



PHYSICS

BOOKS - GK PUBLICATIONS PHYSICS (HINGLISH)

PHOTOELECTRIC EFFECT AND MATTER WAVES

Illustration

1. The photoelectric threshold of the photo electric effect of a certain metal is 2750\AA . Find

- (i) The work function of emission of an electron from this metal,
- (ii) Maximum kinetic energy of these electrons.
- (iii) The maximum velocity of the electrons ejected from the metal by light with a wavelength 1800\AA .



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2. Light quanta with energy 4.9 eV eject photoelectrons from metal with work function 4.5 eV. Find the maximum impulse transmitted to the surface of the metal when each electron flies out.

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3. In an experiment tungsten cathode which has a threshold 2300\AA is irradiated by ultraviolet light of wavelength 1800\AA . Calculate

(i) Maximum energy of emitted photoelectron and

(ii) Work function for tungsten.

(Mention both the results in electron-volts)

Given Planck's constant $h = 6.6 \times 10^{-34} \text{ joule-sec}$,

$\leq V = 1.6 \times 10^{-19} \text{ joule}$ and velocity of light $c = 3 \times 10^8 \text{ m/sec}$

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4. Light of wavelength 180 nm ejects photoelectrons from a plate of metal whose work - function is 2 eV. If a uniform magnetic field of 5×10^{-5} T be applied parallel to the plate, what would be the radius of the path followed by electrons ejected normally from the plate with maximum energy.

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5. The radiation emitted when an electron jumps from $n = 3 \rightarrow n = 2$ orbit in a hydrogen atom falls on a metal to produce photoelectron. The electron from the metal surface with maximum kinetic energy are made to move perpendicular to a magnetic field of $(1/320)T$ in a radius of $10^{-3}m$. Find (a) the kinetic energy of the electrons, (b) Work function of the metal , and (c) wavelength of radiation.

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6. Photoelectrons are emitted when $400nm$ radiation is incident on a surface of work - function 1.9 eV . These photoelectrons pass through a region containing α -particles. A maximum energy electron combines with an α -particle to form a He^+ ion, emitting a single photon in this process. He^+ ions thus formed are in their fourth excited state. Find the energies in eV of the photons lying in the 2 to 4 eV range, that are likely to be emitted during and after the combination. [Take, $h = 4.14 \times 10^{-15} eV - s$]

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7. Find the frequency of light which ejects electrons from a metal surface. Fully stopped by a retarding potential of $3V$, the photoelectric effect begins in this metal at a frequency of $6 \times 10^{14} Hz$. Find the work function for this metal. (Given $h = 6.63 \times 10^{-34} Js$).

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8. Electrons with maximum kinetic energy 3eV are ejected from a metal surface by ultraviolet radiation of wavelength 1500\AA . Determine the work function of the metal, the threshold wavelength of metal and the stopping potential difference required to stop the emission of electrons.

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9. Calculate the velocity of a photo-electron if the work function of the target material is 1.24eV and the wavelength of incident light is $4.36 \times 10^{-7}\text{m}$. What retarding potential is necessary to stop the emission of the electrons?

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10. Determine the Planck's constant h if photoelectrons emitted from a surface of a certain metal by light of frequency 2.2×10^{15} Hz are fully retarded by a reverse potential of 6.6 V and those ejected by light of frequency 4.6×10^{15} Hz by a reverse potential of 16.5 eV.

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11. When a surface is irradiated with light of wavelength 4950 \AA , a photocurrent appears which vanishes if a retarding potential appears which vanishes if a retarding potential greater than 1.2 volt is applied across the phototube. When a different source of light is used, it is found that the critical retarding potential is changed to 2.1 volt. Find the work function of the emitting surface and the wavelength of second source. If the photoelectrons (after emission from the surface) are subjected to a magnetic field of 10 tesla, what changes will be observed in the above two retarding potentials. Use $h = 6.6 \times 10^{-34} \text{ Js}$, $e = 1.6 \times 10^{-19} \text{ C}$.

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12. (a) If the wavelength of the light incident on a photoelectric cell be reduced from λ_1 to $\lambda_2 \text{ \AA}$, then what will be the change in the cut-off potential?

(b) Light is incident on the cathode of a photocell and the stopping voltages are measured for light of two different wavelengths. From the data given below, determine the work function of the metal of the cathode in eV and the value of the universal constant hc/e .

Wavelength \AA	Stopping voltage (volt)
4000	1.3
4500	0.9

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13. Ultraviolet light of wavelength 2271 \AA from a 100W mercury source irradiates a photocell made of molybdenum metal. If the stopping potential is 1.3V, estimate the work function of the metal. How would

the photocell respond to a high intensity ($\sim 10^5 \text{ W m}^{-2}$) red light of wave length 6328 \AA produced by a He-Ne laser?

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14. The peak emission from a black body at a certain temperature occurs at a wavelength of 9000 \AA . On increase its temperature, the total radiation emitted is increased its 81 times. At the initial temperature when the peak radiation from the black body is incident on a metal surface, it does not cause any photoemission from the surface. After the increase of temperature, the peak from the black body caused photoemission. To bring these photoelectrons to rest, a potential equivalent to the excitation energy between $n = 2$ and $n = 3$ bohr levels of hydrogen atoms is required. Find the work function of the metal.

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15. Find the number of photons emitted per second by a 25 watt source of monochromatic light of wavelength 6000\AA .

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16. Calculate the number of photons emitted in 10 h by a 60 W sodium lamp ($\lambda = 5893\text{\AA}$).

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17. A cylindrical rod of some laser material $5 \times 10^{-2}\text{m}$ long and 10^{-2}m in diameter contains 2×10^{25} ions per m^3 . If on excitation all the ions are in the upper energy level and de-excite simultaneously emitting photons in the same direction, calculate the maximum energy contained in a pulse of radiation of wavelength $6.6 \times 10^{-7}\text{m}$. If the pulse lasts for 10^{-7}s , calculate the average power of the laser during the pulse.

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18. A 100 W sodium lamp is radiating light of wavelength 5890\AA , uniformly in all directions,

a. At what rate, photons are emitted from the lamp?

b. At what distance from the lamp, the average flux is $1 \text{ photon}(\text{cm}^2 - \text{s})^{-1}$?

c. What are the photon flux and photon density at 2m from the lamp?

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19. One milliwatt of light of wavelength 4560 \AA is incident on a cesium surface. Calculate the photoelectric current liberated assuming a quantum efficiency of 0.5 %. Given Planck's constant $h = 6.62 \times 10^{-34} \text{ J} - \text{s}$ and velocity of light $c = 3 \times 10^8 \text{ ms}^{-1}$.

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20. A uniform monochromatic beam of light of wavelength $365 \times 10^{-9} \text{ m}$ and intensity 10^{-8} W m^{-2} falls on a surface having absorption coefficient 0.8 and work function 1.6 eV. Determine the rate of number of electrons emitted per m^2 , power absorbed per m^2 and the maximum kinetic energy of emitted photoelectrons.

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21. When a beam of 10.6 eV photons of intensity 2.0 W / m^2 falls on a platinum surface of area $1.0 \times 10^4 \text{ m}^2$ and work function 5.6 eV , 0.53 % of the incident photons eject photoelectrons. find the number of photoelectrons emitted per second and their minimum and maximum energies (in eV) Take $1 \text{ eV} = 1.6 \times 10^{-19} \text{ J}$

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22. A beam of light has three wavelengths 4144\AA , 4972\AA and 6216\AA with a total intensity of $3.6 \times 10^{-3} \text{Wm}^{-2}$ equally distributed amongst the three wavelengths. The beam falls normally on an area 1.0cm^2 of a clean metallic surface of work function 2.3eV . Assume that there is no loss of light by reflection and that each energetically capable photon ejects an electron. Calculate the number of photoelectrons liberated in two seconds.



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23. A small plate of a metal (work function = 1.7eV) is placed at a distance of 2m from a monochromatic light source of wavelength $4.8 \times 10^{-7}\text{m}$ and power 1.0watt . The light falls normally on the plate. Find the number of photons striking the metal plate per square meter per second. If a constant magnetic field of strength 10^{-4}T is applied parallel to the metal surface, find the radius of the largest circular path followed by the emitted photoelectrons. (use

$h = 6.63 \times 10^{-34} \text{Js}$, mass of electron = $9.1 \times 10^{-31} \text{kg}$, charge of electron = $1.6 \times 10^{-19} \text{C}$

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24. A 40 W ultraviolet light source of wavelength 2480\AA . Illuminates a magnesium (Mg) surface placed 2 m away. Determine the number of photons emitted from the surface per second and the number incident on unit area of Mg surface per second . The photoelectric work function for Mg is 3.68eV . Calculate the kinetic energy of the fastest electrons ejected from the surface. Determine the maximum wavelength for which the photoelectric effects can be observed with Mg surface.

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25. A mercury arc lamp provides 0.10 W of UV radiation at a wavelength of $\lambda = 2537 \text{\AA}$ (all other wavelengths having been

absorbed by filters). The cathode of photoelectric device (a photo-tube) consists of potassium and has an effective area of 4cm^2 . The anode is located at a distance of 1 m from radiation source. The work function for potassium is $\phi_0 = 2.22\text{eV}$.

(a) According to classical theory, the radiation from the arc spreads out uniformly in space as spherical wave. What time of exposure to the radiation should be required for a potassium atoms(radius 2 \AA) in the anode to accumulate sufficient energy to eject a photo-electron?

(b) What is the energy of a single photon from the source?

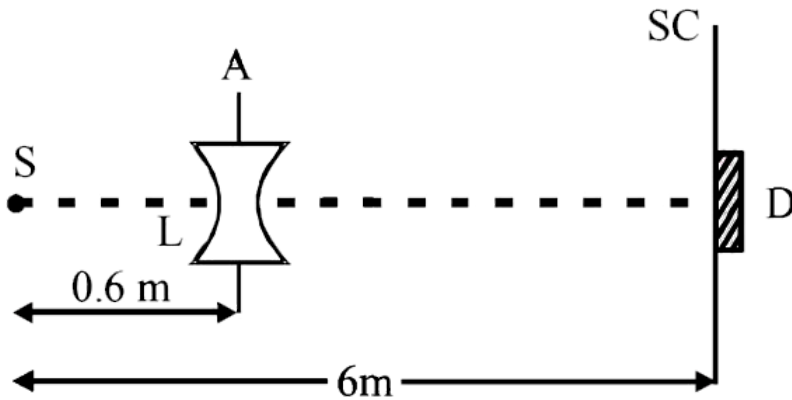
(c) What is the flux of photons(number per second) at the cathode?

To what saturation current does this flux correspond if the photo-conversion efficiency is 5% (i.e. if each photon has a probability of 0.05 of ejecting an electron).

(d) what is the cutt off potential V_0 ?

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26. A monochromatic point source radiating wavelength 6000\AA with power 2 watt, an aperture A of diameter 0.1m and a large screen SC are placed as shown in fig, A photoemissive detector D of surface area 0.5cm^2 is placed at the centre of the screen . The efficiency of the detector for the photoelectron generation per incident photon is 0.9



(a) Calculate the photon flux at the centre of the screen and the photo current in the detector.

(b) If the concave lens L of focal length 0.6 m is inserted in the aperture as shown . find the new values of photon flux and photocurrent Assume a uniform average transmission of 80% from the lens .

(c) If the work function of the photoemissive surface is $1eV$. calculate the values of the stopping potential in the two cases (within and with the lens in the aperture).

