

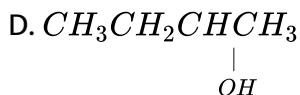
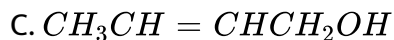
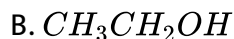
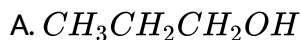
CHEMISTRY

BOOKS - GRB CHEMISTRY (HINGLISH)

ALCOHOLS AND ETHERS

EXERCISE 1

1. An alkene obtained by the dehydration of an alcohol (A), on ozolysis gives two molecules of acetaldehyde for ever molecule of alkene. The alcohol (A) is



Answer: D

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EXERCISE 2

1. $R \rightarrow COOH \rightarrow R \rightarrow CH_2OH$. This mode of reduction of an acid to alcohol can be effected by:

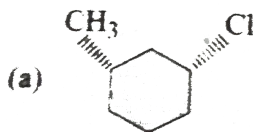
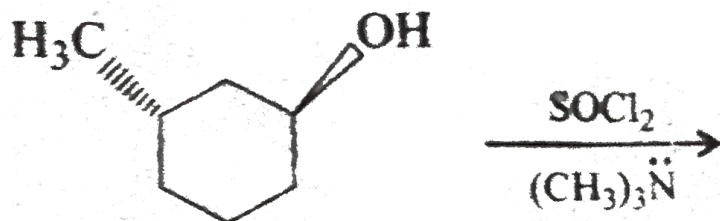
- A. Zn/HCl
- B. Na-alcohol
- C. Aluminium isopropoxide and isopropyl alcohol
- D. $LiAlH_4$

Answer: D

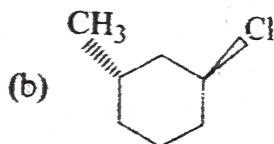
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EXERCISE 3

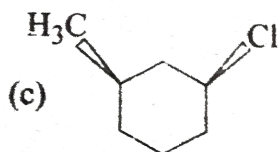
1. Find the major product of the following reaction.



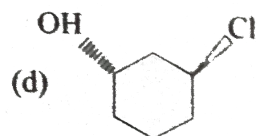
A.



B.



C.



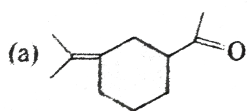
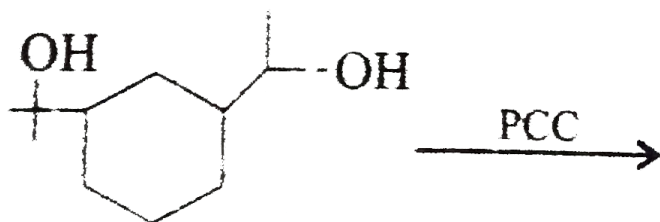
D.

Answer: A

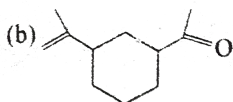
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EXERCISE 4

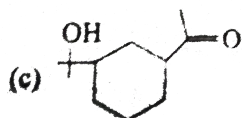
1. What is product of the following reaction:



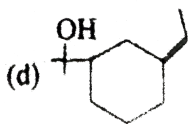
A.



B.



C.



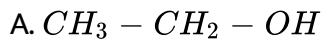
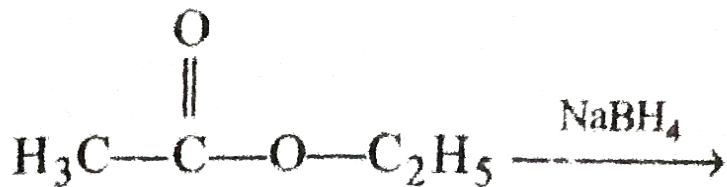
D.

Answer: C

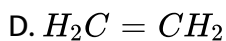
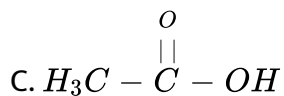
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EXERCISE 5

1. Predict product of the following reaction,



B. No reaction



Answer: B



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EXERCISE 6

1.1 – Propanol and 2 – propanal can be best distinguished by

- A. oxidation with alkaline $KMnO_4$ followed by reaction with Fehling's solution
- B. oxidation with alkaline dichromate followed by reaction with Fehling's solution
- C. oxidation by heating with copper followed by reaction with Fehling's solution
- D. oxidation with alkaline H_2SO_4 followed by reaction with Fehling's solution

Answer: C

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EXERCISE 7

1. On heating glycerol with $KHSO_4/\Delta$, a compound is obtained, which has a bad odour. The compound is:

- A. Acrolein
- B. Formic acid
- C. Allyl alcohol
- D. Methyl isocyanide

Answer: A

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EXERCISE 8

1. A compound X with molecular formula C_3H_8O can be oxidized to a compound Y with the molecular formula $C_3H_6O_2$. X is most likely to be a:

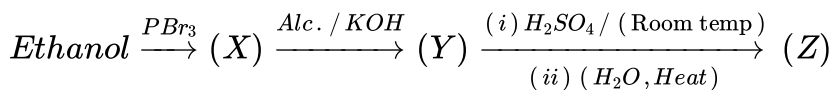
- A. primary alcohol
- B. secondary alcohol
- C. aldehyde
- D. ketone

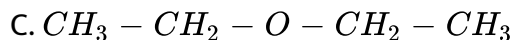
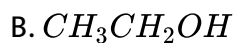
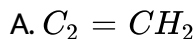
Answer: A

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EXERCISE 9

1. Identify (Z) in the following series.





Answer: B

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EXERCISE 10

1. Which of the following is not characteristic of alcohols?

A. Their boiling points rise fairly uniformly with a rise in molecular weight

B. Lower member have a pleasant smell but burning taste and the higher ones are odourless and tasteless

C. These are lighter than water

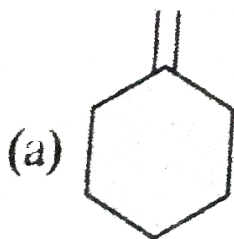
D. Lower member are insoluble in water and organich solvents bu the
solubility goes on increasing with the rise of molecular weight

Answer: D

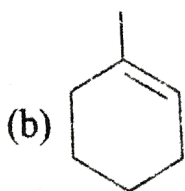
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EXERCISE 11

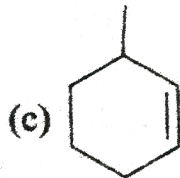
1. A $\xrightarrow{\text{Dil. H}_2\text{SO}_4 / \text{Hg}^+}$ 1 - Methylcyclohexanol. Here A is:



A.



B.



C.

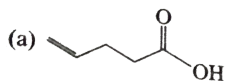
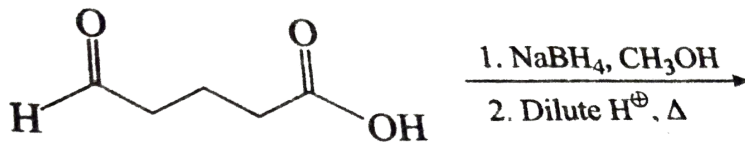
D. (d) (a) or (b)

Answer: D

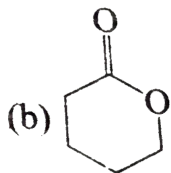
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EXERCISE 12

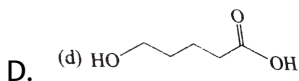
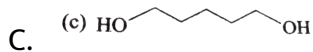
1. Find the product of reaction



A.



B.

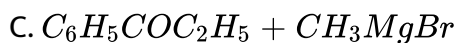
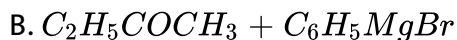
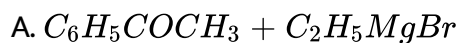


Answer: B

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EXERCISE 13

1. 2-Phenylbutan-2-ol can be prepared by which of the following combinations?



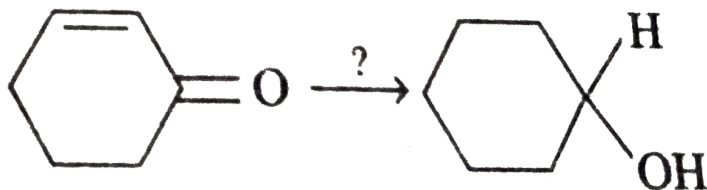
D. All of these

Answer: D

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EXERCISE 14

1. Predict the nature of reducing agent in the following reaction.



A. $LiAlH_4$

B. $NaBH_4$

C. H_2 / Pt

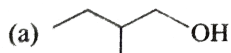
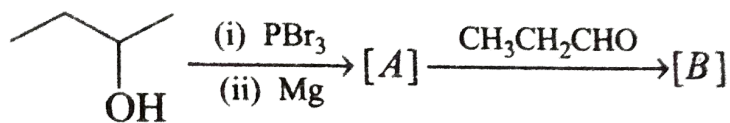
D. Both a and c

Answer: C

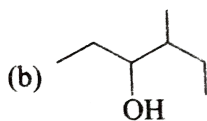
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EXERCISE 15

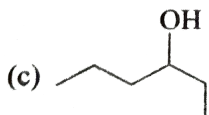
1. The correct structure for compound B will be:



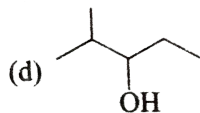
A.



B.



C.



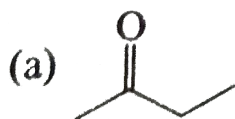
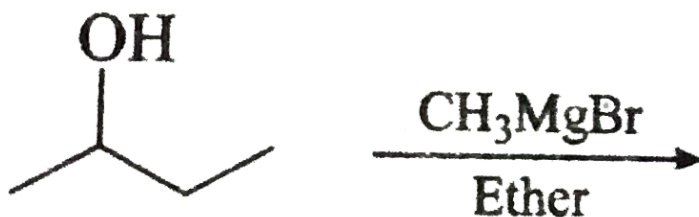
D.

Answer: B

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EXERCISE 16

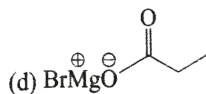
1. Find the product of following reaction,



A.

B. CH_2

C. 



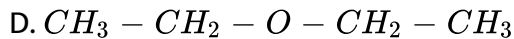
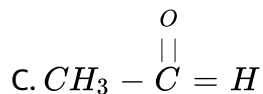
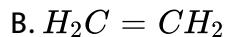
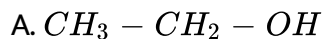
D.

Answer: C

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EXERCISE 17

1. $CH_3 - CH_2 - Br \xrightarrow{\text{Dry } Ag_2O}$ Product of reaction is :



Answer: D

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EXERCISE 18

1. The product when glycerol reacts PCl_5 is:

- A. 1,2,3-trichloropropane
- B. glycerol monochlorohydrin
- C. glycerol dichlorohydrin
- D. All of these

Answer: A

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EXERCISE 19

1. Glycerol $\xrightarrow{KHSO_4}$ A $\xrightarrow{LiAlH_4}$ B.

- A. Acrolein, Allyl Alcohol
- B. glycerol, sulphate, acrylic acid
- C. allyl alcohol, acrolein

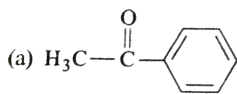
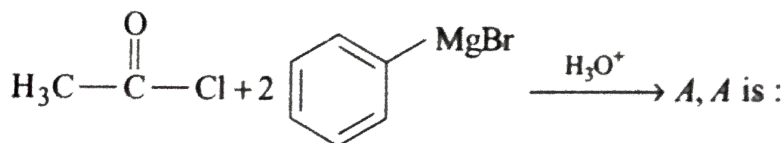
D. only acrolein (B is not formed)

Answer: A

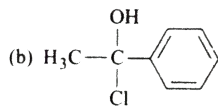
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EXERCISE 20

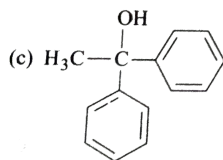
1. Complete the following reaction



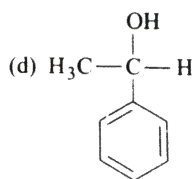
A.



B.



C.



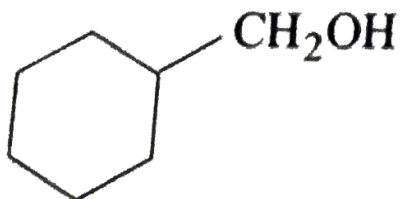
D.

Answer: C

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EXERCISE 21

1. Choose the correct option for the given structure

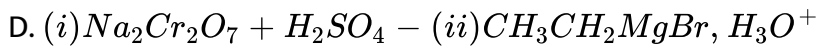


is changed into

A. (i) Cu , 300°C – (ii) $\text{CH}_3\text{CH}_2\text{MgBr}$, H_3O^+

B. (i) CrO_3 – (ii) $\text{CH}_3\text{CH}_2\text{MgBr}$, H_3O^+

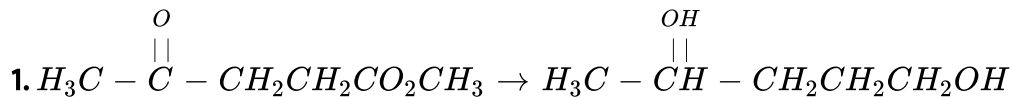
C. (i) KMnO_4 – (ii) $\text{CH}_3\text{CH}_2\text{MgBr}$, H_3O^+



Answer: A

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EXERCISE 22



can be effected using:

A. $LiAlH_4$ and then H^+

B. $NaBH_4$ and then H^+

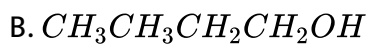
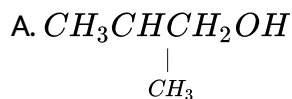
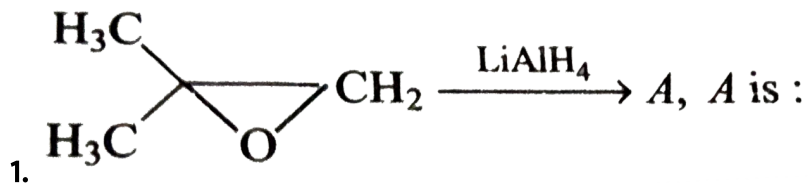
C. H_2 / Pt carbon

D. All of these

Answer: A

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EXERCISE 23



C. No reaction

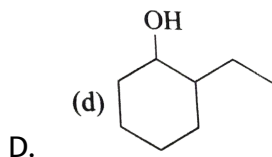
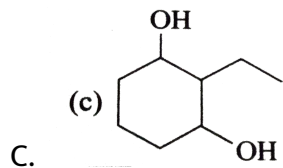
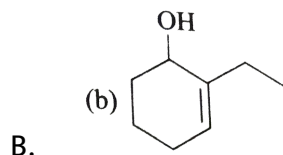
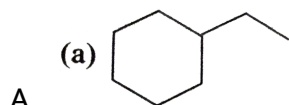
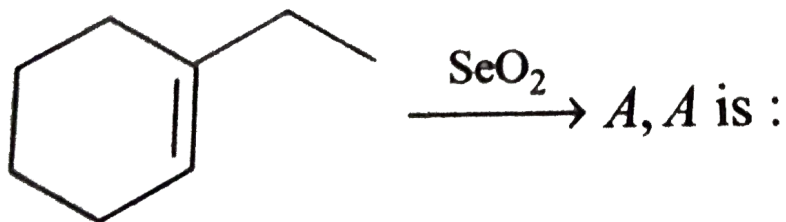
D.

Answer: C

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EXERCISE 24

1. Complete the following reaction

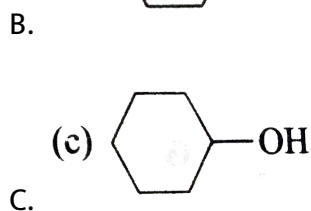
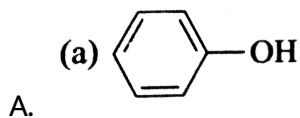
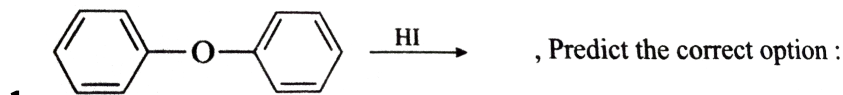


Answer: B



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EXERCISE 25



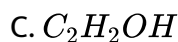
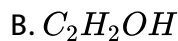
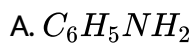
D. No reaction

Answer: D

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EXERCISE 26

1. An aromatic amine (X) was treated with alcoholic potash and another compound (Y) when foul smelling gas was formed C_6H_5NC . The compound (Y) was formed by reacting a compound (Z) with Cl_2 in the presence of slaked lime. The compound (Z) is:



Answer: B

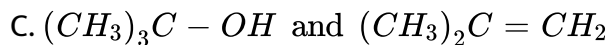
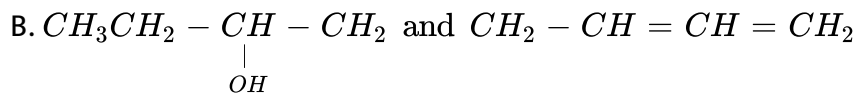
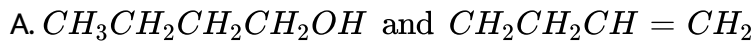


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EXERCISE 27

1. An alcohol (a) on dehydration gives (B), which on ozonolysis gives acetone and formaldehyde. (B) decolourises alkaline $KMnO_4$ solution but

(A) does not .(A) and (B) are respectively:

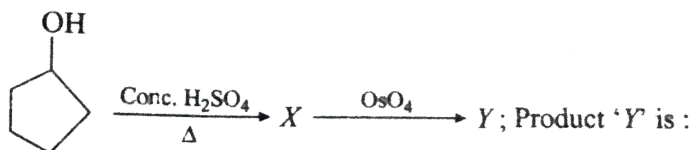


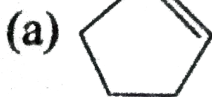
Answer: C

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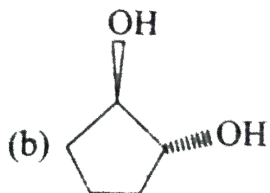
EXERCISE 28

1. Complete the following reaction

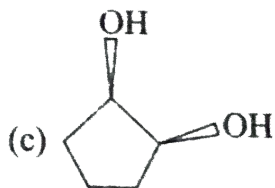




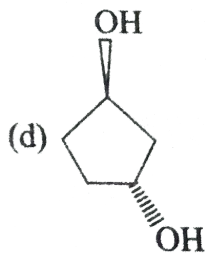
A.



B.



C.



D.

Answer: C



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A. $CH_3CH_2CH_2MgBr$ and hydrolysis

B. $CH_3CH(Br)CH_3$. $AlCl_3$

C. $(CH_3)_2CHMgBr$ and acid hydrolysis

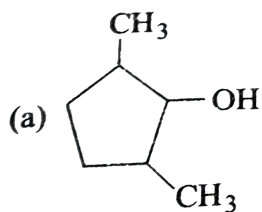
D. $CH_3CHCHCH_3$, Zn

Answer: C

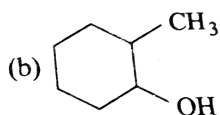
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EXERCISE 31

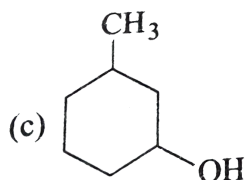
1. Complete the following reaction 



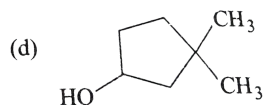
A.



B.



C.



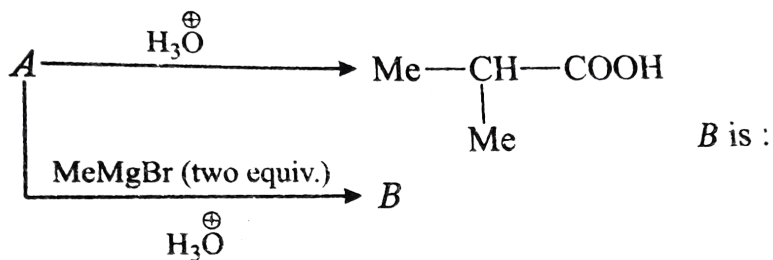
D.

Answer: B

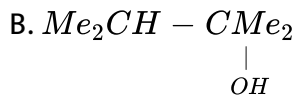
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EXERCISE 32

1. Complete the following reaction



A. Me_2CHCOMe

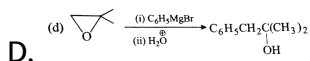
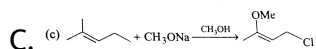
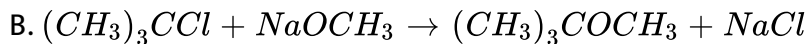
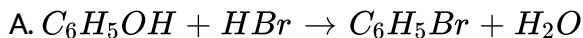


Answer: B

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EXERCISE 33

1. Which of the following reactions is possible?

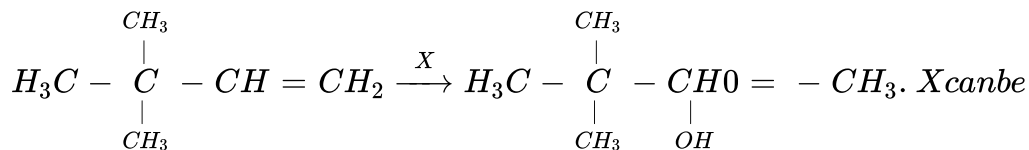


Answer: D

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EXERCISE 34

1.



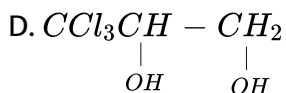
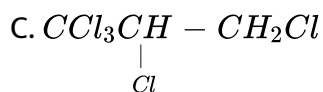
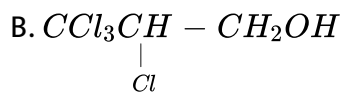
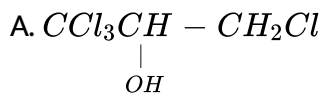
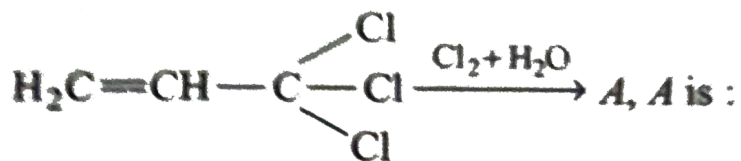
- A. $\text{BH}_3 / \text{THF}, \text{H}_2\text{O}_2 / \text{OH}^-$
- B. H_3O^+
- C. $\text{Hg}(\text{OAc})_2 / \text{NaBH}_4, \text{NaOH}$
- D. All of these

Answer: C

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EXERCISE 35

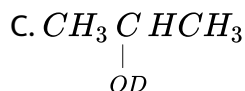
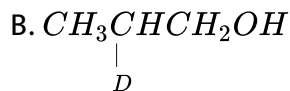
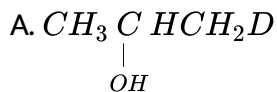
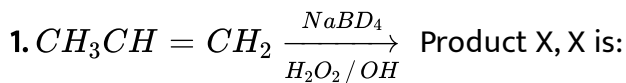
1. Complete the following reaction



Answer: B



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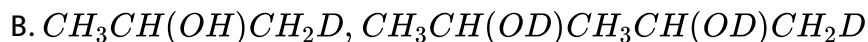
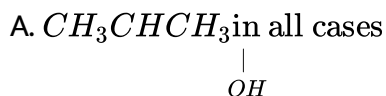
D. None is correct

Answer: B

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EXERCISE 37

1. Identify end product A, B and C of the following:



C. CH_3CHCH_3 in all cases



D. $\text{CH}_3\text{C}^*\text{HCH}_3\text{D}$ in all cases

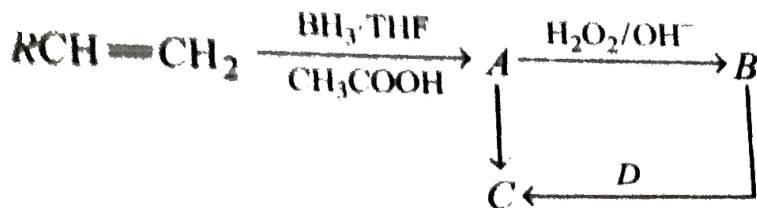


Answer: B

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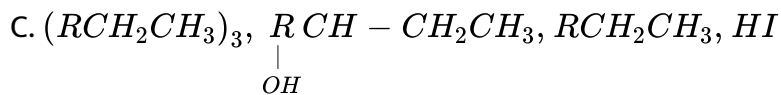
EXERCISE 38

1. Complete the following reaction



A. $(\text{RCH}_2\text{CH}_2)_3$, $\text{RCH}_2\text{CH}_2\text{OH}$, RCH_2CH_3 , HI

B. $(\text{RCH}_2\text{CH}_2)_3$, $\begin{array}{c} \text{RCHCH}_3 \\ | \\ \text{OH} \end{array}$, RCH_2CH_3 , HI

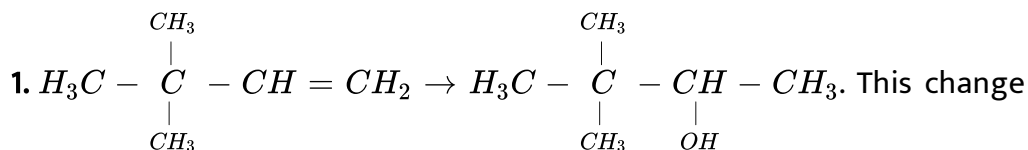


D. None is correct

Answer: A

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EXERCISE 39



can be done by.

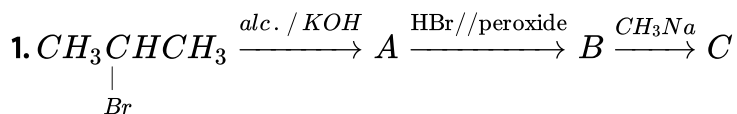
- A. acid catalysed hydration
- B. oxymercuration-demercuration
- C. hydroboration-oxidation
- D. any method mentioned above

Answer: B



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EXERCISE 40



In the above reaction sequence, the final product is:

- A. diethyl ether
- B. 1-methoxypropane
- C. isopropyl alcohol
- D. propylene glycol

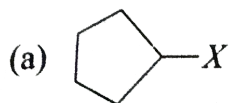
Answer: B



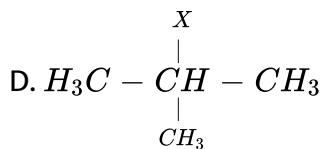
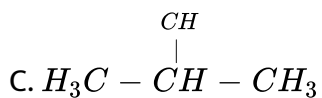
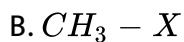
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EXERCISE 41

1. Sodium tertiary butoxide forms ether only with:



A.



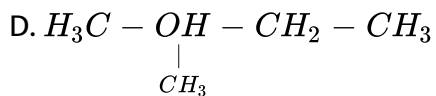
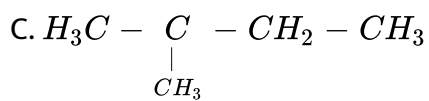
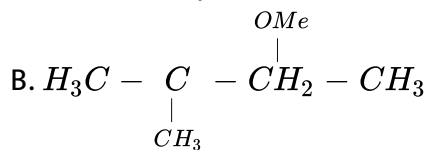
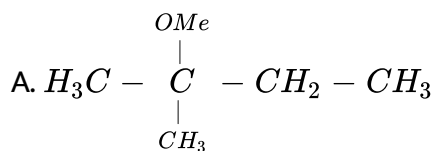
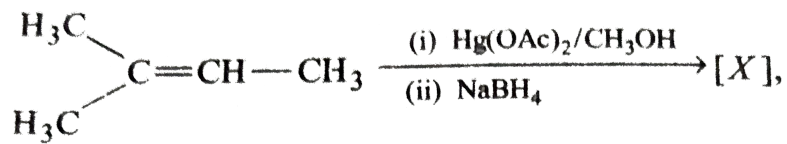
Answer: B



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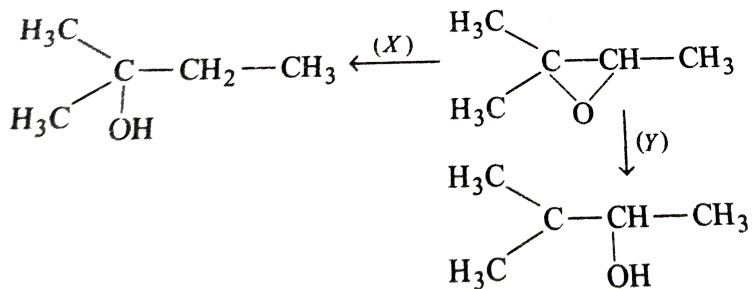
EXERCISE 42

1. In the give reaction



Answer: A

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1.

A. LiAlH_4 and NaBH_4

B. $\text{LiAl} \frac{\text{H}_4}{\text{A}} \text{lCl}_3$ and LiAlH_4

C. LiAlH_4 and $\text{LiAl} \frac{\text{H}_4}{\text{A}} \text{lCl}_3$

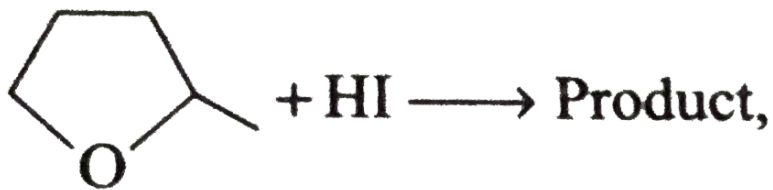
D. H_2 / Ni and H_2 / Pt

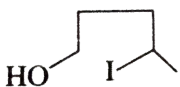
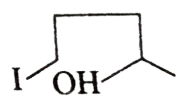
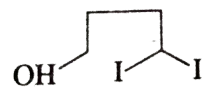
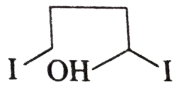
Answer: C

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EXERCISE 44

1. Complete the following reaction



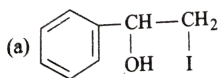
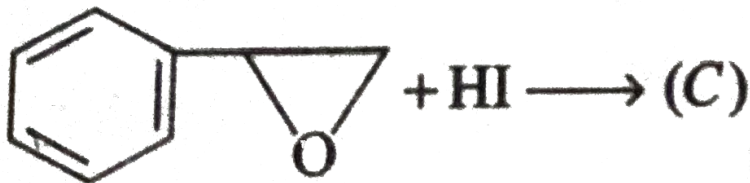
- A. (a)  HO-CH₂-C(CH₃)₂-I
- B. (b)  I-CH₂-C(CH₃)₂-OH
- C. (c)  HO-CH₂-C(CH₃)₂-I₂
- D. (d)  I-CH₂-C(CH₃)(OH)-I

Answer: B

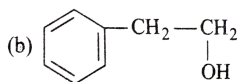


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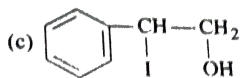
1. Complete the following reaction



A.



B.



C.

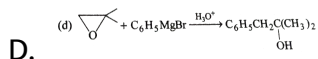
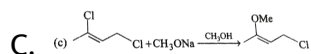
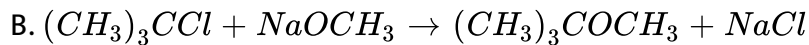
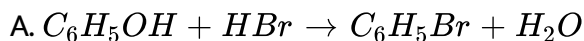
D. None is correct

Answer: A



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1. Which of the following reactions is possible?

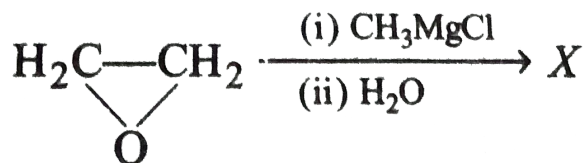


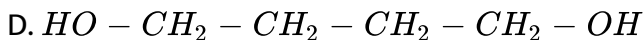
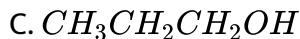
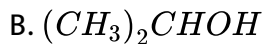
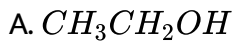
Answer: D

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EXERCISE 47

1. Complete the following reaction



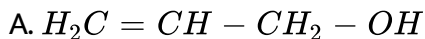
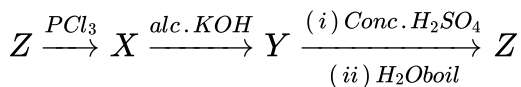


Answer: C

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EXERCISE 48

1. What is Z in the following sequence of reactions?



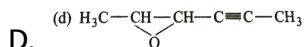
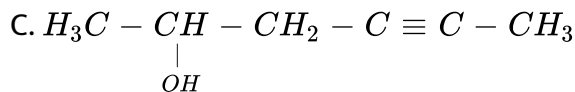
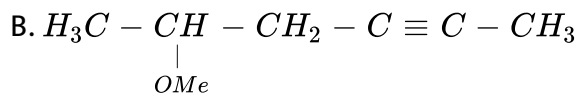
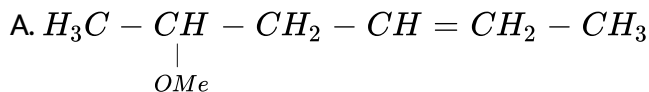
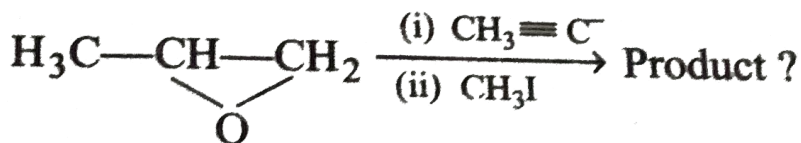


Answer: B

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EXERCISE 49

1. Complete the following reaction



Answer: B

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EXERCISE 50

1. Which of the following reagents cannot be used for the oxidation of 1° alcohol aldehyde?

A. PCC

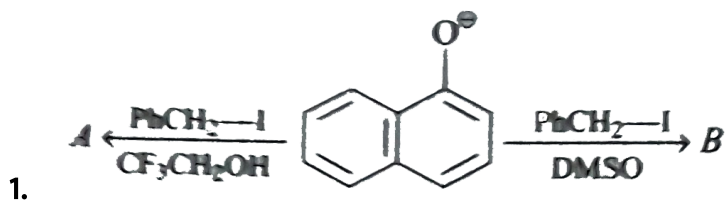
B. Collin's reagent

C. MnO_2

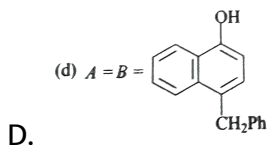
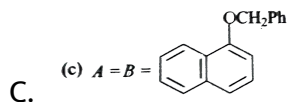
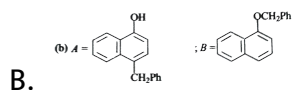
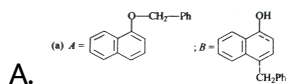
D. MnO_2

Answer: D

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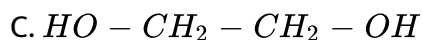
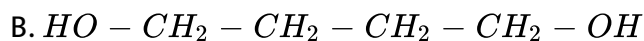
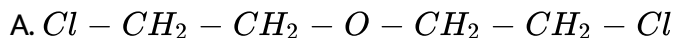
A and B respectively:



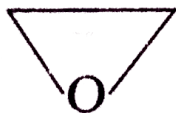
Answer: B

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1. When 2-chloroethanol is reacted with dilute NaOH, the major product formed is:



D. (d)



Answer: D



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level 4

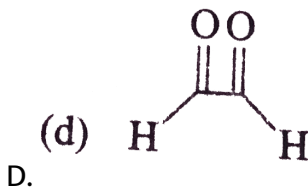
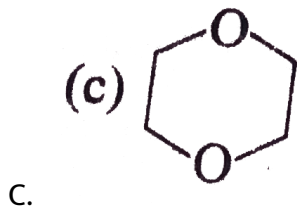
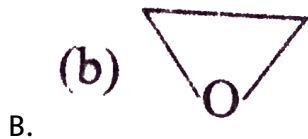
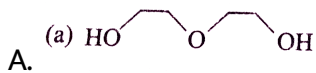
1.

