

India's Number 1 Education App

### **CHEMISTRY**

## **BOOKS - BITSAT GUIDE**

## NITROGEN CONTAINING COMPOUNDS

**Practice Exercise** 

1. Gabriel phthalimide synthesis can be used

for the preparation of amine from

#### A. $CH_3CH_2Br$

### $\mathsf{B.}\left(CH_3\right)_3 CBr$

### $\mathsf{C.}\,p-CH_3OC_6H_4Br$

#### D. $p-CH_3C_6H_4Br$

#### Answer: A



2. Hofmann's bromamide degradation reaction

is shown by

#### A. $ArNH_2$

#### B. $ArCONH_2$

#### $\mathsf{C}.ArNO_2$

#### D. $ArCH_2NH_2$

#### Answer: B



#### **3.** A positive carbylamine test is given by:

A. p-methylbenzylamine

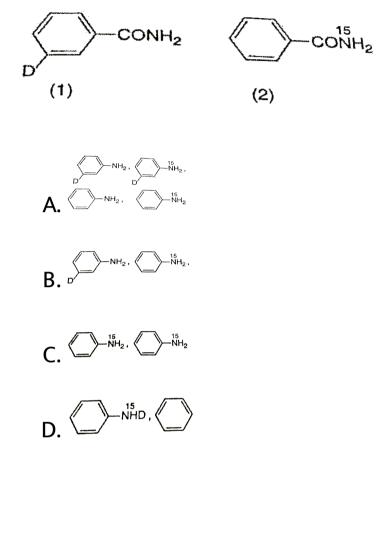
- B. N, N-dimethylanilline
- C. 2, 4-dimethylaniline
- D. N-methyl-o-methyl aniline

Answer: A

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**4.** What are the constituent amines formed when the mixture of I and II undergoes

Hofmann's bromamide degradation?



#### **Answer: B**



5. The primary, secondary and tertiary amines

can be best distinguished by

A. mustard oil reaction

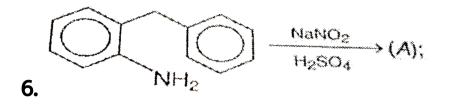
B. carbylamine reaction

C. exhaustive alkylation

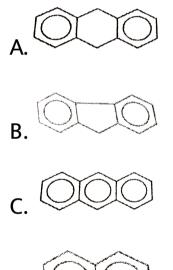
D.  $HNO_2$  treatment

Answer: D

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Product of this reaction is





#### Answer: B



7. Which of the following compounds does not liberate nitrogen with  $HNO_2$ ?

A. Carbamide

B. Primary amine

C. Secondary amine

D. Alkanamide

Answer: C

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**8.** Among the following compounds the one that is most reactive towards electrophilic nitration is

A. toluene

B. benzene

C. benzoic acid

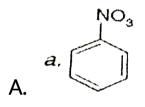
D. nitrobenzene

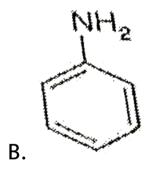
#### Answer: A

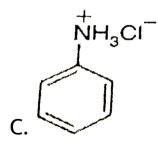
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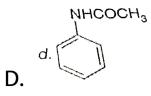
#### 9. Towards electrophilic substitution, the most

reactive species will be









#### Answer: B

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# **10.** When aniline is treated with fuming sulphuric acid at 475 K, it gives

A. sulphanilic acid

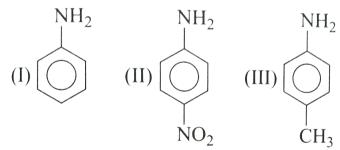
B. aniline sulphate

C. o-aminobenzene sulphonic acid

D. m-aminobenzene sulphonic acid

#### Answer: A





11.

