



CHEMISTRY

BOOKS - DISHA CHEMISTRY (HINGLISH)

ALDEHYDES, KETONES AND CARBOXYLIC ACIDS

Mcqs

1. Which of the following compounds is most reactive towards nucleophilic addition reactions?

A.
$$CH_3 - \overset{O}{\overset{||}{C}} - H$$

B. $CH_3 - \overset{O}{\overset{||}{C}} - CH_3$



Answer: A



2. Arrange the following in order of decreasing

acidity



A. B gt A gt C

B. C gt B gt A

C. A gt C gt B

D. A gt B gt C

Answer: A



3. A and B in the following reactions are









D. $A = RR'CH_2CN, B = NaOH$

Answer: A

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4. Acetaldehyde reacts with

A. Electrophiles only

B. Nucleophiles only

C. Free radicals only

D. Both electrophiles and nucleophiles

Answer: B



5.

$C_6H_5CH = CHCHO \xrightarrow{X} C_6H_5CH = CHCH_2OH$

In the above sequence X can be :

A. $H_2 \,/\, Ni$

B. $NaBH_4$

C. $K_2 C r_2 O_7 \,/\, H^{\,+}$

D. Both (a) and (b)

Answer: B



6. Which one of the following can be oxidised to the

corresponding carbonyl compound?

A. 2-hydroxy-propane

B. Ortho-nitro-phenol

C. Phenol

D. 2-methyl-2 hydroxy-propane



7. In the following reaction sequence, the correct structures of E, F and G are











Answer: C



8. ketones $[R-C-R_1]$, where $R=R_1$ = alkyl $\parallel
ho$

groups] can be obtained in one step by

A. oxidation of primary alcohols

- B. hydrolysis of esters
- C. oxidation of tertiary alcohols
- D. reaction of acid halides with alcohols

Answer: C



9. The compound that neither forms semicarbazone

nor oxime is

A. HCHO

 $\mathsf{B.}\,CH_3COCH_2Cl$

 $\mathsf{C.}\,CH_3CHO$

D. $CH_3CONHCH_3$

Answer: D





11. Which of the following compounds when heated

with CO at $150^{\,\circ}\,C$ and 500 atm pressure in presence

of BF_3 forms ethyl propionate ?

A. C_2H_5OH

 $\mathsf{B.}\,CH_3OCH_3$

 $\mathsf{C.}\, C_2H_5OC_2H_5$

 $\mathsf{D.}\, CH_2 OC_2 H_5$

Answer: C

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12. Benzaldehyde is obtained from Rosenmund's reduction of









Answer: B



13. Acetone oxime is obtained by reacting acetone

with

A. NH_3

B. NH_2OH

 $\mathsf{C}. NH_2Na$

D. NH_2NH_2

Answer: B



14.

 $2C_6H_5CHO \xrightarrow[H_2O]{OH^-} C_6H_5CH_2OH + C_6H_5COO^-$ Which of the following statements are correct regarding the above reduction of benzaldehyde to benzyl alcohol? (i)One hydrogen is coming from H_2O as H^+ and another from C_6H_5CHO as H^- (ii)One hydrogen is coming from H_2O as H^- and another from C_6H_5CHO as H^+ (iii) One hydrogen from H_2O and another from C_6H_5CHO , both in the form of H^- (iv) The reduction is an example of disproportionation reaction

A. (i),(ii) and (iii)

B. (i) and (iv)

C. (ii),(iii) and (iv)

D. (iii) and (iv)



15. A carboxylic acid can best be converted into acid chloride by using

A. PCl_5

B. $SOCl_2$

C. HCl

D. CICOCOCI

Answer: D





16. Among the given compounds, the most susceptible to nucleophilic attack at the carbonyl group is

A. MeCOCl

B. MeCHO

C. MeCOOMe

D. MeCOOCOMe

Answer: A



17. Pinacolone is

A. 2, 3-Dimethyl-2, 3-butanediol

B. 3,3-Dimethyl-2-butanone

C. 1-Phenyl-2-propanone

D. 1, 1-Diphenyl-1, 2-ethandiol

Answer: B



18. The correct sequence of reagents for the following conversion will be :



A.

 $ig[Ag(NH_3)_2ig]^+OH^-,\,H^+\,/CH_3OH,\,CH_3MgBr$ B.

 $CH_{3}MgBr,$ H^{+} / $CH_{3}OH,$ $\left[Ag(NH_{3})_{2}
ight]^{+}OH^{-}$ C.

 $CH_{3}MgBr,\left[Ag(NH_{3})_{2}
ight]^{+}OH^{-},H^{+}/CH_{3}OH$ D.

 $\left[Ag(NH_3)_2
ight]^+OH^-, CH_3MgBr, H^+/CH_3OH$



19. Benzaldehyde reacts with ethanoic KCN to give

A. $C_6H_5CHOHCN$

B. $C_6H_5CHOHCOC_6H_5$

 $\mathsf{C.}\, C_6H_5CHOHCOOH$

D. $C_6H_5CHOHCHOHC_6H_5$

Answer: B



20. Which gives lactic acid on hydrolysis after reacting with HCN?

A. HCHO

 $\mathsf{B.}\,CH_3CHO$

 $\mathsf{C.}\, C_6H_5CHO$

D. CH_3COCH_3

Answer: B

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21. Reduction of 📄 can be carried out with

A. catalytic reduction

 $\operatorname{B.}\operatorname{Na}/\operatorname{C_2H_5OH}$

C. Wollf-Kishner reduction

D. $LiAlH_4$

Answer: C

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22. The end product B in the sequence of reactions

