

CHEMISTRY

BOOKS - GRB CHEMISTRY (HINGLISH)

ALCOHOLS AND ETHERS

EXERCISE 1

1. An alkene obtained by the dehydration of an alcohol (A), on ozolysis gives two molecules of acetaldegyde for ever molecule of alkene. The alcohol (A) is

A.
$$CH_3CH_2CH_2OH$$

B.
$$CH_3CH_2OH$$

$$\mathsf{C.}\,CH_3CH=CHCH_2OH$$

D.
$$CH_3CH_2CHCH_3$$

Answer: D



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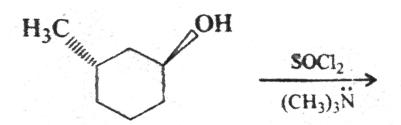
EXERCISE 2

- **1.** $R o COOH o R o CH_2OH.$ This mode of reductoin of an acid to alcohol can be effected by:
 - A. Zn/HCl
 - B. Na-alcohoi
 - C. Aluminium isopropoxide and isopropyl alcohol
 - D. $LiAlH_4$

Answer: D



1. Find the major product of the following reaction.



A.

В.

C.

D.



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EXERCISE 4

1. What is product of the followin reaction:

$$\begin{array}{c}
OH \\
-OH
\end{array}$$

Answer: C



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EXERCISE 5

1. Perdict product of the following reaction,

A.
$$CH_3 - CH_2 - OH$$

B. No reaction

C.
$$H_3C-\stackrel{|}{C}-OH$$

$$\mathsf{D}.\,H_2C=CH_2$$

Answer: B



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EXERCISE 6

- **1.** 1 Propanol and 2 propanal can be best distinguished by
 - A. oxidation with alkaline $KMnO_4$ followed by reaction with Fehling's
 - solution
 - B. oxidation with alkaline dichromate followed by reaction with Fehling's solution
 - C. oxidation by heating with copper followed by reaction with
 - Fehling's solution
 - D. oxidation with alkaline H_2SO_4 followed by reaction with Fehling's solution

Answer: C



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EXERCISE 7

- **1.** On heating glycerol with $KHSO_4/\Delta$, a compound is obtained, which has a bad odour. The compound is:
 - A. Acrolein
 - B. Formic acid
 - C. Allyl alcohol
 - D. Methyl isocyanide

Answer: A



EXERCISE 8

1. A compound X with moleuclar formula C_3H_8O can be oxidized to a compound Y with the molecular formula $C_3H_6O_2$. X is most likely to be a:

A. primary alcohol

B. secondary alcohol

C. aldehyde

D. ketone

Answer: A



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EXERCISE 9

1. Identify (Z) in the following series.

 $Ethanol \stackrel{PBr_3}{\longrightarrow} (X) \stackrel{Alc.\,/\,KOH}{\longrightarrow} (Y) \stackrel{(\,i\,)\,H_2SO_4\,/\,\,(\, ext{Room temp}\,)}{(\,ii\,)\,\,(H_2O\,,Heat\,)} \;(Z)$

A.
$$C_2=CH_2$$

B. CH_3CH_2OH

 $\mathsf{C.}\,CH_3-CH_2-O-CH_2-CH_3$

D. $CH_3-CH_2-SO_3H$

Answer: B



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EXERCISE 10

- 1. Which of the following is not characteristic of alcohols?
- A. Their boiling points rise fiarly uniformly with a rise in molecular

weight

B. Lower member have a pleasant smell but buring taste nad the

higher ones are odourless and tasteless

C. These are lighter than water

D. Lower member are insoluble in water and organicn solvents bu the solublity goes on increasing with the rise of molecular weight

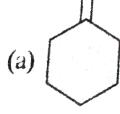
Answer: D



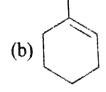
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EXERCISE 11

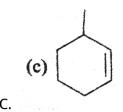
1. A $\xrightarrow{Dil.H_2SO_4/Hg^+}$ 1- Methylcyclohexanol. Here A is:



A.



В.



 $\mathsf{D}.\left(d\right)\left(a\right)$ or $\left(b\right)$

Answer: D



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EXERCISE 12

1. Find the product of reaction

В.

(c) HO

(d) HO

Answer: B



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EXERCISE 13

1. 2-Phenylbutan-2-ol can eb prepared by which of the following combinations?

A.
$$C_6H_5COCH_3+C_2H_5MgBr$$

B.
$$C_2H_5COCH_3+C_6H_5MgBr$$

C.
$$C_6H_5COC_2H_5+CH_3MgBr$$

D. All of these

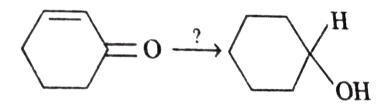
Answer: D



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EXERCISE 14

1. Predict tha nature of reducing agent in the following reaction.



- A. $LiAlH_4$
- B. $NaBH_4$
- $\mathsf{C.}\,H_2/Pt$
- D. Both a and c

Answer: C



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EXERCISE 15

1. The correct structure for compound B will be:

$$(i) \text{ PBr}_3 \longrightarrow [A] \xrightarrow{\text{CH}_3\text{CH}_2\text{CHO}} [B]$$

D.

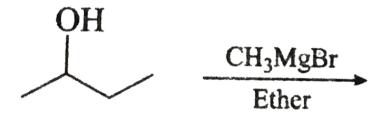
Answer: B



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EXERCISE 16

1. Find the product of following reaction,



- B. CH_2
- C. 📝

Answer: C



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EXERCISE 17

1. $CH_3 - CH_2 - Br \xrightarrow{DryAg_2O}$ Product of reaction is:

A.
$$CH_3-CH_2-OH$$

$$\mathsf{B.}\,H_2C=CH_2$$

$$\overset{O}{\mathsf{C}}.\,CH_3-\overset{|}{C}=H$$

$$\mathsf{D.}\,CH_3-CH_2-O-CH_2-CH_3$$

Answer: D



1. The product when glyerol reacts PCl_5 is:

A. 1,2,3-trichoropropane

B. glycero monochlorophydrin

C. glycero dichlorohydrin

D. All of these

Answer: A



EXERCISE 19

1. Glycerol $\stackrel{KHSO_4}{\longrightarrow} A \stackrel{LiAlH_4}{\longrightarrow} B$.

A. Acrolein, Allyl Alcohol

B. glyceryl, sulphate, acrylic acid

C. allyl alcohol, acrolein

D. only acrolein (B is not formed)

Answer: A



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EXERCISE 20

1. Complete the following reaction

$$H_3C$$
— $Cl+2$ $\longrightarrow A, A \text{ is }:$

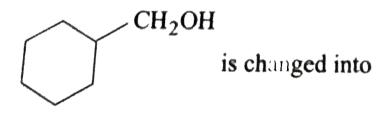
Answer: C



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EXERCISE 21

1. Choose the correct option for the given structure



A. $(i)Cu,300\,^{\circ}\,C-(ii)CH_3CH_2MgBr,H_3O^+$

 $\mathsf{B.}\,(i)CrO_3-(ii)CH_3CH_2MgBr,H_3O^+$

 $\mathsf{C.}\,(i)KMnO_4-(ii)CH_3CH_2MgBr,H_3O^+$

D.
$$(i)Na_{2}Cr_{2}O_{7}+H_{2}SO_{4}-(ii)CH_{3}CH_{2}MgBr,H_{3}O^{+}$$

Answer: A



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EXERCISE 22

1. $H_3C-CH_2CH_2CO_2CH_3
ightarrow H_3C-CH-CH_2CH_2CH_2OH$ can be effected using:

OH

A. $LiAlH_4$ and $thenH^+$

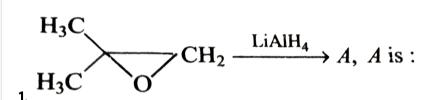
B. $NaBH_4$ and $thenH^+$

C. H_2/Pt carbon

D. All of these

Answer: A

EXERCISE 23



A.
$$CH_3CHCH_2OH$$
 CH_3

B. $CH_3CH_3CH_2CH_2OH$

C. No reaction

D.

Answer: C



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 $\xrightarrow{\text{SeO}_2} A, A \text{ is}:$

A. (a)

В.

Answer: B



O———

OH

, Predict the correct option :

(a) (

(b) \bigg__I

В.

(c) —OH

D. No reaction

Answer: D



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1. An aromatic amine (X) was treated with alcoholic potash and another compound (Y) when foul smelling gas was formed C_6H_5NC . The compound (Y) was formed by reacting a compound (Z) with Cl_2 in the presence of slaked lime. The compound (Z) is:

- A. $C_6H_5NH_2$
- B. C_2H_2OH
- $\mathsf{C}.\,C_2H_2OH$
- D. $CHCl_3$

Answer: B



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EXERCISE 27

1. An alcohol (a) on dehydration gives (B), which on ozonolysis gives acetone and formaldehyde.(B) decolourises alkaline $KMnO_4$ solution but

(A) doies not .(A) and (B) are respectively:

A.
$$CH_3CH_2CH_2CH_2OH$$
 and $CH_2CH_2CH = CH_2$

B.
$$CH_3CH_2-CH-CH_2$$
 and $CH_2-CH=CH=CH_2$

$$\mathsf{C.}\,(CH_3)_3C-OH$$
 and $(CH_3)_2C=CH_2$

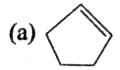
D.
$$(CH_3)_3 CHCH_2 - OH$$
 and $(CH_3)_2 C = CH_2$

Answer: C



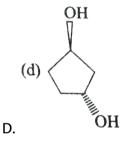
EXERCISE 28

$$\frac{\text{Conc. H}_2\text{SO}_4}{\Delta} X \xrightarrow{\text{OsO}_4} Y; \text{Product '}Y' \text{ is :}$$



A.

В.



Answer: C



1. If phenol1 magnesium bromide and acetaldehyde are the reactants, the product formed after hydrolysis would be:

A. benzyl alcohol

B. 1-Phenylethanol

C. 2-Phenylethonal

D. Acetone

Answer: B



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EXERCISE 30

$$C_7H_{14}O(X) \xrightarrow{H^+} C_7H_{12}(Y) \xrightarrow{1.B_2H_6} C_7H_{14}O(Z) \cdot (A \text{ 3° alcohol}) \qquad O$$

$$Y \xrightarrow{O_3} O$$

$$Y \xrightarrow{O_3} O$$

A. $CH_3CH_2CH_2MgBr$ and hydrolysis

B. $CH_3CH(Br)CH_3$. $AlCl_3$

 $\mathsf{C.}\,(CH_3)_2CHMgBr\,$ and acid hydrolysis

D. $CH_3CHCHCH_3$, Zn

Answer: C



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EXERCISE 31

1. Complete the following reaction

$$CH_3$$

$$CH_3$$

$$CH_3$$

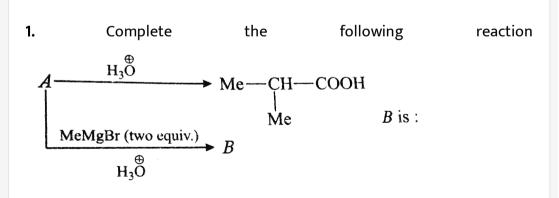
$$CH_3$$

Answer: B



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EXERCISE 32



A.
$$Me_2CHCOMe$$

B. $Me_2CH - CMe_2$ OH

C. $Me_2CHCOCHMe_2$

D. Me_2 C $HCHCHMe_2$ OH

Answer: B



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EXERCISE 33

1. Which of the following reactions is possiblw?

A.
$$C_6H_5OH+HBr
ightarrow C_6H_5Br+H_2O$$

B. $(CH_3)_3CCl + NaOCH_3 \rightarrow (CH_3)_3COCH_3 + NaCl$

 $C. (c) \xrightarrow{+ CH_3ONa} \xrightarrow{CH_3OH} \xrightarrow{OMe} CI$

Answer: D



EXERCISE 34

1.

$$H_3C-egin{pmatrix} CH_3 & CH_3 & CH_3 \ C & CH_3 & CH_3 \ C & CH_3 & CH_3 \ CH_3 & CH_3 & CH_3 \ \end{pmatrix} CH_3 - CH_0 = -CH_3. \ X can be$$

A.
$$BH_3/THF, H_2O_2/OH^-$$

B. H_3O^+

C. $Hg(OAc)_2/NaBH_4, NaOH$

D. All of these

Answer: C



1. Complete the following reaction

$$H_2C = CH - C \xrightarrow{Cl} \xrightarrow{Cl_2 + H_2O} A, A \text{ is}$$

A.
$$CCl_3CH - CH_2Cl$$
 OH

B.
$$CCl_3CH - CH_2OH$$

C.
$$CCl_3CH - CH_2Cl$$

D.
$$CCl_3CH - CH_2$$
 $|$
 OH
 OH

Answer: B



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EXERCISE 36

1.
$$CH_3CH=CH_2 \xrightarrow[H_2O_2/OH]{NaBD_4} ext{Product X, X is:}$$

A.
$$CH_3$$
 C HCH_2D

B. CH_3CHCH_2OH

C. $CH_3 C H C H_3$

D. None is correct

Answer: B



EXERCISE 37

1. Identify end product A,B and C of the following:

A. CH_3CHCH_3 in all cases OH

B. $CH_3CH(OH)CH_2D$, $CH_3CH(OD)CH_3CH(OD)CH_2D$

C.
$$CH_3CHCH_3$$
 in all cases $\stackrel{|}{OD}$

D. CH_3 C HCH_3D in all cases OH

Answer: B



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EXERCISE 38

1. Complete the following reaction

$$RCH = CH_2 \xrightarrow{BH_3'THF} A \xrightarrow{H_2O_2/OH^-} A$$

$$C \leftarrow D$$

A. $\left(RCH_{2}CH_{2}\right)_{3},RCH_{2}CH_{2}OH,RCH_{2}CH_{3},HI$

B. $\left(RCH_2CH_2\right)_3, RCHCH_3, RCH_2CH_3, HI$

C.
$$\left(RCH_{2}CH_{3}
ight)_{3}, \ \underset{OH}{R}\ CH-CH_{2}CH_{3}, RCH_{2}CH_{3}, HI$$

D. None is correct

Answer: A



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EXERCISE 39

1.
$$H_3C-\stackrel{CH_3}{\stackrel{|}{C}}-CH=CH_2 o H_3C-\stackrel{CH_3}{\stackrel{|}{C}}-CH-CH_3.$$
 This change $\stackrel{CH_3}{\stackrel{|}{C}}-CH_3$

 CH_3

can be done by.