The correct increasing order of basic strength for the following compounds is :

#### A. II < III < I

#### B. III < I < II

 $\mathsf{C}.\,III < II < I$ 

D. II < I < III

#### Answer: D

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12. Amino group,  $-NH_2$  is ortho, paradirecting group in case of aromatic electrophilic substitution but nitration of aniline produce a good amount of mnitroaniline. This is because A.  $-NH_2$  gets converted into  $NH^{-}NO_{2}^{+}$  which is m-directing B.  $NH_2$  gets converted into  $\stackrel{+}{N}H_3$  which is m-directing  $C. - NH_2$  gets converted into  $-NH^+SO_4^-$  which is m-directing D. ortho, para activity of  $-NH_2$  group is completely destroyed during nitration

#### Answer: B



**13.** Reduction of aromatic nitro compounds using Fe and HCl gives

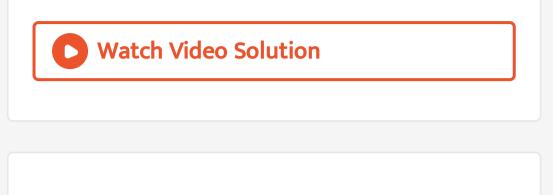
A. aromatic oxime

B. aromatic hydrocarbon

C. aromatic primary amine

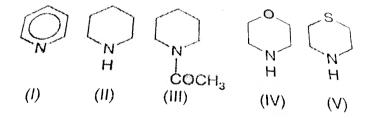
D. aromatic amide

#### Answer: C



14. The relative order of basic character of the

following compounds is



A. II > I > III > IV > V

 $\mathsf{B}.\,II > III > IV > V > I$ 

#### $\mathsf{C}.\,II > V > IV > I > III$

#### $\mathsf{D}.\,II > IV > V > III > I$

#### Answer: C



#### 15. Which of the following amine does not

react with Hinsberg reagent-

A. Neopentyl amine

B. Isopropyl amine

C. Triethyl amine

D. Ethyl methyl amine

Answer: C



16. When primary aminereacts with chloroform

in ehthanolic KOH then the product is .

A. an isocyanide

B. an aldehyde

C. a cyanide

D. an alcohol

#### Answer: A



## **17.** Tertiary nitro compounds do not tautomerise because

A. there is no double bond

B. there is no `alpha-hydrogen

C. oxygen is more electronegative than H

D. All of the above

Answer: B



**18.** When p-toluidine reacts with sodium nitrite and hydrochloric acid at 274 K, a crystalline precipitate is formed, with is boiled with water. The resulting compound obtained is A. p -cresol

B. p -nitro toluene

C. phenol toluic acid

D.

#### **Answer: A**



**19.** The reagent used to distinguish pmethylaniline from N-methylaniline is

#### A. benzenesulphonyl chloride

#### B. iodoform in aic. KOH

C. AgCl

D.  $AgNO_3$ 

#### Answer:

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20. The best reagent for converting 2-

phenylpropanamide into 2-

phenylpropanamine is

A. excess  $H_2$ 

B.  $Br_2$  in (aq) NaOH

C. iodine in the presence of phosphorus

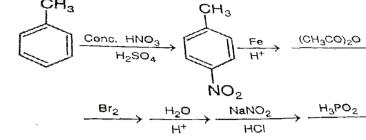
D.  $LiAlH_4$  in ether

**Answer:** 

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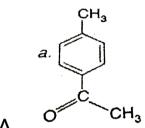
**21.** Consider the following reaction sequence.

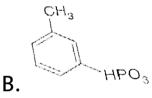
The final product of this reaction sequence is



5

\$





CH3

Br

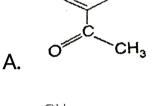
ŅНСОСН₃

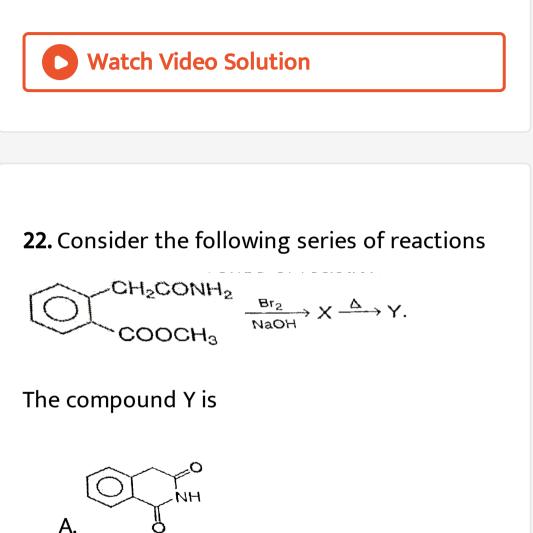
CH2NH2

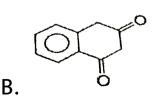
C.

