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## CHEMISTRY

## BOOKS - GR BATHLA \& SONS CHEMISTRY (HINGLISH)

## AMINES

## Level 1 (Q.1 To Q.25)

1. Acetamide is treated separately with the following reagens. Which one of these given methylamine?
A. $P C I_{5}$
B. Sodalime
C. $\mathrm{NaOH}+\mathrm{Br}_{2}$
D. Hot concentrated $\mathrm{H}_{2} \mathrm{SO}_{4}$

## Answer: C

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2. Reaction of $\mathrm{RCONH}_{2}$ with a mixture of $\mathrm{Br}_{2}$ and KOH gives $\mathrm{RNH}_{2}$ as the main product. The intermediates involved in the reaction are:
A. $R-\stackrel{\stackrel{O H}{\mid}}{C}=N-B r$
B. $R-\mathrm{NHBr}$
C. $R-N=C=O$
(d) $\left.R-C-N{ }^{\prime \prime}\right\rangle_{\mathrm{Br}}^{\mathrm{Br}}$
D.

## Answer: C

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3. Which is the best method of preparing $2^{\circ}$ amine ?
A. $\mathrm{CH}_{3} \mathrm{CI}+\mathrm{NH}_{3} \rightarrow$
B. $\mathrm{CH}_{3} \mathrm{CI} \xrightarrow{\mathrm{KCN}} \xrightarrow{\mathrm{Sn} / \mathrm{HCI}}$
C. $\mathrm{CH}_{3} \mathrm{CI} \xrightarrow{\mathrm{KCN}} \xrightarrow{\mathrm{LiAIH}_{4}}$
D. $\mathrm{CH}_{3} \mathrm{NH}_{2} \xrightarrow[\Delta]{\mathrm{CHCI}_{3} / \mathrm{KOH}} \xrightarrow{\mathrm{Sn} / \mathrm{HCI}}$

Answer: D
4. Ethyl cyanide (A) can be converted to ethyl amine (B) by:
A. $A \xrightarrow{S n / H C I} B$
B. $A \xrightarrow{\mathrm{H}_{3} \mathrm{O}^{+}} \xrightarrow{\mathrm{NH}_{3} / \Delta} \xrightarrow{\mathrm{KBrO} / \Delta} B$
C. $A \xrightarrow{\mathrm{LiAIH}_{4}} B$
D. both (a),(c) are correct

## Answer: B

## D Watch Video Solution

5. In Gabriel synthesis, amine is always:
A. alphatic primary amine
B. aliphatic secondary amine
C. aromatic primary amine
D. aromatic secondary amine

## Answer: A

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6. In Gabriel synthesis, halide may be:
A. benzyl halide
B. allyl halide
C. both (a), (c) are correct
D. tertiary alkyl halide

## Answer: C

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7. In the given reaction sequence
$\mathrm{C}_{6} \mathrm{H}_{5}-\mathrm{CH}_{2}-\mathrm{NH}_{2} \xrightarrow[\Delta]{\mathrm{CHCl}_{3} / \mathrm{Alc.KOH}}[X] \xrightarrow{\mathrm{H}_{2} \mathrm{O} / \mathrm{NaOH}}[Y],[Y]$
will be:
A. $C_{6} H_{5}-C N$
B. $C_{6} H_{5} N C$
C. $\mathrm{C}_{6} \mathrm{H}_{5}-\mathrm{CH}_{2}-\mathrm{NH}_{2}$
D. $\mathrm{C}_{6} \mathrm{H}_{5}-\mathrm{CH}_{2} \mathrm{OH}$

## Answer: C

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8. Predict the nature of the product
$\mathrm{PC}_{6} \mathrm{H}_{5} \mathrm{CONH}_{2} \xrightarrow{\mathrm{Br}_{2} / P D^{-}} P$
A. $\mathrm{C}_{6} \mathrm{H}_{5} \mathrm{NH}_{2}$
B. $C_{6} H_{5} \mathrm{NHD}$
C. $C_{6} H_{5} N D_{2}$
D. All of these

Answer: C
9. Which of the following statement is not correct?
A. Aliphatic amines are stronger bases than ammonia
B. Aromatic amines are stronger bases than ammonia
C. The alkyl group in alkyl ammonium ion more
stabilizes the ion relative to the amine
D. The aryl group in aryl ammonium ion less stabilizes the ion relative to the amine

## Answer: B

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10. The correct sequence regarding base strength of aliphatic amines in aqueous solution is:
A. $\mathrm{R}_{3} \mathrm{~N}>\mathrm{R}_{2} \mathrm{NH}>\mathrm{RNH}_{2}>\mathrm{NH}_{3}$
B. $\mathrm{R}_{2} \mathrm{NH}>\mathrm{RNH}_{2}>\mathrm{R}_{3} \mathrm{~N}>\mathrm{NH}_{3}$
C. $R_{2} \mathrm{NH}>\mathrm{R}_{3} \mathrm{~N}>\mathrm{RNH}_{2}>\mathrm{NH}_{3}$
D. $\mathrm{RNH}_{2}>\mathrm{R}_{2} \mathrm{NH}>\mathrm{R}_{3} \mathrm{~N}>\mathrm{NH}_{3}$

## Answer: B

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11. Decreasing order of basicity of the three isomers of nitro aniline is:
A.p-nitroaniline $>o$-nitroaniline $>m$-nitroaniline
B. p-nitroaniline $>m$-nitroaniline $>o$-nitroalinine
C. m-nitroanilin
nitroaniline
D. m-nitroaniline $\quad>o-\quad$ nitroaniline $\quad>p-$
nitroaniline

Answer: C

D Watch Video Solution
12. Strongest base is:
A.

(b) ${ }_{\mathrm{H}_{2} \mathrm{~N}^{\prime}}^{\mathrm{H}_{2} \mathrm{~N}} \mathrm{C}=\stackrel{\text { + }}{\mathrm{NH}}{ }_{2}$
B. $\quad \mathrm{H}_{2} \mathrm{~N}^{\prime}$
C. ${ }^{(\mathrm{H})}{ }_{2}^{\mathrm{H}_{2} \mathrm{~N}}$ C=O
D.
(d) $\mathrm{H}_{2} \mathrm{~N}_{2} / \mathrm{C}-\mathrm{OH}$

Answer: A

## D Watch Video Solution

13. Which is the best leaving group ?
A. $-\stackrel{\oplus}{N} \equiv N$
B. $O H^{-}$
C. $\mathrm{NH}_{2}^{-}$
D. $\mathrm{CH}_{3} \mathrm{COO}^{-}$

Answer: A

## - Watch Video Solution

14. Which is most volatile?
A. $\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{NH}_{2}$
B. $\left(\mathrm{CH}_{3}\right)_{3} \mathrm{~N}$

D. $\mathrm{CH}_{3} \mathrm{OH}$

Answer: B
15. Which one of the following is used as phase transfer catalyst?
A. Primary amine
B. Quaternary ammonium salt
C. Tertiary nitroalkane
D. Tertiary amine

## Answer: B

## - Watch Video Solution

16. Which of the following is most basic?
(a)

A.
(b)

B.

H
(c)

C.

D.


Answer: C

## - Watch Video Solution

17. Predict about the relative boiling point of the following two amines.
(I)

(II)

A. Boiling point of $I>I I$
B. Boling point of $I I>I$
C. Both should have equal boiling points
D. It can't be predicted

## Answer: B

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18. Carbylamine test is performed in alcoholic $K O H$ by
heating a mixture of:
A. chloroform and silver silver powder
B. chloroform and a primary amine
C. an alkyl halide and a primary amine
D. an alkyl cyanide and a primary amine

## Answer: B

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19. Which of the following compounds is an imine ?

D.
(d)


## Answer: B

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20. Which of the following statements is not correct?
A. Primary amines show intermolecular hydrogen bonding.
B. Secondary amines show intermolecular hydrogen bonding
C. Tertiary amines show intermolecular hydrogen bonding
D. Amines have lower boiling points as compared to those of alcohols and carboxylic acids of comparable molar masses.

## Answer: C

## - Watch Video Solution

21. Which of the following amines form $N$ - nitroso derivative when treated with $\mathrm{NaNO}_{2}$ and HCI ?
A. $\mathrm{CH}_{3} \mathrm{NH}_{2}$
B.

C.

(d)
D.

## Answer: C

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22. Hinsberg's reagent is:
A. phenylisocyanide
B. benzensulphonyl chloride
C. p-toluenesulphonic acid
D. o-dichlorobenzene

Answer: B
23. Which of the following compounds is an emamine?

$$
\begin{aligned}
& \text { A. } \\
& \text { B. }{ }_{\text {(b) }>\mathrm{N}-\mathrm{CH}_{3}}^{\text {C. } P h-N=N-P h} \text { D. } P h-N=N-H
\end{aligned}
$$

Answer: A

## ( Watch Video Solution

24. Complete the following
reaction
$\square-\mathrm{CH}_{2} \stackrel{\stackrel{\mathrm{~N}}{+}}{\mathrm{O}^{-}}\left(\mathrm{CH}_{3}\right)_{2} \xrightarrow{\Delta} \longrightarrow \mathrm{CH}_{2}+\left(\mathrm{CH}_{3}\right)_{2} \mathrm{NOH}$. This is called :
A. Hofmann elimination
B. Cope reaction
C. Saytzeff reaction
D. Carbyl amine reaction

## Answer: B

## D Watch Video Solution

25. Cope reaction is:
A. $S_{N} 1$ intramolecular
B. $S_{N} 2$ intramolecular
C. $E_{1}$ intramolecular
D. $E_{c}$ or $E_{i}$ intramolecular

## Answer: D

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## Level 1 (0. 26 To Q.41)

1. Which of the following is Hofmann mustard oil reaction?
A. Reaction of primary amine with $\mathrm{CHCI}_{3}$
B. Reaction of primary amine with $\mathrm{CHCI}_{3}+\mathrm{KOH}$
C. Reaction of primary amine with $\mathrm{CS}_{2}+\mathrm{HgCI}_{2}$

## D. Reaction of armoatic amine with iodoform

## Answer: C

## - Watch Video Solution

## 2.



B.

