



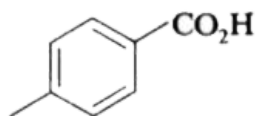
# CHEMISTRY

## BOOKS - MS CHOUHAN CHEMISTRY (HINGLISH)

### CARBOXYLIC ACIDS AND THEIR DERIVATIVES

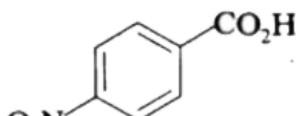
**Solved Problem**

1. Which carboxylic acid would you expect to be stronger A or B ?



**A**

or



**B**



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2. Suggest explanation for the following

The  $pK_{a1}$  for all of the dicarboxylic acids in Table 17.2 is smaller than the  $pK_a$  for a

monocarboxylic acid with the same number of carbon atoms.



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3. Suggest explanation for the following

The difference between  $pK_{a1}$  and  $pK_{a2}$  for dicarboxylic acid of the type  $HO_2C(CH_2)_nCO_2H$  decreases as  $n$  increases.



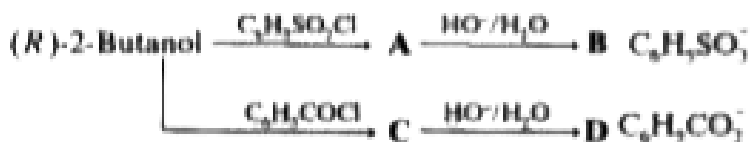
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4. N,N-Diethyl-3-methylbenzamide (also called N,N-diethyl-m-toluamide. Or DEET) is used many insect repellants. Write its structure.



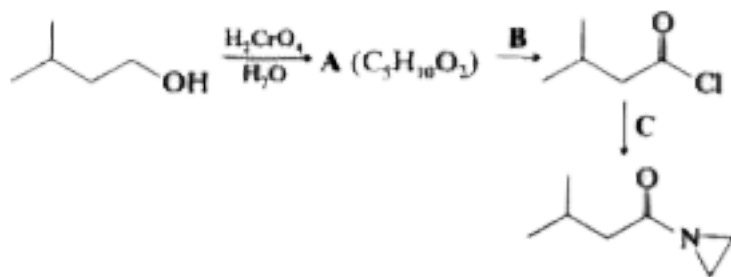
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5. Give stereochemical formulas for A-D. are enantiomers of each of other.



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6. Provide the missing compound A-C, in the following synthesis



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7. At first glance the conversion of bromobenzene to benzonitrile looks simple just carry out a nucleophilic substitution using

cyanide ion as the nucleophile. Then we remember that bromobenzene does not undergo either an  $S_N1$  or an  $S_N2$  reaction (Section 6.14A). The conversion can be accomplished, however, though it involves several steps. Outline possible steps.



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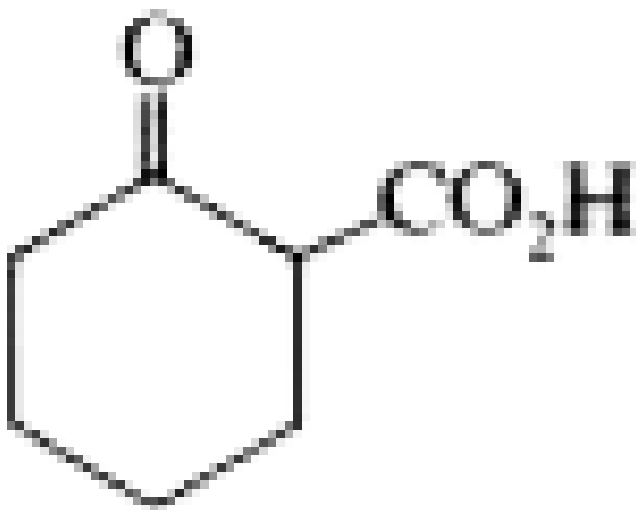
**8. Provide structures for A and B.**





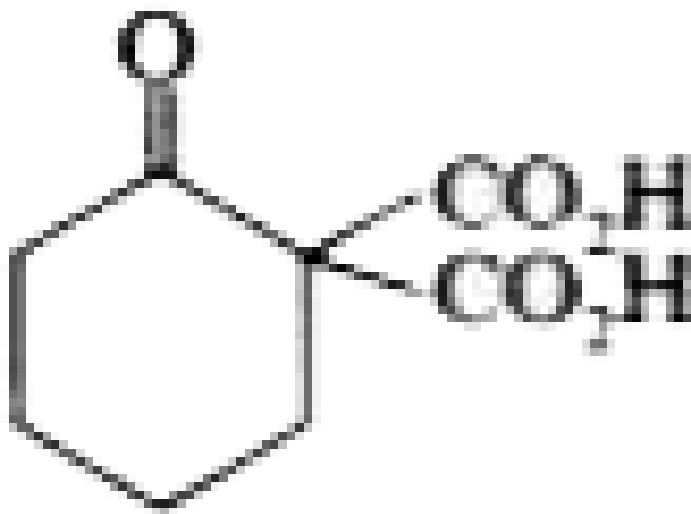
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9. Identify the product obtained when following acids, undergo prolonged heating,