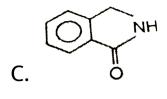
D.

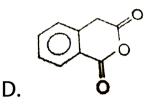
d.











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**23.** Compound A  $(C_3H_9N)$  reacts with benzene sulphonyl chloride to form a solid insoluble in alkali. The structure of compound

A. 
$$CH_3 - \mathop{N}\limits_{| \atop CH_3} - CH_3$$

 $\mathsf{B.}\,CH_3-CH_2-NH-CH_3$ 

 $\mathsf{C.}\,CH_3-CH_2-CH_2-NH_2$ 

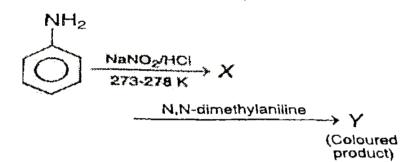
D. All of the above

#### **Answer:**

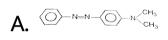


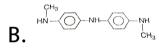
24. Aniline yields a coloured product Y through

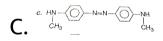
the following series of reaction:



#### The structure of Y is







D. d. 
$$H_3C - O - N = N - O - NH_2$$

#### Answer:



**25.** Consider the following statements:

Phenyl diazonium salts form azo dye with

- I. aniline
- II. Phenol
- III. N, N dimethyl aniline
- IV. Anisole (methoxybenzene)

The correct statements is

- A. II, III and IV are correct
- B. I, III and IV are correct

C. I, II and IV are correct

D. I, II and III are correct

#### Answer:

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26. Benzylamine may be alkylated as shown in the following equation  $C_6H_5CH_2NH_2 + R - X \rightarrow C_6H_5CH_2NHR$ Which of the following alkyl halides is best suited for this reaction through  $S_N$  1 mechanism?

#### A. $CH_3Br$

#### $\mathsf{B.}\, C_6 H_5 Br$

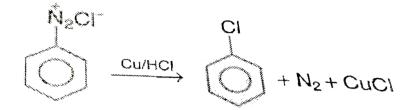
 $\mathsf{C.}\, C_6H_5CH_2Br$ 

D.  $C_2H_5Br$ 

#### **Answer:**

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**27.** Consider the following reaction :



The above reaction is called

A. carbylamine reaction

B. Gattermann synthesis

C. Sandmeyer's reaction

D. Balz-Schiemann reaction

#### **Answer:**



**28.** Identify the final product (z) in the following sequence of reactions  $C_6H_5COOH \xrightarrow{(i) LiAlH_4}_{(ii) PBr_3} X \xrightarrow{KCN} Y \xrightarrow{LiAlH_4} Z$ 

A.  $C_6H_5CH_2NH_2$ 

B.  $C_6H_5CH_2CH_2NH_2$ 

 $\mathsf{C.}\, C_6H_5CH_2CH_2NH_2$ 

D.  $C_6H_5 - CH - NH_2$ 

Answer:



**29.** Hydrolysis of phenyl isocyanide forms :

A. benzoic acid

B. formic acid

C. acetanilide

D. acetic acid

Answer:

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**30.** Which of the following on reaction with nitrous acid followed by treatment with NaOH produces a blood red colouration?

A.  $RCH_2NO_2$ 

 $\mathsf{B.}\,R_3CNO_2$ 

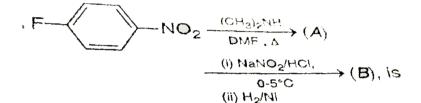
 $\mathsf{C.}\,R_2CHNO_2$ 

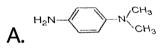
D.  $PhNO_2$ 

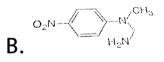
#### Answer:

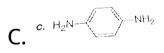
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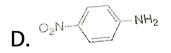
#### 31. Complete the following reaction











