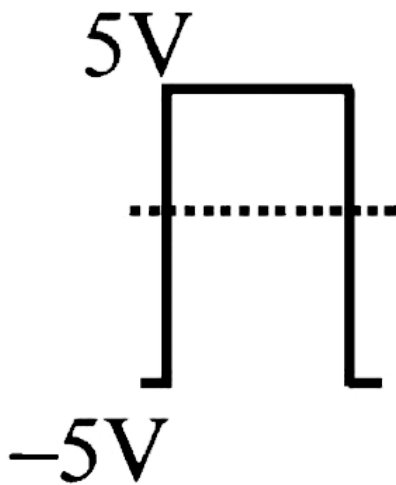
**Answer: C**



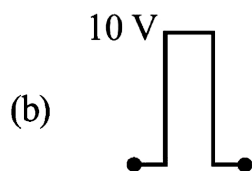
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**245.** If in a  $p - n$  junction diode , a square input signal of  $10V$  is applied as shown



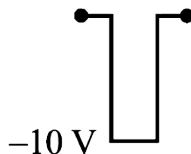


A.



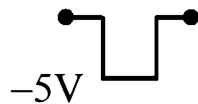
B.

(c)



C.

(d)



D.

**Answer: A**



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**246.** Photon of frequency  $\nu$  has a momentum associated with it . If  $c$  is the velocity of light , the momentum is

A.  $hc/\lambda$

B.  $\nu/c$

C.  $h\nu c$

D.  $h\nu/c^2$

**Answer: A**



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**247.** The half-life period of a radioactive element  $x$  is same as the mean life time of another radioactive element  $y$ . Initially, both of them have the same number of atoms. Then,

- (a)  $x$  and  $y$  have the same decay rate initially
- (b)  $x$  and  $y$  decay at the same rate always
- (c)  $y$  will decay at a faster rate than  $x$
- (d)  $x$  will decay at a faster rate than  $y$

A.  $X$  and  $Y$  decay at same rate always

B.  $X$  will decay faster than  $Y$

C.  $Y$  will decay faster than  $X$

D.  $X$  and  $Y$  have same rate initially

**Answer: C**



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**248.** Carbon , silicon and germanium have four valence electrons each . At room temperature which one of the following statements is most appropriate ?

A. the number of free electron for conduction is significant is

*Si* and *Ge* but small in *C*

B. the number of free conduction electron is significant in *C* but small

in *Si* and *Ge*

C. the number of free conduction electron is negligibly small in all the three.

D. the number of free electron for conduction is significant in all the three

**Answer: A**



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**249.** Which of the following transition in hydrogen atom emit photons of highest frequency ?

A.  $n = 1 \rightarrow n = 2$

B.  $n = 2 \rightarrow n = 6$

C.  $n = 6 \rightarrow n = 2$

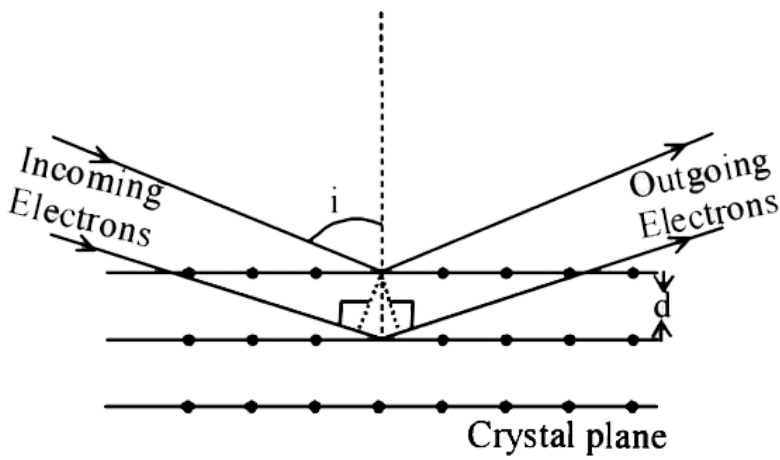
D.  $n = 2 \rightarrow n = 1$

**Answer: D**



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**250.** Wave property of electron implies that they will show diffraction effected . Davisson and Germer demonstrated this by diffracting electron from crystals . The law governing the diffraction from a crystals is obtained by requiring that electron waves reflected from the planes of atoms in a crystal interfere constructively