 CH_3

- A. acid catalysed hydration
- B. oxymercuation-demercuation
- C. hyrdroboration-oxidation
- D. any method mentioned above

Answer: B



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EXERCISE 40

In the above reaction sequence, the final product is:

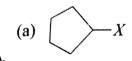
- A. diethyl ether
- B. 1-methoxypropane
- C. isopropyl alcohol
- D. propylene glycol

Answer: B



EXERCISE 41

1. Sodium teritary butoxide forms ether only with:



B.
$$CH_3-X$$

C.
$$H_3C-CH-CH_3$$

D.
$$H_3C-\mathop{CH}\limits_{CH_3}^{X}-CH_3$$

Answer: B



1. In the give reaction

$$C = CH - CH_3 \xrightarrow{\text{(i) Hg(OAc)}_2/CH_3OH} [X],$$

A.
$$H_3C-egin{array}{c}OMe\\ C\\CH_3\end{array}-CH_2-CH_3$$
 OMe B. $H_3C-egin{array}{c}C\\CH_3\end{array}-CH_2-CH_3$ CH_3

B.
$$H_3C-\stackrel{C}{\stackrel{}{\stackrel{}{CH}}}-CH_2-CH_3$$

C.
$$H_3C-C \atop \mid CH - CH_2 - CH_3$$

D.
$$H_3C-OH-CH_2-CH_3$$
 CH_3

Answer: A



1.

A. $LiAlH_4$ and $NaBH_4$

B. $LiAl\frac{H_4}{A}lCl_3$ and $LiAlH_4$

C. $LiAlH_4$ and $LiAl\frac{H_4}{A}lCl_3$

 $\mathsf{D}.\,H_2\,/Ni\,\,\,\mathrm{and}\,\,H_2\,/Pt$

Answer: C

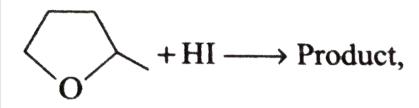


EXERCISE 44

1.

Complete the following

reaction



Answer: B

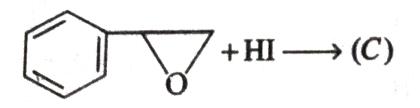


Complete

the

following

reaction



A.

$$\text{(b)} \begin{array}{|c|c|c|} \hline \text{CH}_2 - \text{CH}_2 \\ \hline \text{OH} \\ \hline \end{array}$$

В.

D. None is correct

Answer: A



1. Which of the following reactions is possible?

A.
$$C_6H_5OH + HBr
ightarrow C_6H_5Br + H_2O$$

$$\operatorname{B.}(CH_3)_3CCl + NaOCH_3 \rightarrow (CH_3)_3COCH_3 + NaCl$$

C. (c) $Cl + CH_3ONa \xrightarrow{CH_3OH} OMe$

 $D. \xrightarrow{\text{(d)} \begin{subarray}{c} (d) \begin{subarray}{c} C_6H_5MgBr \xrightarrow{H_5O^*} C_6H_5CH_2C(CH_3)_2 \\ OH \end{subarray}}$

Answer: D



EXERCISE 47

1. Complete the following reaction

$$H_2C$$
 CH_2 (i) CH_3MgCl X (ii) H_2O

A. CH_3CH_2OH

B. $(CH_3)_2CHOH$

 $\mathsf{C}.\,CH_3CH_2CH_2OH$

 $\mathsf{D}.\,HO-CH_2-CH_2-CH_2-CH_2-OH$

Answer: C



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EXERCISE 48

1. What is Z in the following sequence of reactions?

$$Z \overset{PCl_3}{\longrightarrow} X \overset{alc\,.\,KOH}{\longrightarrow} Y \overset{(\,i\,)\,Conc\,.\,H_2SO_4}{(\,ii\,)\,H_2Oboil} Z$$

$$A.\,H_2C=CH-CH_2-OH$$

B. $CH_3CHOHCH_3$

C. $(CH_3CH_2)COOH$

D.
$$CH_3CH = CH_2$$

Answer: B



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EXERCISE 49

1. Complete the following reaction

$$H_3C$$
— CH — CH_2 $\xrightarrow{(i)}$ CH_3 = C $\xrightarrow{(ii)}$ Product?

A.
$$H_3C-CH-CH_2-CH=CH_2-CH_3$$
 OMe

B.
$$H_3C-CH-CH_2-C\equiv C-CH_3$$

C.
$$H_3C-CH-CH_2-C\equiv C-CH_3$$
 OH

Answer: B



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EXERCISE 50

- **1.** Which of the following reagents cannot be used for the oxidation of 1° alcohol aldehyde?
 - A. PCC
 - B. Collin's reagent
 - $\mathsf{C.}\,MnO_2$
 - D. MnO_2

Answer: D



$$A \leftarrow \underbrace{\frac{\text{PhCH}_2 - 1}{\text{CF}_3\text{CH}_2\text{OH}}}_{\text{DMSO}} \xrightarrow{\frac{\text{PhCH}_2 - 1}{\text{DMSO}}} E$$

A and B respectively:

(a)
$$A = \bigcup_{CH_2 = Ph} \bigcup_{CH_2 Ph} OH$$

(b)
$$A = \bigcup_{CH_2Ph} OCH_2Ph$$

(c)
$$A = B =$$

(d)
$$A = B = CH_2Ph$$

Answer: B

D.



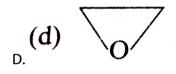
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1. When-2-chloroethenol is wanted with dilute NaOH, the major product formed is:

A.
$$Cl-CH_2-CH_2-O-CH_2-CH_2-Cl$$

$$\mathsf{B}.\,HO-CH_2-CH_2-CH_2-CH_2-OH$$

$$\mathsf{C.}\ HO-CH_2-CH_2-OH$$



Answer: D



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level 4

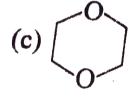


on heating

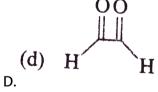
with conc. H_2SO_4 gives mainly:



В.



C.



Answer: C



level 5

1. Which of the following reaction would give the best yield of t- butyl methly ether ?

A.
$$(CH_3)_3C-OH \stackrel{H_2SO_4}{\underset{140^{\circ}C}{\longrightarrow}}$$

B.
$$(CH_3)_3C-Br+CH_3OH
ightarrow$$

C.
$$(CH_3)_3C-Br+CH_3\overset{\Theta}{O}Na
ightarrow$$

D.
$$(CH_3)_3C-\overset{\Theta}{CK}+CH_3Br
ightarrow$$

Answer: D



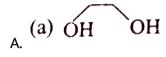
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level 6

1. Consider the following reactions:

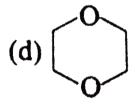
$$CH_3-CH_2 \stackrel{NaOH}{\underset{Br}{ert}} A \stackrel{NaH}{\underset{Br}{
ightarrow}} B, B+CH_2-CH_2
ightarrow C$$

The major product formed is:



B. (b) Br O ONa

С. (с) НО О ОН



Answer: D

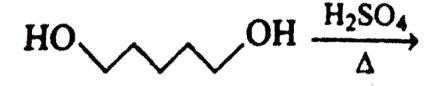
D.



In

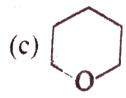
the

reaction



The major product formed is:

_R (b) /



C.

(d) HO OHO OH

Answer: C



1. The major product formed in the reaction is"

$$OH - CH_3CH_2 - I \xrightarrow{\text{Na}_2CO_3} \xrightarrow{\text{acetone, } \Delta}$$

D.

Answer: B



1. Consider the following reaction

$$H_2C = CH - CH_2CH_2 - OH \xrightarrow{Br_2/CCl_4} A \xrightarrow{Dil.KOH} B$$

The product B is:

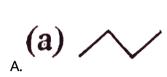
$$(d) \bigcirc Br$$

Answer: D

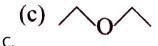
D.



1. Find out correct product of reaction:



B. $CH_2 + CH_2$



D. CH_(3)CH_(2)OH`

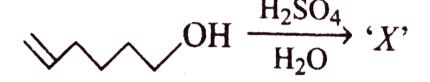
Answer: C

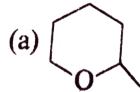


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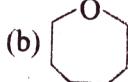
level 11

1. The major product X of the reaction





A.



В.



Answer: A



1. In the reaction

$$H_3C$$
 $C \longrightarrow CH_2 \xrightarrow{CH_3OH} (X)$

The product X has the structure"

A.
$$H_3C- CH_3$$
 CH_3
 CH_3CH_3

B. $H_3C- CCH_3$
 CH_3
 CH_3

C. $H_3C- CCH_3$
 CH_3

C. $H_3C- CCH_3$
 CCH_3
 CCH_3

Answer: A



level 13

1. Consider the following sequence of reactions

$$H_2C = CH - COOCH_3 + Br_2 \xrightarrow{CCl_4} A$$
OH
$$+ A \xrightarrow{K_2CO_3} B$$
OH

Answer: B



level 14

1. In the reaction:

$$Me_3C-O-CH_2CH_3+HI \xrightarrow{\Delta}_{1 ext{ mole}}$$

A.
$$Me_3C-OH+CH_3CH_2I$$

$$\mathsf{B.}\, Me_3C - I + CH_3CH_2OH$$

C.
$$Me_3C-I+CH_3CH_2I$$

$$\mathsf{D.}\,Me_3C-OH+CH_3CH_2OH$$

Answer: B



Watch Video Solution

level 15

1. Which of the following ethers ethers is the most unreactive to cleavage

with conc. HBr?

A.
$$Ph-CH_2-O-CH_3$$

$$B. Ph - O - Ph$$

$$(d) \rightarrow O \leftarrow$$

Answer: B



Watch Video Solution

level 16

$$O \longrightarrow H^{\oplus}$$
 Major product

Major Product:

1.

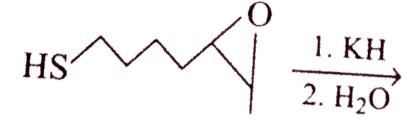
A.

В.

Answer: C



1. The product of the reaction is:



Answer: A

level 18

1. Which of the following reactions will not result in the formatio of anisole?

(a)
$$\sim$$
 OH + (CH₃)₂SO₄ $\xrightarrow{\text{NaOH}}$

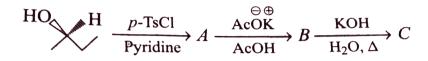
$$\mathbf{B.} \stackrel{\text{(a)}}{\longleftrightarrow} \mathrm{OH} + (\mathrm{CH_3})_2 \mathrm{SO_4} \xrightarrow{\mathrm{NaOH}}$$

D. (d)
$$\langle OH + CH_3MgI \longrightarrow$$

Answer: D



1. Consider the following sequence of reactions



D. $(\pm)-2$ butanol

Answer: A



1.

Complete

the

following

reaction

$$\begin{array}{c}
O \\
O \\
O \\
COOH
\end{array}$$

$$O$$
; A is CH_2OH

A. $B_2H_6\,/\,H_2O$

B. $LiAIH_4$

C. CH_3OH/Na

D. P/HI

Answer: A



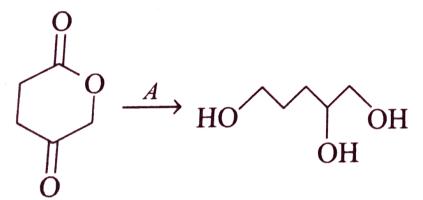
1.

Complete

the

following

reaction



A. B_2H_6

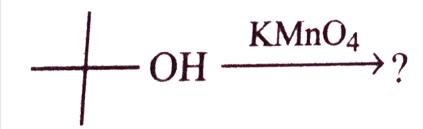
B. $LiAlH_4$

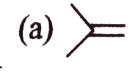
 $\mathsf{C.}\,Sn\,/\,HCl$

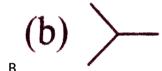
D. $NaBH_4$

Answer: B









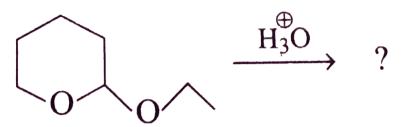
D. No reaction

Answer: D

C.



1. The major product formed in the reaction is"



A. (a)
$$CH_3CH_2OH$$

Answer: C



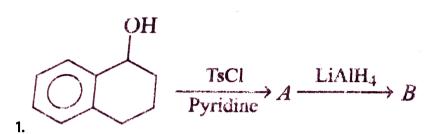
Complete the following

reaction

$$\xrightarrow{\Delta} A \xrightarrow{\text{OsO}_4} B \xrightarrow{\text{HIO}_4} C \xrightarrow{\text{LiAlH}_4}$$
 excess

Answer: A





Product B of the above reaction is:

Answer: B



1. Find out the product when compound reacts with $NaBH_4$:

В.

Answer: B



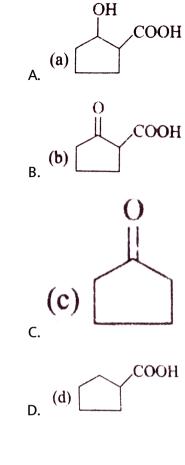
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level 27

$$\begin{array}{c}
\text{OH} \\
\text{COOH} \\
\xrightarrow{\text{COOH}}
\end{array}
\xrightarrow{\text{K}_2\text{Cr}_2\text{O}_7} (X)$$

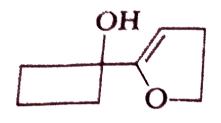
Find out X:

1.

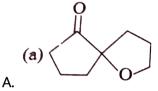


Answer: C





$$\xrightarrow{H^{\oplus}}$$
 Product



Answer: A



level 29

$$H_3CO$$
 OCH₃

$$NaBH_4 \rightarrow (X), Product (X) is:$$
1. OCI

X, Product is:

Answer: C



level 30

Et O—C—CH₃
$$\stackrel{?}{\longrightarrow}$$
 Et O—C—OH

Which of the following is best set of reagents to performs to the above conversion?

