



### **CHEMISTRY**

# BOOKS - MS CHOUHAN CHEMISTRY (HINGLISH)

## **ALKENES AND ALKYNES I**

**Solved Problem** 

**1.** The two stereoisomers of 1 - bromo-1, 2dichloroethene can - not be designated as cis and trans in the normal way because the double bond is trisubstituted. They can, however, be given (E) and (Z) designations. Write a structural formula for each

isomer and give each the proper designation.

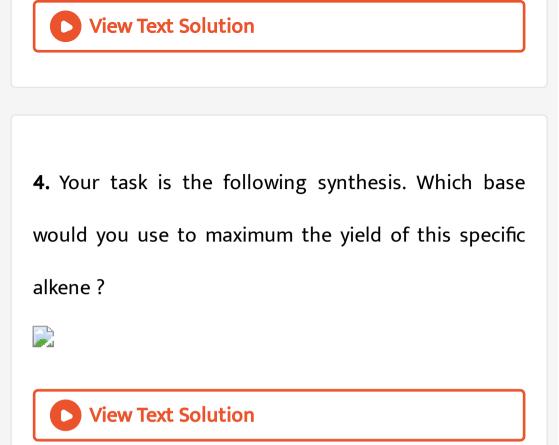
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**2.** Cosider the two alkenes 2-methhyl-1-pentene and 2methyl-2-pentene and decide which would be most stable.



**3.** Using Zaitsev's rule, predict which would be the major product of the following reaction:





5. Predict the major product formed when the following compound is subjected to dehydrochlorination with sodium ethoxide in ethanol.







**6.** Explain why the major product of the dehydration above is 1,2 - dimethylcyclohexene (as shown) and not

2,3 - dimethyl -1-cyclohexene.





7. As we shall soon see, sidium amide  $(NaNH_2)$  is useful, especially when a reaction requires a very strong base. Explain why a solvent such as methanol cannnot be used to carry out a reaction in which you

might want to use sodium amide as a base.

**View Text Solution** 8. Outline a synthesis of 4-phenyl-2-butyne from 1propyne. **View Text Solution** 

**9.** Outline a retrosynthetic pathway that leads from 'muscalure', the sex attrantant pheromone of the common housefly back to the simplest alkyne, ethyne

(acetylene). Then show the synthesis. You may use any inorgannic compounds. Or solvents, you need and alkyl halides of any length neccesary.



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**10.** The two stereoisomers of 1 - bromo-1, 2dichloroethene can - not be designated as cis and trans in the normal way because the double bond is trisubstituted. They can, however, be given (E) and (Z) designations. Write a structural formula for each isomer and give each the proper designation.



**11.** Cosider the two alkenes 2-methhyl-1-pentene and 2methyl-2-pentene and decide which would be most

stable.



**12.** Using Zaitsev's rule, predict which would be the major product of the following reaction:



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**13.** Your task is the following synthesis. Which base would you use to maximum the yield of this specific alkene ?



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14. Predict the major product formed when the following compound is subjected to dehydrochlorination with sodium ethoxide in ethanol.





15. Explain why the major product of the dehydration

above is 1,2 - dimethylcyclohexene (as shown) and not

2,3 - dimethyl -1-cyclohexene.





**16.** As we shall soon see, sidium amide  $(NaNH_2)$  is useful, especially when a reaction requires a very strong base. Explain why a solvent such as methanol cannnot be used to carry out a reaction in which you might want to use sodium amide as a base.



17. Outline a synthesis of 4-phenyl-2-butyne from 1-propyne.

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**18.** Outline a retrosynthetic pathway that leads from 'muscalure', the sex attrantant pheromone of the common housefly back to the simplest alkyne, ethyne (acetylene). Then show the synthesis. You may use any inorgannic compounds. Or solvents, you need and alkyl halides of any length neccesary.

