

CHEMISTRY

BOOKS - MTG WBJEE CHEMISTRY (HINGLISH)

ALIPHATIC AMINES

Wbjee Workout Category 1 Single Option Correct Type 1

Mark

- 1. Amines are basic in character because they have
 - A. a lone pair of electrons on the nitrogen atom
 - B. a hydroxyl group in the n1olecule

- C. replaceable hydrogen atom
- D. tetrahedral structure.

Answer: A



- 2. The basic character of amines can be explained
 - A. only in terms of Lowry-Bronsted concepT
 - B. only in tenns of Lewis concept
 - C. both in tenns of Arrhenius and Lewis concepts
 - D. both in terms of Lewis and Lowry-Bronsted concepts.

Answer: D



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- 3. Which of the foil owing is most basic?
 - A. $CH_3CH_2NH_2$
 - B. CH_3CONH_2
 - C. $C_6H_5CONHCH_3$
 - D. NH_2NH_2

Answer: A



4. Increasing order of basic nature of

 $NH_3(I), CH_3NH_2(II)$ and $C_6H_5NH_2(III)$ is

A.
$$III < I < II$$

$$\mathrm{B.}\,I < II < III$$

$$\mathsf{C}.\,III < II < I$$

$$\mathrm{D.}\,I < III < II$$

Answer: A



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5. In aqueous solutions, the basic strength of amines decreases in the order

A. $CH_3NH_2 > (CH_3)_2NH_2 > (CH_3)_3N$

 $\mathsf{B.}\,(CH_3)_2NH > (CH_3)_3N > CH_3NH_2$

 $\mathsf{C.}\ (CH_3)_3N > (CH_3)_2NH > CH_3NH_2$

D. $(CH_3)_2NH > CH_3NH_2 > (CH_3)_3N$

Answer: D



6. In chlorobenzene solutions, the basic strength of amines increases in the order

A.
$$(C_2H_5)_3N < (C_2H_5)_2NH < C_2H_5NH_2$$

B. $C_2H_5NH_2<(C_2H_5)_2NH<(C_2H_5)_3N$

C. $(C_2H_5)_2NH < C_2H_5NH_2 < (C_2H_2)_3N$

D. $(C_2H_5)N < C_2H_5NH_2 < (C_2H_5)_2NH$

Answer: B



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7. The conjugate base of $(CH_3)_2NH_2^+$ is

A. $CH_3)_3$ N

 $\mathsf{B.}\left(CH_{3}\right)_{2}\!NH$

C. $\left(CH_3
ight)_2N^-$

D. $(CH_3)_2N^{\,+}$

Answer: B

8. Which of the following has the most stable conjugate acid?

A.
$$(CH_3)_2$$

B.
$$(CH_3)_3N$$

$$\mathsf{C.}\ C_2H_5NH_2$$

D.
$$C_2H_5NH_2$$

Answer: C



9.	The	amine	salts	commonly	used	for	determination	of
m	olecu	ular ma	sses c	of amines ar	·c			

- A. nitrates
- B. sulphates
- C. chlorides
- D. chloroplatinates

Answer: D



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10. Which one of the fo llowing amines gives carbylamine reaction'?

A. $CH_3CH_2NHCH_3$

 $\mathsf{B.}\, C_6H_5N(CH_3)_2$

C. CH_3)₂ $CHNH_2$

D. $(CH_3)_3N$

Answer: C



11. When ethyl amine is treated with CH_3MgBr , the product is

A. CH_3CH_3

B. CH_4

- $\mathsf{C}.\,CH_3CH_2CH_3$
 - $\mathsf{D.}\,CH_3CH_2CH_2CH_3$

Answer: B



View Text Solution

12. Methane is produced when CH_3MgBr reacts with

- - A. $CH_3CH_2NH_2$
 - B. $(CH_3CH_2)NH$
 - $\mathsf{C}.\,CH_3CH_2OH$
 - D. all of these

Answer: D

13. Primary aliphatic amines can be distinguished from secondary and tertiary arnmes by heating with

A. chloroform and alcoholic KOH

B. chloroform alone

C. soda-lime

D. zinc dust.

Answer: A



14. One of the distinguishing reactions of aliphatic and aromatic primary amines is

- A. Carbylamine reaction
- B. reaction with red fitmus paper
- C. Hofmann bromamide reaction
- D. Coupling reaction.