A. $LiAlH_4$

B. $NaBH_4$

 $\mathsf{C.}\, K_2 C r_2 O_7$

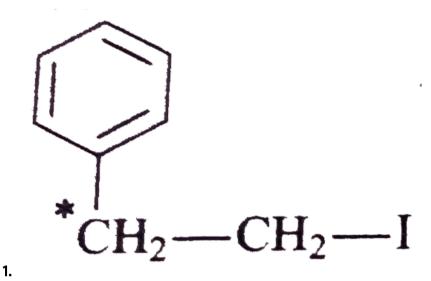
D. None of these

Answer: B



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level 31



Product of the reaction is:

A.
$$Ph-\overset{*}{C}H_2-CH_2-SH$$

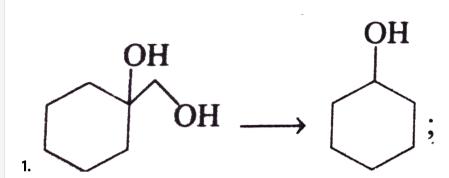
B.
$$Ph-CH_2-\overset{*}{C}H_2-SH$$

C. Both of these

D. None of these

Answer: C





This conversion can be carried out by:

A. $H_2SO_4/\Delta, HIO_4$

B. $NaIO_4, H^{\,\oplus}$ $/\Delta$

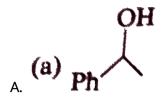
C. HIO_4 , $NaBH_4$

D. $H^{\,\oplus}$ / De < s, Zn(Hg-HCl)

Answer: C

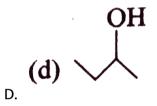


1. Which of the following alcohols will show positive iodoforms test?



В. 📄

 $C.CH_3 - CH_2 - OH$

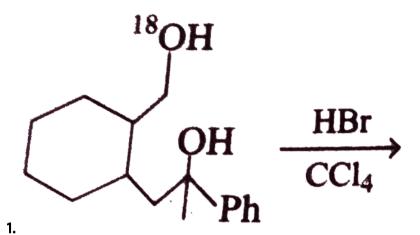


Answer: C



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level 34



Major product obtained in this reaction is:

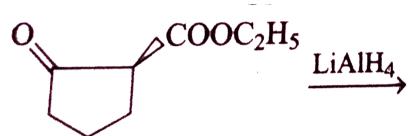
. .

Answer: B



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level 35



1. , Products of

the reaction is:

A. racemic

B. diastereomers

C. meso

D. optically pure

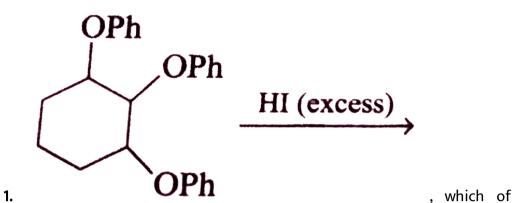
Answer: B

level 36

- **1.** Reduction of $R-CH_2OH-RCH_3$ can be carried out by:
 - A. $LiAlH_4$
 - B. H_2-Ni
 - $\mathsf{C}.\,RedP+HI$
 - D. $NaBH_4/AlCl_3$

Answer: C





the following is major product?

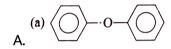
A.

В.

D. None of these

level 38

1. Which of the following ethers will get hydrolysed by $H^{\,\oplus}$ / H_2O ?



Answer: D



1. Which of the following alcohols will not react with Cu / Δ

A.
$$CH_3 - CH_2 - OH$$

B.
$$Ph- {\stackrel{Ph}{\stackrel{|}{C}}}-OH$$

$$\overset{|}{\overset{Ph}{CH_3}} CH_3 \ C.\ H_3C - \overset{|}{\overset{C}{\overset{C}{C}}} - OH$$

$$(d)$$
 Ph—CH CH_3

Answer: B



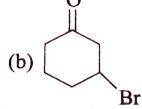
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level 40

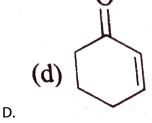
OH
$$AgNO_3 \rightarrow Y'; Y' \text{ is}$$

A.

1.



В.



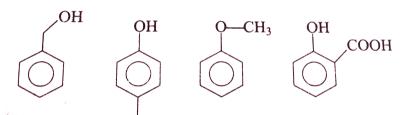
Answer: A



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level 41

1. Which of the following can give purple colour with neutral $FeCl_3$



- A. ii and iv
- B. I and iii
- C. ii and iii
- D. iii and iv

Answer: A



1. The hydrolsed

in aqueous

of

acetone gives.

$$K = CH_3O$$
 H_3C
 H
 CH_3
 H_3C
 $H_$

$$M = H_3CO$$
 H_3C
 $CH_3 CH_3$
 NO_2
 H
 $CH_3 OH$

A. K and L

B. Only K

C. L and M

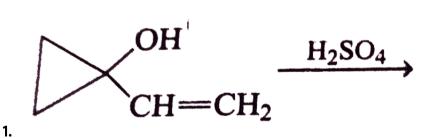
D. Only M

Answer: A

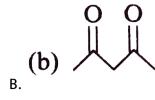


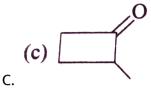
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level 43



identify 'P' in the reaction:



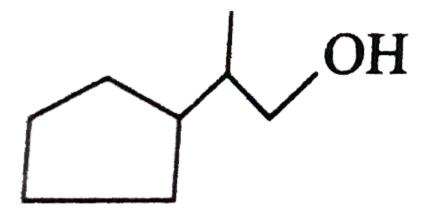


Answer: C



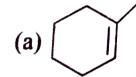
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level 44



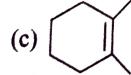
1.

identify 'P' in the reaction:



A

В.



C.

Answer: C



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level 45

$$+ CH_3 - CH = CH_2 \xrightarrow{H_2SO_4} P \xrightarrow{O_2} Q \xrightarrow{H^{\oplus}/H_2O} R + S$$

1.

identify 'P' in the reaction:

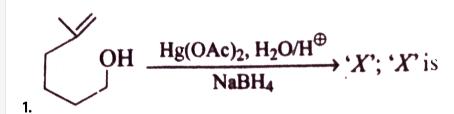
A.

Answer: C

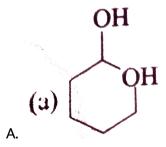


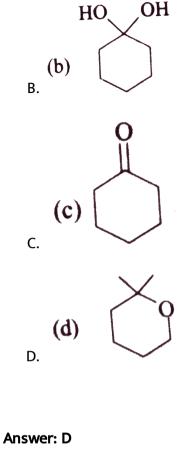
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level 46



identify 'P' in the reaction:







$$+ CH_3MgBr \xrightarrow{H^{\oplus}/H_2O} P \xrightarrow{HBr} Q \xrightarrow{Mg} R \xrightarrow{HCHO} S,$$

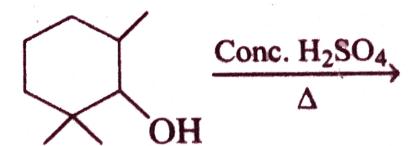
identify 'P' in the reaction:

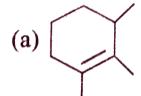
Answer: B

D.



1. Identify the major product of the following reation:

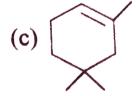




A.



В.



C.

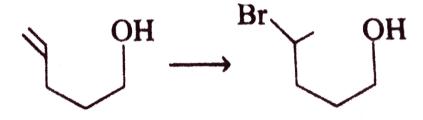
Answer: A



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level 49

1. Find the correct method for the following conversion:



A. H, HBr

B. $Conc.\ H_2SO_4, \Delta$

C. $H^{\,\oplus}$, HBr

D. None of these

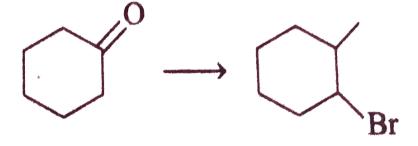
Answer: C



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level 50

1. Which combination of reagents will bring about the following conversion?



A. $MeMgBr/H^{\oplus}$, H_2SO_4/Δ , HBr/H_2O_2

B. $MeMgBr/H^{\oplus}, H_2SO_4/\Delta, HBr$

C. $MeMgBr/H^{\oplus}$, HBr/CCl_4

D.
$$HBr/H_{2}O_{2},\,MeMgBr/H^{\,\oplus}$$

Answer: A



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level 52

1. Which of the following alcohols will undergo easiest dehgydration?

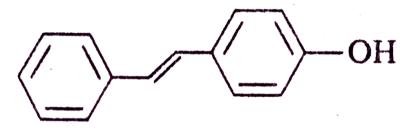
Answer: D



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level 53

1. The reactio of HBr with the followin compound would produce



$$D. \xrightarrow{(d)} Br \longrightarrow Br$$

Answer: B



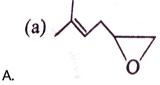
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level 54

1. In the following

В.

$$Me_2C=CH-CH_2-CH=CH_2+C_6H_5CO_3H(ext{1equiv.})
ightarrow X,Xis$$



$$D.$$
 (d) O

Answer: C



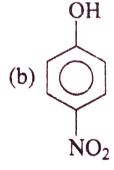
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level 55

1. The most steam volatile species is:

A.

В.



OH
$$O_{2}N \longrightarrow NO_{2}$$

$$NO_{2}$$

$$O_{3}N \longrightarrow NO_{2}$$

Answer: A



Watch Video Solution

level 56

as:

1. In the Libermann nitroso reaction, change in the colour of phenol occur

A. Brown or red green red deep blue

B. Red deep blue green

C. Red brown white

D. White red green

Answer: B



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level 57

- **1.** Which one of the following compounds will be most readily dehydrated?
 - A. 2-butanol
 - B. 1-phenyl-1-propanol

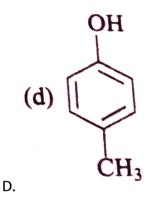
$$D.$$
 (d) O OH



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level 58

1. Compound A, C_7H_8O , is insoluble in water, dilute HCl, and aquenous $NaHCO_3$, it dissolves in dilute NaOH. When A is treated with bromine water is is converted rapidly into a compound of formula $C_7H_5Obr_3$. The structure of A is



Answer: C



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level 59

1. In the following sequence of reaction

CH=CH—CH₃ + Hg (CH₃COO)₂
$$\xrightarrow{\text{CH}_3\text{OH}}$$
 X, X is

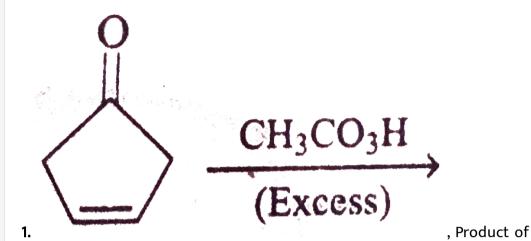
$$\begin{array}{c} \text{(d)} & \\ \hline \\ \text{D.} \end{array}$$

Answer: C

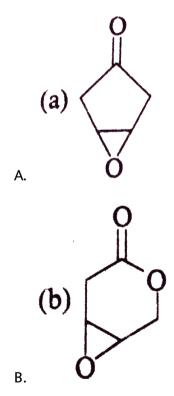


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level 60



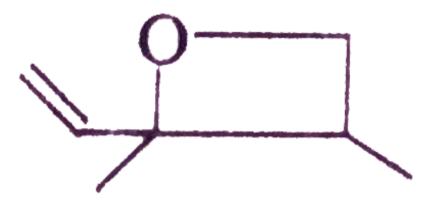
the reaction is:



Answer: B



level 61



1. When is

treated with proton acid, a resonance stablized cation is produced. Which dience listed below when treated with acid will give the same carbocation?

Answer: B

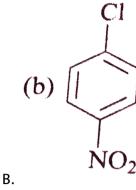


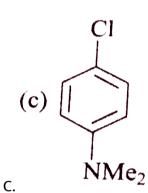
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level 62

1. Which of the following would undergo most rapid hydrolysis with aqueous to furnish the corresponding hydroxy derivatives?

$$NO_2$$





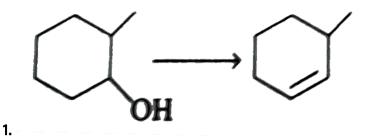
D. 🔀

Answer: A



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Others



Which of the following is best set of reagents to performs to the above

conversion?

A. ThO_2,Δ

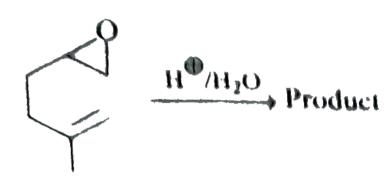
B. H_3PO_4 , Δ

C. $Conc.\ H_2SO_4,\ \Delta$

D. Al_2O_3, Δ

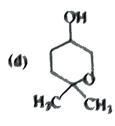
Answer: A





В.

A.



D.

Answer: B



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$$\mathbf{3.}\, 2Ph - \overset{O}{C} - CH_{3} \xrightarrow{Mg - Hg} \overset{Conc.\, H_{2}SO_{4}}{\xrightarrow{\Delta}} \xrightarrow{KMnO_{4},\, H^{\,\oplus}}$$

The final product is

A.

В.

C.

Answer: C



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4.
$$2Ph-C - CH_3 \xrightarrow[H_2O]{Mg-Hg} \xrightarrow[H_2O]{Dil.H_2SO_4} Product$$

The main product is

C.

Answer: C



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$$\begin{array}{c}
OCH_3 \\
\xrightarrow{\text{Mc}_3C-Cl} \\
\xrightarrow{\text{AlCl}_3}
\end{array}
\xrightarrow{\text{Cl}_2, \text{FeCl}_3} \xrightarrow{\text{HBr}, \Delta} A$$

5.

The final product A is:

Answer: C



The product is

A.

В.

C. (c)
$$Ph_3P = CHCH_2CH_3$$
; B_2H_6 , H_2O_2 , OH

$$\label{eq:D.Ph3P} \textbf{D.} \text{ (d) } Ph_3P = CHCH_2CH_3 \text{ ; } H_2SO_4, H_2O$$

Answer: A



7. Which of the following sets of reagents would accomplish the following

conversation

 $\textbf{A.} \hspace{0.2cm} \textbf{(a)} \hspace{0.1cm} \text{CH}_{3} \text{CH}_{2} \text{CH}_{2} \text{MgBr} \hspace{0.1cm} ; \hspace{0.1cm} \text{H}^{\oplus} \hspace{0.1cm} / \hspace{0.1cm} \text{H}_{2} \text{O}, \hspace{0.1cm} \text{PCC}, \hspace{0.1cm} \text{CH}_{2} \text{Cl}_{2}$

 $B_{\bullet}^{~~(b)~CH_3CH_2CH_2MgBr~;~H^{\oplus}/H_2O~;~H_2SO_4,~\Delta;~PCC,CH_2Cl_2}$

C.