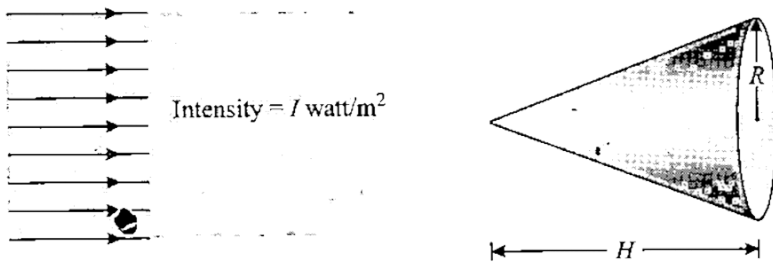
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27. In the path of a uniform light beam of large cross-sectional area and intensity I . a solid sphere of radius R which is perfectly reflecting is placed . Find the force exerted on this sphere due to the light beam.

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28. Figure 2.25 shows a cone of radius R and height H with perfectly reflecting lateral surface, is placed in the path of a light beam of

intensity I . Find the force exerted on this cone.



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29. With what velocity must an electron travel so that its momentum is equal to that of a photon with a wavelength of 5000\AA ($h = 6.6 \times 10^{-34} \text{ Js}$, $m_e = 9.1 \times 10^{-31} \text{ Kg}$)

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30. Hydrogen gas in the atomic state is excited to an energy level such that the electrostatic potential energy of H-atom becomes -1.7eV . Now, a photoelectric plate having work function $w=2.3 \text{ eV}$ is exposed to the emission spectra of this gas. Assuming all the

transitions to be possible, find the minimum de-Broglie wavelength of the ejected photoelectrons.

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31. An α -particle and a proton are fired through the same magnetic field which is perpendicular to their velocity vectors. The α -particles and the proton move such that radius of curvature of their paths is same. Find the ratio of their de Broglie wavelengths.

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32. Find the ratio of de Broglie wavelength of a proton and an α -particle which have been accelerated through same potential difference.

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33. Assume that the de Broglie wave associated with an electron can form a standing wave between the atoms arranged in a one-dimensional array with nodes at each of the atomic sites. It is found that one such standing wave is formed if the distance d between the atoms of the array is 2λ . A similar standing wave is again formed if d is increased to 2.5λ . Find the energy of the electrons in electron volts and the least value of d for which the standing wave type described above can form.



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34. In a photoelectric effect set up, a point source of light of power $3.2 \times 10^{-3} \text{ W}$ emits monochromatic photons of energy 5 eV . The source is located at a distance of a stationary metallic sphere of work function 3 eV and radius $8 \times 10^{-3} \text{ m}$. The efficiency of photoelectron emission is one for every 10^6 incident photons. Assume that the sphere is isolated and initially neutral and the photoelectrons are

initially swept away after emission.

Find the number of photons emitted per second

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Discussion Question

1. When the speed of a particle doubles, its momentum doubles, and its kinetic energy becomes four times greater. When the momentum of a photon doubles, does its energy become four times greater? Provide a reason for your answer.

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2. It is found that yellow light does not eject photoelectrons from a metal. Is it advisable to try with orange light? With green light?

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3. It is found that photosynthesis starts in certain plants when exposed to the sunlight but it does not start if the plant is exposed only to infrared light. Explain.

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4. Photon A has twice the energy of photon B.

(a) If the momentum of A less than, or equal to , or greater than that of B?

(b) is the wavelength of A less than, equal to, or greater than that of B?

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5. If electromagnetic radiation is made up of quanta, why don't we detect the discrete packets of energy, for example, when we listen to

a radio?



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6. The threshold wavelength of a metal is (λ_0). Light of wavelength slightly less than λ_0 is incident on an insulated plate made of this metal. It is found that photoelectrons are emitted for sometimes and after that the emission stops. Explain.



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7. Consider the de Broglie wavelength of an electron and a proton. Which wavelength is smaller if the two particles have (a) the same speed (b) the same momentum (c) the same energy?



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8. If an electron has a wavelength, does it also have a colour?

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9. Human skin is relatively insensitive to visible light, but ultraviolet radiation can cause severe burns. Does this have anything to do with photon energies? Explain

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10. Do all electrons emitted in the photoelectric effect have the same kinetic energy?

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11. If you shine ultraviolet light on an isolated metal plate, the plate emits electrons for a while. Why does it eventually stop?

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12. Is it always true that for two sources of equal intensity, the number of photons emitted in a given time are equal?

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13. The wave theory of radiation cannot explain..... Existence of threshold frequency, dependence of photocurrent on intensity, dependence of stopping potential on frequency, that stopping potential is independent of intensity?

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14. The photoelectrons emitted by an illuminated surface have a maximum kinetic energy of 3.0 eV. If the intensity of the light is tripled, What is the maximum kinetic energy of photoelectrons now?

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15. The threshold frequencies for photoemission for three metals numbered 1,2,3 are respectively ν_1, ν_2, ν_3 and $\nu_1 > \nu_2 > \nu_3$. An incident radiation of frequency $\nu_0 > \nu_2$cause the photoemission from 3 but.....cause photoemission from 1.

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16. Is $p = \frac{E}{c}$ valid for electrons?

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17. Photon A is from an ultraviolet tanning lamp and photon B is from a television transmitter. Which has the greater

- A. Wavelength
- B. energy
- C. frequency and
- D. momentum

Answer:



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18. Of the following statement about the photoelectric effect, which are true and which are false?

- (a) The greater the frequency of the incident light is, the greater is the stopping potential.
- (b) The greater the intensity of the incident light is, the greater is the

cutoff frequency.

(c) The greater the work function of the target material is, the greater is the stopping potential.

(d) The greater the work function of the target material is, the greater is the cutoff frequency.

(e) The greater the frequency of the incident light is, the greater is the maximum kinetic energy of the ejected electrons.

(f) The greater the energy of the photons is, the smaller is the stopping potential.

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19. In an experiment on photoelectric effect, a photon is incident on an electron from one direction and the photoelectron is emitted almost in the opposite direction. Does this violate conservation of momentum ?

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20. Can we find the mass of a photon by the definition $p = mv$?

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21. If the intensity of the light producing a photocurrent is doubled, how is that current affected?

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22. Should the energy of a photon be called its kinetic energy or its internal energy?

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23. In an experiment on photoelectric effect, a photon is incident on an electron from one direction and the photoelectron is emitted

almost in the opposite direction. Does this violate conservation of momentum ?

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24. The de-Broglie hypothesis predicts that a wave associated with any object has momentum. Why do we not observe the wave nature of a moving car?

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25. In the photoelectric effect, if the frequency of the radiation is below a certain cutoff value, no photoelectrons will be observed no matter how intense the radiation is. Why does this fact favour a particle theory of light over a wave theory?

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26. The photons emitted by a source of light do not all have the same energy. Is the source monochromatic? Give your reasoning.

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27. Radiation of a given wavelength causes electrons to be emitted from the surface of one metal but not from the surface of another metal. Explain why this could be

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28. Can a photon be deflected by an electric field ? By a magnetic field ?

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29. In the photoelectric effect, suppose that the intensity of the light is increased, while the frequency is kept constant. The frequency is greater than the minimum frequency f_0 . State whether each of the following will increase, decrease, or remain constant, and explain your choice. (a) the current in the phototube, (b) the number of electrons per second from the metal surface, (c) the maximum kinetic energy that an electron could have, (d) the maximum momentum that an electron could have, and (e) the maximum de-Broglie wavelength that an electron could have.

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30. A hot body is placed in a closed room maintained at a lower temperature. Is the number of photons in the room increasing?

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31. What is the speed of a photon with respect to another photon if (a) the two photons are going in the same direction and (b) they are going in opposite directions?



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32. which coloured light bulb, red, orange, yellow, green, or blue, emits photons with (a) the least energy and (b) the greatest energy? Account for your answers.



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33. A stone is dropped from the top of a building. As the stone falls, does its de-Broglie wavelength increase, decrease, or remains the same? Provide a reason for your answer.



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34. "If the frequency of light incident on a metallic plate be doubled."

Explain if this statements is TRUE or FALSE



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35. The frequency and the intensity of a beam of light falling on the surface of a photoelectric material are increased by a factor of two.

This will

- A. Increase the maximum kinetic energy of the photoelectrons, as well as photoelectric current by a factor of two
- B. Increase the maximum kinetic energy of the photoelectrons and would increase the photo electric current by a factor of two
- C. Increase the maximum kinetic energy of the photoelectrons by a factor of two and will have no effect on the magnitude of the

photo electrons by a factor of two and will have no effect on the magnitude of the photoelectric current produced

D. No produce any effect on the kinetic energy of the emitted electrons but will increase the photoelectric current by a factor of two.

Answer:

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36. Let be the maximum kinetic energy of photoelectrons emitted by light of wavelength λ_1 and λ_2 corresponding to wavelength λ_2 . If $\lambda_1 = 2\lambda_2$ then:

A. $2K_1 = K_2$

B. $K_1 = 2K_2$

C. $K_1 < K_2/2$

D. $K_1 > 2K_2$

Answer:

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37. When stopping potential is applied in an experiment on photoelectric effect, no photo current is observed. This means that

- A. The emission of photo electrons is stopped
- B. The photoelectrons are emitted but are reabsorbed by the emitter metal
- C. The photoelectrons are accumulated near the collector plate
- D. The photoelectrons are dispersed from the sides of the apparatus

Answer:

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38. If the frequency of light in a photoelectric experiment is doubled the stopping potential will

- A. Be doubled
- B. Behalved
- C. Become more than double
- D. Become less that double

Answer:

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39. A cesium photocell, with a steady potential difference of 60 V across it ,is illuminated by a small bright light placed 1m away. When

the same light is placed 2m away, the electrons crossing the photocell

- A. Each carry one quarter of their previous energy
- B. Each carry one quarter of their previous momentum
- C. Are half as numerous
- D. Are one quarter as numerous

Answer:



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40. A point source of light is used in a photoelectric effect. If the source is removed farther from the emitted metal, the stopping potential

- A. Will increase
- B. Will decrease

C. Will remain constant

D. Will either increase or decrease

Answer:



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41. A metal surface is illuminated by a light of given intensity and frequency to cause photoemission. If the intensity of illumination is reduced to one-fourth of its original value, then the maximum KE of emitted photoelectrons will become.

A. Unchanged

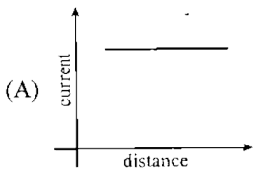
B. $\frac{1}{16}$ th of original value

C. Twice the original value

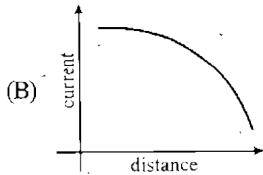
D. Four times the original value

Answer:

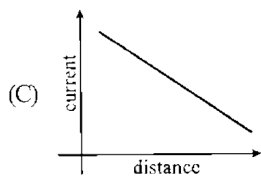
42. A point source causes photoelectric effect from a small metal plate. Which of the following curves may represent the saturation photocurrent as a function of the distance between the source and the metal?



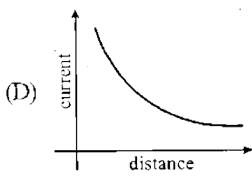
A.



B.



C.



D.

Answer:

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43. The stopping potential for a certain photosensitive metal is V_0 when the frequency of incident radiation is ν_0 . When the frequency of the incident radiation is doubled, what will be the stopping potential?

