





# **CHEMISTRY**

# BOOKS - GR BATHLA & SONS CHEMISTRY (HINGLISH)

# AMINES

Level 1 (Q.1 To Q.25)

**1.** Acetamide is treated separately with the following reagens. Which one of these given methylamine?

A.  $PCI_5$ 

B. Sodalime

 $\mathsf{C.} NaOH + Br_2$ 

D. Hot concentrated  $H_2SO_4$ 

#### Answer: C

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**2.** Reaction of  $RCONH_2$  with a mixture of  $Br_2$  and KOH gives  $RNH_2$  as the main product. The intermediates involved in the reaction are:

$$\stackrel{OH}{\stackrel{|}{=}} N - Br$$
A.  $R - \stackrel{|}{C} = N - Br$ B.  $R - NHBr$ C.  $R - N = C = O$ 



Answer: C

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**3.** Which is the best method of preparing  $2^\circ$  amine ?

$$\begin{array}{l} \mathsf{A.} \ CH_{3}CI + NH_{3} \rightarrow \\ \\ \mathsf{B.} \ CH_{3}CI \xrightarrow{KCN} \stackrel{Sn/HCI}{\longrightarrow} \end{array} \\ \\ \mathsf{C.} \ CH_{3}CI \xrightarrow{KCN} \stackrel{LiAIH_{4}}{\longrightarrow} \end{array} \\ \\ \mathsf{D.} \ CH_{3}NH_{2} \xrightarrow{CHCI_{3}/KOH} \stackrel{Sn/HCI}{\longrightarrow} \end{array}$$

Answer: D



4. Ethyl cyanide (A) can be converted to ethyl amine (B) by:

A. 
$$A \xrightarrow{Sn/HCI} B$$
  
B.  $A \xrightarrow{H_3O^+} \xrightarrow{NH_3/\Delta} \xrightarrow{KBrO/\Delta} B$   
C.  $A \xrightarrow{LiAIH_4} B$ 

D. both (a),(c) are correct

#### Answer: B



5. In Gabriel synthesis, amine is always:

A. alphatic primary amine

- B. aliphatic secondary amine
- C. aromatic primary amine
- D. aromatic secondary amine

Answer: A

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6. In Gabriel synthesis, halide may be:

A. benzyl halide

B. allyl halide

C. both (a), (c) are correct

D. tertiary alkyl halide

#### Answer: C



will be:

A.  $C_6H_5-CN$ 

 $\mathsf{B.}\, C_6H_5NC$ 

 $\mathsf{C}.\, C_6H_5-CH_2-NH_2$ 

 $\mathsf{D.}\, C_6H_5-CH_2OH$ 

# Answer: C



Answer: C



9. Which of the following statement is not correct?

A. Aliphatic amines are stronger bases than ammonia B. Aromatic amines are stronger bases than ammonia

C. The alkyl group in alkyl ammonium ion more

stabilizes the ion relative to the amine

D. The aryl group in aryl ammonium ion less stabilizes

the ion relative to the amine

#### Answer: B



**10.** The correct sequence regarding base strength of aliphatic amines in aqueous solution is:

A. 
$$R_3N>R_2NH>RNH_2>NH_3$$

 $\mathsf{B.}\,R_2NH > RNH_2 > R_3N > NH_3$ 

 $\mathsf{C.}\,R_2NH > R_3N > RNH_2 > NH_3$ 

D.  $RNH_2 > R_2NH > R_3N > NH_3$ 

#### **Answer: B**

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**11.** Decreasing order of basicity of the three isomers of nitro aniline is:

A. p-nitroaniline $> o$ -nitroaniline $> m$ -nitroaniline	
B. p-nitroaniline $> m$ -nitroaniline $> o$ -nitroal	inine
C. m-nitroaniline $> p-$ nitroaniline	> 0 -
nitroaniline	
D. m-nitroaniline > $o$ - nitroaniline	> p -
nitroaniline	

Answer: C

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12. Strongest base is:

A. (a) 
$$H_2N-C-NH_2$$



#### **Answer: A**



13. Which is the best leaving group ?

A. 
$$-\stackrel{\oplus}{N}\equiv N$$

 $\mathsf{B.}\,OH^{\,-}$ 

 $\mathsf{C}.\,NH_2^{\,-}$ 

# D. $CH_3COO^-$

#### Answer: A



14. Which is most volatile ?

A.  $CH_3CH_2CH_2NH_2$ 



D.  $CH_3OH$ 

#### **Answer: B**



**15.** Which one of the following is used as phase transfer catalyst?

A. Primary amine

B. Quaternary ammonium salt

C. Tertiary nitroalkane

D. Tertiary amine

**Answer: B** 



**16.** Which of the following is most basic?







17. Predict about the relative boiling point of the following

two amines.



A. Boiling point of I>II

B. Boling point of II > I

C. Both should have equal boiling points

D. It can't be predicted

#### Answer: B



**18.** Carbylamine test is performed in alcoholic *KOH* by heating a mixture of:

A. chloroform and silver silver powder

B. chloroform and a primary amine

C. an alkyl halide and a primary amine

D. an alkyl cyanide and a primary amine

#### Answer: B

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19. Which of the following compounds is an imine ?

A. 
$$Ph - \overset{O}{\overset{||}{C}} - NH_2$$

B. 
$$CH_3 - CH_2 - N = CH_2$$



#### Answer: B

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20. Which of the following statements is not correct?

