

CHEMISTRY

BOOKS - CENGAGE CHEMISTRY (HINGLISH)

CLASSIFICATION AND NOMENCLATURE OF ORGANIC COMPOUNDS

Illustration

1. Given the IUPAC of the following alkanes:

(i)
$$CH_3-CH-CH_2-CH_3-CH_3$$
 CH_3-CH_3
 CH_3
 CH_3

(ii)
$$CH_3-CH_2-CH_2-CH_3 - CH_3 - CH_3 - CH_3 - CH_3 - CH_3$$

$$CH_3 - {\scriptsize \begin{array}{cccc} C & H - {\scriptsize \begin{array}{cccc} C & H - CH_3 \\ & C_2H_5 & & C_2H_5 \end{array}} \end{array}}$$

(iv)

v.
$$\left(CH_{3}
ight)_{3}CCH_{2}CH_{2}\stackrel{CH_{3}}{C}HCH_{2}CH_{3}$$

viii.

٧i.

$$i.~1, 1-Dimethypen an ce$$
ii. 2- Methyl-2-

proplhexan

`iii. 3-Dimethylpentane

iv. 4,4-Dimethyl-3-ethyplentane

v. 4-(2-Methylethyl) heptane

2. Given the IUPAC names of the following compounds:

(i)
$$CH_3CH_2CH_2 \overset{CH_3}{\overset{C}{C}} CH_2 - \overset{CH_3}{\overset{C}{C}} = CH_2$$

$$\overset{CH_3}{\overset{C}{\overset{C}{C}}} CH_5$$
(ii) $CH_3 - \overset{C}{\overset{C}{\overset{C}{H}}} - C \equiv CH$

(iii)
$$C_6H_5 - CH = CH - CH_2Cl$$

$$CH_3CH - CH - CH_2Cl$$

(iv)
$$CH_3CH=CH-CH_2Br$$

v
$$CH_2=CH-\overset{\circ}{C}-CH=CH_2$$

vi
$$CH_2=CH-{CH_2CH_3\choose C}H-C=CH_2$$
 vii. $(CH_3)_3C-CH=CH_2$

 CH_3



3. $CH_3 - CH - CH - CH_3$

 CH_3

$$Cl_2CH$$
 CCl_3
 V $CH_3CH_2-CH-CH-CH_2CH_3$

$$\begin{array}{ccc} (\mathrm{CH_3})_2\mathrm{CH} & \mathrm{CCl}(\mathrm{CH_3})_2 \\ \downarrow & \downarrow & \\ \mathbf{vi.} & \mathrm{CH_3CH_2CH_2} - \mathrm{CH} - \mathrm{CH} - \mathrm{CH_2CH_2CH_3} \\ \end{array}$$

$$\begin{array}{ccc} \text{Cl}_2\text{CCH}_3 & \text{CHCl} - \text{CH}_2\text{Br} \\ | & | & | \\ \text{vii.} & \text{CH}_3\text{CH}_2\text{CH}_2 - \text{CH} - \text{CH} - \text{CH}_2\text{CH}_2\text{CH}_3 \\ \end{array}$$

$$\begin{array}{ccc} \operatorname{BrCH}_2-\operatorname{ClCH} & \operatorname{CHBr}-\operatorname{CH}_2\operatorname{Cl} \\ \operatorname{viii.} & \operatorname{CH}_3\operatorname{CH}_2\operatorname{CH}_2-\operatorname{CH} ---\operatorname{CH}-\operatorname{CH}_2\operatorname{CH}_2\operatorname{CH}_3 \end{array}$$



4.
$$CH_2H_5-C - CH_2OH$$

ii

CH (3)-CH=CH-CHO

$$NH_2-CH_2-CH_2-CH_2-NH_2$$
 Itbrlt iv.

$$CH_3 - CH = CH - COOH$$

v.
$$(CH_3)_2C = CHCOCH_3$$

vi.
$$CH_2 = CH - CN$$

viii. HOOC - C = COOH

$$\mathsf{ix}\: CH_2 = egin{pmatrix} \mathit{CH}_3 \ \mathit{C} \ \mathit{COOCH}_3 \ \mathit{C} \ \mathit$$

 CH_3

X.

$$CH_{3}CH_{2}-\stackrel{\mid}{C}(Cl)-CH_{2}-CONHCH_{2}CH_{3}$$

xi.
$$CH_3 - \overset{|}{C} H - CHO$$

xii. $[(CH_3)_2CH]_3COH$

 OCH_3

xiii. $C_6H_5-CH=CH-COOH$



5. Give the IUPAC names for the following polyfunctional compounds:

i.
$$CH_3CH_2O-CH_2-CHOH-CH_3$$

$$CH_3 - u \underbrace{ndereset(\mid)(C)N - CH_2 - \overset{C}{\overset{\mid}{C}}_{OH}}_{OH} - CH_3$$

COOH



1. 1.

6. a. Rewrite the following structual formula in bond line notation.