A. V_0

B. $2V_0$

C. $4V_0$

D. None of the above

Answer:



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44. The collector plate in an experiment of photoelectric effect is kept vertical above the emitter plate. Light source is put on, and a saturation photocurrent is recorded. An electric field is switched on which has a vertically downward direction-

- A. The photo current will increase
- B. The kinetic energy of the electron will increase
- C. The stopping potential will decrease
- D. The threshold wavelength will increase

Answer:



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45. Which electrons are emitted in the photoelectric effect

- A. Electrons in the inner orbits of the atom
- B. Electrons in the outermost orbit of the atom
- C. Electrons from within the nucleus
- D. Electrons freely roaming about in the interatomic space

Answer:

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46. Light of two different frequencies whose photons have energies 1eV and 2.5 eV respectively illuminate a metallic surface whose work function is 0.5 eV successively. Ratio of maximum kinetic energy of emitted electrons will be:

- A. 4: 1

B. 1:4

C. 1:5

D. 1:2

Answer:



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47. The image of the sun is formed on the photosensitive metal with a convex lens and the photoelectric saturation current obtained is I . If, the lens is replaced by another similar lens of half the diameter and double the focal-length, then photoelectric -current will be (Assume all light passing through lens falling on the photosensitive metal):

A. I

B. $\frac{I}{4}$

C. $\frac{I}{8}$

D. $\frac{I}{6}$

Answer:

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48. When monochromatic light of wavelength λ illuminates a metal surface, then stopping potential for photo electric current is $3V_0$. If wavelength changes to 2λ then stopping potential becomes V_0 . Threshold wavelength for photo electric emission is:

- A. 4λ
- B. 8λ
- C. $\frac{4}{3}\lambda$
- D. 6λ

Answer:

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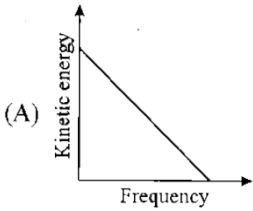
49. According to Einstein in theory of photoelectric effect, light is not behaving like a wave because :

- A. Kinetic energy of ejected electrons does not depend on the intensity of light
- B. Energy absorbed by an electron from a spreading wavefront is negligible
- C. No electron is ejected if the frequency is not more than threshold frequency whatever may be the intensity of light
- D. all the above

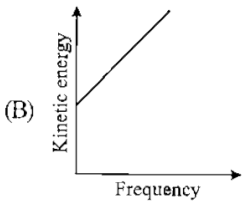
Answer:

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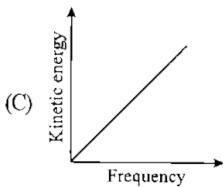
50. According to Einstein's photoelectric equation , the graph between the kinetic energy of photoelectrons ejected and the frequency of incident radiation is



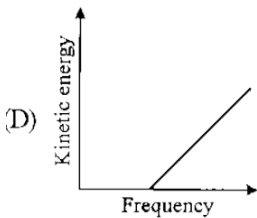
A.



B.



C.



D.

Answer:



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Conceptual Mcq

1. The photoelectric current in a photoelectric cell depends upon:

- A. The nature of the metal used as the emitter
- B. The wavelength of the incident light
- C. The intensity of the incident light
- D. All the above parameters.

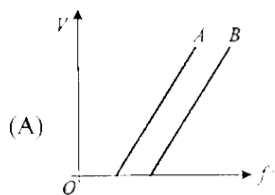
Answer:



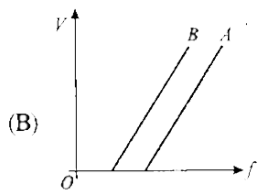
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2. In a photoelectric experiment, two metal pairs plates A and B are used for a given light intensity I and frequency f . work function of

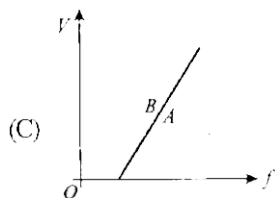
metal B is more than that of A. The correct variation of stopping potential difference versus frequency is given by:



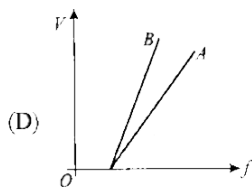
A.



B.



C.



D.

Answer:

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3. The threshold wavelength for photoelectric emission for a material is 5200\AA . Will the photoelectrons be emitted when this material is illuminated with monochromatic radiation from 1 watt ultra violet lamp?

- A. 50-watt infrared lamp
- B. 100-watt red neon lamp
- C. 40-watt sodium lamp
- D. 5-watt ultraviolet lamp

Answer:

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4. A photon of light enters a block of glass after travelling through vacuum. The energy of the photon on entering the glass block

- A. A photon of light enters a block of glass after travelling through a vacuum. The energy of the photon on entering the glass block
- B. Decreases because the speed of the radiation decreases
- C. Stays the same because the speed of the radiation and the associated wavelength do not change
- D. Stays the same because the frequency of the radiation does not change

Answer:

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5. Which one of the following statements is NOT true for de Broglie waves?

- A. All atomic particles in motion have waves of a definite wavelength associated with them
- B. The higher the momentum, the longer is the wavelength
- C. The faster the particle, the shorter is the wavelength
- D. For the same velocity, a heavier particle has a shorter wavelength

Answer:

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6. Moving with the same velocity . One of the following has the longest deBroglie wavelength

- A. α -particle
- B. β -particle
- C. Proton

D. Neutron

Answer:

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7. Two photons having

- A. Equal wavelengths have equal linear momenta
- B. Equal energies have equal linear momenta
- C. Equal frequencies have equal linear momenta
- D. Equal linear momenta have equal wavelengths

Answer:

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8. A proton and an electron move with the same velocity. The associated wavelength for proton is :

- A. Shorter than that of the electron
- B. Longer than that of the electron
- C. The same as that of the electron
- D.) Zero

Answer:

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9. In photoelectric effect when photons of energy $h\nu$ fall on a photosensitive surface (work function $h\nu^0$) electrons are emitted from the metallic surface with a kinetic energy. It is possible to say that:

A. All ejected electrons have same kinetic energy equal to

$$hv - hv_0$$

B. The ejected electrons have a distribution of kinetic energy from

$$\text{zero to } (hv - hv_0)$$

C. The most energetic electrons have kinetic energy equal to hv_0

D. All ejected electrons have kinetic energy hv_0

Answer:



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10. Photoelectric emission occurs only when the incident light has more than a certain minimum

A. Minimum frequency

B. Minimum wavelength

C. Minimum intensity

D. Minimum speed

Answer:

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11. Photo electric effect can be explained only by assuming that light

- A. Is a form of transverse waves
- B. Is a form of longitudinal waves
- C. Can be polarized
- D. Consists of quanta

Answer:

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12. At frequencies of the incident radiation above the threshold frequency, the photoelectric current in a photoelectric

- A. Intensity of incident radiation
- B. Wavelength of incident radiation
- C. Frequency of incident radiation
- D. Speed of incident radiation

Answer:



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13. A metallic surface has a threshold wavelength 5200 Å. This surface is irradiated by monochromatic light of wavelength 4500 Å. Which of the following statements is true ?

- A. The electrons are emitted from the surface having energy between 0 and infinity
- B. The electrons are emitted from the surface having energy between 0 and certain finite maximum value
- C. The electrons are emitted from the surface, all having certain finite energy
- D. No electrons is emitted from the surface

Answer:

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14. Let p and E denote the linear momentum and energy of a photon.

If the wavelength is decreased,

- A. Both p and E increase
- B. p increases and E decreases

C. p decreases and E increases

D. Both p and E decreases

Answer:

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15. The equation $E=pc$ is valid

A. For an electron as well as for a photon

B. For an electron but not for a photon

C. For a photon but not for an electron

D. Neither for an electron nor for a photon

Answer:

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16. Let n_r and n_b be respectively the number of photons emitted by a red bulb and a blue bulb of equal power in a given time.

A. $n_r = n_b$

B. $n_r < n_b$

C. $n_r > n_b$

D. The information is insufficient to get a relation between n_r and n_b

Answer:

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17. The photoelectrons emitted from a metal surface:

A. Are all at rest

B. Have the same kinetic energy

C. Have the same momentum

D. Have speeds varying from zero upto a certain maximum

Answer: D



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18. The graph between, which of the following two factors for photoelectric effect, is a straightline?

A. Intesity of radiation and photoelectric current

B. Potential of anode and photoelectric current

C. Threshold frequency and velocity of photoelectrons

D. Intensity of radiations and stopping potential

Answer:



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19. When ultraviolet light is incident on a photocell, its stopping potential is V_0 and the maximum kinetic energy of the photoelectrons is K_{\max} . When X-rays are incident on the same cell, then:

- A. V_0 and K_{\max} both increase
- B. V_0 and K_{\max} both decrease
- C. V_0 increase but K_{\max} remains the same
- D. K_{\max} increases but V_0 remains the same

Answer:



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20. The slope of the stopping potential versus frequency graph for photoelectric effect is equal to:

A. h

B. he

C. h/e

D. e

Answer: C

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21. A photon of energy $h\nu$ is absorbed by a free electron of a metal having work function $\phi < h\nu$

A. The electron is sure to come out

B. The electron is sure to come out with a kinetic energy $h\nu - \Psi$

C. Either the electron does not come out or it comes out with a kinetic energy $h\nu - \Psi$

D. It may come out with a kinetic energy less than $h\nu - \Psi$

Answer:



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22. Which one of the following statements is NOT true about photoelectric emission ?

- A. For a given emitter illuminated by light of a given frequency, the number of photo-electrons emitted per second is proportional to the intensity of incident light.
- B. For every emitter there is a definite threshold frequency below which no photoelectrons are emitted, no matter what the intensity of light is
- C. Above the threshold frequency, the maximum kinetic energy of photo electrons is proportional to the frequency of incident light

D. The saturation value of the photoelectric current is independent of the intensity of incident light

Answer:

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23. Blue light can cause photo electric emission from a metal, but yellow light cannot. If red light is incident on the metal,

- A. Photoelectric current will increase
- B. Rate of emission of photo electrons will decrease
- C. No photoelectric emission will occur
- D. Energy of the photoelectrons will increase

Answer:

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24. A proton and an electron are accelerated by the same potential difference, let λ_e and λ_p denote the de-Broglie wavelengths of the electron and the proton respectively

A. $\lambda_e = \lambda_p$

B. $\lambda_e < \lambda_p$

C. $\lambda_e > \lambda_p$

D. The relation between λ_e and λ_p depends on the accelerating potential difference.

Answer:

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25. Photoelectrons are being obtained by irradiating zinc by a radiation of λ_0 . In order to increase the kinetic energy of ejected

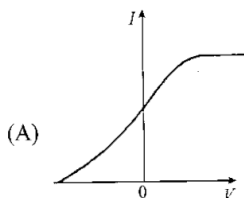
photoelectrons:

- A. The intensity of radiation should be increased
- B. The wavelength of radiation should be increased
- C. The wavelength of radiation should be decreased
- D. Both wavelength and intensity of radiation should be increased.

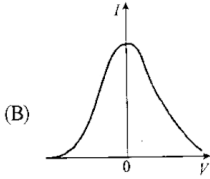
Answer:

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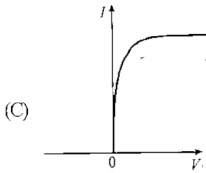
26. Which one of the following graphs in figure shows the variation of photoelectric current (I) with voltage (V) between the electrodes in a photoelectric cell?



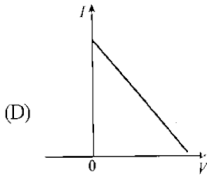
A.



B.



C.



D.