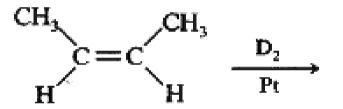






Additional Objective Questions Single Correct Choice Type

1. Major product of the following reaction is



A. racemic mixture

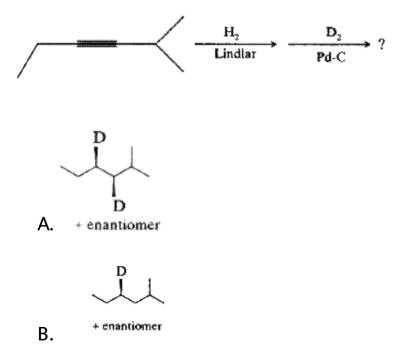
B. diastereomer

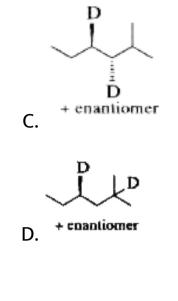
C. meso compound

D. structural isomer

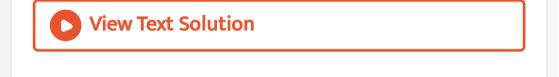
# Answer: C View Text Solution

**2.** Select the major product from the following reaction sequence.



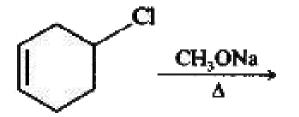


#### Answer: C



#### 3. The following reaction proceeds by which

#### mechanism?



A.  $S_N 1$ 

B.  $S_N 2$ 

 $\mathsf{C.}\,E_2$ 

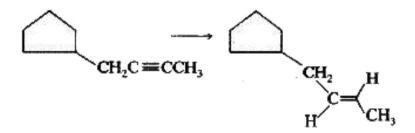
D.  $E_1$ 

#### Answer: C



4. Reagent used to carry out following conversion

from alkyne to alkene is



A.  $Pd-C/H_2$ 

B.  $Na/NH_3$ 

 $\mathsf{C.}\,Pt\,/\,H_2$ 

D.  $Ni/H_2$ 

**Answer: B** 

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5. The reagent(s) for the following conversion is/are



A. alc. KOH

B. alc. KOH followed by  $NaNH_2$ 

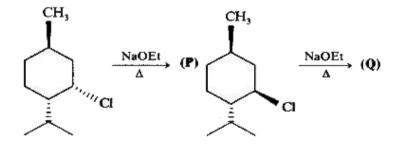
C. aq. KOH followed by  $NaNH_2$ 

D.  $Zn/CH_3OH$ 

#### Answer: B

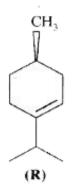


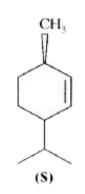
6. Consider the following reaction:



Identify product (P) and (Q) from the folowing

#### compounds:



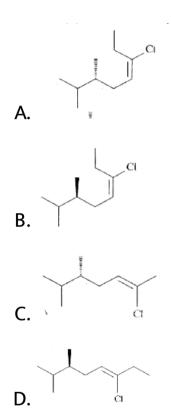


#### Answer: A



7. Choose the compound with the (S) chiral carbon

and the (Z) double bond configuration



#### Answer: D

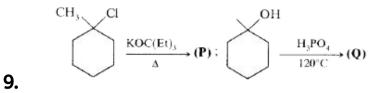


8. The synthesis of 3-octyne is achieved by adding a bromoalkane into a mixture of sodium amide and an alkyne. The bromoalkane and alkyne, respectively, are

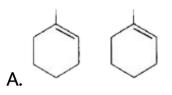
A.  $BrCH_2CH_2CH_2CH_3$  and  $CH_3CH_2C \equiv CH$ B.  $BrCH_2CH_2CH_3$  and  $CH_3CH_2CH_2C \equiv CH$ C.  $BrCH_2CH_2CH_2CH_2CH_3$  and  $CH_3C \equiv CH$ D.  $BrCH_2CH_2CH_2CH_3$  and  $CH_3CH_2C \equiv CH$ 