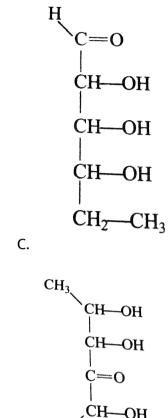
D.

Answer: C



8. An organic compound A (Molecular formula $C_6H_{12}O_4$) on treatment with Na metal liberates H_2 gas and on treatment with HIO_4 gives 2 moles of $CH_3CHO, HCOOH(1 mole)$ and $CO_2(1 mole)$. Find the structure on A.

A.



Answer: D

D.

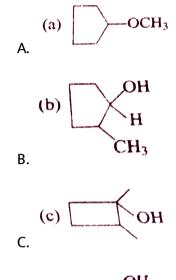


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9. An organic compound A (Molecular formula $C_6H_{12}O_4$) does not change the colour of acidic dichromate solution. Compound A on treatment with

 H_2SO_4 produces alkene, which on oxiative ozonolysis gives a molecule

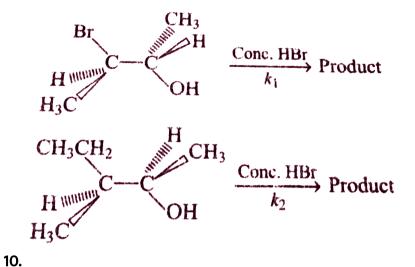
 $(C_6H_{10}O_3)$ which gives positive iodoform test. Find the structure of 'A'.



Answer: D

D.





-OH group is substituted by -Br. The slowest step is dehydration. Which of the following is correct comparison of rate constant K_1 and K_2 ?

A.
$$k_1 = k_2$$

B.
$$k_1 > k_2$$

$$C. k_1 < k_2$$

D. cannot be predict

Answer: C



$$C_{5}H_{12}O_{2} \xrightarrow{K_{2}Cr_{2}O_{7}} C_{5}H_{8}O_{3} \xrightarrow{CH_{3}OH, H^{\bigoplus}} C_{6}H_{10}O_{3} \xrightarrow{LiAiH_{4}} A + CH_{3}OH$$

$$\downarrow H^{\bigoplus}, \Delta$$

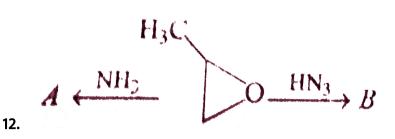
$$CH_{3}$$

$$CH_{3}$$

The molecule A in the sequence reaction is

Answer: D





The product A and B respectively:

(a)
$$A = H_2N$$

$$OH$$

$$OH$$

$$N_3$$

$$N_3$$

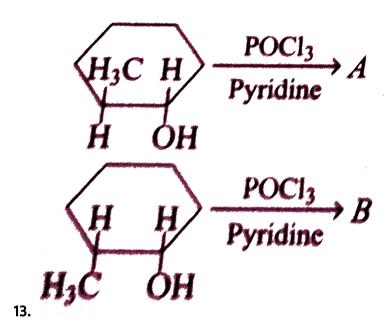
(b)
$$A = {\begin{pmatrix} H_3C & OH \\ NH_2 \end{pmatrix}}; B = {\begin{pmatrix} OH \\ OH \end{pmatrix}}$$

(c)
$$A = \begin{pmatrix} H_3C & OH & N_3 \\ H_2N & \vdots & B = \end{pmatrix}$$
 OH

(d)
$$A = \begin{pmatrix} H_2N & CH_3 \\ OH & OH \end{pmatrix}$$

Answer: C





Product A and B respectively:

(a)
$$A = \langle \rangle$$
 ; $B = \langle \rangle$

(b)
$$A = B = \langle -- \rangle$$

(c)
$$A =$$
 ; $B =$

(d)
$$A = B =$$

Answer: C

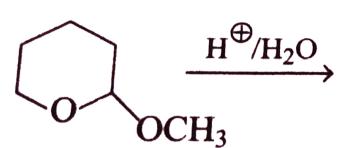
В.

14. Complete

the

following

reaction



$$\mathsf{B.}\,(b)\;CH_3CH_2-OH$$

D. HO

Answer: A



15. The releative rate or acid catalyssed dehydration of following alcohols

would be:

(P)
$$Ph-\stackrel{CH_3}{\stackrel{|}{C}}-CH-CH_3 \ _{OH}^{\stackrel{|}{OH}}$$

(Q)
$$Ph-\stackrel{|}{C}-CH_2-CH_2-OH$$

(R)
$$Ph- \displaystyle \stackrel{CH_3}{\underset{OH}{|}} -CH_2CH_3$$

$$\operatorname{A.}R>P>S>Q$$

$$\operatorname{B.}R>S>P>Q$$

$$\operatorname{C.} P > R > S > Q$$

$$\operatorname{D.}R > S > Q > P$$

Answer: A



16. Which of the following alcohols will show positive iodoforms test?

A.
$$CH_3-\overset{OH}{CH}-CH_2-NO_2$$

$$\overset{OH}{\overset{OH}{\overset{OH}{\downarrow}}}-CH_2-\overset{OH}{\overset{OH}{\overset{\downarrow}{\downarrow}}}-OH$$
 B. $CH_3-\overset{OH}{\overset{OH}{\overset{\downarrow}{\downarrow}}}-CH_2-\overset{OH}{\overset{OH}{\overset{\downarrow}{\downarrow}}}-CH_2CH_3$

D. None is correct

Answer: C



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17. In the given reaction

$$\begin{array}{c}
CH_{3} \\
\xrightarrow{B_{2}H_{6}} \xrightarrow{TsCl} \xrightarrow{Me_{3}CO^{\ominus}K^{\ominus}} \\
(A)
\end{array}$$

$$\xrightarrow{(A)} \xrightarrow{(A)} \xrightarrow{(B)} \xrightarrow{(B)}$$

The product B is:

A. Identical to B

B. Chain isomer of A

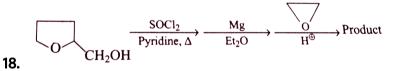
C. a positional isomer of 'A'

D. reduced product of A

Answer: C



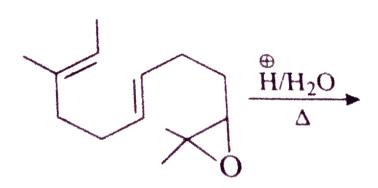
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Answer: B



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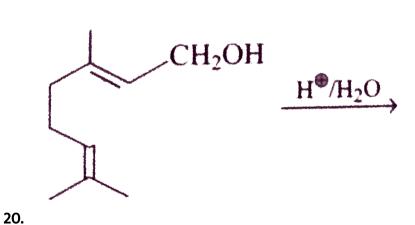
19.

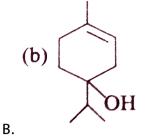
В.

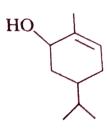
Answer: C



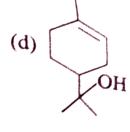
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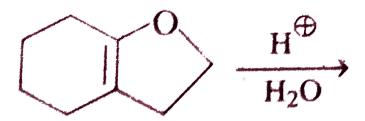
C.



D.

Answer: D





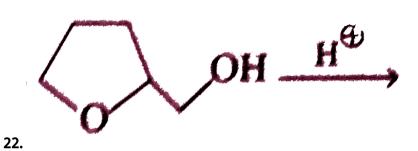
The final product is :

A.

21.

(c) OH OF

Answer: B



The final product is :

A.

В.

C.

D.



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$$-CH-CH_2 \xrightarrow{CH_3CH_2OH}$$

The final product is:

(a)
$$CH_2CH_3$$
 OCH_2CH_3

C.

D.

A.

В.



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$$OH \xrightarrow{Br_2} OH$$

24.

The final product is:

A.

В.

C.

$$(d)$$
 O Br

Answer: A

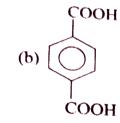
D.



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OH Conc.
$$H_2SO_4$$
 $A \xrightarrow{(CH_3CO)_2O} B \xrightarrow{KMnO_4/OH} C$
OH

25.



В.



C.

Answer: B

D.



$$+ CH_3CH_2OH \xrightarrow{HCl}$$

The final product is:

26.

CH2—O—CH3

$$(c)$$
 OCH₂CH₃

Answer: C

C.



$$CH_2OH$$

$$(a) \qquad NO_2$$
A.

(c)

C.

D.

Answer: D



28. The order of reactivity of the following alcohols towards HCl is:

- A. 1 > 2 > 3 > 4
- ${\rm B.}\,1>3>2>4$
- $\mathsf{C.}\,4>3>2>1$
- $\mathsf{D.}\,4>3>1>2$

Answer: C



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29. The order of solubility of

 $\mathrm{A.}\,I > II > III$

 $\mathrm{B.}\,I < II < III$

 $\mathsf{C}.\,II > III > I$

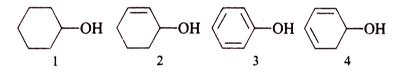
D. II > I > III

Answer: C



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30. Degydration of the following alcohols will be in order:



$${\rm A.}\,1 < 2 < 3 < 4$$

$${\rm B.}\,4 < 3 < 1 < 2$$

$$\mathsf{C.}\,4 > 2 > 1 > 3$$

$$\mathsf{D}.\,1>3>4>2$$

Answer: C

31. Which of the following reactions proceeds with retention of configuration?

(a)
$$H \xrightarrow{\text{Me}} OH \xrightarrow{\text{Na}} \xrightarrow{\text{CH}_3\text{Br}} A$$
.

(b) H OH
$$\xrightarrow{\text{TsCl}}$$
 $\xrightarrow{\text{CH}_5\text{ONa}}$

(c) H
$$\xrightarrow{\text{OH}}$$
 OH $\xrightarrow{\text{PCl}_5}$ $\xrightarrow{\text{CH}_3\text{ONa}}$

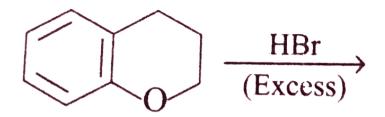
$$(d) H \xrightarrow{Ph} OH \xrightarrow{SOCl_2} \xrightarrow{KCN}$$

$$CH_3$$

Answer: A



32. Find out correct product of reaction:



C.

D.

Answer: B



$$Cl$$
 $\xrightarrow{\text{Mg/ether}}$ $\xrightarrow{\text{HCHO}}$ $A, A \text{ is :}$

Answer: B

33.



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$$\xrightarrow{\text{cold KMnO}_4} A \xrightarrow{\text{CrO}_3} B, A \text{ and } B \text{ are } :$$

34.

$$\xrightarrow{coldKMnO_4} A \xrightarrow{CrO_3} B, A \text{ and } Bare$$

A.

В.

$$(d)$$
 OH OH OH

Answer: A

D.



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$$\begin{array}{c}
OH \\
\hline
OH \\
\hline
MnO_2
\end{array}$$
Product

(MnO_(2))` Product

35.

A. (a)
$$OH$$
O

B. (b) OH
O

C. OH
O

PCls
CH3ONa

CH3ONa

D.

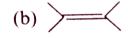
(b)

Answer: C

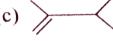


36. Complete the following reaction
$$\frac{OH}{\Delta} \xrightarrow{Conc. H_2SO_4} Major product:$$

A.



(c



D. None of these

Answer: B



37.

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$$\begin{array}{c}
O \quad OH \\
& \downarrow \\
&$$

Identify product D in this reaction

C. CHI_3

$$\begin{array}{c} O & O \\ O & N_{a}^{\oplus} \end{array}$$

Answer: A

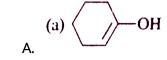


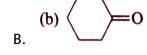
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38. In the given reaction

$$\bigcirc -\text{O}-\text{CH}_2\text{CH}_3 \xrightarrow{\text{H}^{\oplus}/\text{H}_2\text{O}} (P^{\oplus})$$

P will be:

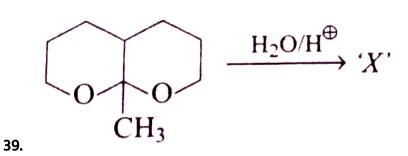




D.
$$(d) \leftarrow CH_2 - O$$

Answer: B





'X' will be

$$\begin{array}{c|c} \text{(c) } H & \xrightarrow{\text{PCI}_5} & \xrightarrow{\text{CH}_3\text{ONa}} \\ \hline \textbf{C.} & \text{Et} \end{array}$$

D. All of these

Answer: A



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$$OH \xrightarrow{MnO_2} \xrightarrow{CH_3CH_2OK} Major product :$$

40.

Answer: B



$$\begin{array}{c}
OH \\
\hline
O & 1. Pb (OAc)_4 \\
\hline
O & 2. OH/H_2O
\end{array}$$
(X)

'X' will be

A.

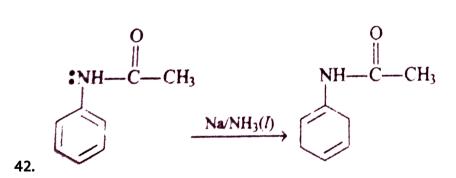
41.

- B. (b) O—C—(CH₂)₃—C—(CH₂)₄—COOF
- C. (c) HO—(CH₂)₃—C—(CH₂)₄—C—OH
- D. (d) HO—(CH₂)₃—C—(CH₂)₄—CH₂OH

Answer: C



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The above reaction is known as:

- A. Clemmensen reduction
- B. Birch reduction
- C. MPV reaction
- D. Wolff-Kishmer reaction

Answer: B



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43. In the following reaction



The main product is:

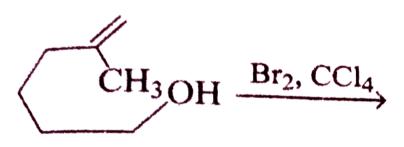
В.

D.

Answer: B



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44.

В.

