

India's Number 1 Education App

CHEMISTRY

BOOKS - MBD

CARBON AND ITS COMPOUNDS



1. What would be the electron-dot structure of

potassium oxide which has the formula K2O?

2. What would be the electron-dot structure of a molecule of sulphur which is made up of eight atoms of sulphur? (HINT. The eight atoms of sulphur are joined together in the from of ring.)

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3. How many structural isomers can you draw

for pentane?





4. What are the two properties of carbon which lead to the huge number of carbon compounds we see around us?

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5. What will be the formula and electron dot

structure of cyclopentane?





10. How would you name the following compounds? CH_3-CH_2-Br

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11. How would you name the following compounds?



12. How would you name the following compounds?



14. A mixture of oxygen and ethyne is brunt for

welding. can you tell why a mixture of ethyne

and air is not used?



15. How would you distinguish experimentally

between an alcohol and a carboxylic acid?



16. What are oxidising agents?



18. People use variety of methods to wash clothes usually after adding the soap. they

beat the clothes on a stone or beat it with a paddle, scrub with a brush or the mixture is agitated in a washing machine. why agitation is necessary to get clean clothes clothes?



19. Ethane, with the molecular formula C_2 H_6

has:

A. 6 covalent bonds

B. 7 covalent bonds

C. 8 covalent bonds

D. 9 covalent bonds

Answer:



20. Butanone is a four- carbon compound with

the functional group:

A. carboxylic acid

B. aldehyde

C. ketone

D. alcohol

Answer:



21. While cooking, if the bottom of the vesel is

getting blackened on the outside, it means that:

A. the food is not cooked completely

B. the fule is not burning completely

C. the fuel is wet

D. the fuel is burning completely

Answer:

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22. Explain the nature of the covalent bond

using the bond formation in CH_3 Cl



propanone



28. How can ethanol and ethanoic acid be differentiated on the basis of their physical and chemical properties?



29. Why does micelle formation take place when soap is added to water? will a micelle be

formed in other solvents such as ethanol also?



30. Why are carbon and its compounds used

as fuels for most application?



31. Explain the formation of scum when hard

water is treated with soap.

32. What change will you observe if you test

soap with litmus paper?

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33. What is hydrogenation? what is its

industrial application?

34. Which of the following hydrocarbons undergo addition reaction : C_2H_6 , C_3H_8 , C_3 H_6 , C_2 H_2 and CH_4.



35. Give a test that can be used to

differentiate between butter and cooking oil.

36. Explain the mechanism of the cleansing

action of soaps

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37. What is allotropy? Name the allotropes of carbon. are they chemically same? compare their physical properties

38. What is covalent bond?



40. Why are some compound called saturated

and other unsaturated compounds?





43. Write the names of the compounds

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44. Draw structure of pentan-2-one





48. what is vinegar?

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49. Give important properties of ethanoi acid

(acetic acid) also give its uses

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50. Define the following terms: esterification



53. Define the following terms: polymerization



54. Descibe a method for the preparation of

soap.

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55. What do you understand by isomers? explain it with an example



58. In order to form large amount of carbon compounds which major elements are used other than hydrogen and oxygen?



59. Carbon is a unique element', why?



60. Explain the structure of diamond and write

why is it so hard?

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61. What are the uses of diamond?

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62. Write the general formula for alkene and give example.



63. Write the general formula for alkyne and

give example.

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64. Write the structure of graphite why is it so

soft?

65. Why are diamonds used in jewellery? give

reasons



66. Why is graphite a good conductor of electricity?

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67. What are the uses of graphite?





70. Write molecular formula of butane?



73. What do you understand by homologous

series? explain with an example



75. What happens when hydrogen gas is passes through vegetable oils in the presence


78. Write combustion of ethanol.



80. Write the IUPAC names ethyl alcohol and acetic acid. write the name of products formed

when ethyl alchol and acetic acid react in the presence of concentreated sulphuric aicd. Write its chemical reaction

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81. How will you prepare:

ethene from ethyl alcohol

acetic acid from ethyl alcohol? give reaction

82. How alcohol is reacted with conc sulphuric

acid?