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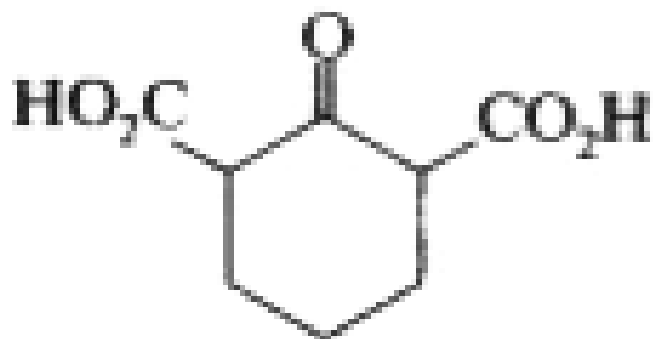
10. Identify the product obtained when following acids, undergo prolonged heating,



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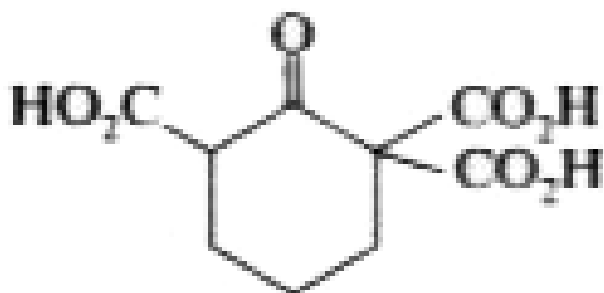
11. Identify the product obtained when following acids, undergo prolonged heating,





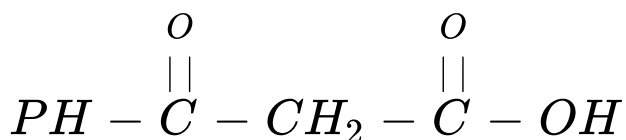
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12. Identify the product obtained when following acids, undergo prolonged heating,

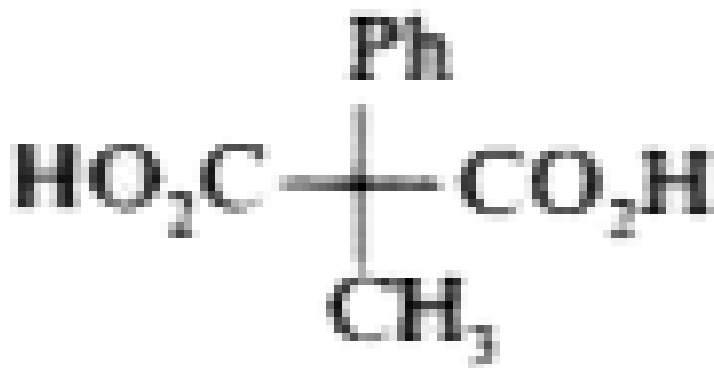


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13. Identify the product obtained when following acids, undergo prolonged heating,

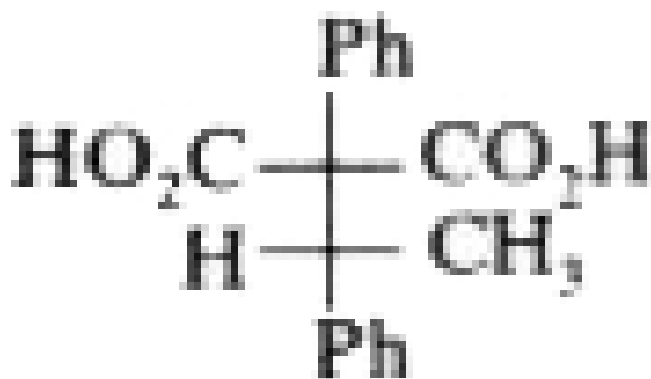
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14. Identify the product obtained when following acids, undergo prolonged heating,



