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## CHEMISTRY

## BOOKS - MS CHOUHAN CHEMISTRY <br> (HINGLISH)

## ALKENES AND ALKYNES I

## Solved Problem

1. The two stereoisomers of 1 - bromo-1, 2dichloroethene can - not be designated as cis and trans in the normal way because the double bond is trisubstituted. They can, however, be given ( E) and (Z)
designations. Write a structural formula for each isomer and give each the proper designation.

## - View Text Solution

2. Cosider the two alkenes 2-methhyl-1-pentene and 2-methyl-2-pentene and decide which would be most stable.

## - View Text Solution

3. Using Zaitsev's rule, predict which would be the major product of the following reaction:
4. Your task is the following synthesis. Which base would you use to maximum the yield of this specific alkene?

## - View Text Solution

5. Predict the major product formed when the following compound is subjected to dehydrochlorination with sodium ethoxide in ethanol.

## D View Text Solution

6. Explain why the major product of the dehydration above is 1,2-dimethylcyclohexene (as shown) and not

2,3-dimethyl -1-cyclohexene.

## - View Text Solution

7. As we shall soon see, sidium amide $\left(\mathrm{NaNH}_{2}\right)$ is
useful, especially when a reaction requires a very
strong base. Explain why a solvent such as methanol
cannnot be used to carry out a reaction in which you might want to use sodium amide as a base.

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8. Outline a synthesis of 4-phenyl-2-butyne from 1propyne.

## - View Text Solution

9. Outline a retrosynthetic pathway that leads from
'muscalure', the sex attrantant pheromone of the common housefly back to the simplest alkyne, ethyne
(acetylene). Then show the synthesis. You may use any inorgannic compounds. Or solvents, you need and alkyl halides of any length neccesary.

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10. The two stereoisomers of 1 - bromo-1, 2dichloroethene can - not be designated as cis and trans in the normal way because the double bond is trisubstituted. They can, however, be given ( E) and (Z) designations. Write a structural formula for each isomer and give each the proper designation.
11. Cosider the two alkenes 2-methhyl-1-pentene and 2-methyl-2-pentene and decide which would be most stable.

## - View Text Solution

12. Using Zaitsev's rule, predict which would be the major product of the following reaction:
13. Your task is the following synthesis. Which base would you use to maximum the yield of this specific alkene ?

## - View Text Solution

14. Predict the major product formed when the following compound is subjected to dehydrochlorination with sodium ethoxide in ethanol.
15. Explain why the major product of the dehydration above is 1,2 - dimethylcyclohexene (as shown) and not 2,3-dimethyl -1-cyclohexene.

## D View Text Solution

16. As we shall soon see, sidium amide $\left(\mathrm{NaNH}_{2}\right)$ is useful, especially when a reaction requires a very strong base. Explain why a solvent such as methanol cannnot be used to carry out a reaction in which you might want to use sodium amide as a base.
17. Outline a synthesis of 4-phenyl-2-butyne from 1propyne.

## - View Text Solution

18. Outline a retrosynthetic pathway that leads from
'muscalure', the sex attrantant pheromone of the common housefly back to the simplest alkyne, ethyne
(acetylene). Then show the synthesis. You may use any
inorgannic compounds. Or solvents, you need and alkyl halides of any length neccesary.

# Additional Objective Questions Single Correct Choice 

 Type1. Major product of the following reaction is

A. racemic mixture
B. diastereomer
C. meso compound
D. structural isomer

## Answer: C

## D View Text Solution

2. Select the major product from the following reaction sequence.


A. + enantiomer

B.

