



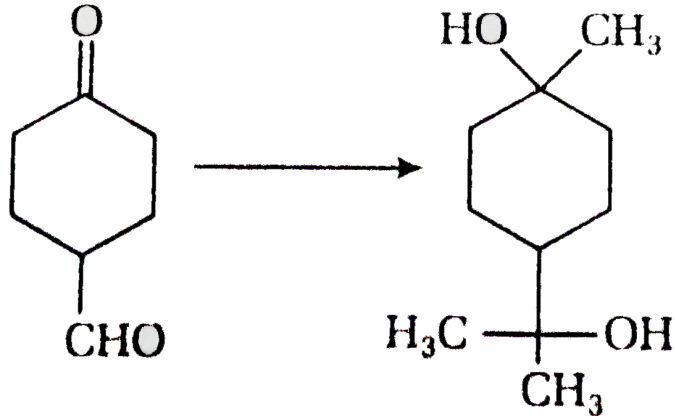
CHEMISTRY

BOOKS - IIT-JEE PREVIOUS YEAR (CHEMISTRY)

ALDEHYDES AND KETONES

Jee Main And Advanced

1. The correct sequence of reagents for the following conversion will be



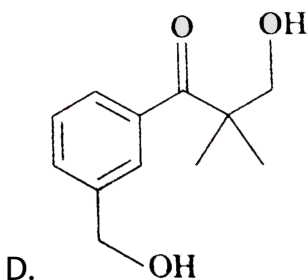
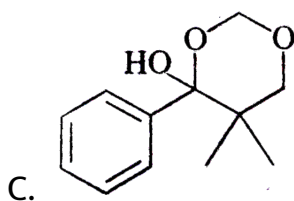
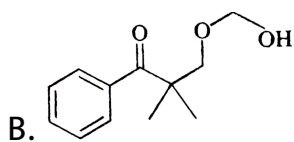
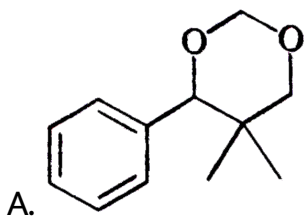
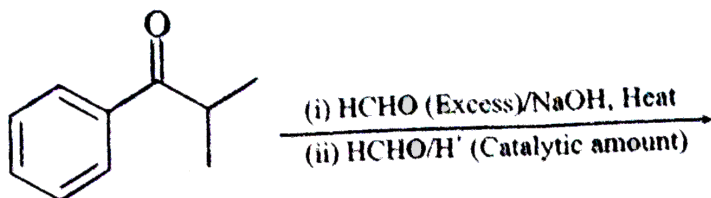
- A. $[Ag(NH_3)_2]^+ OH^-$, H / CH_3OH , CH_3MgBr
- B. CH_3MgBr , H^+ / CH_3OH , $[Ag(NH_3)_2]^+ OH^-$
- C. CH_3MgBr , $[Ag(NH_3)_2]^+ OH^-$, H / CH_3OH
- D. $[Ag(NH_3)_2]^+ OH^-$, CH_3MgBr , H^+ / CH_3OH

Answer: A



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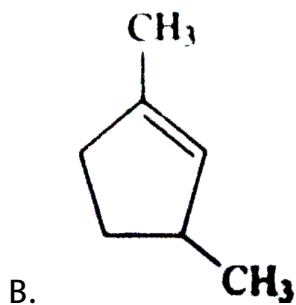
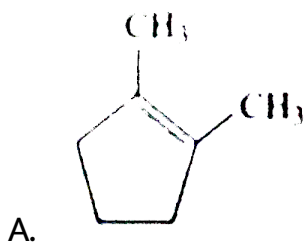
2. The major product of the following reaction sequence is is

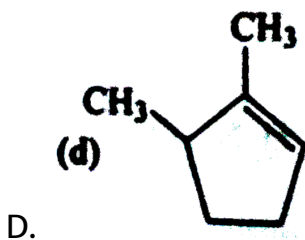
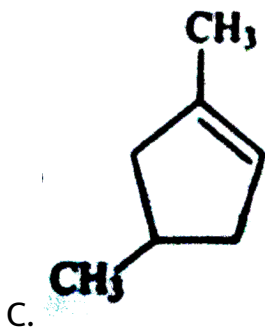


Answer: A

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3. The compound which would yield 5 - Oxo - 2 - methylhexanal on reductive ozonolysis

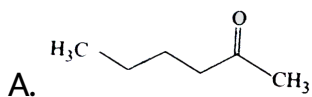
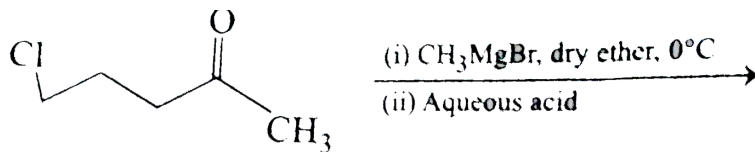


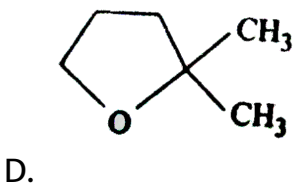
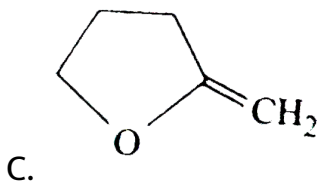
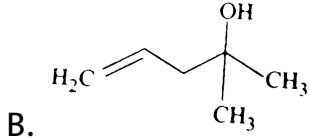


Answer: B

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4. The major product in the following reaction is

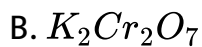
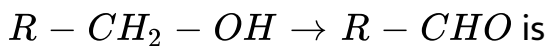




Answer: D

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5. The most suitable reagent for the conversion of



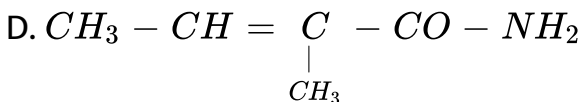
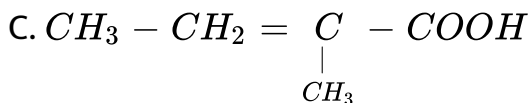
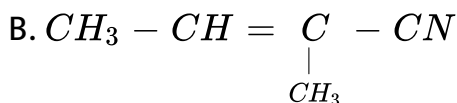
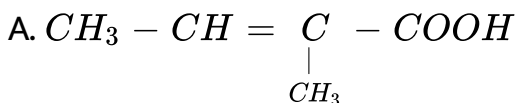
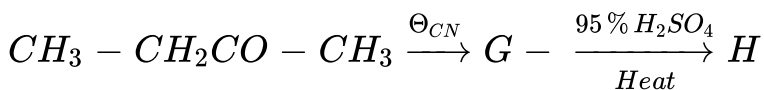
C. CrO_3

D. PC C (pyridinium chlorochromate)

Answer: D

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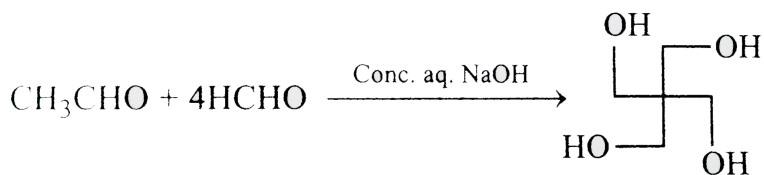
6. The major product H of the given reaction sequence is