Electron accelerated by potential  $V$  are diffracted from a crystal if  $d = 1\text{\AA}$  and  $i = 30^\circ$ ,  $V$  should be about  $(h = 6.6 \times 10^{-34} Js, m_e = 9.1 \times 10^{-31} kg, e = 1.6 \times 10^{-19} C)$

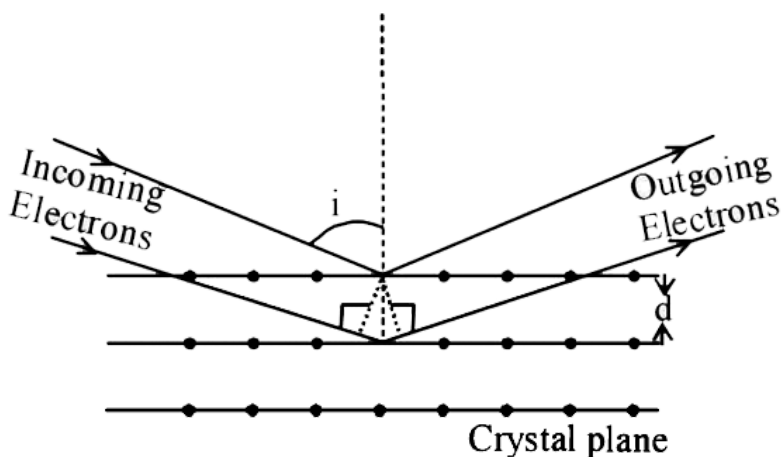
- A.  $2000V$
- B.  $50V$
- C.  $500V$
- D.  $1000V$

**Answer: B**



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**251.** Wave property of electron implies that they will show diffraction effected . Davisson and Germer demonstrated this by diffracting electron from crystals . The law governing the diffraction from a crystals is obtained by requiring that electron waves reflected from the planes of atoms in a crystal inter fere constructiely



If a strong diffraction peak is observed when electrons are incident at an angle  $i$  from the normal to the crystal planes with distance  $d$  between them (see fig) de Brogle wavelength  $\lambda_{dB}$  of electrons can be calculated by the relationship ( $n$  is an intenger)

A.  $d \sin i = n\lambda_{dB}$

B.  $2d \cos i = n\lambda_{dB}$

C.  $2d \sin i = n\lambda_{dB}$

D.  $d \cos i = n\lambda_{dB}$

**Answer: B**



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**252.** This question contains Statement - 1 and Statement -2 Of the four choice given after the Statements , choose the one that best describes the two Statements

Statement- 1:

Energy is released when heavy nuclei undergo fission or light nuclei undergo fusion and

Statement- 2:

for nuclei , binding energy per nucleon increases with increasing  $Z$  while for light nuclei it decreases with increasing  $Z$

A. Statement - 1 is false ,Statement - 2 is true



B. Statement - 1 is true ,Statement - 2 is true, Statement - 2 is a correct explanation for Statement - 1

C. Statement - 1 is true ,Statement - 2 is true, Statement - 2 is not a correct explanation for Statement - 1

D. Statement - 1 is true ,Statement - 2 is false

**Answer: D**



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**253.** A working transistor with its three legs marked  $P$ ,  $Q$  and  $R$  is tested using a multimeter. No conduction is found between  $P$ ,  $Q$  by connecting the common (negative) terminal of the multimeter to  $R$  and the other (positive) terminal to or  $Q$  some resistance is seen on the multimeter. Which of the following is true for the transistor ?

A. it is an npn transistor with  $R$  as base

B. it is a pnp transistor with  $R$  as collector

C. it is a pnp transistor with  $R$  as emitter

D. it is an npn transistor with  $R$  as collector

**Answer: A**



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**254.** Suppose an electron is attracted toward the origin by a force  $\frac{k}{r}$  where  $k$  is a constant and  $r$  is the distance of the electron from the origin. By applying Bohr model to this system the radius of the  $n^{th}$  orbital of the electron is found to be  $r_n$  and the kinetic energy of the electron to be  $T_n$ . Then which of the following is true ?