#### **Answer:**

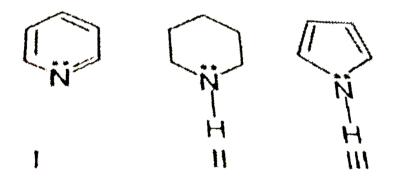




#### **Bitsat Archives**

1. Arrange the following in correct order of

#### basicity



#### A. I > II > III

#### $\mathsf{B}.III > II > I$

 $\mathsf{C}.\,II>I>III$ 

 $\mathsf{D}.\,I > III > II$ 

#### **Answer:**

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$$\textbf{2.} C_6H_5NH_2 \xrightarrow[180°C]{H_2SO_4} NH_2C_6H_4(SO_3H) \\ \underset{\text{Para form}}{\overset{\text{Para form}}{\overset{\text{form}}{\overset{\text{Para form}}{\overset{\text{Para form}}{\overset{Para form}}{\overset{Para form}}}}}}}}}}}$$

#### The true statement about the product is

A. it does not exist as Zwitter ion

B. it does not act as inner salt

 ${\rm C.}-SO_3$  diminishes the basic character of

 $-NH_2$ 

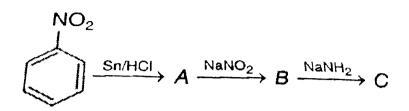
 $\mathsf{D}.-NH_2$  displays a powerful basic

character

**Answer:** 

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3. Identify C in the following reaction:

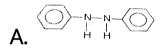


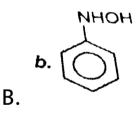
- A. benzamide
- B. benzoic acid
- C. chlorobenzene
- D. aniline

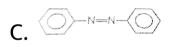
#### Answer:

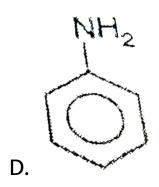


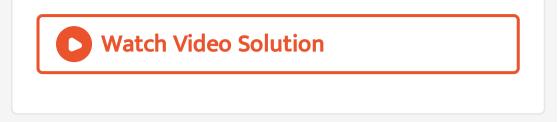
**4.** The structure of the compound formed, when nitrobenzene is reduced by lithium aluminium hydride  $(LiAlH_4)$  is

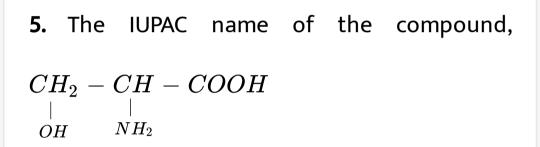










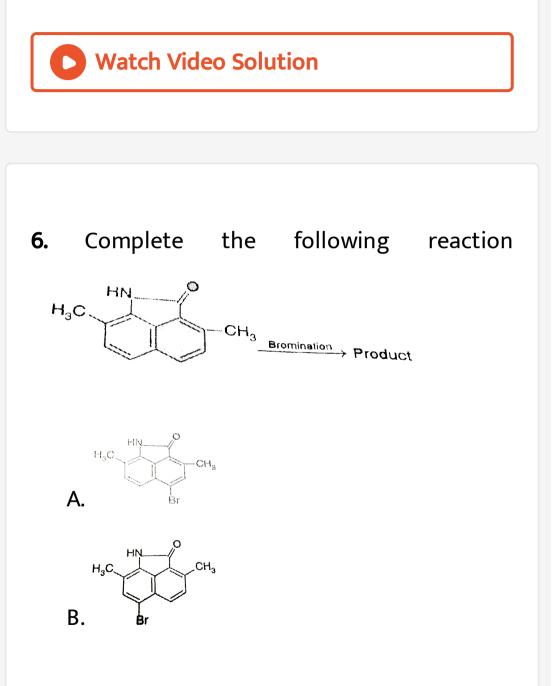


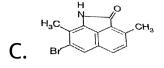
A. 2-amino-3-hydroxy propanoic acid

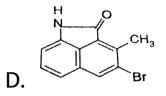
B. 1-hydroxy-2-aminopropan-3-oic acid

C. 1-amino-2-hydroxypropanoic acid

D. 3-hydroxy-2-amino propanoic acid

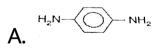


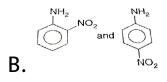


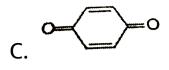


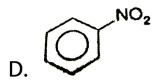
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#### 7. Aniline reacts with conc. $HNO_3$ to give









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