

India's Number 1 Education App

# PHYSICS

# BOOKS - CENGAGE PHYSICS (HINGLISH)

# THERMAL PROPERTIES OF MATTER

**Question Bank** 

**1.** Thegraph AB shown in the figure is a plot

of temperature of a body in degree Celsius

and degree Fahrenheit. If the slope of line'AB

is 
$$\frac{x}{y}$$
, find  $(x + y)$ .

'(##CEN\_KSR\_PHY\_JEE\_C13\_E01\_001\_Q01##)'

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2. Two spherical black bodies, A and B, having radii  $r_A$  and  $r_D$ , where  $r_B = 2r_A$  emit radiations with peak intensities i at wavelengths 400nm and 800nm, respectively. If their temperature are  $T_A$  and  $T_B$ respectively, in Kelvin scale, their emissive powers are  $E_A$  and  $E_B$  and energies emitted per second are  $P_A$  and  $P_g$ , then find  $\left(\frac{P_A}{P_B}\right)$ .



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**3.** A ring consisting of two parts ADB and ACB. of same conductivity K carries an amount of heat H. The ADB part is now replaced with another metal keeping the temperature  $T_1$  and  $T_2$  constant. The heat carried increases to 2H. The conductivity of the new ADB part is  $\frac{x}{3}K$ . Find the value of

$$x$$
, (Given:  $rac{ACB}{ADB}=3$  ) .

'(##CEN\_KSR\_PHY\_JEE\_C13\_E01\_003\_Q02##)'



**4.** Two identical solid spheres have the same temperature. One of the spheres is cut into two identical pieces, The intact spheres radiates energy Q during a given small time interval. During the same interval, the two hemispheres radiate a total energy  $Q^r$ . The ratio  $\frac{Q'}{Q}$  is equal to





5. A cup of tea cools from  $80^{\circ}C$  to  $60^{\circ}C$  in one min. The ambient temperature is  $30^{\circ}C$ . In cooling from  $60^{\circ}C$  to  $0^{\circ}C$  how much time (in s) will it take?

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6. Ice starts forming in lake with water at  $0^{\circ}C$ when the atmospheric temperature is  $-10^{\circ}C$ . If the time taken for 1cm of ice be 7h, then the time taken (in hours) for the thickness of ice

to change from 1cm to 2cm is



7. There is a small hole in a container. At what temperature (in K) should it be maintained in order that it emits one. calorie of energy per second per  $metre^2$ ?

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8. Emissivity e is a property of surface. Suppose for t surface emissivity e varies with Kelvin temperature T as e = CT (C is constant ). If energy emission rate at temperature 600, K from the surface is 160W, what will be the energy emission rate (in watt) at 300K?



9. The energy radiated by a black body at 2000K is found to have the maximum value at a wavelength  $1.5\mu m$ . Its emissive power being  $8000Wm^{-2}$ . When the body is cooled at a temperature T, the emissive power decreases to  $500Wm^{-2}$ . At this temperature T, the maximum of energy distribution occurs at a wavelength  $\mu(m)$ .



**10.** In a 10m deep lake, the bottom is at a constant temperature of  $4^{\circ}C$ . The air temperature is constant at  $-4^{\circ}C$ . The thermal conductivity of ice is 3 times that of water. Neglecting the expansion of water on freezing, the maximum thickness (in m ) of ice will he



**11.** A system receives heat continuously at the rate of 10 W. The temperature of the system becomes constant at  $70^{\,\circ}C$  when the temperature of the surrounding is  $30^{\,\circ}\,C$ . After the heat is switched off, the system cools from  $50^{\,\circ}\,C$  to  $49.9^{\,\circ}\,C$  in  $1\,\min$  . Find the heat capacity  $\Big( \in rac{kJ}{\circ} C$  of the system.