A.  $T_n \propto \frac{1}{n^2}, r_n \propto n^2$

B.  $T_n$  independent of  $n, r_n \propto n$

C.  $T_n \propto \frac{1}{n}, r_n \propto n$

D.  $T_n \propto \frac{1}{n}, r_n \propto n^2$

**Answer: B**



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**255.** The transition from the state  $n = 4$  to  $n = 3$  in a hydrogen-like atom results in ultraviolet radiation. Infrared radiation will be obtained in the transition

A.  $3 \rightarrow 2$

B.  $4 \rightarrow 2$

C.  $5 \rightarrow 4$

D.  $2 \rightarrow 1$

**Answer: C**



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**256.** The surface of a metal is illuminated with the light of  $400\text{nm}$ . The kinetic energy of the ejected photoelectron was found to be  $1.68\text{eV}$ . The work function of the metal is :

A.  $1.41\text{eV}$

B.  $1.51\text{eV}$

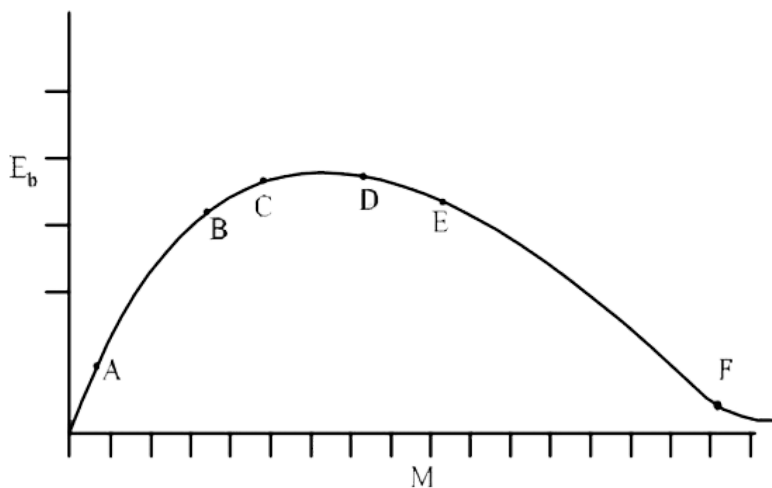
C.  $1.68\text{eV}$

D.  $3.09\text{eV}$

**Answer: A**

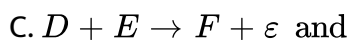
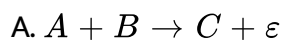


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257.

The above is a plot of binding energy per nucleon  $E_b$  against the nuclear mass  $M$ , A, B, C, D, E, F correspond to different nuclei. Consider four reactions:

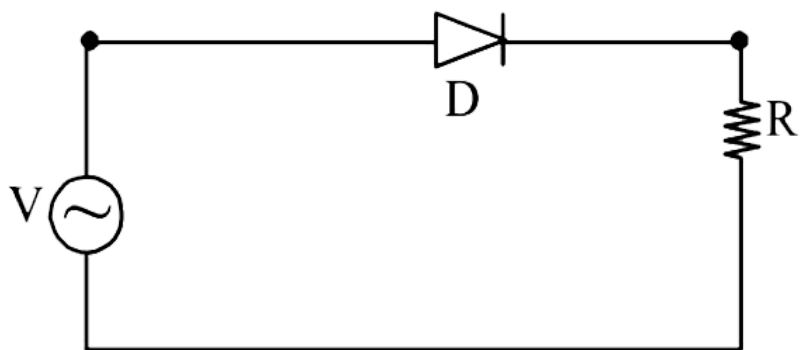


**Answer: D**



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258. a p -n junction (D) shown in the figure can act an a rectifier An alternatting current source (V) is connected in the circuit



The corrent (I ) in the resistor® can be shown by :

- (a)

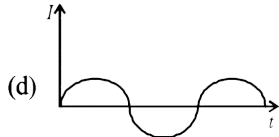
A graph with current  $I$  on the vertical axis and time  $t$  on the horizontal axis. It shows three consecutive positive half-sine waves starting from the origin, indicating full-wave rectification.
- A.

(b)

A graph with current  $I$  on the vertical axis and time  $t$  on the horizontal axis. It shows two positive half-sine waves, each followed by a gap where the current is zero, indicating half-wave rectification.
- B.