A. Primary amines show intermolecular hydrogen

bonding.

- B. Secondary amines show intermolecular hydrogen bonding
- C. Tertiary amines show intermolecular hydrogen

bonding

D. Amines have lower boiling points as compared to

those of alcohols and carboxylic acids of comparable

molar masses.

Answer: C

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**21.** Which of the following amines form N - nitroso derivative when treated with  $NaNO_2$  and HCI?

A.  $CH_3NH_2$ 







## Answer: C



- **22.** Hinsberg's reagent is:
  - A. phenylisocyanide
  - B. benzensulphonyl chloride
  - C. p-toluenesulphonic acid
  - D. o-dichlorobenzene

#### Answer: B



23. Which of the following compounds is an emamine?



$$B. (b) >= N - CH_3$$

$$\mathsf{C}.\, Ph-N=N-Ph$$

D. 
$$Ph-N=N-H$$

#### Answer: A



A. Hofmann elimination

B. Cope reaction

C. Saytzeff reaction

D. Carbyl amine reaction

Answer: B



25. Cope reaction is:

- A.  $S_N 1$  intramolecular
- B.  $S_N 2$  intramolecular
- C.  $E_1$  intramolecular
- D.  $E_c$  or  $E_i$  intramolecular

#### Answer: D

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# Level 1 (Q.26 To Q.41)

1. Which of the following is Hofmann mustard oil reaction?

A. Reaction of primary amine with  $CHCI_3$ 

B. Reaction of primary amine with  $CHCI_3 + KOH$ 

C. Reaction of primary amine with  $CS_2 + HgCI_2$ 

D. Reaction of armoatic amine with iodoform

#### Answer: C



#### product B is:



### Answer: B



A. R - CONHBrB.  $\begin{bmatrix} 0 \\ || \\ R - C - \overline{N} - Br \end{bmatrix} Na^+$ 

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$$\mathsf{C}.\,R-N=C=O$$

D. all of these

Answer: D



Product















5. Which of the following compounds can form alcohol with  $NaNO_2/HCl$ ?



$$\mathsf{C}.\,CH_3-CH_2-NH_2$$

D. All of these

#### Answer: D



**6.** Which of the following will not react with  $CS_2$ ?



#### Answer: D



A.  $CH_2 = CH_2$ 

 $\mathsf{B.}\,CH_2=CHCI$ 

C. 1: 1 ratio of (a) and (b)

 $\mathsf{D}.\,CH_3-CH_2-CI$ 

#### Answer: B

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8. Predict the nature of P in the following reaction:  $Me_3CCH_2NH_2 \xrightarrow{HOMO} P$  (main product)

A.  $Me_3CCH_2OH$ 

 $\mathsf{B}.\, Me_2CCH=CH_2$ 

 $\mathsf{C}.\, Me_2C(OH)C_2H_5$ 

D.  $Me_3CCH_2NH(NO)$ 

#### Answer: C





#### Answer: C







D. none of these

#### Answer: C



# 11. Which of the following is an enamine?





12. Which of the following amines will react with cyclohexanone to give enamine?



(b)



 $NH_2$ 



## Answer: D



$$\stackrel{NaN_3}{ ext{Heat}} R - NH_2$$
 are:

A. 
$$R-\overset{O}{\overset{||}{C}}-\overline{N}-\overset{\oplus}{N}\equiv N$$

$$\mathsf{B}.\,R-N=C=O$$

 $\mathsf{C.}\,R-CNO$ 

D. none of these

#### Answer: A::B

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**14.** Compound  $[X]C_4H_{11}N$  reacts with p-toluence sulphonyl chloride in aqueous NaOH to give a solid. The compound [X] is:

A. 
$$CH_3-CH_2-CH_2-CH_2-NH_2$$

B. 
$$H_3C-CH-CH_2-NH_2$$
  
 $\downarrow_{CH_3}$   
C.  $CH_3CH_2-NH-CH_2-CH_3$   
D.  $H_3C-CH_2-NH-CH_3$   
 $\downarrow_{CH_3}$ 

#### Answer: D



**15.** Cyclohexanol can be converted into cyclohexylamine by following two routes. Which of the following methods is expected to give good yield of cyclohexyalmine?

A. (a) 
$$- \operatorname{OH} \xrightarrow{K_{LOO}} = 0 \xrightarrow{\operatorname{NH}_{1} \operatorname{H}_{2} \operatorname{NH}_{2}} - \operatorname{NH}_{2}$$


C. both are equally suitable

D. neither of the two

#### Answer: A

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**16.** Which of the following will give unsymmetrical disubstituted urea after reaction with  $CH_3NH_2$ ?

A.  $COCI_2$ 

 $\mathsf{B.}\,CH_3CH_2NCS$ 

 $\mathsf{C.}\,CH_3CH_2NCO$ 

D. all of these



## Level 2 (Q.1 To Q.25)

1. Which of the following compounds is an amine?







**2.** Which of following compounds exists as non-resolvable racemic mixture?







#### Answer: D



3. Which of the following compounds loses optical activity

due to pyramidal inversion?



#### Answer: D



 $\xrightarrow{AgCN} A \xrightarrow{H_2, Ni} B$ Br The final 4.

product (B) is:









D.