The major product obtained in this reaction is:

$$CH_3$$
 CH_2Br
(a) O

$$CH_3$$
 Br O

Answer: A



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Ph
$$\longrightarrow$$
 Conc.H₂SO₄ OH OH

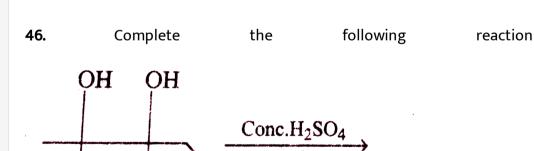
reaction

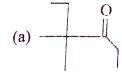
(a)
$$Ph$$
— C — CH_3

(d)
$$CH_3$$
— C — Ph
OH

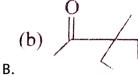
Answer: C







A.



٥.

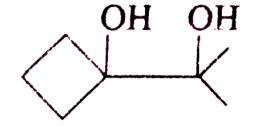


C.

D. None of these

Answer: A

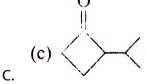








(b)



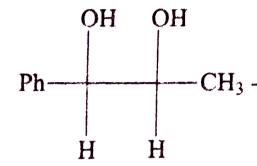
Answer: B

D.

Complete the

following

reaction



A.

C.

D.



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49. Complete the following reaction

$$\begin{array}{c}
CH_3\\
OH\\
OH\\
CH_3
\end{array}$$

В.

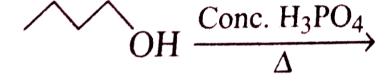
C.

Answer: B



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50. Complete the following reaction



$$_{\Delta}$$
 (a) $/\!\!\!/$

В.

C.

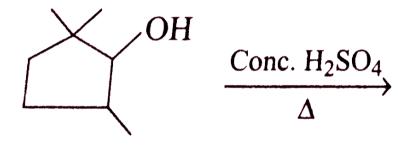
D. None of thses

Answer: C

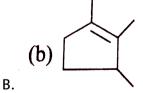


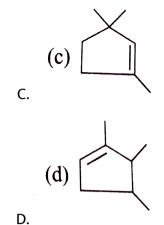
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51. Complete the following reaction



(a)





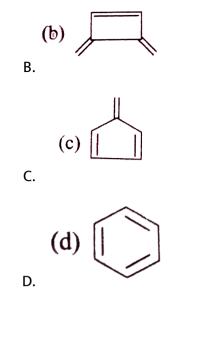
Answer: A



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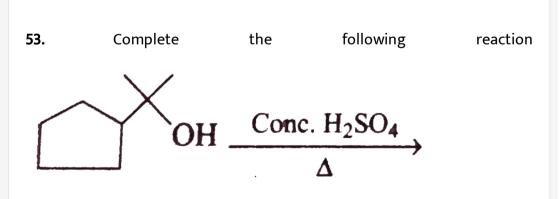
52. Complete the following reaction

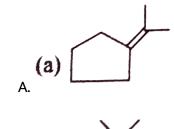
$$CH_2$$
— OH
 H^{\oplus}, Δ
 CH_2 — OH

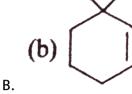


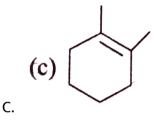
Answer: D





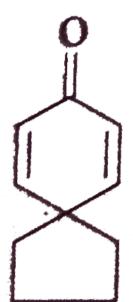




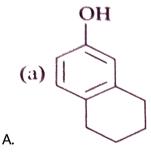


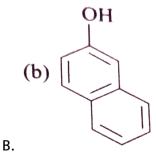
Answer: C

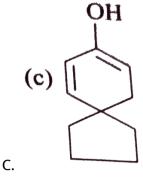


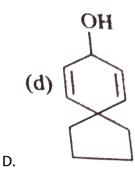


$\xrightarrow{\Lambda}$









Answer: A



55. When ethylene glycol is heated with oxalic acid in this presence of conc. H_2SO_4 , the product formed is:

$$(c) \bigcirc O$$

Answer: A

В.



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56. An organic compoun having molecular formula C_3H_6O does not react with 2,4-dintrophenol hydrazine and does not react Na metal. The compound is expected to be:

A.
$$CH_3CH_2CHO$$

$$\operatorname{B.}CH_{3}COCH_{3}$$

$$\mathsf{C.}\,CH_2=CH-CH_2-OH$$

 $\operatorname{D.}CH_2=CH-O-CH_3$

Answer: D



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- **57.** Which of the following statements is correct?
 - A. Phenol is less acidic than ethanol
 - B. Phenol is more acidic than ethanol
 - C. Phenol is more acidic than p-nitrophenol
 - D. Phenol is more acidic than acetic acid

Answer: B



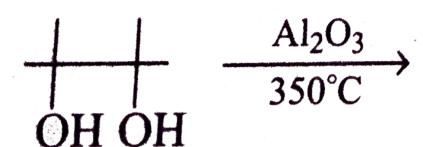
58. The vapour of an alcohol X are passed over Cy heated at $300^{\circ}\,C$ whereby an alkene is formed as prduct . The alcohol X is expected to be:

$$_{\rm B.}$$
 (b) $\bigwedge_{\rm OH}$

$$(d)$$
 \rightarrow OH

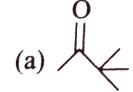
Answer: D





59.

The major product formed in the reaction is:



A.



c (c)

 $(d) \rightarrow COH$

Answer: C





60.

The major product formed in the reaction is:

A.
$$(CH_3)_3C-Br+(CH_3)_3COK
ightarrow$$

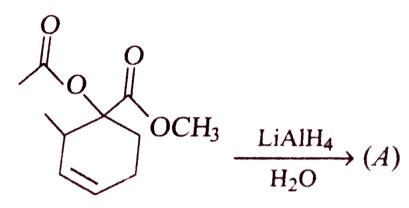
$$\mathsf{B.}\,(CH_3)_3C-OH \xrightarrow[140°C]{H_2SO_4}$$

C.
$$(CH_3)_3C-OH \stackrel{Al_2O_3}{\underset{240^{\circ}C}{\longrightarrow}}$$

D. overset

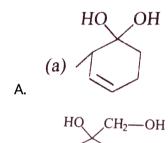
Answer: D





61.

Find out 'A' of the reaction:



В.

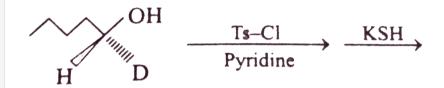
(b)

Answer: C



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62. Identify the major product of the following reation:



D. No reaction

Answer: B

63. Find the product of the following reaction with sterechemistry.

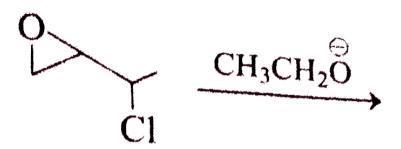
(a)
$$H_3C \longrightarrow C - C$$

C.
$$H_{3C} \longrightarrow C \longrightarrow C$$

Answer: C



64. Select the major product of the following reaction:



(c)
$$\overset{\ominus}{O}$$
 $\overset{OCH_2CH_3}{\smile}$ C.

D.
$$CH_3CH_2-OCH_2$$
 CH_3

Answer: D



65. What would be the major product of the following reaction?

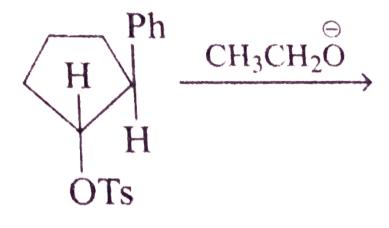
$$H_3C$$
 H_3C
 CH_3OH
 CH_3OH

D. None of these

Answer: A



66. Find out the major product of the following reaction:



$$(a)$$
 Ph

(c)
$$\sim$$
 Ph \sim OCH₂CH₃

D. No reaction

Answer: B



$$\begin{array}{ccc} CH_3 & & & & & \\ \hline CH_3CH_2 & & & & & \\ H & & & & & \\ \end{array} \xrightarrow{PBr_3} & \xrightarrow{\bigoplus \bigoplus \\ KCN & \\ \end{array} \xrightarrow{\text{final product.}}$$

67.

(b)
$$CH_3$$
 $C=C$

В.

C.
$$CH_3$$
 CH_3CH_2
 CH_3

Answer: C



68. What would be the major product of the following reaction?

$$CH_3$$
 CH_3
 CH_3
 $Conc. H_2SO_4$
 COH
 COH

Answer: B



The unexpected product B is:

69.

Answer: C

D.



70. The final product in the following reaction is :

$$\frac{1. \text{ Mg, ether}}{2. \text{ He} 0^{\oplus} \Lambda}$$

A.

В.

Answer: A

D.



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71. How many structure of final products are possible?

$$\stackrel{Conc.H_2SO_4}{\longrightarrow} \stackrel{Br_2\,/\,CCl_2}{\longrightarrow} C_4H_8Br_2$$

A. 2

B. 5

C. 6

D. 3

Answer: B



72. Chosse the correct statements regarding the following reaction:



- A. Syn addition of -H (from BH_3) and -OH (from solution) occur.
- B. Syn addition of -H (from BH_3) and -OH (from H_2O_2) occur.
- C. The product is optically active.
- D. Addition follows anti Markownikoff orientation

Answer: A::C



$$OH \xrightarrow{HBr} Product$$

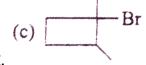
Which of the following are possible are possible products in significant amounts?



Α.



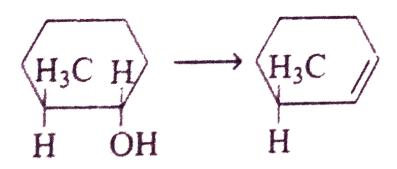
В.



_ (d

Answer: B::C::D





74.

Which of the following represent conditions to perform given conversation?

A. $POCl_3$, pyridine

B. Na - metal, CS_2heat

$$H_3C$$
— SO_2C1

D. CF_3SO_2Cl , pryidine, $Me_3CO^{\Theta}K^{\oplus}$

Answer: B::C::D



75. Which of the following alcohols do not give white turbidity on treatment $HCl\slash ZnCl_2$?

A.
$$CH_3CH_2OH$$

C.
$$N=C-CH-OH$$
 CH_3 CH_3

D.
$$CH_3 - \overset{|}{\overset{\circ}{CH_2}} - OH$$

Answer: A::C

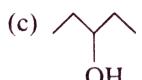


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76. Which of the followin wil give iodoform?

A.
$$CH_3CH_2OH$$

$$\stackrel{OH}{\stackrel{|}{\stackrel{}{=}}}$$
 B. $CH_3-CH-Ph$



C.

Answer: A::B::D



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77. which of the following ethers will get hydrolysed by HI?

Answer: C::D



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78. Which of the following reactions are correctly matched?

A.
$$CH_3 - \overset{CH_3}{\overset{}{\overset{}{\underset{CH_3}{\bigcup}}}} - O - CH_3 \overset{HI}{\overset{}{\overset{}{\longrightarrow}}} S_N 1$$

B.
$$CH_3 - CH - O - CH_3 \stackrel{H^{\,\oplus}\,/\,H_2O}{\longrightarrow} S_N 1$$

$$\mathsf{C.}\,CH_3 - \underset{CH_3}{CH} - O - CH_3 \stackrel{HI}{\longrightarrow} S_N 2$$

D.
$$CH_3 - O - CH_2 \stackrel{HI}{\longrightarrow} S_N 1$$

Answer: A::B::C



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79. Which of the following compounds will give positive Victor Meyer test?

A.
$$CH_3CH_2OH$$

B.
$$CH_3-\stackrel{CH_3}{\overset{|}{\underset{CH_3}{CH_3}}}-OH$$

C.
$$CH_3CH_2-I$$

D.
$$CH_3 - \overset{CH_3}{\overset{|}{C}} - NO_2$$

Answer: A::C::D



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80. Which of the followin alcohols undergo rearrangement during degydration reaction?

Answer: A::B::D



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81.

 $C_2H_50C_2H_5$

and



can be

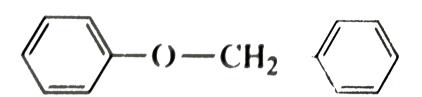
distnguished by:

- A. aq. $FeCl_3$
- B. Na metal
- C. Tollen's reagent
- D. $K_2Cr_2O_7$

Answer: B::D



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82. The ether

when treated with HI produces:

$$CH_2I$$

$$CH_2OH$$

Answer: B::D

83. Which of the following reactions will give ether as main product?

A. OH
$$\frac{Na}{C_2H_5B_1}$$

B.
$$Me_3C-OH \xrightarrow[C_6H_5Br]{Na}$$

C.
$$Me_3C-OH \xrightarrow[CH_3CH_2CH_2Br]{Na}$$

D.
$$CH_3CH_2CH_2OH \xrightarrow{Na}_{Me_3C-Br}$$

Answer: A::D



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84. C_2H_5Br can be converted into $C_2H_5-O-C_2H_5$ by:

A. reacting by C_2H_5ONa

B. heating with moist $Ag_2{\cal O}$

C. heating with dry Ag_2O

D. treating with C_2H_5MgBr

Answer: A::C



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85. $1^{\circ}, 2^{\circ} \ \ {
m and} \ \ 3^{\circ}$ alcohols can be distinguished by:

A. Cu/573K

B. Victor Meyer test

C. $ZnCl_2$ / HCl

D. $Br_2 + H_2O$

Answer: A::B::C



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86. Alcohols can be replaced by-Cl group by the followin reagents:

Answer: B::C::D **Watch Video Solution** 87. Glycerol can be converted to acrolein by dehydration in presence of A. $Conc.\ H_2SO_4$ B. $KHSO_4$ C. $CaCl_2$ D. $Anhyd. ZnCl_2$ Answer: A::B

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A. Cl_2

B. $SOCl_2$

 $\mathsf{C}.\,PCl_5$

D. $HCl + ZnCl_2$

88. CH_3CH_2-OH can be converted to CH_3CH_2CN by the following reaction:

A.
$$CH_3CH_2OH + KCN \stackrel{\Delta}{\longrightarrow}$$

B.
$$CH_3CH_2OH + HCN \stackrel{\Delta}{\longrightarrow}$$

$$\mathsf{C.}\,\mathit{CH}_{3}\mathit{CH}_{2}\mathit{OH} + \mathit{HCN} \xrightarrow{\mathit{TsCl}} \xrightarrow{\mathit{KCl}}$$

$$\mathsf{D.}\, CH_3CH_2OH + HCN \xrightarrow{SOCl_2} \ \stackrel{KCN}{\longrightarrow}$$

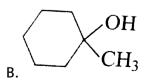
Answer: C::D



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89. Which of the following will oxidise to salt of acid by $Br_2 + KOH$?