(c)

A graph with current  $I$  on the vertical axis and time  $t$  on the horizontal axis. It shows a constant horizontal line at a positive value, indicating a steady-state DC current.
- C.



D.

**Answer: B**



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**259.** Statement - 1 : When ultraviolet light is incident on a photocell , its stopping potential is  $V_0$  and the maximum kinetic energy of the photoelectrons is  $K_{\max}$  when the ultraviolet light is replaced by X- rays both  $V_0$  and  $K_{\max}$  increase

Statement - 2 : photoelectrons are emitted with speeds ranging from zero to a maximum value because of the range of frequencies present in the incident light

A. Statement - 1 is true , statement - 2 is true , statement - 2 is the correct explanation of statement - 1

- B. Statement - 1 is true , statement - 2 is true , statement - 2 is not the correct explanation of statement - 1
- C. Statement - 1 is false , statement - 2 is true
- D. Statement - 1 is true , styatement - 2 is false

**Answer: D**



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**260.** A nuclear of mass  $M + \delta m$  is at rest and decay into two daughter nuclei of equal mass  $\frac{M}{2}$  each speed is  $c$

The binding energy per nucleon for the nucleus is  $E_1$  and that for the daughter nuclei is  $E_2$  Then

A.  $E_2 = 2E_1$

B.  $E_1 > E_2$

C.  $E_2 > E_1$

D.  $E_1 = 2E_2$



**Answer: C**



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**261.** A nuclear of mass  $M + \delta m$  is at rest and decay into two daughter nuclei of equal mass  $\frac{M}{2}$  each speed is  $c$

The speed of daughter nuclei is

A.  $c \frac{\delta m}{M + \delta m}$

B.  $c \sqrt{\frac{2\delta m}{M + \delta m}}$

C.  $c \sqrt{\frac{\delta m}{M}}$

D.  $c \sqrt{\frac{\delta m}{M + \delta m}}$

**Answer: B**



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**262.** A radioactive nucleus (initial mass number  $A$  and atomic number  $Z$ ) emits  $3\alpha$  - particles and  $2\text{positrons}$ . The ratio of number of neutrons to that of proton in the final nucleus will be

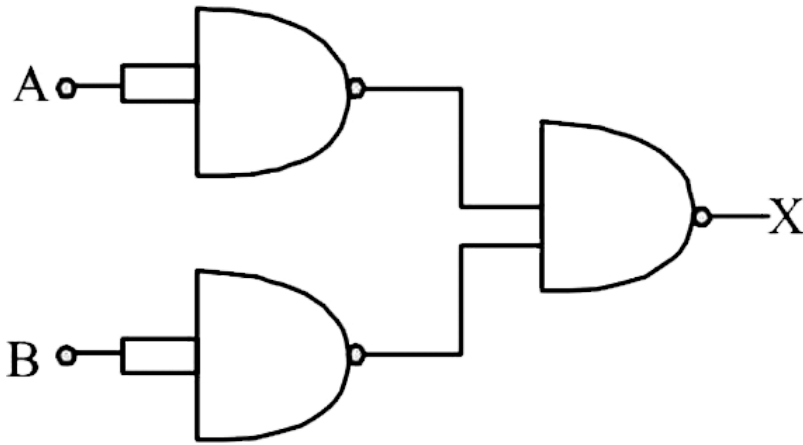
- A.  $\frac{A - Z - 8}{Z - 4}$
- B.  $\frac{A - Z - 4}{Z - 8}$
- C.  $\frac{A - Z - 12}{Z - 4}$
- D.  $\frac{A - Z - 4}{Z - 2}$

**Answer: B**



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263. The combination of shown below yields



A. *ORgate*

B. *NOTgate*

C. *XORgate*

D. *NANDgate*

**Answer: A**



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**264.** If the source of power  $4kW$  product  $10^{20}$  photons //second , the radiation belongs to a part spectrum called

- A. X- rays
- B. ultraviolet rays
- C. microwaves
- D.  $\gamma$  rays

**Answer: A**



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**265.** The quwstion has statement - 1 and statement - 2 Of the four choices given after the statements , choose the one that best describes the two statements

statement - 1 : Sky wave signals are used for long distance radio communication . These signals are in generel , less stable then ground wave signals

statement - 2 : The state of inosphere varies from to hour day and season to season .