C. + enantiomer

D. + enantiomer

Answer: C

- View Text Solution

3. The following reaction proceeds by which mechanism?

A. $S_{N} 1$
B. $S_{N} 2$
C. $E_{2}$
D. $E_{1}$

## Answer: C

## - View Text Solution

4. Reagent used to carry out following conversion from alkyne to alkene is

A. $P d-C / H_{2}$
B. $\mathrm{Na} / \mathrm{NH}_{3}$
C. $\mathrm{Pt} / \mathrm{H}_{2}$
D. $\mathrm{Ni} / \mathrm{H}_{2}$

## Answer: B

## - View Text Solution

5. The reagent(s) for the following conversion is/are $\xrightarrow{\mathrm{Br}} \xrightarrow{?} \mathrm{HC} \equiv \mathrm{CH}$
A. alc. KOH

# B. alc. KOH followed by $\mathrm{NaNH}_{2}$ 

C. aq. KOH followed by $\mathrm{NaNH}_{2}$
D. $\mathrm{Zn} / \mathrm{CH}_{3} \mathrm{OH}$

## Answer: B

## - View Text Solution

6. Consider the following reaction:


Identify product $(\mathrm{P})$ and ( Q ) from the folowing

## compounds:


(R)

(S)
A. $P=R, Q=S$
B. $P=S, Q=S$
C. $P=S, Q=R$
D. $P=R, Q=R$

Answer: A
7. Choose the compound with the (S) chiral carbon and the $(Z)$ double bond configuration

B.

C.

D.


Answer: D
8. The synthesis of 3 -octyne is achieved by adding a bromoalkane into a mixture of sodium amide and an alkyne. The bromoalkane and alkyne, respectively, are
A. $\mathrm{BrCH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3}$ and $\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{C} \equiv \mathrm{CH}$
B. $\mathrm{BrCH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3}$ and $\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{C} \equiv \mathrm{CH}$
C. $\mathrm{BrCH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3}$ and $\mathrm{CH}_{3} \mathrm{C} \equiv \mathrm{CH}$
D. $\mathrm{BrCH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3}$ and $\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{C} \equiv \mathrm{CH}$

## Answer: D

## D View Text Solution



Products $(P)$ and $(Q)$ are


B.


D.


Answer: B
10. The following reaction proceeds by which mechanism?

A. $S_{N} 1$
B. $S_{N} 2$
C. $E_{2}$
D. $E_{1}$

## - View Text Solution

11. Consider the following reaction:


If ( P ) on heating isomerizes to $(\mathrm{Q})$. What is the
structure of (Q) ?



Answer: B

- View Text Solution

12. Choose the structure that has the name ( $\mathrm{R}, \mathrm{Z}$ )-2-bromo-4-mrthyl-2-phenylhex-4-ene.


Answer: D
13. Identify the final product in the reaction


D. Stilbene

Answer: B
14. Which of the following alkene is most stable?
A. $\mathrm{CH}_{3}-\mathrm{CH}_{2}-\mathrm{CH}=\mathrm{CH}_{2}$
B. $\mathrm{CH}_{3}-\mathrm{CH}-\mathrm{CH}=\mathrm{CH}_{3}$

D. ${ }^{\mathrm{CH}_{3}}{ }^{\mathrm{CH}} \mathrm{C}=\mathrm{C}-\underset{\mathrm{CH}_{3}}{-\mathrm{CH}_{3}}$

Answer: D

- View Text Solution

15. How many halogen atoms will be removed in the

## following $E_{2}$ reaction ?


A. 4
B. 6
C. 8
D. 10

Answer: B

- View Text Solution

16. What is the major product of the following reaction?
 $\xrightarrow[\text { Heat }]{\mathrm{H}_{2} \mathrm{SO}_{4}}$ ?
OH

A.
B.

C.
D.

## Answer: C

## - View Text Solution

17. The following reaction proceeds by which mechanism?

A. $S_{N} 1$
B. $S_{N} 2$
C. E2
D. E1

## - View Text Solution

18. Which of the following carbocation will undergo rearrangemment ?
A.
B.
C. $\mathrm{CH}_{3}-\underset{\substack{\mathrm{l} \\ C H_{3}}}{\mathrm{CH}}-\stackrel{+}{\mathrm{C}}=\mathrm{O}$
D. $\mathrm{CH}_{3}-\mathrm{NH}-\stackrel{+}{\mathrm{C}} \mathrm{H}-\underset{\substack{\mathrm{C} \\ \mathrm{CH}_{3}}}{\mathrm{CH}} \mathrm{H}-\mathrm{CH}_{3}$

Answer: B
19. The most reactive conformation for an E2 reaction is
A. syn periplanar
B. anti periplanar
C. gauche staggered
D. gauche eclipsed