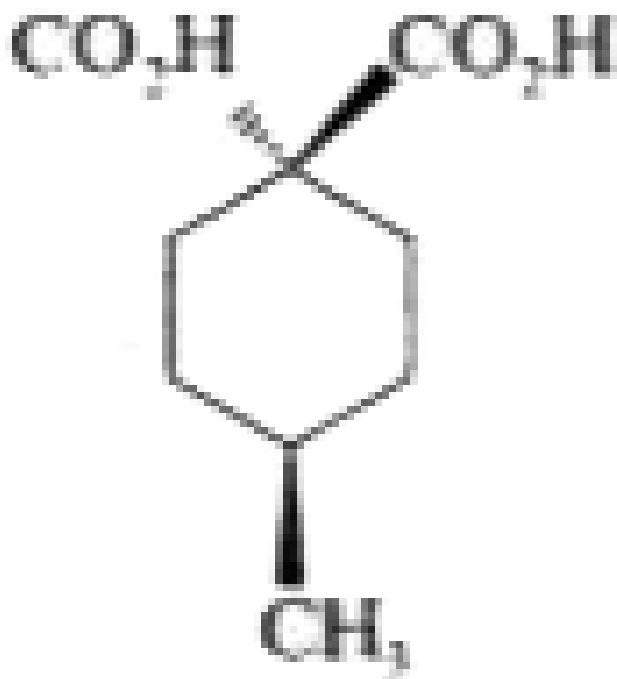
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15. Identify the product obtained when following acids, undergo prolonged heating,



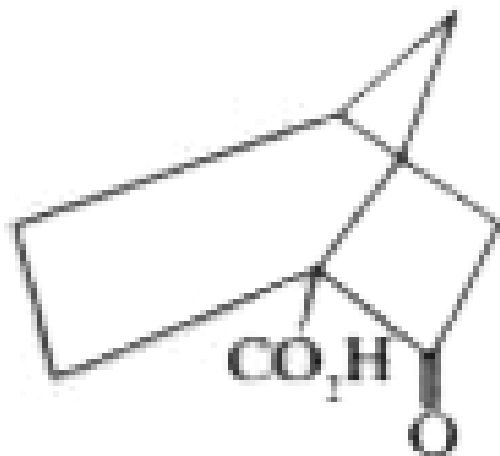
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16. Identify the product obtained when following acids, undergo prolonged heating,



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17. Following  $\beta$  ketoacid will not undergo decarboxylation on heating Give reasons.



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**18.** Provided detailed curved arrow mechanisms for the following reactions:



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19. Provided detailed curved arrow mechanisms for the following reactions:



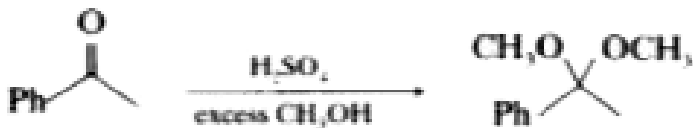
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20. Provided detailed curved arrow mechanisms for the following reactions:



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21. Provided detailed curved arrow mechanisms for the following reactions:



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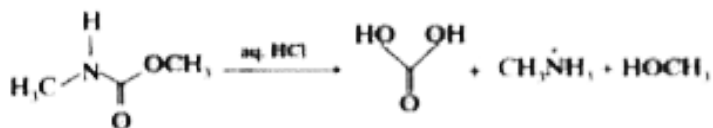


22. Provided detailed curved arrow mechanisms for the following reactions:



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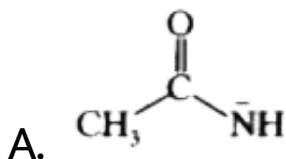
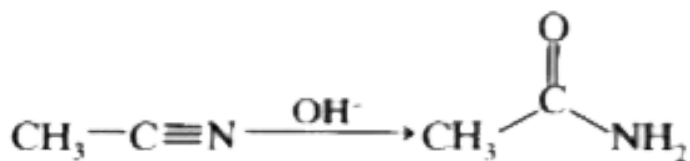
23. Provided detailed curved arrow mechanisms for the following reactions:

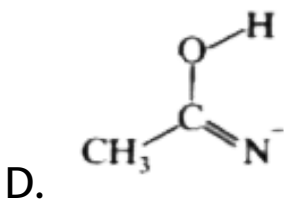
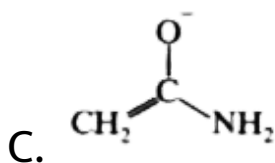
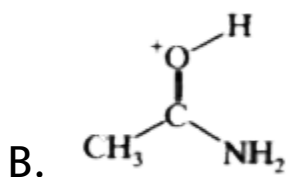


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## Additional Objective Questions Single Correct Choice Type

