

CHEMISTRY

BOOKS - DISHA CHEMISTRY (HINGLISH)

HYDROCARBONS

Mcq

1. When neopentyl bromide is subjected to Wurtz reaction, the product formed is

A. 2,2,4,4-tetramethy lhexane

B. 2,2,4,4-tetramethy Ipentane

- C. 2,2,5,5-tetrametllylhexane
- D. 2,2,3,3-tetrametllylhexane

Answer: C



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2. The conversion of 2,3,-dibromobutane to 2-butene with

Zn and alcohol is

- A. redox reaction
- B. α -climination
- C. β -elimination
- D. Both (a) and (b)

Answer: C



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- **3.** 1,3-Butadiene when treated with Br_2 gives
 - A. 1,4-dibromo-2-butene
 - B. 1,3-bibromo-2-butene
 - C. 3,4-dibromo-1-butene
 - D. 2,3-dibromo-2-butene

Answer: A



4. An alkene having molecular formula C_2H_{14} was subjected to ozonolysis in the presence of zinc dust. An equimolar amount of the following two compounds was obtained CH_3COCH_3 and $CH_3COC_2H_5$

The IUPAC name of the alkene is

- A. 3,4-dimethyl-3-pentene
- B. 3,4-dimethyl-2-pentene
- C. 2,3-dimethyl-2-pentene
- D. 2,3-dimethyl-2-pentene

Answer: D



5. Acid catalyzed hydration ofalkenes except ethene leads to the formation of

A. mixture of secondary and tertiary alcohols

B. mixture of primary and secondary alcohols

C. secondary or tertiary alcohol

D. prirnaryalcohol

Answer: C



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6. In a reaction

$$CH_2 = CH_2 \stackrel{ ext{Hypochlorous}}{\longrightarrow} M \stackrel{R}{\longrightarrow} egin{pmatrix} CH_2 - OH \ & CH_2 - OH \end{pmatrix}$$

Where M =molecule, R=reagent, M and R are

A. CH_3CH_2Cl and NaOH

B. $CH_2Cl - CH_2OH$ and $aq. NaHCO_3$

 $C. CH_3CH_2OH \text{ and } HCl$

D. $CH_2=CH_2$ and heat

Answer: D



7. The negative part of an addendum adds on to the carbon atom joined to the least number of hydrogen atoms. This statement is called

A. Thiele's theory

- B. Peroxide effect
- C. Markownikoff's rule
- D. Baeyer's strain theory

Answer: C



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8. Match the columns



- A. A-III, B-IV, C-I, D-II
- B. A-IV, B-III, C-II, D-I
- C. A-II, B-I, C-IV, D-III
- D. A-III, B-IV, C-II, D-I

Answer: A



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9. Which of the following will yielda mixture of 2-chlorobutene and 3-chlorobutene on treatment with HCl?

A.
$$CH_2=C=CH-CH_3$$

$$\operatorname{B.}H_2C = \mathop{C}\limits_{CH_3} - CH = CH_2$$

$$C. CH_2 = CH - CH - CH_2$$

D.
$$HC = C - CH = CH_2$$

Answer: A



10. Lindlar's catalyst is

A. Na in alcohol

B. Rancy nickcl

C. $Pd/BaSO_4$

D. Na/liq. NH_3

Answer: C



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11. The treatment of CH_3MgX with CH_3C-C-H produces

 $A. CH_3 - CH = CH_2$

$$B. CH_3C = C - CH_3$$

$$C. CH_2 - HC = CH - CH_3$$

D. CH_4

Answer: D



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12. A group which deactivates the benzene ring towards electrophilic substitution but which directs the incoming group principally to the o- and p-positions is

A.
$$-NH_2$$

$$B.-Cl$$

$$\mathsf{C.}-NO_2$$

$$\mathsf{D.}-C_2H_5$$

Answer: B



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13. Isopropyl alcohol is obtained by reacting which of the following alkenes with conc. $H_2SO_4 \ {
m and} \ H_2O$

- A. Ethylene
- B. Propylene
- C. 2-methyl propene
- D. Isoprene

Answer: B

- **14.** In the preparation of alkanes from hydrogenation of alkenes and alkynes. Finely divided catalysts are used which of the following statement(s) is/are correct regarding these catalysts
- (i) Platinum and palladium catalyse the reaction at room temperature.
- (ii) Nickel catalyse the reaction at relatively higher temperature and pressure.
- (iii) Platinum and palladium catalyse the reaction at higher temperature.

A. I and iii

B. I and ii

- C. ii and iii
- D. I only

Answer: B



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15. Which one of the following reactions is expected to readily give a hydrocarbon product in good yields?

A. RCOOK
$$\xrightarrow{\text{Elegtrolytic}}$$
 oxidation

B.
$$RCOO^-Ag^+ \stackrel{Br_2}{\longrightarrow}$$

$$\mathsf{C.}\ CH_3CH_3 \xrightarrow[hv]{Cl_2}$$

D.
$$(CH_3)_3CCl \xrightarrow{C_2H_5OH}$$

Answer: A



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16. Which of the following change is correct

- A. 🖳
- В. 📝
- C. Both (a) and (b) one correct
- D. Neither (a) nor (b)

Answer: C



17. The products obtained via oxymer curation $(HgSO_4 + H_2SO_4)$ of butyne would be

A.
$$CH_3-CH_2-CO-CH_3$$

$$\mathsf{B.}\,CH_3-CH_2-CH_2-CHO$$

C.
$$CH_3 - CH_2 - CHO + HCHO$$

$$\mathsf{D.}\,CH_3CH_2COOH + HCOOH$$

Answer: A



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18. The number of optically active products obtained from the complete ozonolysis of the given compound is :



A. 0
B. 1
C. 2
D. 4
Answer: A
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19. Propyne on polymerisation gives
A. Mesitylene
B. Benzene
C. Ethyl benzene

D. Propyl benzene

Answer:



- **20.** The electrophilic substitutions reactions of benzene takes place via
- (i) generation of electrophile
- (ii) generation of nucleophile
- (iii) formation of carbocation intermediate
- (iv) removal of proton from the carbocation inter mediate
 - A. I, iii and iv
 - B. ii, iii and iv

C. I and iv

D. ii and iv

Answer: A



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21. Two organic compounds A and B both containing only carbon and hydrogen, on quantitative analysis gave the same percentage composition by weight :

$$C = (12/13) imes 100 \, \% \,, H = (1/13) imes 100 \, \%$$

A decolourises bromine water but B does not. A and B respectively are

A. C_2H_2 and C_6H_6

- $B. C_6H_6$ and C_2H_2
- $C. C_2H_4$ and C_2H_6
- D. C_2H_2 and C_2H_6

Answer: A



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22. The major product of the following reaction



- A. 🖳
- В. 🖳
- C. 📝
- D. 🖳

Answer: D



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23. 2,3-Dimethyl-2-butene can be prepared by heating which of the following compounds with a strong acid?

A.
$$(CH_3)_2CH-CH-CH=CH_2$$
 $_{CH_3}^{\parallel}$

B.
$$(CH_3)_3C - CH = CH_2$$

$$\mathsf{C.}\,(CH_3)_2C=CH-CH_2-CH_3$$

D.
$$(CH_3)_2CH - CH_2 - CH = CH_2$$

Answer: B



24. Which C-atom is the most electronegative in this

structure?

$$\overset{III}{C}H_3 - \overset{II}{C}H_2 - C = \overset{I}{C}H$$

A. I

B. II

C. III

D. all are equal electronegative

Answer: A



25. Which of the following will have least hindered rotation around carbon- carbon bond ?

- A. Ethane
- B. Ethylene
- C. Acctylene
- D. Hexachlorocthane

Answer: A



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26. Of the three isomeric C_3H_4 hydrocarbons shown below how many can exist with all carbon and hydrogen nuclci

located in a single plane A. 0 B. 1 C. 2 D. 3 **Answer: A View Text Solution 27.** Excess of CH_3COOH is reacted with $CH\equiv CH$ in presence of Hg^{2+} , the product is A. $CH_3CH(OCOCH_3)_2$

$$B. CH_2 = CH(OCOCH_3)$$

$$C.(CH_3COO)CH_2 - CH_2(OOCCH_3)$$

D. None of these

Answer: A



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28. Which one of the following contain isopropyl group?

- A. 2,2,3,3-tetramethyl pentane
- B. 2,4-dimethy lhexone
- C. 2,2,3-trimethylpentane
- D. 3,3- dimethylpentane

Answer: B



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29. 1-Pentyne



X and Y can be distinguished by

- A. Silver-mirror test
- B. lodoform test
- C. Both
- D. None

Answer: C



30. The major product obtained in the photo catalysed bromination of 2-methylbutane is :

- A. 1-bromo-2-methylbutane
- B. 1-bromo-3-methylbutane
- C. 2-bromo-2-methyl butane
- D. 2-bromo-2-methyl butane

Answer: D



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31. Acctylenic hydrogens are acidic because

- A. Sigma electron density of C-H bond in acetylene is nearer to carbon, which has 50% s-character
- B. Acetylene has only open hydrogen in each carbon
- C. Acetylene contains least number of hydrogens among the possible hydrocarbons having two carbons
- D. Acetylene belongs to the class of alkynes with molecular formula, $C_n H_{2n-2}$

Answer: A



32. Which of the following will be most easily attacked by an electrophilic?

A. C_6H_6

 $\operatorname{B.} C_6H_5Cl$

 $\mathsf{C.}\, C_6 H_5 O H$

D. $C_6H_5CH_3$

Answer: C



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33. Name of the following reaction is



A. Claisen Condensation

B. Diel's Alder reaction

C. Dieckmann cyclisation

D. Michael addition reaction

Answer: B



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34. The most suitable catalyst for the hydrogenation of 2-

Hexyne \rightarrow 2-cis-Hexene is

A. $Pd-BaSO_4$

 $\mathsf{B.}\,(Ph_3P)_3RhCl$

C. 10% Pd-C

D. Raney Ni

Answer: A

35.

- A. 1 mole of Ethane
- B. 3 mole of ethane
- C. $EtC \equiv CCH_2CH_2NHEt$
- D. 4 mole of ethane

Answer: B



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36. Among the following compounds (I-III), the correct order of reactivity with an electrophile is

$$C_6H_5OCH_3 \qquad C_6H_6 \qquad C_6H_5NO_2 \ _{III}$$

A.
$$II > III > I$$

$$\operatorname{B.}III < I < II$$

$$\mathsf{D}.\,I=II>III$$

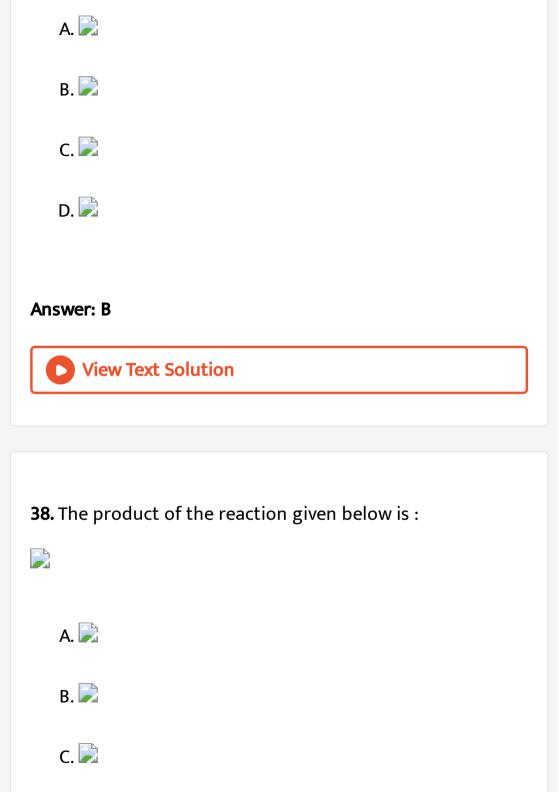
Answer: C



37. What is the major product expected from the following reaction ?



Where D is an isotope of hydrogen



Answer: D



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39.
$$CH_3C\equiv CCH_3 \xrightarrow[(ii)\,H_2O\,/\,Zn]{(ii)\,H_2O\,/\,Zn}$$

 $H_3C \overset{C}{||} - \overset{C}{||}CH_3$ In the above reaction X is

A. HNO_3

 $B.O_2$

 $\mathsf{C}.\,O_3$

D. $KMnO_4$

Answer: C

40. Which of the following represent the correct order of acidic strength ?

$$(i)HC\equiv CH>H_2C=CH_2>CH_3-CH_3$$

(ii)
$$HC \equiv CH > CH_3 - CH_3 > H_2C = CH_2$$

(iii)

$$CH_3C\equiv CH>HC\equiv CH>CH_3-C\equiv C-CH_3$$

$$(iv)HC \equiv CH > CH_3 - C \equiv CH > CH_3 - C \equiv C - CH_3$$

A. I and iii

B. ii and iv

C. I and iv

D. I and iv

Answer: C



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41. Which one of the following compounds would have the highest heat of hydrogenation ?

A.
$$CH_2=CH_2$$

$$\operatorname{B.}CH_3-CH_2-CH=CH_2$$

$$C. CH_3CH = CHCH_3$$

D.
$$(CH_3)_2C - C(CH_3)_2$$

Answer: A



42. On mixing a certain alkane with chlorine and irradiating it with ultraviolet light, it forms only one monochloroalkaane. This alkane could be

- A. pentane
- B. isopentane
- C. neopentane
- D. propane

Answer: C



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43. Acetylene gives

- A. white ppt with $AgNO_3$ and red ppt with Cu_2Cl_2
- B. white ppt with Cu_2Cl_2 and red ppt with $AgNO_3$
- C. white ppt with both
- D. red ppt with both

Answer: A



- **44.** Sodium ethoxide is a specific reagent for
 - A. dehydration
 - B. dehydrogenation
 - C. dehydrohalogenation

D. dehalogenation

Answer: C