Ethylene

glycol



on heating

with conc. H_2SO_4 gives mainly:

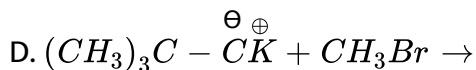
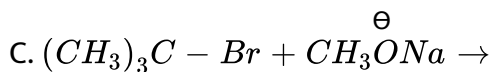
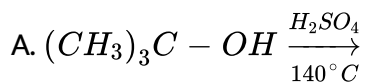
Answer: C



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level 5

1. Which of the following reaction would give the best yield of t-butyl methyl ether ?



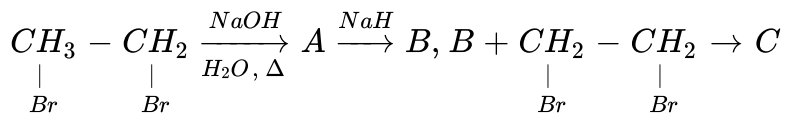
Answer: D



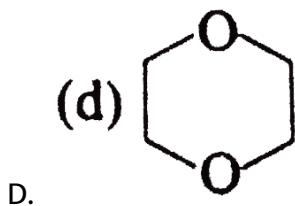
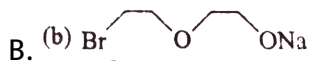
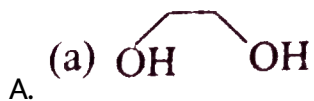
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level 6

1. Consider the following reactions:



The major product formed is:



Answer: D

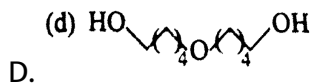
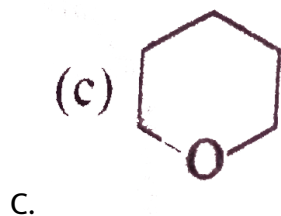
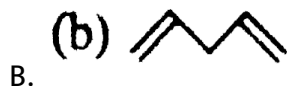
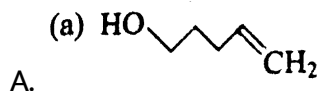


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1. In the reaction



The major product formed is:

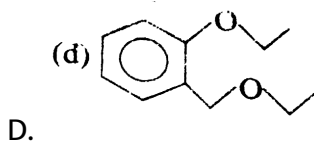
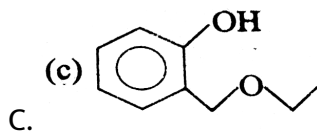
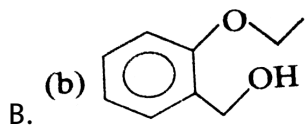
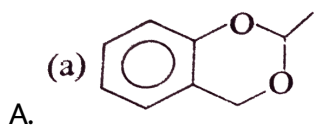


Answer: C



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1. The major product formed in the reaction is"

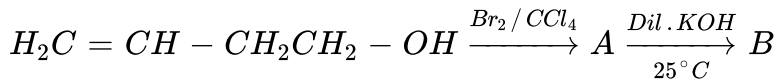


Answer: B

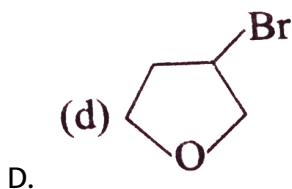
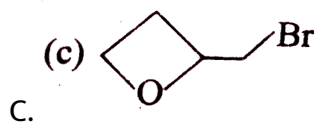
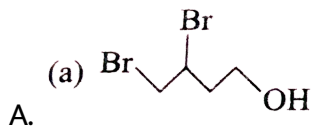


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1. Consider the following reaction



The product B is:



Answer: D

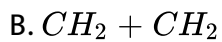


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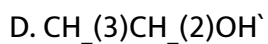
1. Find out correct product of reaction:



A.



C.

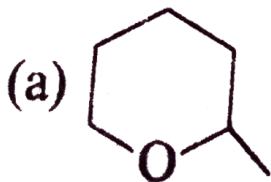
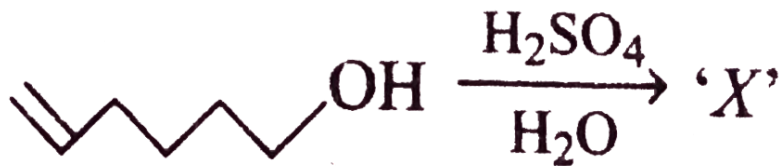


Answer: C

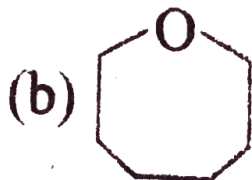


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1. The major product X of the reaction



A.



B.



C.

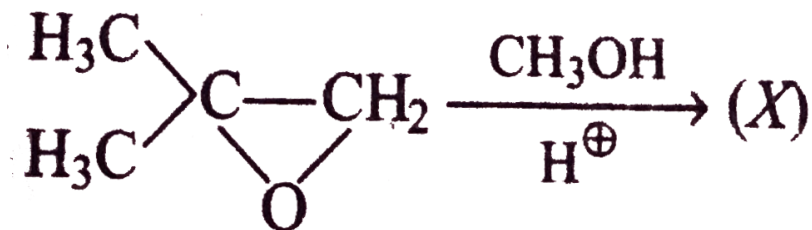


D.

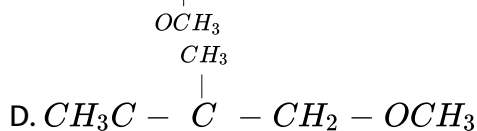
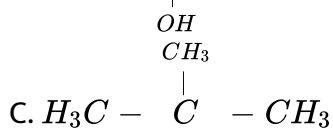
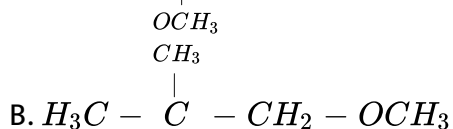
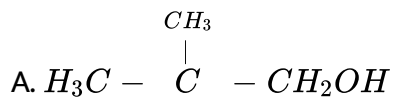
Answer: A

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1. In the reaction



The product X has the structure"

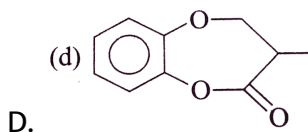
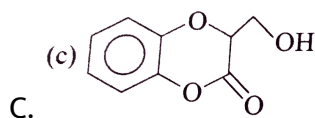
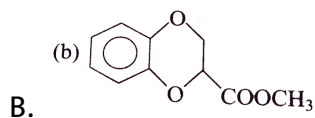
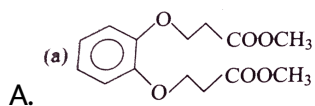
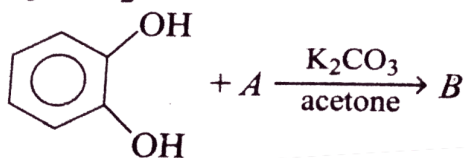


Answer: A



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1. Consider the following sequence of reactions



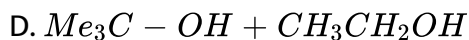
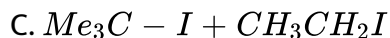
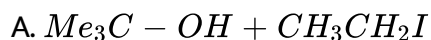
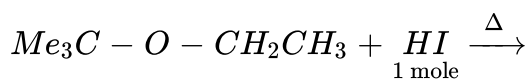
Answer: B



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level 14

1. In the reaction:



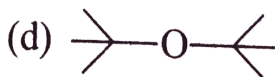
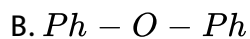
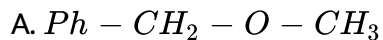
Answer: B



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level 15

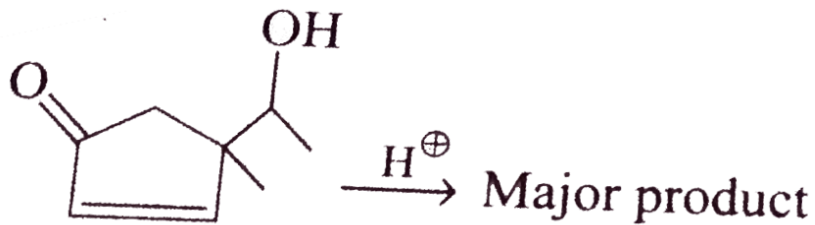
1. Which of the following ethers is the most unreactive to cleavage with conc. HBr?



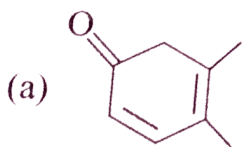
Answer: B



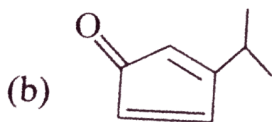
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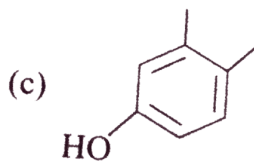
Major Product:



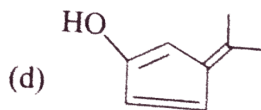
A.



B.



C.



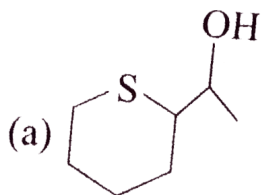
D.

Answer: C

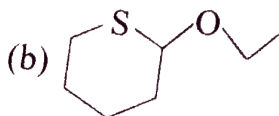


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1. The product of the reaction is:



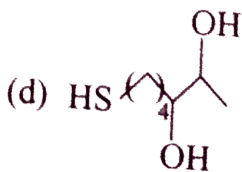
A.



B.



C.



D.

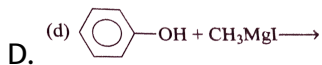
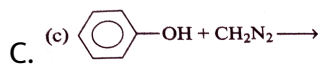
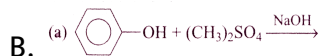
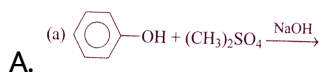
Answer: A



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level 18

1. Which of the following reactions will not result in the formation of anisole?



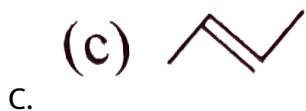
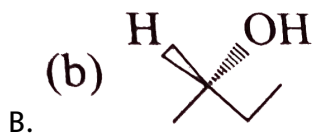
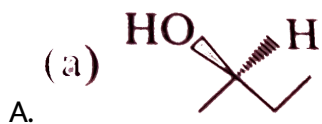
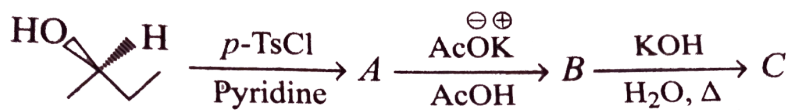
Answer: D



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level 19

1. Consider the following sequence of reactions

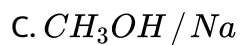
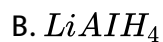
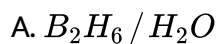
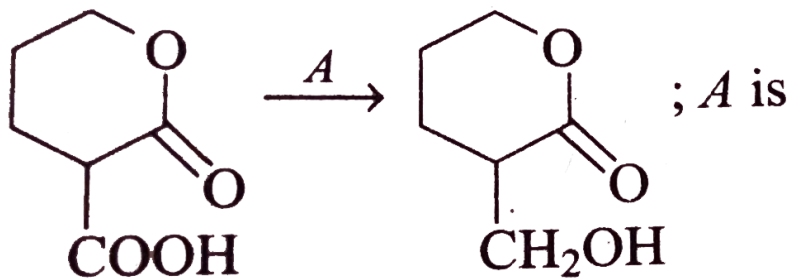


D. (\pm) - 2 butanol

Answer: A

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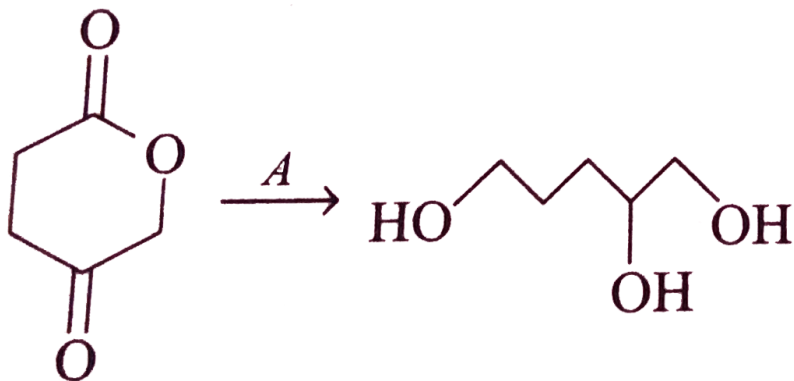
1. Complete the following reaction



Answer: A

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1. Complete the following reaction



A. B_2H_6

B. $LiAlH_4$

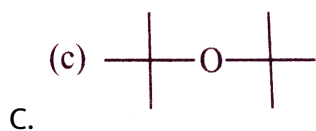
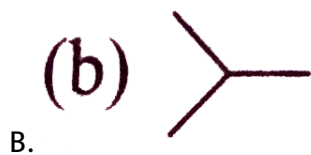
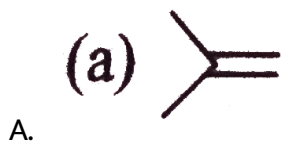
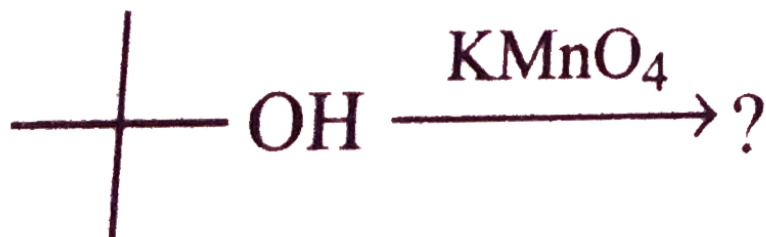
C. Sn/HCl

D. $NaBH_4$

Answer: B

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1. Complete the following reaction

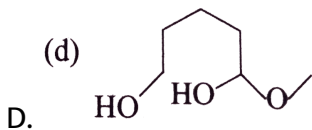
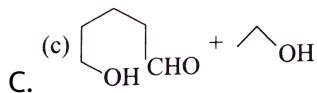
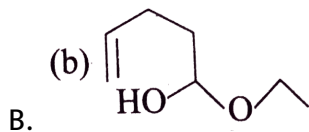
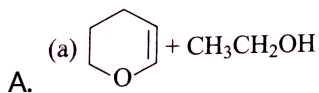
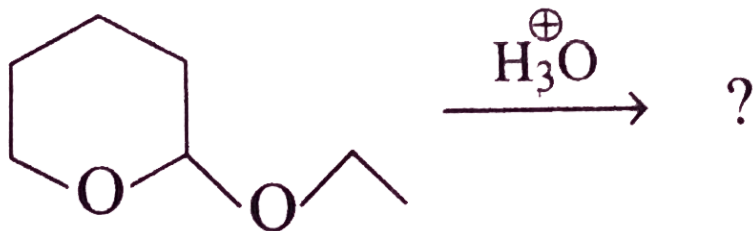


D. No reaction

Answer: D

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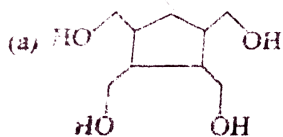
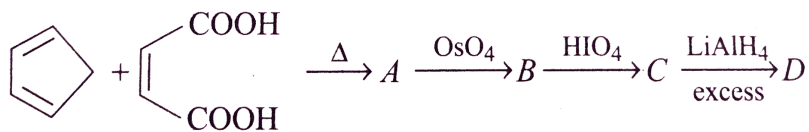
1. The major product formed in the reaction is"



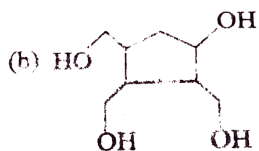
Answer: C

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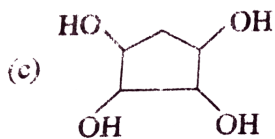
1. Complete the following reaction



A.



B.



C.

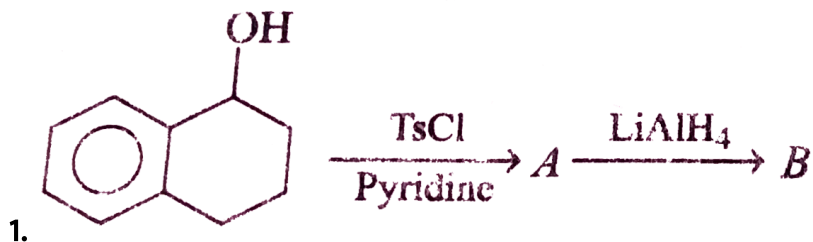


D.

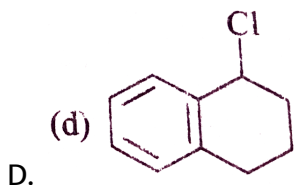
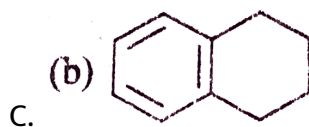
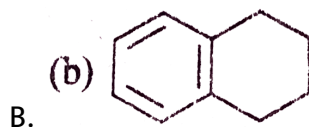
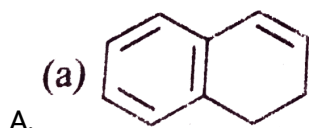
Answer: A



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Product B of the above reaction is:

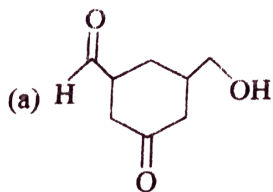
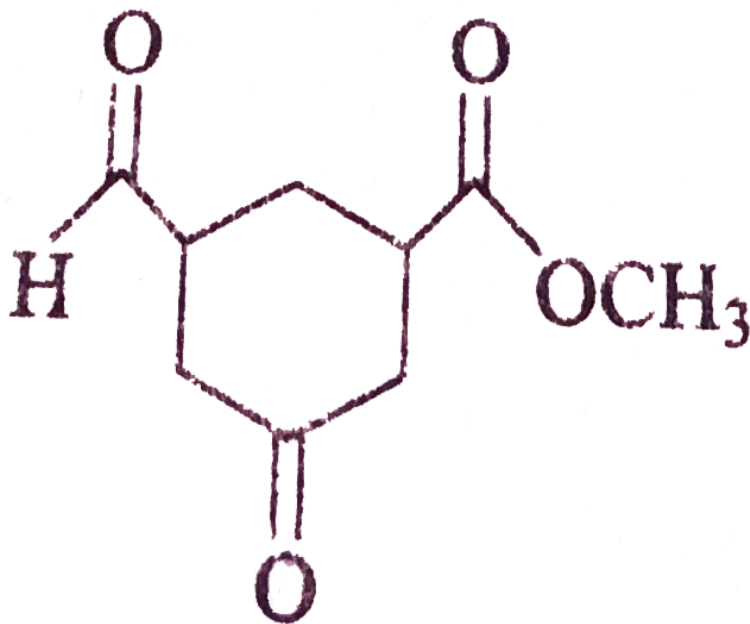


Answer: B

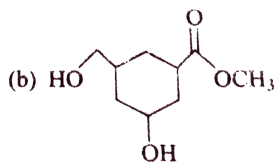


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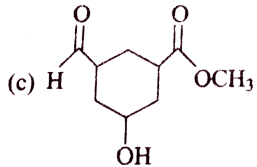
1. Find out the product when compound reacts with NaBH_4 :



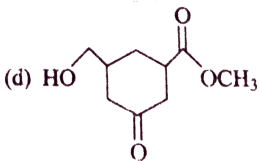
A.



B.



C.

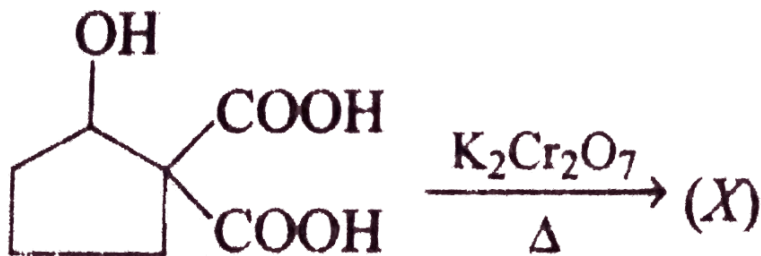


D.

Answer: B

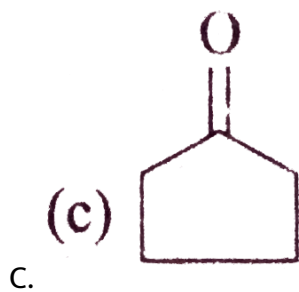
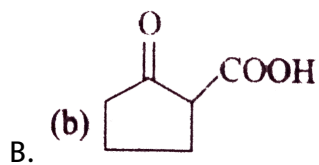
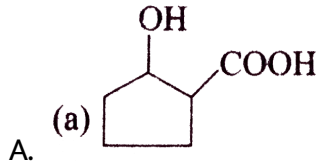
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level 27



1.

Find out X:

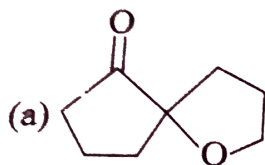
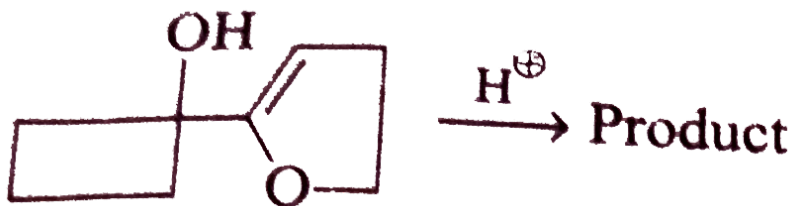


Answer: C

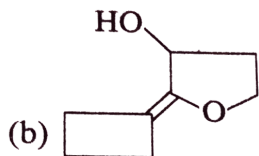


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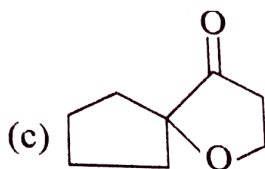
1. Complete the following reaction



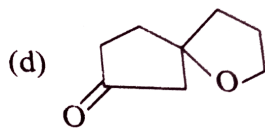
A.



B.



C.

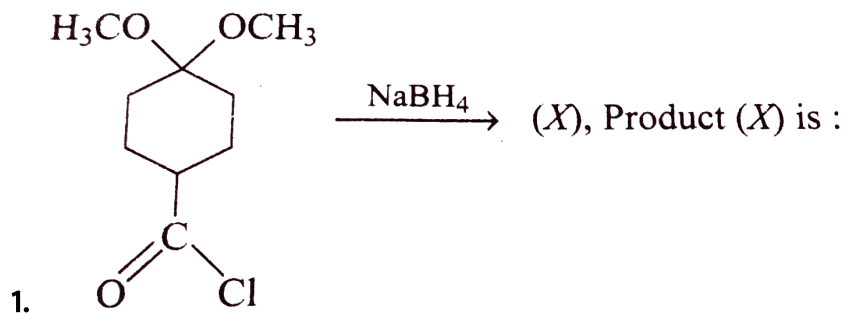


D.

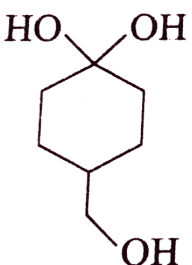
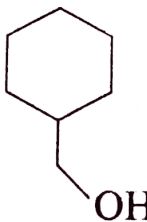
Answer: A

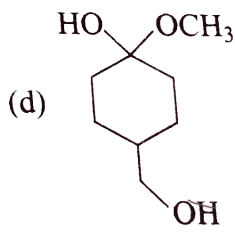
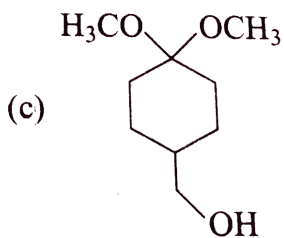


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X, Product is:

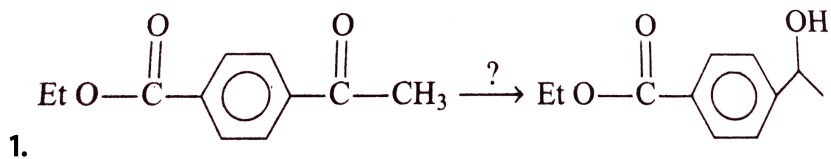
- (a)  A cyclohexane ring with two hydroxyl groups (HO and OH) on the same carbon atom. Attached to the ring is a carboxylic acid group (CH_2OH).
- A.
- (b)  A cyclohexane ring with a hydroxymethyl group (CH_2OH) attached to one of the ring carbons.
- B.



Answer: C

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level 30



Which of the following is best set of reagents to performs to the above conversion?

A. $LiAlH_4$

B. $NaBH_4$

C. $K_2Cr_2O_7$

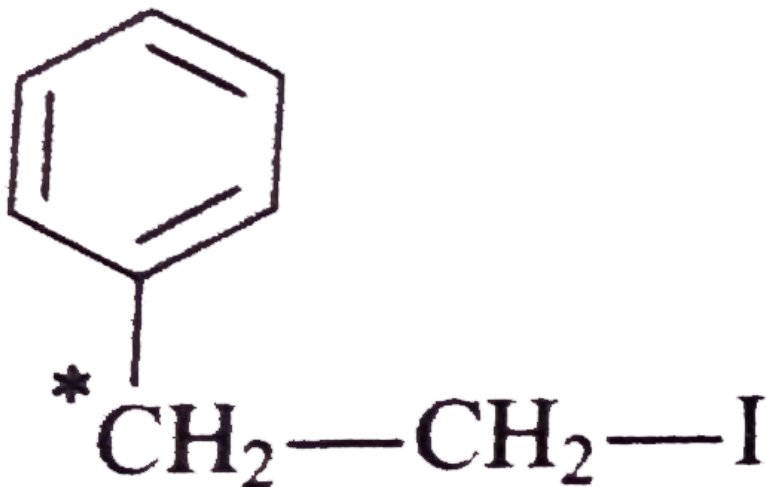
D. None of these

Answer: B

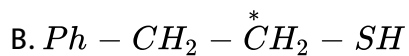
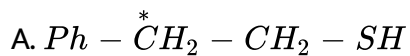


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level 31



Product of the reaction is:



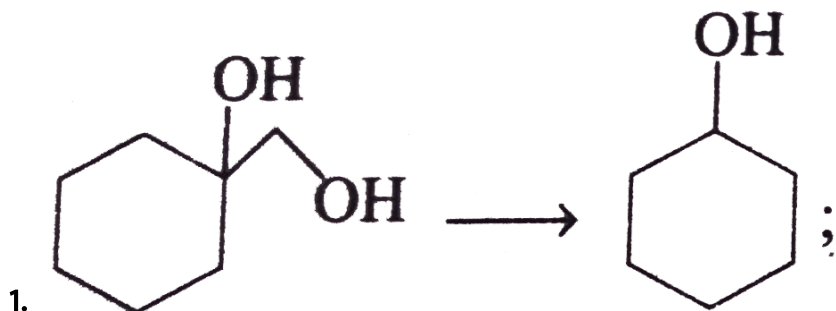
C. Both of these

D. None of these

Answer: C



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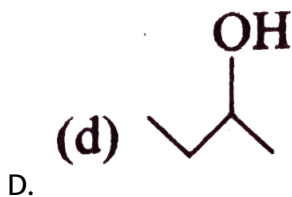
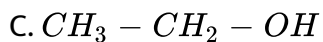
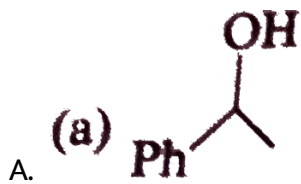
This conversion can be carried out by:

- A. $H_2SO_4 / \Delta, HIO_4$
- B. $NaIO_4, H^{\oplus} / \Delta$
- C. $HIO_4, NaBH_4$
- D. $H^{\oplus} / De < s, Zn(Hg - HCl)$

Answer: C

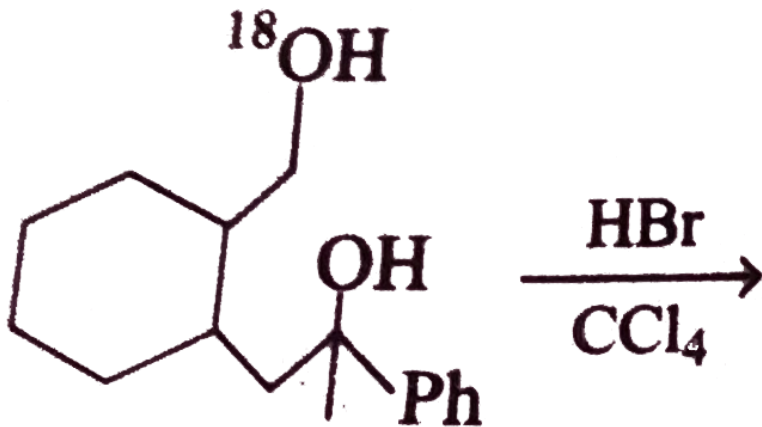
 Watch Video Solution

1. Which of the following alcohols will show positive iodoforms test?

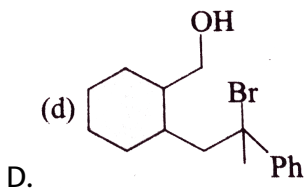
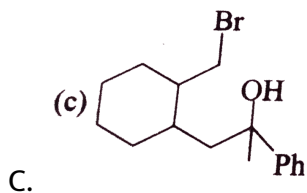
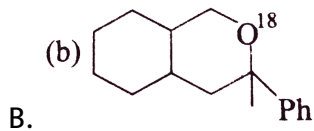
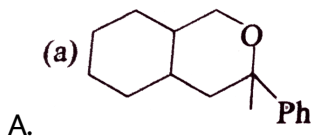


Answer: C

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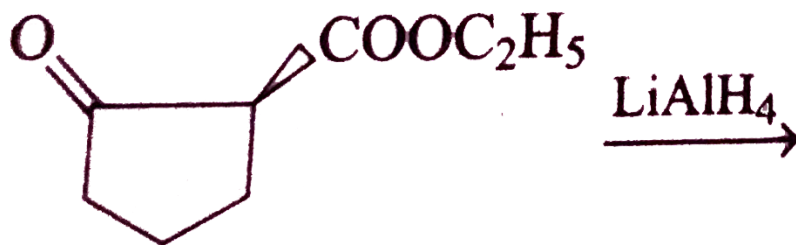
Major product obtained in this reaction is:



Answer: B

 Watch Video Solution

level 35



1. _____, Products of the reaction is:

- A. racemic
- B. diastereomers
- C. meso
- D. optically pure

Answer: B



Watch Video Solution

level 36

1. Reduction of $R - CH_2OH - RCH_3$ can be carried out by:

A. $LiAlH_4$

B. $H_2 - Ni$

C. $RedP + HI$

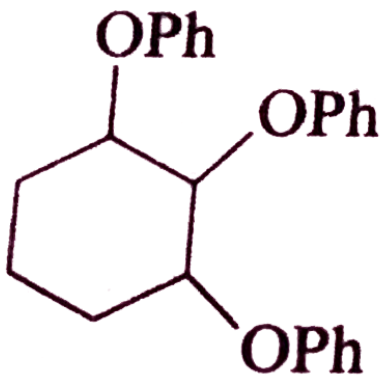
D. $NaBH_4 / AlCl_3$

Answer: C



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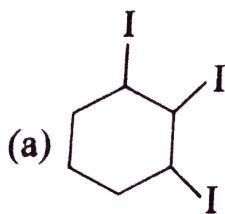
level 37



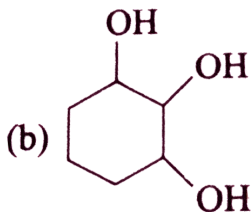
1.

, which of

the following is major product?



A.



B.



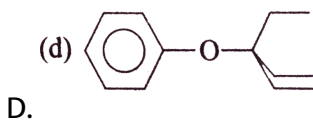
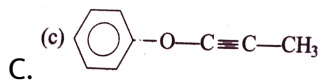
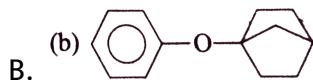
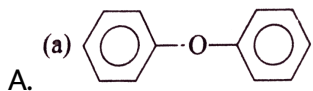
C.