#### Answer: B



The end product B of the above reaction is:





**6.** 
$$CH_3CH_2Br \xrightarrow{AgCN} A \xrightarrow[H_3O]{NaOH, \Delta} B, (B)$$
 is:

A.  $CH_3CH_2NHCH_3$ 

B.  $CH_3CH_2CH_2NH_2$ 

#### $\mathsf{C.}\,CH_3CH_2NH_2$



#### Answer: C





A. 
$$Ph - \underset{|_{H}}{N} - CH_{3}$$
  
B.  $Ph - CH_{2} - NH_{2}$   
C.  $Ph - \underset{|_{H}}{N} - \overset{O}{\overset{||}{C}} - H$   
D.  $Ph - \overset{\oplus}{N} \equiv \overset{O}{C}$ 

#### Answer: A



8. The major product formed in the reaction:





#### Answer: A



product (Y) is:





#### Answer: B



**10.** Among the following compounds which one will produce a Schiff base on reaction with cyclopentanone?





11. In which of the following reactions does the amine

behaves as an acid ?

A.  $(C_2H_5)_2NH + H_2PtCI_6$ 

 $\mathsf{B.}\,CH_3NH_2+H_2O$ 

C.  $(Me_2CH)_2NH + n - C_4H_9Li$ 

D.  $(C_2H_5)_3\overset{\cdot\cdot}{N}+BF_3$ 

Answer: C

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**12.** Consider the following sequence of reactions:











#### Answer: A



## 13. Consider the following sequence of reactions :

$$H_2C = CH - CH = CH_2 \xrightarrow{Br_1(1 ext{mole})} A \xrightarrow{1.KCN( ext{excess})} B$$

The end product (B) is:

A. 
$$H_2N - \left( CH_2 
ight)_2 - CH = CH - \left( CH_2 
ight)_2 - NH_2$$

B. 
$$H_2N - (CH_2)_6 - NH_2$$

$$\mathsf{C}.\,NC-CH_2-CH=CH-CH_2-CN$$

D. 
$$H_2C=CH-CH-(CH_2)_2-NH_2$$

#### **Answer: B**



14.

$$H_3C-CH_2- \overset{CH_3}{\overset{|}{igcap}}_{igcap} - CH - \overset{CH_3}{\overset{|}{igcap}}_{O} - NH_2 \stackrel{Br_2+KOH}{\longrightarrow} A \stackrel{1.CH_3I\,(\, ext{excess}\,)}{\overset{2.AgOH\,,\,\Delta}{\longrightarrow}} B$$

The major product (B) is:

A.  $CH_3 - CH_2 - CH = CH_2$ 

В. 📄

 $\mathsf{C}.\,H_3C-CH=CH-CH_3$ 



#### Answer: A





#### 15.



The major product (B) is:







**16.** Consider the following sequence of reaction:

$$HO- \stackrel{O}{C} - \stackrel{CH_3}{ \stackrel{||}{C} - \stackrel{|}{C} - \stackrel{C}{C} = N \stackrel{200^{\circ}C}{ \longrightarrow} A \stackrel{LiAIH_4}{ \stackrel{H_2O}{ \longrightarrow} B, (B) ext{ is:}$$







#### Answer: D



## **17.** The major product (X) of the reaction is:





**18.** Which of the following compounds does not liberate  $N_2$  on treatement with  $HNO_2$ ?



#### Answer: D





**19.** The product formed in the reaction is:





C. 
$$H_{3}C - \frac{N}{|H|} - (CH_{2})_{3} - \overset{O}{C} - H$$
  
H  
D.  $H_{3}C - \frac{N}{|H|} - (CH_{2})_{3} - CH_{2}OH$ 

#### Answer: B



20. The major product (B) formed in the reaction sequence

is:













#### Answer: A



**21.** An organic compound (A) on reduction gives a compound (B) which on reaction with  $CHCI_3$  and NaOH

form (C). The compound (C) on catalytic reduction gives Nmethylaniline. The compound (A) is:



$$\overset{(d)}{\bigcup}\overset{\frown}{\longrightarrow} \operatorname{NH}_2$$

#### Answer: A



**22.** The major end product (B) of the reaction:









**23.** Which one among the following is expected to form a secondary alcohol on treatment with  $HNO_2$ ?



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#### 24. The end product (B) of the reaction sequence:





#### Answer: B

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## 25.

 $\xrightarrow{1.Br_2^- + KOH, \Delta}$  The product of above reaction is:

2 .  $H_3O$   $\oplus$ 



#### Answer: D

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1. The major product (X) of the reaction is:











# **Answer: B** Watch Video Solution 2. The major product of the reaction is: NH2 $\xrightarrow{\text{NaNO}_2}$ dil. HCl, 0°C $NH_2$ ${\stackrel{\oplus}{\underset{\sim}{N_2}}} Cl {\stackrel{\ominus}{\longrightarrow}}$ (a) ${\stackrel{\tiny{\tiny{\oplus}}}{\scriptstyle{N_2}Cl}} {\stackrel{\scriptstyle{\tiny{\oplus}}}{\scriptstyle{N_2}cl}}$ A. ${\stackrel{\oplus}{\mathbb{N}}}_2 Cl^{\ominus}$ - (b) $NH_2$ Β.


# Answer: C

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**3.** The reaction of p-aminophenol with one mole of actyl chloride in presence of pyridine gives:









C.



