

CHEMISTRY

BOOKS - SWAN PUBLICATION

MATTER IN OUR SURROUNDINGS

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1. Which of the following are matter?

Chair, air, love, smell, hate, almonds, thought,

cold, cold-drink, smell of perfume.



2. Give reason for the following observation:

The small of hot sizzling food reaches you several metres away, but to get the smell from cold food you have to go close.



3. A diver is able to cut through water in a swimming pool, which property of matter does

this observation show.



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4. What are the characteristics of the particles of matter?



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1. The mass per unit volume of a substance is called density.(density=mass/volume). Arrange the following in order of increasing density - air, exhaust from chimneys, honey water, chalk, cotten and iron.



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2. Tabulate the diffrence in the characteristics of states of matter.



3. Comment upon the following : rigidity, compressibility, fluidity, filling a gas container, shape, kinetic energy and density.



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4. Give reasins:

A gas fills completely the vessel in which it is kept.



5. Give reason : a gas exerts pressure on the walls of the container.



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6. Give reason : A wooden table should to be called a solid.



7. Give a reason: We can easily move our hand in air but to do the same in solid block of wood we need a Karate expert.



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8. Liquids generally have lower density as compared to solids. But you must have observed that ice floats on water. Find out why?



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1. Convert the following temperature to celsius scale: 300 K?



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2. Convert the following temperature to celsius scale: 573 K?



3. What is the physical state of water at:

 $250^{\circ}C$?



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4. What is the physical state of water at:

 $100^{\circ}C$?



5. For any substance, Why does the temperature remain constant during the change of state?



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6. Suggest a method to liquefy atmospheric gases.



1. Why does a desert cooler cool better on a hot dry day?



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2. How does the water kept in an earthen pot (matka) become cool during summer?



3. Why does our palm feel cold when we put some acetone or petrol or perfume on it?



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4. Why are we able to sip hot tea or milk faster from saucer rather than a cup?



5. What type of clothes should we wear in summer?



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Exercises

1. Convert the following temperature to the

Celsius scale: 293 K



2. Convert the following temperature to the Celsius scale :470 K



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3. Convert the following temperatures to the

Kelvin scale : $25^{\circ}C$



4. Convert the following temperatures to the Kelvin scale : $1373^{\circ}C$



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5. Give reason for the following observations: Naphthalene balls disappear with time without leaving any solid.



6. Give reason for the following observations: We can get the smell of perfume sitting several metres away.



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7. Arrange the following substances in increasing order of forces of attraction between the particles— water, sugar, oxygen.



8. What is the physical state of water at:

 $250^{\circ}C$?



9. What is the physical state of water at:

 $100^{\circ}C$?



10. What is the physical state of water at: $100^{\circ} C$?



11. Give two reasons to justify that :Water at room temperature is a liquid.



12. Give two reasons to justify that :An iron almirah is a solid at room temperature.



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13. Why is ice at 273 K more effective in cooling than water at the same temperature?

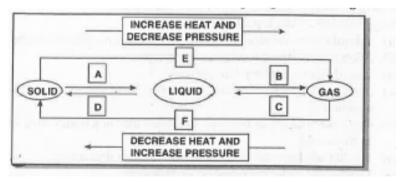


14. What produces more severe burns : boiling water or steam ?



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15. Name A, B, C, D, E and F in the following diagram showing state change





Important Q A

1. Name the three states of matter.



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2. What is the full form of CNG?



3. Define melting point.



4. What is melting point of ice?



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5. Define latent heat.



6. Define boiling point. What is elevation in boiling point?



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7. What is the boiling point of water?



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8. How can we change the state of matter?



9. What is dry ice?



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10. Why is solid carbon-dioxide called dry ice ?



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11. What is Evaporation?



12. Discuss the factors affecting evaporation.



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13. Evaporation causes cooling. Explain why?



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14. Why do the people sprinkle water on the roof or open ground after a hot sunny day?



15. What is plasma?