(c)

D.
(d)


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3. In the Hofmann-Bromamide rearrangement intermediate compounds are:
A. $R-\mathrm{CONHBr}$
B. $\left[\begin{array}{c}\stackrel{O}{\|} \\ R-\bar{N}-\bar{N}-B r\end{array}\right] N a^{+}$
C. $R-N=C=O$
D. all of these

## Answer: D


4.

HCI

Product
a)
A.


C.


Answer: C
5. Which of the following compounds can form alcohol with $\mathrm{NaNO}_{2} / \mathrm{HCl}$ ?
A.
(a)

B.
(b) $\mathrm{H}_{3} \mathrm{C}-\stackrel{\mathrm{CH}_{\mathrm{H}}^{\mathrm{NH}}}{\mathrm{CH}_{3}}$
C. $\mathrm{CH}_{3}-\mathrm{CH}_{2}-\mathrm{NH}_{2}$
D. All of these

Answer: D
6. Which of the following will not react with $C S_{2}$ ?
A. $\mathrm{C}_{6} \mathrm{H}_{5}-\mathrm{NH}_{2}$
B. $\mathrm{CH}_{5}-\mathrm{NH}-\mathrm{CH}_{3}$
c.
(c) $\quad \mathrm{N}-\mathrm{H}$
D.
(d) $\square \mathrm{N}-R$

Answer: D

## - Watch Video Solution

$$
\left.\begin{array}{l}
\text { 7. } \begin{array}{c}
\text { In } \\
\text { the }
\end{array} \text { given }
\end{array} \begin{array}{l}
\text { reaction } \\
\mathrm{CH}_{3}-\mathrm{CH}_{2}-\mathrm{NH}-\mathrm{CH}_{2}-\mathrm{CH}_{2}-\mathrm{CI} \xrightarrow\left[(\text { ii) } \mathrm{AgOH} / \Delta]{(i) \mathrm{CH}_{3} I(\text { excess })}\right.
\end{array} \mathrm{X}\right] .
$$

is the major product, $[\mathrm{X}]$ will be:
A. $\mathrm{CH}_{2}=\mathrm{CH}_{2}$
B. $\mathrm{CH}_{2}=\mathrm{CHCI}$
C. 1:1 ratio of (a) and (b)
D. $\mathrm{CH}_{3}-\mathrm{CH}_{2}-\mathrm{CI}$

## Answer: B

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8. Predict the nature of $P$ in the following reaction:
$\mathrm{Me}_{3} \mathrm{CCH}_{2} \mathrm{NH}_{2} \xrightarrow{\mathrm{HOMO}} P$ (main product)
A. $\mathrm{Me}_{3} \mathrm{CCH}_{2} \mathrm{OH}$
B. $\mathrm{Me} \mathrm{e}_{2} \mathrm{CCH}=\mathrm{CH}_{2}$
C. $\mathrm{Me}_{2} \mathrm{C}(\mathrm{OH}) \mathrm{C}_{2} \mathrm{H}_{5}$
D. $\mathrm{Me}_{3} \mathrm{CCH}_{2} \mathrm{NH}(\mathrm{NO})$

## Answer: C

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9.
is:
A.
(a)

B.
(b)

(c)

C. H
(d)

D.

H

Answer: C

## D Watch Video Solution


(a)

A.
B.
(b)


## (c) <br>  <br> C.

D. none of these

Answer: C

## D Watch Video Solution

11. Which of the following is an enamine?
A.
(a)

B.
(b)

C.
(c) $\square-\mathrm{NH}_{2}$
D.
(d)


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12. Which of the following amines will react with cyclohexanone to give enamine?
A. $\mathrm{CH}_{3} \mathrm{NH}_{2}$
B.

(c) $\|_{N}$
c. $\quad \mathrm{H}$
(d)
D.

Answer: D

## (D) Watch Video Solution

13. The intermediates obtained in the reaction


NaN 3
$R-\mathrm{NH}_{2}$ are:
Heat
A. $R-\stackrel{\|}{C}-\bar{N}-\stackrel{\oplus}{N} \equiv N$
B. $R-N=C=O$
C. $R-C N O$
D. none of these

Answer: A::B

## D Watch Video Solution

14. Compound $[X] C_{4} H_{11} N$ reacts with p-toluence sulphonyl chloride in aqueous $N a O H$ to give a solid. The compound $[\mathrm{X}]$ is:
A. $\mathrm{CH}_{3}-\mathrm{CH}_{2}-\mathrm{CH}_{2}-\mathrm{CH}_{2}-\mathrm{NH}_{2}$

$$
\begin{aligned}
& \text { B. } \mathrm{H}_{3} \mathrm{C}-\underset{\text { | }}{\mathrm{CH}}-\mathrm{CH}_{2}-\mathrm{NH}_{2} \\
& \text { C. } \mathrm{CH}_{3} \mathrm{CH}_{2}-\mathrm{NH}-\mathrm{CH}_{2}-\mathrm{CH}_{3} \\
& \text { D. } \mathrm{H}_{3} \mathrm{C}-\mathrm{CH}_{2}-\underset{\substack{\mathrm{CH}}}{\mathrm{~N}}-\mathrm{CH}_{3}
\end{aligned}
$$

## Answer: D

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15. Cyclohexanol can be converted into cyclohexylamine by following two routes. Which of the following methods is expected to give good yield of cyclohexyalmine?
A.
B.

C. both are equally suitable
D. neither of the two

Answer: A

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16. Which of the following will give unsymmetrical disubstituted urea after reaction with $\mathrm{CH}_{3} \mathrm{NH}_{2}$ ?
A. $C O C I_{2}$
B. $\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{NCS}$
C. $\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{NCO}$
D. all of these

## Answer: C

## D Watch Video Solution

## Level 2 (0.1 To Q.25)

1. Which of the following compounds is an amine?
(a)

A.

C.

H
(d)

H

## Answer: C

## - Watch Video Solution

2. Which of following compounds exists as non-resolvable racemic mixture?
(a)

$\mathrm{NH}_{2}$
A.
(b)

H
B.
(c) $\mathrm{NH}_{2}$
D.


Answer: D
(D) Watch Video Solution
3. Which of the following compounds loses optical activity due to pyramidal inversion?
(a)
A.

$\mathrm{NH}_{2}$
(b)

B.

(c)


OH
C.
(d)


Answer: D

$$
\widehat{\mathrm{Br}}^{\mathrm{AgCN}} A \xrightarrow{\mathrm{H}_{2}, \mathrm{Ni}} B
$$

4. 

The final product (B) is:
A.
(a) $\sim \mathrm{NH}_{2}$
(b) $\mathrm{N}^{\prime}$

H
B.
C.

(d) $\Omega \mathrm{NH}_{2}$
D.

Answer: B
5.


The end product B of the above reaction is:
(a) $\mathrm{Ph} \mathrm{NH}_{2}$
A.

B.

(c)

C.


H
D.

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6. $\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{Br} \xrightarrow{\mathrm{AgCN}} A \xrightarrow[\substack{\oplus \\ H_{3} O}]{\mathrm{NaOH}, \Delta} B,(B)$ is:
A. $\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{NHCH}_{3}$
B. $\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{NH}_{2}$
C. $\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{NH}_{2}$


Answer: C
7. Complete the following reaction $\mathrm{NH}_{2}$

$$
\xrightarrow[\Delta]{\mathrm{CHCl}_{3}, \mathrm{KOH}} X \xrightarrow[\mathrm{H}_{2} \mathrm{O}]{\mathrm{LiAlH}_{4}} Y ;(Y) \text { is : }
$$

A. $\mathrm{Ph}-\underset{\mathrm{l}}{\mathrm{N}} \mathrm{N}-\mathrm{CH}_{3}$
B. $\mathrm{Ph}-\mathrm{CH}_{2}-\mathrm{NH}_{2}$

D. $P h-\stackrel{\oplus}{N} \equiv \stackrel{\ominus}{C}$

Answer: A

- Watch Video Solution

8. The major product formed in the reaction:

$p-T s O H$ (Trace)
$C_{6} H_{6} . \Delta$
A.
(a)

(b)

B.

(c)

C.


Answer: A

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9.