(i)
$$CH_3CH_2CH_2CH_2CH_3$$

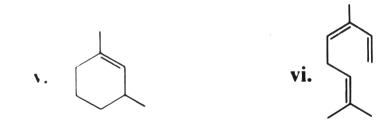
ii.
$$(CH_3)_2CHCH_2CH(CH_3)NH_2$$

iii.
$$(CH_3CH_2)NCH_3$$

iv. $CH_3CHBrCH_2CH_2CHO$

$$(CH_3)_2CHC(CH_3)CH=CH_2$$

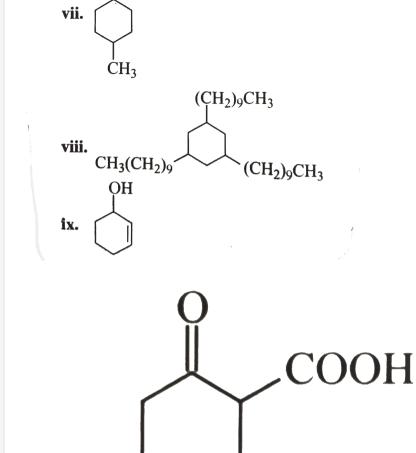
b.



7. Given the IUPAC names of the following compounds:

CH₃

 C_2H_5



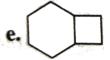


OHC

8. Give the IUPAC names of the following compounds:









Solved Example

1.

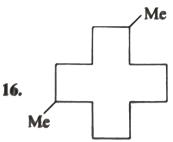
Mc-

Me

O

H

Me () Me



17. Me NH

СООН

21.

ŞН

35.

MeO

Ö



- 2. 1.6-Diethy1 cyclohexene
- 2. 4-Hydrocy-5-isopropy1 hept-6-yn-2-one
- 3. 6-Ethy1-4-isopropy1-2-methy1 decane
- 4. 2,-Dimethy1 cyclo pent-3-en-1-one
- 5. Benzonitrile
- 6. Bicyclo [4.1.0] hept-3-en-2-one
- 7. Sprio [3.5] non-5-en-2-ol
- 8. 1-Ethy1-2-methy1 cyclo hexane
- 9. 4-Bromo-2-ethy-1-methy1 cyclohexane
- 10. 3-Isopropy1 cyclopetane carbaldehyde
- 11. 4-cyclo propy1-3,6-diethy1 undecane
- 12. 6-(Cyclobut-2-enl1) hex-2-ene
 - 13. 1,4-Dicyclopenty1 butane

14. Ethylidene bromide 15. Ethylene dichloride 16. trans.-1,3 -Dibromo cyclobutance 17. 2-exo-3-endo Dichloro bicylco [2.2.1] heptane 18. Benzocyclopentene 19. 2,5-Dimethy1 oxalone or 2,5-dimethy1 oxa cyclopentane 20. Methy1 viny1 carbino1 21. 3-Cyclopenteny1 ethy1 ether 22. sec-Buty1 isopropy1 ether 23. N,N-Diethy1 butan-1-amine 24. o-Toluidine 25. Isoamy1 school 26. sec-Buty1 alcohol

- 27. isobuty1 alcohol
- 28. Maleic acid
- 29. Caproic acid

Z-Crotonalddehyde



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- **3.** Write the correct name of the following:
 - 1. Pent-1-yn-5-ol
- 2. 2-Ethy1pent-2-ene
- 3. 4-Cyclopenty1 butane
- 4. Buty1 cyclo propane
- 5.1-Bromo cyclo hex-3-ene

- 6. 3,5,9-Trimethy1 decane
- 7. 1,3,4-Trimethy1 cyclohexan
- 8. 2-(2-Methy1 buty1) cyclohexane-1-ol
- 1,1-Dimethy1 pentane



4. Write the structure of cety*ethy*1 dimethy1 ammonium bromide, a compound with anitseptic property and also used as a cationic detergent.

Predict its solubity behaviour in water and diethy ether.

b. Write structure of $\mathsf{cety} trimethy$ ammonium

bromide, a popular cationic dtergent used in hair condittiners.



5. Write the structure (s) of the simplest alkane (s), with fewest number of C atoms, possessing $1^\circ, 2^\circ, 3^\circ$, and $4^\circ C$ atoms.





- **1.** write the structural formula for each of the following
- a. A 3° amine with the formula C_3H_9N .
- b. Three ethers with the formula $C_4H_{10}O$.
- c. A 3° alcohol with the formula C_4H_8O .
- d. Three 2° alky1 halides with the formula $C_5H_{11}C1$.
- e. Three $1^{\circ}\,$ alcohols with the formula $C_4H_8O.$
- f. A 2° wiht the foumula C_3H_6O .
- g. Two esters with the formula C_3H_6O .
- g. Two esters with the formula $C_3H_6O_2$.
- h. Four 1° alky1 halides with the formula $C_5H_{11}C1$

i. A 3° alky1 halide with the formula $C_5H_{11}CI$.

j. Three aldehydes with the formula $C_5H_{10}O_{f \cdot}$

k. Three ketones with the formula $C_5H_{10}O$.

l. $A2^\circ$ amine with the formula C_3H_9N .

m. Two amides with the formula C_2H_5NO .

n. Two 1° amines with the formula C_3H_9N .



