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CHEMISTRY

BOOKS - BITSAT GUIDE

NITROGEN CONTAINING COMPOUNDS

Practice Exercise

1. Gabriel phthalimide synthesis can be used

for the preparation of amine from

A. CH_3CH_2Br

$\mathsf{B.}\left(CH_3\right)_3 CBr$

$\mathsf{C.}\,p-CH_3OC_6H_4Br$

D. $p-CH_3C_6H_4Br$

Answer: A



2. Hofmann's bromamide degradation reaction

is shown by

A. $ArNH_2$

B. $ArCONH_2$

$\mathsf{C}.ArNO_2$

D. $ArCH_2NH_2$

Answer: B



3. A positive carbylamine test is given by:

A. p-methylbenzylamine

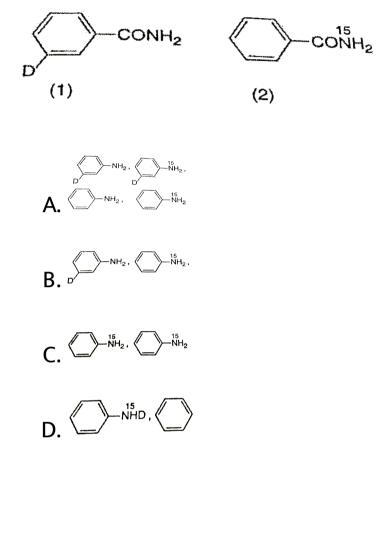
- B. N, N-dimethylanilline
- C. 2, 4-dimethylaniline
- D. N-methyl-o-methyl aniline

Answer: A

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4. What are the constituent amines formed when the mixture of I and II undergoes

Hofmann's bromamide degradation?



Answer: B



5. The primary, secondary and tertiary amines

can be best distinguished by

A. mustard oil reaction

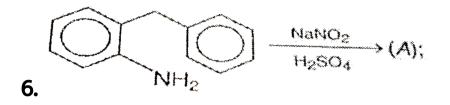
B. carbylamine reaction

C. exhaustive alkylation

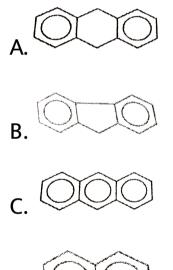
D. HNO_2 treatment

Answer: D

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Product of this reaction is





Answer: B



7. Which of the following compounds does not liberate nitrogen with HNO_2 ?

A. Carbamide

B. Primary amine

C. Secondary amine

D. Alkanamide

Answer: C

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8. Among the following compounds the one that is most reactive towards electrophilic nitration is

A. toluene

B. benzene

C. benzoic acid

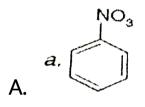
D. nitrobenzene

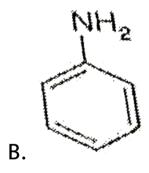
Answer: A

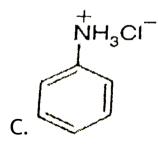
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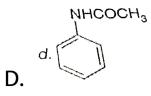
9. Towards electrophilic substitution, the most

reactive species will be









Answer: B

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10. When aniline is treated with fuming sulphuric acid at 475 K, it gives

A. sulphanilic acid

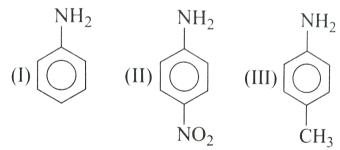
B. aniline sulphate

C. o-aminobenzene sulphonic acid

D. m-aminobenzene sulphonic acid

Answer: A





11.