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84. Distinguish between soaps and detergents



86. Write the name of the reaction in which ethanoic acid and ethanol react together to form product write the common name of the product





87. What is shown in the figure given a head?

Also indicate1 and 2 in the figure





90. write molecular formula of butanoic acid?

91. Write structure for 2-mehtylbutanal?



93. On which element do all living processes depend?



95. What precentage of CO_2 os present in

atmosphere?

96. How many electrons are presents in the

outer shell of carbon?



97. Why does carbon need four electrons to

gain or lose?

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98. How does carbon gain four electrons?





100. Which type of bond is shared by a pair of

two hydrogen atoms?



Γ



106. Write the full form C.N.G.



109. Which of these two is a good conductor

of electricity-diamond or grphite?



112. Which element has the ability to from

large number of comopunds



113. give names of two elements which are placed in the group of carbon and have same valency?



hydrocarbon?



115. By which types of bonds can carbon atoms

join other carbon atom

116. Write the names of five elements which can make bond with carbon to form new compounds



117. Write the formula of ethane?



118. Write the structural formula of propane?



119. Give the next higher homologues of: C_3H_5

Br and C_6 H_10`,

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120. Give the structure of ethanol

121. Name the functional group in propaone.



124. Name the products obtained when

ethanol undergoes complete combustion?



125. Give the name and structure of functional

group present in acetic acid (CH_3 COOH).



126. Which of the following compounds belong

to same homo-logous series?

 $C_2 H_6$ 02, C2H2, C2H6,`CH4O2.

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127. Give the structural formula of an ester



128. Give the names and structural formulas of acid and alcohol from which ehtyl acetate is obtained.



129. Give the electron-dot structure of methane.?

130. Define convalent bond?



132. Name the bond formed between metal atom and non-metal atoms?



135. Give two oxidising agents?



138. Give the electron dot structure of ethene.



140. If two organic compounds have some functional group, predict they have similar

physical or chKHOical properties?

141. Give the general formula or alkanes, and

alkynes?



142. Give the isomers of structural formula C_4

 H_{10} (butane)





148. Give three uses of ethanoic acid?



151. Give name of four carbon compounds?



153. Give the name of simples ketone?

154. Give the name of first member of

homologous series of alkynes



155. Give the name and formula of acid present

in vinegar?

156. Ethanoic acid has the formula:

A. C_2H_5 OH

 $\mathsf{B.}\,CH_3COOH_3$

 $\mathsf{C.}\,CH_3\mathsf{COOH}$

D. C_2 `H_5COOH.

Answer:

157. Carboxylic acids contain functional group:

 $\mathsf{A.}-CHO$

$\mathsf{B.}-CH_2OH$

$\mathsf{C.}-COOH$

D. - OH

Answer:

158. The general fomula of alkyne is:

A.
$$C_n H_{2n-2}$$

 $\mathsf{B.}\, C_n H_{2n+2}$

- $\mathsf{C.}\, C_n H_{2n}$
- D. $C_n H_{2n+2}$

Answer:



159. Propanone has the functional group:

 $\mathsf{A.}-OH$

B.-CHO

- $\mathsf{C}.\, C=O$
- $\mathsf{D.}-COOH$

Answer:



160. Vinegar contain acetic acid:

A.
$$5-8~\%$$

B.
$$15-20~\%$$

- C. 21-29~%
- D. 30-40~%

Answer:

161. The reaction of acetic acid with ethyl alcohol is called

A. Polymerisation

B. saponifaction

C. hydrogenation

D. Esterification

Answer:

162. Fill in the blanks: The functional group

present in ethyl alcohol is____



163. Fill in the blanks:



164. Fill in the blanks: Two atoms of the same element combine to form a molecule. The bond between them is knows as... bond



165. Fill in the blanks: In the formation of

oxygen molecule, the oxygen atoms

share_____ Electrons each

166. Fill in the blanks: The number of single

covalent bond in the molecule of ammonia is