# Answer: D









#### Answer: A



# 5. Which of the following is the stongest Bronsted acid?





# Answer: C



6. Which of the following is the strongest Bronsted base?



# Answer: A Watch Video Solution

# 7. Which of the following is the weakest Bronsted base?



# Answer: A



8. Which of the following is stronger Bronsted base?





#### Answer: D



9. For the following compounds, which is the strongest

base and which is strongest acid?



A. II = Strongest base, I = Strongest acid

- B. IV=Stronger base, III=Stronger acid
- C. III = Strongest base, IV = Strongest acid
- D. II = Strongest base, III = Strongest acid

#### Answer: C



10. Which compound is the likely from following reaction?



#### **Answer: B**











A.  $LiAIH_4, 3CH_3I/AgOH, \Delta$ 

B.  $LiAIH_4, P_2O_5/\Delta$ 

C.  $20\,\%\,H_2SO_4\,/\,\Delta,\,P_2O_5\,/\,\Delta$ 

D.  $H_2, Pd - BaSO_4$ 

Answer: A



**13.** What is the likely product from the following reaction?





# Answer: B

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14. Repeated Hofmann elimination reaction (exhaustive methylation followed by heating with AgOH) will often remove a nitrogen atom from an amine molecule. Which of the following compounds is the likely product in this case?





# Answer: B



**15.** Only one of the following amines will lose its nitrogen atom as trimethyl amine by repeated Hofmann elimination reactions:



D.

# Answer: D



**16.** The nitrogan atom in each of the following tertiary amines may be removed as trimethyl amine by repeated Hoffmann elimination. Which of the following amines requires the greater number of Hofmann sequence to accomplish this?









# Answer: A



17. The Hinsberg test of a  $C_5H_{14}N_2$  compound produces a solid that is insoluble in 10% aq. NaOH. This solid derivative dissolves in 10% aq.  $H_2SO_4$ . Which of the following would best fit these facts ?





Answer: B

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18. What set of conditions would be useful for preparing a

 $2^\circ$  amine?

A.  $2^\circ R - Br + NaNH_2$ 

B.  $2^2R-Br+NaN_3, H_2\,/\,Pt$ 

C.  $1^{\circ} R - NH_2 + 1^{\circ} RCHO, H_2$  and Pt











# Answer: D

C.



**20.** Consider the following sequence of reactions:



Identify

# product C:









D.



 $\mathsf{B}.\,H_3C-CH=CD_2$ 

 $\mathsf{C}.\,H_2C=CH-CD_3$ 

 $\mathsf{D}.\,H_2C=N-CH_3$ 

#### Answer: C



22. The product formed in the reaction is :









# Answer: B





B:





D.

# Answer: D



final product (B) is:









# Answer: A



**25.** The final major product of the reaction is:



$$\mathsf{D}. \ Ph = egin{array}{ccc} OH & OH \ & OH \ & & | \ & | \ & | \ & | \ & | \ & | \ & Ph \ & CH_3 \end{array} = CH_3$$

#### Answer: C

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**1.** The major product of the reaction:





# Answer: D



2. The end product of the following reaction is:



![](_page_101_Figure_0.jpeg)

# Answer: A

![](_page_101_Figure_2.jpeg)

# 3. Consider the following sequence of reactions:

$$\underbrace{ \begin{array}{c} & & \\ &$$

The products (B) and (C) are:

![](_page_101_Figure_6.jpeg)

![](_page_102_Figure_0.jpeg)

![](_page_102_Figure_1.jpeg)

#### Answer: B

![](_page_102_Figure_3.jpeg)

**4.** Which of the following compounds will react with cyclopentanone to form an enamine?

![](_page_102_Figure_5.jpeg)

![](_page_103_Picture_0.jpeg)

# Answer: C

![](_page_103_Figure_2.jpeg)

# **5.** Predict the major product 'X' in the following reaction:

![](_page_103_Picture_4.jpeg)

![](_page_104_Figure_0.jpeg)

Answer: A

![](_page_104_Picture_2.jpeg)

![](_page_105_Figure_0.jpeg)

of this reaction is:

![](_page_105_Picture_2.jpeg)

#### Answer: D

![](_page_105_Picture_4.jpeg)

![](_page_106_Figure_0.jpeg)

Compound B is :

7.

![](_page_106_Figure_2.jpeg)

![](_page_106_Figure_3.jpeg)

# Answer: B

D.

![](_page_106_Picture_5.jpeg)

![](_page_107_Figure_0.jpeg)

Product

(X) is :

![](_page_107_Figure_3.jpeg)

![](_page_107_Picture_4.jpeg)

![](_page_107_Picture_5.jpeg)

![](_page_107_Figure_6.jpeg)










(Y) of the reaction:





C. Mixture of (a) and (b)



#### Answer: A



















#### Answer: D





The product M is:







Final product 'O' is:





## Answer: C



The order of reactivity towards diazo coupling with phenol

in presence of dil. NaOH:

A. 
$$P > Q > R > S$$
  
B.  $Q > S > R > P$   
C.  $P > R > S > Q$ 

$$\mathsf{D}.\,S>R>Q>P$$

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16. Reaction I 
$$Ph - \overset{O}{\overset{||}{C}} - NH_2 \stackrel{\overset{\Theta}{\longrightarrow}}{\overset{O}{\longrightarrow}} A$$

Reaction II 
$$Ph - \overset{
ightarrow}{C} - ND_2 \stackrel{\overset{
ightarrow}{OH}, Br_2}{\longrightarrow} B$$