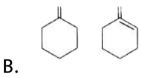
Answer: D

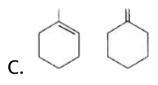
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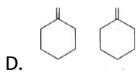


Products (P) and(Q) are



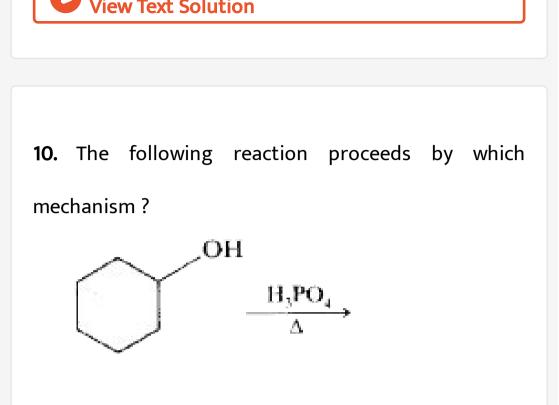






#### **Answer: B**



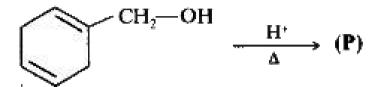


- A.  $S_N 1$
- B.  $S_N 2$
- $\mathsf{C}. E_2$
- D.  $E_1$

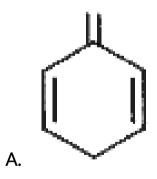
#### Answer: D

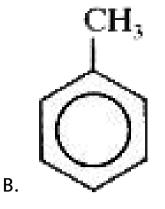


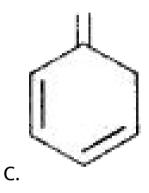
**11.** Consider the following reaction:

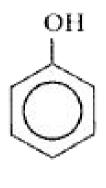


If (P) on heating isomerizes to (Q). What is the structure of (Q) ?







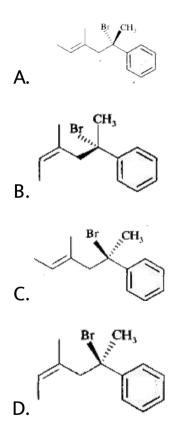


#### Answer: B

D.



**12.** Choose the structure that has the name (R, Z)-2bromo-4-mrthyl-2-phenylhex-4-ene.

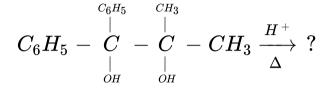


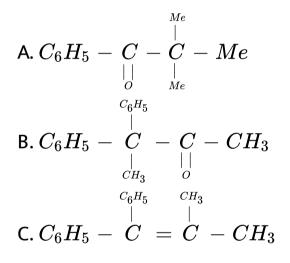
#### **Answer: D**





13. Identify the final product in the reaction





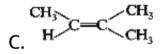
D. Stilbene

#### Answer: B

14. Which of the following alkene is most stable ?

A. 
$$CH_3-CH_2-CH=CH_2$$

$$\mathsf{B}.\,CH_3-CH-CH=CH_3$$

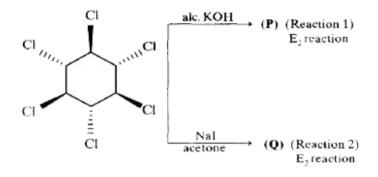


#### Answer: D



15. How many halogen atoms will be removed in the

following  $E_2$  reaction ?



A. 4

B. 6

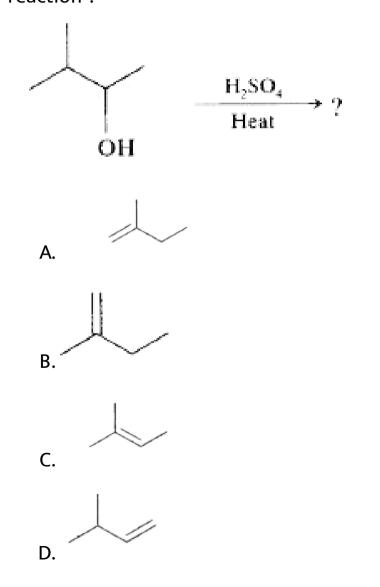
C. 8

D. 10

#### **Answer: B**

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**16.** What is the major product of the following reaction ?