D. None of these

Answer: A

level 38

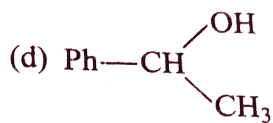
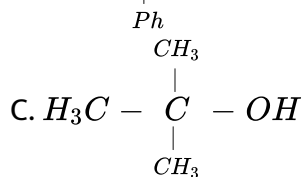
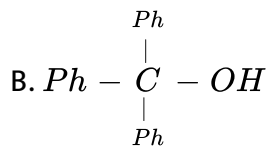
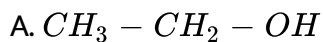
1. Which of the following ethers will get hydrolysed by H^{\oplus} / H_2O ?



Answer: D

level 39

1. Which of the following alcohols will not react with Cu / Δ



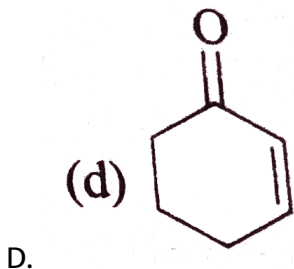
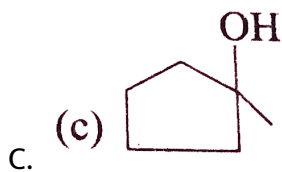
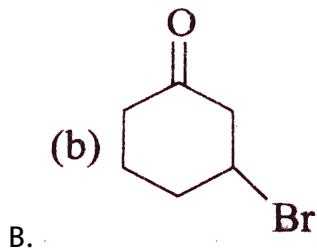
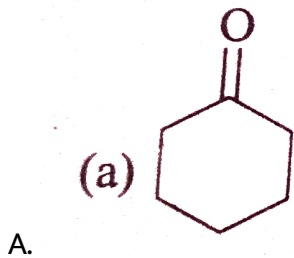
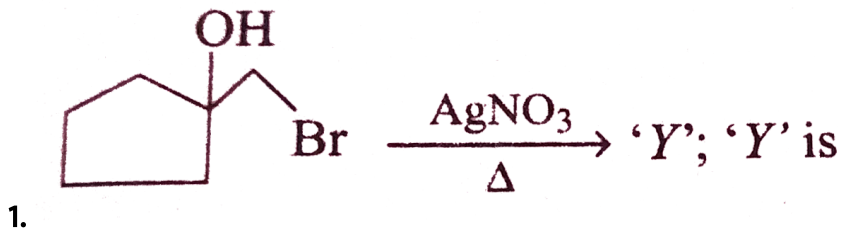
D.

Answer: B



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level 40

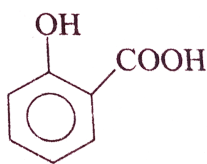
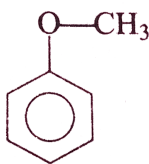
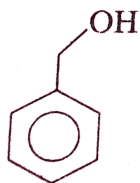


Answer: A

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level 41

1. Which of the following can give purple colour with neutral $FeCl_3$



A. ii and iv

B. I and iii

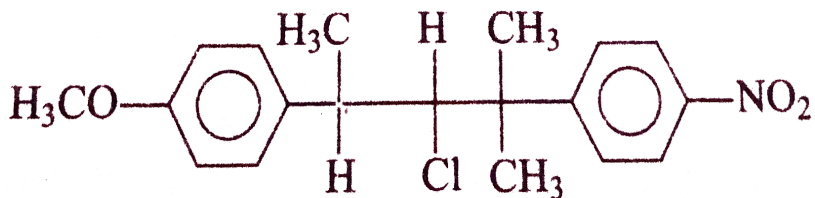
C. ii and iii

D. iii and iv

Answer: A

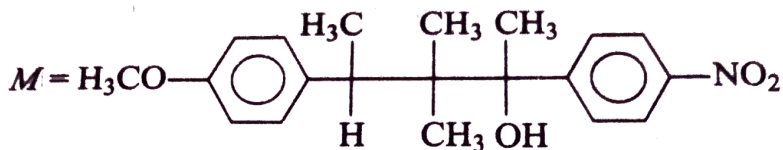
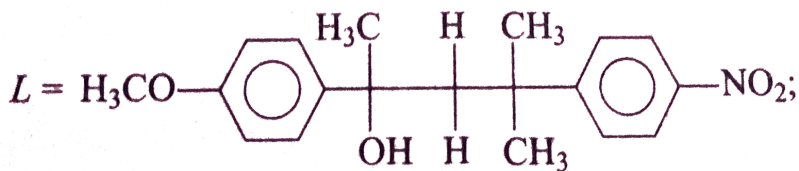
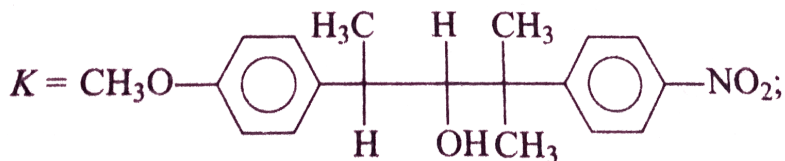
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1. The hydrolysed of



in aqueous

acetone gives.



A. K and L

B. Only K

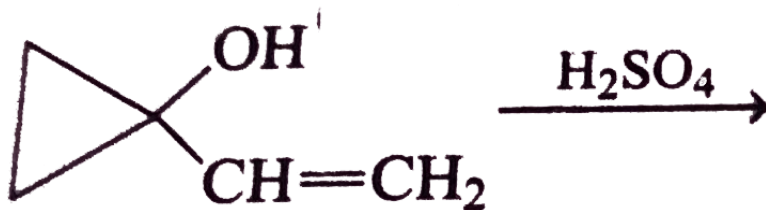
C. L and M

D. Only M

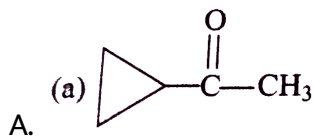
Answer: A

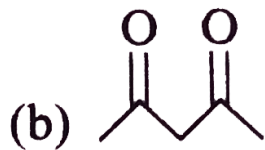
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level 43

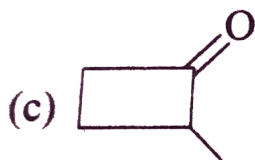


identify 'P' in the reaction:

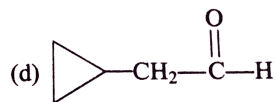




B.



C.

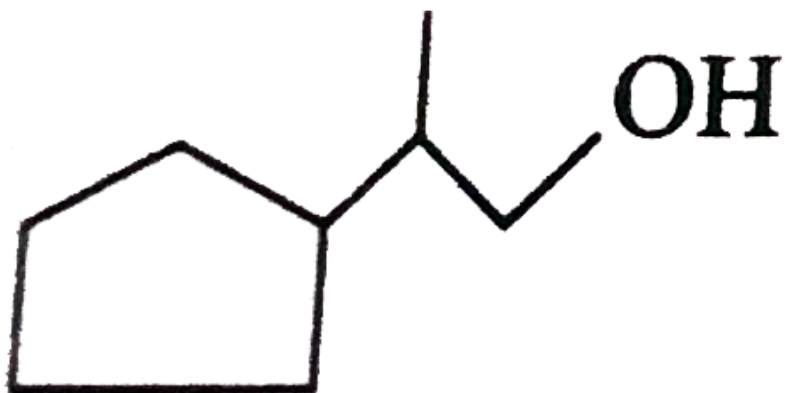


D.

Answer: C

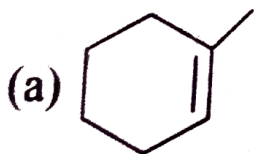
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level 44

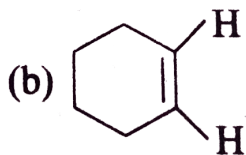


1.

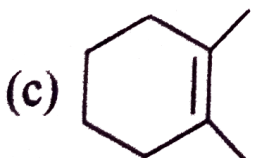
identify 'P' in the reaction:



A.



B.



C.

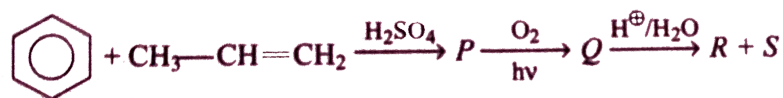


D.

Answer: C

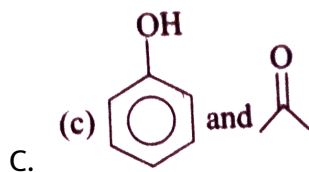
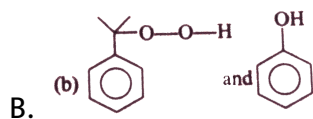
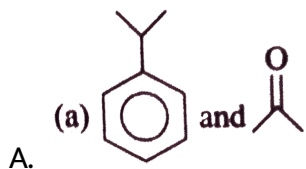
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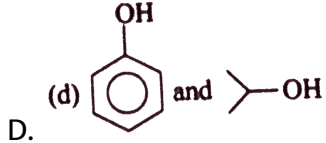
level 45



1.

identify 'P' in the reaction:

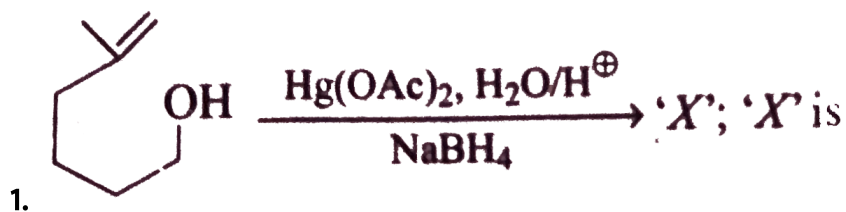




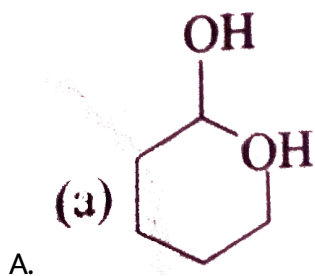
Answer: C

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level 46

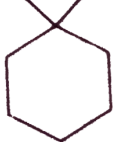


identify 'P' in the reaction:



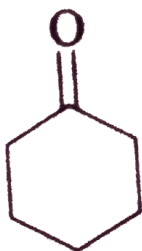
HO OH

(b)



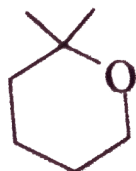
B.

(c)



C.

(d)

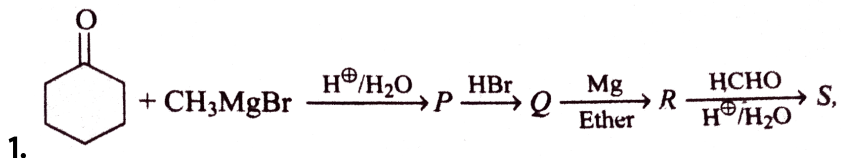


D.

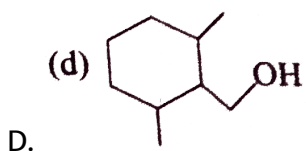
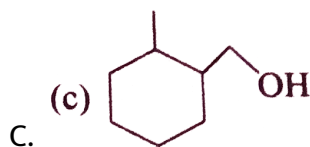
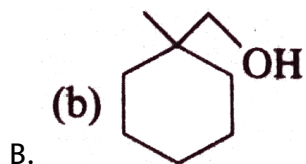
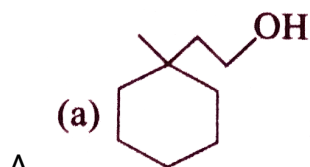
Answer: D



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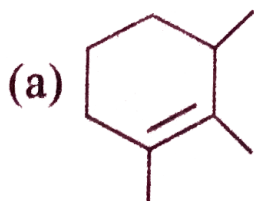
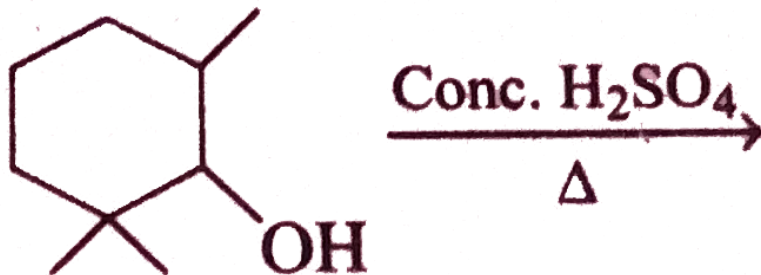
identify 'P' in the reaction:



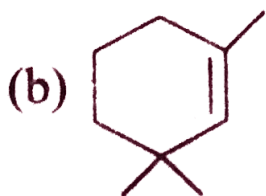
Answer: B

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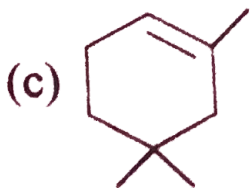
1. Identify the major product of the following reaction:



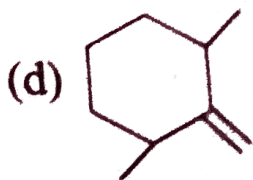
A.



B.



C.



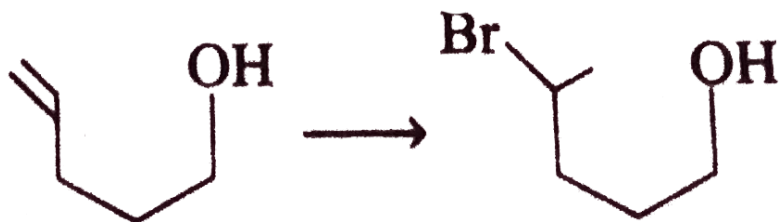
D.

Answer: A

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level 49

1. Find the correct method for the following conversion:



A. H^{\ominus} , HBr

B. *Conc.* H_2SO_4 , Δ

C. H^{\oplus} , HBr

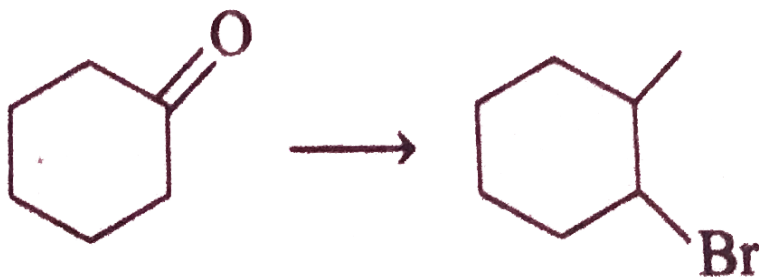
D. None of these

Answer: C

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level 50

1. Which combination of reagents will bring about the following conversion?



A. $MeMgBr / H^{\oplus}$, H_2SO_4 / Δ , HBr / H_2O_2

B. $MeMgBr / H^{\oplus}$, H_2SO_4 / Δ , HBr

C. $MeMgBr / H^{\oplus}$, HBr / CCl_4

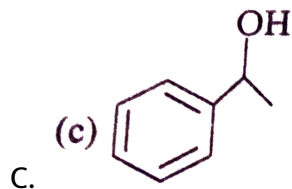
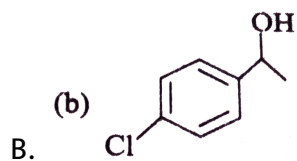
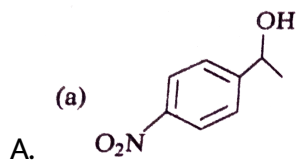
D. HBr / H_2O_2 , $MeMgBr / H^+$

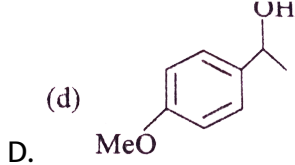
Answer: A

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level 52

1. Which of the following alcohols will undergo easiest dehydration?



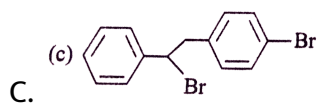
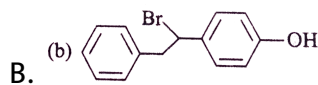
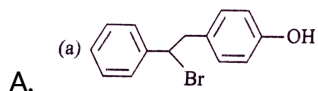
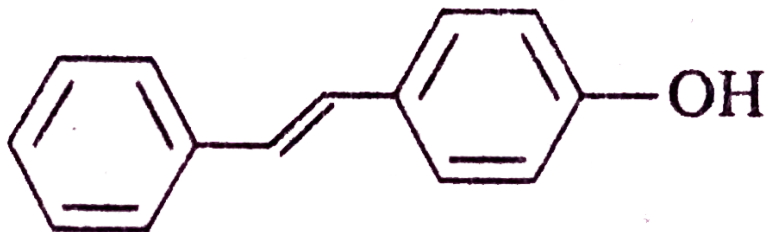


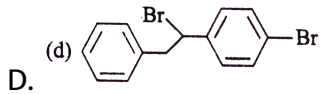
Answer: D

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level 53

1. The reactio of HBr with the followin compound would produce



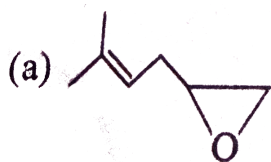
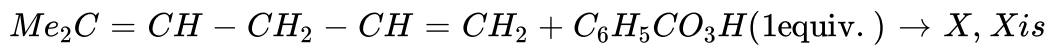


Answer: B

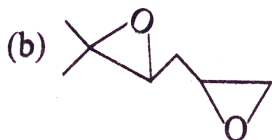
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level 54

1. In the following



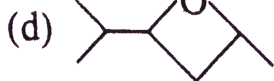
A.



B.



C.



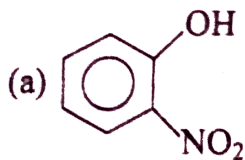
D.

Answer: C

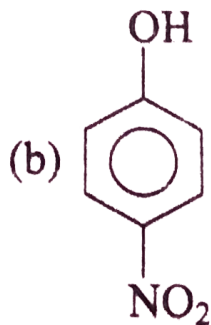
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level 55

1. The most steam volatile species is:

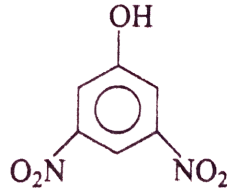


A.



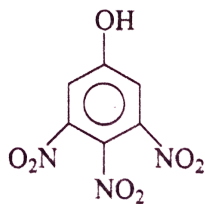
B.

(c)



C.

(d)



D.

Answer: A



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level 56

1. In the Libermann nitroso reaction, change in the colour of phenol occur

as:

A. Brown or red green red deep blue

B. Red deep blue green

C. Red brown white

D. White red green

Answer: B

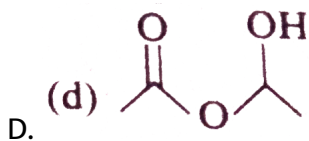
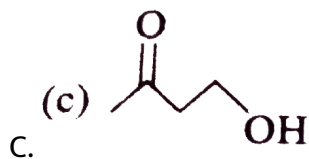
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level 57

1. Which one of the following compounds will be most readily dehydrated?

A. 2-butanol

B. 1-phenyl-1-propanol

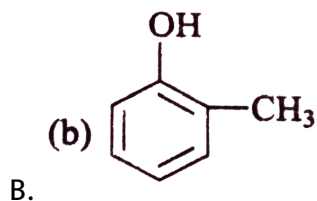
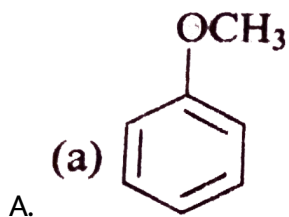


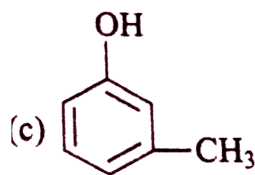
Answer: B

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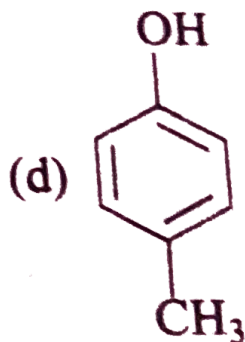
level 58

1. Compound A , C_7H_8O , is insoluble in water, dilute HCl , and aqueous $NaHCO_3$, it dissolves in dilute $NaOH$. When A is treated with bromine water it is converted rapidly into a compound of formula $C_7H_5OBr_3$. The structure of A is





C.



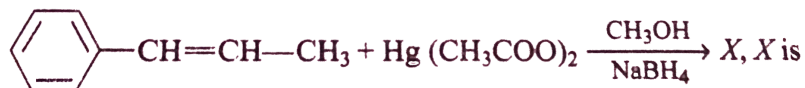
D.

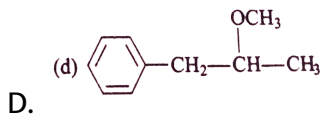
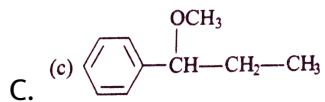
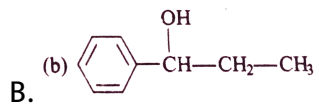
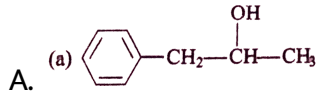
Answer: C

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level 59

1. In the following sequence of reaction

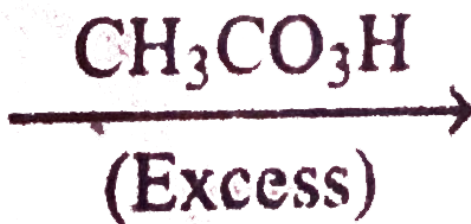




Answer: C

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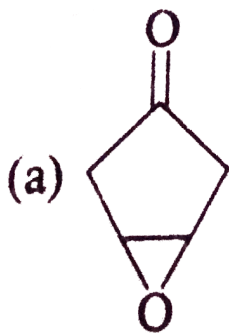
level 60



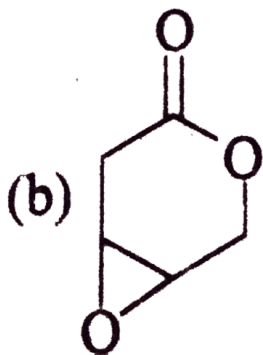
1.

, Product of

the reaction is:



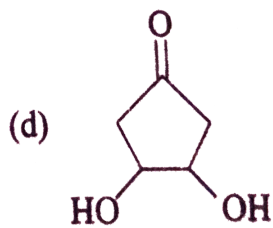
A.



B.



C.

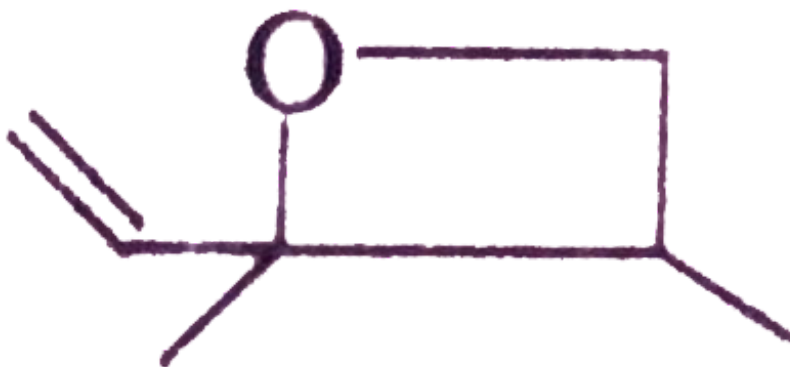


D.

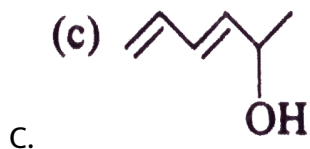
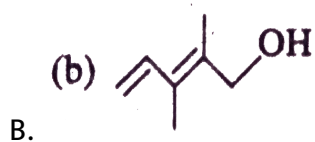
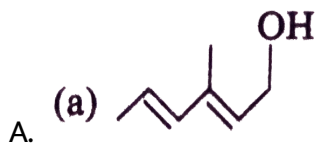
Answer: B

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level 61



1. When _____ is treated with proton acid, a resonance stabilized cation is produced. Which diene listed below when treated with acid will give the same carbocation?





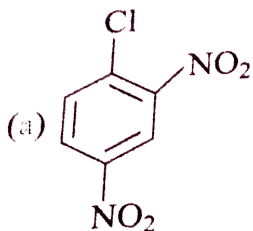
D.

Answer: B

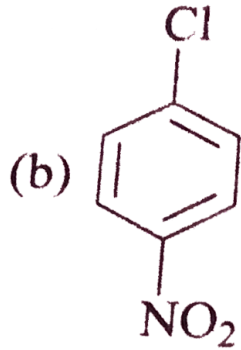
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level 62

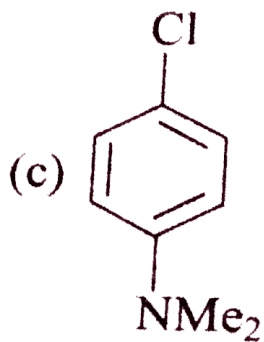
1. Which of the following would undergo most rapid hydrolysis with aqueous to furnish the corresponding hydroxy derivatives?



A.



B.



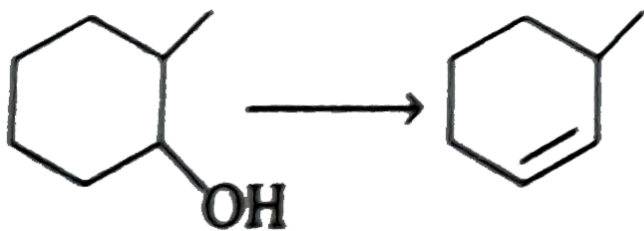
C.

D. 

Answer: A

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Others



1.

Which of the following is best set of reagents to performs to the above conversion?

A. ThO_2, Δ

B. H_3PO_4, Δ

C. *Conc.* H_2SO_4, Δ

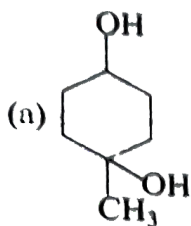
D. Al_2O_3, Δ

Answer: A

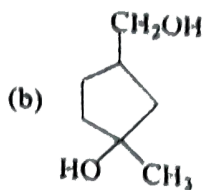


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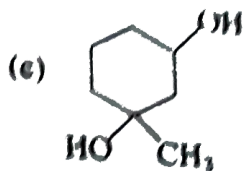
2. Complete the following reaction



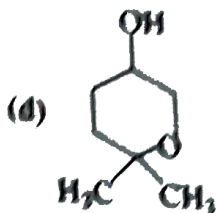
A.



B.



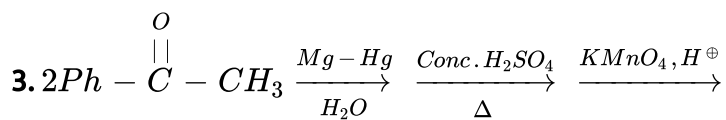
C.



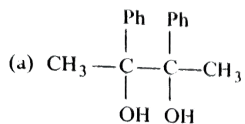
D.

Answer: B

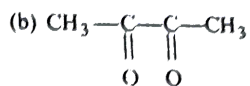
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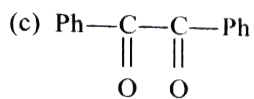
The final product is



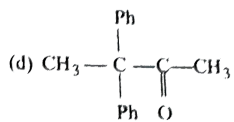
A.



B.



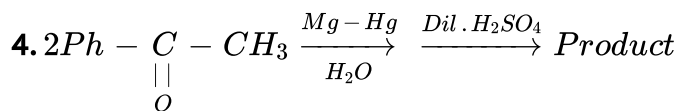
C.



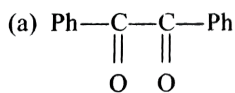
D.

Answer: C

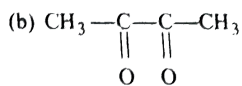
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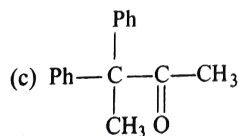
The main product is



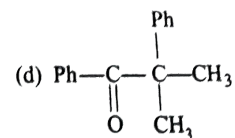
A.



B.



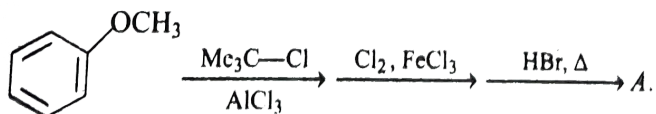
C.



D.

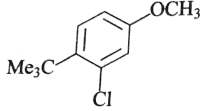
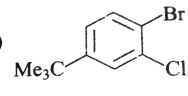
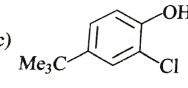
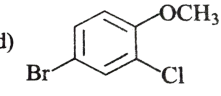
Answer: C

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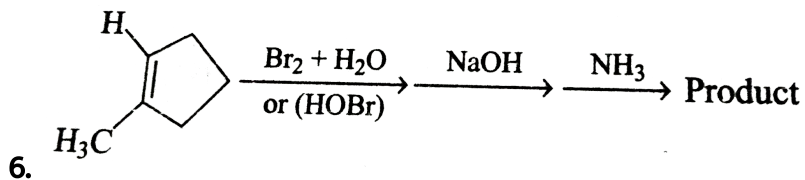
5.

The final product A is:

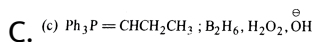
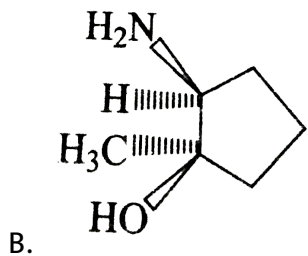
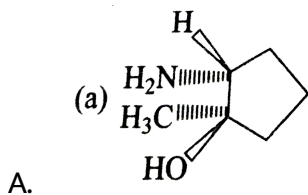
- (a) 
A.
- (b) 
B.
- (c) 
C.
- (d) 
D.

Answer: C

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The product is

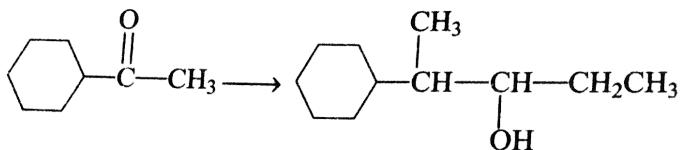


Answer: A



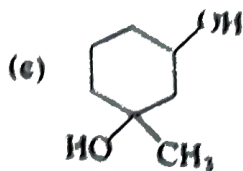
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7. Which of the following sets of reagents would accomplish the following conversation

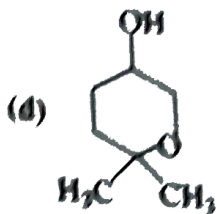


A. (a) $\text{CH}_3\text{CH}_2\text{CH}_2\text{MgBr}; \text{H}^+ / \text{H}_2\text{O}, \text{PCC}, \text{CH}_2\text{Cl}_2$

B. (b) $\text{CH}_3\text{CH}_2\text{CH}_2\text{MgBr}; \text{H}^+ / \text{H}_2\text{O}; \text{H}_2\text{SO}_4, \Delta, \text{PCC}, \text{CH}_2\text{Cl}_2$



C.

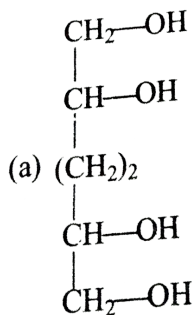


D.

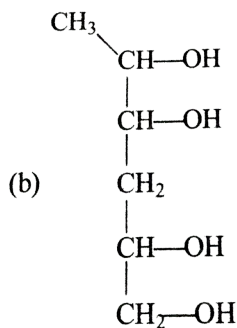
Answer: C

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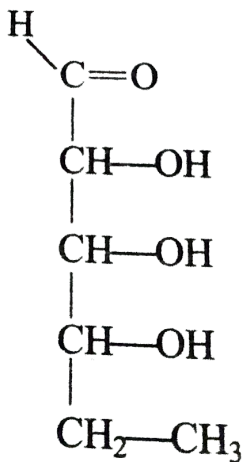
8. An organic compound A (Molecular formula $C_6H_{12}O_4$) on treatment with Na metal liberates H_2 gas and on treatment with HIO_4 gives 2 moles of CH_3CHO , $HCOOH$ (1mole) and CO_2 (1mole). Find the structure on A.



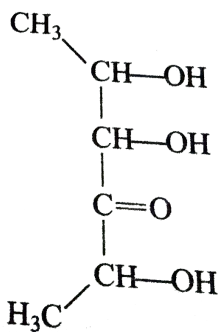
A.



B.



c.



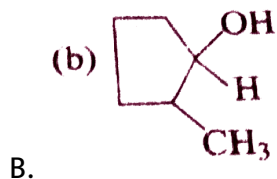
d.

Answer: D

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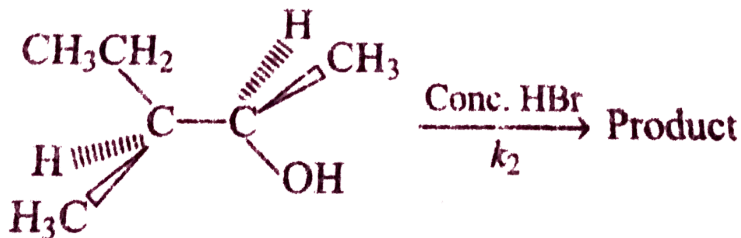
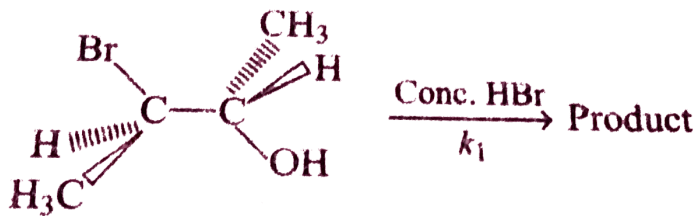
9. An organic compound A (Molecular formula $C_6H_{12}O_4$) does not change the colour of acidic dichromate solution. Compound A on treatment with

H_2SO_4 produces alkene, which on oxidative ozonolysis gives a molecule ($C_6H_{10}O_3$) which gives positive iodoform test. Find the structure of 'A'.