Answer: D



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15. Primary amines can be distinguished from secondary amines by

- A. Gabriel-pbthalimide reaction
- B. Liebermann nilrosoanline reaction
- C. Hofmann bromamidc reaction
- D. all of these.

Answer: B



- **16.** A mixture of $1^{\circ}, 2^{\circ}$ and 3° amines can be separated by Hinsberg's reagent which is
 - A. Benzoyl chloride
 - B. acetyl chloride

- C. benzene sulphonyl chloride
- D. benzyl chloride.

Answer: C



- **17.** In the Hofmann 's method for separation of $1^\circ, 2^\circ$ and 3° amines, the reagent used is
 - A. acetyl chloride
 - B. benzene sulphonyl chloride
 - C. diethyl oxalate
 - D. nitrous acid.

Answer: C



18. Which of the following amines docs not react with Hinsberg's reagent?

A.
$$CH_3CH_2NH_2$$

B.
$$(CH_3CH_2)_2NH$$

C.
$$(CH_3CH_2)_3N$$

D. all will react.

Answer: C



19. An amine (X) reacts with benzene sulphonyl chloride and the product thus obtained is soluble in KOH . The amine (X) is

- A. 1° amine
- B. 2° amine
- $\mathsf{C.}\,3^\circ$ amine
- D. None of thesen

Answer: A



20. How many primary formula of $C_4H_{11}N$?	,
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A. 1

B. 2

C. 3

D. 4

Answer: D



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21. When Tl-butyl trimethyl amine is heated strongly then major product obtruned is

A.
$$CH_3CH_2CH = CH_2$$

$$\operatorname{B.}CH_3CH=CH-CH_3$$

$$\mathsf{C}.\,CH_2=CH_2$$

$$\mathsf{D.}\,CH_3-CH=CH_2$$

Answer: A



22. Which one of the following is called a carbylamine?

A. RCN

B. $RCONH_2$

C. RCH= NH

D. RNC

Answer: D



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23. Hofmann rearrangement during the conversion of an amide to amme is

- A. intenn olecular
- B. intramolecular
- C. both (a) and (b)
- D. none of these.

Answer: B

24. Thennal decomposition of

$$egin{bmatrix} NH_3 & & & & \ CH_3CH_2 - N & - C \left(CH_3
ight)_2 \ & & & \ \ CH_3 & & CH_3 \end{bmatrix} OH^-$$
 gives

A.
$$CH_3CH = CH_2$$

$$\mathsf{B.}\,CH_2=CH_2$$

$$C. CH_3 - CH_3$$

D.
$$CH_3CH_2CH_3$$

Answer: B



25. Ethyl amine can be obtained from methyl iorude by reaction with alcoholic KCN followed by

- A. hydrolysis
- B. reduction
- C. oxidation
- D. reaction with ammonia.

Answer: B



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26. Which one of the following methods is neither meant for the synthesis nor for separation of amfoes?

- A. Hinsberg method
- B. Hofmann method
- C. Wurtz reaction
- D. Curtius reaction

Answer: C



- **27.** Which of the following is Hofmann mustard oil reaction?
 - A. Reaction of primary amine with $CHCl_3$
 - B. Reaction of primary amine with $CHCI_3 + KOH$

- C. Reaction of primary amine with $CS_2 + HgCl_2$
- D. Reaction of aromatic amine with iodoform.

Answer: C



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28. The reagent that reacts with nitromethane to methylhydroxyl amine is

- A. Zn//HCI
- B. Zn/NH_4CI
- C. Zn/NaOH
- D. Sn/HCI

Answer: B



- **29.** Treatment of ammonia with excess of ethyl iodide will yield
 - A. diethyl amine
 - B. ethyl amine
 - C. triethyl amine
 - D. tetraethyl ammonium iodide.

Answer: D



30. $CH_3NH_2+CHCI_3+KOH o$ nitrogen containing compound $+KCI+H_2O$. Nitrogen containing compound is

A.
$$CH_3 - C \equiv N$$

B.
$$CH_3NH - CH_3$$

C.
$$CH_3-N\equiv C^+$$

D.
$$CH_3N^+\equiv C^-$$

Answer: D



Wbjee Workout Category 2 Single Option Correct Type 2
Marks

1. A compound (X) with molecular formula C_2H_5ON forms a compound Y when treated with Br_2 and KOH. Y on beating with chloroform and alcoholic KOH produces a compound Z with very olTensive smell. The compound 'X' is

A. an amine

B. an amide

C. a hydroxylamine

D. an aminoaldchyde.

Answer: B

- 2. Gas evolved in Bhopal gas tragedy was
 - A. methane
 - B. CO_2
 - C. methyl isocyanide
 - D. butene

Answer: C



3. Condensation of ethyl amine with acctaldehyde gives a compound A which upon reduction with hydrogen and nickel gives

- A. diethyl amine
- B. ethyl methyl amine
- C. triethyl amine
- D. dimethyl amine.

Answer: A



4. What is the end product in the following sequence of operations?

$$C_2H_5NH_2\stackrel{HNO_2}{\longrightarrow} A\stackrel{PCI_5}{\longrightarrow} B\stackrel{NH_3}{\longrightarrow} C$$

- A. Ethyl cyanide
- B. Mcthylaminc
- C. Ethylamine
- D. Acetamide

Answer: C



5. Considering the basic strength of amines in aqueous solution, which one has the smallest pK_b value?

A.
$$C_6H_5NH_2$$

B.
$$(CH_3)_2NH$$

C.
$$CH_3NH_2$$

D.
$$(CH_3)_3N$$

Answer: B



View Text Solution

6. Identify A in the following sequence of reactions:

$$A \xrightarrow{NH_3} B \xrightarrow{CHCI_3} C \xrightarrow{\text{Reduction}} (CH_3)_2 CHNHCH_3$$

- A. Ethyl halide
- B. Iso-propylamine
- C. n-Propyl halide
- D. Iso-propyl halide

Answer: D



View Text Solution

7.
$$CH_3CH_2CI \stackrel{NaCN}{\longrightarrow} X \stackrel{Ni/H_2}{\longrightarrow} Y \stackrel{ ext{Acetic anhydride}}{\longrightarrow} Z$$

Z in the above reaction sequence is

- A. $CH_3CH_2CH_2NHCOCH_3$
- B. $CH_3CH_2CH_2NH_2$

 $C. CH_3CH_2CH_2CONHCH_3$

D. $CH_3CH_2CH_2CONHCOCH_3$

Answer: A



View Text Solution

8. Iso-propylamine caimot be obtained by

$$B_{\bullet} (CH_3)_2CO + NH_3 \xrightarrow{\Delta} ? \xrightarrow{H_2/N_i}$$

$$C$$
. CH_3 $CH - Br + NaNH_2$ CH_3

Answer: C

- **9.** Amines are well known to be stronger bases and nucleophiles than alkenes. Why do enamines, such as 1-dimethylaminocyclopentene, preferentially react with electrophiles at a double bond carbon rather than at nitrogen?
 - A. The nitrogen is stcrically hindered by alkyl subst.ituents.
 - B. Nitrogen is more electonegative than carbon.
 - C. The carbocation formed by electrophilic attack at C-
 - 2 is stabilized by π bonding with the lone pair of

electrons on nitrogen.

D. Ammonium cations are less stable than carbocations.

Answer: C



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10. What sequence of reactions would best accomplish the following reaction?

$$C = N - ?$$
 CH_2

A. $I.\ LiAlH_4$ in ether

II. CH_3I followed by heating with AgOH

B. $I.\ LiAIH_4$ in ether

 $II. P_2O_5$ and heat

C. $I.~20~\%~H_2SO_4$ and heat

 $II. P_2O_5$ and heat

D. $I.\ H_2$ and Lindlar's catalyst

Answer: A



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11. Identify product D in the following reaction sequence:

$$CH_{3} \xrightarrow{C} CH_{2}CH_{2}OH \xrightarrow{K_{2}Cr_{2}O_{7}, H^{+}} A \xrightarrow{SOCl_{2}} B$$

$$CH_{3} \xrightarrow{C} CH_{3}$$

$$\xrightarrow{(CH_{3})_{2}NH} C \xrightarrow{I. LiAiH_{4}, ether} D$$

CH₃

$$CH_3 - C - CH_2C = N$$
A.
$$CH_3$$

$$\begin{array}{c|c} \operatorname{CH}_3 & \operatorname{N}(\operatorname{CH}_3)_2 \\ \operatorname{CH}_3 - \operatorname{C}-\operatorname{CH}_2 \operatorname{CHN}(\operatorname{CH}_3)_2 \\ \operatorname{CH}_3 \end{array}$$

В.

$$\begin{array}{c} CH_3 & O \\ | & | \\ CH_3 - C - CH_2 & CN(CH_3)_2 \end{array}$$

 C.
$$\begin{array}{c} CH_3 & O \\ | & | \\ CH_3 & CH_3 \end{array}$$

$$CH_3$$
 CH_3
 $CH_2CH_2N(CH_3)_2$
 CH_3

Answer: D



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12. An optically active amine (A) of molecular formula $C_4H_{11}N$ is subjected to Hofmann 's exhaustive methylation process and following hydrolysis an alkene

(B) is produced which upon ownolysis and subsequent hydrolysis yields fonnaldehyde and propanal. The amine 'A' is

A.
$$CH_3CHCH_2CH_3$$
 $\mid \atop NH_2$

B.
$$CH_3NH-CH-CH_3$$
 $\mid \atop CH_3$

D.
$$CH_3CH_2CH_2CHNH$$

Answer: A



13. Product (B) of the following reaction is

$$O$$
 + NH₂-OH $\longrightarrow A \xrightarrow{\text{LiAlH}_4} B$

$$\bigcap^{NH_2}$$

Answer: B

$$N \xrightarrow{C_6 H_5 CH_2 CH_2 I} \xrightarrow{AgOH} X + 3^\circ \text{ amine}$$

The final product X is

$$\mathsf{B.}\, CH_2 = CH_2$$

$$\mathsf{C.}\, C_6H_5CH=CH_2$$

D.
$$H_3C$$
 CH_2

Answer: C



15. An organic compound 'P (mol mass -75) on reduction gives compound 'Q' which yields compound 'R' when treated with nitrous acid. Compound R responds to iodofonn reaction. Compound Q is

- A. ethyl a lcohol
- B. nitroethanc
- C. ethanamine
- D. propanol

Answer: C



Wbjee Workout Category 3 One Or More Than One Option Correct Type 2 Marks

1.
$$C_5H_{13}N + HNO_2
ightarrow C_5H_{12}O(3^\circ ext{alcohol})$$

Hence (A) gives

- A. carbylamine reaction
- B. diazonium salt
- C. Hofmann mustard oil
- D. Zerewitinoff's reaction.

Answer: A::B::C::D



2. Which are correct reactions'?

A.
$$\bigcirc$$
 -Cl + NH₃ \rightarrow \bigcirc

$$\mathbf{B.} \xrightarrow{\mathsf{CI} + \mathsf{NH}_3} \longrightarrow -\langle$$

$$CH_3-CH_2-Cl+NH_{3\,(\,l\,)}\,
ightarrow CH_3-CH_2-NH_2$$

D. None of these

Answer: A::B::C



View Text Solution

3. The products ofreaelion of alcoholic silver nitrite witht ethyl bromide are

B. ethene
C. nitroethane
D. ethyl nitrite.
Answer: C::D
View Text Solution
4. Which of the following statements are correct regarding N-methyl aniline?
A. It is less basic than aniline.