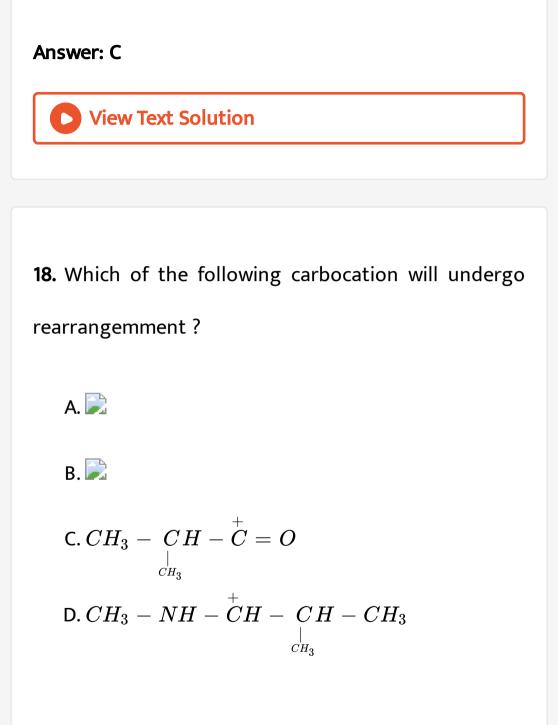


Answer: C		
View Text Solution		
17. The following reaction proceeds by which		
mechanism ?		
A. $S_N 1$		

B.  $S_N 2$ 

C. E2

D. E1



#### **Answer: B**





#### 19. The most reactive conformation for an E2 reaction

is

- A. syn periplanar
- B. anti periplanar
- C. gauche staggered
- D. gauche eclipsed

#### Answer: B



20. In the following sequence of reactions

 $CH_3CH_2CH_2Br \xrightarrow{KOH\,(\,alc\,)} (A) \xrightarrow{HBr} (B) \xrightarrow{KOH\,(\,aq\,.\,)} (C),$ 

The product (C) is

A. propan-2-ol

B. propan-1-ol

C. propyne

D. propene

Answer: A



**21.** Which of the reactions can not be used to synthesize propyne ?

A. 
$$CH_3 - CH = CH - Br \xrightarrow[t-ButOH]{t-ButOH} X$$

B.  $CH \equiv CNa + CH_3 - I 
ightarrow$ 

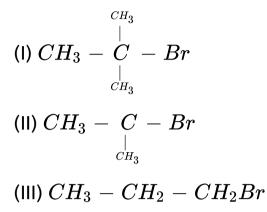
C.  $HC \equiv CH + CH_3Na 
ightarrow$ 

$$\mathsf{D.}\,CH_3 - CHBrCHBr \xrightarrow{(1)\,NaNH_2(\,excess\,)}{(2)\,H^+}$$

Answer: C



#### 22. Consider the following compounds



Their reactivity toward E1 is

A. I gt II gt III

B. II gt I gt III

C. II gt III gt I

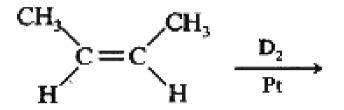
D. III gt II gt I

#### **Answer: A**





23. Major product of the following reaction is



A. racemic mixture

B. diastereomer

C. meso compound

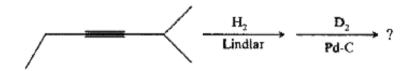
D. structural isomer

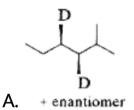
Answer: C



24. Select the major product from the following

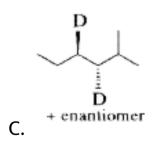
reaction sequence.