## Answer: B

- View Text Solution

20. In the following sequence of reactions
$\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{Br} \xrightarrow{\mathrm{KOH}(\text { alc })}(A) \xrightarrow{\mathrm{HBr}}(B) \xrightarrow{\mathrm{KOH}(a q .)}(C)$,
The product ( $C$ ) is
A. propan-2-ol
B. propan-1-ol
C. propyne
D. propene

Answer: A
21. Which of the reactions can not be used to synthesize propyne?

$$
\begin{aligned}
& \text { A. } \mathrm{CH}_{3}-\mathrm{CH}=\mathrm{CH}-\mathrm{Br} \xrightarrow[t-\mathrm{ButOH}]{t-\mathrm{ButO}^{-} / \Delta} \\
& \text { B. } \mathrm{CH} \equiv \mathrm{CNa}+\mathrm{CH}_{3}-\mathrm{I} \rightarrow \\
& \text { C. } \mathrm{HC} \equiv \mathrm{CH}+\mathrm{CH}_{3} \mathrm{Na} \rightarrow \\
& \text { D. } \mathrm{CH}_{3}-\mathrm{CHBrCHBr} \xrightarrow[(2) H^{+}]{(1) \mathrm{NaNH}_{2}(\text { excess })}
\end{aligned}
$$

## Answer: C

## D View Text Solution

22. Consider the following compounds
(I) $\mathrm{CH}_{3}-\stackrel{\mathrm{CH}_{3}^{\mathrm{I}} \mathrm{C}}{\substack{\mathrm{I} \\ \mathrm{CH}_{3}}}-\mathrm{Br}$
(II) $\mathrm{CH}_{3}-\mathrm{C}-\mathrm{Br}$

$$
{ }_{C H_{3}}
$$

(III) $\mathrm{CH}_{3}-\mathrm{CH}_{2}-\mathrm{CH}_{2} \mathrm{Br}$

Their reactivity toward E1 is
A. I gt II gt III
B. II gt I gt III
C. II gt III gt I
D. III gt II gt I

Answer: A
23. Major product of the following reaction is

A. racemic mixture
B. diastereomer
C. meso compound
D. structural isomer

## Answer: C

24. Select the major product from the following reaction sequence.


A. + enantiomer

B.

C.

D. + enantiomer

## Answer: C

## - View Text Solution

25. The following reaction proceeds by which mechanism?

A. $S_{N} 1$
B. $S_{N} 2$
C. $E_{2}$
D. $E_{1}$

Answer: C

## - View Text Solution

26. Reagent used to carry out following conversion from alkyne to alkene is

A. $P d-C / H_{2}$
B. $\mathrm{Na} / \mathrm{NH}_{3}$
C. $\mathrm{Pt} / \mathrm{H}_{2}$
D. $\mathrm{Ni} / \mathrm{H}_{2}$

## Answer: B

## D View Text Solution

27. The reagent(s) for the following conversion is/are $\xrightarrow{\text { Br }} \xrightarrow{?} \mathrm{HC} \equiv \mathrm{CH}$
A. alc. KOH
B. alc. KOH followed by $\mathrm{NaNH}_{2}$
C. aq. KOH followed by $\mathrm{NaNH}_{2}$
D. $\mathrm{Zn} / \mathrm{CH}_{3} \mathrm{OH}$

## Answer: B

## - View Text Solution

28. Consider the following reaction:


Identify product ( P ) and ( Q ) from the folowing

## compounds:


(R)

(S)
A. $P=R, Q=S$
B. $P=S, Q=S$
C. $P=S, Q=R$
D. $P=R, Q=R$

Answer: A
29. Choose the compound with the (S) chiral carbon and the $(Z)$ double bond configuration

B.

C.

D.


Answer: D
30. The synthesis of 3 -octyne is achieved by adding a bromoalkane into a mixture of sodium amide and an alkyne. The bromoalkane and alkyne, respectively, are
A. $\mathrm{BrCH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3}$ and $\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{C} \equiv \mathrm{CH}$
B. $\mathrm{BrCH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3}$ and $\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{C} \equiv \mathrm{CH}$
C. $\mathrm{BrCH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3}$ and $\mathrm{CH}_{3} \mathrm{C} \equiv \mathrm{CH}$
D. $\mathrm{BrCH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3}$ and $\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{C} \equiv \mathrm{CH}$

## Answer: D

## D View Text Solution

31. 