Answer: D

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10.

-OH group is substituted by -Br. The slowest step is dehydration. Which of the following is correct comparison of rate constant k_1 and k_2 ?

A. $k_1 = k_2$

B. $k_1 > k_2$

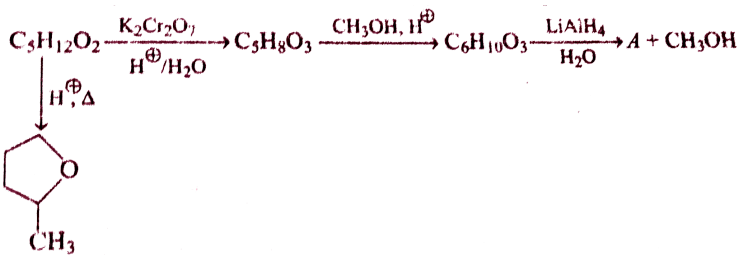
C. $k_1 < k_2$

D. cannot be predict

Answer: C

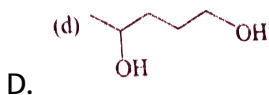
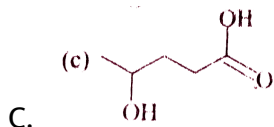
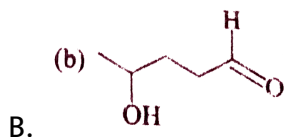


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11.

The molecule A in the sequence reaction is

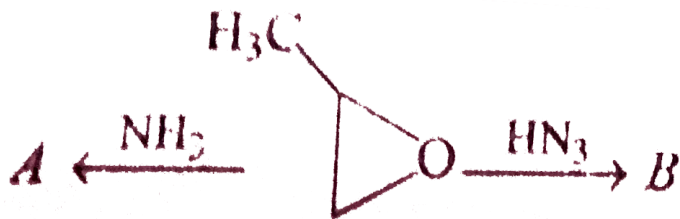


Answer: D

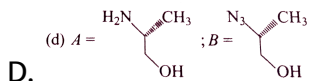
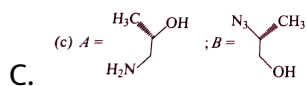
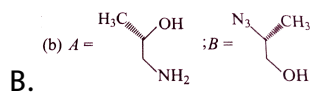
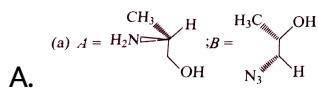


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12.

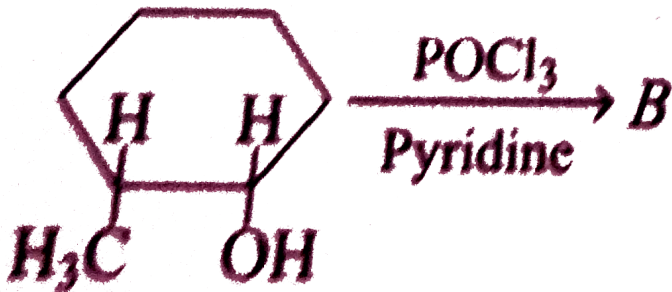
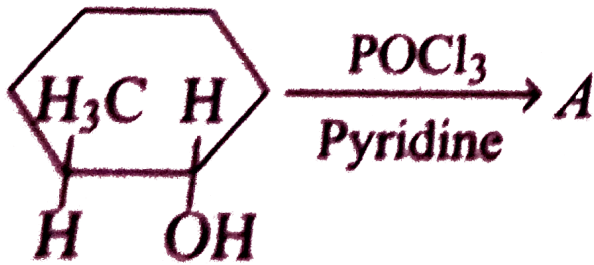


The product A and B respectively:



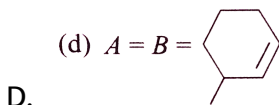
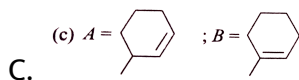
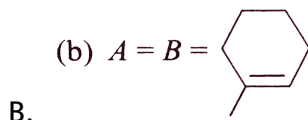
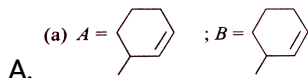
Answer: C

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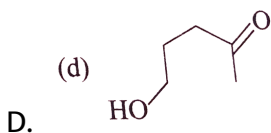
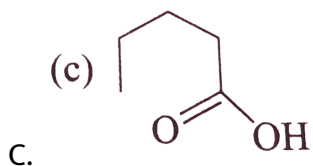
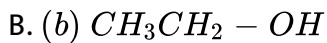
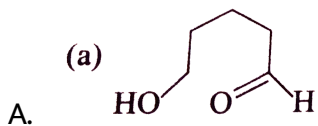
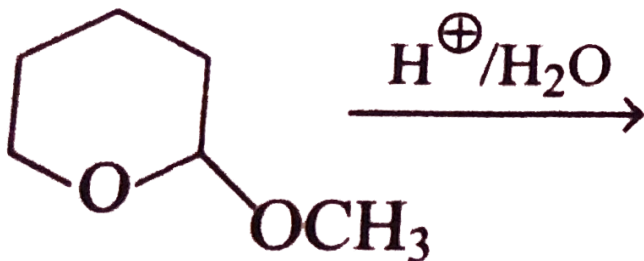
13.

Product A and B respectively :



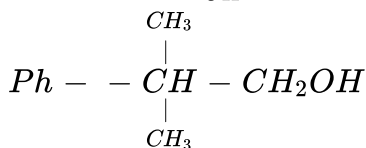
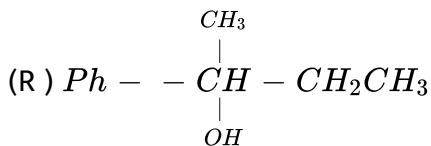
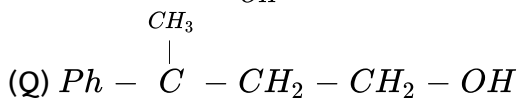
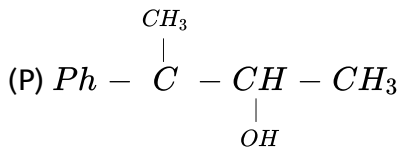
Answer: C

14. Complete the following reaction



Answer: A

15. The relative rate or acid catalysed dehydration of following alcohols would be:



A. $R > P > S > Q$

B. $R > S > P > Q$

C. $P > R > S > Q$

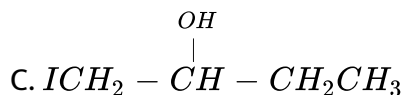
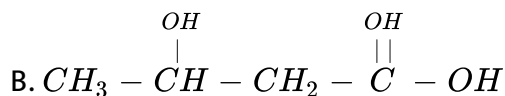
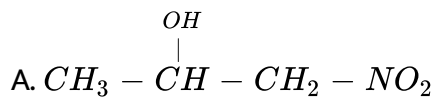
D. $R > S > Q > P$

Answer: A



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16. Which of the following alcohols will show positive iodoforms test?

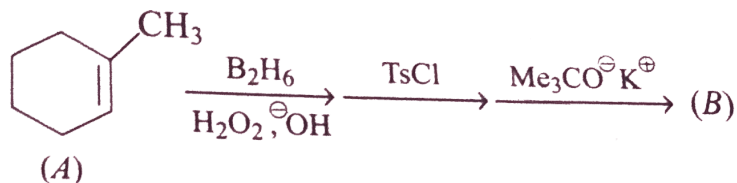


D. None is correct

Answer: C

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17. In the given reaction



The product B is:

A. Identical to B

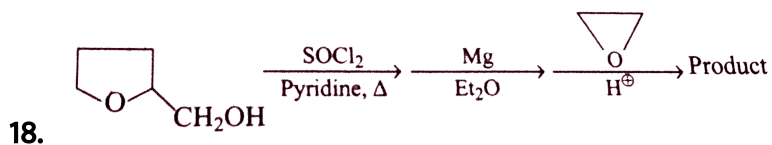
B. Chain isomer of A

C. a positional isomer of 'A'

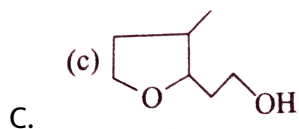
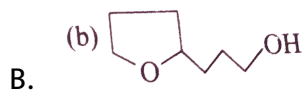
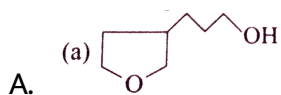
D. reduced product of A

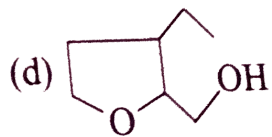
Answer: C

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The final product is :

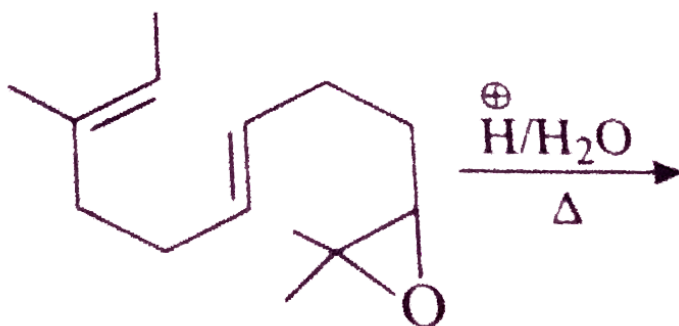




D.

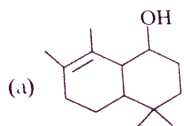
Answer: B

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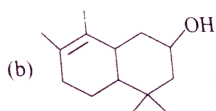


19.

The final product is :

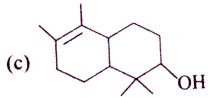


A.

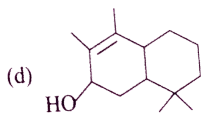


B.

C.

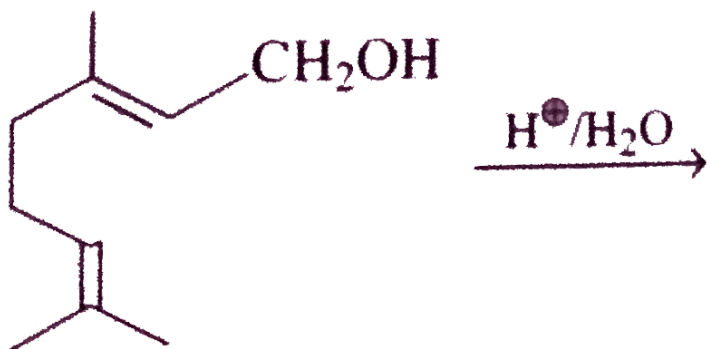


D.



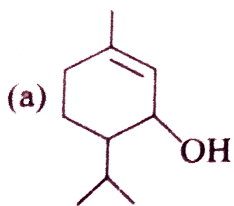
Answer: C

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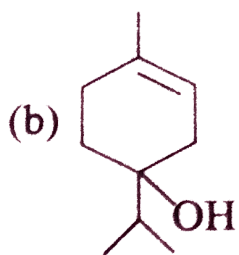


20.

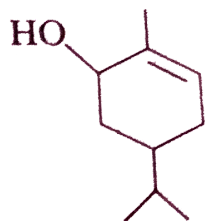
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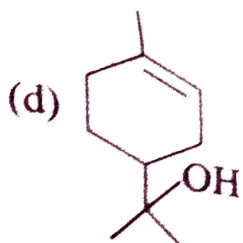
A.



B.



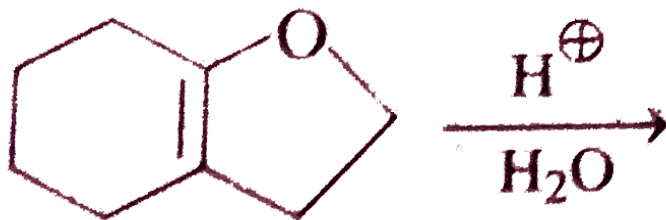
C.



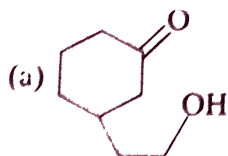
D.

Answer: D

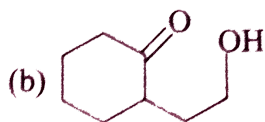
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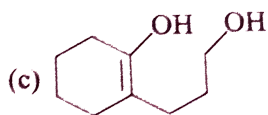
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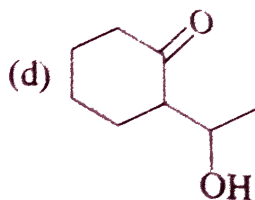
A.



B.



C.



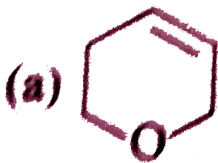
D.

Answer: B

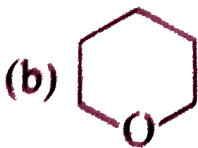


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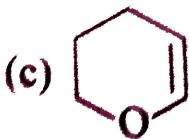
The final product is :



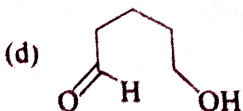
A.



B.



C.



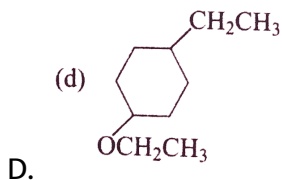
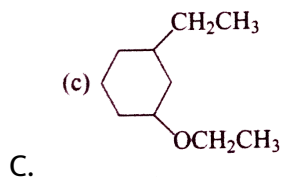
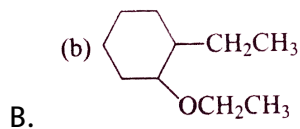
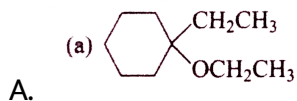
D.

Answer: C

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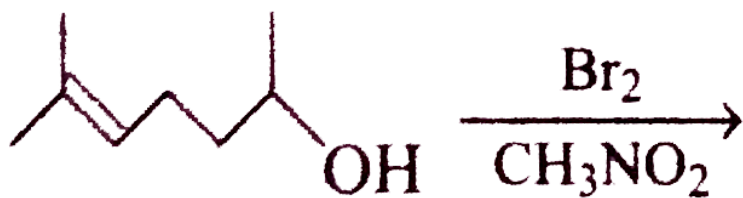


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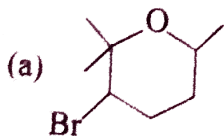
Answer: A

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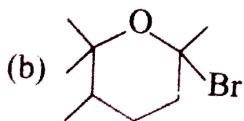


24.

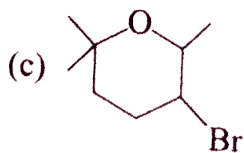
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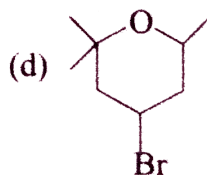
A.



B.



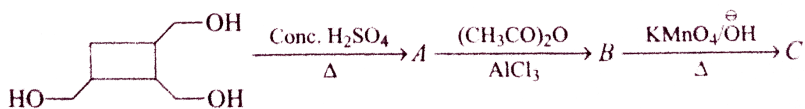
C.



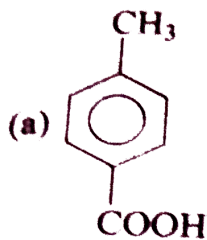
D.

Answer: A

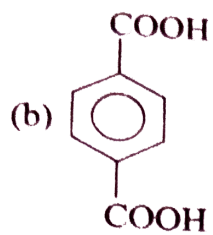
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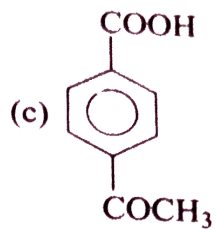
The final product is :



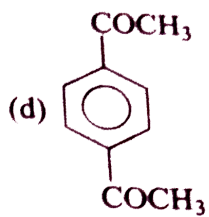
A.



B.



C.

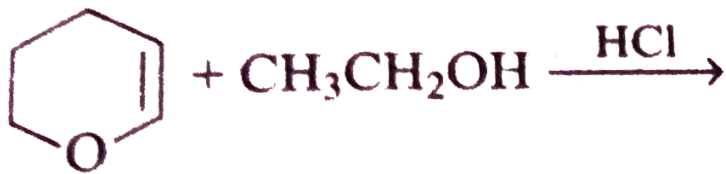


D.

Answer: B

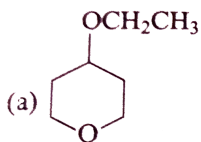


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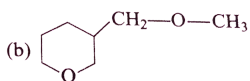


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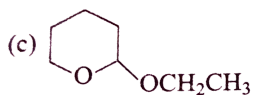
The final product is :



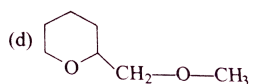
A.



B.



C.



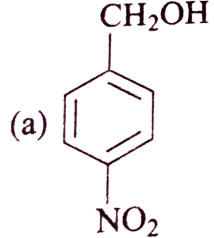
D.

Answer: C

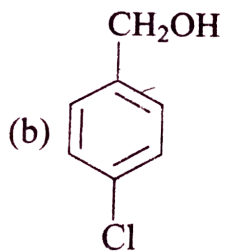


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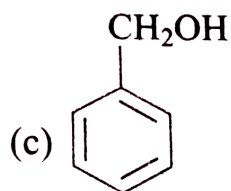
27. Which of the following reacts fastest with HBr?



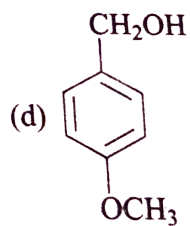
A.



B.



C.



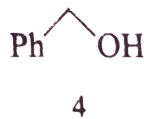
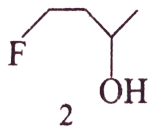
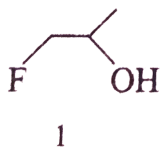
D.

Answer: D



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28. The order of reactivity of the following alcohols towards HCl is :



A. $1 > 2 > 3 > 4$

B. $1 > 3 > 2 > 4$

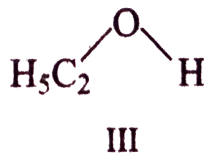
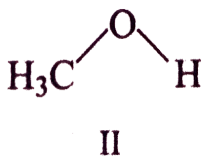
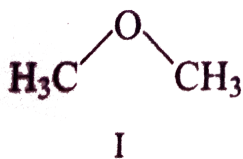
C. $4 > 3 > 2 > 1$

D. $4 > 3 > 1 > 2$

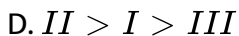
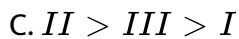
Answer: C

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29. The order of solubility of



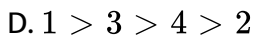
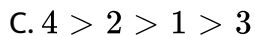
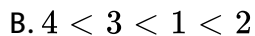
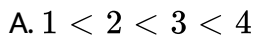
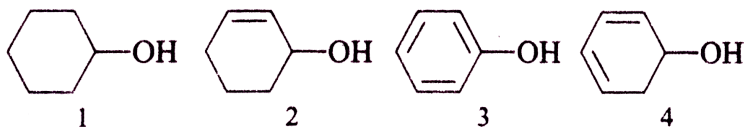
A. $I > II > III$



Answer: C

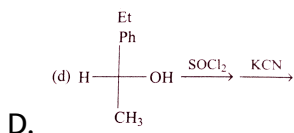
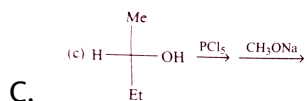
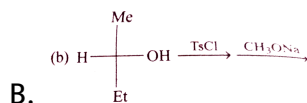
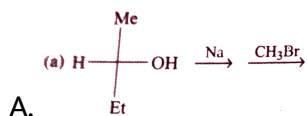
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30. Deydration of the following alcohols will be in order:



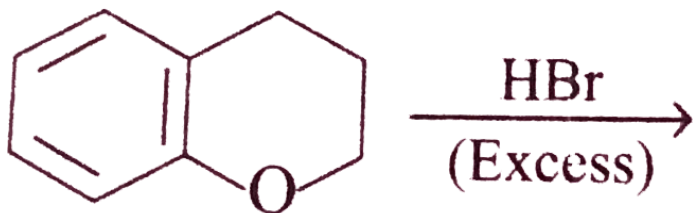
Answer: C

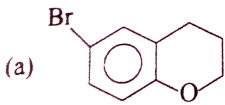
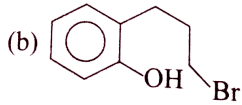
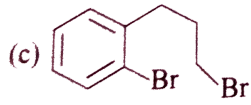
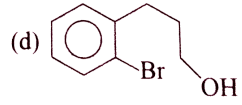
31. Which of the following reactions proceeds with retention of configuration?



Answer: A

32. Find out correct product of reaction:

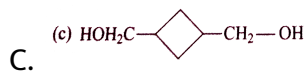
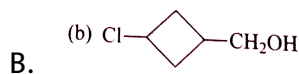
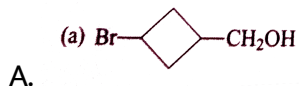
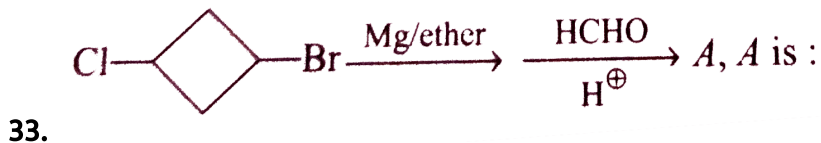


- A. (a) 
- B. (b) 
- C. (c) 
- D. (d) 

Answer: B

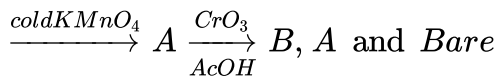
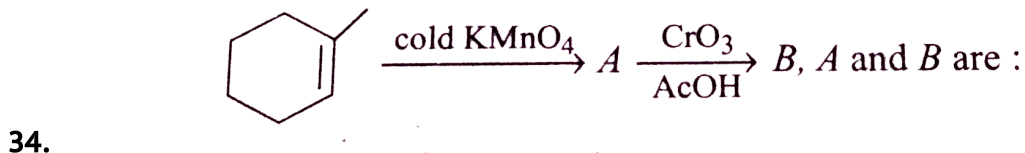


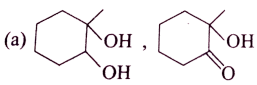
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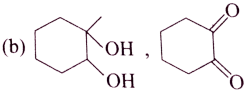
Answer: B

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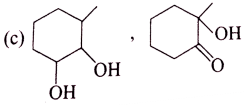




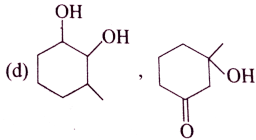
A.



B.



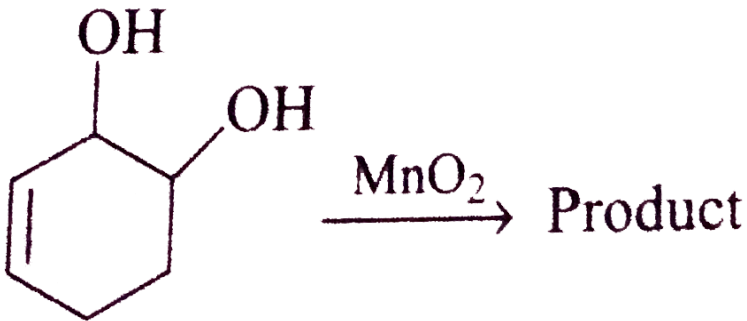
C.



D.

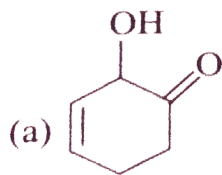
Answer: A

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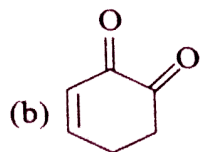


35.

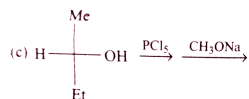
(MnO₂) Product



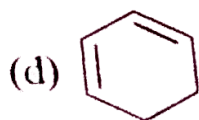
A.



B.



C.

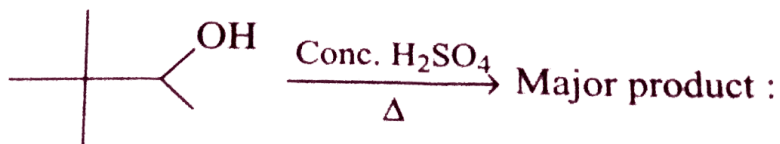


D.

Answer: C

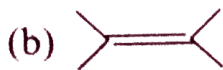
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36. Complete the following reaction





A.



B.

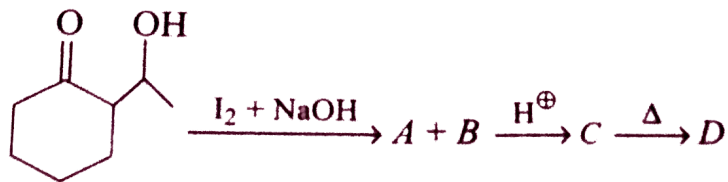


C.

D. None of these

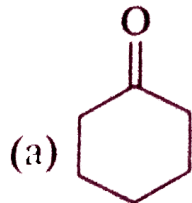
Answer: B

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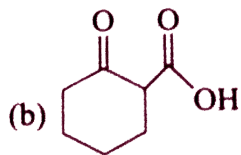


37.

Identify product D in this reaction

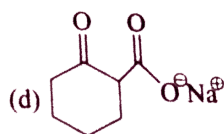


A.



B.

C. CHI_3



D.

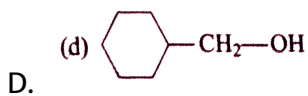
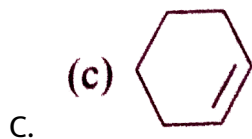
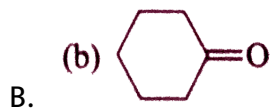
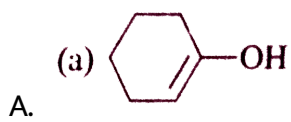
Answer: A

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38. In the given reaction

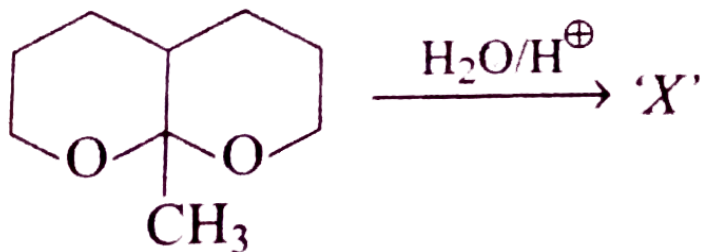


P will be:

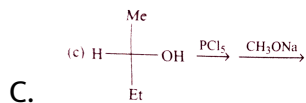
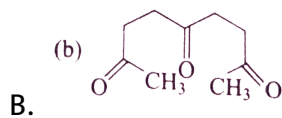
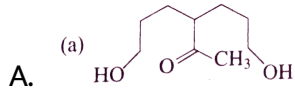


Answer: B

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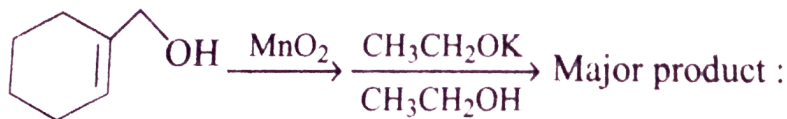
'X' will be



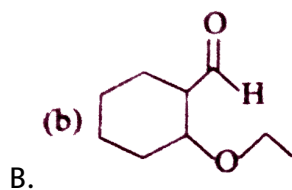
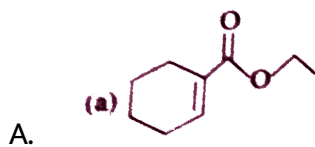
D. All of these

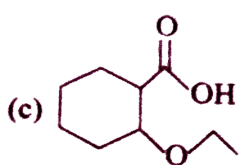
Answer: A

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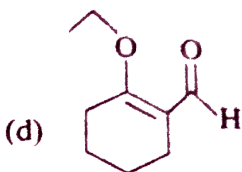


40.





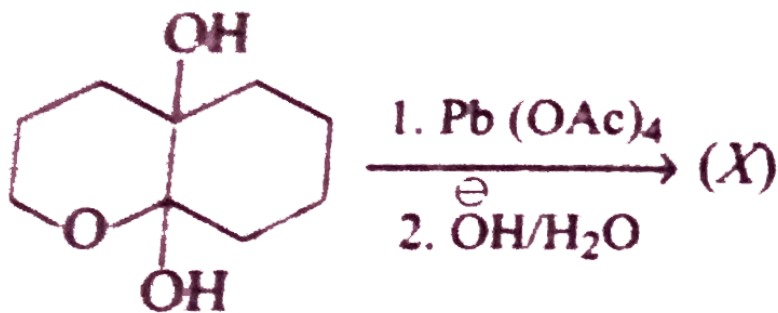
C.



D.

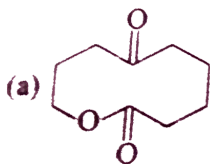
Answer: B

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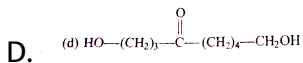
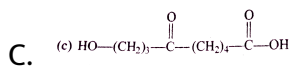
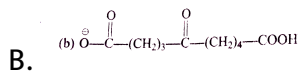


41.

'X' will be

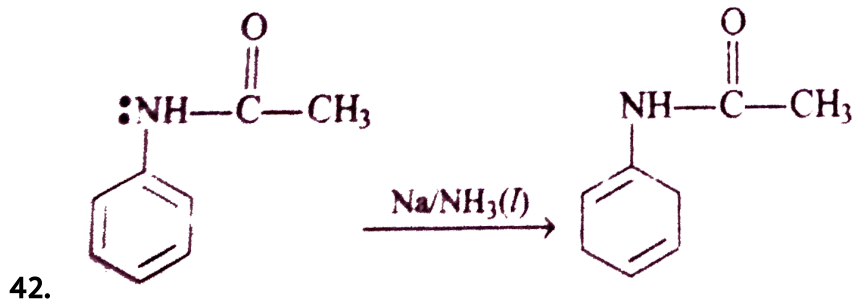


A.



Answer: C

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The above reaction is known as:

A. Clemmensen reduction

B. Birch reduction

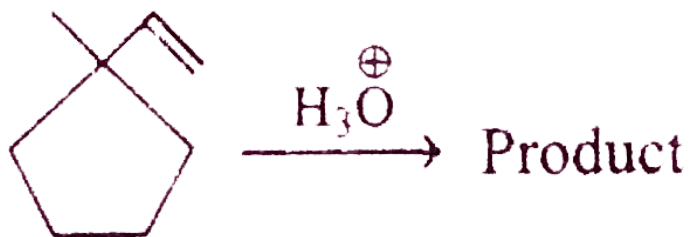
C. MPV reaction

D. Wolff-Kishner reaction

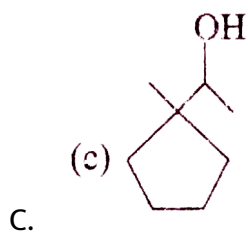
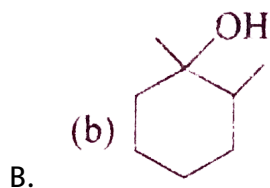
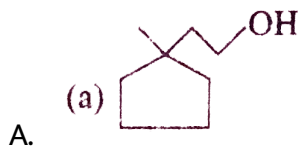
Answer: B

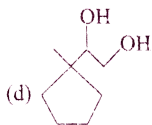
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43. In the following reaction



The main product is:

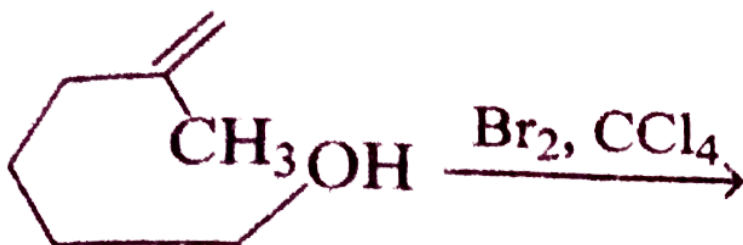




D.

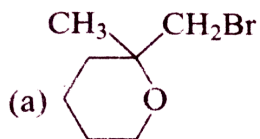
Answer: B

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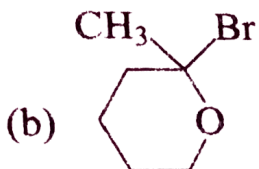


44.

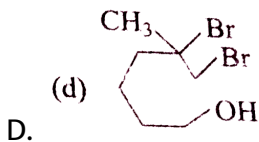
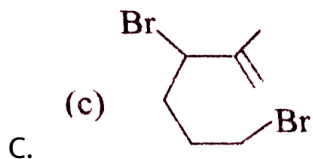
The major product obtained in this reaction is:



A.



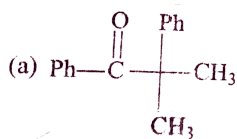
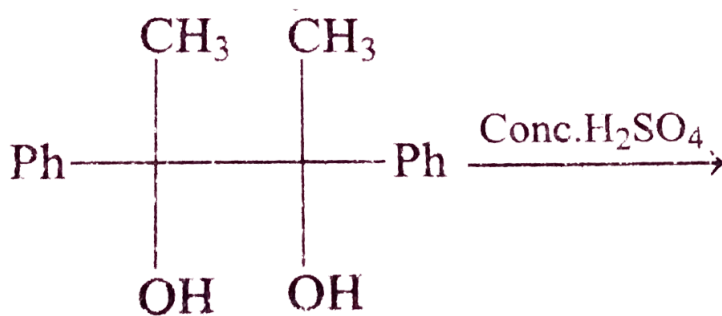
B.



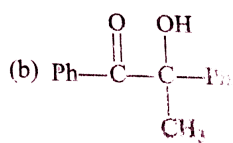
Answer: A

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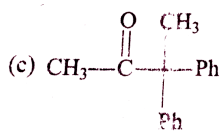
45. Complete the following reaction



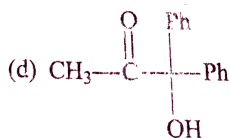
A.



B.



C.