- A. Statement -1 is true ,Statement -2 is true , Statement -2 is the correct explanation of Statement -1
- B. Statement -1 is true ,Statement -2 is true , Statement -2 is not the correct explanation of Statement -1
- C. Statement -1 is false ,Statement -2 is true
- D. Statement -1 is true ,Statement -2 is false

**Answer: B**



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**266.** Energy required for the electron excitation in  $Li^{++}$  from the first to the third Bohr orbit is

- A.  $36.3eV$

B.  $108.8\text{eV}$

C.  $122.4\text{eV}$

D.  $12.1\text{eV}$

**Answer: B**



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**267.** The half life of a radioactive substance is 20 minutes . The approximate time interval  $(t_1 - t_2)$  between the time  $t_2$  when  $\frac{2}{3}$  of it had decayed and time  $t_1$  when  $\frac{1}{3}$  of it had decay is

A. 14 min

B. 20 min

C. 28 min

D. 7 min

**Answer: B**



**268.** This question has statement - 1 and statement - 2 of the four choice given after the statements choose the one that best describes the two statements

statement - 1 : A metallic surface is irradiated by a monochromatic light of frequency

$\nu > \nu_0$  (the threshold frequency). The maximum kinetic energy  $K_{\text{max}}$  and the stopping potential  $V_0$  are

$K_{\text{max}}$  and  $V_0$  respectively if the frequency  $\nu \in \text{certain range}$  and  $\nu \leq \nu_0$ ,  $K_{\text{max}}$  and  $V_0$  are zero

respectively if the frequency  $\nu \in \text{certain range}$  and  $\nu \leq \nu_0$ ,  $K_{\text{max}}$  and  $V_0$  are zero

$K_{\text{max}}$  and  $V_0$  are also doubled

statement - 2 : The maximum kinetic energy and the stopping potential of photoelectron emitted from a surface are linearly dependent on the frequency of incident light

- A. Statement -1 is true ,Statement -2 is true , Statement -2 is the correct explanation of Statement -1

- B. Statement -1 is true ,Statement -2 is true , Statement -2 is not the correct explanation of Statement -1
- C. Statement -1 is false ,Statement -2 is true
- D. Statement -1 is true ,Statement -2 is false

**Answer: C**



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**269.** Hydrogen atom is excited from ground state to another state with principal quantum number equal to 4 Then the number of spectral lines in the emission spectra will be

- A. 2
- B. 3
- C. 5
- D. 6



**Answer: D**



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**270.** A radio has a power of  $1kW$  and is operating at a frequency of  $10GHz$  it is located on a mountain top of height  $500m$ . The maximum distance upto which it can detect object located on the surface of the earth (Radius of earth  $= 6.4 \times 10^6m$ ) is

A.  $80km$

B.  $16km$

C.  $40km$

D.  $64km$

**Answer: A**



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**271.** Assume that a neutron breaks into a proton and an electron . The energy released during this process is (mass of neutron  $= 1.6725 \times 10^{-27} kg$  mass of proton  $= 1.6725 \times 10^{-27} kg$  mass of electron  $= 9 \times 10^{-31} kg$ )

A.  $0.73 MeV$

B.  $7.10 MeV$

C.  $6.30 MeV$

D.  $5.4 MeV$

**Answer: A**



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**272.** A diatomic molecule is made of two masses  $m_1$  and  $m_2$  which are separated by a distance  $r$  . If we calculate its rotational energy by applying Bohr's rule of angular momentum quantization its energy will be (  $n$  is an integer )

- A.  $\frac{(m_1 + m_2)^2 n^2 h^2}{2m_1^2 m_2^2 r^2}$
- B.  $\frac{n^2 h^2}{2(m_1 + m_2) r^2}$
- C.  $\frac{2n^2 h^2}{m_1 + m_2} r^2$
- D.  $\frac{(m_1 + m_2) n^2 h^2}{2m_1 m_2 r^2}$