Answer: A

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7. The number of aldol reaction (s) that occurs in the given transformation is



A. 1

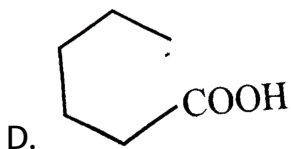
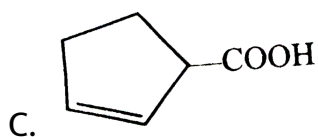
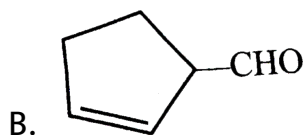
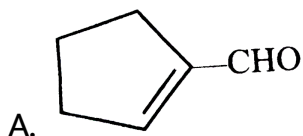
B. 2

C. 3

D. 4

Answer: C

8. Cyclohexene on ozonolysis followed by reaction with zinc dust and water gives compound E. Compound E on further treatment with aqueous KOH yields compound F. Compound F is



Answer: A

9. The smallest ketone and its next homologue are reacted with NH_2OH to form oxime.

- A. two different oximes are formed
- B. three different oximes are formed
- C. two oximes are optically active
- D. all oxime are optically active

Answer: B



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10. Butan-2-one can be converted to propanoic acid by which of the following ?

A. NaOH , NaI / H^+

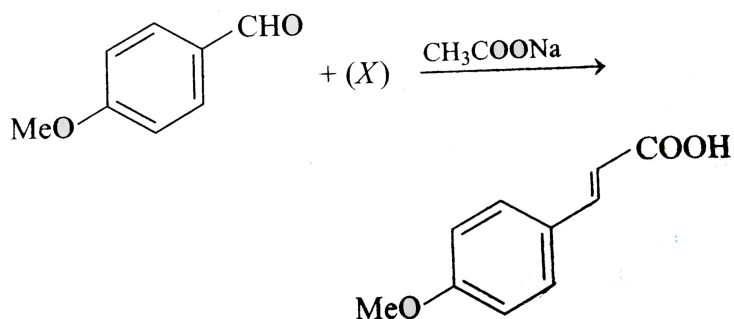
B. Fehling's solution

C. NaOH , I_2 / H^+

D. Tollen's reagent

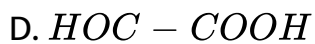
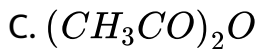
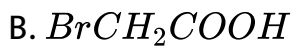
Answer: C

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What is X?

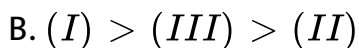
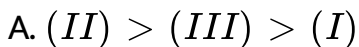
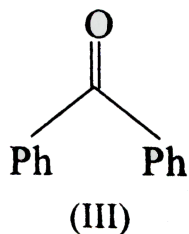
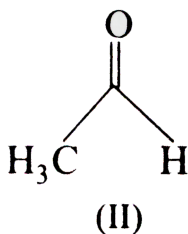
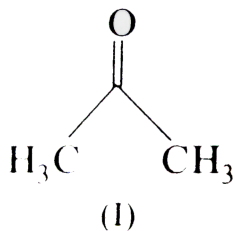
A. CH_3COOH



Answer: C

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12. The order of reactivity of phenyl magnesium bromide with the following compounds is



C. $(II) > (I) > (III)$

D. All of the above

Answer: C



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13. A mixture of benzaldehyde and formaldehyde on heating with aqueous NaOH solution gives

A. benzyl alcohol and sodium formate

B. sodium benzoate and methyl alcohol

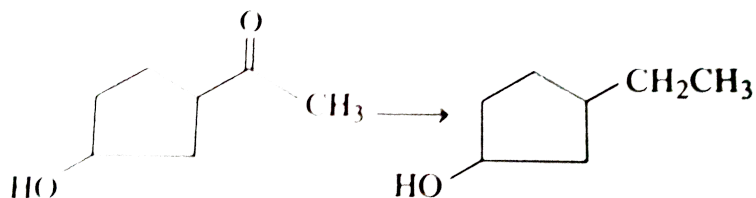
C. sodium benzoate and sodium formate

D. benzyl alcohol and methyl alcohol

Answer: A

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14. The appropriate reagent for the following transformation:



A. Zn(Hg), HCl

B. $\text{NH}_2\text{NH}_2, \text{OH}^-$

C. H_2 / Ni

D. NaBH_4

Answer: B

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15. which of the following has the most acidic hydrogen?

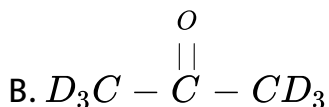
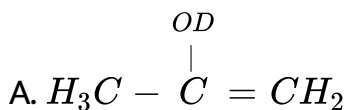
- A. 3 – hexanonoe
- B. 2, 4 – hoxanedione
- C. 2, 5 – heaxnedione
- D. 2, 3 – hoxanedione

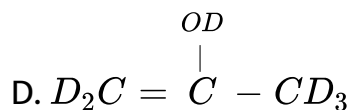
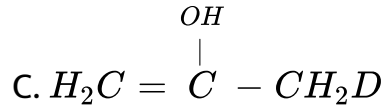
Answer: B



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16. The enol form of acetone after treatment with D_2O gives:

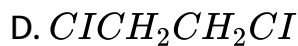
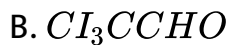




Answer: B

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17. Which of the following will reacts with water ?



Answer: B

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18. Which of the following compounds is oxidised to prepare methyl ethyl ketone?

A. 2 – propanol

B. 1 – butanol

C. 2 – butanol

D. *t* – buty alcohol

Answer: C

19. The compound that will not give iodoform on treatment with alkali and iodine is

- A. acetone
- B. ethanol
- C. diethyl ketone
- D. isopropyl alcohol

Answer: C



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20. The Cannizzaro's reaction is not given by

- A. trimethyl acetaldehyde
- B. acetaldehyde
- C. benzaldehyde
- D. formaldehyde

Answer: B

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21. When acetaldehyde is treated with Fehling's solution, it gives a precipitate of

A. Cu

B. CuO

C. Cu_2O

D. $Cu + Cu_2O + CuO$

Answer: C

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22. A compound that gives a positive iodoform test is

- A. 1-pentanol
- B. 3-pentanone
- C. 2-pentanone
- D. pentanal

Answer: C



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23. The reagent with which both acetaldehyde and acetone react easily is

- A. Tollen's reagent
- B. Schiff's reagent

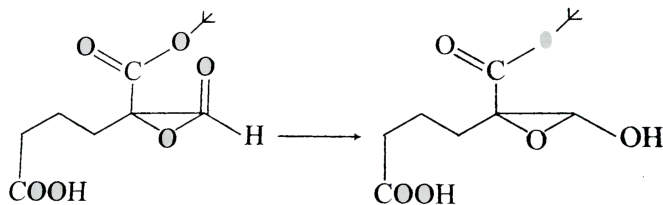
C. Grignard's reagent

D. Fehling's reagent

Answer: C

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24. Reagent (s) which can be used to bring about the following transformation is (are)