Answer:

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27. When monochromatic light falls on a photosensitive material, the number of photoelectrons emitted per second is n and their maximum kinetic energy is K_{\max} . If the intensity of the incident light is doubled keeping the frequency same, then:

- A. Both n and K_{\max} are doubled
- B. Both n and K_{\max} are halved
- C. n is doubled but K_{\max} remains the same
- D. K_{\max} is doubled but n remains the same

Answer:

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Numerical Mcqs

1. Light of wavelength 400nm is incident continuously on a Cesium ball. (work function 1.9eV). The maximum potential to which the ball will be charged is

- A. 3.1 V
- B. 1.2V

C. Zero

D. infinite

Answer:



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2. Electrons are accelerated in television tubes through potential differences of about 10 kV. The highest frequency of the electromagnetic waves emitted when these electrons strike the screen of the tube is :

A. 2.4×10^{18} Hz

B. 3.6×10^{18} Hz

C. 2.2×10^{17} Hz

D. 3.2×10^{16} Hz

Answer:



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3. 1.5 mW of 400 nm light is directed at a photoelectric cell. If 0.1% of the incident photons produce photoelectrons, find the current in the cell.

A. $0.36\mu A$

B. $0.48\mu A$

C. $0.42mA$

D. $0.32mA$

Answer:



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4. No. of identical photons incident on a perfectly black body of mass m kept at rest on smooth horizontal surface. Then the acceleration of

the body if n no. of photons incident per sec. is (Assume wavelength of photon to be λ):

A. $\frac{nh}{2\pi\lambda m}$

B. $\frac{nh}{\lambda m}$

C. $\frac{2\pi nh}{\lambda m}$

D. $\frac{\lambda m}{nh}$

Answer:



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5. No photoelectrons are emitted from a metal if the wavelength of light exceeds 6000 \AA . The work function of the metal is approximately equal to:

A. $3.315 \times 10^{-6} \text{ J}$

B. $3.315 \times 10^{-19} \text{ J}$

C. $2.07 \times 10^{19} \text{J}$

D. $2.07 \times 10^{-22} \text{J}$

Answer:

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6. If 5% of the energy supplied to a bulb is irradiated as visible light, how many quanta are emitted per second by a 100 W lamp? Assume wavelength of visible light as $5.6 \times 10^{-5} \text{ cm}$.

A. 0.75×10^{19}

B. 1.39×10^{19}

C. 2.16×10^{19}

D. 2.83×10^{19}

Answer:

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7. All electrons ejected from a surface by incident light of wavelength 200nm can be stopped before traveling 1m in the direction of a uniform electric field of $4NC^{-1}$. The work function of the surface is

- A. 4ev
- B. 6.2 eV
- C. 2eV
- D. 2.2eV

Answer:

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8. Radiation pressure on any surface is :

- A. dependent on wavelength of the light used

B. dependent on nature of surface and intensity of light used

C. dependent on frequency and nature of surface

D. depends on the nature of source from which light is coming
and on nature of surface on which it is falling

Answer:

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9. The maximum energy of the photoelectrons emitted in a photocell is 5eV . For no photoelectron to reach the anode, the potential difference of the anode with respect to the photo sensitive plate should be :

A. Zero

B. $+2\text{V}$

C. $+5\text{V}$

D. None of the above

Answer:

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10. The frequency and the intensity of a beam of light falling on the surface of a photoelectric material are increased by a factor of two.

This will

A. increase both, the maximum kinetic energy of the photo electrons, as well as photoelectric saturation current by a factor of two.

B. increase the maximum kinetic energy of the photo-electrons by a factor greater than two and would increase the photoelectric saturation current by a factor of two

C. increase the maximum kinetic energy of the photoelectrons by a factor greater than two and will have no effect on the magnitude of the photoelectric saturation current produced.

D. increase the maximum kinetic energy of the emitted photoelectrons by a factor of two but will have no effect. on the saturation photoelectric current.

Answer:

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11. Minimum light intensity that can be perceived by normal human eye is about $10^{-10} \text{ W m}^{-2}$. What is the minimum number of photons of wavelength 660 nm that must enter the pupil in one second, for one to see the object? Area of cross-section of the pupil is 10^{-4} m^2 ?

A. 3.318×10^3

B. 1.453×10^3

C. 3.38×10^4

D. 1.453×10^5

Answer:



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12. The wavelength of de-Broglie wave associated with a thermal neutron of mass m at absolute temperature T is given by (here, k is the Boltzmann constant)

A. $\frac{h}{\sqrt{mkT}}$

B. $\frac{h}{\sqrt{2mkT}}$

C. $\frac{h}{3mkT}$

D. $\frac{h}{2\sqrt{mkT}}$

Answer:

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13. What percentage increase in wavelength leads to 75% loss of photon energy in a photon-electron collision?

A. 2

B. 1

C. 0.677

D. 3

Answer:

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14. A parallel beam of light of intensity I and cross section area S is incident on a plate at normal incidence. The photoelectric emission efficiency is 100%, the frequency of beam is ν and the work function of the plate is ϕ ($h\nu > \phi$). Assuming all the electrons are ejected normal to the plane and with same maximum possible speed. The net force exerted on the plate only due to striking of photons and subsequent emission of electrons is

A. $\frac{IS}{h\nu} \left(\frac{h}{\lambda} \right) + \sqrt{m(h\nu - \phi)}$

B. $\frac{IS}{h\nu} \left(\frac{h}{\lambda} + \sqrt{2m(h\nu - \phi)} \right)$

C. $\frac{h\nu S}{I} \left(\frac{h}{\lambda} \right) + \sqrt{2m(h\nu - \phi)}$

D. None of these

Answer:

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15. A desk lamp illuminates a desktop with light of wavelength λ . The amplitude of this electromagnetic wave is E_0 . Assuming illumination to be normally on the surface, the number of photons striking the desk per second per unit area N

A. $N = \frac{\lambda \epsilon_0 E_0^2}{h}$

B. $N = \frac{2\lambda \epsilon_0 E_0^2}{h}$

C. $N = (\lambda \epsilon_0 E_0^2)$

D. Data insufficient.

Answer:

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16. If a hydrogen atom at rest, emits a photon of wavelength λ , the recoil speed of the atom of mass m is given by :

A. $\frac{h}{m\lambda}$

B. $\frac{mh}{\lambda}$

C. $mh\lambda$

D. None of these

Answer: A

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17. A point source of radiation of power P is placed on the axis of completely absorbing disc. The distance between the source and the disc is 2times the radius of the disc. The force that light exerts on the disc is $\frac{Px}{40c}$. Then the value of x

A. $\frac{P}{c}$

B. $\frac{P}{5c}$

C. $\frac{P}{10c}$

D. $\frac{P}{20}C$

Answer:

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18. A beam of light has an power of 144 W equally distributed among three wavelengths of 4100\AA , 4960\AA and 6200\AA . The beam is incident at an angle of incidence of 60° on an area of 1cm^2 of a clean sodium surface, having a work function of 2.3 eV. Assuming that there is no loss of light by reflection and that each energetically capable photon ejects a photoelectron, find the saturation photocurrent. (Take $hc/e = 12400\text{eV}\text{\AA}$)

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19. The work function of a substance is 4.0eV . The longest wavelength of light that can cause photo-electron emission form this

substance is approximately :

A. 540 nm

B. 400 nm

C. 310 nm

D. 220 nm

Answer:



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20. When a metallic surface is illuminated with monochromatic light of wavelength λ , the stopping potential is $5V_0$. When the same surface is illuminated with light of wavelength 3λ , the stopping potential is V_0 . Then the work function of the metallic surface is

A. $\frac{hc}{6\lambda}$

B. $\frac{hc}{5\lambda}$

C. $\frac{hc}{4\lambda}$

D. $\frac{2hc}{4\lambda}$

Answer:

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21. In a photoelectric experiment, with light of wavelength λ , the fastest electron has speed v . If the exciting wavelength is changed to $\frac{3\lambda}{4}$, the speed of the fastest emitted electron will become

A. $v\sqrt{\frac{3}{4}}$

B. $v\sqrt{\frac{4}{3}}$

C. less than $v\sqrt{\frac{3}{4}}$

D. greater than $v\sqrt{\frac{4}{3}}$

Answer:

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22. The work function of a certain metal is $\frac{hC}{\lambda_0}$. When a monochromatic light of wavelength $\lambda < \lambda_0$ is incident such that the plate gains a total power P . If the efficiency of photoelectric emission is $\eta\%$ and all the emitted photoelectrons are captured by a hollow conducting sphere of radius R already charged to potential V , then neglecting any interaction of potential of the sphere at time t is:

A. $V + \frac{100\eta Pet}{4\pi\epsilon_0 RhC}$

B. $V + \frac{\eta\lambda Pet}{4\pi\epsilon_0 RhC}$

C. V

D. $\frac{\lambda Pet}{4} \pi \epsilon RhC$

Answer:

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23. Radiation of frequency 1.5 times the threshold frequency is incident on a photosensitive material. If the frequency of incident radiation is halved and the intensity is doubled, the number of photoelectron ejected per second becomes:

- A. Zero
- B. half of its initial value
- C. one fourth the initial value
- D. three fourth the initial value

Answer: A

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24. The maximum kinetic energy of photoelectrons emitted from a surface when photons of energy 6eV fall on it is 4eV. The stopping potential is:

A. 2V

B. 4V

C. 6V

D. 10V

Answer:



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25. When a certain metallic surface is illuminated with monochromatic light of wavelength λ , the stopping potential for photoelectric current is $3V_0$. When the same surface is illuminated with light of wavelength 2λ the stopping potential is V_0 . The threshold wavelength for this surface for photoelectric effect is.

A. 6λ

B. 4λ

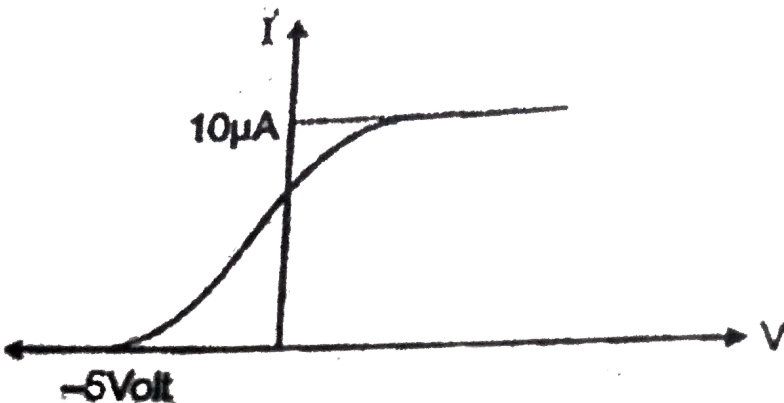
C. $\lambda/4$

D. 8λ

Answer:

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26. In the photoelectric experiment, if we use a monochromatic light, the I-V curve is as shown. If work function of the metal is 2eV, estimate the power of light used. (Assume efficiency of photo emission = $10^{-3}\%$, i.e. number of photoelectrons emitted are $10^{-3}\%$ of number of photons incident on metal.)



A. 2W

B. 5W

C. 7W

D. 10W

Answer:

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27. The eye can detect 5×10^4 photons $(m^2s)^{-1}$ of green light ($\lambda = 5000\text{\AA}$), while ear can detect $10^{-13} Wm^2$. As a power detector, which is more sensitive and by what factor?

A. Sensitivity of eye is one fifth of the ear.

B. Sensitivity of eye is five times that of ear

C. Both are equally sensitive

D. Eye cannot be used as power detector

Answer:



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28. A metal whose work function is $3.3eV$ is illuminated by light of wavelength $3 \times 10^{-7}m$. What is the threshold frequency for photoelectric emission? Plank's constant $= 6.6 \times 10^{-34} Js$?