2. Indicate the following as 1° , 2° , and 3°



$$\mathbf{b}.$$
 $N-Me$

 \mathbf{d} . Et₃N

e. Me
$$NH_2$$

- **3.** Indicate the following as 1° , 2° , and 3°
 - a. OH
 - · Me OH
 - e. Me OH

- b. OH
- $\mathbf{d.} \ \mathbf{Me} \ \mathbf{Me} \ \mathbf{Me}$

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4. Write the structural formula for seven compounds with the formula C_3H_6O and identify the funcitonal groups.

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5. There are seven isomeric compounds with the formula $C_4H_{10}O$ Write their structures and identify their functional groups



6. There are four alky1 chlorides with the formula C_4H_9CI . Writre their structures and identify then as $1^\circ, 2^\circ$, and 3° alky1 chlorides



7. There are four amides with the formula C_3H_7NO .

Write their structures.

Identify the amide which has lower melting point and boiling point than the other three.



8. Write the IUPAC name of the compound (A) which is a 2-methy1 branched alkane having a molecular mass of 245. This compound is a sexattractant and is isolated from female tiger moths.

9. Write the IUPAC name of the compound (A)which is a 2-methyl alkane with molecular mass of 72.

A. 2-methyl butane

B. 2-methyl cyclobutane

C. 2-methyl pentane

D. 2-methyl propane

Answer: A



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10. Write the IUPAC name of the compound (A) in which the molar rations of C, H, and O of equal having a molecular mass of 58

- A. Propanal
- B. Ethane-1,2-diol
- C. Ethane-1,2-dial
- D. None of the above

Answer: C



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Concept Application Type

1. Given the hybridistaiton state of each carbon in the following compounds :

I.
$$CH_2=C=O$$
 II. $CH_3-CH=CH_2$

III.
$$(CH_3)_2CO$$
 IV. $CH_2=CH=C\equiv N$

v. C_6H_6



- **2.** Indicate the $\sigma-$ and $\pi-$ bonds in the following molecules:
 - I. C_6H_6 II. C_6H_{12} III. CH_2CI_2

 $\mathsf{IV}.\,CH_2 = C = CH_2\,\mathsf{V}.\,CH_3NO_2$

VI. $HCOMHCH_3$



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3. Write the bond line formula for the following compounds:

`I. Ilsopropy1 alcohol II. 2,3-Dimethy1 butanal

III. Heptan-4-one



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4. Give the IUPAC names of the following compounds

c.

b. CN

d. Cl Br

e. Cl

f. Cl₂CHCH₂OH

b. 尾

c. 📝





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5. Which of the following represents the correct IUPAC name for the compounds concerned?

a. 2,2-Dimethy1 pentane or 2-Dimethy1 pentane

b. 2,4,7-Trimethy1octane or 2,5,7-Trimethyloctane

c. 2-Chloro-4-methy1pentane or 4-Chloro-2-melthypentane

d. But-3-yn-1-ol or But-4-ol-1-yne



6. Draw the formulae for the first five numbers of each homologous series beginnig with the following compounds:

a. H-COOH

b. CH_3COCH_3

 $\mathsf{c.}\,H-CH=CH_2$



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7. Identify the functional groups in the following compounds.



Linked Comprehesion Type

1. The analgesic drug ibuprofen (A) is chiral and exists in (+) and (-) froms. One enantiomer is physiologically active, while the other is inactive. The other is inactive. The structure of ibuprofen is given below.

The principal functional group in (A) is:

A. Pheny1

B. - COOH group

C. Isopropy1

D. Both (a) and (b)

Answer: B



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2. The analgesic drug ibuprofen (A) is chiral and exists in (+) and (-) froms. One enantiomer is physiologically active, while the other is inactive. The structure of ibuprofen is given below.

The IUPAC name of (A) is:

A. 3-(p-Isobutyl phenyl) propanoic acid

B. 2-(p-Isobutyl phenyl) propanoic acid

C. 3-(p-sec-Butyl phenyl) propanoic acid

D. 2-(p-sec-Buty1 pheny1) propanoic acid

Answer: B



3. The analgesic drug ibuprofen (A) is chiral and exists in (+) and (-) froms. One enantiomer is physiologically active, while the other is inactive. The other is inactive. The other is inactive. The structure of ibuprofen is given below.

The number of π - bonds in (A) is :

A. 2

B. 3

C. 4

D. 5

Answer: C



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4. The analgesic drug ibuprofen (A) is chiral and exists in (+) and (-) froms. One enantiomer is physiologically active, while the other is inactive. The other is inactive. The structure of ibuprofen is given below.