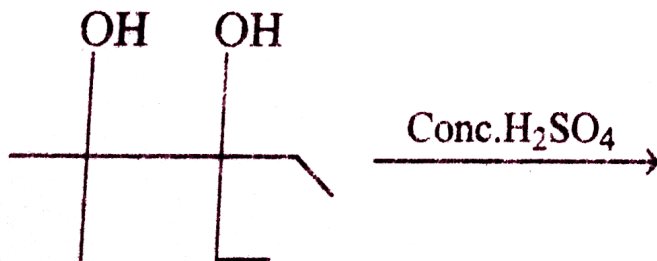


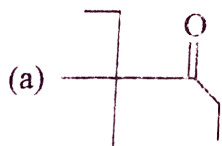
D.

Answer: C

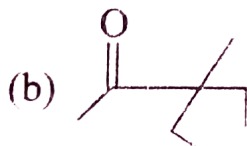
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46. Complete the following reaction

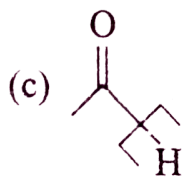




A.



B.



C.

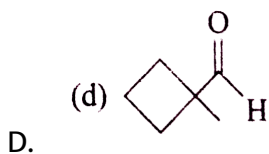
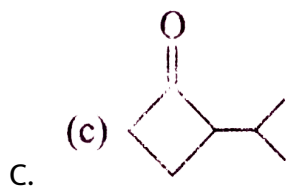
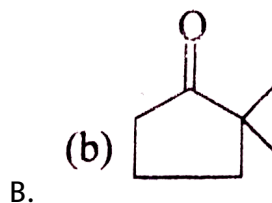
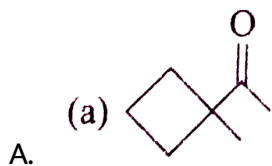
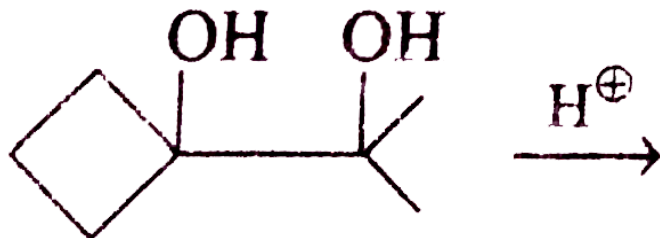
D. None of these

Answer: A



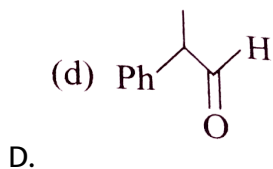
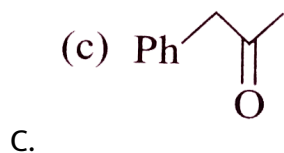
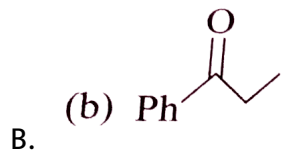
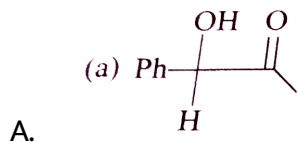
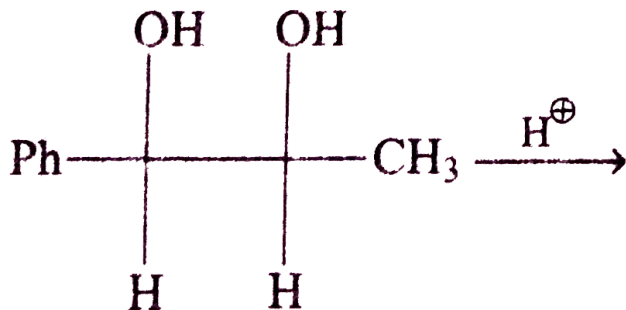
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47. Complete the following reaction



Answer: B

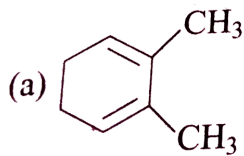
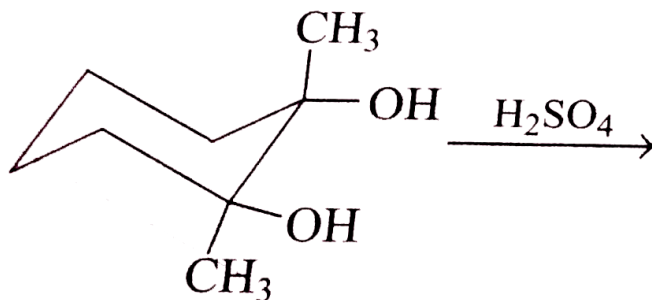
48. Complete the following reaction



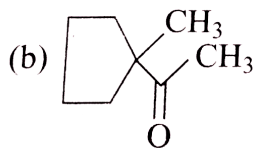
Answer: C

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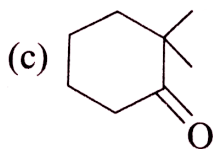
49. Complete the following reaction



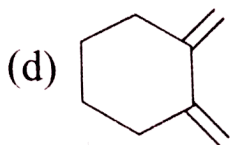
A.



B.



C.

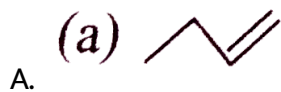


D.

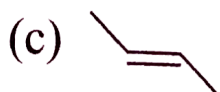
Answer: B

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50. Complete the following reaction



B.



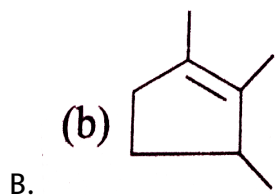
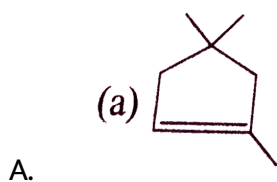
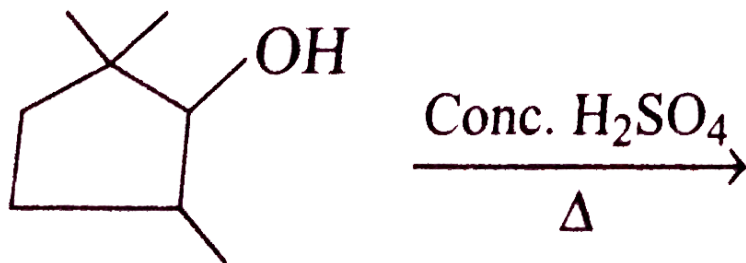
C.

D. None of thses

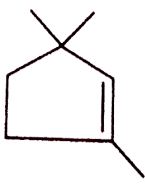
Answer: C

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51. Complete the following reaction

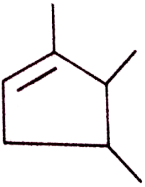


(c)



C.

(d)



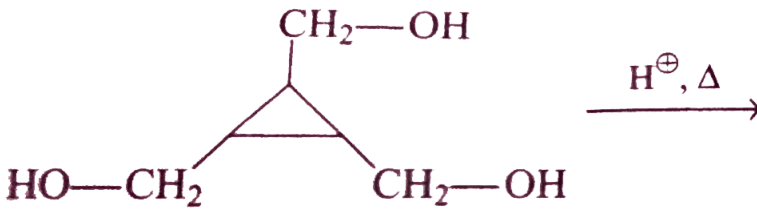
D.

Answer: A



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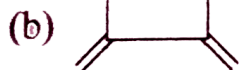
52. Complete the following reaction



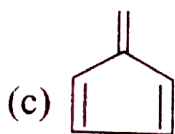
(a)



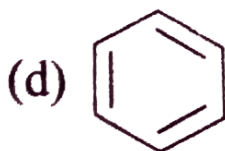
A.



B.



C.

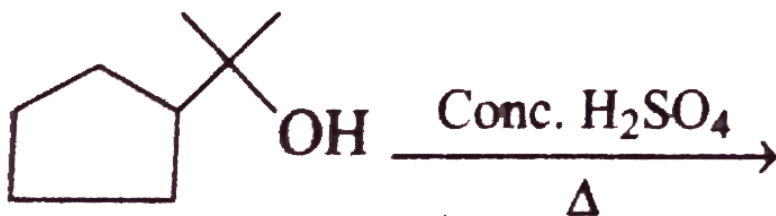


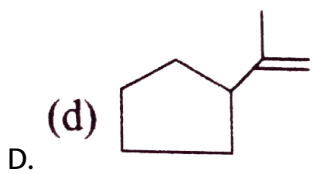
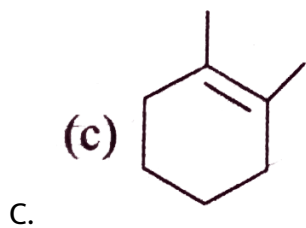
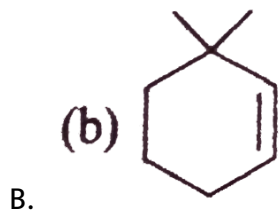
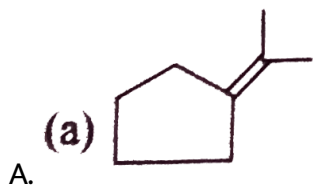
D.

Answer: D

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53. Complete the following reaction

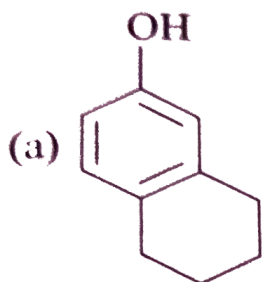
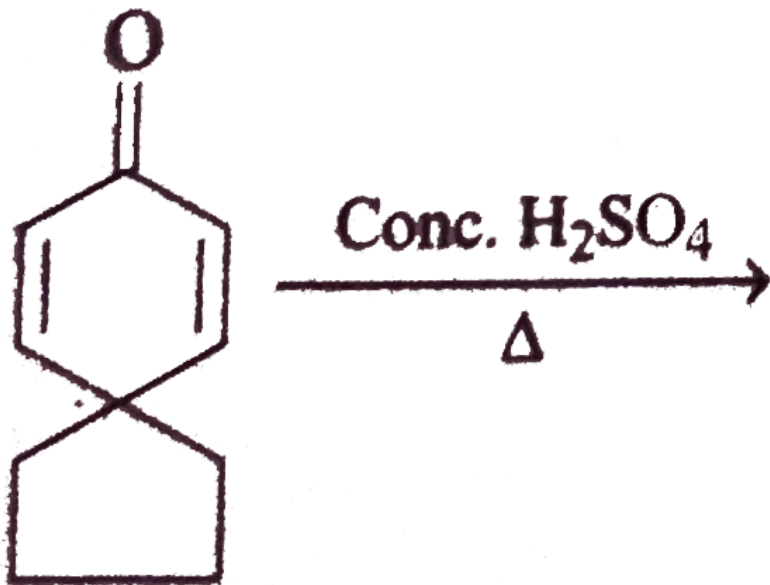




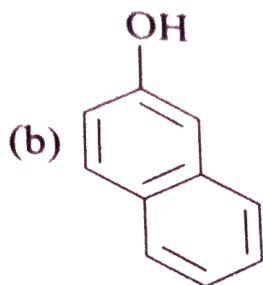
Answer: C

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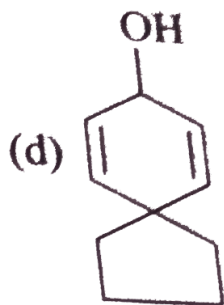
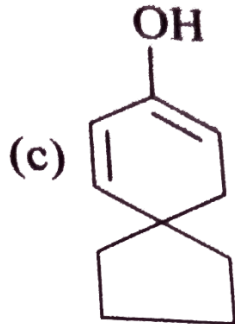
54. Complete the following reaction



A.



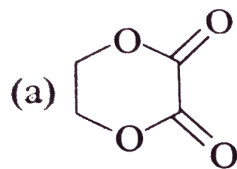
B.

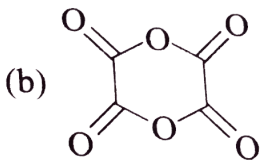


Answer: A

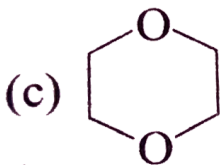
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55. When ethylene glycol is heated with oxalic acid in the presence of conc. H_2SO_4 , the product formed is:

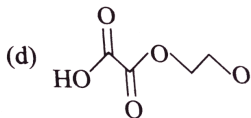




B.



C.

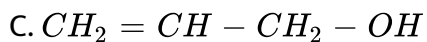
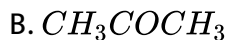


D.

Answer: A

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56. An organic compound having molecular formula C_3H_6O does not react with 2,4-dinitrophenol hydrazine and does not react Na metal. The compound is expected to be:





Answer: D

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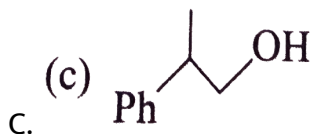
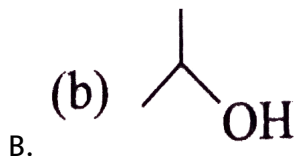
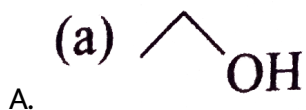
57. Which of the following statements is correct?

- A. Phenol is less acidic than ethanol
- B. Phenol is more acidic than ethanol
- C. Phenol is more acidic than p-nitrophenol
- D. Phenol is more acidic than acetic acid

Answer: B

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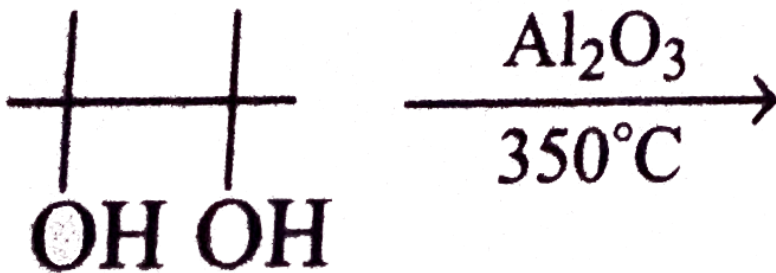
58. The vapour of an alcohol X are passed over Cy heated at $300^{\circ}C$ whereby an alkene is formed as product . The alcohol X is expected to be:



Answer: D

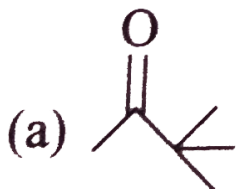


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59.

The major product formed in the reaction is:



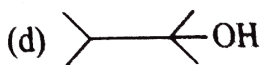
A.



B.



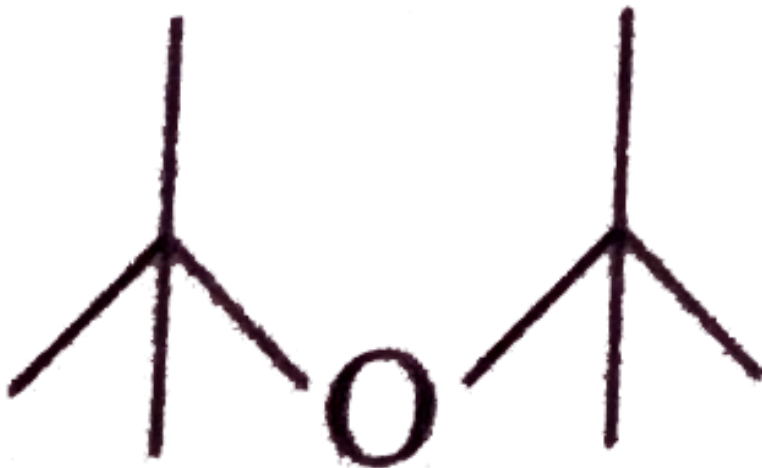
C.



D.

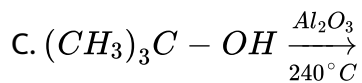
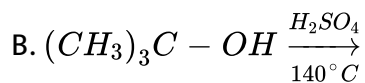
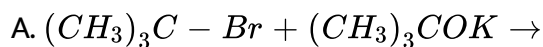
Answer: C

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60.

The major product formed in the reaction is:

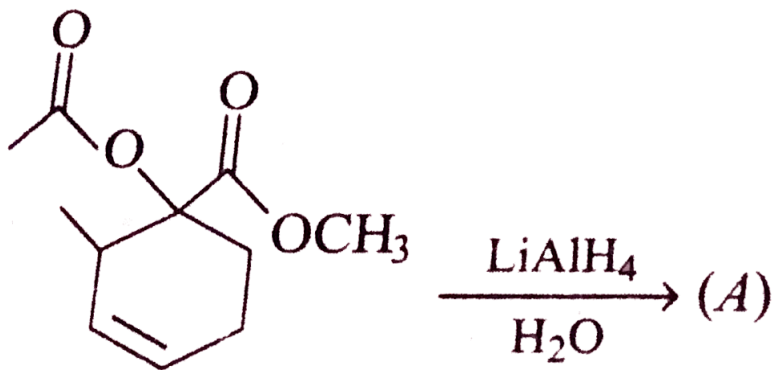


D. oversight

Answer: D

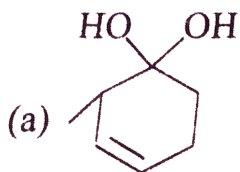


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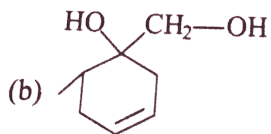


61.

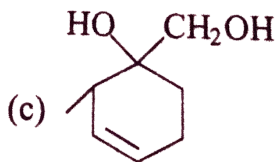
Find out 'A' of the reaction:



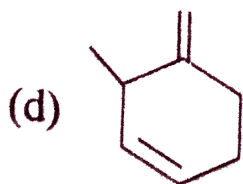
A.



B.



C.

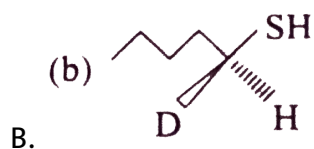
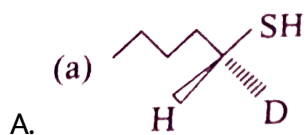
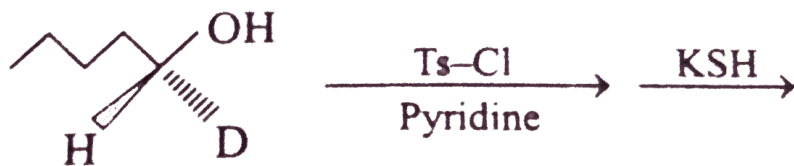


D.

Answer: C

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62. Identify the major product of the following reaction:

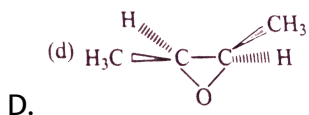
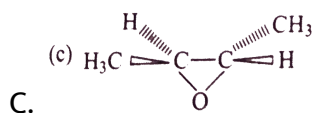
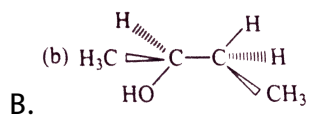
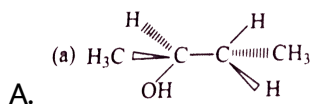
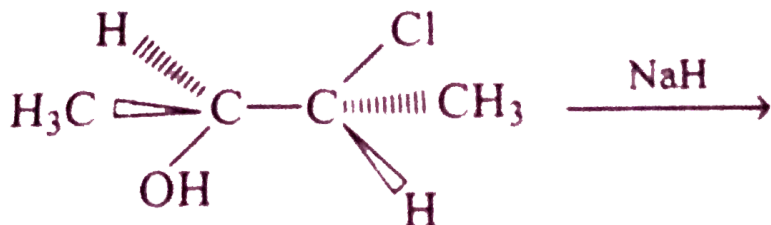


D. No reaction

Answer: B

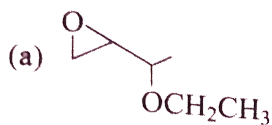
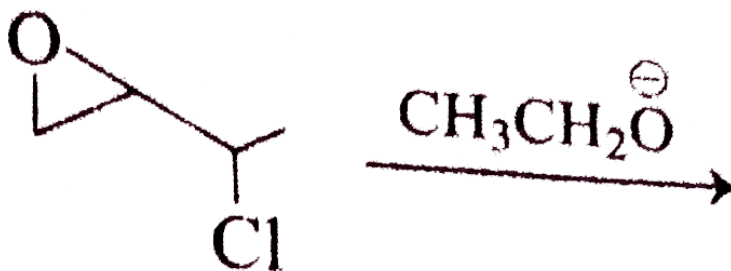


63. Find the product of the following reaction with sterechemistry.



Answer: C

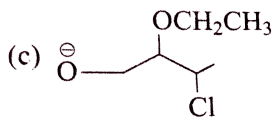
64. Select the major product of the following reaction:



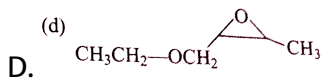
A.



B.



C.



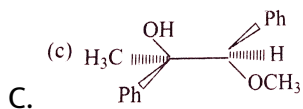
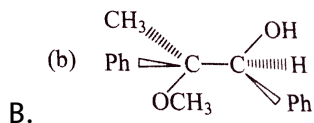
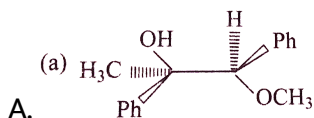
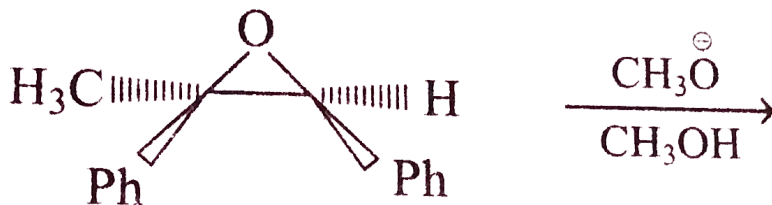
D.

Answer: D



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65. What would be the major product of the following reaction?



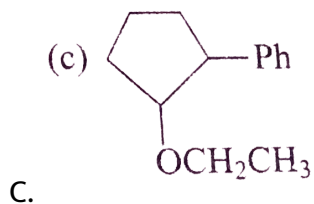
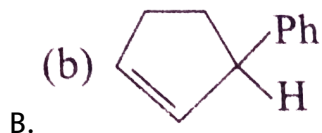
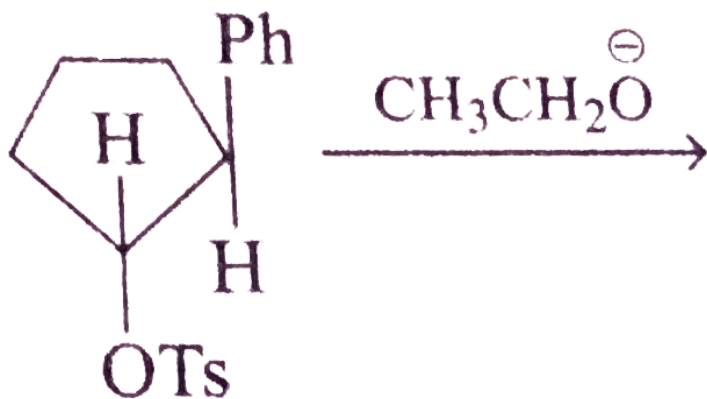
D. None of these

Answer: A



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66. Find out the major product of the following reaction:

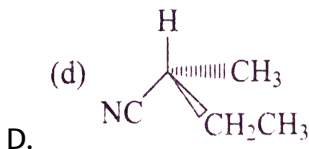
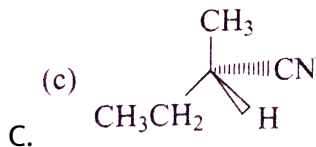
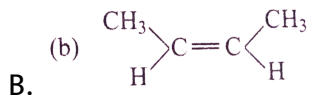
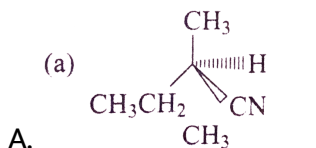
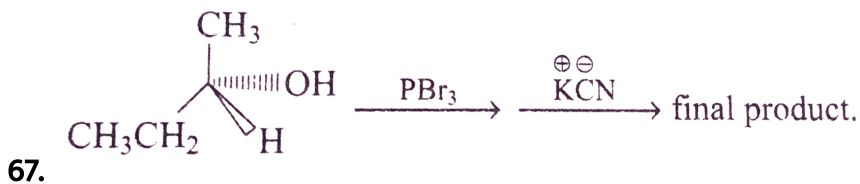


D. No reaction

Answer: B



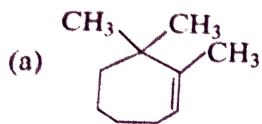
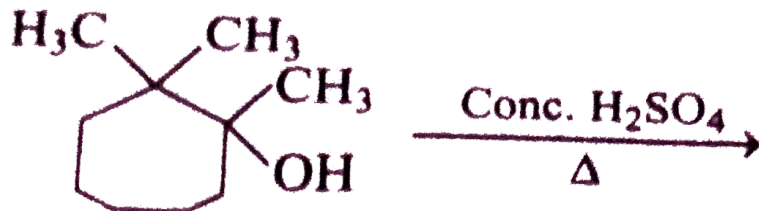
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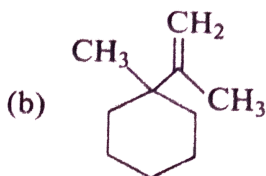
Answer: C

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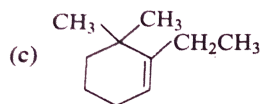
68. What would be the major product of the following reaction?



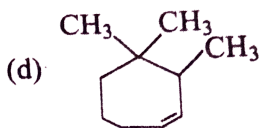
A.



B.



C.

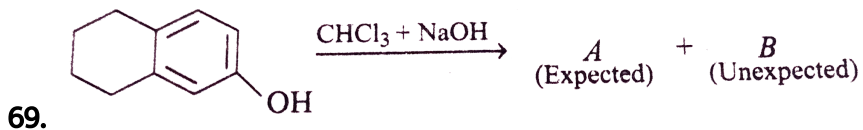


D.

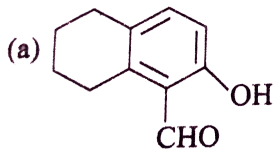
Answer: B



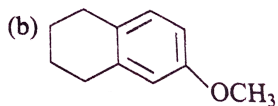
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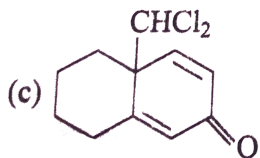
The unexpected product B is:



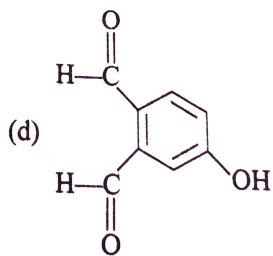
A.



B.



C.

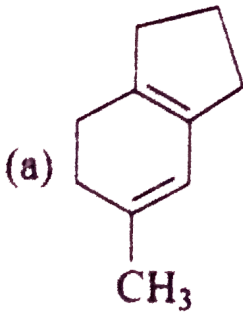
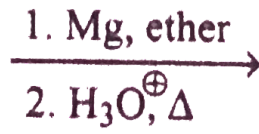
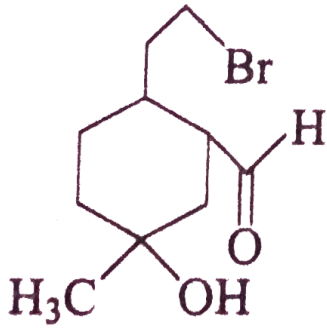


D.

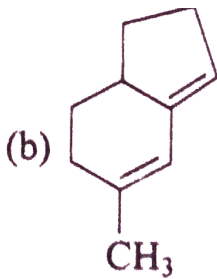
Answer: C

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70. The final product in the following reaction is :



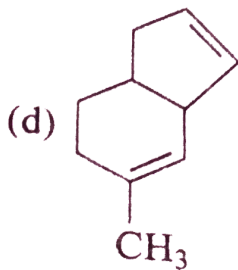
A.



B.



C.

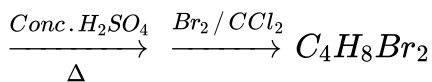
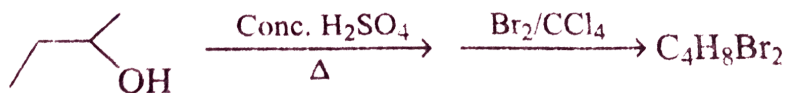


D.

Answer: A

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71. How many structure of final products are possible?



A. 2

B. 5

C. 6

D. 3

Answer: B

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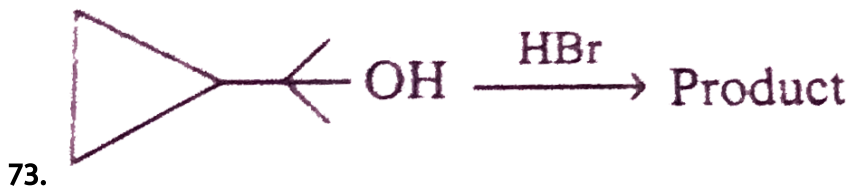
72. Chosse the correct statements regarding the following reaction:



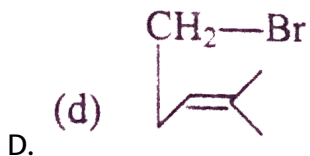
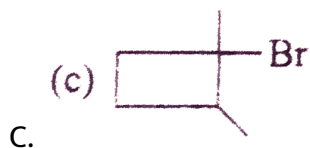
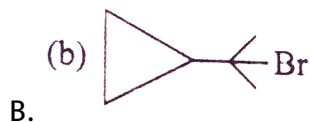
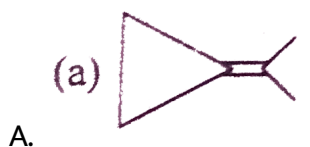
- A. Syn addition of -H (from BH_3) and -OH (from solution) occur.
- B. Syn addition of -H (from BH_3) and -OH (from H_2O_2) occur.
- C. The product is optically active.
- D. Addition follows anti Markownikoff orientation

Answer: A::C

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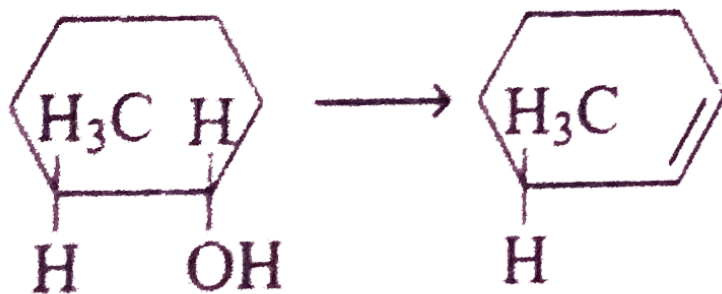
Which of the following are possible are possible products in significant amounts?



Answer: B::C::D



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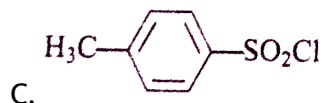


74.

Which of the following represent conditions to perform given conversation?

A. $POCl_3$, pyridine

B. Na – metal, CS_2 heat

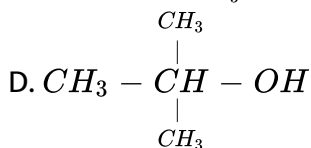
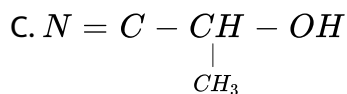
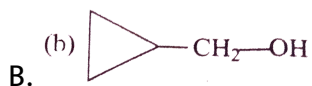
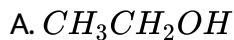


D. CF_3SO_2Cl , pyridine, $Me_3CO^{\ominus} K^{\oplus}$

Answer: B::C::D

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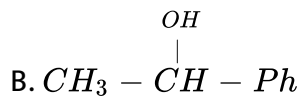
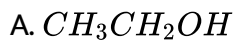
75. Which of the following alcohols do not give white turbidity on treatment $HCl / ZnCl_2$?



Answer: A:C

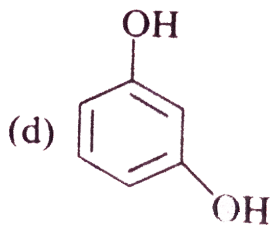
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76. Which of the followin wil give iodoform?





c.

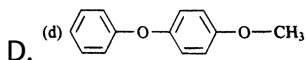
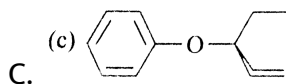
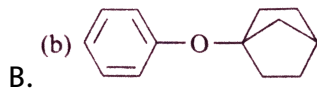
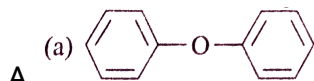


D.

Answer: A::B::D

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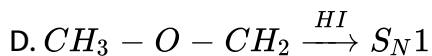
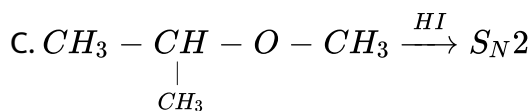
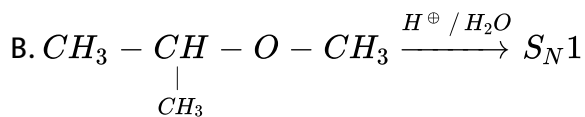
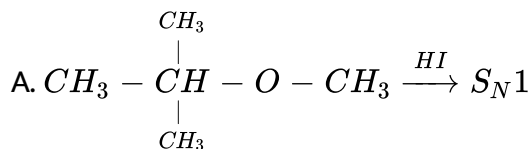
77. which of the following ethers will get hydrolysed by HI?



Answer: C::D

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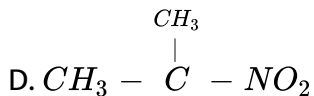
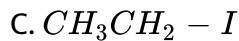
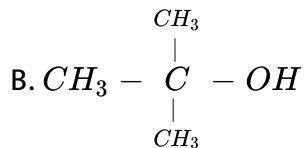
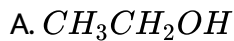
78. Which of the following reactions are correctly matched?



Answer: A::B::C

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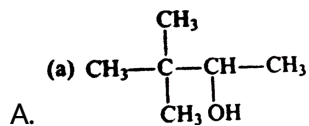
79. Which of the following compounds will give positive Victor Meyer test?