A.
$$CH_3 - CH_2 - OH$$

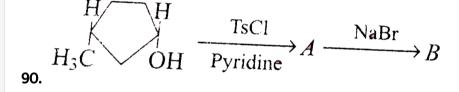


$$\begin{array}{c} \text{(d)} \\ \hline \\ \text{C.} \end{array}$$

D. 📝

Answer: A::B





$$A. \qquad (a) A is H_{3}C \qquad OT$$

B.
$$B ext{ is } H ext{Br}$$

C.
$$H_2C$$
 OT

$$B \text{ is } H_{3}C$$
 $H_{3}C$
 $H_{3}C$

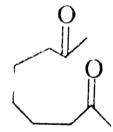
Answer: A::B



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(a)
$$A$$
 is CH_3

A.



В.

$$C$$
 is \bigcirc O

Answer: A::B::C



92.

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$$OH \xrightarrow{\mathbf{PBr_3}} A \xrightarrow{\mathrm{Mg}} B \xrightarrow{CH_3 - CH - C - H} C$$

A. (a)
$$A$$
 is \bigcirc Br

B. (b)
$$B$$
 is \bigcirc MgBr

$$C$$
 is CH

$$C$$
 is CH_2 — C — CH_2 — C — CH_3
OH

D.

Answer: A::B::C

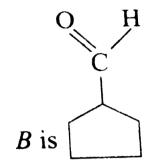


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$$OH \xrightarrow{Conc. H_2SO_4} A + B$$
93.

$$A$$
 is OH

В.



C.

Answer: B::C::D



OH
$$CrO_{3} \longrightarrow A \xrightarrow{CH_{3}MgBr} B \xrightarrow{H_{2}SO_{4}} C \xrightarrow{B_{2}H_{6}} D$$

$$A \xrightarrow{H_{2}O_{2} \cap OH} D$$

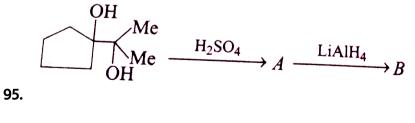
(a)
$$A$$
 is

(b)
$$B$$
 is

В.

Answer: C::D





ΗÓ

Answer: A::C

C.



96.
$$\begin{array}{c}
Conc. HI \\
CO_{2}, H^{\oplus}/H_{2}O
\end{array}
\xrightarrow{KMnO_{4}} D$$

B.
$$COC$$

$$COC$$

$$COC$$

$$COC$$

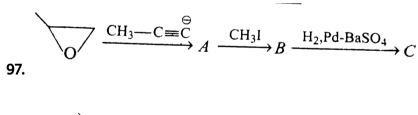
$$COC$$

$$COC$$

$$COC$$

Answer: A::B::C::D





A.
$$A ext{ is } O ext{CH}_2 ext{C} = C ext{C} ext{CH}_3$$

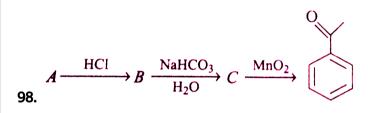
B. $A ext{ is } O ext{CH}_2 ext{C} ext{E} ext{C} ext{C} ext{CH}_3$

C is
$$H_3CO$$
 CH_3 $C=C < CH_3$ C

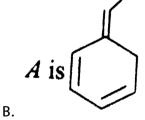
Answer: A::B::C

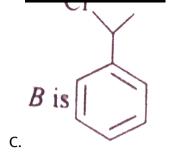


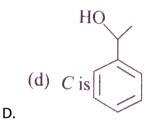
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A.

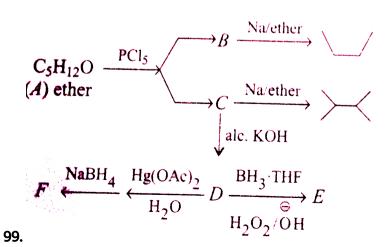






Answer: A::C::D





A.
$$Dsi$$
CH (3)-CH=CH (2)`

B. E is
$$CH_2-CH_2-CH_2-OH$$

C. FisCH_(3)underset(OH)underset(|)(CH)-CH_(3)`

D. B is CH_3CH_2CI

Answer: A,B,C,D



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H₂C=CH—CH₂Br
$$\xrightarrow{1. \text{ Mg}}$$
 $\xrightarrow{2. \text{ HCHO}}$ $\xrightarrow{3. \text{ H}_3^{\oplus} \text{O}}$ $A \xrightarrow{\text{Br}_2}$ $B \xrightarrow{\text{KOH}}$ $C \xrightarrow{\text{alc. KOH}}$ $C \xrightarrow{\text{O}}$

B.
$$BisCH_2 - CH - CH_2 - CH_2 - OH$$

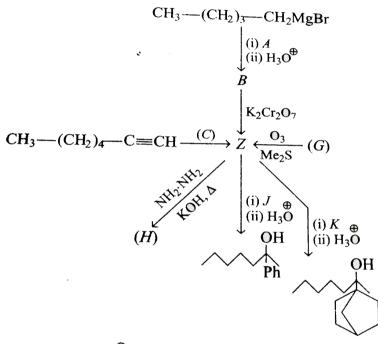
(c) C is

A. $AisH_2C = CH - CH_2CH_2OH$

$$C$$
 is C



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101.

$$\Delta$$
 (a) Z is

B. J is PhMgBr

C.
$$CisHg^{2+} \, / H_2 SO_4$$

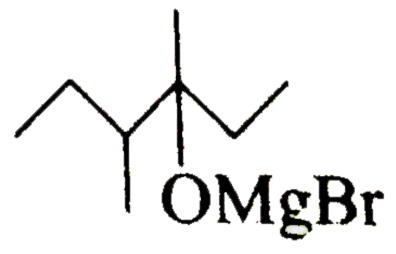
 $\begin{array}{c} MgBr \\ \hline D. \end{array}$

Answer: A,B,C,D



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102. Compound A is an optically active alcohol. Treatment with oxidising agent converts it to a ketone B. In a separate reaction A is treatment with PBr_3 , converting it into C. C on reaction with Mg is added to B to yield



Identify the

correct option.

- A. A is 2-butanol.
- B. A is 1-butanol
- C. C is 2-bromobutane
- D. C is 1-bromobutane

Answer: A,C



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103. Alcohol A $(C_{10}H_{18}O)$ is converted into mixture of alkene B and C on heating with conc. H_3PO_4 . Catalytic hydrogenation of B and C yields the same product. Assuming that dehydration of alcohol A proceed without rearrangement. Alkene B on ozonolysis form cyclopentanone. Identify the correct options.

Answer: A,B,D



D.

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104. A compound 'X' $(C_{14}H_{14}O)$ on mild oxidation yields $C_{14}H_{12}(Y)$. If X is treated with a dehydratingg agent, it loses a molecule of H_2O and resulting product on vigoporus oxidation yields two molecule of benzoic acid. Identify the structure of X and Y.

B. X is
$$Ph-\overset{|}{CH}-CH_2-Ph$$

OH

$$\stackrel{O}{\stackrel{||}{\stackrel{}{\circ}}} ext{C. y is } Ph-\stackrel{C}{C}-CH_2 \ _Ph$$

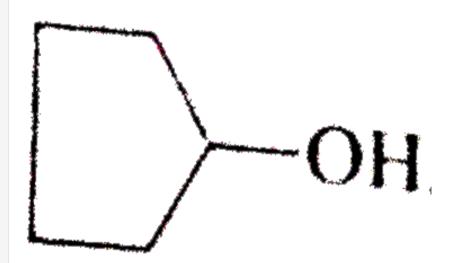


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105. Compounds A, B and C are isomeric alcohols with formula $C_5H_{12}O$. A on oxidation given ketone, B gives acid while C is not oxidised, A gives test with $I_2/NaOH$. The three isomerice alcohols react with HBr with decreasing rates C>A>B. Identify A and B.

$$B$$
 is OH

106. An optically active alcohol A $(c_8H_{16}O)$ on oxidation gives B. A on heating gives $C(C_8H_{14})$ as major product. C on ozonlysis produces $D(C_5H_8O)$ and CH_3-C-CH_3 . D on reduction with $LiAlH_4$ gave C_5H_8O



Identify

correct answers.

$$D$$
 is O

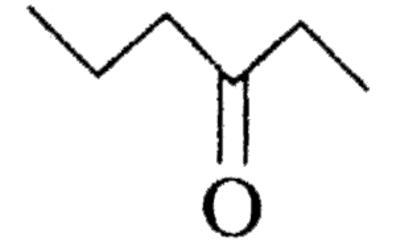
B is
$$\longrightarrow$$
 COOH

$$A$$
 is CH_3 CH_2OH

Answer: A,B,C,D



107. Compound A (C_7H_{14}) decolouries Br_2 in CCl_4 and reacts with $Hg(Oac)_2$ followed by ruduction with $NaBH_4$ to produce a resolvable compound B.A undergoes redutive ozonolysis to give



as one of the

compound

$$B$$
 is OH

Answer: A,C



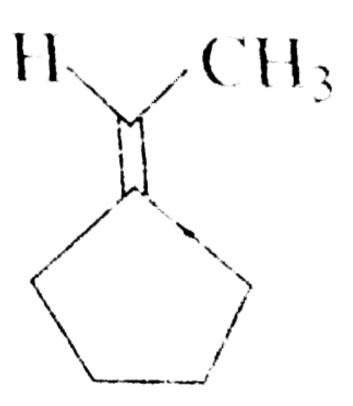
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108. A 3° optically active alcohol $C_9H_{18}O$ 'A' on dehydration with conc. H_2SO_4 produces $B(C_9H_{16})$ which exists in two stereoisomeric forms. For ozonolysis of B followed by work up with $Zn-H_2O$ produces CH_3-C-H and $C(C_7H_{12}O)$. C on treatment with $LiAlH_4$ produces

on

hehydration

produced



Identify the correct the correct answers.

 $D(C_7H_{14}O)$.

C.
$$D$$
 is C — C H.

D. (d)
$$D$$
 is OH

Answer: B::D

В.



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109. A $(C_5H_{12}O)$ produces, on reaction PCl_5 from alkyl chloride B and C.

B and C both on reaction with aqueous KOH form alcohol D and E . Both D and E give iodoform test. Identify the correct answers.

$$A$$
 is CH_3CH_2 —O— CH_3
A.

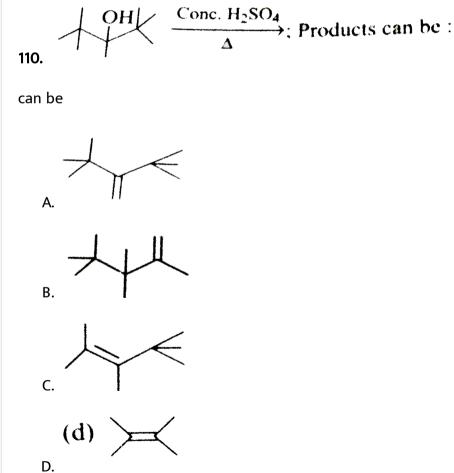
C. C is
$$CH_3CH_2CH_2CI$$

Answer: A::B::D



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, Product





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111. Among of the following gemdiols which are stable with respect to corresponding carbonyls:

A.
$$CI_3CC - H$$
OH
OH
 $F_3C - CF_3$
B.

Answer: A::B::C::D



C.

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112. Which of the following reactions are correct

C.
$$OC_2H_5 \xrightarrow{NaBH_4} OH OC_2H_5$$

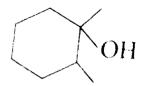
D.
$$CH_3-C\equiv N \xrightarrow{LiAIH_4} CH_3CH_2NH_2$$

Answer: A::C::D

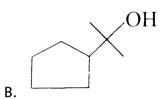


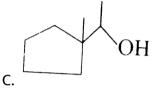
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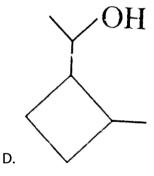
113. Which of the following alcohols will give same alkene on reaction with conc.



A.







Answer: A::B::C



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114. Which of the following reactions would produce same product?

$$(a) \qquad \underbrace{CCl_4 - NaOH}_{H^{\oplus}H_2O}$$

Α

$$\begin{array}{c}
OH \\
\hline
CO_2 + NaOH \\
\hline
H'H_2O
\end{array}$$

В.

C.
$$\frac{\text{CHCl}_4 + \text{NaOH}}{\text{H}^{\oplus}\text{H}_2\text{O}}$$

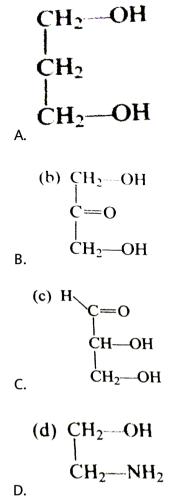
$$\begin{array}{c}
OH \\
HCN + ZnCl_2 \\
\hline
H/H_2O
\end{array}$$

Answer: A::B



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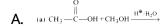
115. Which of the following compound are oxidised by HIO_4 ?

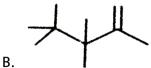


Answer: B::C::D

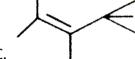


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ı



(d) >

Answer: B::C::D

D.



117. Which of the following reaction involve rearrangement?

Answer: B::C::D



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118. Which of the following paris can be distinguished by using Lucas reagent?

Answer: A::B::C



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119. Which of the following compounds are soluble in $NaHCO_3$?

В.

A.

D.

C.

Answer: A::B::D

120. Which of the following reactions are correctly interpreted?

A. (a)
$$\rightarrow$$
 OH $\xrightarrow{\text{IsCI}}$ $\xrightarrow{\text{KSH}}$ \rightarrow SH

$$\mathsf{B.} \xrightarrow{\mathsf{Hg}(\mathsf{OAc})_2} \mathsf{OH}$$

$$\begin{matrix} CH_3CH_3 & & O & CH_4 \\ (c) & Ph - C - C - Ph & AgNO_3 & CH_3 - C - Ph \\ & & & & & & & & \\ C. & & & & & & & \\ C. & & & & & & & \\ \end{matrix}$$

$$\mathsf{D.}\overset{(d)}{\longleftarrow}\mathsf{CH}_2\mathsf{-OH}\overset{\mathsf{H}_2,\mathsf{Pt}}{\longrightarrow}\mathsf{CH}_3$$

Answer: A::C::D



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121. Which of the following reagents can be used for identification of phenol?