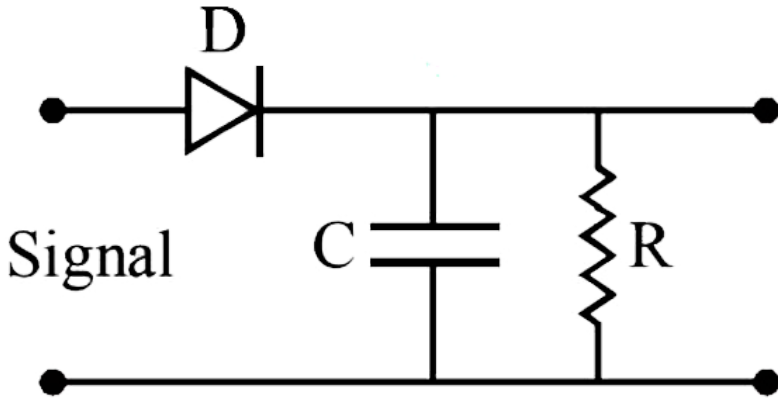
**Answer: D**



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**273.** A diode detector is used to detect an amplitude modulated wave of 60 % modulation by using a condenser of capacity 250 picofarad in parallel with a load resistance 100 kilo ohm find the maximum modulated frequency which could be

detected by it



A.  $10.62MHz$

B.  $10.62kHz$

C.  $5.31MHz$

D.  $5.31kHz$

**Answer: B**



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**274.** The magnetic field in a travelling electromagnetic wave has a peak value of  $20\text{ nT}$ . The peak value of electric field strength is :

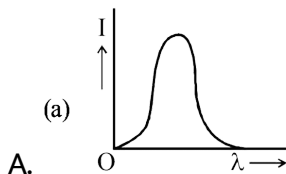
- A.  $3\text{ Vm}$
- B.  $6\text{ Vm}$
- C.  $9\text{ Vm}$
- D.  $12\text{ Vm}$

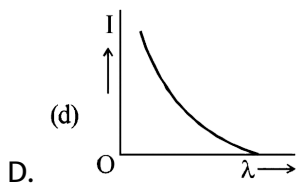
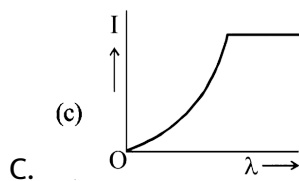
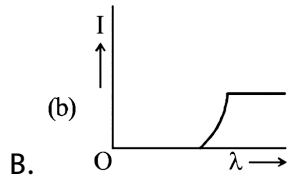
**Answer: B**



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**275.** The anode voltage of a photocell is kept fixed . The wavelength  $\lambda$  of the light falling on the cathode varies as follows



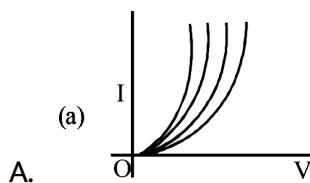


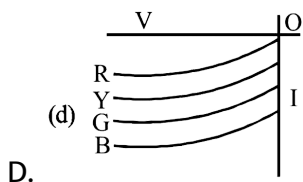
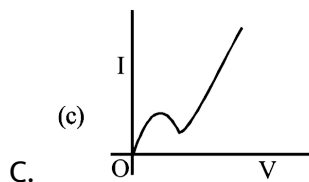
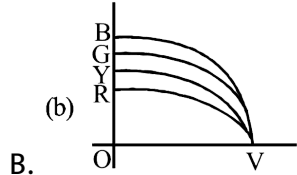
**Answer: D**



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**276.** The  $I - V$  characteristic of an LED is





**Answer: A**



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**277.** In a hydrogen like atom electron make transition from an energy level with quantum number  $n$  to another with quantum number  $(n - 1)$  if  $n > 1$ , the frequency of radiation emitted is proportional to :

A.  $\frac{1}{n}$

B.  $\frac{1}{n^2}$

C.  $\frac{1}{n^3/2}$

D.  $\frac{1}{n^2}$

**Answer: D**



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**278.** The current voltage relation of a diode is given by  $i = \left( e^{v_{anv}/T} - 1 \right) mA$  where the applied volied  $V$  is in volts and the tempetature  $T$  is in degree kelvin if a student make an error meassurting  $\pm 01V$  while measuring the current of  $5mA$  at  $300K$  what be the error in the value of current in  $mA$