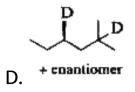




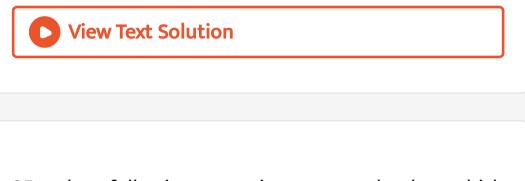


B. + enantiomer

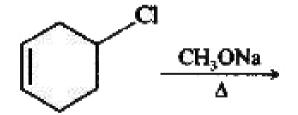




## Answer: C



**25.** The following reaction proceeds by which mechanism ?



A.  $S_N 1$ 

 $\mathsf{C}.\,E_2$ 

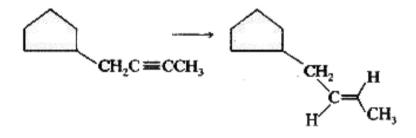
D.  $E_1$ 

## Answer: C



26. Reagent used to carry out following conversion

from alkyne to alkene is

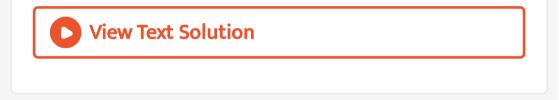


B.  $Na/NH_3$ 

 $\mathsf{C}. Pt/H_2$ 

D.  $Ni/H_2$ 

Answer: B



27. The reagent(s) for the following conversion is/are



A. alc. KOH

B. alc. KOH followed by  $NaNH_2$ 

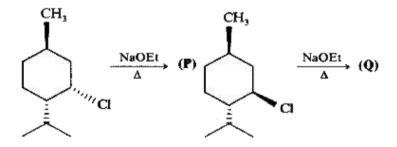
C. aq. KOH followed by  $NaNH_2$ 

D.  $Zn/CH_3OH$ 

## Answer: B

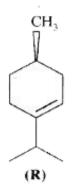


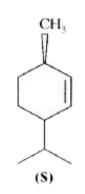
**28.** Consider the following reaction:



Identify product (P) and (Q) from the folowing

# compounds:



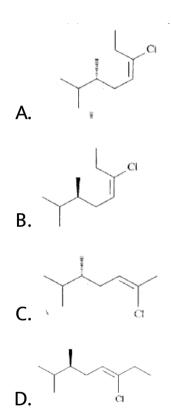


## Answer: A



**29.** Choose the compound with the (S) chiral carbon

and the (Z) double bond configuration



### Answer: D

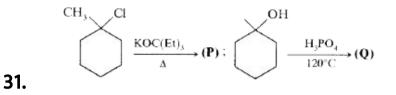


**30.** The synthesis of 3-octyne is achieved by adding a bromoalkane into a mixture of sodium amide and an alkyne. The bromoalkane and alkyne, respectively, are

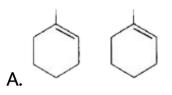
A.  $BrCH_2CH_2CH_2CH_3$  and  $CH_3CH_2C \equiv CH$ B.  $BrCH_2CH_2CH_3$  and  $CH_3CH_2CH_2C \equiv CH$ C.  $BrCH_2CH_2CH_2CH_2CH_3$  and  $CH_3C \equiv CH$ D.  $BrCH_2CH_2CH_2CH_3$  and  $CH_3CH_2C \equiv CH$ 

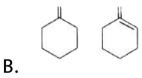
Answer: D

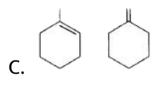
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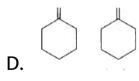