Answer: A::C::D

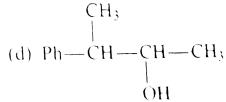
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80. Which of the following alcohols undergo rearrangement during dehydration reaction?



B. 

C. 

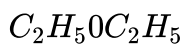


D.

Answer: A::B::D

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81.



and



can be

distinguished by:

A. aq. $FeCl_3$

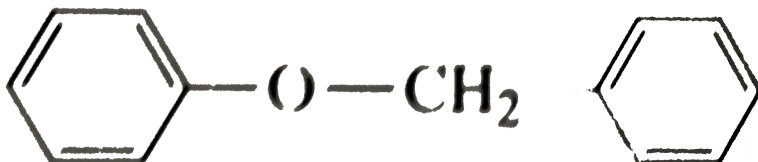
B. Na metal

C. Tollen's reagent

D. $K_2Cr_2O_7$

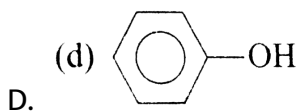
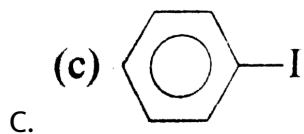
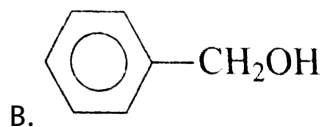
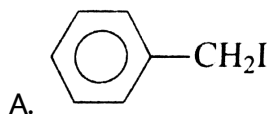
Answer: B::D

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82. The ether

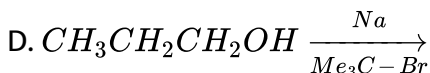
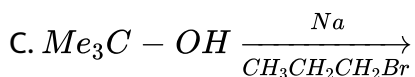
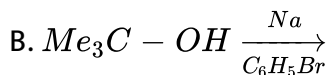
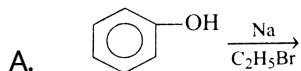
when treated with HI produces:



Answer: B::D

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83. Which of the following reactions will give ether as main product?



Answer: A::D

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84. $\text{C}_2\text{H}_5\text{Br}$ can be converted into $\text{C}_2\text{H}_5 - \text{O} - \text{C}_2\text{H}_5$ by:

A. reacting by $\text{C}_2\text{H}_5\text{ONa}$

B. heating with moist Ag_2O

C. heating with dry Ag_2O

D. treating with C_2H_5MgBr

Answer: A::C

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85. 1° , 2° and 3° alcohols can be distinguished by:

A. $Cu / 573K$

B. Victor Meyer test

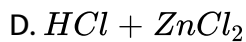
C. $ZnCl_2 / HCl$

D. $Br_2 + H_2O$

Answer: A::B::C

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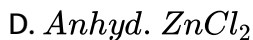
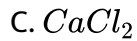
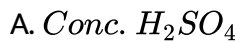
86. Alcohols can be replaced by-Cl group by the followin reagents:



Answer: B::C::D

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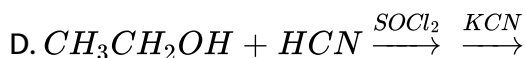
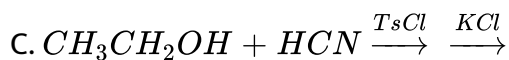
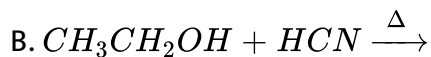
87. Glycerol can be converted to acrolein by dehydration in presence of



Answer: A::B

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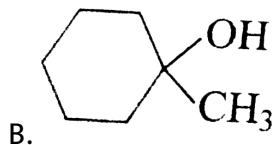
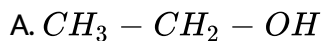
88. $CH_3CH_2 - OH$ can be converted to CH_3CH_2CN by the following reaction:

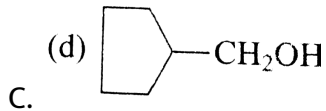


Answer: C::D

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89. Which of the following will oxidise to salt of acid by $Br_2 + KOH$?



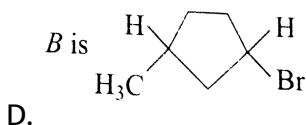
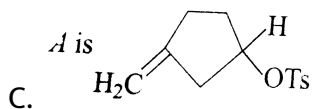
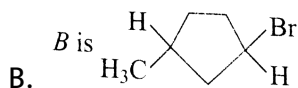
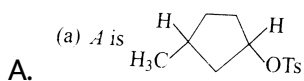
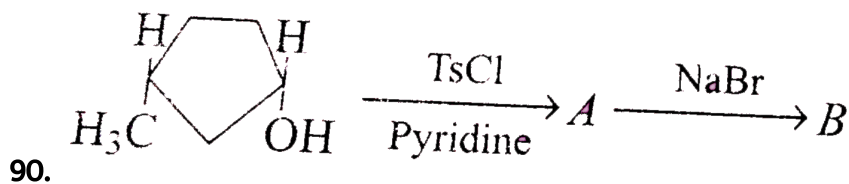


C.

D. 

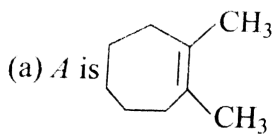
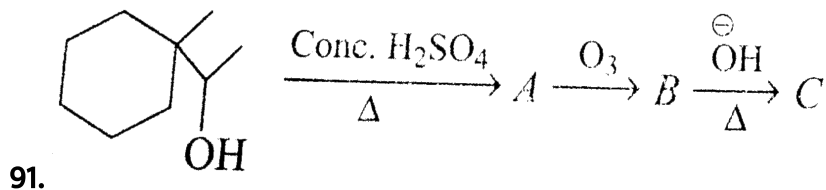
Answer: A:B

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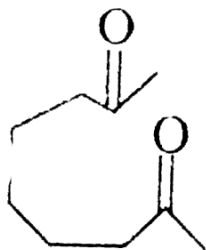


Answer: A::B

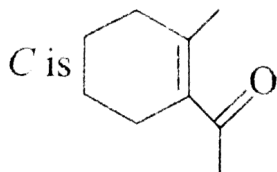
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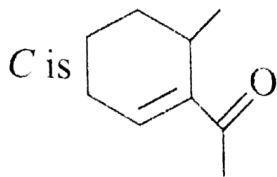
A.



B.



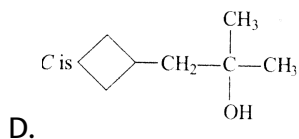
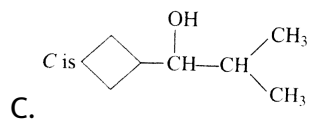
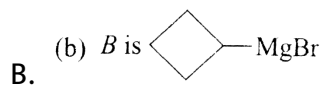
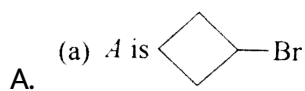
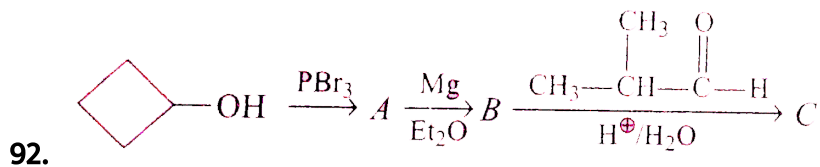
C.



D.

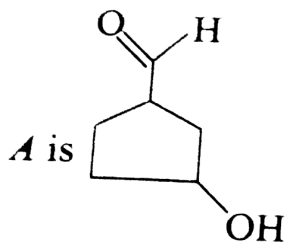
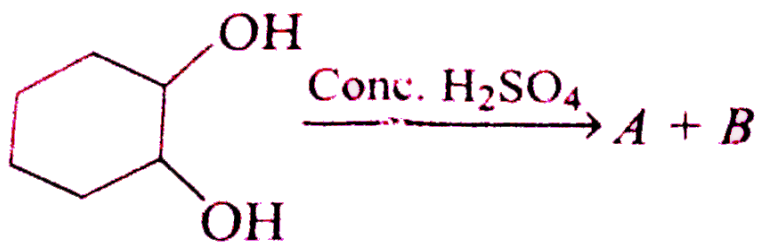
Answer: A::B::C

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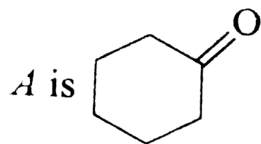


Answer: A::B::C

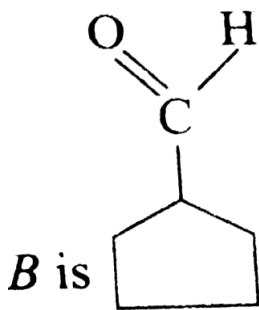
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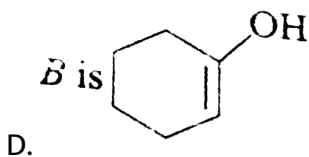
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B.



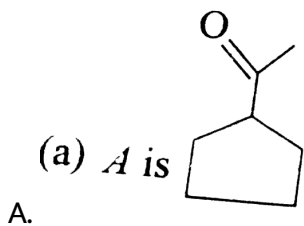
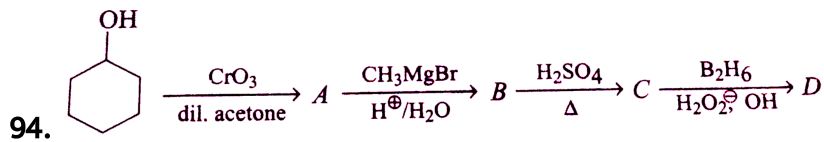
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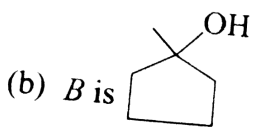


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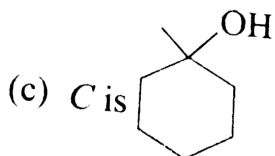
Answer: B::C::D

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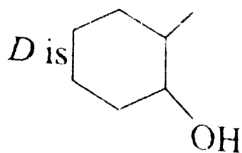




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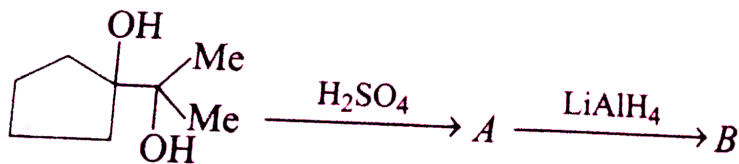
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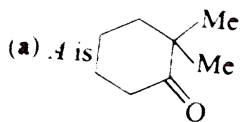
D.

Answer: C::D

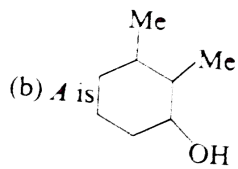
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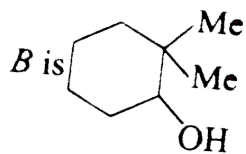
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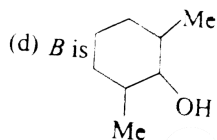
A.



B.



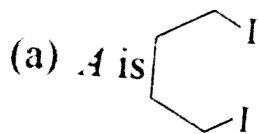
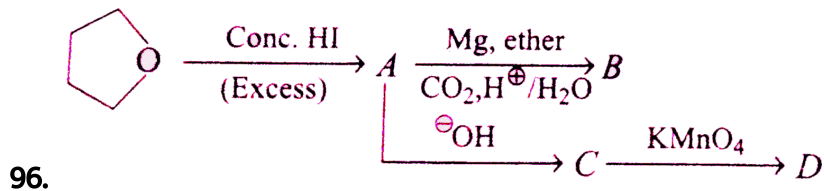
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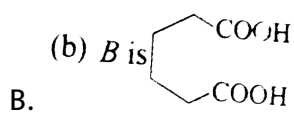
D.

Answer: A::C

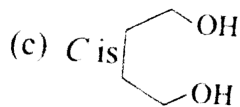
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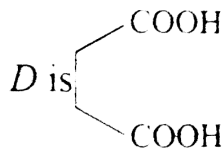
A.



B.



C.

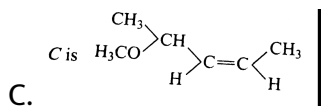
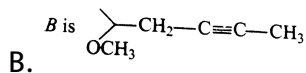
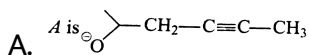
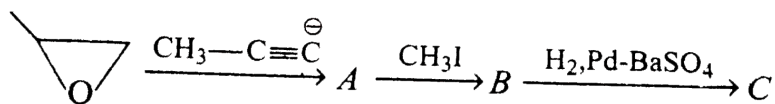


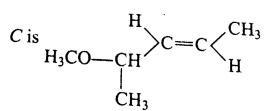
D.

Answer: A::B::C::D

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97.

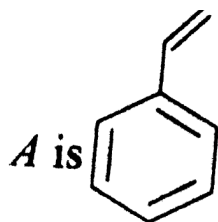
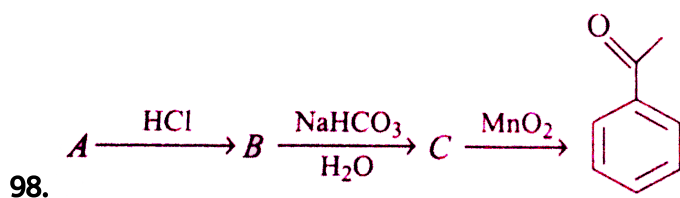




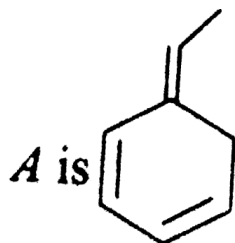
D.

Answer: A::B::C

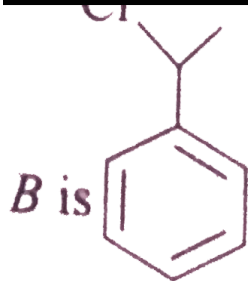
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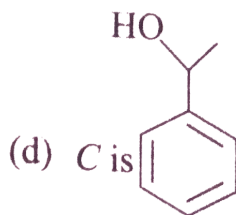
A.



B.



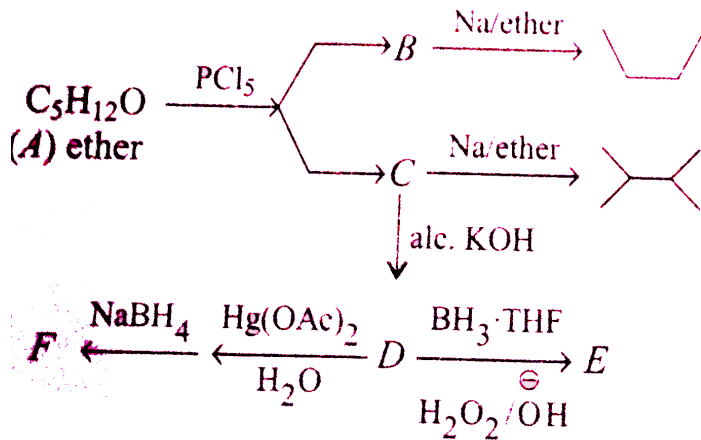
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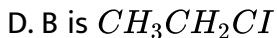
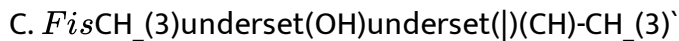
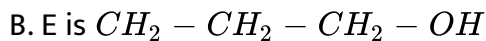
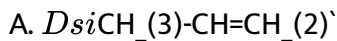
D.

Answer: A::C::D

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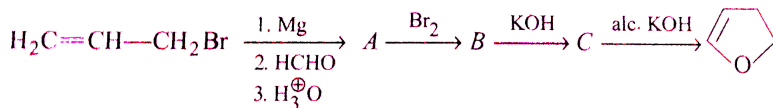


99.

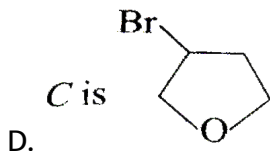
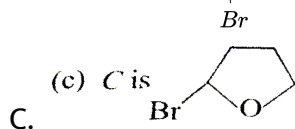
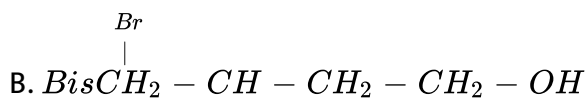


Answer: A,B,C,D

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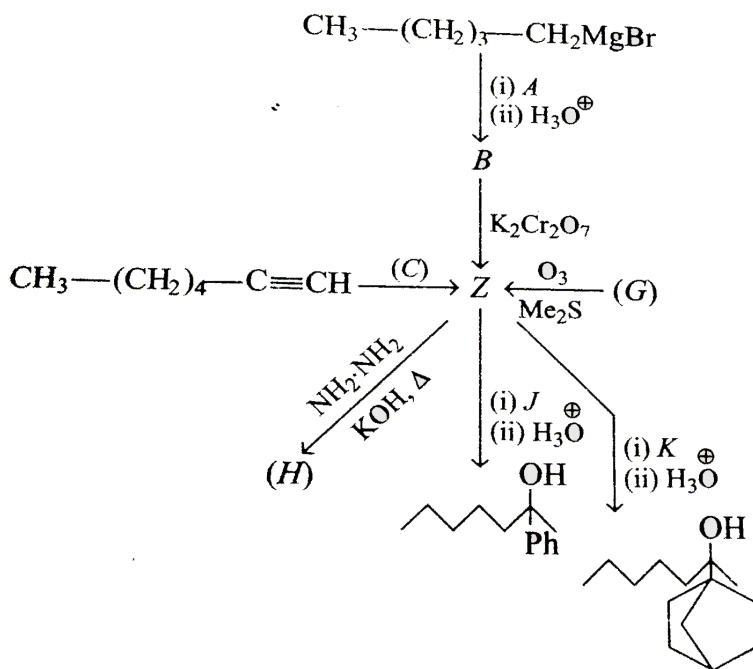


100.

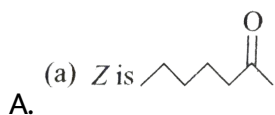


Answer: A,B,D

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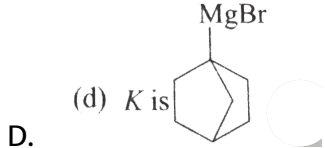


101.



B. J is PhMgBr

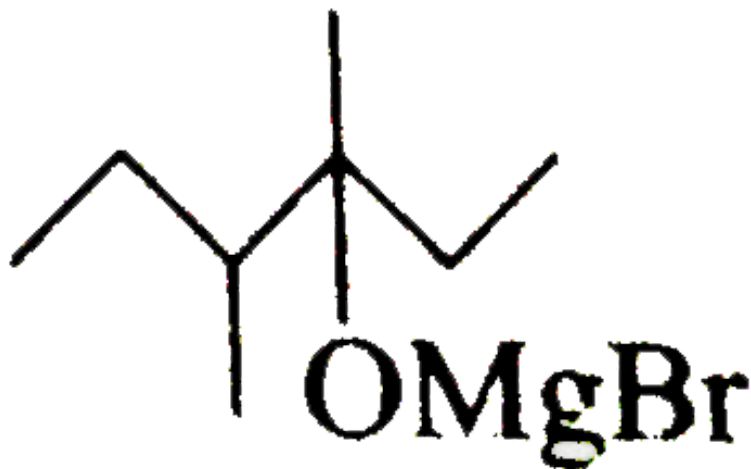
C. $\text{CisHg}^{2+} / \text{H}_2\text{SO}_4$



Answer: A,B,C,D

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102. Compound A is an optically active alcohol. Treatment with oxidising agent converts it to a ketone B. In a separate reaction A is treated with PBr_3 , converting it into C. C on reaction with Mg is added to B to yield



. Identify the correct option.

A. A is 2-butanol.

B. A is 1-butanol

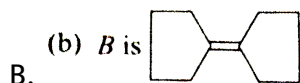
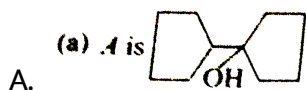
C. C is 2-bromobutane

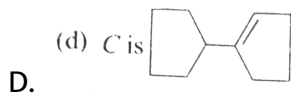
D. C is 1-bromobutane

Answer: A,C

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103. Alcohol A ($C_{10}H_{18}O$) is converted into mixture of alkene B and C on heating with conc. H_3PO_4 . Catalytic hydrogenation of B and C yields the same product. Assuming that dehydration of alcohol A proceed without rearrangement. Alkene B on ozonolysis form cyclopentanone. Identify the correct options.

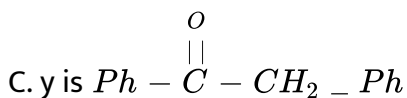
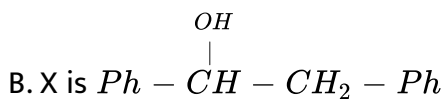
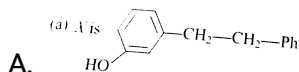




Answer: A,B,D

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104. A compound 'X' ($C_{14}H_{14}O$) on mild oxidation yields $C_{14}H_{12}(Y)$. If X is treated with a dehydrating agent, it loses a molecule of H_2O and resulting product on vigorous oxidation yields two molecules of benzoic acid. Identify the structure of X and Y.

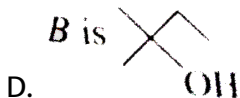
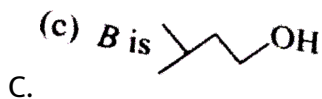
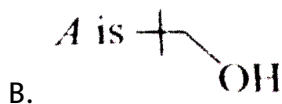
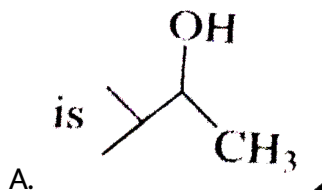


D.

Answer: B,C,D

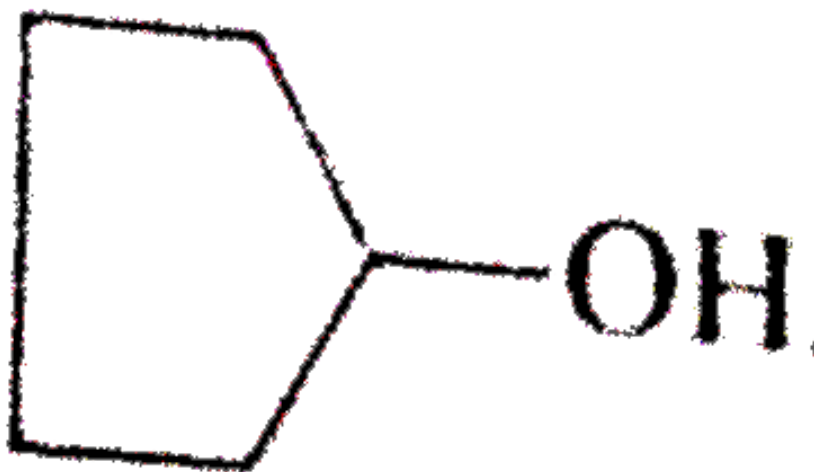
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105. Compounds A, B and C are isomeric alcohols with formula $C_5H_{12}O$. A on oxidation gives ketone, B gives acid while C is not oxidised, A gives test with $I_2/NaOH$. The three isomeric alcohols react with HBr with decreasing rates $C > A > B$. Identify A and B.



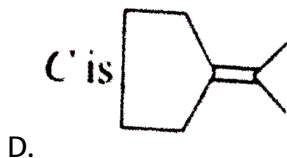
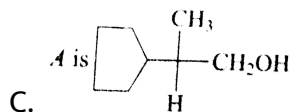
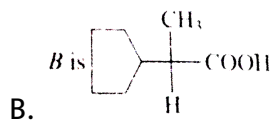
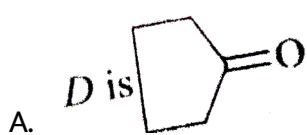
Answer: A,C

106. An optically active alcohol A ($C_8H_{16}O$) on oxidation gives B. A on heating gives $C(C_8H_{14})$ as major product. C on ozonolysis produces $D(C_5H_8O)$ and $CH_3 - \underset{\begin{array}{c} || \\ O \end{array}}{C} - CH_3$. D on reduction with $LiAlH_4$ gave



Identify

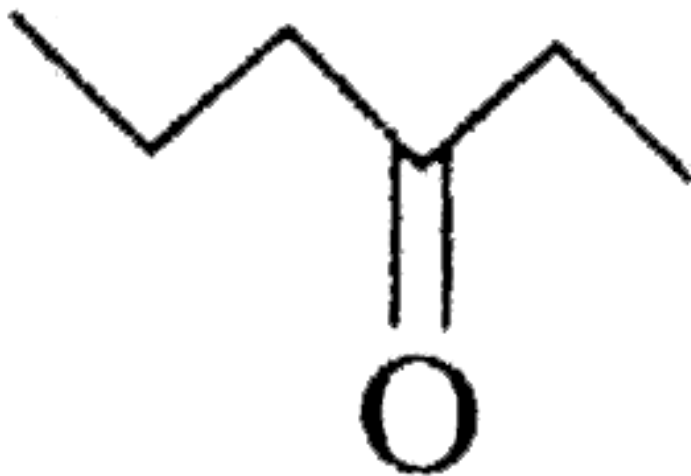
correct answers.



Answer: A,B,C,D

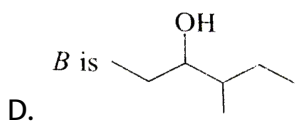
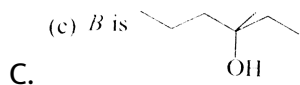
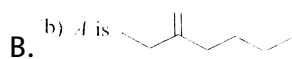
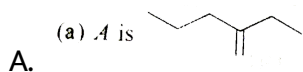
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107. Compound A (C_7H_{14}) decolourises Br_2 in CCl_4 and reacts with $Hg(OAc)_2$ followed by reduction with $NaBH_4$ to produce a resolvable compound B. A undergoes reductive ozonolysis to give



as one of the

compound



Answer: A,C



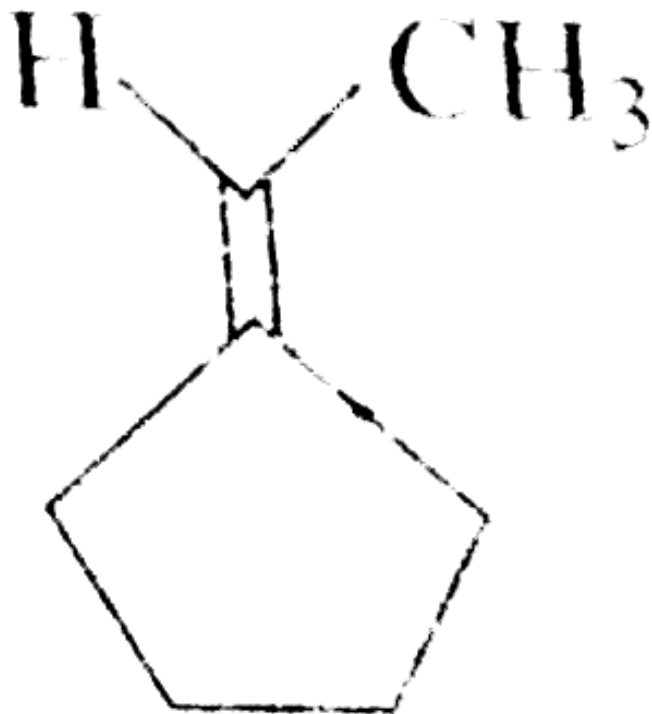
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108. A 3° optically active alcohol $C_9H_{18}O$ 'A' on dehydration with conc. H_2SO_4 produces $B(C_9H_{16})$ which exists in two stereoisomeric forms.

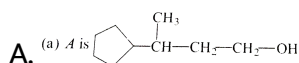
For ozonolysis of B followed by work up with $Zn - H_2O$ produces

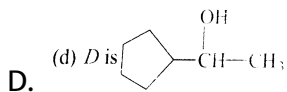
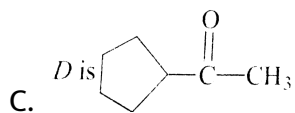
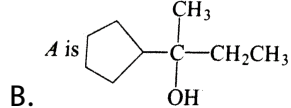
$CH_3 - \underset{\begin{array}{c} || \\ O \end{array}}{C} - H$ and $C(C_7H_{12}O)$. C on treatment with $LiAlH_4$ produces

$D(C_7H_{14}O)$. D on hydration produced



Identify the correct the correct answers.

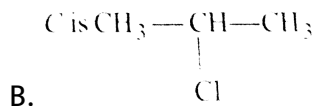
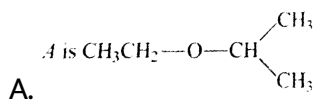




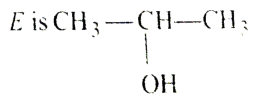
Answer: B::D

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109. A ($C_5H_{12}O$) produces, on reaction PCl_5 from alkyl chloride B and C. B and C both on reaction with aqueous KOH form alcohol D and E. Both D and E give iodoform test. Identify the correct answers.




C. C is $CH_3CH_2CH_2Cl$

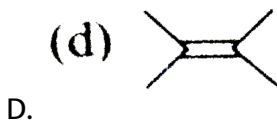
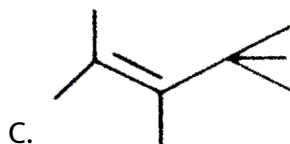
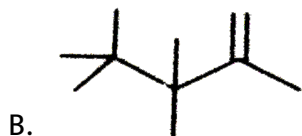
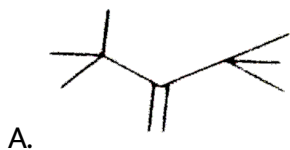


D.

Answer: A::B::D

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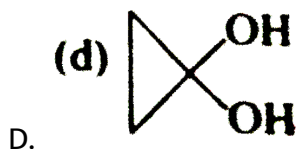
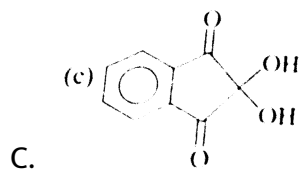
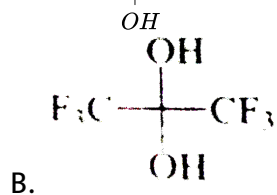
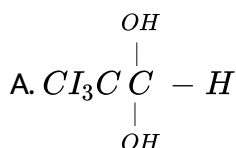
110.  $\xrightarrow[\Delta]{\text{Conc. H}_2\text{SO}_4}$; Products can be : _____, Product can be



Answer: A::B

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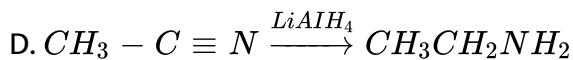
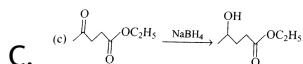
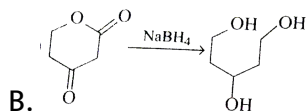
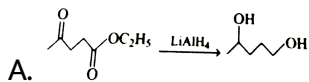
111. Among of the following gemdiols which are stable with respect to corresponding carbonyls:



Answer: A::B::C::D

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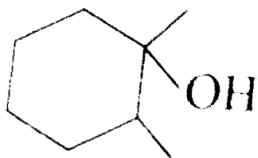
112. Which of the following reactions are correct



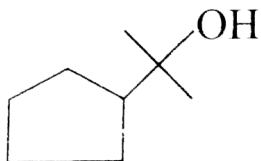
Answer: A::C::D

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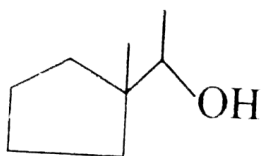
113. Which of the following alcohols will give same alkene on reaction with conc.



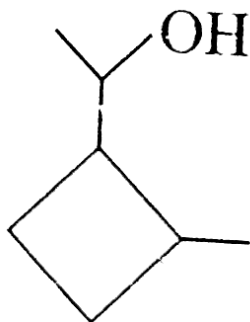
A.



B.



C.

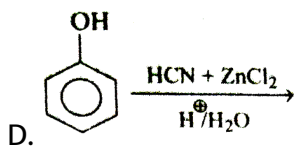
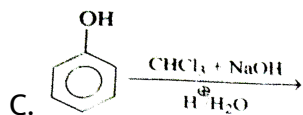
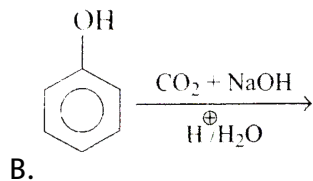
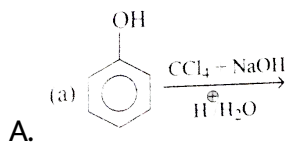


D.

Answer: A::B::C

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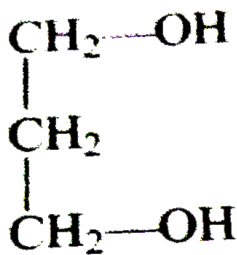
114. Which of the following reactions would produce same product?



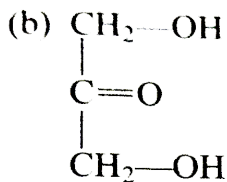
Answer: A::B

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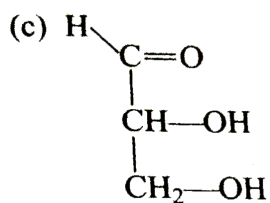
115. Which of the following compound are oxidised by HIO_4 ?



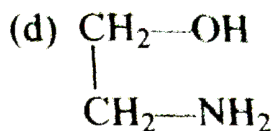
A.



B.



C.



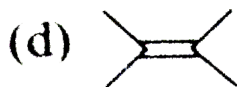
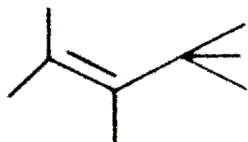
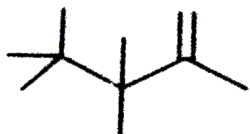
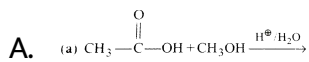
D.

Answer: B::C::D



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116. Which of the following esterification reactions are unimolecular?