1. Choose the structure that is an intermediate in the base catalyzed hydrolysis of acetonitrile into acetamide.

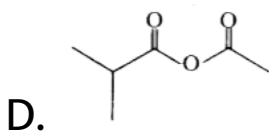
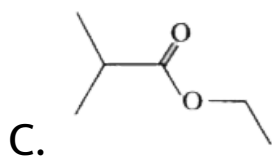
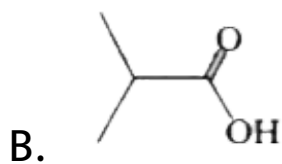
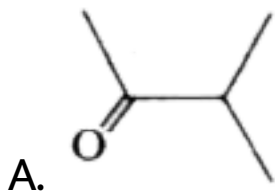
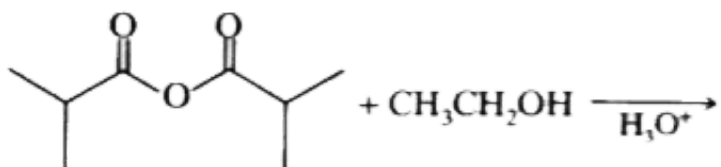




**Answer: D**

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2. What is the major product of the following reaction?

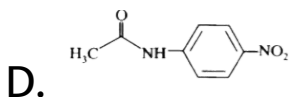
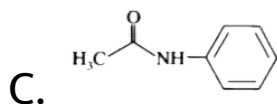
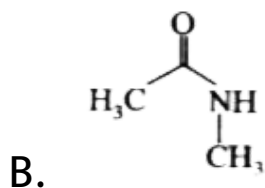
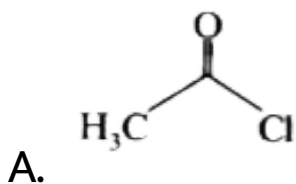


**Answer: C**



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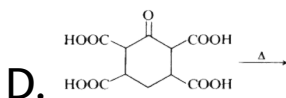
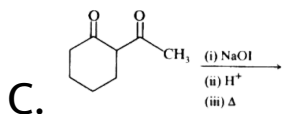
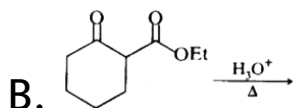
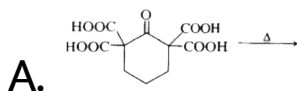
3. Which of the following are most reactive compounds for nucleophilic acyl substitution?



**Answer: D**



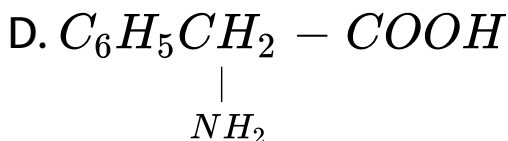
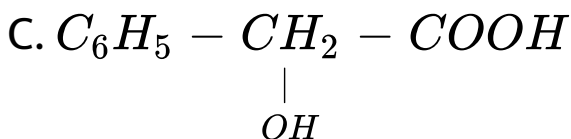
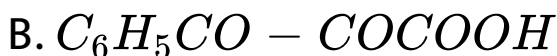
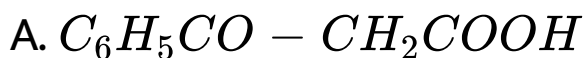
4. Which of the following reaction do not yield cyclohexanone as major product?



Answer: D

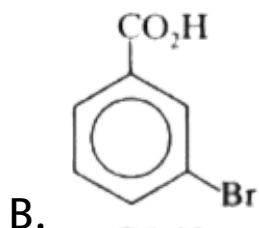
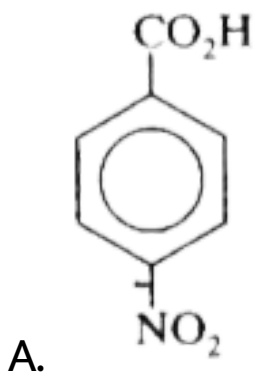
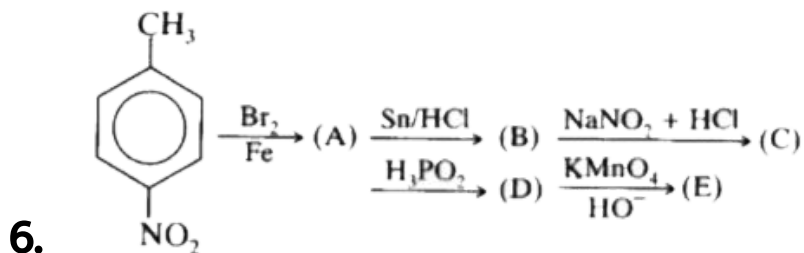