The number of σ -bonds in (A) is:

A. 30

B. 31

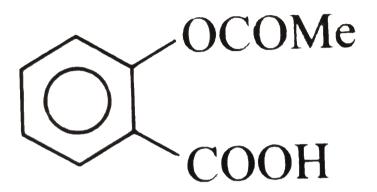
C. 32

D. 33

Answer: D



5. Aspirin is widely used as an analgesic drug. It is optically inactive. The structure of aspirin is:



The principal functional group $\operatorname{in}(A)$ is :

A. Phenyl

B.-COOH

C. Ester

D. All

Answer: B



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6. Aspirin is widely used as an analgesic drug. It is optically inactive. The structure of asprim is:

Which of the following is not the correct name for aspirin?

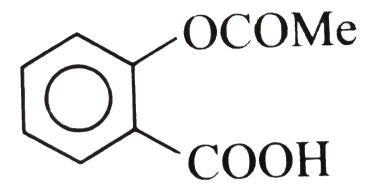
- A. 2-Acety salicylic acid
- B. 2-Acetoxy benzoc acid
- C. 2-Acetoxy salicylic acid
- D. None

Answer: C



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7. Aspirin is widely used as an analgesic drug. It is optically inactive. The structure of asprim is:



The number of π - bonds in (A) is:

A. 3

B. 4

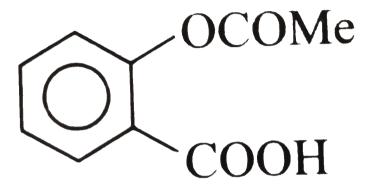
C. 5

D. 6

Answer: C



8. Aspirin is widely used as an analgesic drug. It is optically inactive. The structure of aspirin is:



The number is σ - bonds in (A) is:

A. 19

B. 20

C. 21

Answer: C



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9. Crixivan, a drug produced by Merck and Co., is widely used in the fight at against AIDS (acquied immune dificiency syndrome). The sturcture of cirxivan is given below:

How many 2° alcohol groups are present in the above compound? A. Zero

B. 1

C. 2

D. 3

Answer: C



10. Crixivan, a drug produced by Merck and Co., is widely used in the fight against AIDS (acquired immune deficiency syndrome). The structure of crixivan is given below:

How many amide groups are present in the compound?

A. Zero

B. 1

C. 2

D. 3

Answer: C



11. Crixivan, a drug produced by Merck and Co., is widely used in the fight at against AIDS (acquied immune dificiency syndrome). The sturcture of cirxivan is given below:

How many 3° amine groups are present in the compound?

A. Zero

B. 1

C. 2

D. 3

Answer: C



12. Crixivan, a drug produced by Merck and Co., is widely used in the fight at against AIDS (acquied immune dificiency syndrome). The sturcture of cirxivan is given below:

How many 2° amine groups are present in the compound ?

A. Zero

B. 1

C. 2

D. 3

Answer: A



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Multiple Corrcct Answers Types

1. Which of the following statements is/arewrongs?

A. C_nH_{2n} is the general formula of alkanes.

- B. In holologous series, all member have the same physical properties.
- C. IUPAC means Internatioal Union of Physics and Chemisty.
- D. Butane contains two $1^{\circ}C$ atoms and two $2^{\circ}C$ atoms.

Answer: A::B::C



- **2.** Which of the following statements is/are correct?
 - A. Homologous series can be represented by a general formula.
 - B. The chemical properties of an organic compound depend on the funcitonal group.
 - C. Groups obtained by the removal of one ${\cal H}$ atom from the alkane are called alky1 group.
 - D. Alkynes consist of one doulbe-bond in their molelues.

Answer: A::B::C



- **3.** Which of the following statements is/are wrong?
 - A. Acetic acid is the systematic name of vinegar.
 - B. Me-C-OH is an unsaturated compound.
 - C. Prefixes like n-,iso, sec-,tert,neo-,etc., are used in IPUAC system.

D. The systemic names of acids are formed by dropping-e of the name of parent alkane and adding -oic acid.

Answer: A::B::C



4. Which of the following statements is/are correct?

A.
$$\left(R-C-O-C-R
ight)$$
 is an usaturated compound.

- B. Neohydrocarbons contain a $3^{\circ}\,C$ atom.
- C. The IUPAC name of isopropy alcohol is propan-2-ol.
- D. The IUPAC name of (CH_3CN) ie ethanenitrile.

Answer: C::D



- **5.** Which of the following statements is/are correct?
 - A. Methane was named as fire damp as it froms explosive mixture with air.
 - B. Primary suffixes are added to root word to show struction or unsturation in a ${\cal C}$ atom.
 - C. The IUPAC name of valeric acid is pentanoic acid.
 - D. The common name of hexanoic acid is caproic acid.