A. $0.4 \times 10^{15} Hz$

B. $0.8 \times 10^{15} Hz$

C. $1.6 \times 10^{15} Hz$

D. $3.2 \times 10^{15} Hz$

Answer:



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29. In Q 28, the maximum energy of photoelectrons is:

A. $0.84eV$

B. $1.05eV$

C. $1.25eV$

D. $1.54eV$

Answer:



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30. In Q.28, what is the stopping potential?

A. 0.84 V

B. 1.05 V

C. 1.25 V

D. 1.54 V

Answer:

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31. A point source of radiation power P is placed on the x -axis of an ideal plane mirror. The distance between the source and the mirror is n times the radius of the mirror. Find the force that light exerts on the mirror.

A. $\frac{2P}{C(n^2 + 1)}$

B. $\frac{P(n^2 + 1)}{2C}$

C. $\frac{P}{2C(n^2 + 1)}$

D. $\frac{P}{4C(n^2 + 1)}$

Answer:

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32. A sensor is exposed for time t to a lamp of power P placed at a distance l . The sensor has an opening that is $4d$ in diameter. Assuming all energy of the lamp is given off as light, the number of photons entering the sensor if the wavelength of light is λ is:

A. $N = \frac{P\lambda d^2 t}{hcl^2}$

B. $N = \frac{4P\lambda d^2 t}{hcl^2}$

C. $N = \frac{p\lambda d^2 t}{4hcl^2}$

D. $N = \frac{P\lambda d^2 t}{16hcl^2}$

Answer:

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33. The de-Broglie wavelength of an electron moving with a velocity $1.5 \times 10^8 \text{ ms}^{-1}$ is equal to that of a photon. The ratio of the kinetic energy of the electron to that of the photon is:

A. 2

B. 4

C. $\frac{1}{2}$

D. $\frac{1}{4}$

Answer:

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34. Monochromatic light of frequency ν_1 irradiates a photocell and the stopping potential is found to be V_1 . What is the new stopping potential of the cell if it irradiated by monochromatic light of frequency ν_2 ?

A. $V_1 + \left(\frac{h}{e}\right)(\nu_2 - \nu_1)$

B. $V_1 - \frac{h}{e}(\nu_2 - \nu_1)$

C. $V_1 + \left(\frac{h}{e}\right)(\nu_1 + \nu_2)$

$$D. V_1 - \frac{h}{e}(v_1 + v_2)$$

Answer:

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35. An electromagnetic radiation of frequency 3×10^{15} cycles per second falls on a photo electric whose work function is 4.0eV. Find out the maximum velocity of the photo electrons emitted by the surface.

A. $13.4 \times 10^{-19} m/s$

B. $19.8 \times 10^{-19} m/s$

C. $1.72 \times 10^6 m/s$

D. None of these

Answer:

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36. Two identical photocathode receive light of frequencies f_1 and f_2 . If the maximum velocities of the photoelectrons (of mass m) coming out are respectively v_1 and v_2 then:

A. $v_1^2 - v_2^2 = \frac{2h}{m}(f_1 - f_2)$

B. $v_1 + v_2 = \left[\frac{2h}{m}(f_1 + f_2) \right]^{\frac{1}{2}}$

C. $v_1^2 + v_2^2 = \frac{2h}{m}(f_1 + f_2)$

D. $v_1 - v_2 = \left[\frac{2h}{m}(f_1 - f_2) \right]^{\frac{1}{2}}$

Answer:

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37. A metal plate is exposed to light with wavelength λ . It is observed that electrons are ejected from the surface of the metal plate. When a retarding uniform electric field E is imposed, no electron can move

away from the plate farther than a certain distance d . Then the threshold wavelength λ_0 for the material of plate is (e is the electronic charge, h is Planck's constant and c is the speed of light)

$$\text{A. } \lambda_0 = \left(\left(\frac{1}{\lambda} \right) - \frac{hc}{eEd} \right)^{-1}$$

$$\text{B. } \lambda_0 = \left(\frac{1}{\lambda} - \frac{eEd}{hc} \right)^{-1}$$

$$\text{C. } \lambda_0 = \lambda - \frac{hc}{eEd}$$

$$\text{D. } \lambda_0 = \lambda - \frac{eEd}{hc}$$

Answer:

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38. The energy of an α -particle, whose de-Broglie wavelength is 0.004\AA , will be:

A. 1297eV

B. 1245 KeV

C. 1205 MeV

D. 1288 GeV

Answer:



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39. A small potassium foil is placed (perpendicular to the direction of incidence of light) a distance r ($= 0.5m$) from a point light source whose output power P_0 is $1.0W$. Assuming wave nature of light how long would it take for the foil to soak up enough energy ($= 1.8eV$) from the beam to eject an electron? Assume that the ejected photoelectron collected its energy from a circular area of the foil whose radius equals the radius of a potassium atom ($1.3 \times 10^{-10}m$)

A. 11s

B. 14s

C. 17s

Answer:
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40. A photon and an electron possess same de-Broglie wavelength. Given that c = speed of light and v = speed of electron, which of the following relations is correct? Here $E_e = KE$ of electron, $E_{ph} = KE$ of photon, $p_e = \text{momentum of electron}$, $p_{ph} = \text{momentum of photon}$:

A. $\frac{E_e}{E_{ph}} = \frac{3c}{v}$

B. $\frac{E_e}{E_{ph}} = \frac{v}{2c}$

C. $\left(\frac{p_e}{p_{ph}} = (2c)v \right)$

D. $\left(\frac{p_e}{p_{ph}} = \frac{c}{2v} \right)$

Answer:



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41. The maximum velocity of photoelectrons emitted by a photoemitter is $1.8 \times 10^6 \text{ m/sec}$. Taking $e/m = 1.8 \times 10^{11} \frac{C}{k} g$ for electrons, the stopping potential of the emitter is:

A. $1.82V$

B. $9.21V$

C. $11.82V$

D. $23.64V$

Answer:



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42. A proton with KE equal to that of a photon ($E = 100 \text{ keV}$). λ_1 is the wavelength of proton and λ_2 is the wavelength of photon. Then

$\frac{\lambda_1}{\lambda} \Big)_2$ is proportional to:

A. $E^{\frac{1}{2}}$

B. $E^{-\frac{1}{2}}$

C. E

D. E^{-1}

Answer:



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43. How much potential is to be applied to accelerate an electron, so that its de-broglie wavelength should 0.4\AA ?

A. 9434 V

B. 94.34V

C. 9.434 V

D. 943.4 V

Answer:

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44. A metallic surface is irradiated with monochromatic light of variable wavelength. Above a wavelength of 5000\AA no photoelectrons are emitted from the surface. With an unknown wavelength, stopping potential of 3V is necessary to eliminate the photo current. Find the unknown wavelength.

A. 2268 A

B. 1116 A

C. 3426 A

D. 4801 A

Answer:



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45. Illuminating the surface of a certain metal alternately with light of wavelengths $\lambda_1 = 0.35\mu\text{ m}$ and $\lambda_2 = 0.54\mu\text{ m}$, it was found that the corresponding maximum velocities of photo electrons have a ratio $\eta = 2$. Find the work function of that metal:

A. 3.22 eV

B. 1.88 eV

C. 5.64 eV

D. 6.28 eV

Answer:



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46. The de-Broglie wavelength associated with a material particle when it is accelerated through a potential difference of 150 volt is 1 \AA . The de-Broglie wavelength associated with the same particle when it is accelerated through a potential difference of 1350 V will be:

A. $\frac{1}{4}\text{ \AA}$

B. $\frac{1}{3}\text{ \AA}$

C. 1 \AA

D. 0

Answer:



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47. An electron of mass m , when accelerated through a potential difference V , has de-Broglie wavelength λ . The de-Broglie wavelength

associated with a proton of mass M accelerated through the same potential difference, will be:

A. $\lambda \frac{M}{m}$

B. $\lambda \frac{m}{M}$

C. $\lambda \sqrt{\frac{M}{m}}$

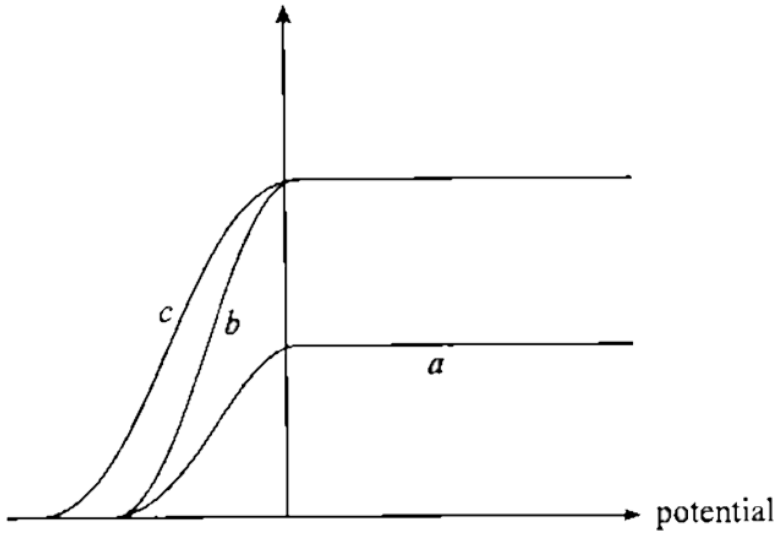
D. $\lambda \sqrt{\frac{m}{M}}$

Answer:

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48. In photoelectric experiment the plot between anode potential and photoelectric current is shown in figure 2.3.1. Which of the

following is correct:



- A. Frequency of light corresponding to "a" is same as that of "b" and is different that corresponding to "c".
- B. Frequency of light corresponding to "a" is different from "b" & intensities are the same.
- C. Frequency corresponding to "b" is same as that of "c", but intensities are differeent.
- D. Frequency corresponding to "b" is different from that of "c", and intensities are different.

Answer: A



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49. If the momentum of electron is changed by P_m then the de-Broglie wavelength associated with it changes by 0.50%. The initial momentum of electron will be:

A. $\frac{P_m}{200}$

B. $\frac{P_m}{100}$

C. $200P_m$

D. $400P - (m)$

Answer:



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50. Light described at a place by the equation $E = (100V(m^{-1})) [\sin(5 \times (10^{15})m(s^{-1}))t + \sin(8 \times (10^{15})(s^{-1})t)]$ falls on a metal surface having work function $2.0eV$. Calculate the maximum kinetic energy of the photoelectrons.

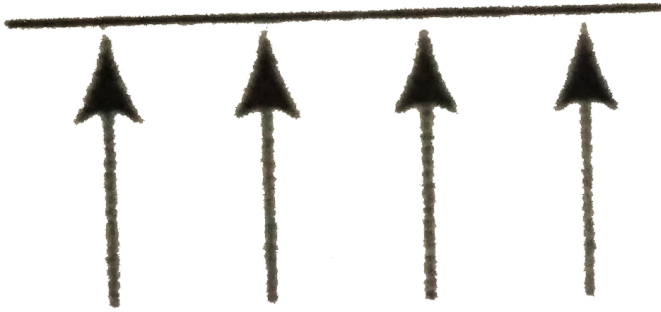
- A. 3.27 eV
- B. 4.54 eV
- C. 5.86 eV
- D. 6.54 eV

Answer:

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51. A plate of mass 10g is in equilibrium in air due to the force exerted by a light beam on the plate. Calculate power of the beam. Assume

that the plate is perfectly absorbing.