5. Which of the following carboxylic acids undergo decarboxylation easily?

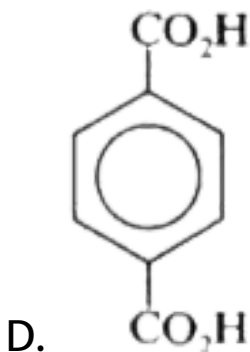
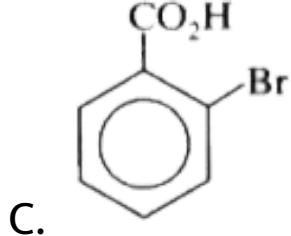


**Answer: A**









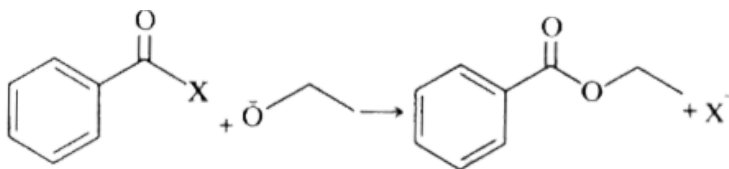
**Answer: C**



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7. Choose the order that has the following benzoyl compounds correctly arranged with

respect to increasing reactivity.



A.  $(i) < (ii) < (iii)$

B.  $(i) < (iii) < (ii)$

C.  $(ii) < (i) < (iii)$

D.  $(i) < (iii) < (i)$

**Answer: B**



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8. When propionic acid is treated with aqueous  $NaHCO_3$ ,  $CO_2$  is liberated. The .C. of  $CO_2$  comes from

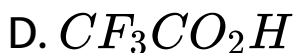
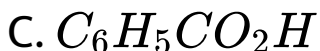
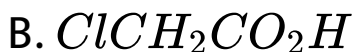
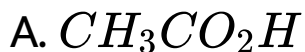
- A. methyl group
- B. carboxylic acid group
- C. methylene group
- D. bicarbonate.

**Answer: D**



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9. Which of the following carboxylic acids will have the largest  $K_a$  value ?



**Answer: D**



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**10.** The reactivity of carboxylic acid derivatives from highest reactivity to lowest reactivity is

A. acid anhydride > acid chloride > ester > amide

B. acid chloride > acid anhydride > ester > amide

C. acid anhydride > ester > amide > acid chloride

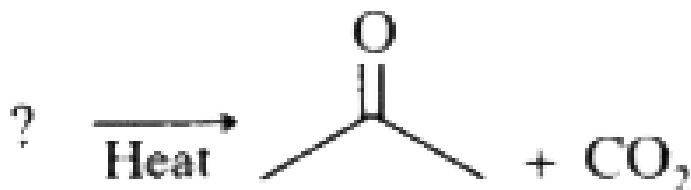
D. amide > ester > acid anhydride >  
acid chloride

**Answer: B**

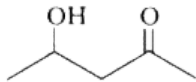


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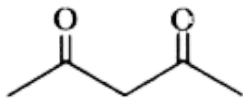
**11.** What starting material is required for the following reaction?



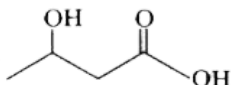
A.



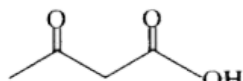
B.



C.



D.

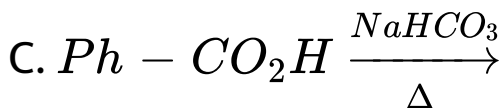
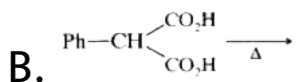
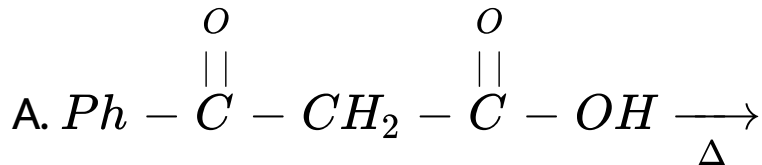


**Answer: D**



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12. In which of the reaction  $CO_2$  gas will evolve?

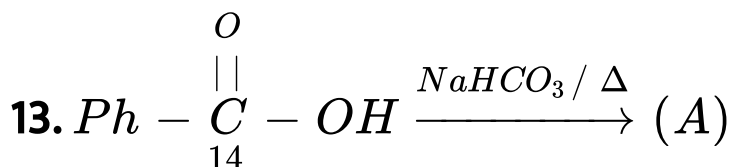


D. All

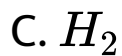
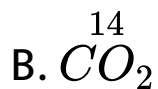
**Answer: D**



