



# PHYSICS

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### (ASSAMESE ENGLISH)

## THERMAL PROPERTIES OF MATTER

### Example

1. The thermal capacity of 8 liters of a liquid A is equal to the thermal capacity of 10 liters of

liquid B. If the densities of the liquids A and B are  $0.5 \text{ g/cm}^3$  and  $0.6 \text{ g/cm}^3$  respectively, find the ratio of their specific heats.



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2. The temperature of equal masses of three liquids A, B and C are  $12^\circ\text{C}$ ,  $19^\circ\text{C}$  and  $28^\circ\text{C}$  respectively. The temperature, when A and B are mixed is  $16^\circ\text{C}$  and when B and C are mixed, it is  $23^\circ\text{C}$ . What will be temperature when A and C are mixed.



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3. A brass sheet has a circular hole of diameter 3 cm at  $30^{\circ}\text{C}$ . What will be the change in diameter of the hole when it is heated to  $120^{\circ}\text{C}$ ? Given  $\alpha_{brass} = 2 \cdot 10^{-5} \text{ }^{\circ}\text{C}^{-1}$ .



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4. The co-efficient of line expansion brass is  $2 \cdot 10^{-5} \text{ }^{\circ}\text{C}^{-1}$ , By how much the temperature

of a brass rod be increased so as increase its length by 4%.



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5. A rod brass and a rod of iron differ by 14 cm in length at all temperature .What are their length at  $0^{\circ}\text{C}$ ? ( $\alpha$  for iron  $=11 \cdot 10^{-6}/^{\circ}\text{C}$  and  $\alpha$  for brass  $=18 \cdot \frac{10^{-6}}{^{\circ}\text{C}}$ )



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6. The length of a copper rod at  $20^{\circ}\text{C}$  is found to be 150cm when measured with a steel scale which gives correct reading at this temperature. What length of the same copper rod be found when measured with the same rod at  $60^{\circ}\text{C}$ ?



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7. The co-efficient of volumn expansion of a liquid is  $49 \cdot 10^{-5} \text{C}^{-1}$ , By what percentage

does its density change on raising the temperature by  $50^{\circ}\text{C}$ ?



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8. A glass container of volume  $200\text{cm}^3$  is full of mercury at  $20^{\circ}\text{C}$ . The temperature is raised to  $100^{\circ}\text{C}$ . Find how much mercury will be going out of the container. Given that cubical expansion of glass  $\gamma_g = 12 \cdot \frac{10^{-5}}{^{\circ}\text{C}}$ , cubical expansion Hg  $\gamma_g = 18 \cdot \frac{10^{-4}}{^{\circ}\text{C}}$ .



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9. An electric heater takes 15 min to raise a certain quantity of water from  $0^{\circ}\text{C}$  to its boiling point. If it takes another 80 min to turn all water into steam, find the latent heat of vaporisation of water.



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10. How much heat is required to convert 5 kg of ice at  $-10^{\circ}\text{C}$  into steam at  $100^{\circ}\text{C}$ ?



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**11.** When 0.2 kg of ice at  $0^{\circ}\text{C}$  is mixed with 0.40 kg water at  $50^{\circ}\text{C}$ , the resulting temperature is  $7^{\circ}\text{C}$ . Find latent heat of fusion of ice, (Give  $s^{-1}$   $w=4186 \text{ Jkg}^{-1} \text{ K}^{-1}$ )



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**12.** Assume the thermal conductivity of copper is 4 times that of brass. Two rods of copper and brass having the same length and cross



section are joined end to end. The free end of copper is at  $0^{\circ}\text{C}$  and the free end of brass is at  $100^{\circ}\text{C}$ . Calculate the temperature of the junction of the two rods at equilibrium.



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**13.** A black body at  $27^{\circ}\text{C}$  surrounds another at  $-73^{\circ}\text{C}$ . Calculate the net heat transferred per square metre of the body at higher temperature.



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## Exercise

1. Define co-efficient of linear expansion. Dose it depend on unit of length?



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2. How does the density of a body change with temperature ?



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3. Why is some space left in between two rails?



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4. Why does a thick glass tumbler crack on pouring hot water into it?



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5. Why are telegraph wires left loose between two poles?



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**6. Why do lakes freeze first at the surface ?**



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**7. What is the principle of calorimetry?**



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**8.** What are the units of heat in CGS and SI system. How are they related?