Products (P) and(Q) are



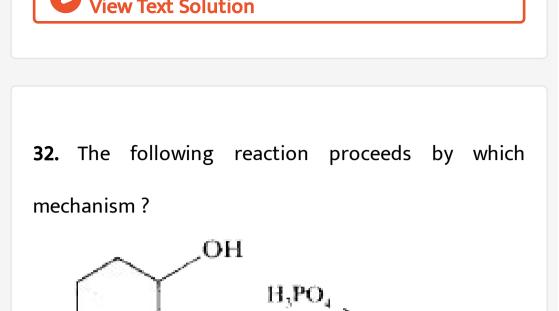






## **Answer: B**



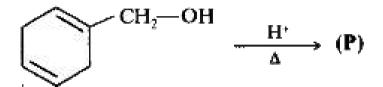


- A.  $S_N 1$
- B.  $S_N 2$
- $\mathsf{C}. E_2$
- D.  $E_1$

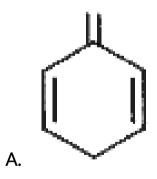
## Answer: D

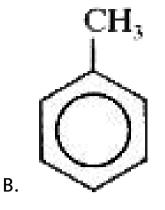


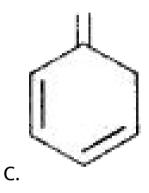
**33.** Consider the following reaction:

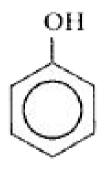


If (P) on heating isomerizes to (Q). What is the structure of (Q) ?







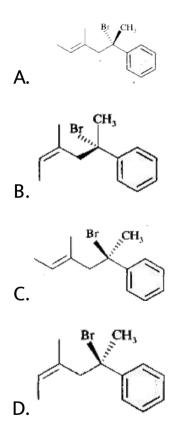


## Answer: B

D.



**34.** Choose the structure that has the name (R, Z)-2bromo-4-mrthyl-2-phenylhex-4-ene.

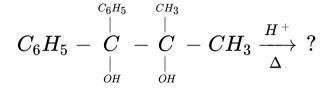


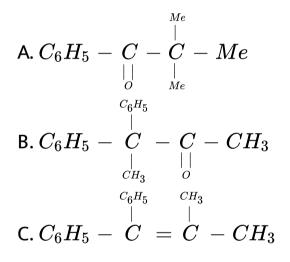
#### **Answer: D**





35. Identify the final product in the reaction





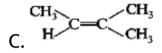
D. Stilbene

#### Answer: B

36. Which of the following alkene is most stable ?

A. 
$$CH_3 - CH_2 - CH = CH_2$$

$$\mathsf{B}.\,CH_3-CH-CH=CH_3$$

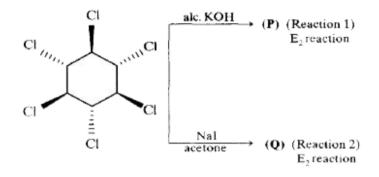


## Answer: D



37. How many halogen atoms will be removed in the

following  $E_2$  reaction ?



A. 4

B. 6

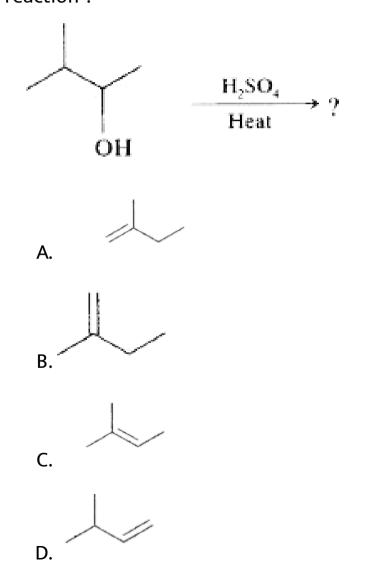
C. 8

D. 10

#### **Answer: B**

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**38.** What is the major product of the following reaction ?