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12. A and B are two points on a uniform metal ring whose centre is C. The angle  $ACB = \theta$ , A and B are maintained at two different constant temperatures. When  $\theta = 180^{\circ}$  i the rate of total heat flow from A to B is 1.2W. When  $\theta = 90^{\circ}$ , this rate (in watt) will be

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**13.** A piece of ice (heat capacity  $=2100^{\circ}(J)(kg)^{-1}(\circ)C^{-1}$  and latent heat  $s=3.36 imes 10^5 Jkg^{-1}$  ) of mass m grams is at  $-5^{\circ}C$  at atmospheric pressure. It is given 420J of heat so that the ice starts melting. Finally when the ice-water mixture is in equilibrium, it is found that 1g of ice has melted. Assuming there is no other heat exchange in the process, the value of m is



**14.** Two identical rectangular rods of metal are welded end to end in series between temperature  $0^oC$  and  $100^\circ C$  and 10J of heat is conducted (in steady state process) through the rod in 2.00 min. If 5 such rods are taken and joined as shown in the figure maintaining the same temperature difference between Aand B, then the time in min in which 20J of heat will flow through the rods is

'(##CEN\_KSR\_PHY\_JEE\_C13\_E01\_014\_Q03##)'



**15.** 5g of steam at  $100^{\circ}C$  is mixed with 10g of ice at  $0^{\circ}C$ . At equilibrium, the mixture contains p gram of steam. Find (p + q). (Given:  $s_{water}) = 1$ cal  $rac{g^{-1}}{2}C^{-1}, L_f = 80rac{cal}{a}, L_v = 540$ (cal) / (g)`) **View Text Solution 16.** The variation of lengths of two metal rods A and B with change in temperature are as

shown in the figure. The ratio of  $\frac{\alpha_B}{\alpha_A}$  is (Here  $\alpha_A$  = Linear coefficient of thermal expansion

of rod  $A_3 lpha_B =$  Linear coefficient of thermal

expansion of rodB .

'(##CEN\_KSR\_PHY\_JEE\_C13\_E01\_016\_Q04##)'

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**17.** Two taps, A and B supply water at temperatures  $10^{\circ}$  and  $50^{\circ}C$ , respectively. Tap A alone fills the tank in 1h and tap B alone fills the tank in 3h. If both the taps are opened simultaneously in an empty tank, then the final temperature of the water in the completely

filled tank is found to be 5a ( $^{o}C$ ). Find the value of a. Neglect loss of heat to the surrounding and heat capacity of the tank,

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**18.** Water of volume 2 liter in a container is heated with a coil of 1kW at  $27^{\circ}C$ . The lid of the container is open and energy dissipates at rate of  $160\frac{J}{s}$ . In how much time (in s) temperature will rise from  $27^{\circ}C$  to  $77^{\circ}C$ ? (Given: Specific heat of water is  $4.2k\frac{J}{k}g$ )





**19.** A baby's bath should be at a temperature of  $30^{\circ}C$ . There is 10kg water at  $12^{\circ}C$  in the bath tub. How much water (in kg ) with temperature  $50^{\circ}C$  should be added to achieve the desired temperature?

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**20.** 2g steam at  $100^{\circ}C$  is mixed with 5g ice at  $-40^{\circ}C$  in an ideal calorimeter. The final

temperature (in 
$${}^{o}C$$
 ) of system is (Givea: $L_r=500calg,\,S_{ice}=0.5calg{}^{-1}{}^{o}C{}^{-1}$  $S_w=1calrac{g{}^{-1}}{{}^{o}}C{}^{-1},\,L_f=80rac{cal}{g}
ight)$ 

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**22.** A refrigerator converts 100g of water at  $25\,^\circ C$  into ice at  $-\,10\,^\circ C$  in 1h and  $50\,\min$  . The quantity of heat (in calorie) removed per min is (specific heat of ice = 0.5 cal  $(g)^{-1} C^{-1}$ , specific heat of water  $=1(cal)rac{\left(g
ight)^{-1}}{\hat{\phantom{aaaa}}}\circ C^{-1}$ , latent. heat of fusion  $-80\frac{cal}{a}$  )