A. $LiAlH_4$ in $(C_2H_5)_2O$

B. BH_3 in THF

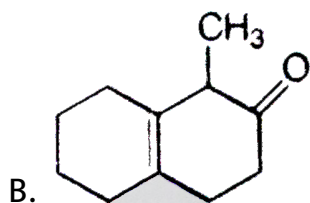
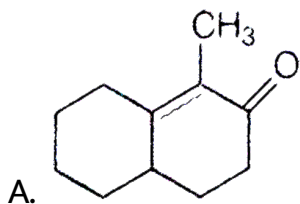
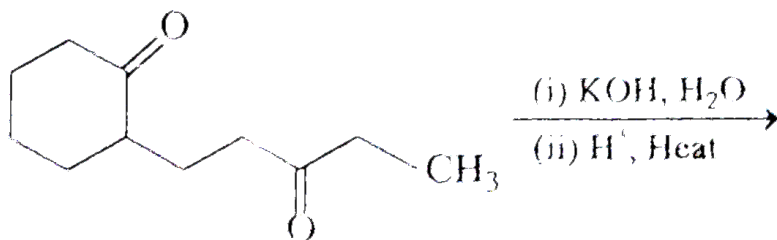
C. $NaBH_4$ "in" C_2H_5OH

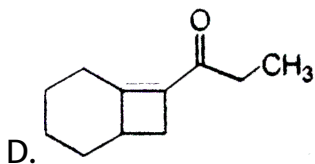
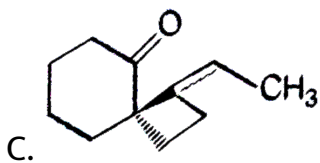
D. Raney Ni / H_2 in THF

Answer: C

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25. The major product of the following reaction is



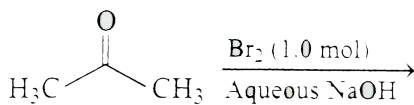


Answer: A

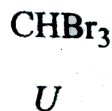
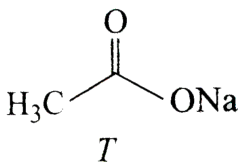
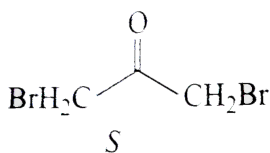
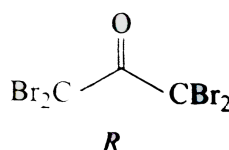
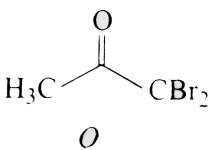
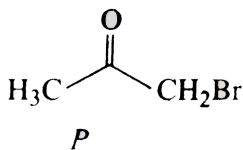
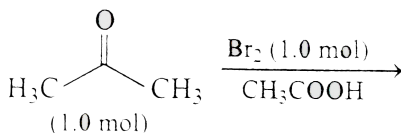
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26. After completion of the reactions (I and II) , the organic compound (s) in the reaction *is / are*

Reaction I



Reaction II



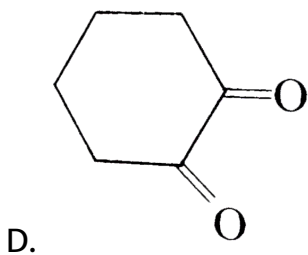
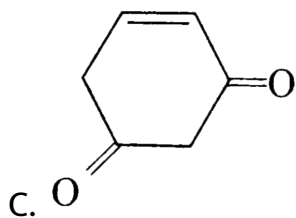
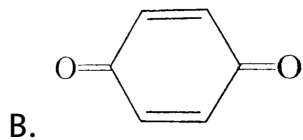
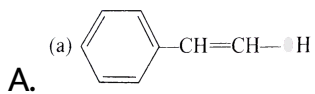
- A. reaction I: *P* and reaction II: *P*
- B. reaction I: *U*, acetone and reaction II: *Q*, acetone
- C. reaction I: *T*, *U* acetone and reaction II: *P*
- D. reaction I: *R*, acetone and reaction II: *S*, acetone

Answer: C



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27. Tautomerism is exhibited by



Answer: A::C::D



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28. A new carbon-carbon bond formation is possible in

- A. Cannizzaro's reaction
- B. Friedel-Crafts's reaction
- C. Clemmensen's reduction
- D. Remier-Tiemann reaction

Answer: B::D



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29. Which of the following will undergo aldol condensation?

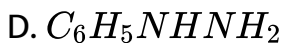
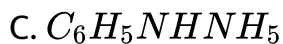
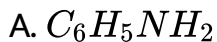
- A. Acetaldehyde
- B. Propanaldehyde
- C. Benzaldehyde

D. Trideutero acetaldehyde

Answer: A::B::D

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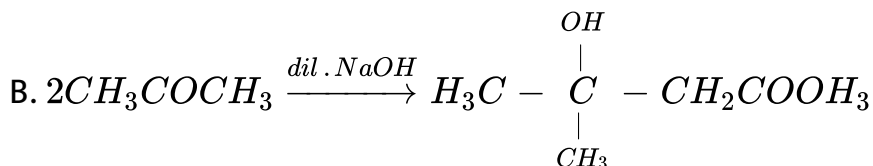
30. Among the following compounds , which will react with acetone to give a product containing



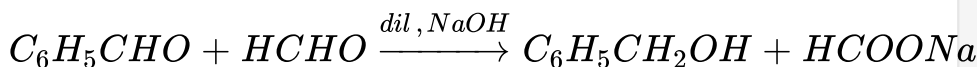
Answer: A::D

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31. Which of the following is an example of aldol condensation?



D.



Answer: A::B

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32. Which of the following compounds will react with ethanolic KCN?

- A. Ethyl chloride
- B. Acetyl chloride
- C. Chlorobenzene
- D. Benzaldehyde

Answer: A::B::D



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33. Which of the following compounds will give a yellow precipitate with iodine alkali?

- A. 2-hydroxy propane

B. Acetophenone

C. Methyl acetate

D. Acetamide

Answer: A::B



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34. Base catalysed aldol condensation occurs with

A. propionaldehyde

B. benzaldehyde

C. 2-methyl propionaldehyde

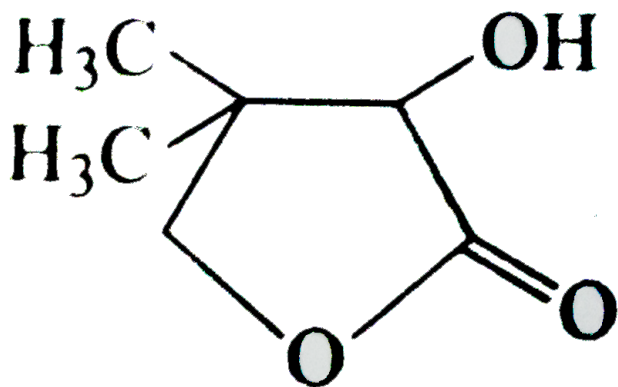
D. 2,2-dimethyl propionaldehyde

Answer: A::C

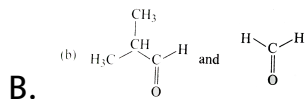
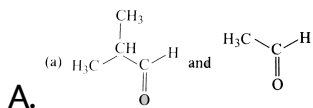


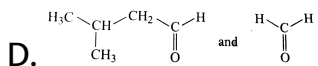
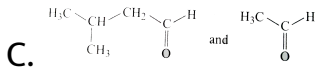
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35. Two aliphatic aldehydes P and Q react in the presence of aqueous K_2CO_3 to give compound R, which upon treatment with HCN provides compound S. On acidification and heating, S gives the product shown below:



The compounds P and Q respectively are

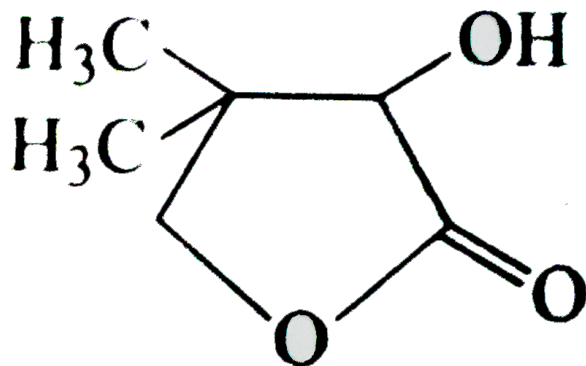




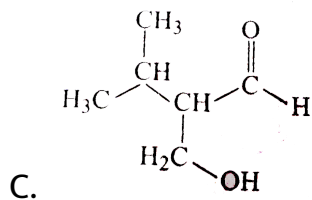
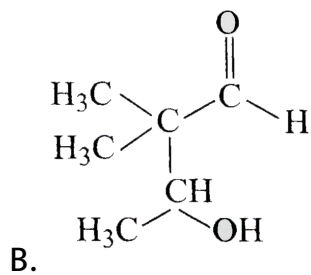
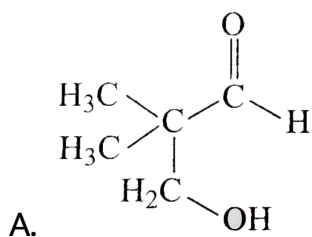
Answer: B

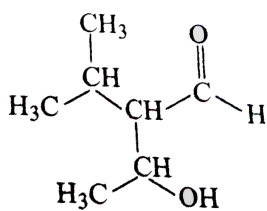
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36. Two aliphatic aldehydes P and Q react in the presence of aqueous k_2CO_3 to give compound R, which upon treatment with HCN provides compound S. On acidification and heating, S gives the product shown below:



The compound R is



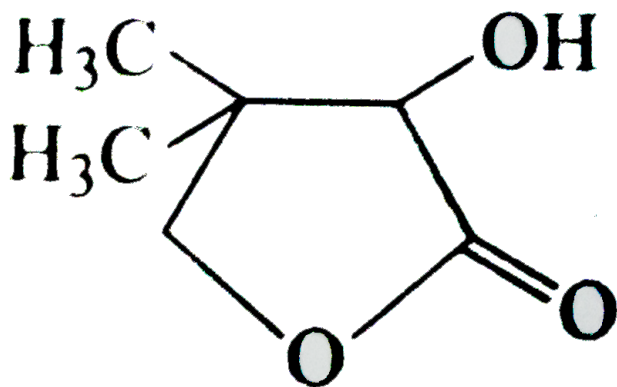


D.

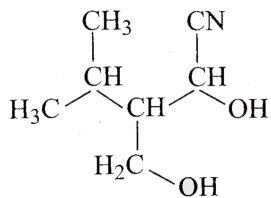
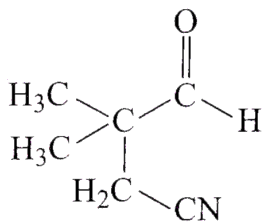
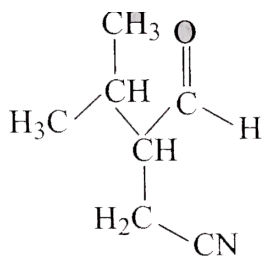
Answer: A

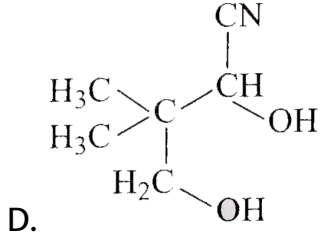
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37. Two aliphatic aldehydes P and Q react in the presence of aqueous K_2CO_3 to give compound R, which upon treatment with HCN provides compound S. On acidification and heating, S gives the product shown below:



The compound S is

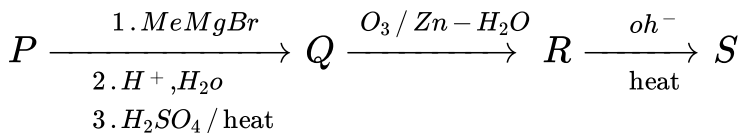




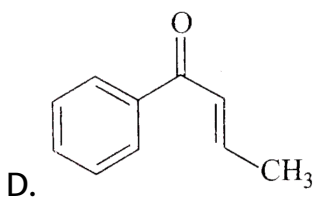
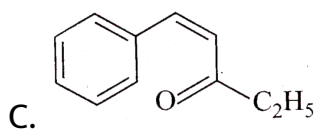
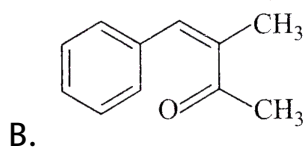
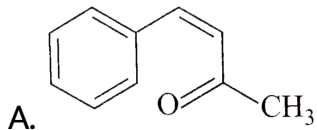
Answer: D

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38. A carbonyl compound P, which gives positive iodoform test, undergoes reaction with MeMgBr followed by dehydration to give olefin Q. Ozonolysis of Q leads to a dicarbonyl compound R, which undergoes intramolecular aldol reaction to give predominantly S.



The structure of the carbonyl compound P, is

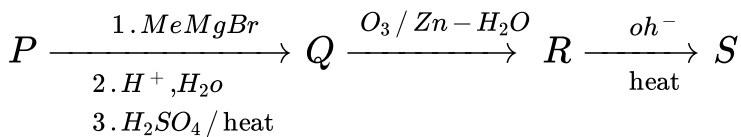


Answer: B

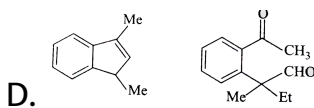
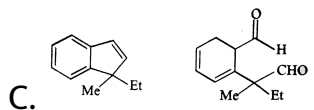
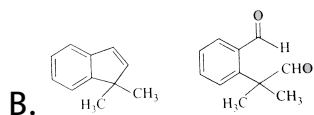
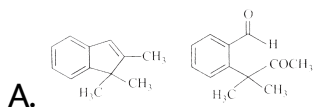
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39. A carbon1 compoundj P, which gives positive idofrom test, undergoes reaction with MeMgBr followed by dehydration to give olefin Q Ozonolysis of Q leads to a dicarbony1 compound R,

which undergoes intramolecular aldol reaction to give predominantly S.



The structures of products Q and R respectively, are

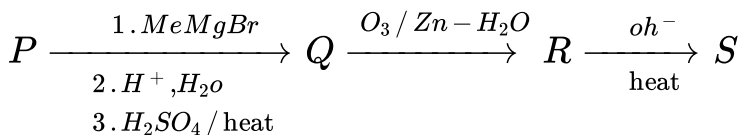


Answer: A

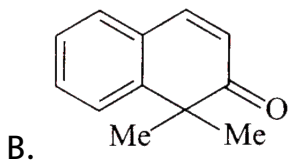
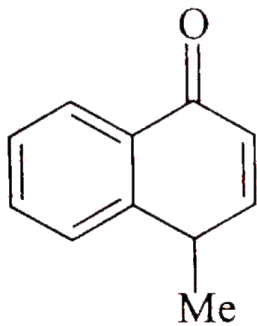


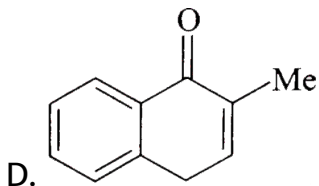
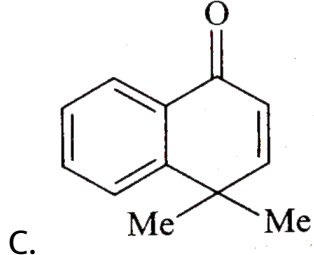
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40. A carbon1 compoundj P, which gives positive idofrom test, undergoes reaction with MeMgBr followed by dehydration to give olefin Q Ozonolysis of Q leads to a dicarbony1 compound R, which undergoes intramolecular aldol reaaction to give predominantly S.