 $R-X \stackrel{CN^{\,-}}{\longrightarrow} A \stackrel{NaOH}{\longrightarrow} B$ is

A. an alkane

B. a carboxylic acid

C. sodium salt of carboxylic acid

D. a ketone

Answer: C



23. Phenylmethyl ketone can be converted into ethylbenzene in one step by which of the following reagents?

A. $LiAlH_4$

B. Zn-Hg/HCl

C. $NaBH_4$

D. CH_3MgI

Answer: B



24. Conversion of acetaldehyde into ethyl acetate in

presence of aluminium ethoxide is called

A. Aldol condensation

B. Cope reaction

C. Tischenko reaction

D. Benzoin condensation

Answer: C



25. An organic compound A upon reacting with NH_3 gives B. On heating B gives C. C in presence of KOH reacts with Br_2 to given $CH_3CH_2NH_2$. A is :

A. CH_3COOH

B. $CH_3CH_2CH_2COOH$

 $\mathsf{C.}\,CH_3 - \mathop{C}_{|}_{CH_3}H - COOH$

D. CH_3CH_2COOH

Answer: D



26. Which one of the following can be oxidised to

the corresponding carbonyl compound ?

A. 2-hydroxypropane

B. Ortho-nitrophenol

C. Phenol

D. 2-methyl-2 hydroxypropane

Answer: A



27. The reagent which can be used to distinguish acetophenone from benzophenone is

A. 2,4- dinitrophenylhydrazine

B. aqueous solution of $NaHSO_3$

C. benedict reagent

D. I_2 and Na_2CO_3

Answer: D



28. $R - CH_2 - CH_2OH$ can be converted into RCH_2CH_2COOH . The correct sequence of reagent is

A. PBr_3, KCN, H^+

B. PBr_3 , KCN, H_2

C. KCN, H^+

D. HCN, PBr_3, H^+

Answer: A



29. Sodium salt of an organic acid 'X' produces effervescence with conc. H_2SO_4 . 'X' reacts with the acidified aqueous $CaCl_2$ solution to give a white precipitate which decolourises acidic solution of $KMnO_4$. 'X' is :

A. C_6H_5COONa

B. HCOONa

C. CH_3COONa

D. $Na_2C_2H_4$

Answer: D



30. In a set of the given reactions, acetic acid yielded a product C.

 $CH_3COOH + PCl_5
ightarrow A \xrightarrow[Anh\,.\,AlCl_3]{C_6H_6} B \xrightarrow[Ether]{C_2H_5MgBr} C$

Product C would be

A.
$$CH_3- \stackrel{C_2H_5}{\stackrel{|}{C}}(OH)C_6H_5$$



$\mathsf{C.}\,CH_3COC_6H_5$

D. $CH_3CH(OH)C_6H_5$

Answer: A



31. Which one of the following will most readily be

dehydrated in acidic condition ?









Answer: A



32. Which of the following contain an aldehyde?

A. Vanilla beans

B. Meadow sweet

C. Cinnamon

D. All of these



33. Heating mixture of sodium benzoate and soda-

lime gives

A. benzene

B. methane

C. sodium phenoxide

D. calcium benzoate

Answer: A





34. Observe the following structures and pick up the

correct statement

A. Carbonyl carbon of I is more electrophilic than

that of II

B. Carbonyl carbon of I is less electrophilic than

that of II

C. Carbonyl carbon of both structures have equal

electrophilic character

D. It depends upon the complete structure of the

compound

Answer: B



35. An enantiomerically pure acid is treated with a racemic mixture of an alcohol having one chiral carbon. The ester formed will be

A. Optically active mixture

B. Pure enantiomer

- C. Meso compound
- D. Racemicmixture

Answer: A



36. m-Chlorobenzaldehyde on reaction with conc.

KOH at room temperature gives

A. Potassium m-chlorobenzoate and m-

hydroxybenzaldehyde

B. m-hydroxybenzaldehyde and m-chlorobenzyl alcohol C. m-chlorobenzyl alcohol and m-hydroxybenzyl alcohol D. potassium m-chlorobenzoate and mchlorobenzyl alcohol Answer: D



37. The correct order of increasing acid strength of

the compounds

(A) CH_3CO_2H , (B) $MeOCH_3CO_2H$

(C) CF_3CO_2H

A. D lt A lt B lt C

B. A lt D lt B lt C

C. B lt D lt A lt C

D. D lt A lt C lt B

Answer: A



38. The increasing order of the rate of HCN addition to compound A-D is (A)HCHO , (B) CH_3COCH_3 , (C) $PhCOCH_3$, (D)PhCOPh

A. D lt Clt Blt A

B. C It D It B It A

C. A lt B lt C lt D

D. D lt B lt C lt A

Answer: A



39. The carboxyl functional group (-COOH) is present in

A. pictic acid

B. barbituric acid

C. ascorbic acid

D. aspirin

Answer: D

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 $CH_{3}CH_{2}CHO$ and $CH_{3}CCH_{3}$



 $\mathsf{B.}\,CH_3CH_2CH=CHCH_2CH_3$

$\mathsf{C}.\,CH_3CH_2CH=CHCH_3$

D.
$$CH_3 - \mathop{C}\limits_{\substack{\mid \ CH_3}} = CHCH_3$$

Answer: A

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41. Which one of the following is reduced with zinc and hydrochloric acid to give the corresponding hydrocarbon ?

A. Acetamide

B. Acetic acid

C. Ethyl acetate

D. Butan-2-one

Answer: D



42. Acetal is produced by reacting an alcohol in the

presence of dry HCl with

A. acetaldehyde

B. ketone

C. ether

D. carboxylic acid

Answer: A