The final
product $(\mathrm{Y})$ is:
(a)

A.
(b) Ph
B.
(c) $\mathrm{Ph}_{\mathrm{N}^{\prime}}$
C.
(d) $\mathrm{Ph}^{\mathrm{N}}$
D.

Answer: B

## - Watch Video Solution

10. Among the following compounds which one will produce a Schiff base on reaction with cyclopentanone?
A.
B.

C.
D.
(d)


## Answer: C

## - Watch Video Solution

11. In which of the following reactions does the amine behaves as an acid?
A. $\left(\mathrm{C}_{2} \mathrm{H}_{5}\right)_{2} \mathrm{NH}+\mathrm{H}_{2} \mathrm{PtCI}_{6}$
B. $\mathrm{CH}_{3} \mathrm{NH}_{2}+\mathrm{H}_{2} \mathrm{O}$
C. $\left(\mathrm{Me}_{2} \mathrm{CH}\right)_{2} \mathrm{NH}+n-\mathrm{C}_{4} \mathrm{H}_{9} \mathrm{Li}$
D. $\left(C_{2} H_{5}\right)_{3} \ddot{N}+B F_{3}$

## Answer: C

## - Watch Video Solution

12. Consider the following sequence of reactions:

$\xrightarrow[\Delta]{\mathrm{CCI}_{4}} A \xrightarrow[2 \cdot H_{3} \mathrm{O}^{\oplus}, \Delta]{1 . \mathrm{CH}_{2}=\mathrm{CH}-\mathrm{CH}_{2} \mathrm{Br}} B$ The end product (B) is:

A.

B.
C.
(d)

D.

## Answer: A

## - Watch Video Solution

13. Consider the following sequence of reactions :

$$
H_{2} C=C H-C H=C H_{2} \xrightarrow[C C I_{4}]{B r_{1}(1 \mathrm{~mole})} A \xrightarrow[2 \cdot H_{2}, N I]{1 . K C N(\text { excess })} B
$$

The end product $(B)$ is:
A. $\mathrm{H}_{2} \mathrm{~N}-\left(\mathrm{CH}_{2}\right)_{2}-\mathrm{CH}=\mathrm{CH}-\left(\mathrm{CH}_{2}\right)_{2}-\mathrm{NH}_{2}$
B. $\mathrm{H}_{2} \mathrm{~N}-\left(\mathrm{CH}_{2}\right)_{6}-\mathrm{NH}_{2}$

$$
\text { C. } \mathrm{NC}-\mathrm{CH}_{2}-\mathrm{CH}=\mathrm{CH}-\mathrm{CH}_{2}-\mathrm{CN}
$$

$$
\text { D. } \mathrm{H}_{2} \mathrm{C}=\mathrm{CH}-\underset{\substack{\mid \\ \mathrm{NH}}}{\mathrm{CH}}-\left(\mathrm{CH}_{2}\right)_{2}-\mathrm{NH}_{2}
$$

## Answer: B

## - Watch Video Solution

14. 

$$
\begin{aligned}
& \mathrm{CH}_{3} \\
& \mathrm{H}_{3} \mathrm{C}-\mathrm{CH}_{2}-\mathrm{CH}-\underset{\substack{\| \\
\mathrm{C}_{\mathrm{O}}}}{\mathrm{C}}-\mathrm{NH}_{2} \xrightarrow[\Delta]{\mathrm{Br}_{2}+\mathrm{KOH}} A \xrightarrow[2 . \mathrm{AgOH}, \Delta]{1 . \mathrm{CH}_{3} I(\text { excess })} B
\end{aligned}
$$

The major product (B) is:
A. $\mathrm{CH}_{3}-\mathrm{CH}_{2}-\mathrm{CH}=\mathrm{CH}_{2}$
B.
C. $\mathrm{H}_{3} \mathrm{C}-\mathrm{CH}=\mathrm{CH}-\mathrm{CH}_{3}$
D.

Answer: A

- Watch Video Solution

15. 

$\xrightarrow[2 .\left(C H_{3}\right)_{2} N H \text { (excess) }]{\text { 1. } \mathrm{Br}_{2} \text { (1mole) } \mathrm{CCI}_{4}} A \underset{2 . \mathrm{AgOH}, \Delta}{1 . C H_{3} I(\text { excess })} B$ (Major product)
The major product ( $B$ ) is:
(a)
A.

B.
(b)

C.
(d)

D.

## Answer: C

## - Watch Video Solution

16. Consider the following sequence of reaction:

(a)

A.
B.
(b)


C. $\mathrm{HO}-\mathrm{H}_{2} \mathrm{C}-\stackrel{\stackrel{\mathrm{CH}_{3}}{\mathrm{C}}}{\stackrel{\mathrm{C}}{\mathrm{C}} \mathrm{H}_{3}}-\mathrm{CH}_{2}-\mathrm{NH}_{2}$
D. ${ }_{\mathrm{H}_{3} \mathrm{C}}^{\mathrm{C}}$ (H-CH2-NH2

## Answer: D

## - Watch Video Solution

17. The major product $(X)$ of the reaction is:


B.
(b) $\mathrm{H}_{2} \mathrm{~N} \times \mathrm{NOH}^{\mathrm{OH}}$
C.
(c)

D.


## Answer: C

## - Watch Video Solution

18. Which of the following compounds does not liberate $N_{2}$ on treatement with $\mathrm{HNO}_{2}$ ?

$$
\begin{aligned}
& \text { A. } \mathrm{H}_{3} \mathrm{C}-\stackrel{\stackrel{O}{\mathrm{C}}}{\mathrm{C}}-\mathrm{NH}_{2} \\
& \text { B. } \mathrm{H}_{2} \mathrm{~N}-\stackrel{\stackrel{O}{\mathrm{C}}-\mathrm{NH}_{2}}{ } \\
& \text { C. } \\
& \text { (c) } \\
& \text { D. } \mathrm{H}_{3} \mathrm{C}-\underset{\mid}{\mathrm{N}} \mathrm{~N}-\mathrm{CO}_{3}
\end{aligned}
$$

Answer: D
19. The product formed in the reaction is:

B.
(b)

c. $\mathrm{H}_{3} \mathrm{C}-\stackrel{\stackrel{O}{\mathrm{\mid}} \underset{\mathrm{H}}{\mathrm{N}}-\left(\mathrm{CH}_{2}\right)_{3}-\stackrel{\|}{\mathrm{C}}-\mathrm{H}}{ }$
D. $\mathrm{H}_{3} \mathrm{C}-\underset{\mathrm{H}}{\mathrm{N}} \mathrm{N}-\left(\mathrm{CH}_{2}\right)_{3}-\mathrm{CH}_{2} \mathrm{OH}$

## Answer: B

## - Watch Video Solution

20. The major product (B) formed in the reaction sequence is:

$\xrightarrow[\text { dil . NaOH }]{\stackrel{\stackrel{O}{\|}}{\mathrm{Ph}-\mathrm{C}-\mathrm{CI}}} A \xrightarrow[2 \cdot \mathrm{H}_{3}^{\oplus} \mathrm{O}]{\text { 1. } \mathrm{CH}_{3} \mathrm{MgBr}} B$

form (C). The compound (C) on catalytic reduction gives N methylaniline. The compound (A) is:
(d)

B.
(b) $\backslash-\mathrm{C} \equiv \mathrm{N}$
c.

(d)

D.

## Answer: A

## - Watch Video Solution

22. The major end product $(B)$ of the reaction:

$\xrightarrow[Z_{n C l}]{\mathrm{HCI}} A \xrightarrow[2 \cdot H_{2}, N i]{1 . N a C N(\text { excess })} B$
A.
${ }^{(a)} \mathrm{HO} \sim \sim^{\mathrm{Cl}}$
B.
(b) $\mathrm{NC}^{\sim} \sim^{\mathrm{CN}}$
(c) $\mathrm{H}_{2} \mathrm{~N}$
C.
D.
(d) $\mathrm{HO}^{\text {NH2 }}$

## Answer: C

## - Watch Video Solution

23. Which one among the following is expected to form a secondary alcohol on treatment with $\mathrm{HNO}_{2}$ ?

A.

(b) $\triangle \mathrm{NH}_{2}$
B.
(c)

C.

D.

## Answer: C

## - Watch Video Solution

24. The end product ( $B$ ) of the reaction sequence:

$$
\mathrm{C}_{2} \mathrm{H}_{5}-\underset{\substack{\mid \\ H}}{\mathrm{~N}}-\mathrm{CH}_{3} \xrightarrow[\mathrm{NaOH}]{\stackrel{\stackrel{O}{\mathrm{H}}-\mathrm{Cl}-\mathrm{Cl}}{\longrightarrow}} A \underset{\mathrm{H}_{2} \mathrm{O}}{\mathrm{LiAlH}_{4}} B
$$




OH
C.


Answer: B
( Watch Video Solution

25.

1. $\mathrm{Br}_{2}^{-}+\mathrm{KOH}, \Delta$
2. $\mathrm{H}_{3} \mathrm{O}^{\oplus}$
A.
(a)


(c) $\mathrm{NH}_{2} \quad \mathrm{NH}_{2}$
C.


Answer: D

## - Watch Video Solution

1. The major product $(X)$ of the reaction is:

A.

(a)

(b)

B.

(c)

C.

D.

Answer: B

## - Watch Video Solution

2. The major product of the reaction is:

(a)
A.

B.
C.



## Answer: C

## - Watch Video Solution

3. The reaction of $p$-aminophenol with one mole of actyl chloride in presence of pyridine gives:
(a)

A.

B.
(c)

C.


Answer: D
4. The major product $(X)$ formed in the reaction:



B.

(b)

(c)

C.
(d)

D.

## Answer: A

## - Watch Video Solution

5. Which of the following is the stongest Bronsted acid?
A.
(a)

B.
(b)

(c)

C.


## D.

## Answer: C

## - Watch Video Solution

6. Which of the following is the strongest Bronsted base?
A.

(a)

(b)

B.