The structure of the product S , is

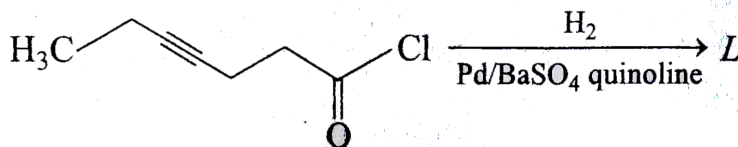




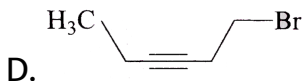
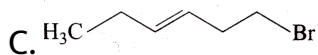
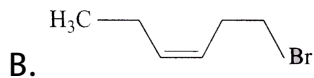
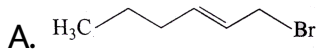
Answer: B

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41. In the following sequence, product I, J and L are formed. K represents a reagent.



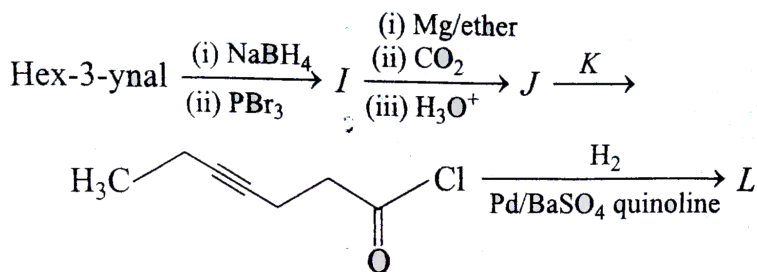
The structure of the product I is



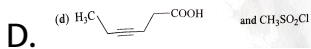
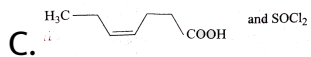
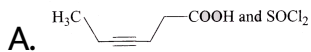
Answer: D

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42. In the following sequence, product I, J and L are formed. K represents a reagent.



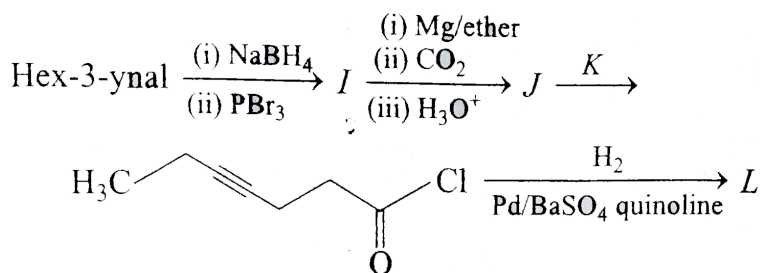
The structure of compound J and K, respectively are



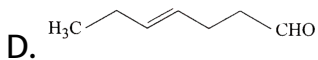
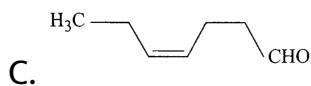
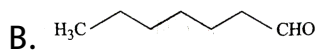
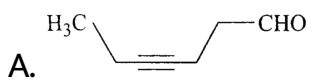
Answer: A

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43. In the following sequence, product I, J and L are formed. K represents a reagent.



The structure of products L is



Answer: C

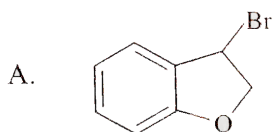


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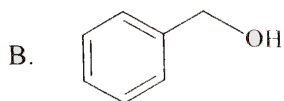
44. Match each of the compounds given in Column I with reaction (s) that they can undergo, given in Column II.

Column I

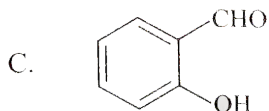
Column II



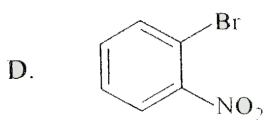
p. Nucleophilic substitution



q. Elimination



r. Nucleophilic addition



s. Esterification with acetic anhydride

t. Dehydrogenation



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45. Match the compounds /ions Column I with their properties /reactions in Column II.

	Column I		Column II
A.	C_6H_5CHO	p.	gives precipitate with 2, 4-dinitrophenylhydrazine
B.	$CH_3C\equiv CH$	q.	gives precipitate with $AgNO_3$
C.	CN^-	r.	is a nucleophile
D.	I^-	s.	is involved in cyanohydrin formation

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46. Fehling's solution A consists of an aqueous solution of copper sulphate while Fehling's B consist of an alkaline solution....

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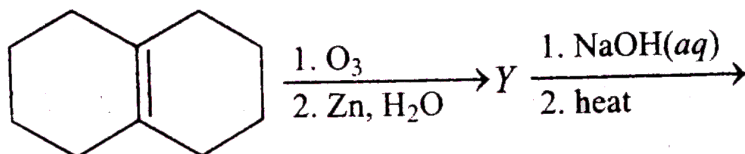
47. The reaction of methyl magnesium iodide with acetone followed by hydrolysis give secondary butanol.

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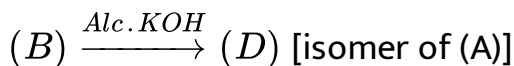
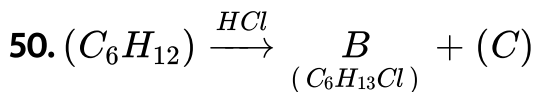
48. Benzaldehyde undergoes aldol condensation in an alkaline medium.

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49. In the scheme given below. The total number of intramolecular aldol condensation products formed from Y is



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$(D) \xrightarrow{Ozonolysis} (E)$ (It gives negative test with Fehling's solution but responds to iodoform test)

$(A) \xrightarrow{Ozonolysis} (F) + (G)$ (Both give positive Tollens test but do not give iodoform test)



Identify A to G.

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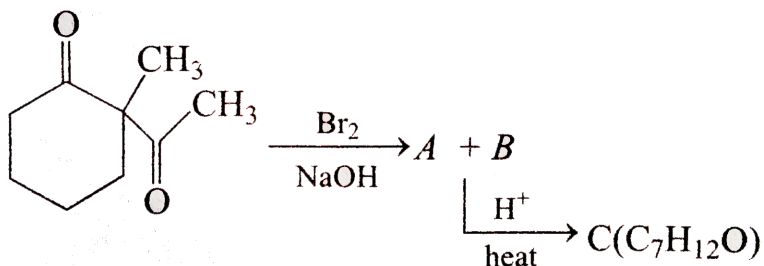
51. A compound $C_9H_7O_2Cl$ exists predominantly in enol form (A) and also in keto form (B). On oxidation with $KMnO_4$ it gives m-chlorobenzoic acid as one of the products. Identify the compounds (A) and (B).