Answer: A::B::C::D



- **6.** Which of the following statements is/are
- (i)The IUPAC name of amyl alcohol is pentanol.
- (ii)The IUPAC name of isoamyl alcohol is 3-methyl butanol.
- (iii)Wood spirit is methanol.
- (iv)Methyl alcohol is also called carbinol

A. The IUPAC name of amyl` alcohol is pentanol.

B. The IUPAC name of isoamyl alcohol is 3-methyl butanol.

C. Wood spirit is methanol.

D. Methyl alcohol is also called carbinol

Answer: A::B::C::D



7. Which of the following statements is/are correct?

A. The trivial name of organic compounds are called common names.

- B. The systematic name of organic compounds are obtained from the IUPAC system.
- C. The systematic names of alkanes are based on the numebe of ${\cal C}$ atoms in the longest continuous chain of ${\cal C}$ atoms.

D. The maximum number of functional groups ${\sf must}$ be included in the ${\sf C}$ atom chain selected even if it does not satisfy the longest chain rule.

Answer: A::B::C::D



8. Which of the following statements is/are correct?

A. The common name of

$$(HOOC-CH_2-COOH)$$
 is malonic acid.

B. The common name of

C. The IUPAC name of

$$(CH_2=CH-OCOCH_3)$$
 is viny1 acetate.

D. The IUPAC name of acrylonitrile is Prop-2-ene-nitrile.

Answer: A::B::D



9. Which of the following statements is/are correct?

A. The common name of benzene-1,2-diol is catechol.

B. The common name of benzene-1,3-diol is resorinol.

C. The common name of benzene-1,4-diol is quinol.

D. The common name of benzene -1,4-diol is

hydroquinone

Answer: A::B::C::D



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10. Which of the following statements is/are correct?

A. The common name of benzene-1,2,3-triol is pyrogallol.

B. The common name of benzene-1,2,4-triol is hydroxquinol.

C. The common name of benzene-1,3,5-triol is phoroglucinol.

D. The common name of $\left(CH_2=CH-Ph\right)$ is styrene.

Answer: A::B::C::D



11. The compounds in which C uses its sp^3 - hybrid orbitals for bond formation are:

A. HCOOH

B. $(H_2N)_2CO$

 $C.(CH_3)_3COH$

D. CH_3CHO

Answer: C::D



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Single Correct Answer Type

1. The decreasing order of priority of the following functional group is:

I.-COOH

 $II. -SO_3H$

III. -COOR

IV. -COCl

$$\mathsf{A.}\left(IV\right)>\left(III\right)>\left(II\right)>\left(I\right)$$

$$\mathsf{B.}\left(I\right)>\left(II\right)>\left(III\right)>\left(IV\right)$$

$$\mathsf{C.}\left(II\right)>\left(I\right)>\left(III\right)>\left(IV\right)$$

$$\mathsf{D}.\left(IV\right)>\left(III\right)>\left(I\right)>\left(II\right)$$

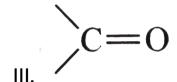
Answer: B



2. The decreasing order of priority for the following functional group is

I.
$$C \equiv N$$

$$II. -CONH_2$$



$$IV - CHO$$

A.
$$(II) > (I) > (IV) > (III)$$

B.
$$(III) > (IV) > (I) > (II)$$

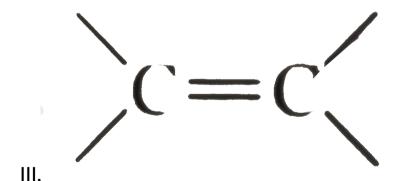
Answer: A



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3. The decreasig order of priority for the following functional groups is:

$$\operatorname{I.-}OH\operatorname{II}\operatorname{gt}-C\equiv C-$$



 $-NH_2$

$$\mathsf{A.}\left(IV\right)>\left(I\right)>\left(II\right)>\left(III\right)$$

IV.

$$\mathsf{B.}\left(IV\right)>\left(I\right)>\left(III\right)>\left(II\right)$$

$$\mathsf{C.}\left(I\right)>\left(IV\right)>\left(III\right)>\left(II\right)$$

$$\mathsf{D}.\left(I\right)>\left(III\right)>\left(IV\right)>\left(I\right)$$

Answer: C



4. The number of 1° , 2° , and $3^{\circ}H$ atoms in 2,5,6-trimethy1 octane, respectively,is

A. 16,5,3

B. 15,5,3

- C. 16,6,3
- D. 15,5,2

Answer: B



- **5.** The number of $1^{\circ}, 2^{\circ}$, and $3^{\circ}H$ atoms in 3ethy1-5-methy1 heptane, respectively, is:
 - A. 12,8,1
 - B. 14,4,2
 - C. 12,6,2

D. 12,8,2

Answer: D



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6. The number of σ - and π -bond in hexan-2,4-diol, respectively,is:

A. 18,2

B. 17,2

C. 17,1

D. 18,1

Answer: B



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7. The number of σ - and π -bond in 5-oxohexanoic acid, respectively, is :