**23.** A calorimeter of negligible heat capacity contains 50m. of water at  $40^{\circ}C$ . The water cools to  $35^{\circ}C$  in 50 min . The water is now replaced by another liquid of equal volume at  $40^{\circ}C$ . The time taken (in min) for the temperature to become  $35^{\circ}C$  under similar condition is [density of liquid = 0.8 ' (density of water), specific heat of liquid 0.5 (specific heat of water)]



**24.** A calorimeter contains 50g of water at  $50^{\circ}C$ . The temperature falls to  $45^{\circ}C$  in 10 min. When the calorimeter contain 100g of water at  $\cdot 50^{\circ}C$ , it takes 15 min' for the temperature to become  $45^{\circ}C$ . Find the water equivalent (in g) of the calorimeter. (Assume Newton's law of cooling)

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**25.** A heating curve has been plotted for a solid object as. shown in the figure. If the mass

of the object is 200g. 'then latent heat of vaporization for the material of the object is  $\frac{n}{2} \times 10^6 \frac{J}{k}g$ .If power supplied to the object is constant and equal to 1kW, then find the value of n.

'(##CEN\_KSR\_PHY\_JEE\_C13\_E01\_025\_Q05##)'



**26.** In an experiment of measuring specific heat of a liquid, a stream of liquid flows at a steady rate of  $5\frac{g}{s}$  over an electrical beater

dissipating 135W and a temperature rise of 5K is observed. On increasing the rate of flow to  $10\frac{g}{s}$ , the same temperature rise is produced with (a) dissipation of 235W. Find the specific heat  $Jg^{-1}K^{-1}$ ) of the liquid. (Assume heat loss to the surrounding in both the cases is same.)

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**27.** The specific heat of a metal at low temperatures varies according to

 $S=\left(rac{4}{5}
ight)T^3$ , where T is the absolute temperature, Find the heat energy (in SI unit) needed to rate the temperature of unit mass of the metal from T=1K to T=2K



28. Two identical square rods of metal are welded end to end as shown in figure (i), 20 calories of heat flows through it in 4 minutes. If the rods are welded as shown in figure (ii), the same amount of heat will flow through the



**29.** Heat is supplied to a certain amount of ice at a constant rate for 9 min. For' the first 1 min, the temperature rises uniformly with time, Then the temperature remains constant for the next 5 min and after that again temperature rises at uniform rate for last  $3 \min$ . If the ratio of magnitude of final temperature to initial temperature is  $P \circ C$ , then find the 'value of 4P. (Given: `S\_ (ice )=0.5 (cal) (g)^(-1)/overset(o) \ C^(-1), L\_(f)=80 (cal g).

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**30.** Two rods shown in the figure have identical geometrical dimensions. They are in contact with two heat bath at temperature  $100^{\circ}C$  and  $0^{\circ}C$ . The temperature of the junction is  $70^{\circ}C$ . If the rods are interchanged, then temperature of the junction is found to be

 $10a^{\,\circ}\,C$ . Find the value of a.

'(##CEN\_KSR\_PHY\_JEE\_C13\_E01\_030\_Q07##)'



**31.** Two different rods A and B are kept as shown in the figure. The variation of temperature of different cross-sections is plotted Find the ratio" of thermal conductivities of B to A.

'(##CEN\_KSR\_PHY\_JEE\_C13\_E01\_031\_Q08##)'



**32.** A clock pendulum made of invar has a period of 2s at  $20^{\circ}C$ . If the clock is used in a climate where average temperature is  $40^{\circ}C$ , what correction (in seconds) may be necessary at the end of 10 days to the time given by clock?

$$(lpha_{\, \in \, var}) = 7 imes rac{10^{-7}}{o} C^{\, -1}, 1 day = 8.64 imes 10^4 s$$

. Give answer in nearest integer

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33. A steel tube, whose coefficient of linear expansion is  $18 imes 10^{-6}$  per  $^oC$  contains mercury, whose coefficient of volumetric expansion is  $180 \times 10^{-6}$  per  $^{o}C$ . The volume of mercury contained in the tube is  $10^{-5}m^3$  at  $0^{\circ}C$ , and it is desired that the length of the mercury column should remain constant at all normal temperatures. This is achieved by inserting into the mercury column a rod of silica, whose thermal expansion is negligible. The volume of silica is given as  $\eta imes 10^{-5} m^3$ . Find the value of  $\eta$ .