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52. An alkane (A) $C_{16}H_{16}$ on ozonolysis gives only one product (B) C_8H_8O . Compound (B) on reaction with $NaOH / I_2$ yields sodium benzoate. Compound (B) reacts with KOH / NH_2NH_2 yielding a hydrogen (C) $C_8H(10)$. Write the structures of compounds (B) and (C). Based on this information two their structures and identify the isomer which on catalytic hydrogenation ($H_2 / Pd - C$) gives a racemic mixture.

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53. Identify A, B and C, and give their structures.





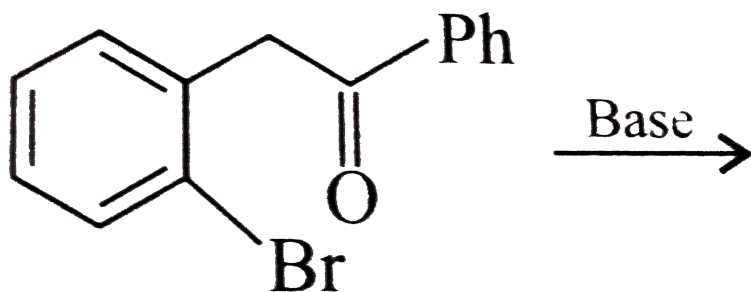
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54. An organic compound (A) ($C_6H_{10}O$) on reaction with CH_3MgBr followed by acid treatment gives compound (B). The compound (B) on ozonolysis gives compound (C), which in the presence of a base gives 1-acetyl cyclopentene (D). The compound (B) on reaction with HBr gives compound (E). Write the structures of (A), (B), (C), (D), and (E). Show how (D) is formed from (C).



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55. What would be the major product in the following reaction ?



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56. (a) Compound A (C_8H_8O) on treatment with NH_2OH . HCl gives B and C. B and C rearrange to give D and E, respectively on treatment with acid. B, C, D and E are all isomers of molecular formula (C_8H_9N) When D is boiled with alcohol KOH an oil F (C_6H_7N) separates out F, reacts rapidly with CH_3COOI to give back D. On the other hand. E on boiling with alkali followed by acidification gives a white solid G ($C_7H_6O_2$). Identify A – G

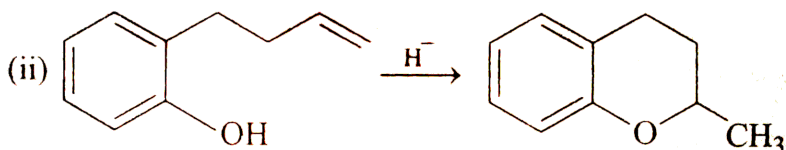
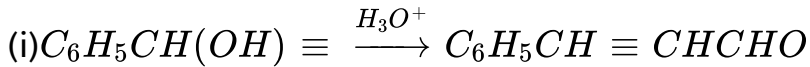
(b) Carry out the following transformation in not more than three steps. 1-butyne "to" 2-pentanone

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57. An ester $A(C_4H_8O_2)$, on treatment with excess of methyl magnesium bromide followed by acidification, gives an alcohol B as the sole organic product. Alcohol B on oxidation with $NaOCl$ followed by acidification gives acetic acid. Deduce the structures of A and B . Show the reactions involved.

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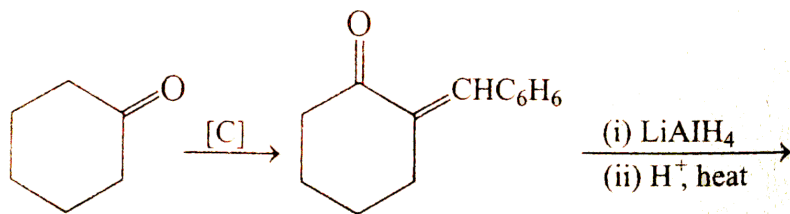
58. Write the intermediate steps for each of the following reactions



(ii)

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59. Complete the following reactions with appropriate structures of products //reagents.



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60. An aldehyde (A) ($C_{11}H_8O$), which does not undergo self aldol condensation, gives benzaldehyde and 2 mol of (B) on

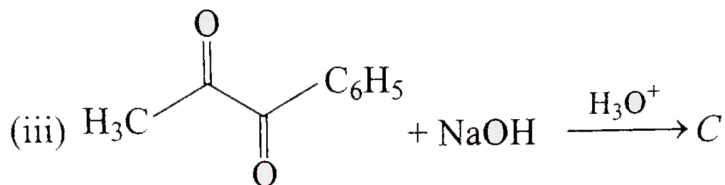
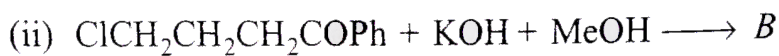
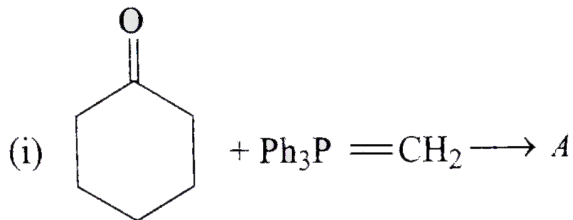
ozonolysis. Compound (B) on oxidation with silver ion gives oxalic acid. Identify the compounds (A) and (B).

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61. Acetophenone on reaction with hydroxylamine lamine hydrochloride can produce two isomeric oximes. Write structures of the oxime.

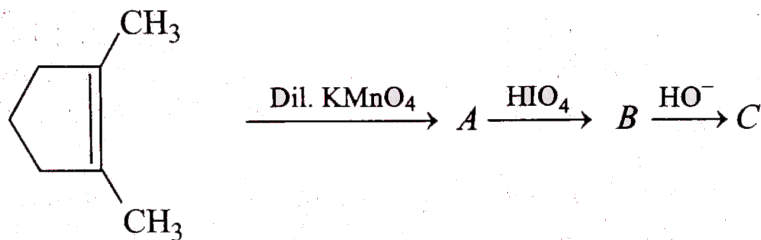
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62. Complete the following, gives the structures of the principal organic prodcuts.



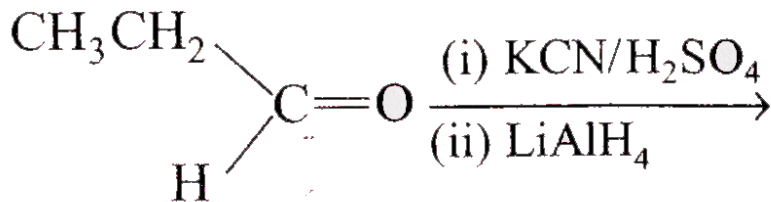
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63. Suggest appropriate structures for the missing compounds. (the number of carbon atoms remains the same throughout the reaction)



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64. Complete the following reaction with appropriate structure:



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65. Complete the following reaction with appropriate structure.



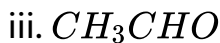
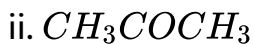
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66. Write the structure of major organic product expected from the following reaction.



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67. Arrange the following in the increasing order of expected enol content.



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68. Give reason in one or two sentence:

Iodoform is obtained by the reaction of acetone with hypiodite but not with iodide



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69. A ketone A, which undergoes haloform reaction, gives compound B on reduction B on heating with sulphuric acid gives compound C, which forms mono-ozonide D. D on hydrolysis in the presence of zinc dust gives only acetaldehyde. Identify A, B and C. Write down the reaction involved.