A. 18,2

B. 18,1

C. 17,2

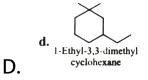
D. 17,1

Answer: A

8. Which of the following is correctly named?

4-chloro-1,3-dinitro benzene

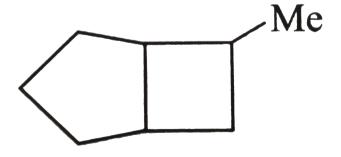
4-methyl-5-chloronitrobenzene



Answer: C



9. The systematic naming of the following cycloalkane is



- A. 6-Methy1 bicyclo [3.2.0] heptane
- B. 7-Methy1 bicyclo [3.2.0] heptane
- C. 2-Methy1 bicyclo [3.2.0] heptane
- D. 3-Methy1 bicyclo [3.2.0] heptane

Answer: A



10. In which of the following reactions, the principal group loses its preferences?

I.
$$Me$$
 $COOH$
 Δ
 Me
 Me
 Me

II. Me
$$COOH \xrightarrow{\Delta} Me$$

III. Me

COOH

COOH

$$A \rightarrow A$$

COOH

COOH

a. I

c. I, II, III

d. I, II

A. I

B. I, II

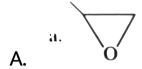
C. I, II, III

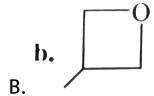
D. I,II

Answer: B



11. Which of the following is oxetane?

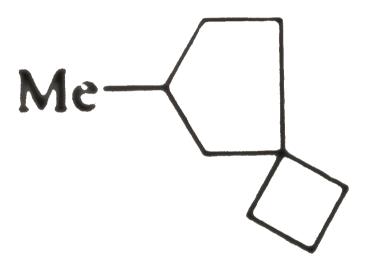




Answer: B



12. The systematic nomencalutre of the following spiro-compound is:



- A. 2-Methy1 spiro [3.4] octane
- B. 3-Methy1 spiro [3.4] octane
- C. 6-Methy1 spiro [3.4] octane

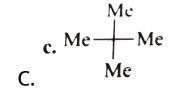
D. 7-Methy1 spiro [3.4] octane

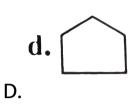
Answer: C



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13. An alkane (A) having a molecular mass of 72 produces one monochlorination product. Compound (A) is:



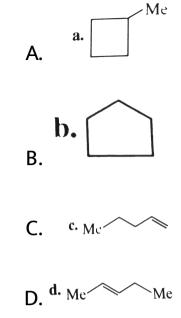


Answer: C



14. A compound (A) with molecular formula C_5H_{10} gives one monochlorination product.

Compound (A) is:



Answer: B



15. Which of the following the following in a 3° amine?

- A. Propan-2-amine
- B. N- Methy1 ehtanamine
- C. Ally1 amine
- D. N, N-Diethy1 butan-1-amine

Answer: D



- **16.** Which of the following as a 3° alcohol?
 - A. t-Buty1 carbinol
 - B. 2-Methy1 propan-2-ol

C. 2-Methy1 butan-1-ol

D. isoamy1 alcohol

Answer: B



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17. Which of the following is zerone?

A. MeOH

B. **b.** HO

D. EtOH

Answer: A



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18. Which of the following is pyrogallol?

Answer: A



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19. In 3-chloro cyclohexanol, the primary prefix is:

A. 3-Chloro

B. Cyclo

C. ane

D.-ol

Answer: B



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20. In 2-Chloro-3-methyl hexanoic acid, the primary suffix is:

A. 2-Chloro-

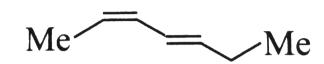
B. -3- Methyl

C. -an(e)

D. oic acid

Answer: C

21. The correct name of the compound (I) is:



A. $(E-2),\,(E-4)$, Hepta-2,4-diene

B. (Z-2), (Z-4), Hepta-2,4-diene

C. (E-2), (Z-4), Hepta-2,4-diene

D. (Z-2), (E-4), Hepta-2,4-diene

Answer: D



22. Which of the following is not a cumulated diene?

A. Hexa-1,2-diene

B. Hexa-2,3-diene

C. Penta-2,3-diene

D. Penta-1,3-diene

Answer: D



23. Which of the following has only 1° and $2^\circ C$ atoms?

A. 2-Methyl butane

B. Butane

C. 2,2-Dimethyl butane

D. 2,2,3,3-Tetramethyl pentane

Answer: B



24. The IUPAC name of vinyl acetylene is:

- A. Pent-1-en-4-yne
- B. Pent-4-yn-1-ene
- C. But-1-en-3-yne
- D. But-1-yn-3-ene

Answer: C



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25. Which of the following structures represents cyclopentyl methyl carbinol?