'(##CEN\_KSR\_PHY\_JEE\_C13\_E01\_033\_Q09##)'

**34.** An experiment is carried out under pressure  $P_0 = 100 cm$  of Hg and consists of a U tube of uniform cross-section 'in 'vertical position as shown in the figure. Now end A of the tube is closed and gas in the tube is heated so that gas expands and mercury spills out. During the process, it is seen that the pressure of enclosed gas is directly proportional to the volume of gas. Find  $l_0$  (in

m ).

'(##CEN\_KSR\_PHY\_JEE\_C13\_E01\_034\_Q10##)'



**35.** A rod has variable coefficient of linear expansion  $\alpha = \frac{x}{5000}$ . If the length of the rod is 1m, then determine the increase in length of the rod (in cm ) on increasing the temperature of the rod by  $100^{\circ}C$ .

'(##CEN\_KSR\_PHY\_JEE\_C13\_E01\_035\_Q11##)'



**36.** A 30.0cm long metal rod expands by 0.0650cm when its temperature is raised from  $0^{\circ}C$  to  $100^{\circ}C$ . A second rod of different metal and of the same length expands by 0.0350cm for the same rise in temperature, A third composite rod, also 30.0cm long, is made up of pieces of each of the above metals placed end to end and expands by 0.0580cmwhen temperature is increased from  $0^{\circ}C$  to  $100^{\,\circ}C$ . Find the length (in cm ) of the smaller portion of the composite rod at  $0^{\circ}C$ .



**37.** Two conductors A and B each of crosssection area  $5cm^2$  are connected in series. Variation of temperature (in  ${}^{o}C$  ) along the length (in cm ) is as shown in the figure. If thermal conductivity of A is  $200Jm^{-1}\frac{s^{-1}}{2}C^{-1}$  'and thermal conductivity of B is k (in  $Jm^{-1}\frac{s^{-1}}{c}C^{-1}$  ), then find  $\sqrt{\frac{k}{8\sqrt{3}}}$ .

'(##CEN\_KSR\_PHY\_JEE\_C13\_E01\_037\_Q12##)'



**38.** A rod of negligible heat capacity has length 40cm, area of cross-section  $1.0cm^2$  and thermal conductivity  $100Wm^{-1}{}_{\odot}C^{-1}$ . The temperature of one end is maintained at  $0^{o}C$ and that of the other end is slowly and linearly varied 'from  $0^{\circ}C$  to  $60^{\circ}C$  in 10 min. Assuming no loss of heat through the sides', find the total heat transmitted (in J) through the rod in these 10 min.



**39.** Three rods AB, BC and BD of same length l and cross-sectional area A are arranged, The end D is immersed in ice whose mass is 440g. Heat is being supplied at constant rate of  $200 \frac{cal}{c}$  from the end A. The time (in s) in which whole ice will melt is (Given:k (thermal conductivity) = 100 cal  $m^{-1} \frac{s^{-1}}{2} C^{-1}, A = 10$  cm<sup>2</sup>, l=1 m<sup>,</sup> latent heat of fusion of ice is 80 cal g.)

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**40.** As shown in the figure, two large black plane surfaces are maintained at constant temperature  $T_1$  and  $T_2(T_1 > T_2)$ . Two thin black plates are placed between the two surfaces and in parallel to them. After some time, steady conditions are obtained. The ratio of heat transfer rate between plate-1 and plate-3 to the ratio of original (when plate-3) and plate- 4 was not present) heat transfer rate between plate-1 and plate-2 in steady state is  $\eta$ . Find  $\frac{1}{\eta}$ .

'(##CEN\_KSR\_PHY\_JEE\_C13\_E01\_040\_Q14##)'

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