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**9.** What is meant by thermal capacity?



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**10.** Define water equivalent of a body. What is its unit?



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11. What is the relation between thermal capacity and specific heat?



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12. Define specific heat of substance. Does it depend on pressure?



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**13.** What is molar specific heat? How is it related to specific heat?



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**14.** What is fusion?



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**15.** Define a melting point. Does it depend on pressure?





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**16.** What is the effect of increase of pressure on melting point of ice?



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**17.** Define a boiling point. Does it depend on pressure?



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**18.** What do you mean by emissive power of a body?



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**19.** What is a black body?



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**20.** What is the emissivity of a perfectly black body? What is meant by absorbing power?



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**21.** What is meant by absorptive power?



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**22.** What is the relation between the emissivity and absorbing power of a black body?



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**23.** State stefan's law of radiation?



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**24.** Show that the co-efficient of area expansion of a rectangular sheet is twice its linear expansivity.



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**25.** What are the co-efficient of expansion of solid? How are they related?



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**26.** What is anomalous expansion of water?

How is it useful in preserving aquatic life?



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**27.** Find a relation to show the effect of temperature on the density .



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**28.** What is the principle of calorimetry?



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**29.** Explain why at high altitude, water boils below 100?



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**30.** Explain what is meant by latent heat.



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**31.** Latent heat of vaporisation of water is 540 cal/gm.Explain the meaning of the statement.



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**32.** Woolen clothes are warmer than cotton cloths why?



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**33.** Describe the construction of a black body.



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**34.** What is the emissivity of a perfectly black body? What is meant by absorbing power?



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**35.** Define coefficient of linear, superficial and volume expansions.



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**36.** What is the principle of calorimetry?



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**37.** Define unit heat. What is specific heat of solid? How does it differ from thermal capacity?



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**38.** Two solids of same mass but the specific heat of one being greater than that of the other are heated equally. Explain in which body the rise of temperature will be more.



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**39.** Define melting point of a substance. State the effect of pressure on it.



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**40.** Explain the different modes of transmission of heat.



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**41.** What is meant by the thermal radiation?



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**42.** What is meant by the co-efficient of thermal conductivity of a material?state its unit and dimension.



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**43.** State Newton's law of cooling. Deduce it from Stefan's law of radiation.



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**44.** The difference between the length of a certain brass rod and that of a steel rod is claimed to be constant at all temperatures. Is it possible?



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**45.** Does the co-efficient of linear expansion depend on the scale of temperature and unit of length?



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**46.** Explain when small pieces of ice pressed together, form a single block.



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**47.** Explain why steam at  $100^{\circ}\text{C}$  usually produces more severe burn than water at the same temperature.



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**48.** Explain why does drinking water kept in an earthen pot cool down ?



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**49.** Why do we feel cooler under the fan?



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**50.** Explain why at high altitude, water boils below 100?



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**51.** Explain why food does not cook speedily at high altitudes.



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52. Why does food get cooked more quickly in a pressure cooker?



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53. Why white cloths are preferred in summer ?



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54. Why is the bottom of a tea kettle blackened while its upper part is so polished?



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55. Explain why a piece of iron appears to be cooler than a piece of wood in winter.



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**56.** Explain why two layer of cloths of equal thickness provide warmer covering than a single layer of double thickness.



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**57.** Cloudy nights are warmer than clear nights-why?



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**58.** At what common temperature would a block of wood and a block of metal feel equally cold or equally hot when touched with?



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**59.** Explain why a piece of paper warped tightly on a wooden rod is found to get charred quickly.



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**60.** Two thermometers are constructed in the same way except that one has a spherical bulb and the other an elongated cylindrical bulb. which one will respond quickly to temperature changes?



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**61.** A metallic solid sphere and a hollow sphere of same material and external radius are heated to the same temperature and allowed

to cool in the same environment .Which of the two has a greater cooling rate?



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**62.** A sphere ,a cube and a thin circular plate,all made of the same material and having the same mass are initially heated to a temperature of  $200^{\circ}\text{C}$  .Which of them will cool fastest and which one slowest when left in air at room temperature?



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**63.** Heat is generated continuously in an electric heater, but its temperature becomes constant after some time. Why?