The correct increasing order of basic strength for the following compounds is :

A. II < III < I

B. III < I < II

 $\mathsf{C}.\,III < II < I$

D. II < I < III

Answer: D

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12. Amino group, $-NH_2$ is ortho, paradirecting group in case of aromatic electrophilic substitution but nitration of aniline produce a good amount of mnitroaniline. This is because A. $-NH_2$ gets converted into $NH^{-}NO_{2}^{+}$ which is m-directing B. NH_2 gets converted into $\stackrel{+}{N}H_3$ which is m-directing $C. - NH_2$ gets converted into $-NH^+SO_4^-$ which is m-directing D. ortho, para activity of $-NH_2$ group is completely destroyed during nitration

Answer: B



13. Reduction of aromatic nitro compounds using Fe and HCl gives

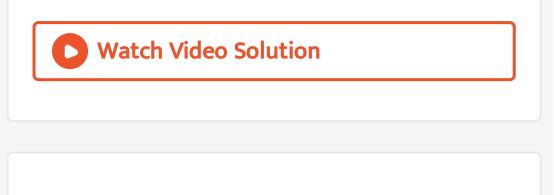
A. aromatic oxime

B. aromatic hydrocarbon

C. aromatic primary amine

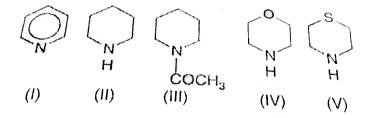
D. aromatic amide

Answer: C



14. The relative order of basic character of the

following compounds is



A. II > I > III > IV > V

 $\mathsf{B}.\,II > III > IV > V > I$

$\mathsf{C}.\,II > V > IV > I > III$

$\mathsf{D}.\,II > IV > V > III > I$

Answer: C



15. Which of the following amine does not

react with Hinsberg reagent-

A. Neopentyl amine

B. Isopropyl amine

C. Triethyl amine

D. Ethyl methyl amine

Answer: C



16. When primary aminereacts with chloroform

in ehthanolic KOH then the product is .

A. an isocyanide

B. an aldehyde

C. a cyanide

D. an alcohol

Answer: A



17. Tertiary nitro compounds do not tautomerise because

A. there is no double bond

B. there is no `alpha-hydrogen

C. oxygen is more electronegative than H

D. All of the above

Answer: B



18. When p-toluidine reacts with sodium nitrite and hydrochloric acid at 274 K, a crystalline precipitate is formed, with is boiled with water. The resulting compound obtained is A. p -cresol

B. p -nitro toluene

C. phenol toluic acid

D.

Answer: A



19. The reagent used to distinguish pmethylaniline from N-methylaniline is

A. benzenesulphonyl chloride

B. iodoform in aic. KOH

C. AgCl

D. $AgNO_3$

Answer:

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20. The best reagent for converting 2-

phenylpropanamide into 2-

phenylpropanamine is

A. excess H_2

B. Br_2 in (aq) NaOH

C. iodine in the presence of phosphorus

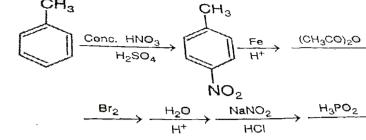
D. $LiAlH_4$ in ether

Answer:

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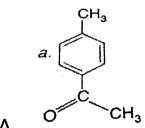
21. Consider the following reaction sequence.

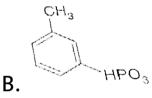
The final product of this reaction sequence is



5

\$





CH3

Br

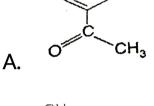
ŅНСОСН₃

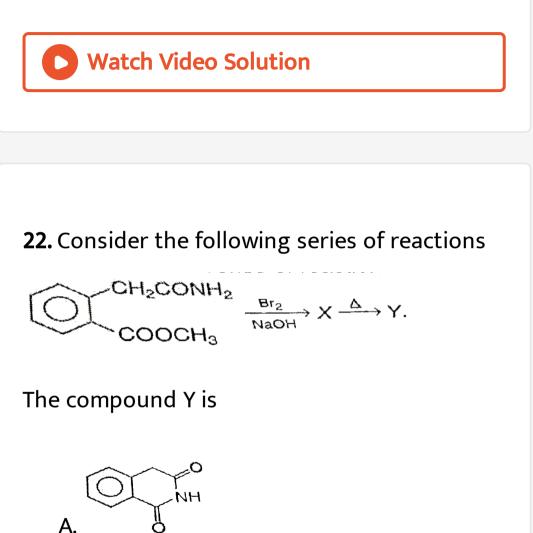
CH2NH2

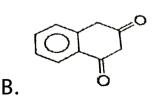
C.