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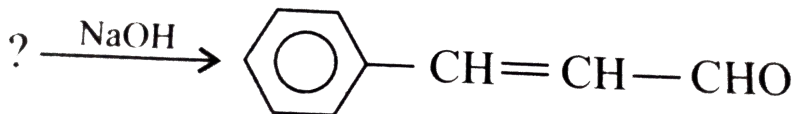
70. Answer the following with suitable equation wherever necessary

(i) suggest a reagent to distinguish acetaldehyde from acetone.

(ii) what happens when excess chlorine is passed through boiling toluene in the presence of sunlight?

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71. Complete the following with appropriate structures.



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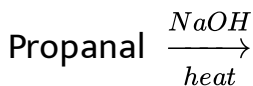
72. In what manner the following transformation might be carried out (in not more than six steps) ? 'Benzaldehyde to cyanobenzene'

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73. Give reasons in one or two sentences for the following: 'Hydrazones of aldehydes and ketones are not prepared in highly acidic medium'.

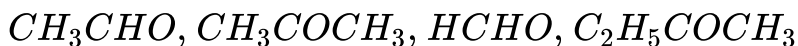
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74. Write down the main product of the following reaction.



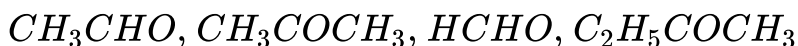
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75. Arrange the following in the order of their increasing reactivity towards HCN:



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76. Arrange the following in the order of their increasing reactivity towards HCN:



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77. Show with balanced equation what happens when the following are mixed: 'Chloral is heated with aqueous hydroxide.'

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78. An alkane A on ozonolysis yields acetone and an aldehyde. The aldehyde is easily oxidised to an acid B. When B is treated with bromine in presence of yields a compound C which on hydrolysis gives a hydroxyle acid D. This acid can also be obtained from acetone by the reaction with hydrogen cyanide followed by hydrolysis. Identify the compounds A, B C and D.

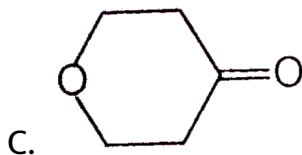
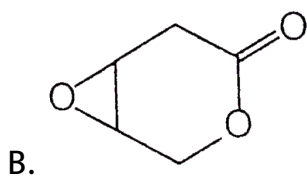
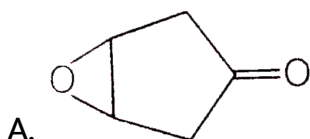
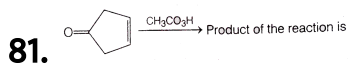
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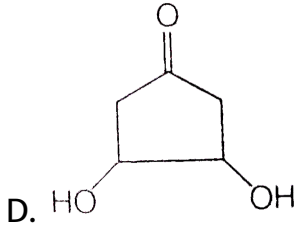
79. Outline the reaction sequence for the conversion of methanal to ethanal (the number of steps should not be more than three).

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80. Write the structural formula of the main organic product formed when methanal reacts with ammonia.

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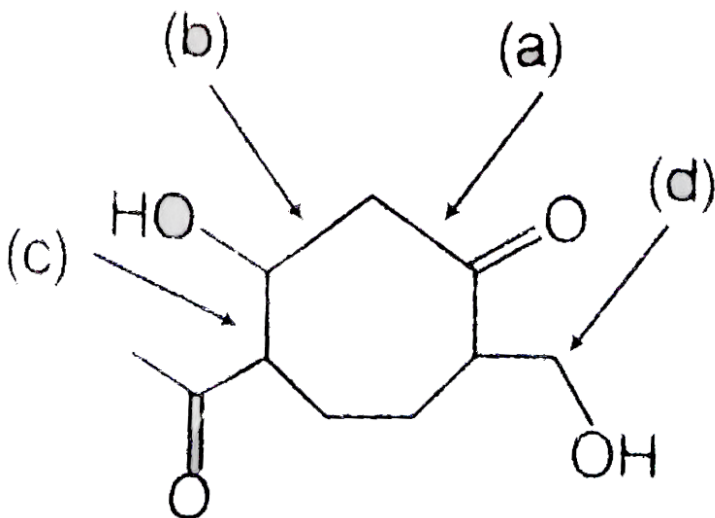




Answer: B

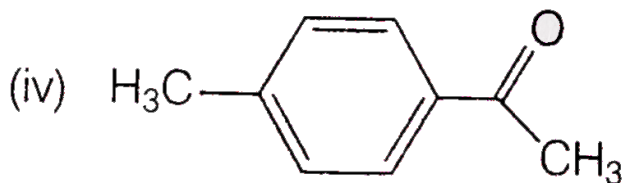
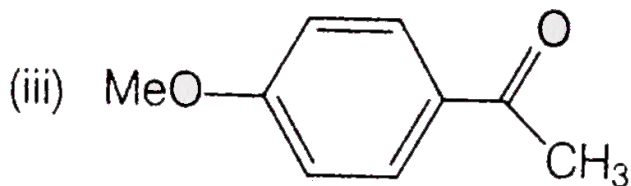
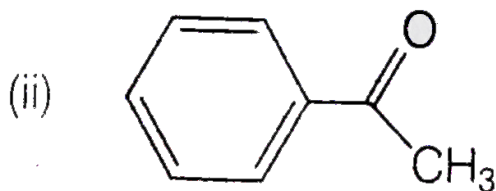
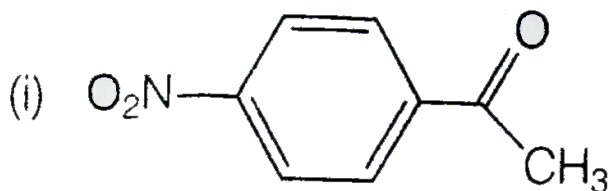
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82. Which carbon-carbon bond could not be formed by an aldol condensation reaction?

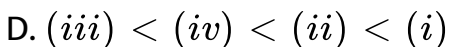
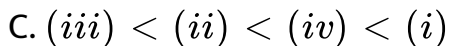
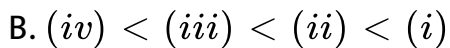


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83. Arrange the following in the increasing order of reactivity towards aldol condensation reaction



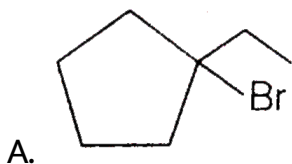
A. (i) < (ii) < (iii) < (iv)

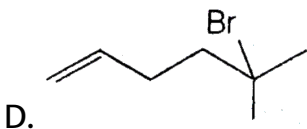
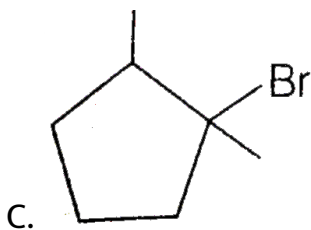
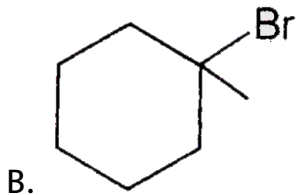


Answer: D

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84. Compound $A(C_7H_{13}Br)$ is a tertiary bromide. On treatment with sodium ethoxide in ethanol. A is treatment to a hydrocarbon B. Ozonolysis of B followed up by work up with $Zn - H_2O$ gives 6-oxoheptanal as the only product The most likely structure of A is





Answer: B

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85. you have two $C_6H_{10}O$ ketones, I and II. Both are optically active, but is racemised by treatment with base then Wolff-Kishner reduction of both ketones gives the same

achiral hydrocarbon, formula C_6H_{12} . What reasonable structures may be assigned to I and II?