C.
(c)

(d)

D.

Answer: A

## - Watch Video Solution

7. Which of the following is the weakest Bronsted base?
(a)
A.

o
(b)
B.

(c)

(d)

D.

Answer: A
8. Which of the following is stronger Bronsted base?
(a)

A.

B.

C.
(c) $\mathrm{O}_{2} \mathrm{~N}-\mathrm{C}$
D.


Answer: D

- Watch Video Solution

9. For the following compounds, which is the strongest base and which is strongest acid?



IV
A. II = Strongest base, I = Strongest acid
B. IV=Stronger base, III=Stronger acid
C. III = Strongest base, IV = Strongest acid
D. II = Strongest base, III = Strongest acid

## Answer: C

## - Watch Video Solution

10. Which compound is the likely from following reaction?
$\mathrm{N}-\mathrm{H}-\mathrm{CH}_{3}-\mathrm{I}$ (excess) $\longrightarrow$
(a)

A.
(b)

B.
C.
(c)

(d)

D.

Answer: B
11. Which of these is the strongest base?
Et
(a) $\mathrm{Et}-\mathrm{N}$ Et
A.
(b)

B.

C.

O
(d)

D.

## - Watch Video Solution

12. What sequence of reaction would best accomplish the following reaction?

A. $\mathrm{LiAIH}_{4}, 3 \mathrm{CH}_{3} \mathrm{I} / \mathrm{AgOH}, \Delta$
B. $\mathrm{LiAIH}_{4}, \mathrm{P}_{2} \mathrm{O}_{5} / \Delta$
C. $20 \% \mathrm{H}_{2} \mathrm{SO}_{4} / \Delta, \mathrm{P}_{2} \mathrm{O}_{5} / \Delta$
D. $\mathrm{H}_{2}, \mathrm{Pd}-\mathrm{BaSO}_{4}$

Answer: A
13. What is the likely product from the following reaction?


A.
(a)


B.
(c)
C.
(d)
D.

## Answer: B

## - Watch Video Solution

14. Repeated Hofmann elimination reaction (exhaustive methylation followed by heating with AgOH ) will often remove a nitrogen atom from an amine molecule. Which of the following compounds is the likely product in this case?

(a)

A.
(b)

B.

D.

Answer: B
(D) Watch Video Solution
15. Only one of the following amines will lose its nitrogen atom as trimethyl amine by repeated Hofmann elimination reactions:
(a)
A.

(b)

B.
(c)

(d)


## Answer: D

16. The nitrogan atom in each of the following tertiary amines may be removed as trimethyl amine by repeated Hoffmann elimination. Which of the following amines requires the greater number of Hofmann sequence to accomplish this?
(a)

A.

(b)

B.

C.
(c)

(d)


Answer: A

## - Watch Video Solution

17. The Hinsberg test of a $C_{5} H_{14} N_{2}$ compound produces a solid that is insoluble in $10 \%$ aq. $N a O H$. This solid derivative dissolves in $10 \%$ aq. $\mathrm{H}_{2} \mathrm{SO}_{4}$. Which of the following would best fit these facts ?
(a)

A.

(c)

C.
(d)

D.

## Answer: B

## - Watch Video Solution

18. What set of conditions would be useful for preparing a
$2^{\circ}$ amine?
A. $2^{\circ} \mathrm{R}-\mathrm{Br}+\mathrm{NaNH} 2$
B. $2^{2} R-B r+N a N_{3}, H_{2} / P t$
C. $1^{\circ} \mathrm{R}-\mathrm{NH}_{2}+1^{\circ} \mathrm{RCHO}, \mathrm{H}_{2}$ and Pt
D.

## - Watch Video Solution

19. Which of the following amines reacts most rapidly with


A.
(b)

B.

C.
(d)


Answer: D
(D) Watch Video Solution
20. Consider the following sequence of reactions:


Identify
product C:
(a)

A.

(b)

B.
C.

(d)

D.

## - Watch Video Solution

21. The major product formed in the reaction is:


B. $H_{3} C-C H=C D_{2}$
C. $\mathrm{H}_{2} \mathrm{C}=\mathrm{CH}-\mathrm{CD}_{3}$
D. $\mathrm{H}_{2} \mathrm{C}=\mathrm{N}-\mathrm{CH}_{3}$

## Answer: C

## - Watch Video Solution

22. The product formed in the reaction is:

(a)

B.

C.
(c)

D.

(d)


## Answer: B

## - Watch Video Solution

23. 



B:
A.

(b)

B.
(c)

C.
(
(d)

D.

Answer: D

## - Watch Video Solution

24. 


A.
(a)


C.

(d)

D.

Answer: A

## - Watch Video Solution

25. The final major product of the reaction is:
$\mathrm{Ph}-\underset{\mathrm{Ph}}{\stackrel{\mathrm{OH}}{\mathrm{C}}} \underset{{ }_{\mathrm{N}}^{\mathrm{C}}}{\mathrm{C}}-\underset{\mathrm{H}_{2}}{\mathrm{C}} \mathrm{H}-\mathrm{CH}_{3} \xrightarrow{\mathrm{NaNO}_{2}+\mathrm{HCI}}$

(b)

B.



## Answer: C

## Level 2 ( $\mathbf{Q} .51$ To 0.75 )

1. The major product of the reaction:

$\mathrm{NaNO}_{2}+\mathrm{HCl}$
A.
(b)

B.

C.
D.

## Answer: D

## - Watch Video Solution

2. The end product of the following reaction is:

A.
(a)

B.

(c)
C.

D.
(d)


## Answer: A

## - Watch Video Solution

3. Consider the following sequence of reactions:


The products (B) and (C) are:
A.

B.

C.

D.


## Answer: B

## - Watch Video Solution

4. Which of the following compounds will react with cyclopentanone to form an enamine?
A.
(a)


(b)

B.
(c)

C.
(d)

D.

## Answer: C

## - Watch Video Solution

5. Predict the major product ' $X$ ' in the following reaction:

(a)
A.

(b)

(c)

D.

Answer: A


Product
of this reaction is:
A.
(a)

B.
(b) OOO
C.
(c)

(d)


Answer: D
(D) Watch Video Solution

7.

Compound $B$ is :
A.

(c)

c.

(a)

B.

(d)

D.

## Answer: B

## - Watch Video Solution



Product
$(X)$ is :
(a)

A.


B.


C.

D.


## - Watch Video Solution

## 9. Find the final product of the reaction


B. (b)


Answer: B

## - Watch Video Solution



Product
$(\mathrm{Y})$ of the reaction:
A.

(b)

B.
C. Mixture of (a) and (b)
(d) $\mathrm{H}_{3} \mathrm{C}{\underset{\mathrm{H}}{-\mathrm{Ph}} \mathrm{O}^{\circ} \mathrm{O}}^{\circ}$
11. Complete the following reaction

A.
(a)

B.

(c)

D.


## - Watch Video Solution

12. Predict the product of following reaction:



B.
(c)

C.
Ts
D.
(d)


## Answer: D

## ( Watch Video Solution

13. 



The product $M$ is:
A.

(b)

(c)

C.
(d)


Answer: B

- Watch Video Solution


14. 

Final product ' O ' is:
A.

(b)

(c)

C.
(d)


## Answer: C

## - Watch Video Solution

15. Consider the following diazonium ion:





The order of reactivity towards diazo coupling with phenol in presence of dil. NaOH :
A. $P>Q>R>S$
B. $Q>S>R>P$
C. $P>R>S>Q$
D. $S>R>Q>P$

## Answer: B

## - Watch Video Solution

16. Reaction I $\mathrm{Ph}-\stackrel{\stackrel{O}{\mathrm{C}}}{\mathrm{\|}}-\mathrm{NH}_{2} \xrightarrow{\stackrel{\ominus}{O D, B r_{2}}} A$

Reaction II $\mathrm{Ph}-\stackrel{\stackrel{O}{\|}}{\mathrm{\|}}-N D_{2} \xrightarrow{\stackrel{\ominus}{\mathrm{OH}, B r_{2}}} B$
Products A and B are:
A. $\mathrm{Ph}-\mathrm{NH}_{2}$ and $\mathrm{Ph}-N D_{2}$
B. $\mathrm{Ph}-\mathrm{ND}_{2}$ and $\mathrm{Ph}-\mathrm{NH}_{2}$
C. Both $\mathrm{Ph}-\mathrm{NH}_{2}$
D. Both $\mathrm{Ph}-N D_{2}$

## - Watch Video Solution

17. An organic compound (A) $\mathrm{C}_{9} \mathrm{H}_{13} \mathrm{~N}$ dissolves in dil. HCI and releases $\mathrm{N}_{2}$ with $\mathrm{HNO}_{2}$ giving an optically active alcohol. Alcohol ox oxidation gives dicarboxylic acid, which on heating form anhydride. The organic compound 'A' is:
A.

(b)

B.
C.
(c)

D.
(d)


Answer: C

## D Watch Video Solution


18.