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**64.** A zinc rod is measured by means of a brass scale (Which is correct at  $0^{\circ}\text{C}$ ) and is found to be 1.0001 metre long at  $10^{\circ}\text{C}$ . What is actual length of the rod at  $0^{\circ}\text{C}$  ? ( $\alpha$  for Zn =  $29 \cdot \frac{10^{-6}}{^{\circ}\text{C}}$

,  $\alpha$  for brass =  $19 \cdot \frac{10^{-6}}{^{\circ}\text{C}}$ ).



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**65.** A hot ball of iron weighing 200 g is dropped into 500 g water at  $10^{\circ}\text{C}$ . The resulting temperature is  $22.8^{\circ}\text{C}$ . Calculate the temperature of the ball



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**66.** The densities of two substance are 2: 3 and their specific heats are 0.12 and 0.09

respectively. compare their capacities per unit volume.



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**67.** Calculate the minimum amount of water at  $15^{\circ}\text{C}$  needed to cool 200g of mercury at  $80^{\circ}\text{C}$  to a temperature of  $24^{\circ}\text{C}$  (specific heat of mercury = 0.033).



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**68.** A piece of metal weighing 50 g is heated to a temperature of  $1000^{\circ}\text{C}$  and quickly dropped into a calorimeter containing 200 g of water at  $25^{\circ}\text{C}$ . The water equivalent of the calorimeter is 10g and specific heat of the metal is 0.1. Calculate the rise in temperature of water.



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**69.** How much steam at  $100^{\circ}\text{C}$  will just melt 3200g of ice at  $-10^{\circ}\text{C}$  ?







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70. Find the result of mixing 100 g of ice at  $-12^{\circ}\text{C}$  with 50g of water at  $92^{\circ}\text{C}$  .



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71. Calculate the amount of heat required to change 5.0 kg of a ice to  $-20^{\circ}\text{C}$  into steam at  $100^{\circ}\text{C}$  .



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72. Find the time in which a layer of ice 3 cm thick on the surface of a pond will increase in thickness by 1 mm when the temperature of the surrounding is  $-20^{\circ}\text{C}$ . The thermal conductivity of ice is  $2.1 \text{ W/m K}$ , latent heat of ice  $3.36 \times 10^5 \text{ J/kg}$  and density is  $900 \text{ kg m}^{-3}$ .



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73. The opposite face of a cubic block of iron of cross-section  $8 \text{ sq.cm}$  are kept in touch with

steam and melting ice. Determine the quantity of ice melted at the end of 5 minutes ( $K=0.2$  CGS units).



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**74.** Calculate the rate of loss heat through a glass window of area  $1000\text{sq.cm.}$  and thickness  $4\text{mm}$  when the temperature inside is  $37^\circ\text{C}$  and outside is  $5^\circ\text{C}$  ( $K$  for glass  $=0.0022$ )



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75. A body which has surface area of  $5.0 \text{ cm}^2$  and a temperature of  $727$  radiates  $300\text{j}$  of energy per minute. What is its emissivity? ( $\sigma = 5.67 \cdot 10^{-8} \frac{\text{W}}{\text{m}^{-2}} \text{K}^{-4}$ )



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76. The temperature of a body is increased from  $27^\circ\text{C}$  to  $127^\circ\text{C}$ . By what fraction would the radiation emitted by it increase?



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77. A body cools from  $80^{\circ}\text{C}$  to  $50^{\circ}\text{C}$  in 5 minutes. If the temperature of the surrounding is  $20^{\circ}\text{C}$ , Calculate the time it takes to cool from  $60^{\circ}\text{C}$  to  $30^{\circ}\text{C}$ .



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78. Two vessels of different materials are similar in size in every respect. The same quantity of ice filled in them gets melted in 20

min. and 40 min respectively. The ratio of their thermal conductivity is

A. 5 : 6

B. 6 : 5

C. 2 : 1

D. 1 : 3

**Answer: A**



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79. The temperature of a black body increases from  $7^{\circ}\text{C}$  to  $287^{\circ}\text{C}$ . The rate of energy radiation increases by

A.  $\left(\frac{287}{4}\right)^4$

B. 16

C. 4

D. 2

**Answer: C**



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80. A black body has maximum  $\omega \cdot L \cdot \lambda m$  at 2000K. Its corresponding  $\omega \cdot L$  at 3000 K is

A.  $\frac{3}{2} \lambda m$

B.  $\frac{2}{3} \lambda m$

C.  $16/81 \lambda m$

D.  $81/16 \lambda m$

**Answer: C**



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81. Two spheres made of same material have radii in the ratio 1:2 both are the same temperature Ratio of heat radiation energy emitted per sec by them is