- A. I is 3-methyl-4-pentanone, II is 4-methyl-1-penten-3-one
- B. I is 2-methylcyclopentanone, II is 3-methylcyclopentanone
- C. I is 3-methylcyclopentanone, II is 2-methylcyclopentanone
- D. I is 2-ethylcyclobutanone, II is 3-ethylcyclobutanone

Answer: B



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86. Treatment of ($C_5H_{12}O$) with concentrated sulphuric acid results in the formation of three alkenes in differing yields. Also, A forms a yellow precipitate on treatment with $NaOH$ / 12

. A turn colour of acidified dichromate solution to blue green forming a new organic compound B which also forms yellow precipitate on treatment with $NaOH$ / 12 The most likely name of A is / are

A. 3-methyl-2-butanol

B. 3-methylbutanol

C. 2-pentanol

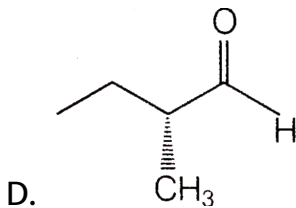
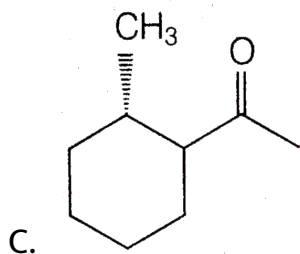
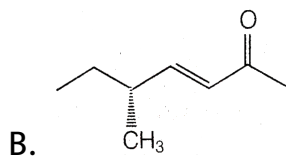
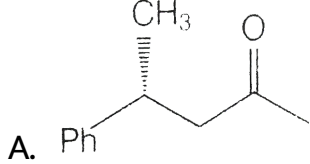
D. either b or c

Answer: A::C



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87. The carbonyl compound(s) that will undergo racemisation on treatment with aqueous KOH is / are

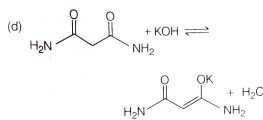
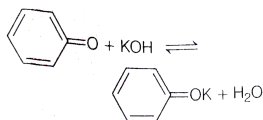
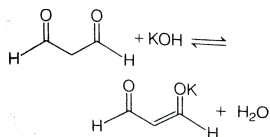
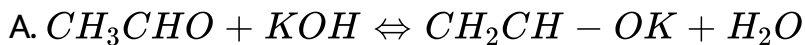


Answer: B::D



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88. Which of the following reactions *has/have* equilibrium constant (k_C) greater than one?

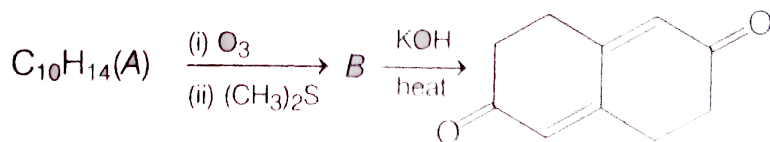


Answer: B::C

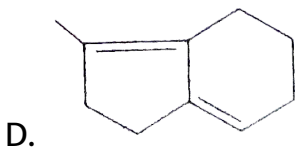
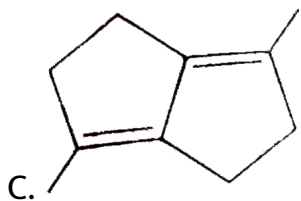
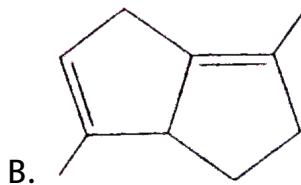
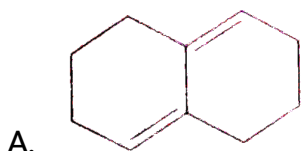


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89. Consider the following reaction to answer the next three question:



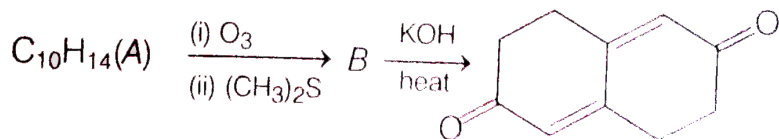
The most likely structure of A is



Answer: C

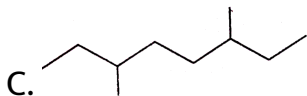
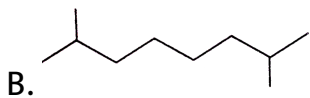
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90. Consider the following reaction to answer the next three question:

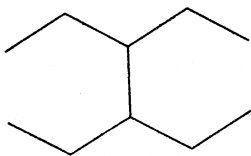


if B is heated with $\text{Zn}(\text{Hg})$ in concentrated HCl solution the product formed is

A. Dance



D.



Answer: A



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91. Assertion When pure $d - 1(+) - 2$ methylbutanal is treated with dilute H_2SO_4 it is racemised completely.

Reason It's α carbon is chiral which undergo keto-enol tautomerism in the presence of acid catalyst.

A. Both assertion and reason are correct and reason is the correct explanation of the assertion.

B. Both assertion and reason are correct butn reason in not the correct explanation of assertion.

C. Assertion is correct but reason is incorrect.

D. Assertion is incorrect but reason is correct.

Answer: A



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92. Assertion When benzaldehyde (C_6H_5CHO) is treated with concentrated NaOD solution in D_2O Cannizzaro reaction occurs but no C-D (bonds with deuterium) is formed.

Reason Cannizzaro reaction involves hydride transfer mechanism.

A. Both assertion and reason are correct and reason is the correct explanation of the assertion.

B. Both assertion and reason are correct butn reason in not the correct explanation of assertion.

C. Assertion is correct but reasn is incorrect.

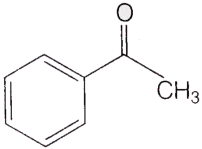
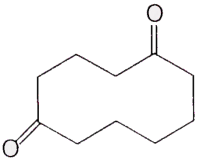
D. Assertion is incorrect but reason is correct.

Answer: A



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93. Match the carbonyls from left column with their characteristics from right column :

	Reaction type	Halides
A.		p. Gives just one aldol only
B.	$\text{H}_3\text{C}-\overset{\text{O}}{\parallel}{\text{C}}-\text{CH}_3$	q. Gives yellow precipitate with I_2/NaOH
C.	$\text{CH}_3\text{CH}_2-\overset{\text{O}}{\parallel}{\text{C}}-\text{CH}_2\text{CH}_3$	r. Produces isomeric oximes with HONH_2
D.		s. Gains more than 4 u in molar mass on treatment with $\text{NaOD}/\text{D}_2\text{O}$
		t. Gives more than one aldol



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94. if ethanedial ($\text{HOC}-\text{COH}$) is treated with excess of HCN (aq) followed by hydrolysis of product results in diacids. How many different diacids would be formed ?



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95. If $CH_2D - CHO$ is treated with dilute alkaline solution how many different aldols (excluding stereoisomers) are expected ?



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Integer Answer Type Questions

1. Consider all possible isomeric ketons including stereoisomers of MW = 100 All these isomers are independently reacted with $NaBH_4$ The total of number of ketones that gives a racemic product (S) *is / are*



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