The final product is:
(a)

(b)

(c)

(d)


Answer: A

## - Watch Video Solution

19. Identify ' $X$ ' in the following sequence of reaction:
$\left\langle\xrightarrow{\stackrel{\oplus}{\mathrm{N}} \equiv \mathrm{NCl}} \xrightarrow[\mathrm{KCN}]{\stackrel{\mathrm{CuCN}}{\longrightarrow}} P \xrightarrow{\mathrm{LiAlH}_{4}} Q \xrightarrow{\mathrm{HNO}_{2}} X\right.$
A. Benzoic acid
B. Phenyl acetic acid
C. Benzyl alcohol
D. Benzamide

## Answer: C

## D Watch Video Solution

20. Which sequence of steps will be able to produce 3,3'-dinitro-biphenyl from benzene?
A. $\mathrm{HNO}_{3} / \mathrm{H}_{2} \mathrm{SO}_{4}, \mathrm{CI}_{2} / \mathrm{FeCI}_{3}, \mathrm{Na} /$ ether
B. $\mathrm{CI}_{2} / \mathrm{FeCI}_{3}, \mathrm{HNO}_{3}, \mathrm{H}_{2} \mathrm{SO}_{4}, \mathrm{Na} /$ ether
C. $\mathrm{CI}_{2} / \mathrm{FeCI}_{3}, \mathrm{H}_{2} \mathrm{SO}_{4}, \mathrm{Na} /$ ether
D. $\mathrm{I}_{2} / \mathrm{HIO}_{3}, \mathrm{CI}_{2} / \mathrm{FeCI}_{3}, \mathrm{C}_{6} \mathrm{H}_{5} \mathrm{NO}_{2}$

## Answer: A

## D View Text Solution

21. $1^{\circ}, 2^{\circ}$ and $3^{\circ}$ nitroalkane can be identified by action of:
A. $\mathrm{HNO}_{3}+\mathrm{NaOH}$ (aq).
B. $\mathrm{CHCI}_{3}+\mathrm{NaOH}$ (aq)
C. $\mathrm{HNO}_{2}+\mathrm{NaOH}(\mathrm{aq})$
```
D. CHCI }+\textrm{KOH}\mathrm{ (aq.)
```


## Answer: C

## - Watch Video Solution

22. A compound ' $X$ ' when reacted with $P C I_{5}$ and then with
$\mathrm{NH}_{3}$ gives ' Y '. When ' Y ' treated with $B r_{2}$ and KOH produced 'Z'. Z on treatement with $\mathrm{NaNO}_{2}+\mathrm{HCI}$ at $0^{\circ} \mathrm{C}$ and then boiling produced ortho-cresol. Compound ' X ' is:
A. o-toluic acid
B. o-chlorotoluene
C. o-bromotoluene
D. m-toluic acid

## Answer: A

## - Watch Video Solution

23. 



For such kind of diazo- coupling reaction the suitable substituents $P$ and $S$ are respectively:
A. $-\mathrm{NH}_{2}$ and $-\mathrm{OCH}_{3}$
B. $-\mathrm{NO}_{2}$ and $-\stackrel{O}{\mathrm{C}}-\mathrm{H}$
C. $-\mathrm{NH}_{2}$ and $-\mathrm{NHCH}_{3}$


Answer: D

## - Watch Video Solution

24. 



Identify ' B ':
(a)

A.
(b)

B.
(c)

C.
(d)
D.

## Answer: D

## - Watch Video Solution

25. The final product $B$ obtained in the reaction is:
$\mathrm{CH}_{3} \mathrm{CH}_{2}$
$\mathrm{CH}_{3} \mathrm{CH}_{2}-\mathrm{N} \xrightarrow{\mathrm{H}_{2} \mathrm{O}_{2}} A \xrightarrow{\Delta} B+\mathrm{H}_{2} \mathrm{C}=\mathrm{CH}_{2}$
$\mathrm{CH}_{3} \mathrm{CH}_{2}$
A.

B. $\left(\mathrm{H}_{2} \mathrm{C}=\mathrm{CH}\right)_{2}$
C. $\mathrm{H}_{3} \mathrm{C}-\mathrm{CH}=\mathrm{CH}-\mathrm{CH}_{2}$

## Answer: A

## - Watch Video Solution

## Level 2 (0.76 To Q.89)


1.

Find out Y of the reaction :
A.
(a)

(b)
B.


D.

## Answer: C

( Watch Video Solution


The compound $C$ is:
(a)

B.

(c)

(d)

D.

Answer: B


## 3.

The compound B is:
A.
(a)


(b)

B.

(d)

D.

Answer: D

## D Watch Video Solution


4.
(a)

A.

B.

C.
(d)

D.

Answer: B
5. Identify the final product of following reaction:

A.

(a)

(b)

B.

B.

C.

(d)

D.

## - Watch Video Solution

6. Consider the following sequence of reaction:


The final product is:
A.
(a)

B.

(c)

(d)

D.

Answer: B

## - Watch Video Solution



The product can be given as:
A.
(a)

(b)

B.

(c)


C.
(d)


Answer: D

## ( Watch Video Solution

8. In a set of reactions acid yielded a compound D.


The strucuture of $D$ would be :
(a) $\wedge_{N} \nearrow^{\mathrm{CH}_{3}}$
H
A.
(b)

B. $\mathrm{NH}_{2}$
C. ${ }^{(c)} \sim^{N H_{2}}$
D.


Answer: B

- Watch Video Solution

9. What would be the final product of reaction?

(a)

A. $\mathrm{CH}_{2} \mathrm{CH}_{3}$
B.
(b) $\mathrm{Br}^{( }$
(c)
C.
D. ${ }^{(\mathrm{d})} \mathrm{Br}^{\sim} \sim^{\mathrm{NH}}$

Answer: A
10. Identify major product of following sequence of reaction:

(a)

(b)

(c)

C. $\mathrm{H}_{3} \mathrm{C}$
(d) HO
D.


Answer: C
11. Identify the final producy of following sequence of reaction:

(a)

B.
A.
(b)

Cl
OH
C.
(c)

(d)

D.

## Answer: B

## - Watch Video Solution

12. Identify the major product of following reaction:

(a)
A.

B.
(b) $\sim \mathrm{CH}_{3}^{\mathrm{CH}_{3}}$
(c)

D. none of these

## Answer: C

## - Watch Video Solution

13. Identify final product of following sequence of reaction:


(b)

B.

D.


Answer: C
( Watch Video Solution
14. What is the product of following reaction sequence?

(a)

A.


B.


C.
(d)

D.

## Answer: B

## - Watch Video Solution

## More Than One Correct (Q.1 To Q.25)

1. The presence of primary amine can be confirmed by its reaction with:
A. $\mathrm{HNO}_{2}$
B. $\mathrm{CHCI}_{3}+\mathrm{NaOH}$
C. $\mathrm{CS}_{2}$ and $H g C I_{2}$
D. $\mathrm{H}_{2} \mathrm{SO}_{4}$

## Answer: A::B::C

## - Watch Video Solution

2. Which of the following reactions can be used to make ethyl isocyanide?
A. $\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{NH}_{2}+\mathrm{CHCI}_{3} \xrightarrow[\Delta]{\mathrm{KOH}}$
B. $\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{Br}+\mathrm{AgCN} \rightarrow$
C. $\mathrm{CH}_{3} \mathrm{CH}_{2}-\mathrm{NH}-\underset{\substack{\| \\ O}}{\mathrm{C}}-\mathrm{H} \xrightarrow{\mathrm{POCI}_{3}}$
D. $\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{Br}+\mathrm{KCN} \rightarrow$

## Answer: A::B::C

## - Watch Video Solution

3. By which of the following reactions can methylcyanide be prepared?
A. $\mathrm{CH}_{3} \mathrm{Br} \xrightarrow[D M F]{K C N}$
B. $\mathrm{CH}_{3} \mathrm{NH}_{2}+\mathrm{CHCI}_{3} \xrightarrow{\mathrm{KOH}}$
c. $\mathrm{CH}_{3}-\mathrm{CH}=\mathrm{N}-\mathrm{OH} \xrightarrow[\Delta]{\mathrm{P}_{2} \mathrm{O}_{5}}$
D. $\mathrm{CH}_{3}-\stackrel{O}{\mathrm{C}}-\mathrm{NH}_{2} \xrightarrow[\Delta]{\mathrm{P}_{4} O_{10}}$

## (D) Watch Video Solution

4. Which of the following compounds react with $\mathrm{HNO}_{2}$ ?
(a) $\stackrel{\mathrm{NO}_{2}}{ }$
A.
(b) $>-\mathrm{NO}_{2}$
B.
c. ${ }^{(c)} \mathrm{NO}_{2}$
D. (d) $\mathrm{NH}_{2}$

## Answer: $\mathrm{B}:: \mathrm{C}:$ :D

5. Consider the following reaction:


The staring substance 'A' can be:

A.

B.


## Answer: A::B::D

## - Watch Video Solution

6. $\mathrm{C}_{6} \mathrm{H}_{5}-\mathrm{CH}_{2}-I \xrightarrow[\Delta]{\mathrm{NaN}_{3}}$ Products

Reaction is assumed to involve nitrene as intermediate, then various possible products are:
A. $\mathrm{C}_{6} \mathrm{H}_{5} \mathrm{CN}_{2} \mathrm{NH}_{2}$
B. $\mathrm{C}_{6} \mathrm{H}_{5} \mathrm{~N}=\mathrm{CH}_{2}$
C. $C_{6} H_{5} C N=N H$
(d) $\mathrm{C}_{6} \mathrm{H}_{5} \mathrm{CH}{ }_{-}^{/ 1} \mathrm{~N}$
D.