Products A and B are:

A.  $Ph-NH_2$  and  $Ph-ND_2$ 

B.  $Ph - ND_2$  and  $Ph - NH_2$ 

C. Both  $Ph - NH_2$ 

D. Both  $Ph - ND_2$ 



**17.** An organic compound (A)  $C_9H_{13}N$  dissolves in dil. *HCI* and releases  $N_2$  with  $HNO_2$  giving an optically active alcohol. Alcohol ox oxidation gives dicarboxylic acid, which on heating form anhydride. The organic compound 'A' is:





## Answer: C





The final product is:









### Answer: A



**19.** Identify 'X' in the following sequence of reaction:

A. Benzoic acid

B. Phenyl acetic acid

C. Benzyl alcohol

D. Benzamide

#### Answer: C



20. Which sequence of steps will be able to produce 3,3'-

dinitro-biphenyl from benzene?

A.  $HNO_3/H_2SO_4, CI_2/FeCI_3, Na/ether$ 

B.  $CI_2$  /  $FeCI_3$ ,  $HNO_3$ ,  $H_2SO_4$ , Na / ether

 $\mathsf{C}.\,CI_2\,/\,FeCI_3,\,H_2SO_4,\,Na\,/\,\mathrm{ether}$ 

 $\mathsf{D}.\,I_2\,/\,HIO_3,\,CI_2\,/\,FeCI_3,\,C_6H_5NO_2$ 

Answer: A

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**21.**  $1^{\circ}, 2^{\circ}$  and  $3^{\circ}$  nitroalkane can be identified by action of:

A.  $HNO_3 + NaOH$  (aq).

B.  $CHCI_3 + NaOH$  (aq)

C.  $HNO_2 + NaOH$  (aq)

D.  $CHCI_3 + KOH$  (aq.)

#### Answer: C



**22.** A compound 'X' when reacted with  $PCI_5$  and then with  $NH_3$  gives 'Y'. When 'Y' treated with  $Br_2$  and KOH produced 'Z'. Z on treatement with  $NaNO_2 + HCI$  at  $0^{\circ}C$  and then boiling produced ortho-cresol. Compound 'X' is:

A. o-toluic acid

B. o-chlorotoluene

C. o-bromotoluene

D. m-toluic acid

#### Answer: A





For such kind of diazo- coupling reaction the suitable substituents P and S are respectively:

A. 
$$-NH_2$$
 and  $-OCH_3$   
B.  $-NO_2$  and  $-C - H$ 

C. 
$$-NH_2$$
 and  $-NHCH_3$ 

(d) 
$$-OCH_3$$
 and  $-N_{0}^{O}$ 

## Answer: D





#### Answer: D

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25. The final product B obtained in the reaction is:

 $\begin{array}{c} \text{CH}_{3}\text{CH}_{2} \\ \text{CH}_{3}\text{CH}_{2} \\ \text{CH}_{3}\text{CH}_{2} \end{array} N \xrightarrow{\text{H}_{2}\text{O}_{2}} A \xrightarrow{\Delta} B + \text{H}_{2}\text{C} = \text{CH}_{2} \\ \text{CH}_{3}\text{CH}_{2} \end{array}$ 



 $\mathsf{B.}\left(H_2C=CH\right)_2$ 

 $\mathsf{C}.\,H_3C-CH=CH-CH_2$ 



#### Answer: A



Level 2 (Q.76 To Q.89)



Find out Y of the reaction :







## Answer: C





## The compound C is:





# The compound B is:









#### Answer: D

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5. Identify the final product of following reaction:













The product can be given as:



## Answer: D



8. In a set of reactions acid yielded a compound D.

$$\bigwedge_{O}^{OH} \xrightarrow{\text{SOCl}_2} B \xrightarrow{\text{NH}_3} C \xrightarrow{\mathbf{K} \ \mathbf{H}} B\mathbf{r}_2 \rightarrow D$$

The strucuture of D would be :





9. What would be the final product of reaction?



#### Answer: A



**10.** Identify major product of following sequence of reaction:



#### Answer: C



**11.** Identify the final producy of following sequence of reaction:

$$\underbrace{\bigcirc} \xrightarrow{N_2O_5} \underbrace{Cl_2}_{AlCl_3} \xrightarrow{H_2}_{Pd-C} \xrightarrow{NaNO_2}_{HCl} \xrightarrow{H_2O}_{HCl}$$











## **12.** Identify the major product of following reaction:









D. none of these

### Answer: C



## **13.** Identify final product of following sequence of reaction:











## Answer: C


14. What is the product of following reaction sequence?





#### Answer: B

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# More Than One Correct (Q.1 To Q.25)

**1.** The presence of primary amine can be confirmed by its reaction with:

A.  $HNO_2$ 

 $\mathsf{B.} CHCI_3 + NaOH$ 

C.  $CS_2$  and  $HgCI_2$ 

D.  $H_2SO_4$ 

#### Answer: A::B::C

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**2.** Which of the following reactions can be used to make ethyl isocyanide?

A. 
$$CH_3CH_2NH_2 + CHCI_3 \xrightarrow{KOH}{\Delta}$$

B.  $CH_{3}CH_{2}Br + AgCN 
ightarrow$ 

$$\mathsf{C.} CH_3CH_2 - NH - \underset{||}{C} - H \xrightarrow[]{POCI_3}{O}$$

D.  $CH_{3}CH_{2}Br+KCN
ightarrow$ 

Answer: A::B::C



3. By which of the following reactions can methylcyanide be

prepared?