A. 1:2

B. 1:4

C. 1:8

D. 1:16

**Answer: B**



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**82.** A body cools from  $60^{\circ}\text{C}$  to  $50^{\circ}\text{C}$  in 10 min. if room temperature is  $25^{\circ}\text{C}$ , the temperature of the body at the end of next 10 min will be

A.  $38.53^{\circ}\text{C}$

B.  $40^{\circ}\text{C}$

C.  $45^{\circ}\text{C}$

D.  $42.86^{\circ}\text{C}$

**Answer: B**



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**83.** A body cools from  $50^{\circ}\text{C}$  to  $40^{\circ}\text{C}$  in 5 min .Its temperature comes down to  $33.3^{\circ}\text{C}$  in next 5 min .the temperature of surrounding is

A.  $15^{\circ}\text{C}$

B.  $20^{\circ}\text{C}$

C.  $25^{\circ}\text{C}$

D.  $10^{\circ}\text{C}$

**Answer: B**



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84. There is a small hole at the centre of metal disc. On heating the size of the hole

A. decreases

B. increases

C. remains same

D. depends on  $\beta$  of the metal

**Answer: B**



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**85.** If a bimetallic strip made of iron and copper is heated ,then

A. it gets twisted

B. it does not bend

C. it bends

D. with copper strip on the convex side

**Answer: A**



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**86.** Which of the following is a correct relation for soils?

A.  $\alpha = \frac{2}{3}\beta$

B.  $\gamma = \frac{3}{2}\beta$

C.  $\gamma = \frac{3}{2}\alpha$

D.  $\gamma = \frac{2}{3}\beta$

**Answer: B**



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87. Sometimes ,a cycle with well inflated tyres left in sun,has its tube burst open.It is because

A. the volume of air inside the tube increases

B. the pressure of air inside the tube increases

C. the tube melts

D. both volume and pressure of air increases

**Answer: B**



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**88.** Two spheres of same size are made of same metal but one is hollow and the other is solid. They are heated to same temperature, then

A. both spheres will expand equally

B. hollow sphere will expand more than the solid one



C. solid sphere will expand more than the  
solid one

D. none of these

**Answer: A**



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**89.** Which of the following has the highest  
specific heat capacity?

A. Water

B. mercury

C. hydrogen

D. steel

**Answer: C**



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**90.** Water is used in hot water bottles, because

A. it is a good conductor

B. it is easily available

C. it has low specific heat

D. it has a high specific heat

**Answer: D**



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**91.** For a gram-molecule of an ideal gas  $(PV)/T$

is

A. 4.2 calories

B. 8.3 calories

C.  $4.2 \times 10^3$  ergs

D. 2 calories

**Answer: D**



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**92.** Water equivalent of a substance is equal to

A. the mass of the substance

B. product of mass and specific heat

C. specific heat of the substance

D. none of these

**Answer: B**



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**93.** The unit of thermal capacity is

A. joule/kelvin

B. joule/kg

C. joule/kg/kelvin

D. none of these

**Answer: A**



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**94.** Specific heat of body depends upon

- A. mass of the body
- B. rise of temperature
- C. amount of heat supplied
- D. none of these

**Answer: D**



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95. The specific heat of water is

A. one

B. zero

C. infinity

D. unknown

**Answer: A**



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96. The water equivalent of a block of solid mass 200 g and of specific heat  $0.4 \text{ cal } g^{-1}(\text{ }^\circ C)^{-1}$  is

A.  $800 \text{ cal}(\text{ }^\circ C)^{-1}$

B. 80 g

C. 500 cal

D. 500g

**Answer: B**



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**97.** With rise of pressure, the boiling point of a liquid

A. decreases

B. increases

C. does not change

D. nothing can be predicted

**Answer: B**



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98. With increase in pressure ,the melting point of substance which contract on melting is

A. lowered

B. raised

C. unaffected

D. nothing can be predicted

**Answer: A**



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99. A closed bottle containing water at  $30^{\circ}\text{C}$  is carried to the moon in a spaceship. If it is placed on the surface of the moon, what will happen to the water as soon as the lid is opened?