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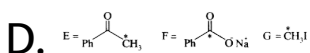
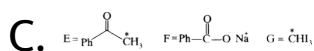
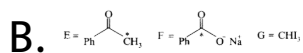
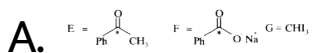
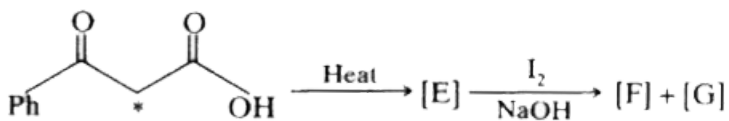


**Answer: A**



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**14.** In the following reaction sequence, the correct structures of E, F and G are



**Answer: C**



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**15.** The product of mixing an acyl halide and an acid is

A. an acid

B. an amide

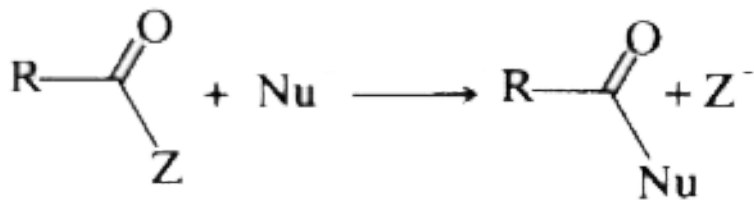
C. anhydride

D. an aldehyde

**Answer: C**



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16.

A. Cl

B.  $\text{OCOCH}_3$

C.  $\text{OC}_2\text{H}_5$

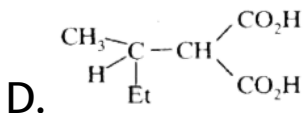
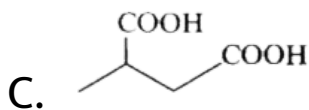
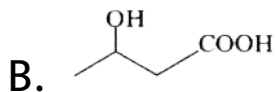
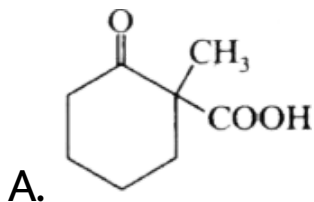
D.  $\text{NH}_2$

**Answer: A**



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17. Which of the following losses chirality when heated ?

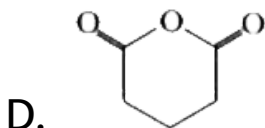
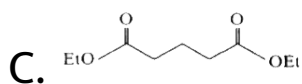
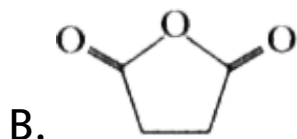
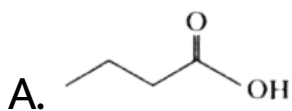
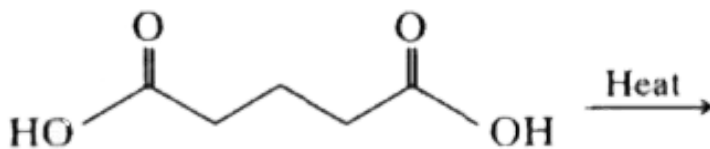


**Answer: B**



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18. What is the product of the following reaction ?



**Answer: D**



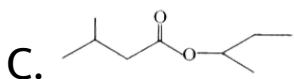
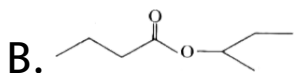
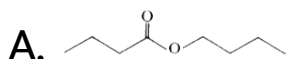
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Additional Objective Questions Linked  
Comprehension Type Paragraph For Questions

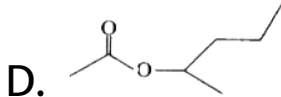
1. 4 A neutral, resolvable organic compound A has molecular formula ( $C_8H_{16}O_2$ ). A on treatment with  $LiAlH_4$  gives isomeric B and C ( $C_4H_{10}O$ ) of which only B is optically active. B on treatment with acidified dichromate solution gives  $D(C_4H_8O)$ , which on refluxing with dilute NaOH followed by acidification of

product gave  $E(C_8H_{14}O)$ . E on heating with  $N_2H_4$  in alkaline medium affords  $F(C_8H_{16})$  which on treatment with B.H/H<sub>2</sub>O/HO produced a resolvable G ( $C_8H_{18}O$ ). G on treatment with acidified dichromate solution produced  $H(C_8H_{16}O)$ , which on treatment with MCPBA produces A.