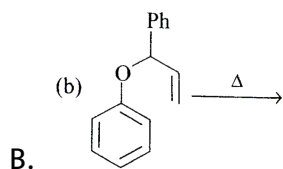
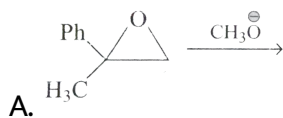


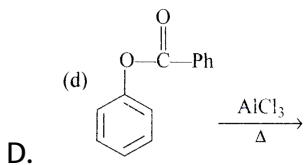
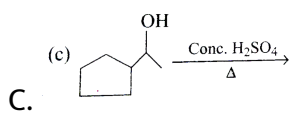
D.

Answer: B::C::D

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117. Which of the following reaction involve rearrangement?

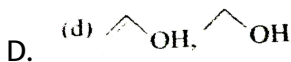
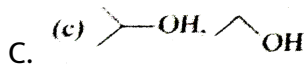
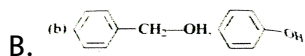
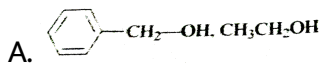




Answer: B::C::D

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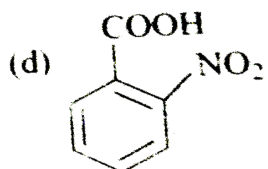
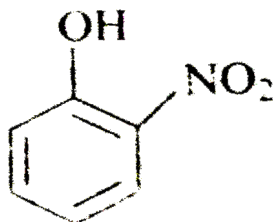
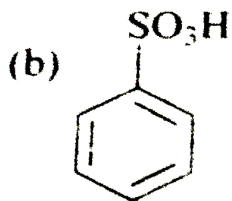
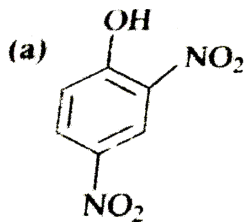
118. Which of the following pairs can be distinguished by using Lucas reagent?



Answer: A::B::C

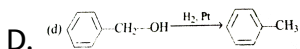
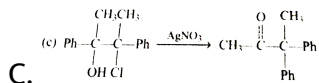
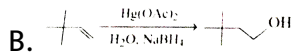
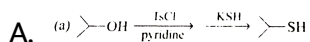
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119. Which of the following compounds are soluble in $NaHCO_3$?



Answer: A::B::D

120. Which of the following reactions are correctly interpreted?



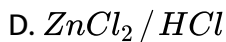
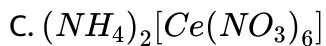
Answer: A::C::D

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121. Which of the following reagents can be used for identification of phenol?

A. Neutral FeCl_3

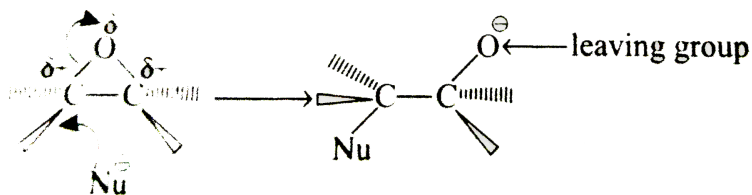
B. $\text{NaNO}_2 + \text{HCl}$



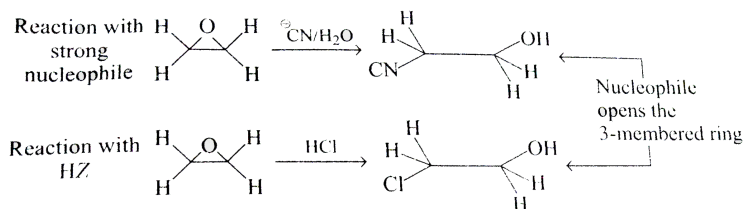
Answer: A::B::C

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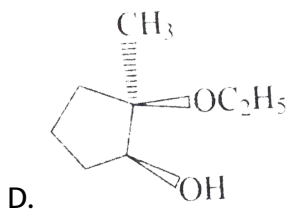
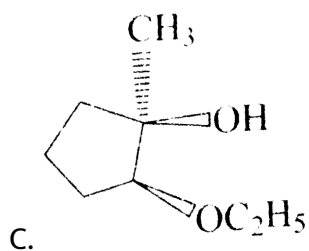
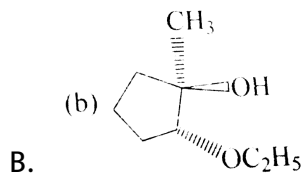
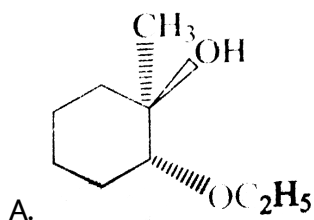
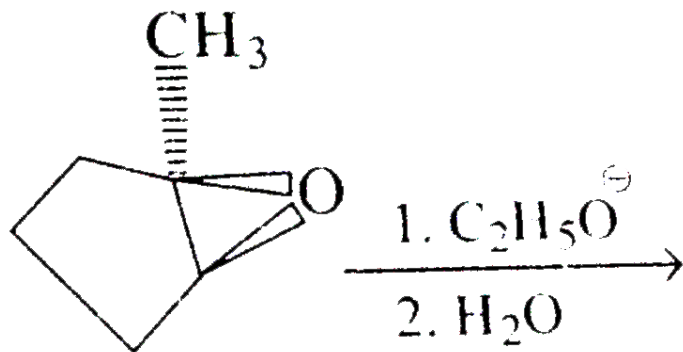
122. Although epoxides do not contain a good leaving group, they contain a strained three membered ring with polar bonds. Nucleophilic attack opens the strained three membered ring making it a favorable process even with the poor leaving group.



This reaction occurs readily with strong nucleophiles, and with acids like HZ , where Z is a nucleophilic atom.



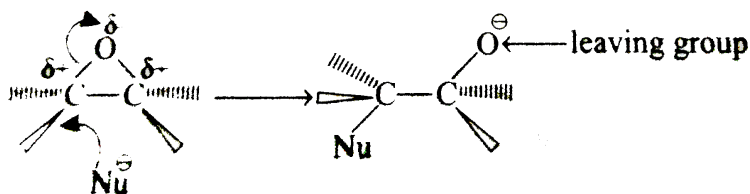
Find out the correct product of the reaction



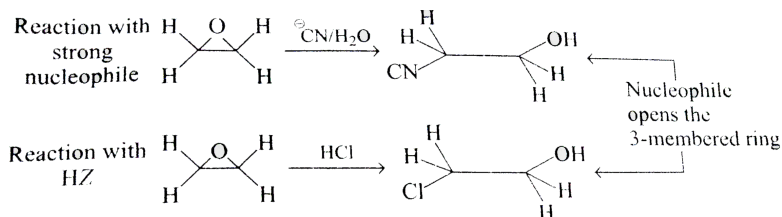
Answer: B

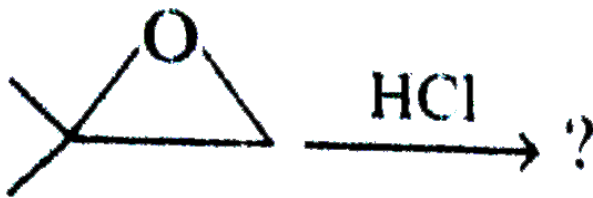
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123. Although epoxides do not contain a good leaving group, they contain a strained three membered ring with polar bonds. Nucleophilic attack opens the strained three membered ring making it a favorable process even with the poor leaving group.

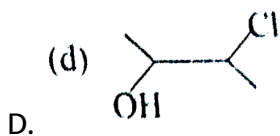
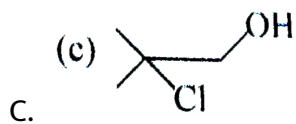
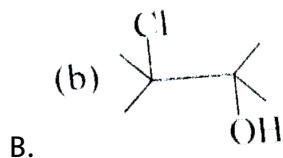
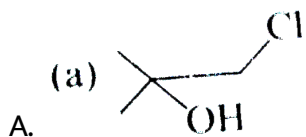


This reaction occurs readily with strong nucleophiles, and with acids like HZ, where Z is a nucleophilic atom.





What would be the major product of reaction?

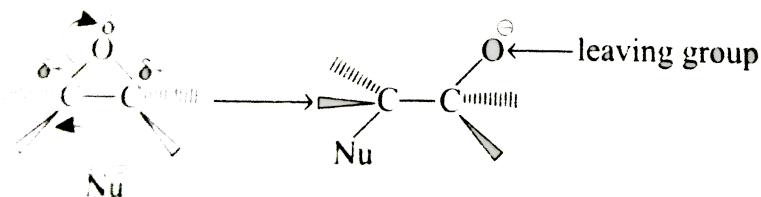


Answer: C

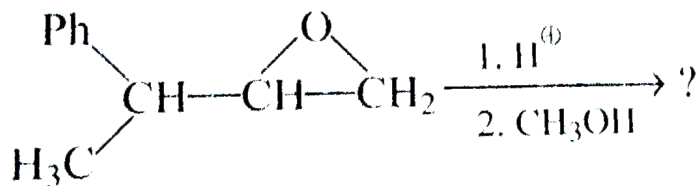
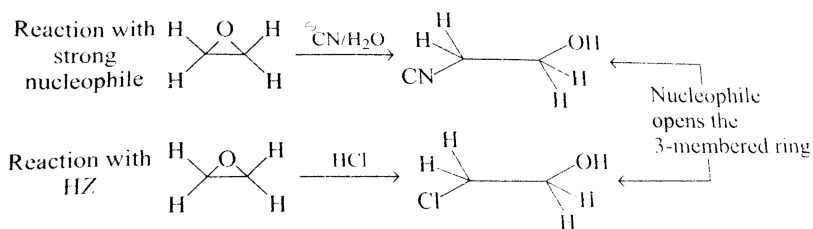


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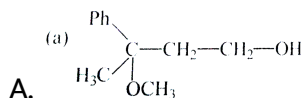
124. Although epoxides do not contain a good leaving group, they contain a strained three membered ring with polar bonds. Nucleophilic attack opens the strained three membered ring making it a favorable process even with the poor leaving group.

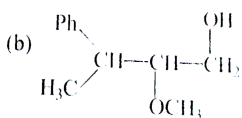


This reaction occurs readily with strong nucleophiles, and with acids like HZ, where Z is a nucleophilic atom.

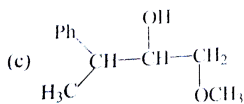


Find out the major product of the reaction:





B.



C.

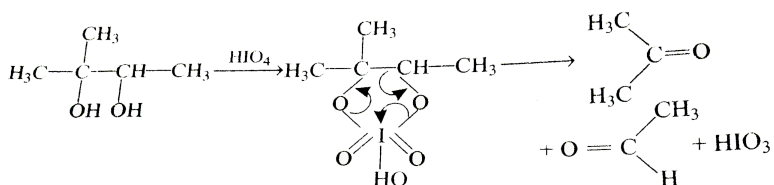
D. None of these

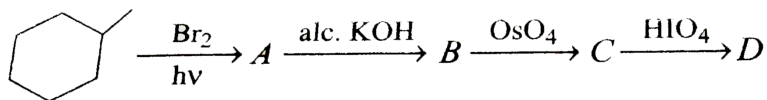
Answer: A

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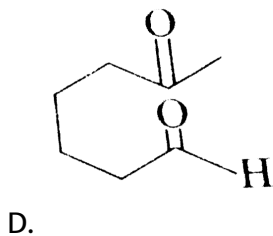
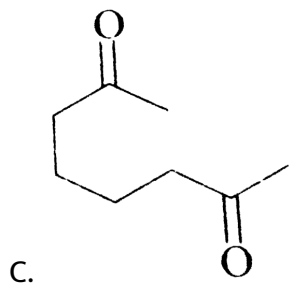
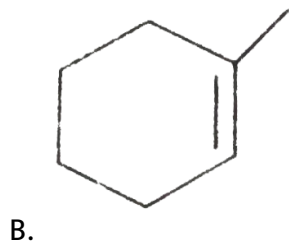
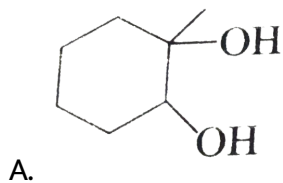
125. 1,2-diols are oxidised to ketones or aldehydes by periodic acid HIO_4 .

Periodic acid reacts with dipol to form a cyclic intermediate. The reaction takes place because iodine is in a highly positive oxidation state, so it readily accepts electrons. When the intermediate breaks down, the bond between the two carbon bonded to the OH group break.



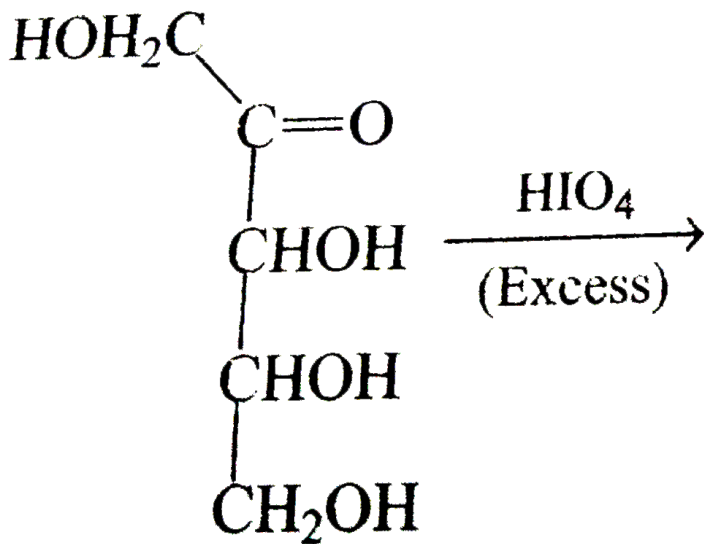
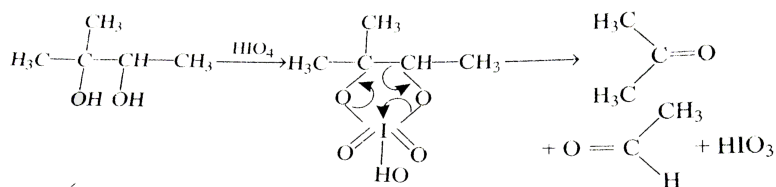


Identify D.

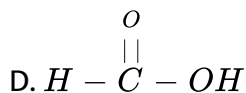
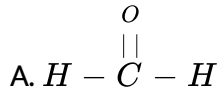


Answer: D

126. 1,2-diols are oxidised to ketones or aldehydes by periodic acid HIO_4 . Periodic acid reacts with diol to form a cyclic intermediate. The reaction takes place because iodine is in a highly positive oxidation state, so it readily accepts electrons. When the intermediate breaks down, the bond between the two carbon bonded to the OH group break.



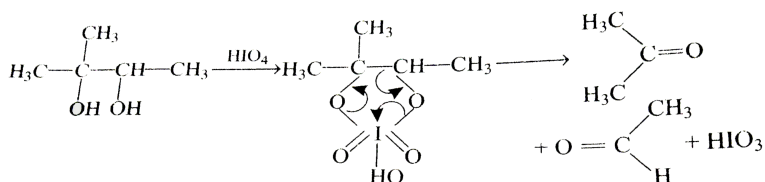
Which of the following will not form by above reaction?



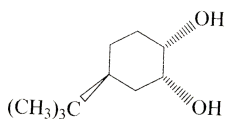
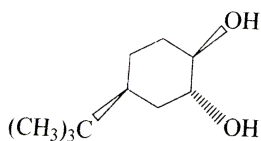
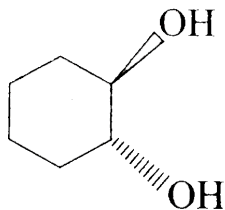
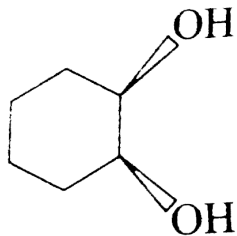
Answer: B

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127. 1,2-diols are oxidised to ketones or aldehydes by periodic acid HIO_4 . Periodic acid reacts with diol to form a cyclic intermediate. The reaction takes place because iodine is in a highly positive oxidation state, so it readily accepts electrons. When the intermediate breaks down, the bond between the two carbon bonded to the OH group breaks.



Which of the following compounds will not react with HIO_4 ?



A.

B.

C.

D.

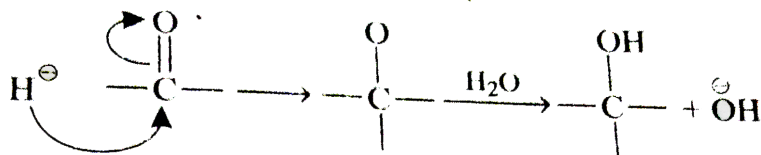
Answer: C



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128. Carbon oxygen double bond are easily reduced by NaBH_4 or LiAlH_4 . The actual reducing agent in these reduction is

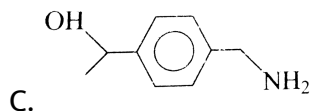
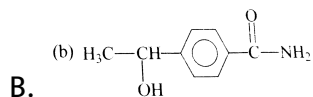
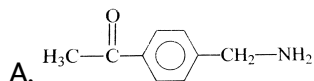
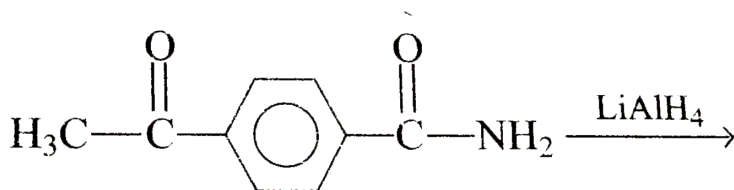
hydride ion (H^-)



The metal hydrogen bond in $LiAlH_4$ is more polar than metal hydrogen bond in $NaBH_4$. As a result $LiAlH_4$ is a stronger reducing agent than $NaBH_4$. Esters, carboxylic acids, amides cannot be reduced by $NaBH_4$.

The carbonyl group of an amide is reduced to a methylene group by $LiAlH_4$.

Find the correct product of the following reaction:

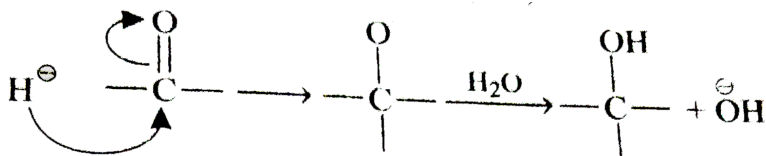


D. No reaction

Answer: C

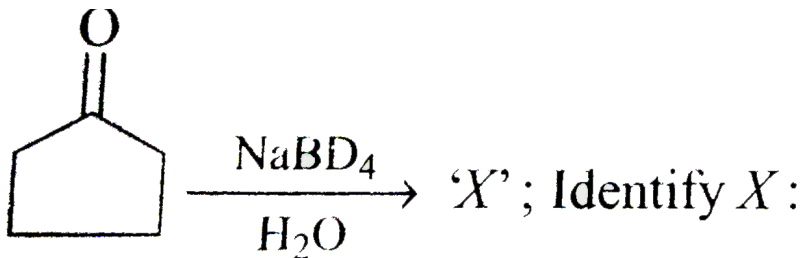
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129. Carbon oxygen double bond are easily reduced by NaBH_4 or LiAlH_4 . The actual reducing agent in these reduction is hydride ion (H^-)

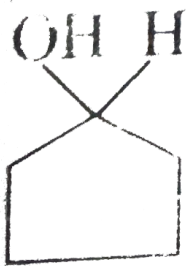


The metal hydrogen bond in LiAlH_4 is more than polar than metal hydrogen bond in NaBH_4 . As a result LiAlH_4 is strong reducing agent than NaBH_4 . Esters, carboxylic acids, amides cannot be reduced by NaBH_4

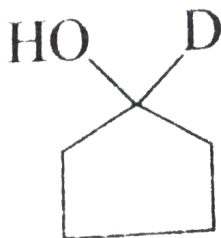
The carbonyl group of amide of reduced to methylene group by LiAlH_4



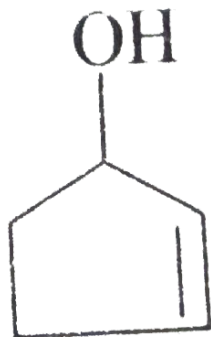
, Identify X:



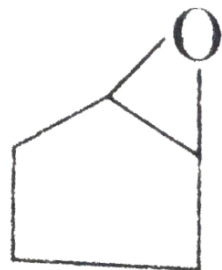
A.



B.



C.

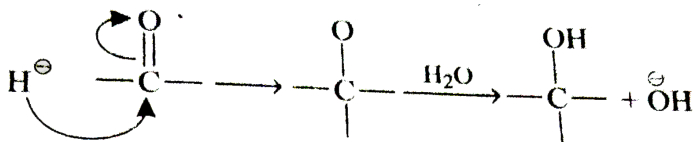


D.

Answer: B

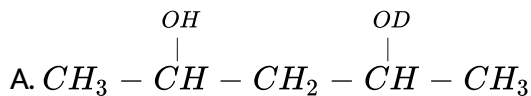
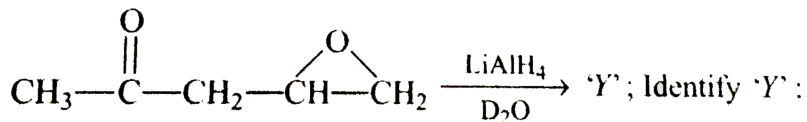
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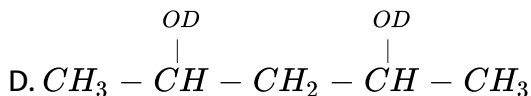
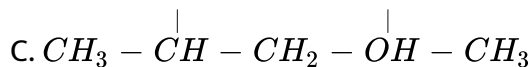
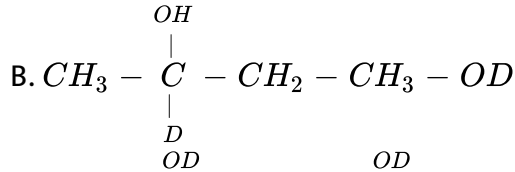
130. Carbon oxygen double bond are easily reduced by NaBH_4 or LiAlH_4 . The actual reducing agent in these reduction is hydride ion (H^-)



The metal hydrogen bond in LiAlH_4 is more than polar than metal hydrogen bond in NaBH_4 . As a result LiAlH_4 is strong reducing agent than NaBH_4 . Esters, carboxylic acids, amides cannot be reduced by NaBH_4

The carbonyl group of amide of reduced to methylene group by LiAlH_4





Answer: D

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131. An organic compound (A) on treatment with CHCl_3 and KOH gives (Y) and (Z) both of which in turn gives the same compound (T) when distilled with Zn. Oxidation of (T) Yields (S) of formula $\text{C}_7\text{H}_6\text{O}_2$. The sodium salt of (S) with sodalime gives (P) which can also be obtained by distilling (X).

The molecular weight of compound (X) is:

A. 122

B. 94

C. 106

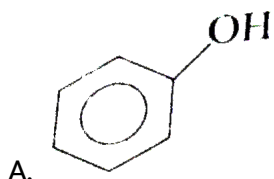
D. 78

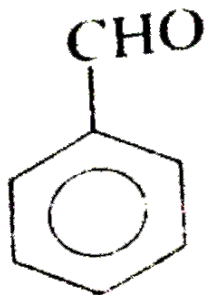
Answer: B

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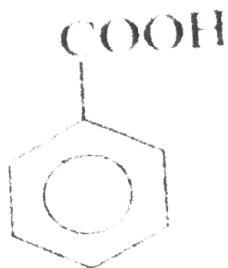
132. An organic compound (A) on treatment with $CHCl_3$ and KOH gives (Y) and (Z) both of which in turn gives the same compound (T) when distilled with Zn. Oxidation of (T) yields (S) of formula $C_7H_6O_2$. The sodium salt of (S) with sodalime gives (P) which can also be obtained by distilling (X).

The compound (T) is





C.



D.

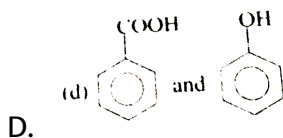
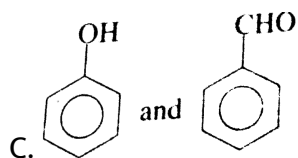
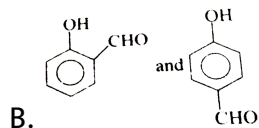
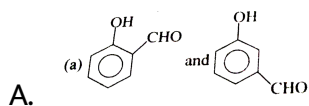
Answer: C

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133. An organic compound (A) on treatment with $CHCl_3$ and KOH gives (Y) and (Z) both of which in turn gives the same compound (T) when distilled with Zn. Oxidation of (T) Yields (S) of formula $C_7H_6O_2$. The sodium salt of (S) with sodalime gives (P) which can also be obtained by

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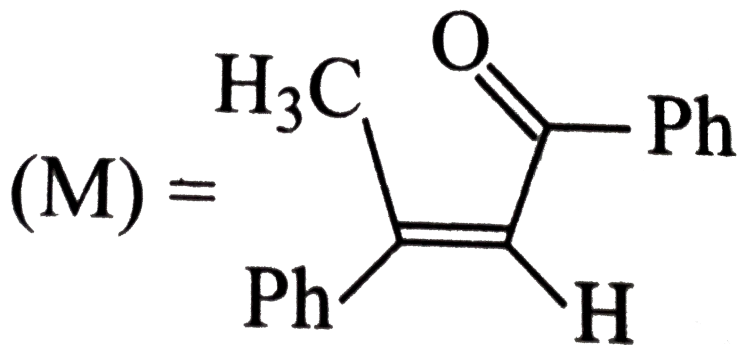
Compounds (Y) and (Z) could be:



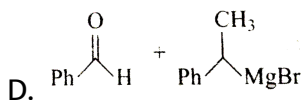
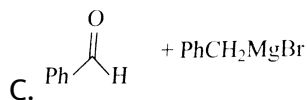
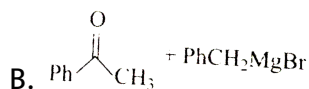
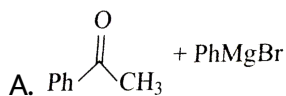
Answer: B

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134. A tertiary alcohol (H) upon acid-catalysed dehydration gives a product (I). Ozonolysis of (I) leads to compounds (J) and (K). Compound (J) upon reaction with KOH gives benzyl alcohol and a compound (L), whereas (K) on reaction with KOH gives only (M).



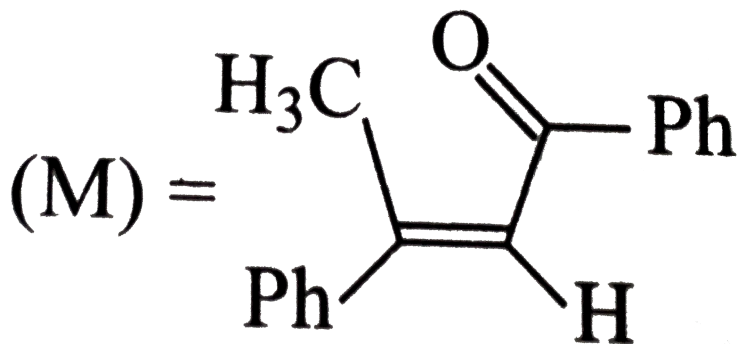
Compound (H) is formed by the reaction of:



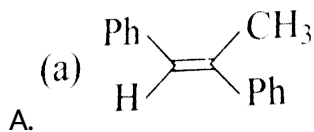
Answer: B

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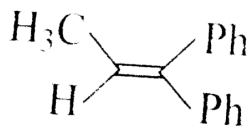
135. A tertiary alcohol (H) upon acid-catalysed dehydration gives a product (I). Ozonolysis of (I) leads to compounds (J) and (K). Compound (J) upon reaction with KOH gives benzyl alcohol and a compound (L), whereas (K) on reaction with KOH gives only (M).



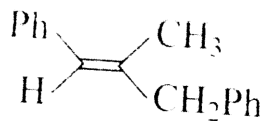
The structure of compound (I) is:



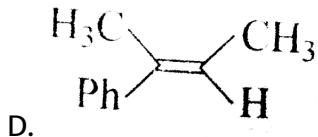
A.



B.



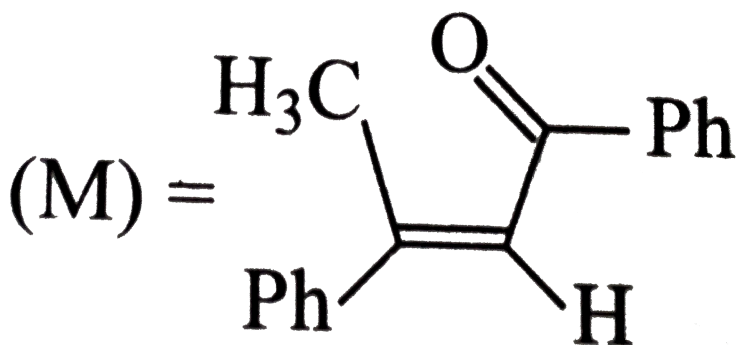
C.



Answer: A

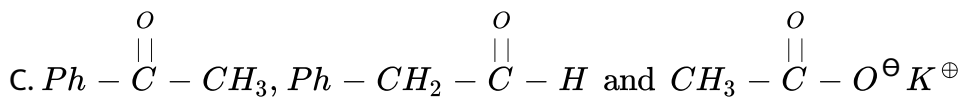
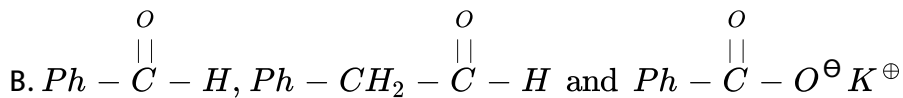
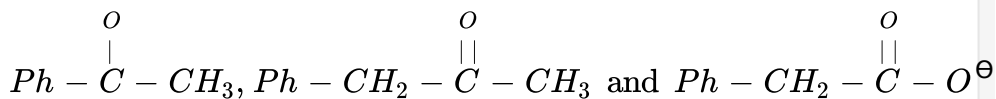
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136. A tertiary alcohol (H) upon acid-catalysed dehydration gives a product (I). Ozonolysis of (I) leads to compounds (J) and (K). Compound (J) upon reaction with KOH gives benzyl alcohol and a compound (L), whereas (K) on reaction with KOH gives only (M).



The structures of compounds (J), (K), and (L), respectively, are:

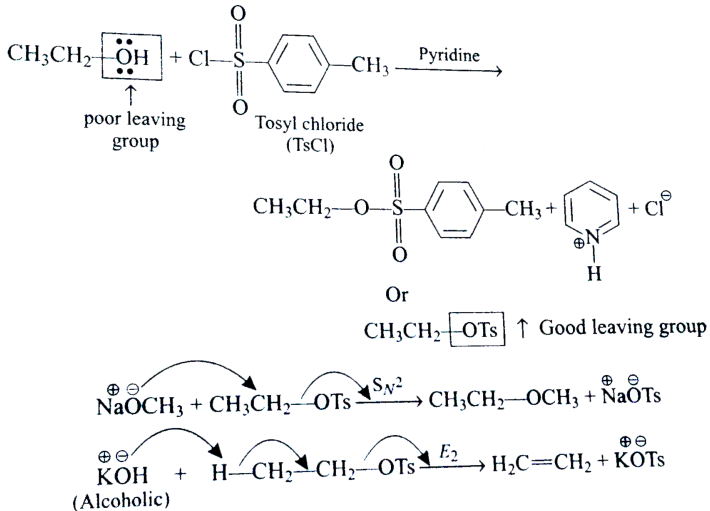
A.



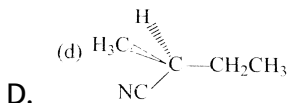
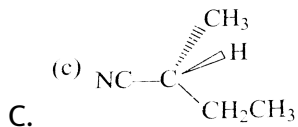
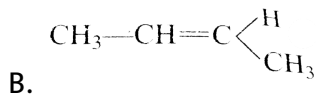
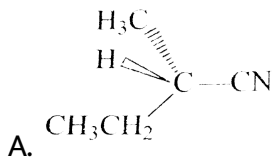
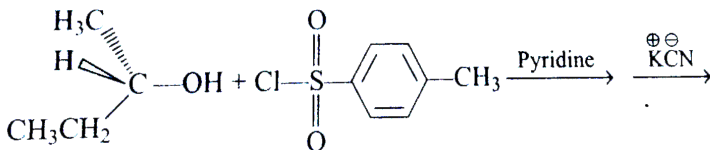
Answer: D

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137. Alcohols are converted to tosylates by treatment with p-toluene sulfonyl chloride (TsCl) in the presence of pyridine. This overall process converts a poor leaving group $\left(\overset{\ominus}{H} \right)$ into good one $\left(\overset{\ominus}{Ts} \right)$. A tosylate is a good leaving group its conjugates acid p-touence sulfonic acid is strong acid. Beacuse alkyl tosylates have food leaving groups, they undergo both nucleophilic substitution and β - elimination.