A. 
$$CH_3Br \xrightarrow{KCN}_{DMF}$$
  
B.  $CH_3NH_2 + CHCI_3 \xrightarrow{KOH}$   
C.  $CH_3 - CH = N - OH \xrightarrow{P_2O_5}{\Delta}$   
D.  $CH_3 - \overset{O}{C} - NH_2 \xrightarrow{P_4O_{10}}{\Delta}$ 

Answer: A::C::D



Answer: B::C::D



5. Consider the following reaction:

 $A \xrightarrow{K_2 \operatorname{Cr}_2 \operatorname{O}_7, \operatorname{H}^{\oplus}} \operatorname{O}^{=}$  $\bigcirc$ 

The staring substance 'A' can be:







Answer: A::B::D

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**6.** 
$$C_6H_5-CH_2-I \stackrel{NaN_3}{\wedge}$$
 Products

Reaction is assumed to involve nitrene as intermediate,

then various possible products are:

A.  $C_6H_5CN_2NH_2$ 

 $\mathsf{B.}\, C_6H_5N=CH_2$ 

 $\mathsf{C.}\, C_6H_5CN=NH$ 



Answer: B::C



## 7. Which of the following can give $1^{\circ}$ amine?

$$\overset{CH_{3}}{\mathsf{A.}}Ph - \overset{|}{\overset{CH}{CH}} - OH \xrightarrow{NaCN\,,H^{\,\oplus}}$$

 $\begin{array}{c} O \\ \mathsf{B}. \, Ph - CH = CH - \overset{O}{CH} - NH_2 \stackrel{NaOCI}{\underset{CH_3OH}{\longrightarrow}} \end{array}$ 

C. 
$$Ph-C\equiv C-\overset{O}{\overset{||}{C}}-NH_{2}\overset{NaOBr}{\longrightarrow}$$

D. 
$$Ph - \overset{|\,|}{C} - CI \stackrel{NaN_3}{\longrightarrow} \stackrel{LiAIH_4}{\longrightarrow}$$

#### Answer: B::C::D



## 8. Which of the following can distinguish?

$$CH_3 egin{array}{ccc} H_3 & H \ dots \\ H_3 - CH - NH_2 ext{ and } CH_3 - N - CH_3 \end{array}$$

A.  $(COOC_2H_5)_2$ 

B.  $NaNO_2 + HCI$ 

 $C. Cs_2, HgCI_2$ 

D.  $Ag_2O/\Delta$ 

### Answer: A::B::C



disease:



Which of the following statements about this compound are correct?

A. It can exist only in optically active forms

B. One mole will react with 3 mole of NaOH to form a

C. It can exist as a zwitter ion in the aqueous solution

D. It given nitroso compound on treatement with

 $HNO_2$ 

Answer: A::B::C

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**10.** Which of the following give nitrosonamine on treatment with  $HNO_2$ ?

$$(a) \qquad N - H$$

B. 
$$H_3C-CH-CH_2-NH_2$$



### Answer: A::C



11. Which of the following sequence of reagent is the good

mean to furnish the conversion?

 $R-CH_2OH 
ightarrow R-CH_2NH_2$ 

A.  $KMnO_4, SOCI_2, NH_3, \Delta, NaOBr$ 

B.  $SOCI_2, NaCN, H_2 / Ni$ 

C.  $CrO_3$  in dilute acetone,  $NH_3, H_2, Ni$ 

D.  $Cu, 300^{\,\circ}C, NH_2, LiAiH_4$ 

#### Answer: A::B::C



12. Choose the correct comparisons of basicity:



Answer: A::B::C

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**13.** Which of the following arrangements are correct with respect to the property of the compounds indicated in the parentheses?

A.  $HCOOH > CH_3COOH > CH_3CH_2COOH$ 

(Acidic strength)

 $B_{\rm c}$  (b) F COOH > CI COOH > Br COOH (Acidic strength)

 $C \xrightarrow{(c) Ph - C - OH > Ph - OH > OH > OH > OH (Acidic strength)}_{O}$ 

 $D_{\bullet} \xrightarrow{(d) Ph - \vec{N}H_2 > \sqrt{-NH_2 > Ph - \vec{C} - \vec{N}H_2 (Base strength)}}$ 

Answer: A::B::C



#### Answer: A::C::D



**15.** Which of the following will give Hofmann-Bromoamide reaction?





### Answer: A::B::C



**16.** Which of the following reactions represent major products?





## Answer: A::B::C



17. Which of the following products will not form by

## following reaction?



#### Answer: A::C::D



18.

$$Ph - \overset{O}{\overset{||}{C}} - NH_2 + Ph - CH_2 - \overset{O}{\overset{||}{C}} - \overset{O}{\overset{NH_2}{NH_2}} \overset{O}{\overset{O}{\longrightarrow}} A + B$$

Products A and B are:

A.  $Ph-NH_2$ B.  $Ph-CH_2\overset{15}{N}H_2$ C.  $Ph-CH_2-NH_2$ D.  $Ph-\overset{15}{N}H_2$ 

Answer: A::B



**19.** Reaction involves isocyanate as intermediate porudct:

- A. Curtius rearrangement
- B. Lossen rearrangement
- C. Schmidt rearrangemnt
- D. Hofmann rearrangement

### Answer: A::B::C::D

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## **20.** Consider the structures:



Which of the following statements are correct?