A. ethane

B. It is optically inactive due to rapid pyramidal inversion.

C. ft is more reactive than aniline towards $Br_2/{\rm Fe}$.

D. It is soluble in dil. HCI.

Answer: B::C::D



5. Which of the following yields propyl amine?

A.
$$CH_3CH_2Br \xrightarrow{1.NaCN} {2.LiAIH_4}$$

B.
$$CH_3CH_2CHO \xrightarrow{NH_3}_{H_2/Ni}$$

C.
$$CH_3CH_2CH_2CNH_2 \xrightarrow{Br_2} OH$$

D.
$$CH_3COOH + N_3H \stackrel{H^+}{\longrightarrow}$$

Answer: A::B::C



View Text Solution

6. Which of the following sequence of reagents is a good means lo furnish the conversion?

 $RCH_2OH
ightarrow RCH_2NH_2$

A. $KMnO_4,\,SOCI_2,\,NH_3,\,$ heat , Obr

B. $PBr_3, NaCN, H_2, Ni$

C. Collin's reagent $NH_3: H_2: Ni$

D. $Cu, 300^{\circ}C, NH_3, NaBH_3:CN$

Answer: C::D



7.
$$A \xleftarrow{H_3O^+}_{\mathrm{heat}} CH_3 \overset{C}{C} HNC \xrightarrow{H_2,Ni}_{\mathrm{heat}}$$
 B. Products 'A' and B

can distinguished by

- A. the distinguished of $CHCI_3$, ^-OH
- B. the action of HNO_2 , A liberates N_2 gas while B does not
- C. the action of $CS_2, HgCI_2.$ A gives odour o f mustard oil while B doesn't

D. the treatment of benzene sulphonyl chloride, 'A' gives alkali soluble product

Answer: A::B::C::D



$$H_3C$$
 NH₂

Answer: A



9.
$$CH_3CH_2\overset{||}{C}-OH o CH_3CH_2NG_2$$

Which of the following is/are effective for this conversion?

- A. NH_3 (excess), heat, Na, C_2H_5OH
- B. N_3H , cold conc . H_2SO_4
- C. NH_3 (excess) heat , $NaOH,\,Br_2$
- D. $SOCI_2$, NH_3

Answer: B::C

10. Which of the following statements is/are correct?

- A. Methylaminc is more basic than NH_3 .
- B. Amines fonn hydrogen bonds.
- C. Ethylamine has higher boiling point than propane.
- D. Dimethylaminc is less basic than methylamine.

Answer: A::B::C



1. The best method for preparation of $Me_3\mathrm{CC}N$ is

A. to react Me_3COH with HCN

B. to reat Me_3CBr with NaCN

C. to react Me_3CM_qBr with CICN

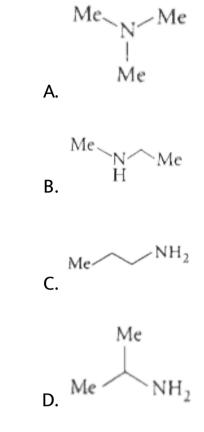
D. to react Me_3Cli with NH^2 CN

Answer: C



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2. An amine C_3H_9N reacts with benzene sulphonyl chloride to fonn a white precipitate which is insoluble in aq. NaOH. The amine is



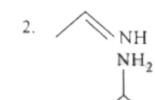
Answer: B



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3. The correct order of basicity of the following compouds is

NH₂



 NH_2

4. H₂N 4

A. 1 < 2 < 3 < 4

B. 1 < 2 < 4 < 3

C.2 < 1 < 3 < 4

 ${\rm D.}\,4 < 3 < 2 < 1$

Answer: C



4. Among Me_3N, C_5H_5N and MeCN (Me = methyl group). the electroncgativity of N is in the order

A.
$$MeCN>C_5H_5N>Me_3N$$

$$\mathrm{B.}\,Me_5N>Me_3N>MeCN$$

C.
$$Me_3NMeCN>C_5H_5N$$

D. electronegativity is same in all .

Answer: A



5. For the reaction below.

the structure of the product Q is

A.

В.

$$\swarrow^{\text{Ph}}_{\text{CN}}$$

C.

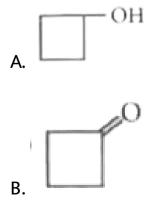
D.

Answer: B



Wb Jee Previous Years Questions Category 3 One Or More Than One Option Corred Type 2 Marks

1. The possible product(s) to be obtained from the reaction of cyclobutyl amine with HNO_2 is/are



C.
$$CH_2OH$$

$$\operatorname{D.} H_2C=CH_2$$

Answer: A::C