## Answer: B::C

## - Watch Video Solution

7. Which of the following can give $1^{\circ}$ amine?

## $\mathrm{CH}_{3}$

A. $\mathrm{Ph}-\stackrel{\mid}{\mathrm{C}} \mathrm{H}-\mathrm{OH} \xrightarrow{\mathrm{NaCN}, \mathrm{H}^{\oplus}}$
B. $\mathrm{Ph}-\mathrm{CH}=\mathrm{CH}-\stackrel{\stackrel{+}{\mathrm{C}}}{\mathrm{C}}-\mathrm{NH}_{2} \xrightarrow[\mathrm{CH}_{3} \mathrm{OH}]{\stackrel{\mathrm{NaOCI}}{\longrightarrow}}$
C. $\mathrm{Ph}-\mathrm{C} \equiv \mathrm{C}-\stackrel{\|}{\mathrm{C}}-\mathrm{NH}_{2} \xrightarrow{\mathrm{NaOBr}}$
D. $\mathrm{Ph}-\stackrel{\stackrel{\mid l}{\mathrm{C}}}{\mathrm{C}}-\mathrm{CI} \xrightarrow[\Delta]{\mathrm{NaN}_{3}} \xrightarrow{\mathrm{LiAIH}_{4}}$

Answer: B::C::D

## - Watch Video Solution

8. Which of the following can distinguish?

A. $\left(\mathrm{COOC}_{2} \mathrm{H}_{5}\right)_{2}$
B. $\mathrm{NaNO}_{2}+\mathrm{HCI}$
C. $C s_{2}, H g C I_{2}$
D. $A g_{2} O / \Delta$

## - Watch Video Solution

9. Dopamine is a drug used in the treatment of Parkinson's disease:


Which of the following statements about this compound are correct?
A. It can exist only in optically active forms
B. One mole will react with 3 mole of NaOH to form a
C. It can exist as a zwitter ion in the aqueous solution
D. It given nitroso compound on treatement with $\mathrm{HNO}_{2}$

Answer: A::B::C

## - Watch Video Solution

10. Which of the following give nitrosonamine on treatment with $\mathrm{HNO}_{2}$ ?
A.
(a) $\quad \mathrm{N}-\mathrm{H}$
B. $\mathrm{H}_{3} \mathrm{C}-\underset{\mathrm{CH}}{\mathrm{CH}} \mathrm{C}-\mathrm{CH}_{2}-\mathrm{NH}_{2}$
C.
(c)

D.
(d)


## Answer: A::C

## - Watch Video Solution

11. Which of the following sequence of reagent is the good mean to furnish the conversion?

R-CH2OH $\rightarrow$ R $-\mathrm{CH}_{2} \mathrm{NH}_{2}$
A. $\mathrm{KMnO}_{4}, \mathrm{SOCI}_{2}, \mathrm{NH}_{3}, \Delta, \mathrm{NaOBr}$
B. $S O C I_{2}, N a C N, H_{2} / N i$
C. $\mathrm{CrO}_{3}$ in dilute acetone, $\mathrm{NH}_{3}, \mathrm{H}_{2}, \mathrm{Ni}$
D. $\mathrm{Cu}, 300^{\circ} \mathrm{C}, \mathrm{NH}_{2}, \mathrm{LiAiH}_{4}$

## Answer: A::B::C

## - Watch Video Solution

12. Choose the correct comparisons of basicity:
A.

B.
(b)

(c)

D.
(d)


## - Watch Video Solution

13. Which of the following arrangements are correct with respect to the property of the compounds indicated in the parentheses?
A. $\mathrm{HCOOH}>\mathrm{CH}_{3} \mathrm{COOH}>\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{COOH}$
(Acidic strength)
B.
C.
D.


## - Watch Video Solution

14. Which of the following products are formed when 1propanamine is treated with $\mathrm{NaNO}_{2}+\mathrm{HCI}$ ?
A.

B.

OH
(c)

C.
(d) $\underset{\sim}{~}$
D.

Answer: A::C::D
15. Which of the following will give Hofmann-Bromoamide reaction?
(a)

A.
(b)

B.



## - Watch Video Solution

16. Which of the following reactions represent major products?
A. (a) $\mathrm{H}_{3} \mathrm{C}-\underset{\mathrm{H}_{3} \mathrm{C}}{\mathrm{O}} \mathrm{C}_{\mathrm{N}}^{-\mathrm{N}} \xrightarrow[\Delta]{\mathrm{H}^{\oplus}} \mathrm{H}_{3} \mathrm{C}-\sqrt{\mathrm{O}}-\stackrel{\mathrm{C}}{\mathrm{C}}-\mathrm{NH}-\mathrm{CH}_{3}$
B.

C. (c) $\mathrm{Ph}-\stackrel{\|}{\mathrm{C}}-\mathrm{NH}_{2} \xrightarrow{\mathrm{Br}_{2}+\mathrm{KOH}} \mathrm{Ph}-\mathrm{NH}_{2}$
D. ${ }^{(d)}$


## Answer: A::B::C

17. Which of the following products will not form by following reaction?

A.
(a) $\mathrm{H}_{3} \mathrm{C}-\mathrm{O}-\mathrm{CN}$
(b)

(c) $\mathrm{H}_{3}$

C.
D.


## Answer: A::C::D

18. 

$\mathrm{Ph}-\stackrel{O}{\stackrel{\mid l}{\mathrm{C}}}-\mathrm{NH}_{2}+\mathrm{Ph}-\mathrm{CH}_{2}-\stackrel{\stackrel{O}{\mathrm{C}}-\stackrel{15}{\mathrm{~N}} \mathrm{H}_{2} \xrightarrow{\stackrel{\ominus}{\mathrm{OH}+B r_{2}}} \mathrm{C}}{\mathrm{O}} \mathrm{B}$

Products $A$ and $B$ are:
A. $\mathrm{Ph}-\mathrm{NH}_{2}$
B. $\mathrm{Ph}-\mathrm{CH}_{2} \stackrel{15}{N} \mathrm{H}_{2}$
C. $\mathrm{Ph}-\mathrm{CH}_{2}-\mathrm{NH}_{2}$
D. $\mathrm{Ph}-\stackrel{15}{N} \mathrm{H}_{2}$

Answer: A::B

## - Watch Video Solution

19. Reaction involves isocyanate as intermediate porudct:
A. Curtius rearrangement
B. Lossen rearrangement
C. Schmidt rearrangemnt
D. Hofmann rearrangement

## Answer: A::B::C::D

## - Watch Video Solution

20. Consider the structures:


Which of the following statements are correct?
A. Basic stength of II is greater than I
B. Basic strength of II is less than that of I
C. Basic strength of IV is greater than III
D. Basic strength of IV is less than that of III

## Answer: A::C

## - Watch Video Solution

21. Which of the following give $L$ Liebermann nitroso reaction?
(a)

A.

B.
(b)

C.
D.
(d)


## Answer: A::B

## D Watch Video Solution

22. Which are related with Curtius rearrangement?
A. $N a N_{3}$
B. $R-N H_{2}$
C. $R-\stackrel{O}{\stackrel{\|}{C}}-C I$
D. $R-\stackrel{O}{\stackrel{I}{C}}-O H$

## - Watch Video Solution


23.

The possible products are:

A.
(b)

B.

(d)


## ( Watch Video Solution

24. Which of the following give Schiff base with aldehyde?
A. $\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{NH}_{2}$
B. $\mathrm{C}_{6} \mathrm{H}_{5}-\mathrm{CH}_{2}-\mathrm{NH}_{2}$
C. $\mathrm{C}_{6} \mathrm{H}_{5}-\mathrm{NH}_{2}$
D. $C_{6} H_{5}-N O_{2}$

Answer: A::B::C

- Watch Video Solution

25. Which of the following give(s) aniline by reduction of nitrobenzene?
A. $H_{2} / P d-C$
B. $S n+H C I$
C. $C u+H C I$
D. $\left(\mathrm{NH}_{4}\right)_{2} \mathrm{~S}$

## Answer: A::B::C::D

## - Watch Video Solution

1. Optically active amine having molecular formula $\mathrm{C}_{5} \mathrm{H}_{13} \mathrm{~N}$ on reaction with $\mathrm{NaNO}_{2}+\mathrm{HCI}$ produces, $3^{\circ}$ optically inative alcohol. Find out structures of amines:
(a)

A.
B.
(b)

c.
(c) $\mathrm{T}^{\mathrm{NH}_{2}}$
D.
(d) $\sim^{N}-H$

## Answer: A::C

2. Find out products which are formed by the following reaction:

A.

(b)

C.

D.


Answer: A::B
3. Which of the following is soluble in dil aqueous $H C I$ ?
A. $\mathrm{C}_{6} \mathrm{H}_{5} \mathrm{NH}_{2}$
B. $\mathrm{C}_{6} \mathrm{H}_{5} \mathrm{CH}_{2} \mathrm{NH}_{2}$
C. $\mathrm{C}_{6} \mathrm{H}_{5} \mathrm{CONH}_{2}$
D.

4. The structural form of a compound $A\left(C_{6} H_{11} N\right)$ is resolvable, dissolve in dil. HCl and reacts with $\mathrm{HNO}_{3}$.