A.
$$a \cdot CH_3$$

Answer: C



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26. The IUPAC name of acrolein is:

- A. But-2-enal
- B. Prop-2-enal
- C. But-3-enal
- D. 2-Methyl prop-2-enal

Answer: B



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27. The IUPAC name of the following compound is

$$\left(\begin{array}{c} NC \\ CN \end{array}\right)$$

- A. Propane-1,2,3-tricarbonitrile
- B. 3-Cyanopentane1,5-dinitrile
- C. Pentane-1,3,5-trinitrile
- D. All

Answer: A



28. Which of the following is not the name of CH_3NC ?

- A. Methy1 isocyanide
- B. Aceto isonitrile
- C. Methy1 carby1 amine
- D. Acetronitrile

Answer: D



29. The IUPAC name of PhCN is :

- A. Phenyl cyanide
- B. Benzonitrile
- C. Benzene carbonitrile
- D. All

Answer: C



30. Give the name of:

A. 2,2-Dimethyl-3-propyl-4-isopropyl heptane

B. 4-isopropyl-5-t-butyl octane

C. 4-t-butyl-5-isopropyl octane

D. 2-Methyl-3-propyl-4-isopropyl heptane

Answer: C

31. Which of the following statements is wrong for homologus series?

A. All members have a general formula.

B. All members have the same functional group.

C. All memberss have the same chemical properties.

D. All members have the same physical properties.

Answer: D



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32. The alkane which has only $1^{\circ}H$ atoms is:

- A. Neppentane
- B. Isopentane
- C. Pentane
- D. 2,2-Dimethy1 betane

Answer: A



33. Which group is always taken as a substituent in the IUPAC system of nomenclature ?

$$A.-NO_2$$

$$B.-C \equiv N$$

$$C=0$$

$$D.-NH_2$$

Answer: A



34. The IUPAC name of

$$(CH_3)_3C-CH=CH_2$$
 is:

A. 2,2-Dimethy1 but-3-ene

B. 2,2-Dimethy1 pent-4-ene

C. 3,3-Dimethy1 but-1-ene

D. Hex-1-ene

Answer: C



35. Which compound is 2,2,3-trimethyl hexane?

$$\mathsf{C}H_3 \quad \overset{C}{C}H_3 \\ -\overset{|}{C} \quad & \overset{|}{C} \quad H - CH_2 - CH_2 - CH_3 \\ \overset{C}{C}H_3 \quad & \overset{C}{C}H_3 \\ & \overset{C}{C}H_3 \quad & \overset{C}{C}H_3 \\ & \overset{|}{C}H_3 \quad & \overset{C}{C}H_3 \\ & \overset{|}{C}H_3 \quad & \overset{C}{C}H_3 \\ & \overset{C}{C}H_3 \quad & \overset{C}{C}H_3 \\ & & \overset{C}{C}H_3 \quad & \overset{C}{C}H_3 \\ & & & & & & & & \\ & & & & & & & \\ & & & & & & & \\ & & & & & & & \\ & & & & & & & \\ & & & & & & & \\ & & & & & & & \\ & & & & & & & \\ & & & & & & \\ & & & & & & \\ & & & & & & \\ & & & & & & \\ & & & & & & \\ & & & & & & \\ & & & & & & \\ & & & & & & \\ & & & & & & \\ & & & & & & \\ & & & & & & \\ & & & & & & \\ & & & & & & \\ & & & & & & \\ & & & & & & \\ & & & & & \\ & & & & & & \\ & & & & & \\ & & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & \\ & & & & & \\ & & & & \\ & & & & \\ & & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & \\ & & & & \\ & & & & \\ & & & \\ & & & \\ & & & & \\ & & & \\ & & & & \\$$

Answer: A



36. The bond between carbon atom (1) and carbon atom (2) in the compound $N\equiv C-CH=CH_2$ in involves the hybridisation as :

- A. sp^2 and sp^2
- B. sp^3 and sp
- C. sp and sp^2
- D. sp and sp

Answer: C



37. If two compounds have the same empirical forumla byt different molecular formula, they must have:

- A. Different percentage composition
- B. Different molecular weights
- C. Same velocity
- D. Same vapour density

Answer: B



38. The number of σ -and π -bond in 1-buten-3-yne is:

A.
$$5\sigma$$
 and 5π

B.
$$7\sigma$$
 and 3π

C.
$$8\sigma$$
 and 2π

D.
$$6\sigma$$
 and 4π

Answer: B



39. Which of the following compounds has isopropyl group?

- A. 2,2,3,3-Trimethyl pentane
- B. 2,2-Dimethyl pentane
- C. 2,2,3-Trimethyl pentane
- D. 2-Methyl pentane

Answer: D



40. The correct IUPAC name of the compound is:

$$CH_3$$

 $CH_3CH_2-C=CH-CH-CH_2-CH_3$
 $CH_3CH_2-CH-CH_2-CH_2-CH_2-CH_3$

- A. 5,6-Dimethy1-3-methy1 dec-4-ene
- B. 5,6-Dimethy1-8-methy1 dec-6-ene
- C. 6-Buty1-5-ethy1-3-methy1 oct-4-ene
- D. 2,4,5-Triethy1-3-nonene