Compound (A) is







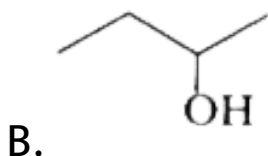
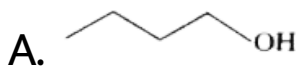
**Answer: B**

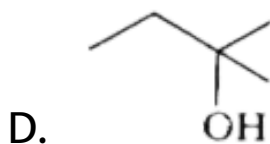
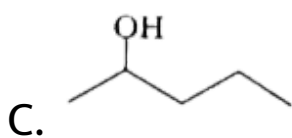
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2. 4 A neutral, resolvable organic compound A has molecular formula ( $C_8H_{16}O_2$ ). A on treatment with  $LiAlH_4$  gives isomeric B and  $C(C_4H_{10}O)$  of which only B is optically active. B on treatment with acidified dichromate solution gives  $D(C_4H_8O)$ , which on refluxing

with dilute NaOH followed by acidification of product gave  $E(C_8H_{14}O)$ . E on heating with  $N_2H_4$  in alkaline medium affords  $F(C_8H_{16})$  which on treatment with B.H/H<sub>2</sub>O/HO produced a resolvable G ( $C_8H_{18}O$ ). G on treatment with acidified dichromate solution produced  $H(C_8H_{16}O)$ , which on treatment with MCPBA produces A.

Compound (B) is





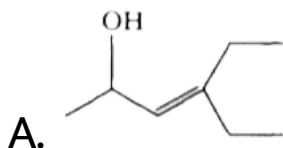
**Answer: B**

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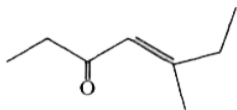
3. 4 A neutral, resolvable organic compound A has molecular formula ( $C_8H_{16}O_2$ ). A on treatment with  $LiAlH_4$  gives isomeric B and C ( $C_4H_{10}O$ ) of which only B is optically active.

B on treatment with acidified dichromate solution gives  $D(C_4H_8O)$ , which on refluxing with dilute NaOH followed by acidification of product gave  $E(C_8H_{14}O)$ . E on heating with  $N_2H_4$  in alkaline medium affords  $F(C_8H_{16})$  which on treatment with B.H/H<sub>2</sub>O/HO produced a resolvable G ( $C_8H_{18}O$ ). G on treatment with acidified dichromate solution produced  $H(C_8H_{16}O)$ , which on treatment with MCPBA produces A.

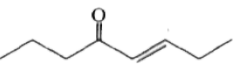
Compound (E)



B.



C.



D. None of these

**Answer: B**

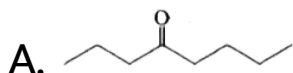


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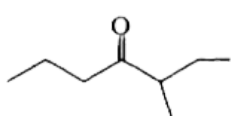
4.4 A neutral, resolvable organic compound A has molecular formula ( $C_8H_{16}O_2$ ). A on treatment with  $LiAlH_4$  gives isomeric B and C ( $C_4H_{10}O$ ) of which only B is optically active.

B on treatment with acidified dichromate solution gives  $D(C_4H_8O)$ , which on refluxing with dilute NaOH followed by acidification of product gave  $E(C_8H_{14}O)$ . E on heating with  $N_2H_4$  in alkaline medium affords  $F(C_8H_{16})$  which on treatment with B.H/H<sub>2</sub>O/HO produced a resolvable G ( $C_8H_{18}O$ ). G on treatment with acidified dichromate solution produced  $H(C_8H_{16}O)$ , which on treatment with MCPBA produces A.

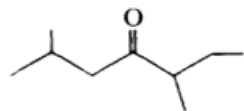
Compound (H) is



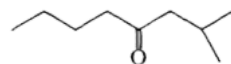
B.



C.



D.



**Answer: B**



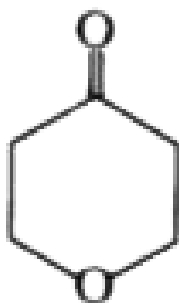
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5. A neutral organic compound  $A(C_5H_8O_2)$  does not decolorize Bayer's reagent and on hydrolysis with dilute  $H_2SO_4$  produces

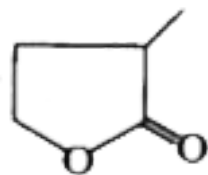
$B(C_2H_{10}O_3)$ , which is diastereomeric. B on heating with concentrated  $H_2SO_4$  undergoes dehydration producing  $C(C_5H_8O_2)$ , which shows geometrical isomerism. Also B on treatment with acidic dichromate solution produced  $D(C_5H_8O_2)$ , which is enantiomeric and gives a yellow precipitate with NaOI. D on gentle heating produces E ( $C_4H_8O$ ), which is non-resolvable.

Compound (A) is

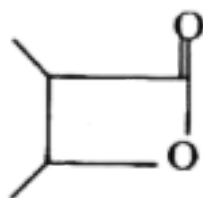




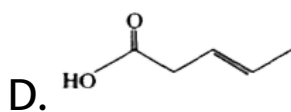
A.