A.  $0.2mA$

B.  $0.02mA$

C.  $0.5mA$

D.  $0.05mA$

**Answer: A**





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**279.** During the propagation wave in a medium

- A. Electric energy density is double of the magnetic energy density
- B. Electric energy density is half of the magnetic energy density
- C. Electric energy equal is double of the magnetic energy density
- D. Both eneric and magnetic energy densities are zero

**Answer:** C



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**280.** The radiation corresponding to  $3 \rightarrow 2$  transition of hydrogen atom falls on a metal surface to produce photoelectrons . These electrons are made to enter circuit a magnitic field  $3 \times 10^{-4}T$  if the ratio of thelargest circular path follow by these electron is  $\sim 10.0$  mm , the work function of the metal is close to

A.  $1.8eV$

B.  $1.1eV$

C.  $0.8eV$

D.  $1.6eV$

**Answer: B**



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**281.** Hydrogen ( ${}_1H^1$ ) Deuterium ( ${}_1H^2$ ) singly ionised helium ( ${}_2He^1$ ) and doubly ionised lithium ( ${}_3Li^{6++}$ ) all have one electron around the nucleus. Consider an electron transition from  $n = 2 \rightarrow n = 1$  if the wavelength of emitted radiation are  $\lambda_1, \lambda_2, \lambda_3$ , and  $\lambda_4$ , respectively then approximately which one of the following is correct ?

A.  $4\lambda_1 = 2\lambda_2 = 2\lambda_3 = \lambda_4$

B.  $\lambda_1 = 2\lambda_2 = 2\lambda_3 = \lambda_4$

C.  $\lambda_1 = \lambda_2 = 4\lambda_3 = 9\lambda_4$

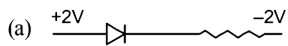
D.  $\lambda_1 = 2\lambda_2 = 3\lambda_3 = 4\lambda_4$

**Answer: C**

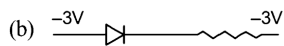


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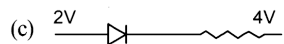
**282.** The forward biased diode connection is



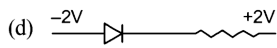
**A.**



**B.**



**C.**



D.

**Answer: A**



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**283.** A redc LED emits light at  $0.1\text{ W}$  uniformly around it . The amplitude of the electric field of the light at a distnce of  $3\text{ m}$  from the diode is

A.  $5.48\text{ Vm}$

B.  $7.75\text{ Vm}$

C.  $1.73\text{ Vm}$

D.  $2.45\text{ Vm}$

**Answer: D**



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**284.** A signal of  $5\text{ kHz}$  frequency is amplitude modulated on a carrier wave of frequency  $2\text{ MHz}$ . The frequency of the resultant signal is //are

A.  $2005\text{ kHz}$ ,  $200\text{ kHz}$  and  $1995\text{ kHz}$

B.  $2000\text{ kHz}$  and  $1995\text{ kHz}$

C.  $2\text{ MHz}$  only

D.  $2005\text{ Hz}$  and  $1995\text{ kHz}$

**Answer: A**



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**285.** As an electron makes a transition from an excited state to the ground state of a hydrogen-like atom //ion

A. kinetic energy decreases potential energy increases but total energy remains same

- B. kinetic energy and total energy decreases but potential energy increases
- C. in kinetic energy increases but potential energy and total energy decreases
- D. kinetic energy potential energy and total energy decreases

**Answer: C**



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**286.** For a common emitter configuration if  $a$  and  $\beta$  have their usual meaning, the incorrect relationship between  $a$  and  $\beta$  is :

A.  $a = \frac{\beta}{1 + \beta}$

B.  $a = \frac{\beta^2}{1 + \beta^2}$

C.  $\frac{1}{a} = \frac{1}{\beta} + 1$

D.  $a = \frac{\beta}{1 - \beta}$

Answer: B::D



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287. if  $a, b, c, d$  are inputs to a gate and  $x$  is its output , then as per the following time graph , the gate is :



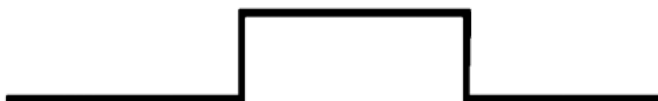
d

c



c

b



b

a



a



x

x



A. *OR*

B. *NAND*

C. *NOT*

D. *AND*

**Answer: A**



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**288.** Choose the correct statement :

A. in frequency modulation the amplitude of the high frequency carrier wave is made to vary in proportion to the amplitude of the audio signal .