A. Basic stength of II is greater than I

B. Basic strength of II is less than that of I

C. Basic strength of IV is greater than III

D. Basic strength of IV is less than that of III

Answer: A::C

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**21.** Which of the following give L Liebermann nitroso reaction?





#### Answer: A::B



## 22. Which are related with Curtius rearrangement?

A.  $NaN_3$ 

B.  $R-NH_2$ C.  $R-\overset{O}{\overset{||}{C}}-CI$ D.  $R-\overset{O}{\overset{||}{C}}-OH$ 

### Answer: A::B::C



#### Answer: A::C

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24. Which of the following give Schiff base with aldehyde?

A.  $CH_3CH_2NH_2$ 

B.  $C_6H_5 - CH_2 - NH_2$ 

 $\mathsf{C.}\,C_6H_5-NH_2$ 

D.  $C_6H_5 - NO_2$ 

Answer: A::B::C

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**25.** Which of the following give(s) aniline by reduction of nitrobenzene?

- A.  $H_2 \,/\, Pd C$
- $\mathsf{B.}\,Sn+HCI$
- $\mathsf{C}.\,Cu+HCI$
- $\mathsf{D}.\,(NH_4)_2S$

Answer: A::B::C::D

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More Than One Correct (Q.26 To Q.30)

**1.** Optically active amine having molecular formula  $C_5H_{13}N$ on reaction with  $NaNO_2 + HCI$  produces,  $3^\circ$  optically inative alcohol. Find out structures of amines:



### Answer: A::C



**2.** Find out products which are formed by the following reaction:



#### Answer: A::B

**3.** Which of the following is soluble in dil aqueous HCI?

A.  $C_6H_5NH_2$ 

 $\mathsf{B.}\, C_6H_5CH_2NH_2$ 

 $\mathsf{C.}\, C_6H_5CONH_2$ 





4. The structural form of a compound  $A(C_6H_{11}N)$  is resolvable, dissolve in dil. HCI and reacts with  $HNO_3$ . Compound A could be:





5. Which of the following basically exist as dipolar ion?



**1.** The conversion of an amide by reaction NaOH and  $Br_2$ to primary amine that has one carbon than the strating amide is known as Hofmann-Bromoamide reaction.

$$\stackrel{O}{R-C} = NH_2 \stackrel{Br_2+NaOH}{\stackrel{ ext{or }NaOBr}{ ext{or }NaOBr}} R - NH_2 + NaBr + Na_2CO_3$$

Mechanism:



Number of moles of NaOH consumed in above reaction:

A. 1

C. 3

D. 4

Answer: D

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2. The conversion of an amide by reaction NaOH and  $Br_2$ to primary amine that has one carbon than the strating amide is known as Hofmann-Bromoamide reaction.

$$\stackrel{O}{R-C} = NH_2 \stackrel{Br_2+NaOH}{ ext{or}} R - NH_2 + NaBr + Na_2CO_3$$

Mechanism:



Find X and Y:









#### Answer: B



**3.** The conversion of an amide by reaction NaOH and  $Br_2$  to primary amine that has one carbon than the strating amide is known as Hofmann-Bromoamide reaction.

$$\stackrel{O}{R-C} = NH_2 \stackrel{Br_2+NaOH}{\longrightarrow} R - NH_2 + NaBr + Na_2CO_3$$

Mechanism:











D. all of these

#### Answer: A



Find out slowest step of the reaction:
A. I

B. II

C. III

D. IV

Answer: B

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**5.** Ketoxime when heated with certain reagents undergoes rearrangement to form amides. This is known as Backmann's rearrangement.





Find out (X):



D. 
$$ph - \overset{|\,|}{C} - NH - Ph$$

# Answer: C

**6.** Ketoxime when heated with certain reagents undergoes rearrangement to form amides. This is known as Backmann's rearrangement.



Find out (X) of the reaction:





### Answer: D





Which of the following amines cannot be prepared by path-I?



# Answer: C







Consider path II, choose the major product for 1 and 2:



A. 1 - Y, 2 - X

B.1 - X, 2 - Y

C.1 - X, 2 - X

D. 
$$1 - Y, 2 - Y$$

# Answer: A





In the path I, if



then the

amine finally formed is :





C. racemic mixture of a and b

D. none of these



10. An organic compound 'A' has molecular formula  $C_9H_{13}NO$  and it can be resolved into enatiomers. A does not decolourise bromine water solution. A on refixing with dilute  $H_2SO_4$  yields another resolable compound  $B(C_9H_{14}O_3)$  which gives effervescence with  $NaHCO_3$ . B on treatemet with  $NaBH_4$  yields  $C(C_9H_{16}O_3)$  on heating concentrated  $H_2O_4$  yields ester  $D(C_9H_{14}O_2)$ . with Compound A on reduction with  $LiAIH_4$ , followed by treatement of  $H_2SO_4$  yields following compound:



Find out structure of compound 'A' :



### Answer: B

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11. An organic compound 'A' has molecular formula  $C_{0}H_{13}NO$  and it can be resolved into enatiomers. A does not decolourise bromine water solution. A on refixing with dilute  $H_2SO_4$  yields another resolable compound  $B(C_9H_{14}O_3)$  which gives effervescence with  $NaHCO_3$ . B on treatemet with  $NaBH_4$  yields  $C(C_9H_{16}O_3)$  on heating concentrated  $H_2O_4$  yields ester  $D(C_9H_{14}O_2)$ . with Compound A on reduction with  $LiAIH_4$ , followed by treatement of  $H_2SO_4$  yields following compound:



The sweet smelling neutral compound D is:



## Answer: C

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12. An organic compound 'A' has molecular formula  $C_{0}H_{13}NO$  and it can be resolved into enatiomers. A does not decolourise bromine water solution. A on refixing with dilute  $H_2SO_4$  yields another resolable compound  $B(C_9H_{14}O_3)$  which gives effervescence with  $NaHCO_3$ . B on treatemet with  $NaBH_4$  yields  $C(C_9H_{16}O_3)$  on heating with concentrated  $H_2O_4$  yields ester  $D(C_9H_{14}O_2)$ . Compound A on reduction with  $LiAIH_4$ , followed by treatement of  $H_2SO_4$  yields following compound:



Due to reduction of optically pure 'B' two isomeric product

'C' form. Isomeric product 'C' are:

A. Enantiomers

**B.** Diastereomers

C. Position isomers

D. Functional isomers

### Answer: B

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**13.** When an primary aromatic amine is treated with  $NaNO_2 + HCI$  at  $0^\circ - 5^\circ C$ , a diazonium salt is formed and the reaction is called diazo reaction. In this reaction mineral acid must be added to prevent the coulping reaction of diazonium salt with excess of aryl amine. diazonium salt is highly in the synthesis of number of coloured dyes.

For the following diazonium ion the decreasing order of reactivity of these ion in azo-coupling reaction:



A. Q > S > R > P

 $\operatorname{B.} Q > S > P > R$ 

 $\mathsf{C}.\, P > Q > R > S$ 

 $\mathsf{D}.\,S>R>Q>P$ 

#### Answer: B

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14. When an primary aromatic amine is treated with  $NaNO_2 + HCI$  at  $0^\circ - 5^\circ C$ , a diazonium salt is formed and the reaction is called diazo reaction. In this reaction mineral acid must be added to prevent the coulping reaction of diazonium salt with excess of aryl amine. diazonium salt is highly in the synthesis of number of

coloured dyes.

In the given reaction.



The final product is



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**15.** When an primary aromatic amine is treated with  $NaNO_2 + HCI$  at  $0^\circ - 5^\circ C$ , a diazonium salt is formed and the reaction is called diazo reaction. In this reaction mineral acid must be added to prevent the coulping reaction of diazonium salt with excess of aryl amine. diazonium salt is highly in the synthesis of number of coloured dyes.

When 2,4-dinitrophenol react with  $NaNO_2 + HCI$  at  $5^{\circ}C$  followed by reaction with anisole, a coloured compound is formed which can be given as:









### Answer: C



Match The Column

### Match the

1.

### following





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# 2. Match

### the

#### following

### columns



#### 

- P. Treatment of NaNO<sub>2</sub>, HCl gives N-nitroso compound
- Q. Treatment of NaNO2, HCl gives diazoniumchloride
- R. Treatment of excess CH
- S. Treatment of Hell Agrices dealkylation







(b) Curtius rearrangement

- (c) Lossen rearrangement
- (d) Hemiaminal





Match

the





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# **7.** Mat

### Match

### the

# following

columns

#### Calmin (h

- (Notice) (a)  $C_2H_5NH_2$  and  $C_6H_5NH_2$
- (a)  $(C_2H_5)_3N$  and  $(C_2H_5)_2NH$
- (b)  $(C_2H_5)_3H$  and  $(C_2H_5)_3N$ (c)  $C_2H_5NH_2$  and  $(C_2H_5)_3N$
- (d)  $(C_2H_5)_3$  N and  $C_6H_5$ NH<sub>2</sub>

#### Column (II) (Distinguished by)

- P. Carbylamine test
- Q. Azo dye test
- R. Hinsberg reagents
- S. Liebermann nitroso reaction





## 9. Match

## the

# following

### columns

#### Column (1)

(a)  $C_2H_5$  —  $NH_2$ (b)  $(C_2H_5)_2NH$ (c)  $(C_2H_5)_3N$ 

 $(d) C_6 H_5 N H_2$ 

#### Column (II)

- P. Reaction with NaN $O_2$  + HCl
- Q. Reaction with CHCl<sub>3</sub> + K $\bullet$ H
- *R*. Formation of N-nitrosodictry, end, with HNO2 *S*. Formation of triethyl ammonium

nitroso with HNO

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### **Integer Answer Type Problems**

**1.** Find out number of reaction which involve electron deficient nitrogen during reaction mechanism.





**2.** Examine the structal formules of following compounds and identify how many compounds are more basic than aniline.





3. Of the following amines how many can give carbyl amine

# reaction?



**4.** Of the following reactions, how many reaction, are used for the preparation of amines?

(a) 
$$R-C\equiv N \xrightarrow{LiAIh_4}$$
 (b)  $R-\overset{O}{C}-NH_2 \xrightarrow{LiAIH_4}$   
(c)  $R-\overset{O}{C}-NH_2 \xrightarrow{Br_2+\overset{\Theta}{O}H}$  (d)



# 5. Of the following amines how many can be separately by

Hofmann's mustard oil reaction?



# **Subjective Type Problems**

## **1.** Find out finla products of following reactions:











