



PHYSICS

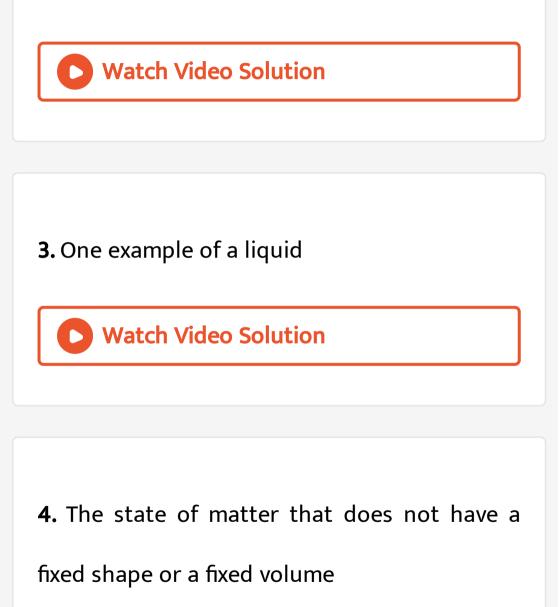
BOOKS - ICSE

MATTER

Exercises Section I Name The Following

1. Anything that has mass and occupies space

2. The three states of matter



5. The state of matter that has limited

movement of particles

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Exercises Section I Choose The Correct Option

1. A solid has

A. a fixed shape and a fixed volume

B. no fixed shape but a fixed volume

C. no fixed shape and no fixed volume

D. a fixed shape but no fixed volume

Answer:

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2. A liquid has

A. a fixed shape and a fixed volume

B. no fixed shape but a fixed volume

C. no fixed shape and no fixed volume

D. a fixed shape but no fixed volume

Answer:

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3. The reason why we are able to smell a leak

from an LPG cylinder

A. because a gas has no fixed shape or fixed

volume

B. because a gas can be compressed easily

C. because a gas spreads out to occupy

space

D. because all gases have a distinct smell

Answer:

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4. If a sample of matter cannot be compressed

easily then it must be:

A. a solid or a liquid

B. a liquid or a gas

C. a gas or a solid

D. a gas

Answer:

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5. Forces of attraction between particles are

strongest in

A. solids

B. gases

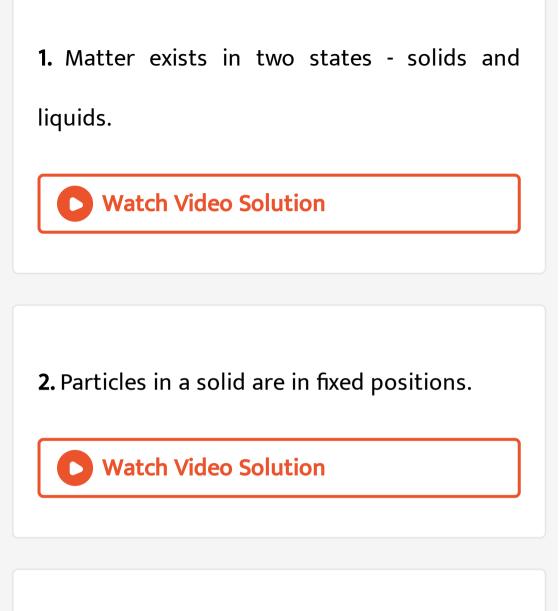
C. solids or liquids

D. solids and gases

Answer:

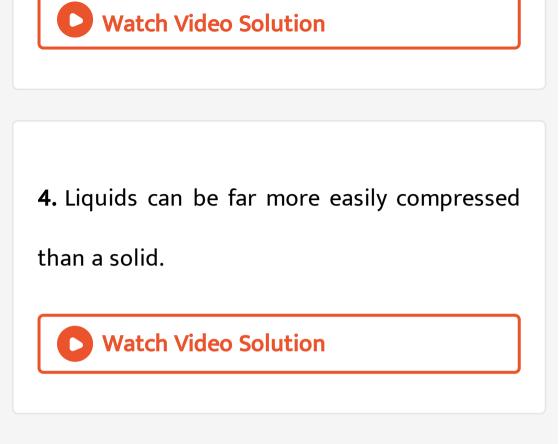
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Exercises Section I Write T For True And F For False Correct The False Statements



3. Particles in a gas have limited space to

move.



5. Particles in a solid have a lot of space

between them.



1. A (brick/petrol) is an example of a solid.



2. (Hydrogen/Kerosene) is an example of

a liquid.

3. We cannot hold a (liquid/gas) in our hand. Watch Video Solution 4. The particles of a solid are packed (loosely/tightly). Watch Video Solution

5. A (liquid/gas) has a fixed

volume but no fixed shape.

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Exercises Section I Match The Following

- Has mass and occupies space
- b. Matter is composed of
- c. Particles can slide over one another
- d. Negligible force of attraction
- e. Particles vibrate in place

1.

- i. Solids
- ii. Particles
- iii. Matter
- iv. Gases
- v. Liquids

1. Particles in a solid can only vibrate in their

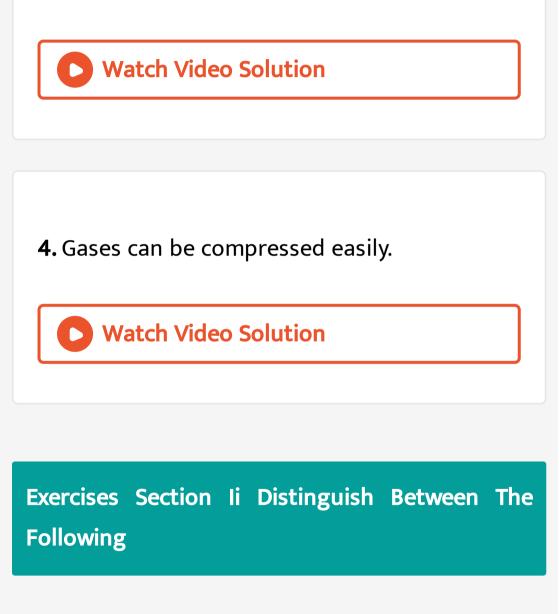
place.

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2. A gas spreads out to occupy all available

space.

3. Liquids cannot be easily compressed.



1. Solids, liquids and gases



Exercises Section Ii Short Answer Questions

1. What is matter? What is it made of?

2. Name the three states of matter, giving two

examples of each.

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3. Explain the arrangement of particles in a solid.

4. Why can liquids take the shape of any vessel

they are poured into?

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Exercises Section Ii Long Answer Questions

1. List the characteristics of solids, liquids and

gases

2. Explain the arrangement of particles in liquids and gases.
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3. Give an example where the characteristic of

gases to occupy all available space is useful to

us.

Questions Write T For True And F For False Correct The False Statements

1. Matter is all around us.

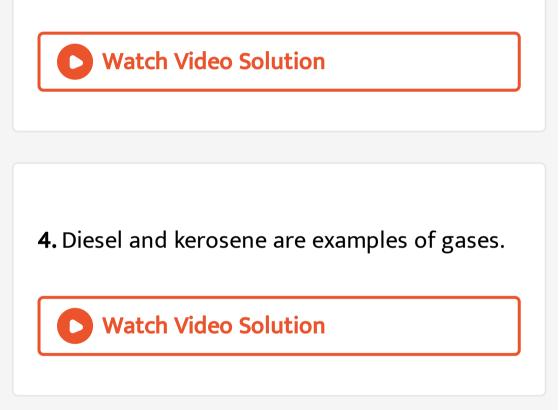
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2. The three states of matter are water, ice and

water vapour.

3. Containers without solids and liquids are

not empty, they are full of gas.



Picture Based Questions

1. Identify a solid , a liquid and a gas in the

picture given below :

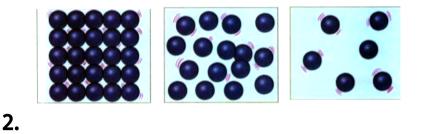
a. Solid :____

b.Liquid:____

c.Gas: ____







a.Identify which of the above pictures shows
particle arrangement in solids, liquids and
gases. Write in the space below each picture.
b. Which state of matter would you expect to
show a fixed volume?
c. Why do you think gases can flow whereas
solids cannot?