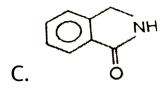
D.

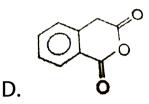
d.











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23. Compound A (C_3H_9N) reacts with benzene sulphonyl chloride to form a solid insoluble in alkali. The structure of compound

A.
$$CH_3 - \mathop{N}\limits_{| \atop CH_3} - CH_3$$

 $\mathsf{B.}\,CH_3-CH_2-NH-CH_3$

 $\mathsf{C.}\,CH_3-CH_2-CH_2-NH_2$

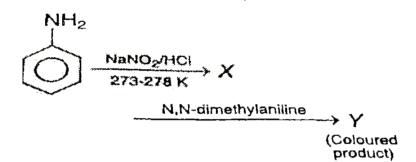
D. All of the above

Answer:

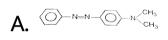


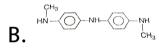
24. Aniline yields a coloured product Y through

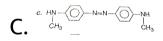
the following series of reaction:



The structure of Y is







D. d.
$$H_3C - O - N = N - O - NH_2$$

Answer:



25. Consider the following statements:

Phenyl diazonium salts form azo dye with

- I. aniline
- II. Phenol
- III. N, N dimethyl aniline
- IV. Anisole (methoxybenzene)

The correct statements is

- A. II, III and IV are correct
- B. I, III and IV are correct

C. I, II and IV are correct

D. I, II and III are correct

Answer:

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26. Benzylamine may be alkylated as shown in the following equation $C_6H_5CH_2NH_2 + R - X \rightarrow C_6H_5CH_2NHR$ Which of the following alkyl halides is best suited for this reaction through S_N 1 mechanism?

A. CH_3Br

$\mathsf{B.}\, C_6 H_5 Br$

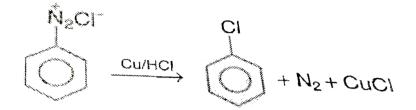
 $\mathsf{C.}\, C_6H_5CH_2Br$

D. C_2H_5Br

Answer:

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27. Consider the following reaction :



The above reaction is called

A. carbylamine reaction

B. Gattermann synthesis

C. Sandmeyer's reaction

D. Balz-Schiemann reaction

Answer:



28. Identify the final product (z) in the following sequence of reactions $C_6H_5COOH \xrightarrow{(i) LiAlH_4}_{(ii) PBr_3} X \xrightarrow{KCN} Y \xrightarrow{LiAlH_4} Z$

A. $C_6H_5CH_2NH_2$

B. $C_6H_5CH_2CH_2NH_2$

 $\mathsf{C.}\, C_6H_5CH_2CH_2NH_2$

D. $C_6H_5 - CH - NH_2$

Answer:



29. Hydrolysis of phenyl isocyanide forms :

A. benzoic acid

B. formic acid

C. acetanilide

D. acetic acid

Answer:

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30. Which of the following on reaction with nitrous acid followed by treatment with NaOH produces a blood red colouration?

A. RCH_2NO_2

 $\mathsf{B.}\,R_3CNO_2$

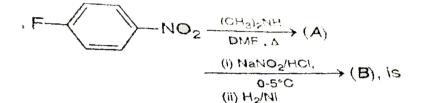
 $\mathsf{C.}\,R_2CHNO_2$

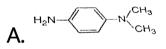
D. $PhNO_2$

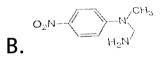
Answer:

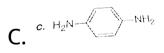
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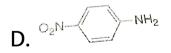
31. Complete the following reaction











Answer:

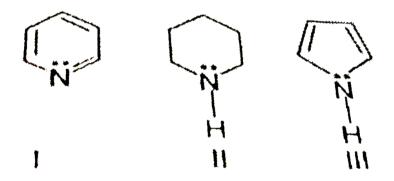




Bitsat Archives

1. Arrange the following in correct order of

basicity



A. I > II > III

$\mathsf{B}.III > II > I$

 $\mathsf{C}.\,II>I>III$

 $\mathsf{D}.\,I > III > II$

Answer:

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$$\textbf{2.} C_6H_5NH_2 \xrightarrow[180°C]{H_2SO_4} NH_2C_6H_4(SO_3H) \\ \underset{\text{Para form}}{\overset{\text{Para form}}{\overset{\text{form}}{\overset{\text{Para form}}{\overset{\text{Para form}}{\overset{Para form}}{\overset{Para form}}}}}}}}}}}$$

The true statement about the product is

A. it does not exist as Zwitter ion

B. it does not act as inner salt

 ${\rm C.}-SO_3$ diminishes the basic character of

 $-NH_2$

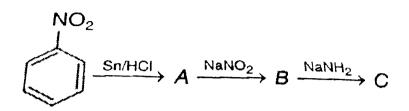
 $\mathsf{D}.-NH_2$ displays a powerful basic

character

Answer:

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3. Identify C in the following reaction:

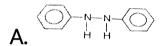


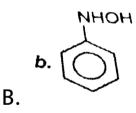
- A. benzamide
- B. benzoic acid
- C. chlorobenzene
- D. aniline

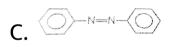
Answer:

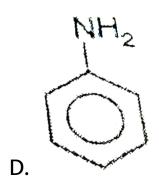


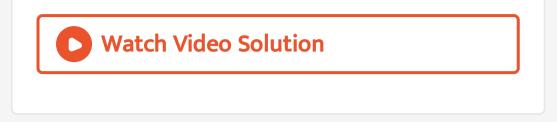
4. The structure of the compound formed, when nitrobenzene is reduced by lithium aluminium hydride $(LiAlH_4)$ is

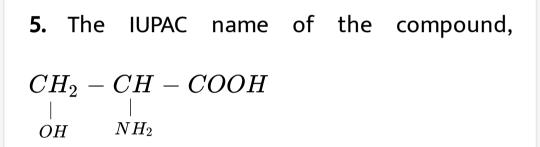










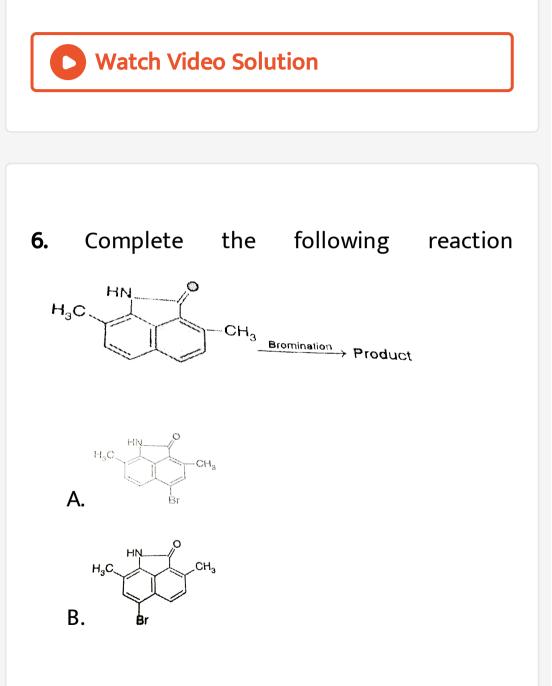


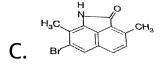
A. 2-amino-3-hydroxy propanoic acid

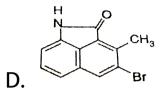
B. 1-hydroxy-2-aminopropan-3-oic acid

C. 1-amino-2-hydroxypropanoic acid

D. 3-hydroxy-2-amino propanoic acid

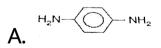


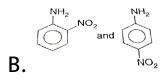


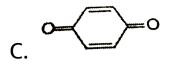


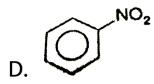
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7. Aniline reacts with conc. HNO_3 to give









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