Compound A could be:
(a)

A.
(b) $-\mathrm{NH} \mathrm{CH}_{3}$
B.
C.
(c) $\square-\mathrm{NH}-\mathrm{CH}_{3}$
D.
(d) $\square^{\mathrm{NHCH}_{3}}$

## - Watch Video Solution

5. Which of the following basically exist as dipolar ion?
$\mathrm{N}\left(\mathrm{CH}_{3}\right)_{2}$
A. $\mathrm{HO}^{\mathrm{C}} \mathrm{O}$
$\mathrm{H}_{3} \mathrm{C} \xlongequal[\mathrm{NH}_{2}]{\mathrm{COOH}} \mathrm{H}$
(c)

(d)


## Linked Comprehension Type

1. The conversion of an amide by reaction NaOH and $\mathrm{Br}_{2}$
to primary amine that has one carbon than the strating amide is known as Hofmann-Bromoamide reaction.
$\mathrm{R}-\stackrel{\stackrel{\mathrm{U}}{\mathrm{C}}}{\mathrm{C}}-\mathrm{NH}_{2} \xrightarrow[\text { or } \mathrm{NaOBr}]{\stackrel{\mathrm{Br}+\mathrm{NaOH}}{\longrightarrow}} \mathrm{R}-\mathrm{NH}_{2}+\mathrm{NaBr}+\mathrm{Na}_{2} \mathrm{CO}_{3}$
Mechanism:


Number of moles of NaOH consumed in above reaction:
A. 1
B. 2
C. 3
D. 4

## Answer: D

## - Watch Video Solution

2. The conversion of an amide by reaction NaOH and $\mathrm{Br}_{2}$ to primary amine that has one carbon than the strating amide is known as Hofmann-Bromoamide reaction.
$R-\stackrel{\text { I| }}{\mathrm{C}}-\mathrm{NH}_{2} \xrightarrow[\text { or } \mathrm{NaOBr}]{\stackrel{\mathrm{Br}_{2}+\mathrm{NaOH}}{\longrightarrow}} R-\mathrm{NH}_{2}+\mathrm{NaBr}+\mathrm{Na}_{2} \mathrm{CO}_{3}$
Mechanism:




Find $X$ and $Y$ :
A.

b) 1

B.

(c) $X=Y=$

D.


## - Watch Video Solution

3. The conversion of an amide by reaction NaOH and $\mathrm{Br} r_{2}$ to primary amine that has one carbon than the strating amide is known as Hofmann-Bromoamide reaction.

$$
R-\stackrel{O}{\|} \mathrm{C}-\mathrm{NH}_{2} \xrightarrow[\text { or } \mathrm{NaOBr}]{\stackrel{\mathrm{Br}}{2}+\mathrm{NaOH}} \mathrm{R}-\mathrm{NH}_{2}+\mathrm{NaBr}+\mathrm{Na}_{2} \mathrm{CO}_{3}
$$

Mechanism:

$\mathrm{NH}_{2}$
(c)
C.
D. all of these

## Answer: A

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4. Ketoxime when heated with certain reagents undergoes
rearrangement to form amides. This is known as Beckman's rearrangement.


Find out slowest step of the reaction:
A. I
B. II
C. III
D. IV

## Answer: B

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5. Ketoxime when heated with certain reagents undergoes rearrangement to form amides. This is known as Backmann's rearrangement.



Find out (X):
A. ${ }^{(\mathrm{a}) \mathrm{H} \cdot \mathrm{C}-()^{-1}-\mathrm{CH}-\mathrm{Ph}}$
B.

C.

D. $p h-\stackrel{\stackrel{O}{\|}}{C}-\mathrm{NH}-\mathrm{Ph}$

Answer: C
6. Ketoxime when heated with certain reagents undergoes
rearrangement to form amides. This is known as

Backmann's rearrangement.



Find out $(X)$ of the reaction:

A.
(b)

B.

C.

D.

Answer: D

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Which of the following amines cannot be prepared by path-I?

B. $\mathrm{Ph} \mathrm{NH}_{2}$
C. (c) $-\mathrm{NH}_{2}$
D. (d) N

Answer: C

8.

Consider path II, choose the major product for 1 and 2:

A. $1-Y, 2-X$
B. $1-X, 2-Y$
C. $1-X, 2-X$

```
D. \(1-Y, 2-Y\)
```


## Answer: A

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9.

In the path I, if

## $\mathrm{CH}_{3}$ <br> $R=$ <br> 

then the
amine finally formed is :
C. racemic mixture of $a$ and $b$
D. none of these

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10. An organic compound ' $A$ ' has molecular formula $\mathrm{C}_{9} \mathrm{H}_{13} \mathrm{NO}$ and it can be resolved into enatiomers. A does not decolourise bromine water solution. A on reflxing with dilute $\mathrm{H}_{2} \mathrm{SO}_{4}$ yields another resolable compound $\mathrm{B}\left(\mathrm{C}_{9} \mathrm{H}_{14} \mathrm{O}_{3}\right)$ which gives effervescence with $\mathrm{NaHCO}_{3}$. B on treatemet with $\mathrm{NaBH}_{4}$ yields $\mathrm{C}\left(\mathrm{C}_{9} \mathrm{H}_{16} \mathrm{O}_{3}\right)$ on heating with concentrated $\mathrm{H}_{2} \mathrm{O}_{4}$ yields ester $D\left(\mathrm{C}_{9} \mathrm{H}_{14} \mathrm{O}_{2}\right)$. Compound A on reduction with $\mathrm{LiAIH}_{4}$, followed by treatement of $\mathrm{H}_{2} \mathrm{SO}_{4}$ yields following compound:


Find out structure of compound 'A' :
A.
(a)

B.

c.
(c)

D.

(d)


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11. An organic compound 'A' has molecular formula $\mathrm{C}_{9} \mathrm{H}_{13} \mathrm{NO}$ and it can be resolved into enatiomers. A does not decolourise bromine water solution. A on reflxing with dilute $\mathrm{H}_{2} \mathrm{SO}_{4}$ yields another resolable compound $\mathrm{B}\left(\mathrm{C}_{9} \mathrm{H}_{14} \mathrm{O}_{3}\right)$ which gives effervescence with $\mathrm{NaHCO}_{3}$. B on treatemet with $\mathrm{NaBH} H_{4}$ yields $\mathrm{C}\left(\mathrm{C}_{9} \mathrm{H}_{16} \mathrm{O}_{3}\right)$ on heating with concentrated $\mathrm{H}_{2} \mathrm{O}_{4}$ yields ester $D\left(\mathrm{C}_{9} \mathrm{H}_{14} \mathrm{O}_{2}\right)$.

Compound A on reduction with $\mathrm{LiAIH}_{4}$, followed by treatement of $\mathrm{H}_{2} \mathrm{SO}_{4}$ yields following compound:


The sweet smelling neutral compound D is:
A.

B.

(c)

D.


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12. An organic compound ' $A$ ' has molecular formula $\mathrm{C}_{9} \mathrm{H}_{13} \mathrm{NO}$ and it can be resolved into enatiomers. A does not decolourise bromine water solution. A on reflxing with dilute $\mathrm{H}_{2} \mathrm{SO}_{4}$ yields another resolable compound $\mathrm{B}\left(\mathrm{C}_{9} \mathrm{H}_{14} \mathrm{O}_{3}\right)$ which gives effervescence with $\mathrm{NaHCO}_{3}$. B on treatemet with $\mathrm{NaBH} H_{4}$ yields $\mathrm{C}\left(\mathrm{C}_{9} \mathrm{H}_{16} \mathrm{O}_{3}\right)$ on heating with concentrated $\mathrm{H}_{2} \mathrm{O}_{4}$ yields ester $D\left(\mathrm{C}_{9} \mathrm{H}_{14} \mathrm{O}_{2}\right)$.

Compound A on reduction with $\mathrm{LiAIH}_{4}$, followed by treatement of $\mathrm{H}_{2} \mathrm{SO}_{4}$ yields following compound:


Due to reduction of optically pure ' B ' two isomeric product 'C' form. Isomeric product ' C ' are:
A. Enantiomers
B. Diastereomers
C. Position isomers
D. Functional isomers

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13. When an primary aromatic amine is treated with $\mathrm{NaNO}+\mathrm{HCI}$ at $0^{\circ}-5^{\circ} \mathrm{C}$, a diazonium salt is formed and the reaction is called diazo reaction. In this reaction mineral acid must be added to prevent the coulping reaction of diazonium salt with excess of aryl amine. diazonium salt is highly in the synthesis of number of coloured dyes.

For the following diazonium ion the decreasing order of reactivity of these ion in azo-coupling reaction:

A. $Q>S>R>P$
B. $Q>S>P>R$
C. $P>Q>R>S$
D. $S>R>Q>P$

## Answer: B

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14. When an primary aromatic amine is treated with $\mathrm{NaNO}+\mathrm{HCI}$ at $0^{\circ}-5^{\circ} \mathrm{C}$, a diazonium salt is formed and the reaction is called diazo reaction. In this reaction mineral acid must be added to prevent the coulping reaction of diazonium salt with excess of aryl amine. diazonium salt is highly in the synthesis of number of
coloured dyes.
In the given reaction.


The final product is

B.

C.

D.


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15. When an primary aromatic amine is treated with $N a N O_{2}+H C I$ at $0^{\circ}-5^{\circ} \mathrm{C}$, a diazonium salt is formed and the reaction is called diazo reaction. In this reaction mineral acid must be added to prevent the coulping reaction of diazonium salt with excess of aryl amine. diazonium salt is highly in the synthesis of number of coloured dyes.

When 2,4-dinitrophenol react with $\mathrm{NaNO}_{2}+\mathrm{HCI}$ at $5^{\circ} C$ followed by reaction with anisole, a coloured compound is formed which can be given as:
A.
(a)

B.