Products (P) and(Q) are



B.


D.


Answer: B
32. The following reaction proceeds by which mechanism?

A. $S_{N} 1$
B. $S_{N} 2$
C. $E_{2}$
D. $E_{1}$

## - View Text Solution

33. Consider the following reaction:


If (P) on heating isomerizes to (Q). What is the
structure of (Q) ?

$\mathrm{CH}_{3}$
B.

C.

D.

Answer: B

- View Text Solution

34. Choose the structure that has the name ( $\mathrm{R}, \mathrm{Z}$ )-2-bromo-4-mrthyl-2-phenylhex-4-ene.


Answer: D
35. Identify the final product in the reaction


D. Stilbene

Answer: B
36. Which of the following alkene is most stable ?
A. $\mathrm{CH}_{3}-\mathrm{CH}_{2}-\mathrm{CH}=\mathrm{CH}_{2}$
B. $\mathrm{CH}_{3}-\mathrm{CH}-\mathrm{CH}=\mathrm{CH}_{3}$



Answer: D

- View Text Solution

37. How many halogen atoms will be removed in the

## following $E_{2}$ reaction ?


A. 4
B. 6
C. 8
D. 10

## Answer: B

38. What is the major product of the following reaction ?
 $\xrightarrow[\text { Heat }]{\mathrm{H}_{2} \mathrm{SO}_{4}}$ ?
OH

A.
B.

C.
D.

## Answer: C

## - View Text Solution

39. The following reaction proceeds by which mechanism?

A. $S_{N} 1$
B. $S_{N} 2$
C. E2
D. E1

## Answer: C

## D View Text Solution

40. Which of the following carbocation will undergo
rearrangemment ?
A.
B.
C. $\mathrm{CH}_{3}-\underset{\substack{\mathrm{C} \\ C H_{3}}}{\mathrm{CH}}-\stackrel{+}{\mathrm{C}}=\mathrm{O}$
D. $\mathrm{CH}_{3}-\mathrm{NH}-\stackrel{+}{\mathrm{C}} \mathrm{H}-\underset{\substack{\mathrm{C} \\ \mathrm{CH}_{3}}}{\mathrm{CH}} \mathrm{H}-\mathrm{CH}_{3}$

Answer: B
41. The most reactive conformation for an E2 reaction is
A. syn periplanar
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## Answer: B

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42. In the following sequence of reactions
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Answer: A
43. Which of the reactions can not be used to synthesize propyne?

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& \text { A. } \mathrm{CH}_{3}-\mathrm{CH}=\mathrm{CH}-\mathrm{Br} \xrightarrow[t-\mathrm{ButOH}]{t-\mathrm{ButO}^{-} / \Delta} \\
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& \text { D. } \mathrm{CH}_{3}-\mathrm{CHBrCHBr} \xrightarrow[(2) H^{+}]{(1) \mathrm{NaNH}_{2}(\text { excess })}
\end{aligned}
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## Answer: C

## D View Text Solution

44. Consider the following compounds
(I) $\mathrm{CH}_{3}-\stackrel{\mathrm{CH}_{3}}{\substack{\mathrm{I} \\ \mathrm{CH}_{3}}}-\mathrm{Br}$
(II) $\mathrm{CH}_{3}-\mathrm{C}-\mathrm{Br}$

$$
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(III) $\mathrm{CH}_{3}-\mathrm{CH}_{2}-\mathrm{CH}_{2} \mathrm{Br}$

Their reactivity toward E1 is
A. I gt II gt III
B. II gt I gt III
C. II gt III gt I
D. III gt II gt I

Answer: A

## Additional Objective Questions Multiple Correct Choice

 Type1. Which of the following alkene has the $Z$ configuration along the double bond ?
A.
B.

C.

R
D.

Answer: A::C
2. Which of the following alkene has the $Z$ configuration along the double bond ?
A.

8
B.

A
C. d
D.

## Answer: A::C

