

### **CHEMISTRY**

# **NCERT - NCERT CHEMISTRY (GUJRATI)**

### **AROMATIC HYDROCARBONS**

**Questions Choose The Best Answer** 

1. Aromatic compounds are

A. benzenoid compounds

- B. non-benzenoid compounds
- C. aliphatic compounds
- D. alicyclic compounds



- 2. Benzene was first isolated by
  - A. Huckel
  - B. Faraday

- C. Hofmann
- D. Barthelot



- **3.** Benzene undergoes
  - A. addition reactions
  - B. oxidation reactions
  - C. polymerisation reactions

D. electrophilic substitution reactions

### **Answer:**



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**4.** The modern theory of aromaticity was introduced by

A. Faraday

B. Hofmann

C. Huckel

D. Berthelot

**Answer:** 



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**5.** Any compound can be aromatic if they have delocalised  $\pi$  electrons.

A. 4n+2

B. 4n + 1

 $\mathsf{C.}\,4n$ 

D. 
$$4n - 2$$



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**6.** The function of FeCl3 in chlorination of benzene is to produce

A. Cl

B.  $Cl^+$ 

 $\mathsf{C}.\,Cl^{\,-}$ 

D. C

### **Answer:**



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7. The ortho and para directing groups are

A. activating group

B. deactivating group

C. both

D. none



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**8.** The purpose of adding conc.  $H_2SO_4$  in nitration of benzene is to produce

A.  $NO_2$ 

 $\mathsf{B}.\,NO_2^-$ 

 $\mathsf{C}.\,NO_2^+$ 

D.  $NO_3^-$ 



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**9.** An example of polycyclic aromatic hydrocarbon

A. pyridine

B. pyrole

C. naphthalene

D. cyclohexane



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**10.** The compound which is used as a solvent for the extraction of fats and oils

- A. naphthalene
- B. benzene
- C. cyclohexane
- D. butane



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## Questions B Fill In The Blanks

**1.** The modern theory of aromaticity was introduced by



2. The ortho and para directing groups are
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3. The ortho and para directing groups are
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4. Alkyl substituted benzenes are prepared by
reaction.
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**5.** Naphtha obtained by fractional distillation of \_\_\_\_\_ is passed over platinum.



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**6.** Aromatic compounds readily undergo substitution reactions.



**7.** In the presence of \_\_\_\_\_benzene reacts with hydrogen to give cyclohexane



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# Questions C Explain Briefly On The Following

1. How is benzene is prepared commercially?



- 2. How would you convert the following?
- a) sodium benzoate to benzene
- b) phenol to benzene
- c) benzene to toluene