A. $1.5 \times 10^7 W$

B. $3 \times 10^7 W$

C. $4.5 \times 10^7 W$

D. $6 \times 10^7 W$

Answer:

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52. The energy that should be added to an electron, to reduce its de-Broglie wavelength from $2 \times 10^{-9} m$ to $0.5 \times 10^{-9} m$ will be:

A. 1.1MeV

B. 0.56MeV

C. 0.56KeV

D. 5.67eV

Answer:

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53. An electron of mass 'm', when accelerated through a potential V has de-Broglie wavelength λ . The de-Broglie wavelength associated with a proton of mass M accelerated through the same potential difference will be:

A. 1.1 MeV

B. 0.56 MeV

C. 0.56 KeV

D. 5.67 eV

Answer:

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54. An electron of mass 'm', when accelerated through a potential V has de-Broglie wavelength λ . The de-Broglie wavelength associated with a proton of mass M accelerated through the same potential difference will be:

A. $\lambda \left(\sqrt{\frac{M}{m}} \right)$

B. $\lambda \sqrt{\frac{m}{M}}$

C. $\lambda \left(\frac{M}{m} \right)$

D. $\lambda \left(\frac{m}{M} \right)$

Answer:

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55. Find the numerical value fo De Broglie wavelength of an electron in the 1st orbit of hydrogen atom assuming Bohr's atomic model. You can use statndard values of the constants. Leave your answer in terms of π .

A. 1.058π

B. 2.223π

C. 2.116π

D. none of these

Answer:



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56. An electron filled with helium is heated to a temperature of 400k. A beam of helium atom emerges out of the enclosure. The mean debroglie wavelength of this beam is:

A. 0.44\AA

B. 2.23π

C. 2.116π

D. none of these

Answer:



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57. If the short wavelength limit of the continous spectrum coming out of a Coolidge tube is 10\AA , then the de Broglie wavelength of the electrons reaching the target metal in the Coolidge tube is approximately

A. 0.3\AA

B. 3\AA

C. 30\AA

D. 10\AA

Answer:

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58. A particle of charge q_0 and of mass m_0 is projected along the y -axis at $t = 0$ from origin with a velocity V_0 . If a uniform electric field E_0 also exist along the x -axis, then the time at which debroglie wavelength of the particle becomes half of the initial value is:

A. $\frac{m_0 v_0}{q_0 E_0}$

B. $2 \frac{m - (0)v_0}{q_0 E_0}$

C. $\sqrt{3} \frac{m_0 v_0}{q_0 E_0}$

$$D. 3 \frac{m_0 v_0}{q_0 E_0}$$

Answer:

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59. Electrons in a sample of gas containing hydrogen-like atom ($Z = 3$) are in fourth excited state. When photons emitted only due to transition from third excited state to second excited state are incident on a metal plate photoelectrons are ejected. The stopping potential for these photoelectrons is 3.95eV . Now, if only photons emitted due to transition from fourth excited state to third excited state are incident on the same metal plate, the stopping potential for the emitted photoelectrons will be approximately equal to

A. 0.85 eV

B. 0.75 eV

C. 0.65 eV

D. None of these

Answer:

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60. Threshold frequency for a certain metal is ν_0 . When light of frequency $2\nu_0$ is incident on it, the maximum velocity of photoelectrons is $4 \times 10^8 \text{ cm s}^{-1}$. If frequency of incident radiation is increased to $5\nu_0$, then the maximum velocity of photoelectrons, in cm s^{-1} , will be

A. $4/5 \times 10^6$

B. 2×10^6

C. 8×10^6

D. 2×10^7

Answer:



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Advance Mcq

1. A collimated beam of light of flux density $30kWm^{-2}$ is incident normally on $100mm^2$ completely absorbing screen. If P is the pressure exerted on the screen and Δp is the momentum transferred to the screen during a 1000 s interval, then

A. $P = 10^{-3}Nm^{-2}$

B. $P = 10^{-4}Nm^{-2}$

C. $\Delta p = 10^{-4}kgms^{-1}$

D. $\Delta p = 10^{-5}kgms^{-1}$

Answer:



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2. In a photoelectric experiment, the collector plate is at 2.0V with respect to the emitter plate made of copper ($\phi = 4.5\text{eV}$). The emitter is illuminated by a source of monochromatic light of wavelength 2000 Ånm.

- A. the maximum kinetic energy of the photo electrons at collector is 1.7eV
- B. the maximum kinetic energy of the photo electrons on the collector is 1.7eV.
- C. if the polarity of the battery is reversed then the minimum kinetic energy of the photo electrons on the collector is 0.
- D. if the polarity of the reversed then the maximum kinetic energy of the photo electrons on the collector is 3.7 eV

Answer:

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3. The work function for aluminium surface is 4.2 eV and that for sodium surface is 2.0 eV. The two metals were illuminated with appropriate radiations so as to cause photo emission. Then :

- A. Both aluminium and sodium will have the same threshold frequency
- B. The threshold frequency of aluminium will be more than that of sodium
- C. the threshold frequency of aluminium will be less than that of sodium
- D. the threshold wavelength of aluminium will be more than that of sodium

Answer:

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4. The stopping potential for the photoelectrons emitted from a metal surface of work function 1.7 eV is 10.4 V. Find the wavelength of the radiation used. Also, identify the energy levels in hydrogen atom, which will emit this wavelength.

A. The wavelength of light used is 1022\AA .

B. The wavelength of light used is 970.6\AA .

C. The light used is emitted by hydrogen gas sample which de-excites from $n=3$ to $n=1$

D. The light used is emitted by hydrogen gas sample which de-excites from $n=4$ to $n=1$

Answer:



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5. H^+ , He^+ and O^{++} all having the same kinetic energy pass through a region in which there is a uniform magnetic field perpendicular to their velocity. The masses of H^+ , He^+ and O^{2+} are $1a\mu$, $4a\mu$ and $16a\mu$ respectively. Then

- A. H^+ will be deflected the most
- B. O^{++} will be deflected the most
- C. He^+ and O^{++} will be deflected equally
- D. All will be deflected equally

Answer:

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6. When an electron of hydrogen like atom jumps from a higher energy level to a lower energy level.

- A. Angular momentum of the electron remains constant
- B. Kinetic energy increases
- C. Wavelength of de-Broglie wave, associated with motion of the electron, decreases
- D. none of these

Answer:



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7. Light rays are incident on an opaque sheet. Then they:

- A. Exert a force energy to the sheet
- B. transfer an energy to the sheet
- C. transfer momentum to the sheet
- D. transfer impulse to the sheet

Answer:



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8. The angular momentum of an electron in an orbit is quantized because:

- A. Bohr's theory is based on this postulate
- B. It is a necessary condition for compatibility with the wave behavior of the electron
- C. It is a necessary condition for compatibility with the particle behavior of the electron
- D. It is a necessary condition for compatibility with Pauli's exclusive principle

Answer:



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9. The intensity of light falling on a phototube is increased while keeping the no of photons falling per seconds constant then for a given experiment.

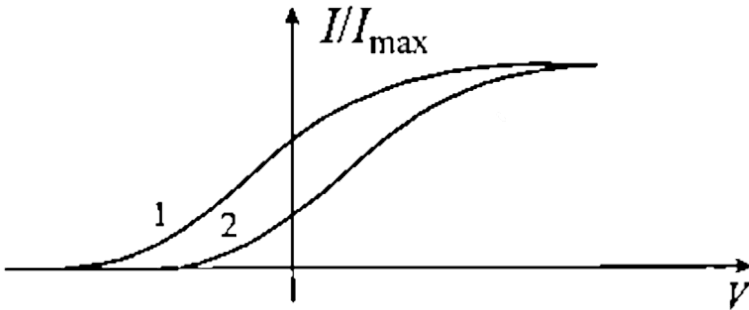
- A. Stopping potential will increase
- B. Saturation current will increase
- C. saturation current will decrease
- D. saturation current will remain unaffected.

Answer:

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10. Two photocathodes are illuminated by the light emitted by a single source. The dependence of photocurrent versus voltages between cathode and anode is shown by curves 1 and 2 as shown in

the figure. (I/I_{max} represents ratio of photocurrent to saturation current):



- A. Photocathode 1 has higher work function than 2
- B. Photocathode 2 has higher work function than 1
- C. Saturation current may be different for 1 and 2
- D. saturation current must be same for 1 and 2

Answer:



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11. Photoelectric effect supports quantum nature of light because

(a) there is a minimum frequency of light below which no photo electrons are emitted

(b) the maximum kinetic energy of photo electrons depends only on the frequency of light and not on its intensity

(c) even when the metal surface is faintly illuminated, the photo electrons leave the surface immediately

(d) electric charge of the photo electrons is quantised

A. There is a minimum frequency below which no photoelectrons are emitted

B. The maximum kinetic energy of photoelectrons depends only on the frequency of light and not on its intensity

C. Even when the metal surface is faintly illuminated the photoelectrons (if $\nu > \nu_{th}$) leave the surface immediately

D. electric charge of the photoelectrons is quantized

Answer:

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12. The momentum of a single photon of red light of frequency 400×10^{12} Hz moving through free surface is given as:

A. Zero

B. $8.8 \times 10^{-28} \text{ kgms}^{-2}$

C. $1.65 \times 10^{-6} \text{ Me} \frac{\text{V}}{\text{c}}$

D. Data insufficient.

Answer:

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13. When a monochromatic point source of light is at a distance of 0.2 m from a photoelectric cell, the cut off voltage and the saturation current are respectively 0.6 V and 18.0 mA. If the same source is placed 0.6 m away from the photoelectric cell, then

- (a) the stopping potential will be 0.2 V
- (b) the stopping potential will be 0.6 V
- (c) the saturation current will be 6.0 mA
- (d) the saturation current will be 2.0 mA

- A. the stopping potential will be 0.2V
- B. the stopping potential will be 0.6V
- C. the saturation current will be 6.0 mA
- D. the saturation current will be 2.0mA

Answer:

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14. In a photoelectric experiment, the collector plate is at 2.0 V with respect to the emitter plate made of copper. The emitter is illuminated by a source of monochromatic light of wavelength 200nm. Find the minimum and maximum kinetic energy of the photoelectrons reaching the collector.

A. The minimum kinetic energy of the photoelectrons reaching the collector is 0.

B. The maximum kinetic energy of the photoelectrons reaching the collector is 3.7 eV

C. If the polarity of the battery is reversed then answer to part A will be 0.

D. If the polarity of the battery is reversed then answer to part B will be 1.7 eV

Answer:

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15. A small mirror of area A and mass m is suspended in a vertical plane by a weightless string. A beam of light of intensity I falls normally on the mirror and the string is deflected from the vertical by a very small angle θ . Assuming the mirror to be perfectly reflecting, obtain an expression for θ .

A. Radiation pressure equal to $2I\frac{A}{c}$

B. Radiation pressure equal to $(I)/(2c)$

C. $\tan \theta = \frac{2IA}{mgc}$

D. $\tan \theta = \frac{IA}{2mgc}$

Answer:

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16. When photons of energy 4.25eV strike the surface of metal A, the ejected photoelectrons have maximum kinetic energy T_A eV and De-broglie wavelength λ_A . The maximum energy of photoelectron liberated from another metal B by photon of energy 4.70 eV is $T_B = (T_A - 1.50)\text{eV}$ if the de Broglie wavelength of these photoelectrons is $\lambda_B = 2\lambda_A$, then

A. the work function of A is 2.25 eV

B. the work function of B is 4.20 eV

C. $T_A = 2.00\text{eV}$

D. $T_B = 2.75\text{eV}$

Answer:

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17. In photoelectric effect, stopping potential depends on

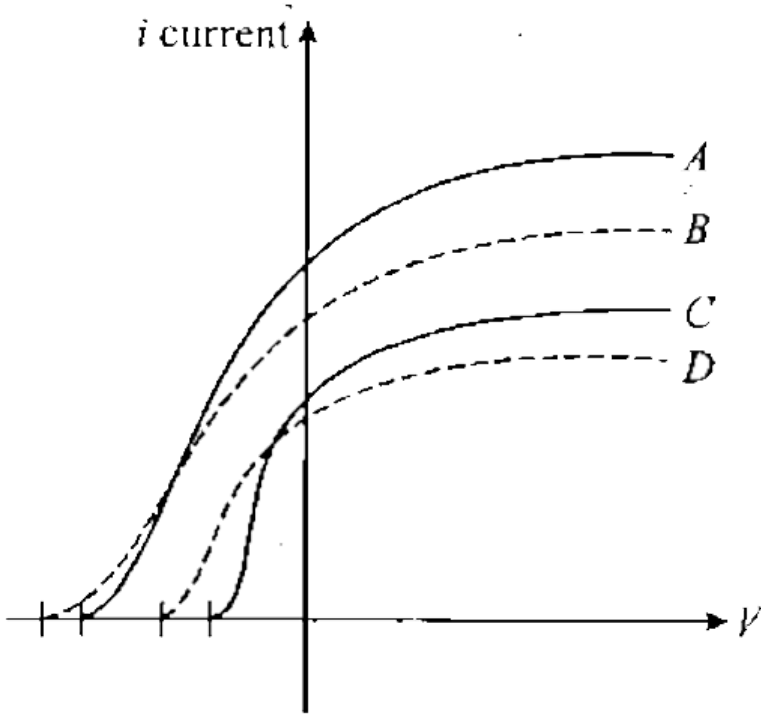
- A. frequency of the incident light
- B. intensity of the incident light by varies source distance
- C. emitter's properties
- D. frequency and intensity of the incident light

Answer:

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18. The figure shows the results of an experiment involving photoelectric effect. The graphs A, B, C and D relate to a light beam

having different wavelengths. Select the correct alternative.



- A. Beam B has highest frequency
- B. Beam C has longest wavelength
- C. Beam A has highest rate of photoelectric emission
- D. Photoelectrons emitted by B have highest momentum.

Answer:

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19. the photoelectric effect can not be explained by the wave theory of light because:

A. the energy carried by the light waves is not given to a particular present on the surface of metal.

B. waves do not have energy

C. energy of the waves becomes zero as it strikes the metal surface.

D. waves do not have sufficient energy which is required for electron emission.

Answer:



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20. A non-monochromatic light is used in an experiment on photoelectric effect. The stopping potential

- A. is related to the mean wavelength
- B. is related to the longest wavelength
- C. is related to the shortest wavelength
- D. is not related to the wavelength

Answer:



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21. In a photoelectron effect energy of photon is directly proportional to the frequency and photons are totally absorbed by the electrons of metals then frequency and photons are totally absorbed by the electrons of metals then photoelectric current is:

- A. increased when the frequency of photon is increased.
- B. decreased when the frequency of photon is increased
- C. independent of the frequency of photon but it only depends on the intensity of incident photons
- D. independent of the intensity of incident photons.

Answer:



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22. Suppose frequency of emitted photon is f_0 when the electron of a stationary hydrogen atom jumps from a higher state m to a lower state n . If the atom is moving with a velocity v ($v < c$) and emits a photon of frequency f during the same transition, then which of the following statements are possible?

A. f may be equal to f_0

B. f may be greater than f_0

C. f may be less than f_0

D. f cannot be equal to f_0

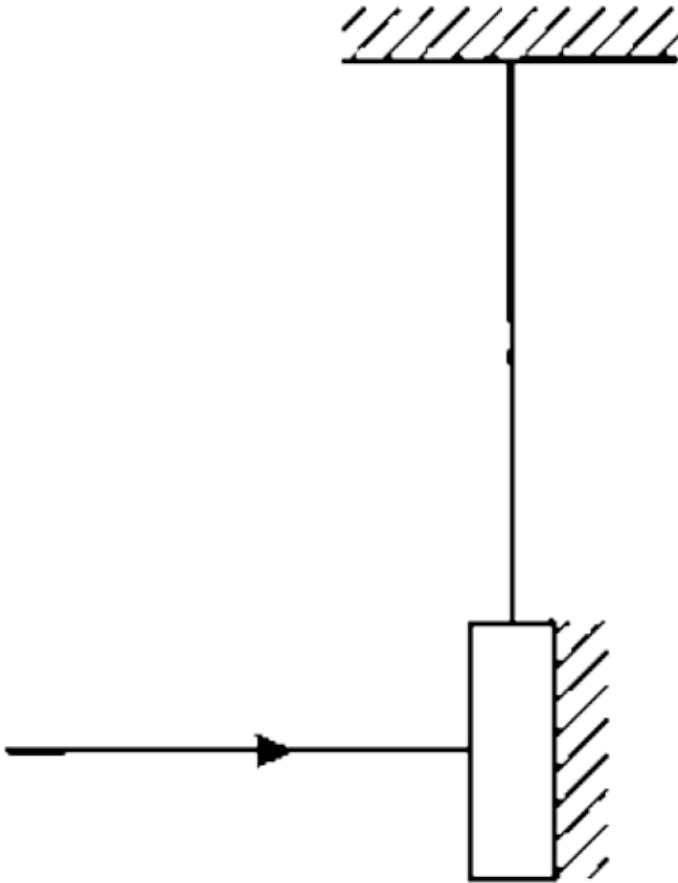
Answer:



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23. A small mirror is suspended by a thread as shown in figure 2.35. A short pulse of mono-chromatic light rays is incident normally on the mirror and gets reflected. Which of the following statements is/are

correct?



A. mirror will start to oscillate

B. wavelengths of reflected rays will be greater than that of incident rays

- C. wavelength of reflected rays may be less than that of incident rays.
- D. none of these

Answer:

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24. When monochromatic light falls on a photosensitive material, the number of photoelectrons emitted per second is n and their maximum kinetic energy is K_{\max} . If the intensity of the incident light is doubled keeping the frequency same, then:

- A. Both n and K_{\max} are doubled
- B. Both n and K_{\max} are halved
- C. n is doubled but K_{\max} remains the same
- D. K_{\max} is doubled but n remains the same

Answer:



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25. Mark the correct statement(s) related to the stopping potential difference.

- A. At the stopping potential, emitter plate is positive with respect to the collector
- B. At the stopping potential, emitter plate is negative with respect to the collector.
- C. At the stopping potential, electrons does not come out of the plates.
- D. At the stopping potential, electrons are able to reach the collector but return before being absorbed.

Answer:



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26. If the wavelength of light in an experiment on photo electric effect is doubled,

- A. The photoelectrons emission will not take place.
- B. the photoemission may or may not take place
- C. The stopping potential will increases
- D. the stopping potential will decrease under the condition that energy of photon of doubled. Wavelength is more than work function of metal.

Answer:



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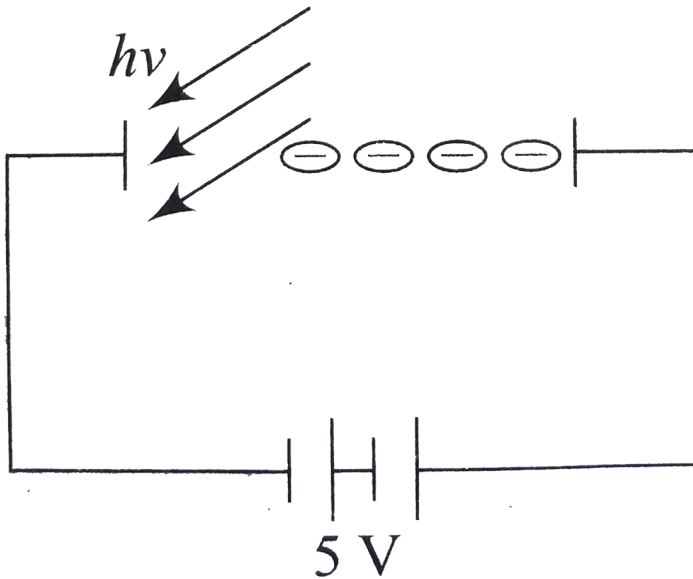
27. Photoelectric effect supports quantum nature of light because

- A. There is minimum frequency of light below which no photoelectrons are emitted
- B. The maximum kinetic energy of photoelectrons depends only on the frequency of light and not on its intensity
- C. Even when the metal surface is faintly illuminated the photoelectrons (if $\nu > \nu_{th}$) leave the surface immediately
- D. electric charge of the photoelectrons is quantized

Answer:



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28.

Photons of energy 5eV are incident on the cathode. Electrons reaching the anode have kinetic energies varying from 6eV to 8eV . Find the work function of the metal and state whether the current in the circuit is less than or equal to saturation current.

- A. Work function of the metal is 2eV
- B. work function of the metal is 3eV
- C. Current in the circuit is equal to saturation value.
- D. Current in the circuit is less than saturation value.

Answer:



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Unsolved Numerical Problems

1. Light from balmer series of hydrogen is able to eject photoelectron from a metal. What can be the maximum work function of the metal?



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2. A photocell is operating in saturation mode with a photocurrent 4.8mA when a monochromatic radiation of wavelength 3000\AA and power of 1mW is incident. When another monochromatic radiation of wavelength 1650\AA and power 5mW is incident, it is observed that maximum velocity of photoelectron increases to two times. Assuming efficiency of photoelectron generation per incident photon to be

same for both the cases, calculate.

(a) the threshold wavelength for the cell

(b) the saturation current in second case

(c) the efficiency of photoelectron generation per incident photon.



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3. Photo electrons are liberated by ultraviolet light of wavelength 3000\AA from a metallic surface for which the photoelectric threshold wavelength is 4000\AA . Calculate the de Broglie wavelength of electrons emitted with maximum kinetic energy.



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4. Lithium has a work function of 2.3eV . It is exposed to light of wavelength 4800\AA . Find the maximum kinetic energy with which electron leaves the surface. What is the longest wavelength which can produce the photoelectrons?



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5. A parallel beam of monochromatic light of wavelength 663 nm is incident on a totally reflecting plane mirror. The angle of incidence is 60° and the number of photons striking the mirror per second is (1.0×10^{19}) . Calculate the force exerted by the light beam on the mirror.



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6. Ultraviolet light of wavelength 800\AA and 700\AA when allowed to fall on hydrogen atoms in their ground states is found to liberate electrons with kinetic energies 1.8eV and 4.0eV , respectively. Find the value of Planck's constant.



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7. The work function of aluminium is 4.2 eV. Calculate the Kinetic energy of the fastest & the slowest photoelectrons, the stopping potential & the cut off wavelength when light of wavelength 2000\AA falls on a clean Aluminium surface.

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8. What potential difference must be applied to stop the fastest photoelectrons emitted by a nickel surface under the action of ultraviolet light of wavelength 2000\AA ? The work function of nickel is 5.01 eV.

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9. A sphere of radius 1.00 cm is placed in the path of a parallel beam of light of large aperture. The intensity of the light is 0.50 W cm^{-2} . If the sphere completely absorbs the radiation falling on it, find the force exerted by the light beam on the sphere.



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10. Find the frequency of light which ejects electrons from a metal surface. Fully stopped by a retarding potential of $3V$, the photoelectric effect begins in this metal at a frequency of $6 \times 10^{14} Hz$. Find the work function for this metal. (Given $h = 6.63 \times 10^{-34} Js$).