B. in frequency modulation the amplitude of the high frequency carrier wave is made to vary in proportion to the frequency of the audio signal .

C. in amplitude modulation the amplitude of the high frequency carrier wave is made to vary in proportion to the amplitude of the audio signal .

D. in amplitude modulation the frequency of the high frequency carrier wave is made to vary in proportion to the amplitude of the audio signal .

**Answer: C**



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**289.** In a photoelectric experiment, with light of wavelength  $\lambda$ , the fastest electron has speed  $v$ . If the exciting wavelength is changed to  $\frac{3\lambda}{4}$ , the speed of the fastest emitted electron will become

A.  $= v \left( \frac{4}{3} \right)^{\frac{1}{2}}$

B.  $= v \left( \frac{3}{4} \right)^{\frac{1}{2}}$

C.  $> v \left( \frac{4}{3} \right)^{\frac{1}{2}}$

D.  $< v \left( \frac{4}{3} \right)^{\frac{1}{2}}$

**Answer: C**

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**290.** Half-lives of two radioactive elements A and B are 20 minutes and 40 minutes respectively. Initially, the samples have equal number of nuclei. After 80 minutes, the ratio of decayed numbers of A and B nuclei will be

A. 1 : 4

B. 5 : 4

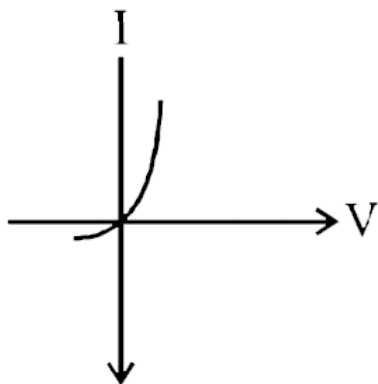
C. 1 : 16

D. 4 : 1

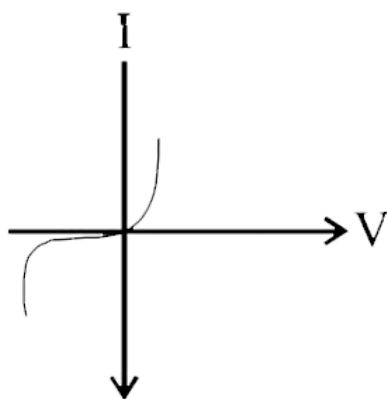
**Answer: c**

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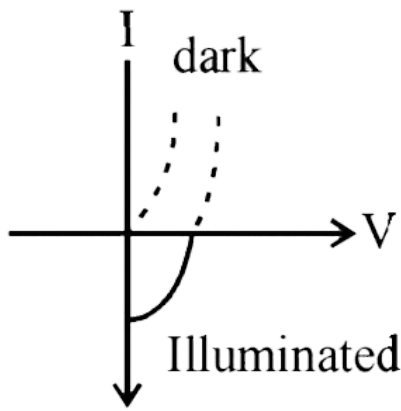
**291.** Identify the semiconductor devices whose characteristics are given below , in the order (a) ,(b) ,(c ) ,(d) :



(a)



(b)



(c)

(c)

(d) 

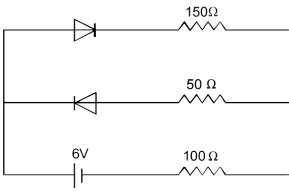
- A. Solar cell , Light dependent resistance, zener diode , simple diode
- B. Zener diode ,Solar cell , simple diode Light dependent resistance
- C. Simple diode,zener diode, Solar cell , Light dependent resistance
- D. Zener diode ,simple diode Light dependent resistance,Solar cell

**Answer: C**



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1. The circuit shown in the contain two diodes each with a forward resistance of  $50\ \Omega$  and with infinite back ward resistance , if the battery voltage is  $6V$  , the current through the  $100\ \Omega$  resistance (in amperes ) is



- A. zero
- B. 0.02
- C. 0.03
- D. 0.036

**Answer: B**



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