B.



C.



D.

**Answer: C**

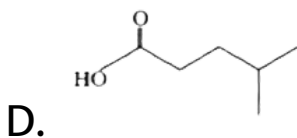
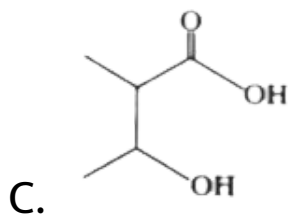
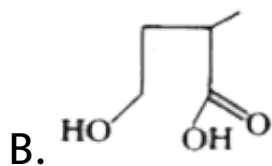
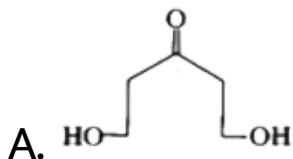


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6. A neutral organic compound  $A(C_5H_8O_2)$  does not decolorize Bayer's reagent and on hydrolysis with dilute  $H_2SO_4$  produces  $B(C_2H_{10}O_3)$ , which is diastereomeric. B on heating with concentrated  $H_2SO_4$  undergoes dehydration producing  $C(C_5H_8O_2)$ , which shows geometrical isomerism. Also B on treatment with acidic dichromate solution produced  $D(C_5H_8O_2)$ , which is enantiomeric and gives a yellow precipitate with NaOI. D on gentle heating produces E ( $C_4H_8O$ ), which is

non-resolvable.

Compound (B) is



**Answer: C**

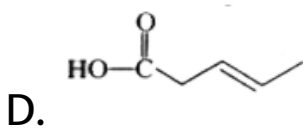
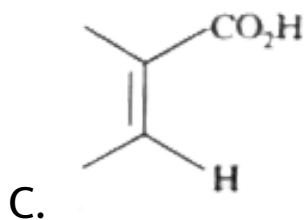
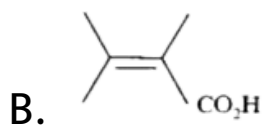
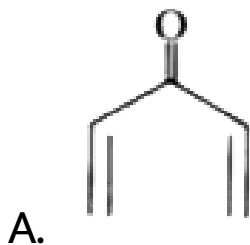


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7. A neutral organic compound  $A(C_5H_8O_2)$  does not decolorize Bayer's reagent and on hydrolysis with dilute  $H_2SO_4$  produces  $B(C_2H_{10}O_3)$ , which is diastereomeric. B on heating with concentrated  $H_2SO_4$  undergoes dehydration producing  $C(C_5H_8O_2)$ , which shows geometrical isomerism. Also B on treatment with acidic dichromate solution produced  $D(C_5H_8O_2)$ , which is enantiomeric and gives a yellow precipitate with NaOI. D on gentle heating produces E ( $C_4H_8O$ ), which is

non-resolvable.

Compound (C) is



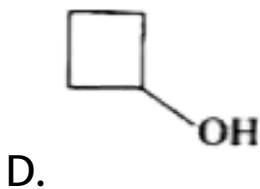
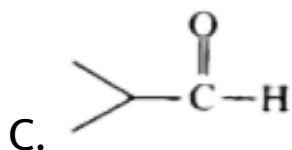
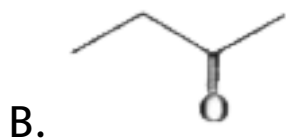
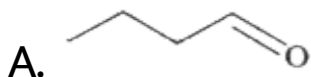
**Answer: C**



8. A neutral organic compound  $A(C_5H_8O_2)$  does not decolorize Bayer's reagent and on hydrolysis with dilute  $H_2SO_4$  produces  $B(C_2H_{10}O_3)$ , which is diastereomeric. B on heating with concentrated  $H_2SO_4$  undergoes dehydration producing  $C(C_5H_8O_2)$ , which shows geometrical isomerism. Also B on treatment with acidic dichromate solution produced  $D(C_5H_8O_2)$ , which is enantiomeric and gives a yellow precipitate with NaOI. D on

gentle heating produces E ( $C_4H_8O$ ), which is non-resolvable.

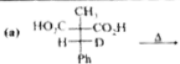
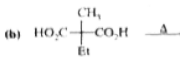
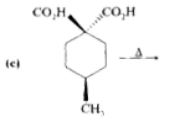
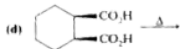
Compound (E) is



**Answer: B**



## Additional Objective Questions Matrix Match Type

Column-I	Column-II
(a) 	(p) Diastereomers
(b) 	(q) Racemic mixture
(c) 	(r) Meso compound
(d) 	(s) CO2 gas will evolve

1.



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