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11. state whether the following statements are TRUE or FALSE, giving reasons in brief to support your answer. In a photo electric emission process the maximum energy of the photo-electron increases with increasing intensity of the incident light.



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12. A point isotropic light source of power $P = 12$ watt is located on the axis of a circular mirror plate of radius $R = 3$ cm. If distance of source from the plate is $\alpha = 39$ cm and reflection coefficient of mirror plate is $\alpha = 0.70$, calculate force exerted by light rays on the plate.

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13. A uniform monochromatic beam of light of wavelength 365×10^{-9} m and intensity 10^{-8} W m^{-2} falls on a surface having absorption coefficient 0.8 and work function 1.6 eV. Determine the rate of number of electrons emitted per m^2 , power absorbed per m^2 and the maximum kinetic energy of emitted photoelectrons.

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14. Light of wavelength 180 nm ejects photoelectrons from a plate of metal whose work - function is 2 eV. If a uniform magnetic field of 5×10^{-5} T be applied parallel to the plate, what would be the radius

of the path followed by electrons ejected normally from the plate with maximum energy.

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15. Suppose the wavelength of the incident light is increased from 3000\AA to 3040\AA . Find the corresponding change in the stopping potential. [Take the product $hc = 12.4 \times 10^{-7} eVm$]

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16. When a metal plate is exposed to a monochromatic beam of light of wavelength 400nm , a negative potential of 1.1 V is needed to stop the photocurrent . Find the threshold wavelength for the metal.

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17. The energy needed to detach the electron of a hydrogen like ion in ground state is a system (a) what is the wavelength of the radiation emitted when the electron jumps from the first excited state to the ground state? (b) What is the radius of the orbit for this atom?

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18. A stationary hydrogen atom emits a photon corresponding to first line of the Lyman series. What velocity does the atom acquire?

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19. In an experiment tungsten cathode which has a threshold wavelength 2300 \AA is irradiated by ultraviolet light of wavelength 1800 \AA . Calculate

(a) maximum energy of emitted photoelectrons &

(b) work function of the tungsten



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20. Electrons with maximum kinetic energy 3eV are ejected from a metal surface by ultraviolet radiation of wavelength 1500\AA . Determine the work function of the metal, the threshold wavelength of metal and the stopping potential difference required to stop the emission of electrons.



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21. A small plate of a metal (work function $= 1.7\text{eV}$) is placed at a distance of 2m from a monochromatic light source of wavelength $4.8 \times 10^{-7}\text{m}$ and power 1.0watt . The light falls normally on the plate. Find the number of photons striking the metal plate per square meter per second. If a constant magnetic field of strength 10^{-4}T is applied parallel to the metal surface, find the radius of the largest circular path followed by the emitted photoelectrons. (use

$h = 6.63 \times 10^{-34} Js$, mass of electron= $9.1 \times 10^{-31} kg$, charge of electron= $1.6 \times 10^{-19} C$

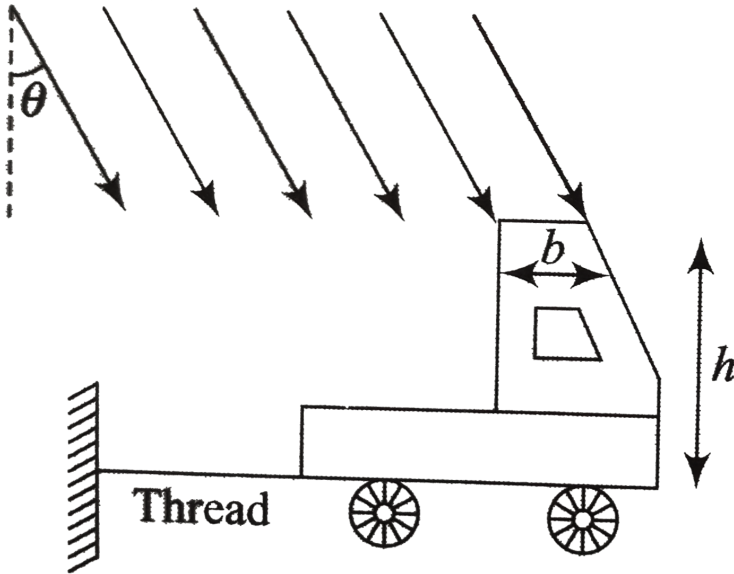
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22. The electric field associated with a monochromatic beam becomes zero 1×10^{15} times per second. Find the maximum kinetic energy of the photoelectron when this light falls on a metal surface whose work function is 2 eV .

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23. Find the frequency of light which ejects electrons from a metal surface. Fully stopped by a retarding potential of $3V$, the photoelectric effect begins in this metal at a frequency of $6 \times 10^{14} Hz$. Find the work function for this metal. (Given $h = 6.63 \times 10^{-34} Js$).

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24.

A toy truck has dimensions as shown in fig. and its width, normal to the plane of this paper, is d . The sun rays are incident on it as shown in the figure. If intensity of rays is I and all surfaces of truck are perfectly black, calculate tension in the thread used to keep the truck stationary. Neglect friction.

25. Light of wavelength 180 nm ejects photoelectrons from a plate of metal whose work - function is 2 eV. If a uniform magnetic field of 5×10^{-5} T be applied parallel to the plate, what would be the radius of the path followed by electrons ejected normally from the plate with maximum energy.

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26. Electron in a hydrogen-like atom ($Z = 3$) make transition from the forth excited state to the third excited state and from the third excited state to the second excited state. The resulting radiations are incident potential for photoelectrons ejected by shorter wavelength is $3.95eV$.

Calculate the work function of the metal and stopping potential for the photoelectrons ejected by the longer wavelength.

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27. An electron in a hydrogen like atom is in an excited state. It has a total energy of -3.4eV . Calculate:

(i) The kinetic energy,

(ii) The de-Broglie wavelength of the electron.

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28. In an experiment on photo electric emission, following observations were made,

(i) Wavelength of the incident light $= 1.98 \times 10^{-7}\text{m}$,

(ii) Stopping potential $= 2.5\text{volt}$.

Find: (a) Kinetic energy of photoelectrons with maximum speed.

(b) Work function and

(c) Threshold frequency,

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29. Light from a discharge tube containing hydrogen atoms falls on the surface of a piece of sodium. The kinetic energy of the fastest photoelectrons emitted from sodium is 0.73 eV. The work function for sodium is 1.82 eV. Find (a) the energy of the photons causing the photoelectrons emission.

(b) the quantum numbers of the two levels involved in the emission of these photons.

(c) the change in the angular momentum of the electron in the hydrogen atom, in the above transition, and

(d) the recoil speed of the emitting atom assuming it to be at rest before the transition. (Ionization potential of hydrogen is 13.6 eV.)

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30. In a photo electric effects set-up, a point source of light of power 3.2×10^{-3} W emits mono energetic photons of energy 5.0 eV. The source is located at a distance of 0.8m from the centre of a stationary

metallic sphere of work function 3.0 eV & of radius $8.0 \times 10^6 \text{ m}$. The efficiency of photo electric emission is one of every 10^6 incident photons. Assume that the sphere is isolated and initially neutral, and that photo electrons are instantly swept away after emission.

(a) Calculate the number of photo electrons emitted, per second.

(b) Find the ratio of the wavelength of incident light to the de-Broglie wave length of the fastest photo electrons emitted. It is observed that the photo electron emission stops at a certain time after the light source is switched on. Why?

(d) Evaluate the time t .

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31. State whether the following statements are true or false. Give brief reasons in support of your answers.

(i) The dimension of (h/e) is the same as that of magnetic flux ϕ . Here h & e represents Planck's constant and electronic charge respectively.

(ii) Two metal plates of same surface area and work function are

irradiated by a beam of light, incident normally. It is found that the photoelectric current from the two metals are different.

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32. A 40 W ultraviolet light source of wavelength 2480 \AA . Illuminates a magnesium (Mg) surface placed 2 m away. Determine the number of photons emitted from the surface per second and the number incident on unit area of Mg surface per second. The photoelectric work function for Mg is 3.68 eV . Calculate the kinetic energy of the fastest electrons ejected from the surface. Determine the maximum wavelength for which the photoelectric effects can be observed with Mg surface.

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33. Photoelectrons are emitted when 400 nm radiation is incident on a surface of work - function 1.9 eV . These photoelectrons pass

through a region containing α -particles. A maximum energy electron combines with an α -particle to form a He^+ ion, emitting a single photon in this process. He^+ ions thus formed are in their fourth excited state. Find the energies in eV of the photons lying in the 2 to 4 eV range, that are likely to be emitted during and after the combination. [Take, $h = 4.14 \times 10^{-15} eV - s$]

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34. At a given instant there are 25 % undecayed radioactive nuclei in a sample. After 10s the number of undecayed nuclei reduces to 12.5 % . Calculate

(a) mean life of the nuclei,

(b) the time in which the number of undecayed nuclei will further reduce to 6.25 % of the reduced number.

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35. When a beam of 10.6eV photons of intensity $2.0\text{W}/\text{m}^2$ falls on a platinum surface of area $1.0 \times 10^4\text{m}^2$ and work function 5.6eV , 0.53% of the incident photons eject photoelectrons. find the number of photoelectrons emitted per second and their minimum and maximum energies (in eV) Take $1\text{eV} = 1.6 \times 10^{-19}\text{J}$

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36. In a photoelectric experiment set up, photons of energy 5eV falls on the cathode having work function 3eV (a) if the saturation current is $i = 4\mu\text{A}$ for intensity $10^{-5}\text{W}/\text{m}^2$, then plot a graph between anode potential and current (b) Also draw a graph for intensity of incident radiation $2 \times 10^{-5}\text{W}/\text{m}^2$

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37. The photons from the Balmer series in Hydrogen spectrum having wavelength between 450nm to 700nm are incident on a metal

surface of work function $2eV$ Find the maximum kinetic energy of ejected electron (Given $hc = 1242 \text{ eV nm}$)`

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38. Monochromatic radiation of wave length 640.2 nm ($1\text{nm} = 10^{-9}\text{m}$) from a neon lamp irradiates a photosensitive material made of calcium or tungsten. The stopping voltage is measured to be 0.54V . The source is replaced by an iron source and its 427.2 nm line irradiates the same photocell. Predict the new stopping voltage.

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39. Irradiating the metal surface successive by radiations of wavelength 3000\AA and 5400\AA , it is found that the maximum velocities of electrons are in the ratio $2:1$. Find the work function of the metal surface, meV .



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40. A filter transmits only the radiation of wavelength greater than 440nm . Radiation from a hydrogen discharge tube goes through such a filter and is incident on a metal of work function 2.0eV . Find the stopping potential which can stop the photoelectrons.



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41. The peak emission from a black body at a certain temperature occurs at a wavelength of 9000\AA . On increase its temperature, the total radiation emitted is increased its 81 times. At the initial temperature when the peak radiation from the black body is incident on a metal surface, it does not cause any photoemission from the surface. After the increase of temperature, the peak from the black body caused photoemission. To bring these photoelectrons to rest, a potential equivalent to the excitation energy between

$n = 2$ and $n = 3$ bohr levels of hydrogen atoms is required. Find the work function of the metal.

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42. The radiation emitted when an electron jumps from $n = 3 \rightarrow n = 2$ orbit in a hydrogen atom falls on a metal to produce photoelectron. The electron from the metal surface with maximum kinetic energy are made to move perpendicular to a magnetic field of $(1/320)T$ in a radius of $10^{-3}m$. Find (a) the kinetic energy of the electrons, (b) Work function of the metal , and (c) wavelength of radiation.

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