Answer: A



41. The hybridisation of C atoms in (C-C) single-bond of $H-C\equiv C-CH=CH_2$ is :

A.
$$H_2C=CH-C\equiv CH$$

B.
$$HC \equiv C - CH_2 - C \equiv CH$$

$$\mathsf{C.}\,H_2C=C=C=CH_2$$

D.
$$sp^3 - sp$$

Answer: C



42. Which of the following represents the given mode of hybridisation $sp^2-sp^2-sp-sp$ from left to right?

A.
$$H_2 = CH - C = CH$$

B.
$$HC \equiv C - CH_2 - C \equiv CH$$

$$\operatorname{C.} H_2C=C=C=CH_2$$

D. d.
$$_{H_2C}$$
 CH_2

Answer: A



43. The IUPAC name of C_6H_5COCl is :

A. Benzoyl chloride

B. Benzene chloro ketone

C. Benzene carbonyl chloride

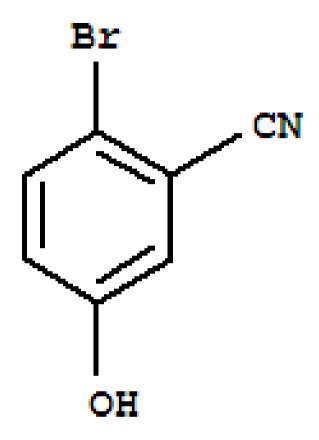
D. Chlorophenyl ketone

Answer: A



44. The IUPAC name of the following compound

is



A. 4-Bromo-3-cyanophenol

B. 2-Bromo-5-hydroxy benzo nitrile

C. 2-Cyano-4-hydroxy bromo benzene

D. 6-Bromo-3-hydroxy benzonitrile

Answer: B



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Assertion Reasoning Type

1. Assertion (A): Pentane and 2-methyl pentane are homologous.

Reason (R): Pentane is a straight-chain alkane, while 2-methyl pentane is a branched-chain alkane.

A. If both (A) and (R) are correct and (R) is the correct explanation for (A).

B. If both (A) and (R) are correct and (R) is not the correct explanation.

C. If (A) is correct and (R) is incorrect.

D. If both (A) and (R) are incorrect.

Answer: B



2. Assertion (A): All the C atoms of but-2-ene lie in one plane

Reason (R): Double-bond C atoms are sp^2 -hypbridised.

A. If both (A) and (R) are correct and (R) is the correct explanation for (A).

B. If both (A) and (R) are correct and (R) is not the correct explanation.

C. If (A) is correct and (R) is incorrect.

D. If both (A) and (R) are incorrect.

Answer: A



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3. Assertion (A): The IUPAC name of the citric acid is 2-hydroxy-propane-1,2,3-tricarboxylic acid Reason (R): When an unbranched C atom directly linked to more than two like-functional groups, then it is named as a derivative of the parent alkane which does not include the C of the

functional groups.

A. If both (A) and (R) are correct and (R) is the correct explanation for (A).

B. If both (A) and (R) are correct and (R) is not the correct explanation.

C. If (A) is correct and (R) is incorrect.

D. If both (A) and (R) are incorrect.

Answer: A



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4. Assertion (A): Rochelle's salt is used as a complexing agent in Tollens reagent.

Reason (R). Sodium potassium salt of tartaric acid is known as Rochelle's salt. The IUPAC name of Rochelle's salt is sodium potassium-2,3-

dihydroxy butane-1,4- dioate.

A. If both (A) and (R) are correct and (R) is the correct explanation for (A).

B. If both (A) and (R) are correct and (R) is not the correct explanation.

C. If (A) is correct and (R) is incorrect.

D. If both (A) and (R) are incorrect.

Answer: D



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5. Assertion (A): The IUPAC name of isoprene is 2-methyl buta-1,3-diene.

Reason (R): Isoprene unit is a monomer of natural rubber

A. If both (A) and (R) are correct and (R) is the correct explanation for (A).

B. If both (A) and (R) are correct and (R) is not the correct explanation.

C. If (A) is correct and (R) is incorrect.

D. If both (A) and (R) are incorrect.

Answer: B



Fill In The Blanks Type

1. The compounds having both sp-1 and sp^2-1 hybridised C atoms is

A. Propene

- B. Propdiene
- C. Propane
- D. Both A and B

Answer: B



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2.ring is the most strained. (Cyclopropane, Cycloputane Cyclopentane)

Cyclobutane, Cyclopentane)



3. The terminal C atom in butane ishybridised.



4. Adiol has two hydroxy1 group on.....C atoms



5. The IUPAC name of succinic acid is......



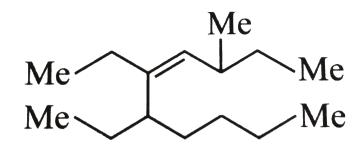
Analytical And Descriptive Type

1. Give the IUPAC name of the following compound.

$$CH_2 = CH - CH(CH_3)_2$$



2. Give the IUPAC name of the following compound.





3. Give the IUPAC name of the following compound.