A. Water will boil

B. Water will freeze

C. nothing will happen to it

D. it will decompose into  $\text{H}_2\text{O}_2$

**Answer: A**



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**100.** The ventilation of building is necessary for

- A. entry of outside oxygen
- B. removing  $CO_2$  exhaled by us
- C. entry of outside moisture
- D. entry of outside heat and light

**Answer: B**



**101.** The melting point of an alloy is usually

- A. lower than those of the constituents
- B. higher than those of the constituents
- C. in between those of the constituents
- D. equal to that of the heaviest constituent

**Answer: A**



**102.** What is the effect of increase of pressure on melting point of ice?

A. increased

B. decreased

C. remaining same

D. increased by  $1^\circ$  for every unit rise of pressure

**Answer: B**



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**103.** Boiling point of water

- A. increases with increase of pressure
- B. decreases with increase of pressure
- C. not affected by pressure
- D. none of these

**Answer: A**



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**104.** A marble floor appears cold than a cemented floor, because marble has

- A. low conductivity
- B. high conductivity
- C. high radiating capacity
- D. high specific heat

**Answer: B**



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**105.** Air is heated by

A. conduction

B. convection

C. radiation

D. none of these

**Answer: B**



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**106.** A perfect black body

A. absorbs all the radiations

B. reflects all radiations

C. absorbs and reflects all radiations

D. absorbs all radiations but reflects none.

**Answer: A**



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**107.** The essence of kirchoff's law is that

A. a good absorber must be a bad radiator

B. a good absorber must be a good radiator

C. a good absorber must be a good conductor

D. a good absorber must be a bad conductor

**Answer: B**



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**108.** The total radiation emitted by a perfectly black body is proportional to

A. the absolute temperature

B. the square of absolute temperature

C. the cube of absolute temperature

D. the fourth power of absolute temperature

**Answer: D**



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**109.** Conversion of heat into electrical energy can be achieved by According to Newton's law of cooling (provided the difference of temperature is small), the rate of loss of heat is proportional to

A. transistor

B. voltmeter

C. thermocouple

D. photo-electric cell

**Answer: C**



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**110.** According to Newton's law of cooling (provided the difference of temperature is small), the rate of loss of heat is proportional to

- A. the excess temperature
- B. the square of excess temperature
- C. the cube of excess temperature
- D. the fourth power of excess temperature

**Answer: A**



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**111.** An ideal material for making cooking vessels must have

A. Small conductivity and large heat capacity

B. large heat capacity and large conductivity

C. small heat capacity and large conductivity

D. small heat capacity and small conductivity

**Answer: C**



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**112.** A hot body will radiate heat most rapidly ,if its surface is



A. White and polished

B. white and rough

C. black and polished

D. black and rough

**Answer: D**



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**113.** If the temperature of sun is doubled, the rate of energy received on earth will be increased by a factor roughly

A. 2

B. 3

C. 8

D. 16

**Answer: D**



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**114.** which of the following cylindrical rods will conduct most heat, when their ends are maintained at the same steady temperature?

A. length 100cm, radius 1 cm

B. length 100 cm, radius 2 cm

C. length 200 cm, radius 2 cm

D. length 200 cm, radius 1 cm

**Answer: B**



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**115.** Two spheres of same material have radii 1 m and 4m and temperature 4000K and 2000K

respectively. The energy radiated per second by the first sphere is

- A. greater than that by the second
- B. less than that by the second
- C. equal in both cases
- D. the information is incomplete to draw any conclusion

**Answer: C**



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**116.** A black body at a high temperature  $T$  radiates energy at the rate of  $E$ . When the temperature falls to  $T/2$ , the radiated energy will be

A.  $E/4$

B.  $E/2$

C.  $2E$

D.  $E/16$

**Answer: D**



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**117.** A spherical black body with radius 12 cm radiates 450 W power at 500K, If the radius were halved and the temperature doubled, the power radiated in watt would be

A. 225

B. 450

C. 900

D. 1800

**Answer: D**



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**118.** A sphere, a cube and a thin circular plate, all made of the same material and having the same mass are initially heated to a temperature of  $3000^{\circ}\text{C}$ . Which of these will cool fastest

A. sphere

B. cube

C. plate

D. All will take same time

**Answer: C**



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**119.** A polished metal plate with a rough black spot on it is heated to about  $1400\text{K}$  and quickly taken to a dark room. Which of the following statements is true?



A. The spot will appear brighter than the plate.

B. The spot will appear darker than the plate

C. The spot and the plate will appear equally bright

D. The spot and the plate will not be visible in the dark room

**Answer: A**



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