

CHEMISTRY

BOOKS - ICSE

TEST PAPER-2 (CHAPTER 6 TO 9)

Tick The Most Appropriate Answer

1. Name: A reaction in which a substance breaks up into two or more simpler substances

B. neutralization reaction.
C. decomposition reaction.
D. endothermic reaction.
Answer:
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2. About 93% of all atoms in the universe are
A. nitrogen atoms.

A. displacement reaction.

- B. hydrogen atoms.
- C. oxygen atoms.
- D. chlorine atoms.

Answer:



- 3. Liquids that dissolve in water are called
 - A. universal solvents
 - B. miscible liquids

- C. colloids
- D. immiscible liquids.

Answer:



- **4.** A solution in which more of the solute can dissolve at a given temperature is called
 - A. homogeneous mixture.
 - B. heterogeneous mixture.

C. unsaturated solution.

D. saturated solution.

Answer:



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Fill In The Blanks

1. Neutralization reactions between acids and bases are reactions.



2. Metals such as lithium, sodium and potassium are called ___ metals.



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3. A substance that contains water of crystallization is called a substances



4. The different physical forms of the same element in the same physical state are called



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Write True Or False Correct The False Statements

1. Metal nitrates decompose on heating to give metal oxides, brown fumes of nitrogen dioxide and carbon dioxide. (T/F)

2. Distilled water does not conduct electricity due to the absence of free ions.



3. Water of crystallization is the definite number of water molecules chemically combined with a salt in its crystalline form. (T/F)



4. Diamond is made up of flat planar layers of carbon atoms. (T/F)



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Answer The Following In Short

1. Give two examples of double displacement reactions with their chemical equations



2. Differentiate between an oxidizing agent and a reducing agent.



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3. Write the reactions of any three metal oxides with water.



Answer The Following In Detail

1. Write about the preparation of hydrogen by Bosch's process.



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2. Write three uses each of wood charcoal and sugar charcoal.



3. How are oxides classified? Describe each type of oxide with the help of chemical reactions.