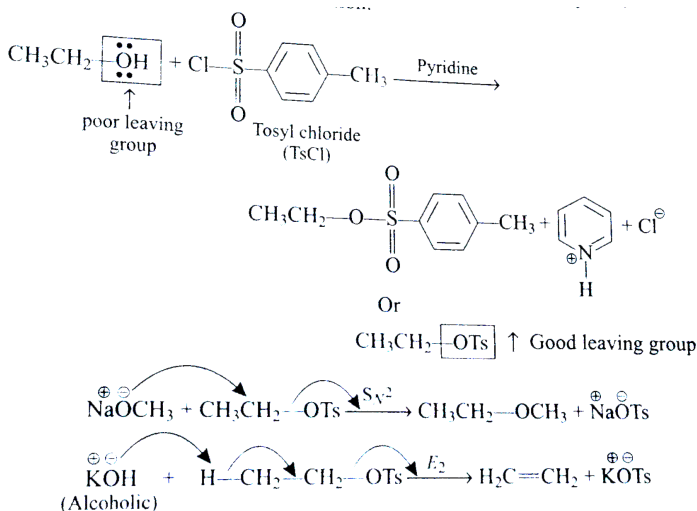
Find the major product of the following reaction:



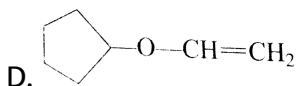
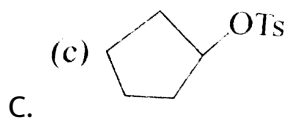
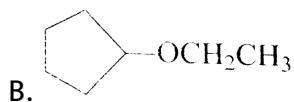
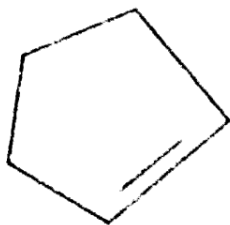
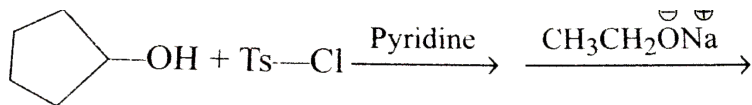
Answer: C

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138. Alcohols are converted to tosylates by treatment with p-toluene sulfonyl chloride (TsCl) in the presence of pyridine. This overall process converts a poor leaving group (H) into good one (OTs). A tosylate is a good leaving group its conjugate acid p-toluene sulfonic acid is strong acid. Because alkyl tosylates have good leaving groups, they undergo both nucleophilic substitution and β - elimination.



What would be the major product of the following reactions?

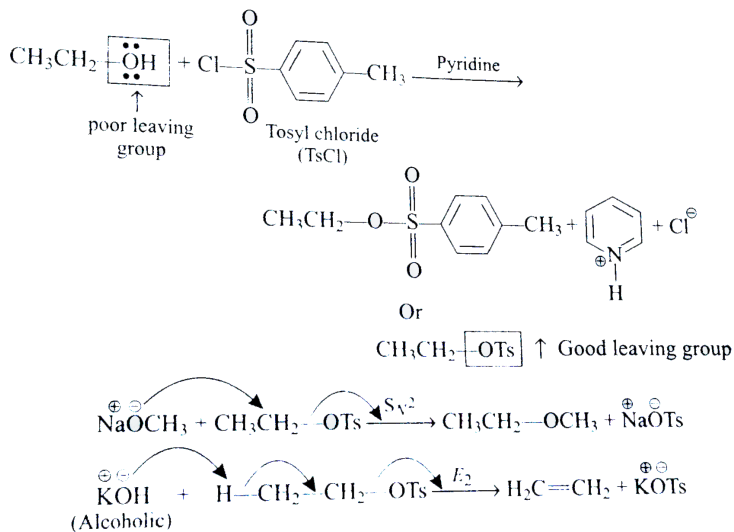


Answer: A

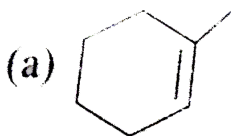
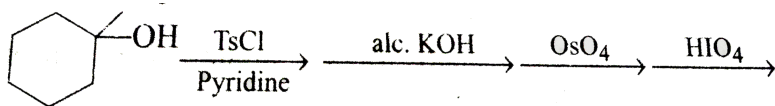
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139. Alcohols are converted to tosylates by treatment with p-toluence sulfonyl chloride (TsCl) in the presence of pyridine. This overall process

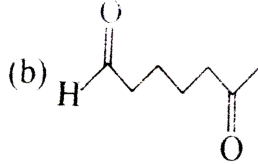
converts a poor leaving group $\left(\overset{\ominus}{H} \right)$ into good one $\left(\overset{\ominus}{Ts} \right)$. A tosylate is a good leaving group its conjugates acid p-touence sulfonic acid is strong acid. Beacuse alkyl tosylates have food leaving groups, they undergo both nucleophilic substitution and β – elimination.



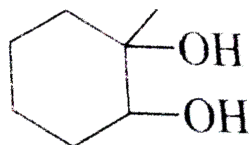
Identify the final product of the following sequences of reactions:



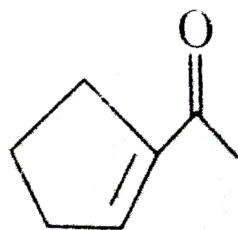
A.



B.



C.

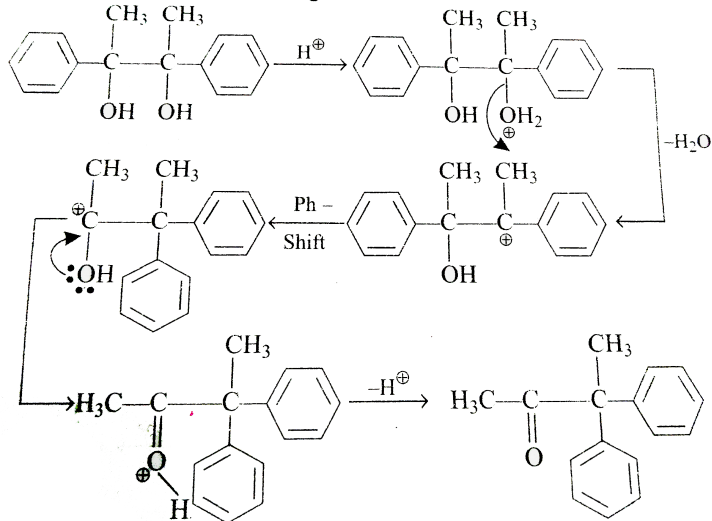


D.

Answer: B

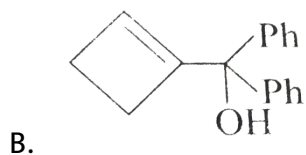
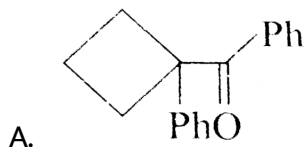
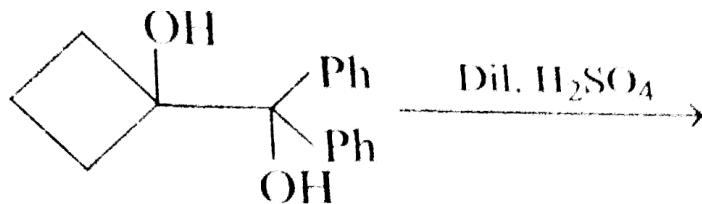
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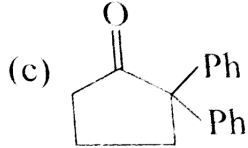
140. Acid catalysed conversion of 1,2-diol or vicinal, into carbonyl compound known as pinacol-pinacolone rearrangement.



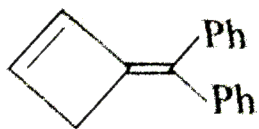
Generally more electron donating group migrate during mechanism, migration of H is faster because of its smaller size.

What would be the major product of reaction?





C.

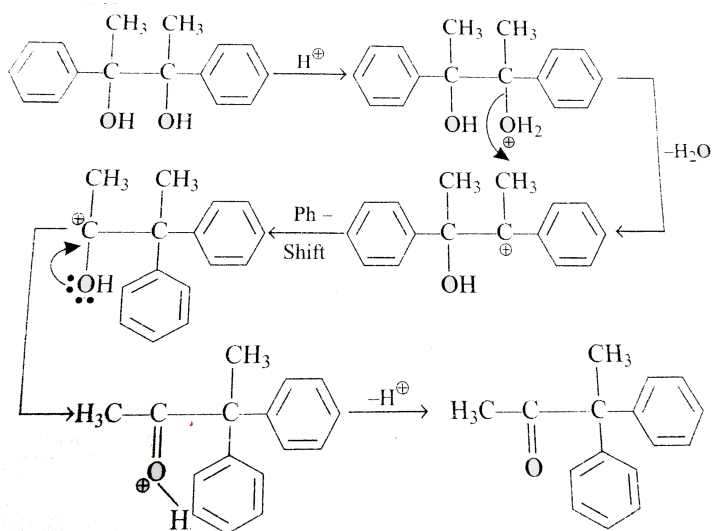


D.

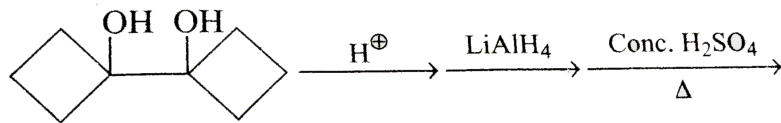
Answer: C

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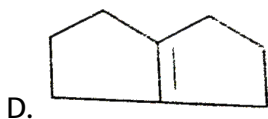
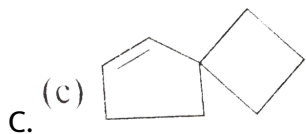
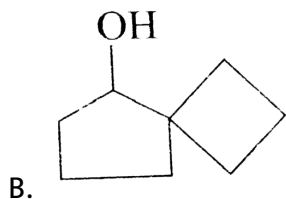
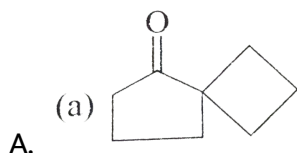
141. Acid catalysed conversion of 1,2-diol or vicinal, into carbonyl compound known as pinacol-pinacolone rearrangement.



Generally more electron donating group migrate during mechanism, migration of H is faster because of its smaller size.



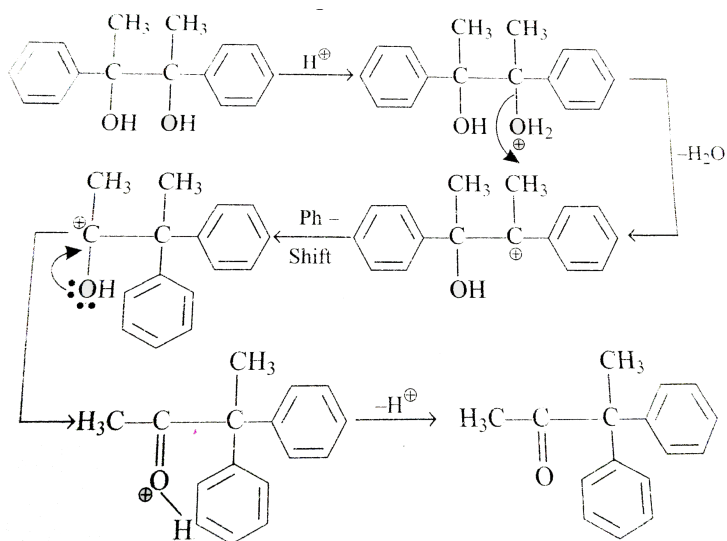
In this sequence of reaction final product is"



Answer: D

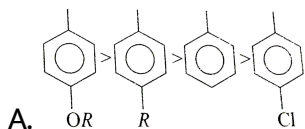
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142. Acid catalysed conversion of 1,2-diol or vicinal, into carbonyl compound known as pinacol-pinacolone rearrangement.



Generally more electron donating group migrate during mechanism, migration of H is faster because of its smaller size.

Which of the following is not correct about this rearrangement?



B. The carboncation is stabilised by 1,2-shift

C. Migration aptitude for substituent is in $R - > H - > C_6H_5$

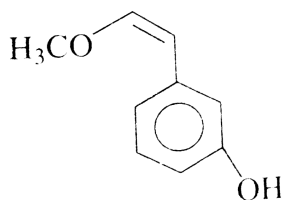
D. Product of reaction is carbonyl compound.

Answer: C

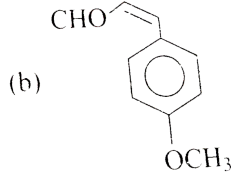
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143. Compound (A) $C_{10}H_{22}O_2$ is insoluble in aq. NaOH but not in $NaHCO_3$. Treatment of (A) with DMSO $\left(CH_3 - \overset{S}{\parallel} - CH_3 \right)$ in alkali gives (B) $C_{11}H_{14}O_2$. Treatment of (A) with strong alkali alone gives an isomeric compound (C). When (A) is refluxed with HI, CH_3I is obtained, compound (B) is insoluble in alkali and decolorises Br_2/CCl_4 . (B) on treatment with strong base gives (D), an isomer of (B). Ozonolysis of (C) gives (E), C_8H_8O and isomer of vanillin. Ozonolysis of (D) gives (F) $C_9H_{10}O_3$, which is identical with the product of methylation of vanillin (4-hydroxy-3-methoxybenzaldehyde).

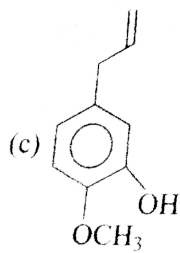
Structure of compound (A) is:



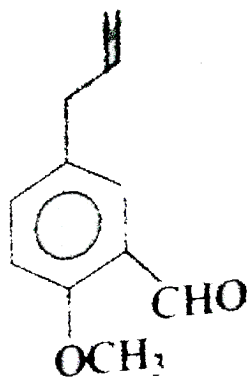
A.



B.



C.



D.

Answer: C



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144. Compound (A) $C_{10}H_{22}O_2$ is insoluble in aq. NaOH but not in $NaHCO_3$

. Treatment of (A) with DMSO $\left(CH_3 - \overset{S}{\parallel} - CH_3 \right)$ in alkali give (B)

$C_{11}H_{14}O_2$. Treatment of (A) with strong alkali alone give an isomeric

compound (C). When (A) is reflux with HI, CH_3I is obtained, compound

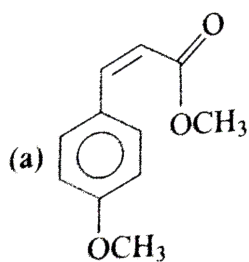
(B) is insoluble in alkali and decolorises Br_2/CCl_4 . (B) on treating with

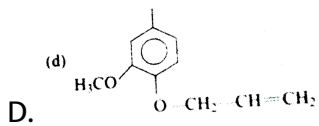
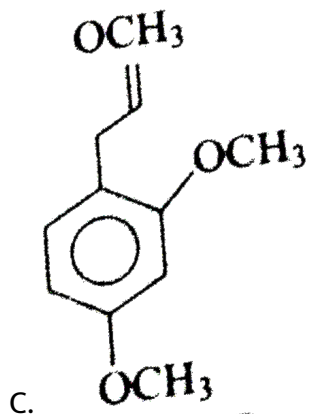
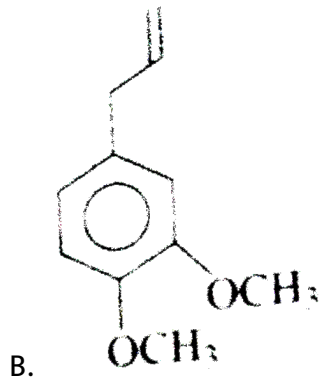
strong base gives (D), an isomer of (B). Ozonolysis (C) of gives (E), C_8H_8O

and isomer of vanilline. Ozolysis of (D) gives (F) $C_9H_{10}O_3$, which is

identical with product of methylation of vanilline (4-hydroxy-3-methoxy benzaldehyde).

Compound (B) is:





Answer: B



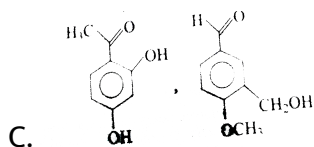
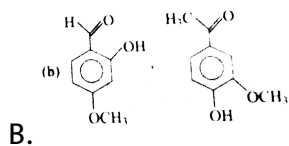
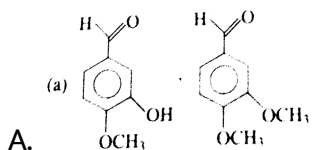
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145. Compound (A) $C_{10}H_{22}O_2$ is insoluble in aq. NaOH but not in $NaHCO_3$

. Treatment of (A) with DMSO $\left(CH_3 - \overset{S}{\parallel} - CH_3 \right)$ in alkali give (B)

$C_{11}H_{14}O_2$. Treatment of (A) with strong alkali alone give an isomeric compound (C). When (A) is reflux with HI, CH_3I is obtained, compound (B) is insoluble in alkali and decolorises Br_2/CCl_4 . (B) on treating with strong base gives (D), an isomer of (B). Ozonolysis (C) of gives (E), C_8H_8O and isomer of vanilline. Ozolysis of (D) gives (F) $C_9H_{10}O_3$, which is identical with product of methylation of vanilline (4-hydroxy-3-methoxy benzaldehyde).

Compound (E) and (F) are respectively:

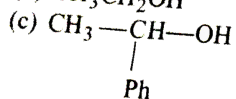
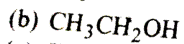
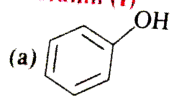


D. None of these

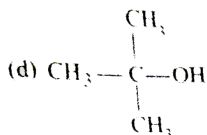
Answer: A

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Column (I)



146.



Column (II)

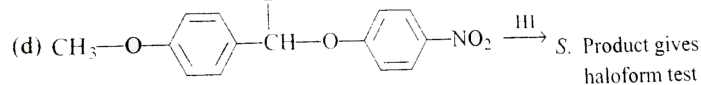
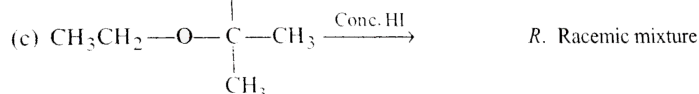
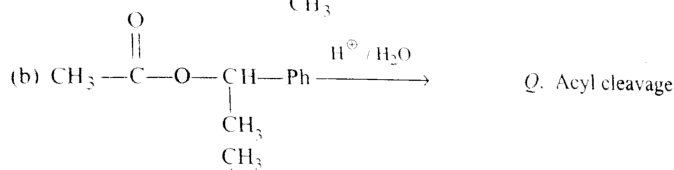
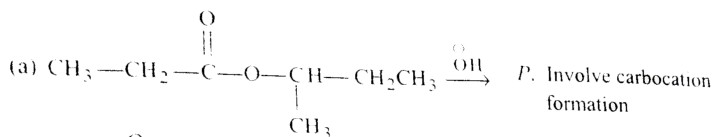
P. White turbidity with HCl/ZnCl_2

Q. Violet colour with FeCl_3

R. Colour change of $\text{Na}_2\text{Cr}_2\text{O}_7, \text{H}^\oplus$

S. $\text{I}_2/\text{O}^\oplus\text{H}$, gives bright yellow ppt.

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147.

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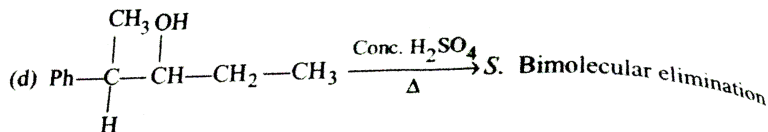
148.

- | | |
|-----------------------------------|--------------------------------------|
| (a) Fries rearrangement | (P) Acid catalysed rearrangement |
| (b) Claisen rearrangement | (Q) Concerned with ester |
| (c) Buyer-Viliger's rearrangement | (R) Involve electrophilic substitut. |
| (d) Pinacole-Pinacolone | (S) Intramolecular rearrangement |

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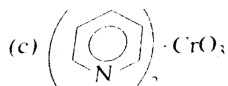
- (a) Fries rearrangement
 (b) Claisen rearrangement
 (c) Bayer-Villiger's rearrangement
 (d) Pinacole-Pinacolone rearrangement
- P.* Acid catalysed rearrangement
Q. Concerned with ester
R. Involve electrophilic substitution
S. Intramolecular rearrangement

149.



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(a) Oxidation of 1° alcohol in aldehyde



P. KMnO₄, Δ, OH

Q. Collin's reagent

R. Jones's reagent

S. PCC

150. (d) Oxidation of alkyne into acid

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151.

- (a) Identification of 1°, 2° and 3° Alcohol (*P*) Oxyme
 (b) Identification of 1°, 2° and 3° Nitroalkane (*Q*) Cu/3C
 (c) Formation of alcohol by anti-Markownikoff's addition (*R*) Victor
 (d) Formation of alcohol by Markownikoff's addition (*S*) Huydr



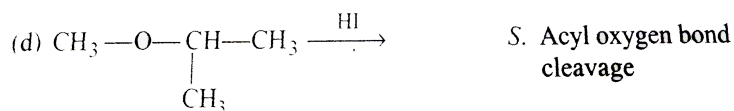
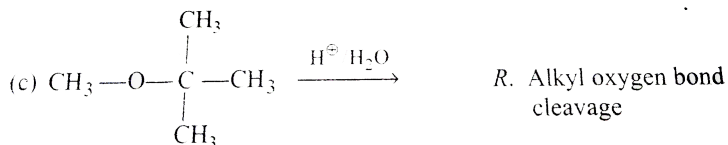
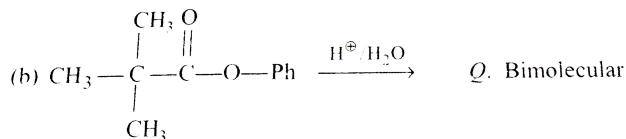
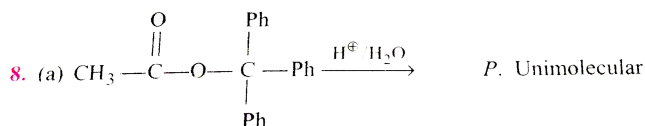
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152.

((a), Phenol+Neutral $FeCl_3$, (P), No reaction), ((b), Phenol Br_2 (aq.), (Q).
((d), Picric acid + $NaHCO_3$, (S), CO_2 gas is evolved)



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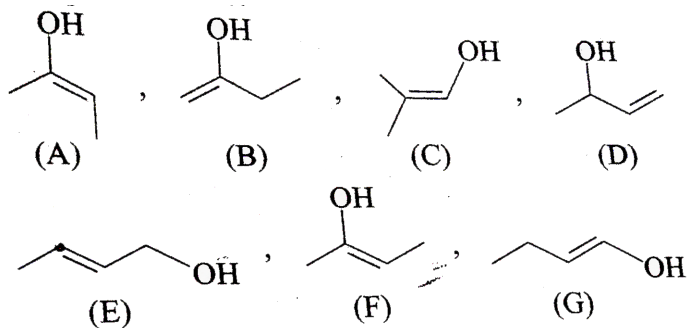


153.



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154. How many compounds A through G are enol tautomer of 2-butanone?



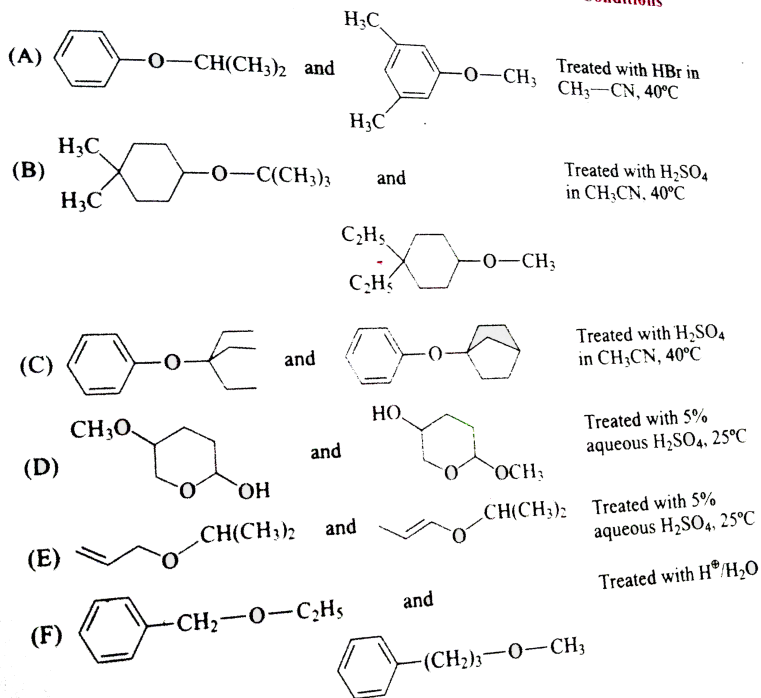
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155. Consider the pairs of ethers A through F shown below. To the right of each pair is a description of reaction conditions to be applied to each. One compound of the pair will react more rapidly than the other. Find out number of reactions in which first ether more rapidly cleaved than

second.

Ether pairs

Conditions



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156. Find out number of moles of HIO_4 that will react with following compound



f

CHOH

CHOH

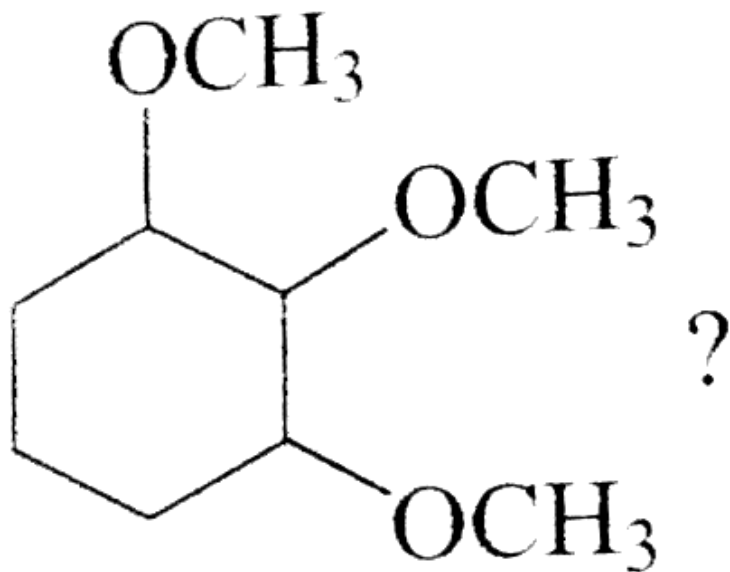
CHOH

CHOH

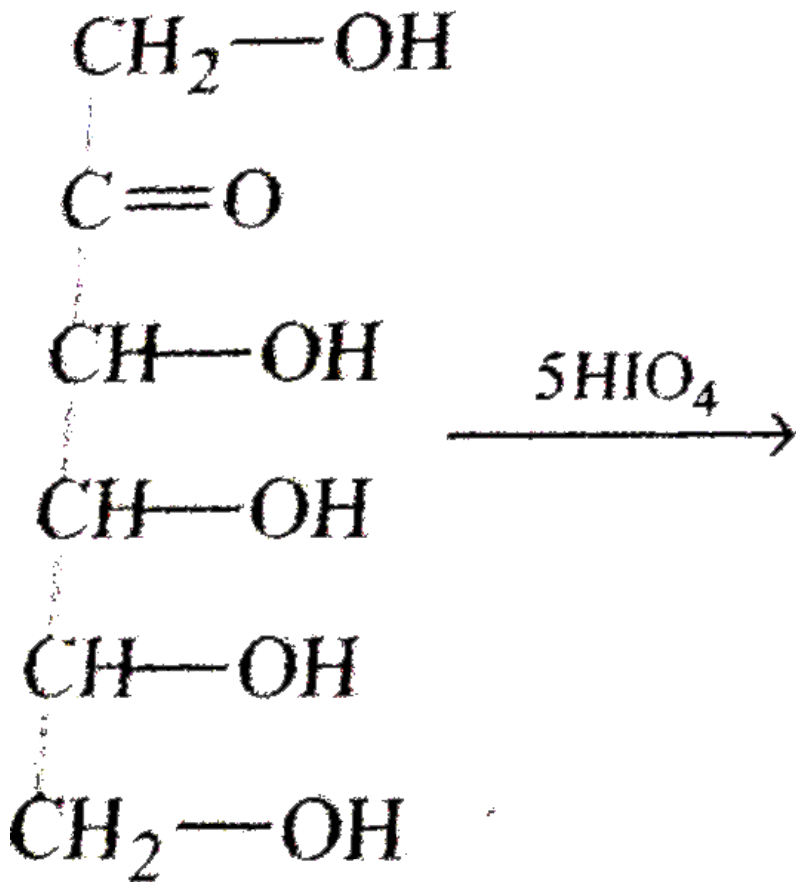
CH₂OH

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157. How many mole of 'H⁺' will react with

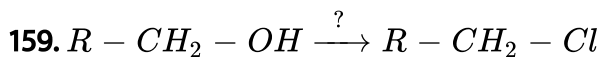


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158. _____, Find out the value of 'X'

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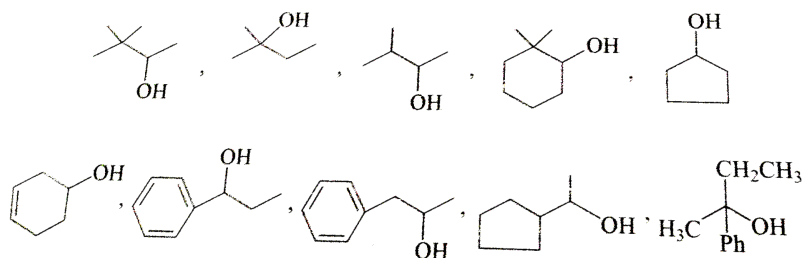
find out the number that can be used for above conversion, from the

following.

HCl , $ZnCl_2$, PCl_3 , PCl_5 , $POCl_3$, $SOCl_2$, $NaCl$, $TsCl$

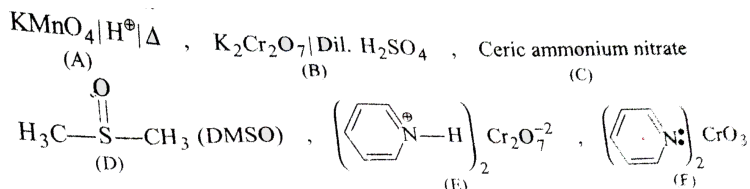
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160. Identify number of alcohols those will show rearrangement during dehydration will concentrate H_2SO_4 .



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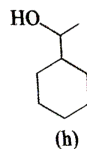
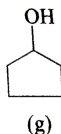
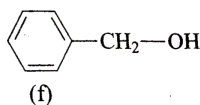
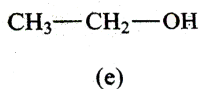
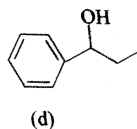
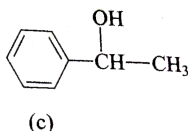
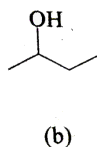
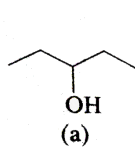
161. Find out number of reagents that converts 1(°) alcohols to



aldehyde.

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162. Find out number of alcohols that can give positive iodoform test.

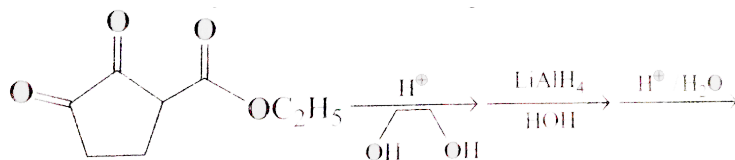


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163. How many mole of HI reacts with glycerol to give 2-iodopropane?

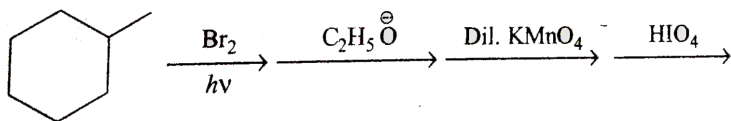
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164. Find out final product of following reactions:



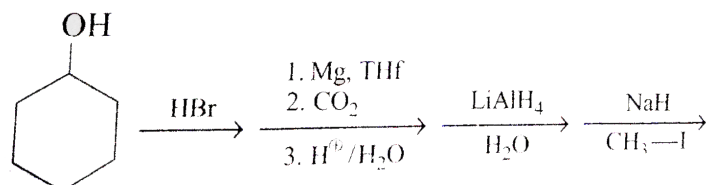
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165. Find out final product of following reactions:



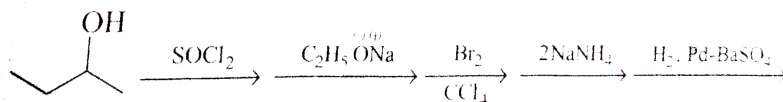
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166. Find out final product of following reactions:



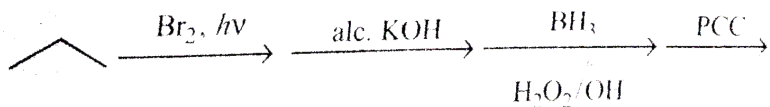
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167. Find out final product of following reactions:



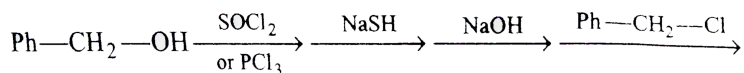
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168. Find out final product of following reactions:



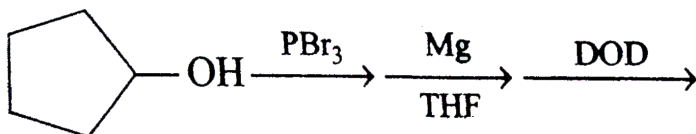
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169. Find out final product of following reactions:



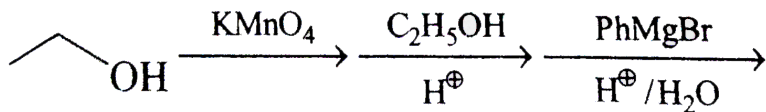
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170. Find out final product of following reactions:



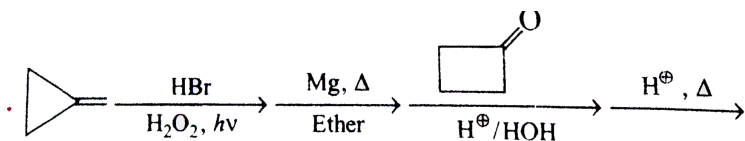
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171. Find out final product of following reactions:



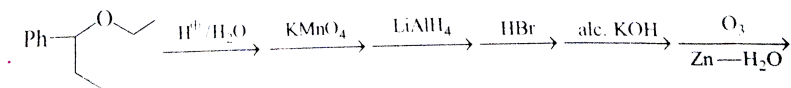
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172. Find out final product of following reactions:



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173. Find out final product of following reactions:



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