A. Neutral $FeCl_3$

B.
$$NaNO_2 + HCl$$

$$\mathsf{C.}\,(NH_4)_2[Ce(NO_3)_6]$$

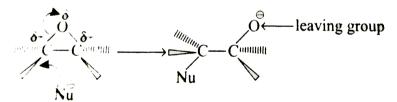
D. $ZnCl_2 \, / \, HCl$

Answer: A::B::C



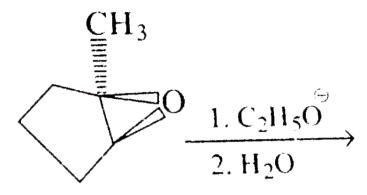
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122. Althought epoxides do not contain a good leaving group, they contains a strained three membered ring with polar bonds. Nucleophilic attack opens the strained three membered ring making it favorable process even with the poor leaving group.



This reaction occurs readily with strong nucleophilic, and with acids like HZ, where Z is nucleophilic atom.

Find out the correct product of the reaction



$$CH_3$$
 OH

A. CH_3 OH

$$CH_3$$
 OC_2H_5
OH



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123. Althought epoxides do not contain a good leaving group, they contains a strained three membered ring with polar bonds. Nucleophilic attack opens the strained three membered ring making it favorable process even with the poor leaving group.

This reaction occurs readily with strong nucleophilic, and with acids like HZ, where Z is nucleophilic atom.

$$\longrightarrow$$
 HCl ?

What would be the major product of reaction?

$$D.$$
 OH

Answer: C



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124. Althought epoxides do not contain a good leaving group, they contains a strained three membered ring with polar bonds. Nucleophilic attack opens the strained three membered ring making it favorable process even with the poor leaving group.

This reaction occurs readily with strong nucleophilic, and with acids like HZ, where Z is nucleophilic atom.

Find out major product of reaction:

В

(c)
$$\begin{array}{c} Ph \\ H_3C \end{array}$$
 $\begin{array}{c} OH \\ -CH - CH_2 \\ OCH_3 \end{array}$

C

D. None of these

Answer: A



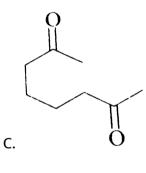
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125. 1,2-diols are oxidised to ketones or aldehydes by periodic acid HIO_4 . Periodic acid reacts with dipol to form a cyclic intermeditate. The reaction takes places because iodine is in a highly positive oxidation state, so it readily accepts electrons. When the intermediate breaks down, the bond between the two carbon bonded to the OH group break.

$$\begin{array}{c}
 & \xrightarrow{\text{Br}_2} A \xrightarrow{\text{alc. KOH}} B \xrightarrow{\text{OsO}_4} C \xrightarrow{\text{HIO}_4} D
\end{array}$$

Identify D.





D.

126. 1,2-diols are oxidised to ketones or aldehydes by periodic acid HIO_4 . Periodic acid reacts with dipol to form a cyclic intermeditate. The reaction takes places because iodine is in a highly positive oxidation state, so it readily accepts electrons. When the intermediate breaks down, the bond between the two carbon bonded to the OH group break.

HOH₂C
$$C=O$$

$$CHOH \xrightarrow{HIO_4}$$

$$CHOH$$

$$CH_2OH$$

Which of the following will not form by above reaction?

A.
$$H-\overset{|\ |}{C}-H$$

B. CH_3OH

 $\mathsf{C}.\,CO_2$

$$\begin{array}{c} O \\ || \\ \text{D.} \ H - C - OH \end{array}$$

Answer: B



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127. 1,2-diols are oxidised to ketones or aldehydes by periodic acid HIO_4 . Periodic acid reacts with dipol to form a cyclic intermeditate. The reaction takes places because iodine is in a highly positive oxidation state, so it readily accepts electrons. When the intermediate breaks down, the bond between the two carbon bonded to the OH group break.

Which of the following compounds will not react with HIO_4 ?

D.

Answer: C



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128. Carbon oxygen double bond are easily reduced by $NaBH_4$ or $LiAlH_4$. The actual reducing agent in these reduction is

hrdride ion (H^-)

$$H \xrightarrow{\Theta} - C \longrightarrow -C \longrightarrow -C \longrightarrow -C \longrightarrow + \Theta H$$

The metal hydrogen bond in $LiAlH_4$ is more than polar than metal hydrogen bond in $NaBH_4$. As a result $LiAlH_4$ is strong reducing agent than $NaBH_4$. Esters, carboxylic acids, amides cannot be reduced by $NaBH_4$

The carbonyl group of amide of reduced to methylene group by $LiAlH_4$ Find the correct product of the following reaction:

$$\mathbf{A} \quad \mathbf{H}_{3}\mathbf{C} - \mathbf{C} \longrightarrow \mathbf{C}\mathbf{H}_{2} - \mathbf{N}\mathbf{H}_{2}$$

D. No reaction



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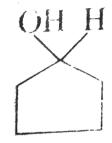
129. Carbon oxygen double bond are easily reduced by $NaBH_4$ or $LiAlH_4$. The actual reducing agent in these reduction is hrdride ion $\left(H^-\right)$

$$H^{\Theta} \xrightarrow{C} - \longrightarrow -C \xrightarrow{H_2O} -C \xrightarrow{OH}$$

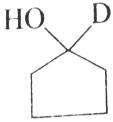
The metal hydrogen bond in $LiAlH_4$ is more than polar than metal hydrogen bond in $NaBH_4$. As a result $LiAlH_4$ is strong reducing agent than $NaBH_4$. Esters, carboxylic acids, amides cannot be reduced by $NaBH_4$

The carbonyl group of amide of reduced to methylene group by $LiAlH_4$

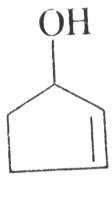
$$\begin{array}{c}
 & \xrightarrow{\text{NaBD}_4} & X'; \text{ Identify } X: \\
 & \xrightarrow{\text{H}_2O} & X'; \text{ Identify } X:
\end{array}$$



A.



В.



D.

C.



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130. Carbon oxygen double bond are easily reduced by $NaBH_4$ or $LiAlH_4$. The actual reducing agent in these reduction is hrdride ion $\left(H^-\right)$

$$H \xrightarrow{\bullet} C \xrightarrow{\bullet} C$$

The metal hydrogen bond in $LiAlH_4$ is more than polar than metal hydrogen bond in $NaBH_4$. As a result $LiAlH_4$ is strong reducing agent than $NaBH_4$. Esters, carboxylic acids, amides cannot be reduced by $NaBH_4$

The carbonyl group of amide of reduced to methylene group by $LiAlH_4$

$$CH_3$$
— C — CH_2 — CH — CH_2 $\xrightarrow{LiAlH_4}$ 'Y'; Identify 'Y':

A.
$$CH_3 - \overset{OH}{CH} - CH_2 - \overset{OD}{CH} - CH_3$$

B.
$$CH_3 - \overset{OH}{\overset{|}{C}} - CH_2 - CH_3 - OD$$

$$\overset{D}{\overset{OD}{\overset{OD}{OD}}} \quad \overset{OD}{\overset{OD}{\overset{OD}{OD}}}$$
C. $CH_3 - \overset{|}{CH} - CH_2 - \overset{|}{OH} - CH_3$
D. $CH_3 - \overset{|}{CH} - CH_2 - \overset{|}{CH} - CH_3$

Answer: D



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131. An organic compound (A) on treatment with $CHCl_3$ and KOH gives (Y) and (Z) both of which in turn gives the same compound (T) when distilled with Zn. Oxidation of (T) Yields (S) of formula $C_7H_6O_2$. The sodium salt of (S) with sodalime gives (P) which can also be obtained by distilling (X).

The molecular weight of compound (X) is:

A. 122

B. 94

C. 106

D. 78

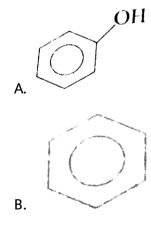
Answer: B

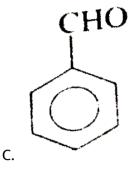


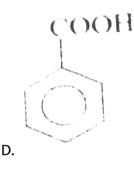
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132. An organic compound (A) on treatment with $CHCl_3$ and KOH gives (Y) and (Z) both of which in turn gives the same compound (T) when distilled with Zn. Oxidation of (T) Yields (S) of formula $C_7H_6O_2$. The sodium salt of (S) with sodalime gives (P) which can also be obtained by distilling (X).

The compound (T) is







Answer: C



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133. An organic compound (A) on treatment with $CHCl_3$ and KOH gives (Y) and (Z) both of which in turn gives the same compound (T) when distilled with Zn. Oxidation of (T) Yields (S) of formula $C_7H_6O_2$. The sodium salt of (S) with sodalime gives (P) which can also be obtained by

distilling (X).

Compounds (Y) and (Z) could be:

Answer: B



134. A tertiary alcohol (H) upon acid-catalysed dehydration gives a product (I). Ozonolysis of (I) leads to compounds (J) and (K). Compound (J) upon reaction with KOH gives benzyl alcohol and a compound (L), whereas (K) on reaction with KOH gives only (M).

$$(M) = \begin{array}{c} H_3C \\ Ph \end{array}$$

Compound (H) is formed by the reaction of:

$$\begin{array}{c} O \\ + PhMgBr \\ CH_3 \end{array}$$

B. Ph
$$CH_3$$
 + PhCH₂MgBr

$$\begin{array}{c} O & + CH_3 \\ D. & Ph \end{array} \begin{array}{c} Ph & + Ph \end{array} \begin{array}{c} CH_3 \\ \end{array}$$

Answer: B



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135. A tertiary alcohol (H) upon acid-catalysed dehydration gives a product (I). Ozonolysis of (I) leads to compounds (J) and (K). Compound (J) upon reaction with KOH gives benzyl alcohol and a compound (L), whereas (K) on reaction with KOH gives only (M).

$$(M) = \begin{array}{c} H_3C \\ Ph \end{array}$$

The structurer of compound (I) is:

A.
$$Ph$$
 CH_3
 Ph
 H_3C
 Ph
 Ph
 Ph
 Ph

C.

$$Ph \xrightarrow{H_3C} CH_3$$

Answer: A



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136. A tertiary alcohol (H) upon acid-catalysed dehydration gives a product (I). Ozonolysis of (I) leads to compounds (J) and (K). Compound (J) upon reaction with KOH gives benzyl alcohol and a compound (L), whereas (K) on reaction with KOH gives only (M).

$$(M) = \begin{array}{c} H_3C \\ Ph \end{array}$$

The structures of compounds (J), (K), and (L), respectively, are:

A.

$$Ph-\overset{\circ}{C}-CH_3, Ph-CH_2-\overset{\circ}{C}-CH_3 \text{ and } Ph-CH_2-\overset{\circ}{C}-O^{\Theta}$$
 B. $Ph-\overset{\circ}{C}-H, Ph-CH_2-\overset{\circ}{C}-H \text{ and } Ph-\overset{\circ}{C}-O^{\Theta}K^{\oplus}$ C. $Ph-\overset{\circ}{C}-CH_3, Ph-CH_2-\overset{\circ}{C}-H \text{ and } CH_3-\overset{\circ}{C}-O^{\Theta}K^{\oplus}$ D. $Ph-\overset{\circ}{C}-H, Ph-\overset{\circ}{C}-CH_3 \text{ and } Ph-\overset{\circ}{C}-O^{\Theta}K^{\oplus}$

Answer: D



137. Alcohols are converted to tosylates by treatment with p-toluence sulfonyl chloride (TsCl) in the presence of pyridine. This overall process converts a poor leaving group $\begin{pmatrix} \Theta \\ H \end{pmatrix}$ into good one $\begin{pmatrix} \Theta \\ Ts \end{pmatrix}$. A tosylate is a good leaving group its conjugates acid p-touence sulfonic acid is strong acid. Beacuse alkyl tosylates have food leaving groups, they undergo both nucleophilic substitution and β — elimination.

$$CH_{3}CH_{2} \xrightarrow{O} H + Cl \xrightarrow{S} CH_{3} \xrightarrow{Pyridine}$$

$$CH_{3}CH_{2} \xrightarrow{O} CH_{3} + CH_{3}CH_{2} \xrightarrow{O} CH_{$$

Find the major product of the following reaction:

$$\begin{array}{c} H_3C \\ H \\ C \\ CH_3CH_2 \end{array} \longrightarrow \begin{array}{c} O \\ \parallel \\ CH_3CH_2 \end{array} \longrightarrow \begin{array}{c} O \\ \parallel \\ CH_3CH_3 \end{array} \longrightarrow \begin{array}{c} O \\$$

$$CH_3-CH=C < \begin{matrix} H \\ CH_3 \end{matrix}$$

$$\begin{array}{ccc} & & & & \\ & & & \\ \text{(d)} & & & \\ \text{H}_3\text{C} & & \\ \text{C} & & \\ \text{C} & & \\ \text{CH}_2\text{CH}_3 & \\ \\ \text{D.} & & \\ \text{NC} & & \\ \end{array}$$

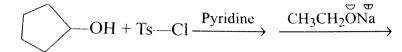
138. Alcohols are converted to tosylates by treatment with p-toluence sulfonyl chloride (TsCl) in the presence of pyridine. This overall process converts a poor leaving group $\begin{pmatrix} \Theta \\ H \end{pmatrix}$ into good one $\begin{pmatrix} \Theta \\ Ts \end{pmatrix}$. A tosylate is a good leaving group its conjugates acid p-touence sulfonic acid is strong acid. Beacuse alkyl tosylates have food leaving groups, they undergo both nucleophilic substitution and β — elimination.

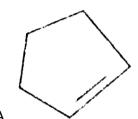
$$CH_{3}CH_{2} \xrightarrow{OH} + CI - S \xrightarrow{O} CH_{3} \xrightarrow{Pyridine}$$

$$O = CH_{3}CH_{2} \xrightarrow{O} CH_{3} + CH_{3}CH_{2} \xrightarrow{OTS} \uparrow Good leaving group$$

$$O = CH_{3}CH_{2} \xrightarrow{OTS} \uparrow Good leaving group$$

What would be the major product of the following reactions?