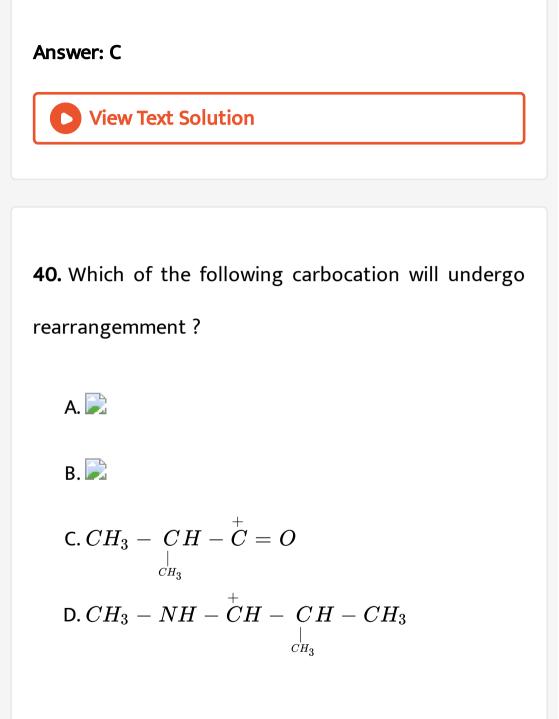


Answer: C
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<b>39.</b> The following reaction proceeds by which
mechanism ?
CI alc. KOH
A. $S_N 1$

B.  $S_N 2$ 

C. E2

D. E1



#### Answer: B





## 41. The most reactive conformation for an E2 reaction

is

- A. syn periplanar
- B. anti periplanar
- C. gauche staggered
- D. gauche eclipsed

## Answer: B



42. In the following sequence of reactions

 $CH_3CH_2CH_2Br \xrightarrow{KOH\,(\,alc\,)} (A) \xrightarrow{HBr} (B) \xrightarrow{KOH\,(\,aq\,.\,)} (C),$ 

The product (C) is

A. propan-2-ol

B. propan-1-ol

C. propyne

D. propene

Answer: A



**43.** Which of the reactions can not be used to synthesize propyne ?

A. 
$$CH_3 - CH = CH - Br \xrightarrow[t-ButOH]{t-ButOH} X$$

B.  $CH \equiv CNa + CH_3 - I 
ightarrow$ 

C.  $HC \equiv CH + CH_3Na 
ightarrow$ 

$$\mathsf{D.}\,CH_3 - CHBrCHBr \xrightarrow{(1)\,NaNH_2(\,excess\,)}{(2)\,H^+}$$

Answer: C



44. Consider the following compounds

(I) 
$$CH_3 - \overset{CH_3}{\overset{C}{\underset{CH_3}{CH_3}} - Br$$
  
(II)  $CH_3 - \overset{C}{\underset{CH_3}{CH_3}} - Br$   
(III)  $CH_3 - CH_2 - Br$ 

Their reactivity toward E1 is

A. I gt II gt III

B. II gt I gt III

C. II gt III gt I

D. III gt II gt I

### **Answer: A**





Additional Objective Questions Multiple Correct Choice Type

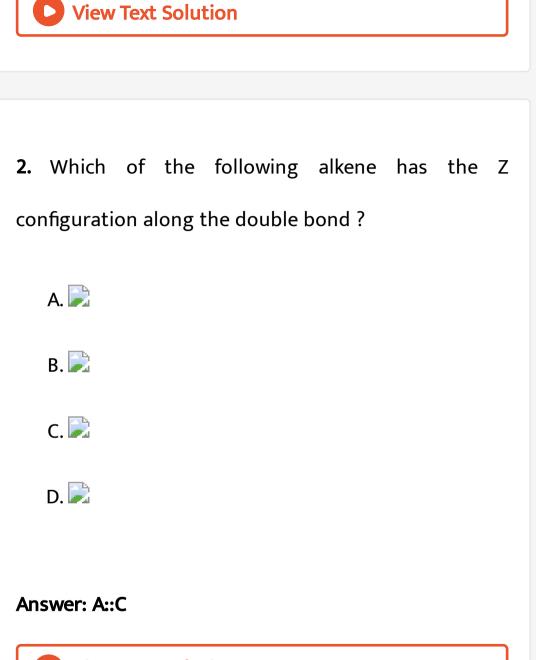
**1.** Which of the following alkene has the Z configuration along the double bond ?





## Answer: A::C





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