C.

D.


Answer: C

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Match The Column
（b）

（c）

（d）


Q． $3^{\circ}$ amine

R．Gives $\mathbf{p}$ vitive Tollen｀s test
Gives pungent smell on treatment with $\mathrm{CHCl}_{3},{ }^{\ominus} \mathrm{OH}$
$S$ ．The amine which is not prepared b Hofmann ammonol us process
T．Hydroxyl imine

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2. 

min（II）


（d）


P．Treatment of $\mathrm{NaNO}_{2}, \mathrm{HCl}$ gives N －nitroso compound

Q．Treatment of $\mathrm{NaN}^{\prime}{ }_{2}, \mathrm{HCl}$ gives diazoniumchloride
R．Treatment of excess $\mathrm{CH}_{3}$ followed by $\mathbf{A}$（1）aminat gives out alke
S．Treatment of $\mathrm{H}_{1} 1$ ：いいと dealkylation

```
3.
    Match
(a) Hofmann degradation
(h) Curtius rearrangement
(c) Lossen rearrangement
(d) Hemiaminal
the following
folu M:
\(P\). Aldchyde \(+1^{\circ}\) amine
Q. I
R. \(\mathrm{Br}_{2}+\) hい 11
S. \(R-{ }_{C}^{C} \cdot N \quad N=N\)
4.
4. Column (I)
\(\mathrm{CH}_{3}\)
(a) \(\mathrm{H}_{3} \mathrm{C}-\mathrm{CH}-\mathrm{OH}\)
(b) \(\mathrm{H}_{3} \mathrm{C}-\mathrm{C}-\mathrm{NH}_{2}\)
(c) \(\mathrm{H}_{3} \mathrm{C}-\mathrm{N}=\mathrm{C}=\mathrm{O}\)
(d) \(R-\stackrel{\oplus}{\mathrm{N}} \equiv \stackrel{\ominus}{\mathrm{C}}\)
5. Column (I)
(a)


II

\section*{Column (II)}

Q. Reduction gives \(2^{\circ}\) amine
R. \(\mathrm{Br}_{2}, \stackrel{\ominus}{\mathrm{O}} \mathrm{H}\), gives bromoform
\(S\). NaOBr gives \(1^{\circ}\) amine
\(T\). Dehydration gives nitrile
Celumn (II)
\(P\). Schmidt reaction
5. Match
the
following

\section*{Column (II)}
(i) \(\mathrm{R}-\mathrm{C}-\mathrm{NH}_{2} \longrightarrow \mathrm{R} \quad\left(\mathrm{H}_{2} \mathrm{NH}_{2}\right.\)
\(P\). Schmidt reaction
(b) \(\mathrm{R}-\mathrm{C}-\mathrm{NH}_{2} \longrightarrow \mathrm{R} \quad \mathrm{NH}_{2}\)
(). \(P_{2}{ }^{\prime}{ }_{5}\)
(c) \(R-\mathrm{C}-\mathrm{NH}_{2} \longrightarrow R-\mathrm{CN}\)
R. Hofmann reaction
(d) \(\mathrm{R}-\mathrm{C}-\mathrm{N}_{3} \longrightarrow \mathrm{RNH}\) 2
S. \(\mathrm{LiAlH}_{4}\)

\section*{- Watch Video Solution}
6. Match the following columns

(b) \(\mathrm{NH}_{2}\)
(c) \(\mathrm{N}^{\prime}\)
(d)

P. Treatment of \(\mathrm{CS}_{2}, \mathrm{HgCl}_{2}\) gives out alkyl isothiocyanate
Q. Treatment of \(\mathbf{P h}-\mathrm{SO}_{2}-\mathrm{Cl}\) produces the compound insoluble in alkali
R. Treatment of \(\mathrm{H}_{\mathbf{2}} \mathrm{O}_{\mathbf{2}}, \boldsymbol{\Delta}\) gives out alkene
\(S\). Treatment of \(\mathrm{CS}_{2}\) produces dithídeltbamic acid

\title{
7. \\ Match \\ the \\ following
}
( 1111111
(a) \(\mathrm{C}_{2} \mathrm{H}_{5} \mathrm{NH}_{2}\) and \(\mathrm{C}_{6} \mathrm{H}_{5} \mathrm{NH}_{2}\)
(b) \(\left(\mathrm{C}_{2} \mathrm{H}_{5}\right)_{3} \mathrm{~N}\) and \(\left(\mathrm{C}_{2} \mathrm{H}_{5}\right)_{2} \mathrm{NH}\)
(c) \(\mathrm{C}_{2} \mathrm{H}_{5} \mathrm{NH}_{2}\) and \(\left(\mathrm{C}_{2} \mathrm{H}_{5}\right)_{3} \mathrm{~N}\)
(d) \(\left(\mathrm{C}_{2} \mathrm{H}_{5}\right)_{3} \mathrm{~N}\) and \(\mathrm{C}_{6} \mathrm{H}_{5} \mathrm{NH}_{2}\)

\section*{Column (II)}
(Distinguished by)
\(P\). Carbylamine test
\(Q\). Azo dye test
R. Hinsberg reagents
\(S\). Liebermann nitroso reaction

\section*{( Watch Video Solution}
8. Match
the
following

\section*{Column (II)}
(a) \(\mathrm{H}, ~ \mathrm{NH}_{3} \mathrm{Cl}\)
h) 110

c) HO

(d)

P. Na extract of compound gives prussian blue colour with \(\mathrm{FeSO}_{4}\)
Q. Positive \(\mathrm{FeCl}_{3}\) test
\(R\). White ppt. with \(\mathrm{AgNO}_{3}\)
S. React with aldehyde to form the corresponding hydrazone derivative
9.
Match
the
following
columns
(atumn (1)
(a) \(\mathrm{C}_{2} \mathrm{H}_{5}-\mathrm{NH}_{2}\)
(b) \(\left(\mathrm{C}_{2} \mathrm{H}_{5}\right)_{2} \mathrm{NH}\)
(c) \(\left(\mathrm{C}_{2} \mathrm{H}_{5}\right)_{3} \mathrm{~N}\)
(d) \(\mathrm{C}_{6} \mathrm{H}_{5} \mathrm{NH}_{2}\)
Column (II)
P. Reaction with \(\mathrm{NaNO}_{2}+\mathrm{HCl}\)
Q. Reaction with \(\mathrm{CHCl}_{3}+\mathrm{KeH}\)
\(R\). Formation of N -nitros, de: with \(\mathrm{HNO}_{2}\)
S. Formation of tricthylammonum nitroso with H NO ,

\section*{- Watch Video Solution}

\section*{Integer Answer Type Problems}
1. Find out number of reaction which involve electron deficient nitrogen during reaction mechanism.
(b) \(\mathrm{H}_{3} \mathrm{C}-\mathrm{C}-\mathrm{Ph}\)
\(\xrightarrow{\mathrm{PhCO}} \mathrm{H}\)

(e) \(\mathrm{Ph}-\mathrm{C}-\mathrm{OH} \xrightarrow{\mathrm{N}_{3} \mathrm{H} \text {. Dilute } \mathrm{H}_{2} \mathrm{SO}_{4}}\)
(g) \(\quad \mathrm{NH}_{2} \xrightarrow{\mathrm{CHCl}_{3}+\mathrm{KOH}}\)
(f)

(h)


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2. Examine the structal formules of following compounds and identify how many compounds are more basic than aniline.

3. Of the following amines how many can give carbyl amine reaction?



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4. Of the following reactions, how many reaction, are used for the preparation of amines?
(a) \(R-C \equiv N \xrightarrow{\text { LiAIG }_{4}}\) (b) \(R-\stackrel{O}{\|} \mathrm{C}-\mathrm{NH}_{2} \xrightarrow{\mathrm{LiAIH}_{4}}\)
(c) \(\quad \stackrel{\stackrel{O}{\mid}}{\mathrm{C}}-\mathrm{NH}_{2} \xrightarrow{\mathrm{Br}_{2}+\stackrel{\ominus}{\mathrm{O}}}\)
\[
R-\stackrel{\|}{\mathrm{C}}-\mathrm{CH}_{3}+\mathrm{H}_{3} \mathrm{C}-\mathrm{NO}_{2} \xrightarrow{\mathrm{NaOH}}
\]
(a) \(\mathrm{R}-\mathrm{C} \equiv \mathrm{N} \xrightarrow{\mathrm{LiAlH}_{4}}\)



(b)

(d)

(f)

(g) \(\mathrm{R}-\stackrel{\mathrm{O}}{\mathrm{C}}-\mathrm{NH}_{2} \underset{D}{\stackrel{\mathrm{P}_{2} \mathrm{O}_{5}}{\longrightarrow}}(\mathrm{~h}) \mathrm{R}-\mathrm{CH}_{2}-\mathrm{NO}_{2} \xrightarrow{\mathrm{H}_{2}, \mathrm{Ni}}\)

\section*{- Watch Video Solution}
5. Of the following amines how many can be separately by Hofmann's mustard oil reaction?





\section*{Subjective Type Problems}
1. Find out finla products of following reactions:


\section*{D Watch Video Solution}

3. Match
the
following
columns

\[
\begin{aligned}
& \mid \mathrm{HNO}_{2} \\
& \mathrm{~S}
\end{aligned}
\]

\section*{( Watch Video Solution}
4.

Match
the
following
columns

5. Match the following columns
```