OCH₂CH₃



$$\sim$$
 O-CH=CH₂

Answer: A



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139. Alcohols are converted to tosylates by treatment with p-toluence sulfonyl chloride (TsCl) in the presence of pyridine. This overall process

converts a poor leaving group $\begin{pmatrix} \Theta \\ H \end{pmatrix}$ into good one $\begin{pmatrix} \Theta \\ Ts \end{pmatrix}$. A tosylate is a good leaving group its conjugates acid p-touence sulfonic acid is strong acid. Beacuse alkyl tosylates have food leaving groups, they undergo both nucleophilic substitution and β — elimination.

$$CH_{3}CH_{2} \xrightarrow{OH} + CH_{3} \xrightarrow{Pyridine}$$

$$CH_{3}CH_{2} \xrightarrow{O} CH_{3} \xrightarrow{Pyridine}$$

$$CH_{3}CH_{2} \xrightarrow{O} CH_{3} + CH_{3} + CH_{3} + CH_{3}$$

$$Or$$

$$CH_{3}CH_{2} \xrightarrow{OTs} \uparrow Good leaving group$$

$$NaOCH_{3} + CH_{3}CH_{2} \xrightarrow{OTs} CH_{3}CH_{2} \xrightarrow{OCH_{3}} + NaOTs$$

$$KOH + H \xrightarrow{CH_{2}} CH_{2} \xrightarrow{OTs} \xrightarrow{E_{2}} H_{2}C \Rightarrow CH_{2} + KOTs$$

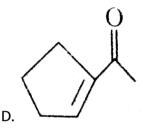
$$(Alcoholic)$$

Idetify the final product of the following sequences of reactions:

$$\begin{array}{c}
\text{OH} \quad \text{TsCl} \\
\hline
\text{Pyridine}
\end{array}$$

$$\begin{array}{c}
\text{alc. KOH} \quad \text{OsO}_4
\end{array}$$

$$\begin{array}{c}
\text{HIO}_4
\end{array}$$



Answer: B



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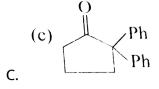
140. Acid catalysed conversation of 1,2-diol or vicinal, into carbonyl compound known as pinocaol-pinacolone rearrangement.

Generally more electron donating group migrate during mechanism, migration of H is faster because of its smaller size.

What would be the major product of reaction?

$$\begin{array}{c}
OH \\
Ph \\
OH
\end{array}$$

$$\begin{array}{c}
OH \\
OH
\end{array}$$



Answer: C



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141. Acid catalysed conversation of 1,2-diol or vicinal, into carbonyl compound known as pinocaol-pinacolone rearrangement.

Generally more electron donating group migrate during mechanism,

migration of H is faster because of its smaller size.

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& &$$

In this sequence of reaction final product is"

В.



Answer: D



Watch Video Solution

142. Acid catalysed conversation of 1,2-diol or vicinal, into carbonyl compound known as pinocaol-pinacolone rearrangement.

Generally more electron donating group migrate during mechanism, migration of H is faster because of its smaller size.

Which of the following is not correct about this rearrangement?

$$A. \quad OR \quad R \qquad CI$$

- B. The carboncation is stabilised by 1,2-shift
- C. Migration aptitude for substituent is in $R-\ > H-\ > C_6 H_5$
- D. Product of reaction is carbonyl compound.



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143. Compound (A) $C_{10}H_{22}0_2$ is insoluble in aq. NaOH bu not is $NaHCO_3$

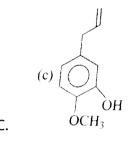
. Treatment of (A) with DMSO
$$\left(CH_3-\overset{S}{S}-CH_3
ight)$$
 in alkali give (B)

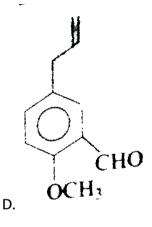
 $C_{11}H_{14}O_2$. Treatment of (A) with strong alkali alone give an isomeric compound (C). When (A) is reflux with HI, CH_3I is obtained, compound (B) is insoluble in alkali and decolurises Br_2/CCl_4 . (B) on treating with strong base gives (D), an isomer of (B). Ozonolysis (C) of gives (E), C_8H_8O and isomer of vanilline. Ozolysis of (D) gives (F) $C_9H_{10}O_3$, which is identical with product of methylation of vanilline (4-hydroxy-3-methoxy benzaldehyde).

Structure of compound (A) is:

A.

В.





Answer: C



View Text Solution

144. Compound (A) $C_{10}H_{22}0_2$ is insoluble in aq. NaOH bu not is $NaHCO_3$

. Treatment of (A) with DMSO $\left(CH_3-\overset{S}{S}-CH_3
ight)$ in alkali give (B)

 $C_{11}H_{14}O_2$. Treatment of (A) with strong alkali alone give an isomeric compound (C). When (A) is reflux with HI, CH_3I is obtained, compound (B) is insoluble in alkali and decolurises Br_2/CCl_4 . (B) on treating with strong base gives (D), an isomer of (B). Ozonolysis (C) of gives (E), C_8H_8O and isomer of vanilline. Ozolysis of (D) gives (F) $C_9H_{10}O_3$, which is identical with product of methylation of vanilline (4-hydroxy-3-methoxy benzaldehyde).

Compound (B) is:

Answer: B



View Text Solution

145. Compound (A) $C_{10}H_{22}0_2$ is insoluble in aq. NaOH bu not is $NaHCO_3$

. Treatment of (A) with DMSO $\left(CH_3-\overset{S}{S}-CH_3
ight)$ in alkali give (B)

 $C_{11}H_{14}O_2$. Treatment of (A) with strong alkali alone give an isomeric compound (C). When (A) is reflux with HI, CH_3I is obtained, compound (B) is insoluble in alkali and decolurises Br_2/CCl_4 . (B) on treating with strong base gives (D), an isomer of (B). Ozonolysis (C) of gives (E), C_8H_8O and isomer of vanilline. Ozolysis of (D) gives (F) $C_9H_{10}O_3$, which is identical with product of methylation of vanilline (4-hydroxy-3-methoxy benzaldehyde).

Compound (E) and (F) are respectively:

D. None of these

Answer: A



View Text Solution



146.

Column (H)

- P. White turbidity with HCl/ZnCl₂
- Q. Violet colour with FeCl₃
- R. Colour change of Na₂Cr₂O₇, H[⊕]

S. $I_2/O^{\odot}H$, gives bright yellow ppt.



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(a)
$$CH_3 - CH_2 - C - O - CH - CH_2CH_3 \xrightarrow{OH} P$$
. Involve carbocation formation

O

CH₃

(b) $CH_3 - C - O - CH - Ph \xrightarrow{H^{\oplus} / H_2O} Q$. Acyl cleavage

CH₃



147.

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148.

- Acid catalysed rearrangement (a)(P)Fries rearrangement
- (Q) Concerned with ester (b) Claisen rearrangement
- Buyer-Viliger's rearrangement Involve electrophilic substitut: (c)(R)
- Pinacole-Pinacolone (d)(S)Intramolecular rearrangement



P. Acid catalysed rearrangement (a) Fries rearrangement O. Concerned with ester (b) Claisen rearrangement R. Involve electrophilic substitution (c) Baver-Villiger's rearrangement S. Intramolecular rearrangement (d) Pinacole-Pinacolone rearrangement 149. (d) Ph—C—CH—CH₂—CH₃ $\xrightarrow{\text{Conc. H}_2\text{SO}_4} S$. Bimolecular elimination **View Text Solution** P_{*} KMnO₄, Δ , Θ H . (a) Oxidation of 1° alcohol in aldehyde O. Collin's reagent (c) $\left(\bigcap_{N}\right)$ · CrO₃ R. Jone's reagent S. PCC 150. (d) Oxidation of alkyne into acid **View Text Solution** 151. Identification of 1°, 2° and 3° Alcohol (a) (P) Oxyme Identification of 1°, 2° and 3° Nitroalkane (b) (Q) Cu/30Formation of alcohol by anti-Markownikoff's additioin (R)Victor (c)Formation of alcohol by Markownikoff's addition (S)(d)Huydr

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152.

 $((d), Picric acid + NaHCO_3, (S), CO_2gas is envoled)$



153.

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8. (a)
$$CH_3 = C = O = C = Ph \xrightarrow{H = H_2O} P$$
. Unimolecular

Ph

CH₃ O

(b) $CH_3 = C = C = O = Ph \xrightarrow{H^{\oplus}/H_2O} Q$. Bimolecular

CH₃

CH₃

CH₃

CH₃

CH₃
 $CH_3 = C = C = C = C$
 $CH_3 = C = C = C$
 $CH_3 = C$

(d) CH_3 —O—CH— CH_3 — $\overset{HI}{\longrightarrow}$

CH₃

cleavage

S. Acyl oxygen bond cleavage

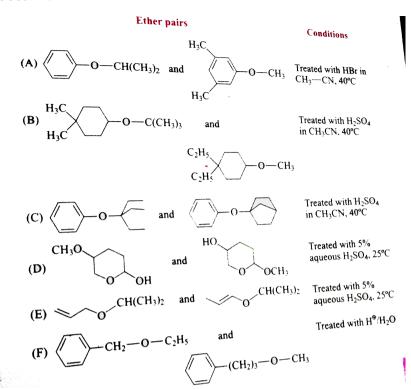
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154. How many compounds A through G are enol tautomer of 2-butanone?

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155. Consider the pairs of ethers A through F shown below. To the right of each pairs is a description of reaction conditions to be applied to each. One compound of the pair will react more rapidly than the other. Find out number of reactions in which first ether more rapidly cleaved than

second.



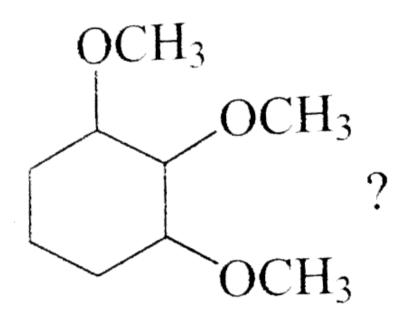
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156. Find out number of moles of HIO_4 that will react with following compund

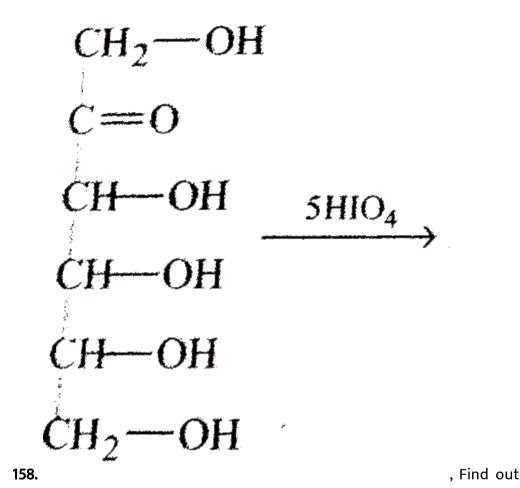
CHOHCHOHCHOHCHOH CH_2OH



157. How many mole of 'HI' will react with



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the value of 'X'

○ Wa

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159. $R-CH_2-OH \stackrel{?}{\longrightarrow} R-CH_2-Cl$

find out the number that can be used for above conversion, from the

following.

 $HCl, ZnCl_2, PCl_3, PCl_5, POCl_3, SOCl_2, NaCl, TsCl$



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160. Identify number of alcohols those will show rearrangement during dehydration will concentrate H_2SO_4 .



161. Find out number of reagents that converts $1(\circ)$ alcohols to

$$\begin{array}{c} KMnO_4|H^{\oplus}|\Delta \quad , \quad K_2Cr_2O_7|\, Dil. \,\, H_2SO_4 \quad , \quad \text{Ceric ammonium nitrate} \\ O \quad \qquad (B) \quad \qquad (C) \\ H_3C-S-CH_3 \,\, (DMSO) \quad , \quad \left(\begin{array}{c} \\ \\ \\ \\ \end{array} \right)^{\oplus} N-H \\ \begin{array}{c} \\ \\ \\ \end{array} Cr_2O_7^{-2} \quad , \quad \left(\begin{array}{c} \\ \\ \\ \end{array} \right)^{\oplus} CrO_3 \end{array}$$

aldehyde.



162. Find out number of alcohols that can give positive iodoform test.

OH OH OH OH

OH
$$CH-CH_3$$
 OH

 CH_3-CH_2-OH OH

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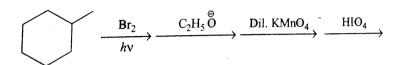
163. How many mole of HI racts with glycerol to give 2-iodopropane?



164. Find out final product of following reactions:

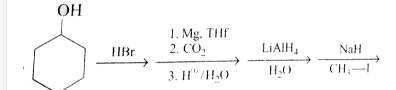


165. Find out final product of following reactions:





166. Find out final product of following reactions:



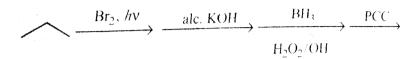


167. Find out final product of following reactions:

$$\begin{array}{c|c}
OH & & & \\
SOCI_2 & \xrightarrow{C_2H_5 \text{ ONa}} & \xrightarrow{Br_2} & \xrightarrow{2\text{NaNH}_4} & \text{H}_2. \text{ Pd-BaSO}_4
\end{array}$$



168. Find out final product of following reactions:



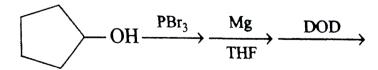


169. Find out final product of following reactions:

$$\text{Ph--CH}_2 - \text{OH} \xrightarrow{\text{SOCl}_2} \xrightarrow{\text{or PCl}_3} \xrightarrow{\text{NaSH}} \xrightarrow{\text{NaOH}} \xrightarrow{\text{Ph--CH}_2 - - \text{Cl}}$$



170. Find out final product of following reactions:





171. Find out final product of following reactions:

$$\begin{array}{c}
 & \xrightarrow{\text{KMnO}_4} \xrightarrow{\text{C}_2\text{H}_5\text{OH}} \xrightarrow{\text{PhMgBr}} \\
 & \xrightarrow{\text{H}^{\oplus}} /\text{H}_2\text{O}
\end{array}$$

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172. Find out final product of following reactions:

O View Text Solution

173. Find out final product of following reactions:

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