

CHEMISTRY

BOOKS - CENGAGE CHEMISTRY (ENGLISH)

CLASSIFICATION AND NOMENCLATURE OF ORGANIC COMPOUNDS

Illustration

1. Given the IUPAC of the following alkanes:

(i)
$$CH_3- {C \atop |\atop CH_2 \atop CH_3} H-CH_2- {CH_3 \atop |\atop CH_3 \atop CH_3}$$



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2. Given the IUPAC names of the following compounds:

(i)
$$CH_3CH_2CH_2 \overset{CH_3}{\overset{C}{C}} CH_2 - \overset{CH_3}{\overset{C}{C}} = CH_2$$

$$\downarrow & \downarrow & \downarrow \\ & \downarrow & \downarrow \\ & C_2H_5 & \downarrow \\ & CH_3 & C & \downarrow \\ & CH_3 & CH_3 & C & \downarrow \\ & CH_3 & CH_3 & CH_3 & CH_3 & CH_4 \\ & CH_3 & CH_4 & CH_4 & CH_5 \\ & CH_4 & CH_5 & CH_5 \\ & CH_5 & CH_5 & CH$$

$$(II) CH_3 - H - C = CH$$

(iii)
$$C_6H_5-CH=CH-CH_2Cl$$

$$CH_3CH - CH - CH_2Cl$$

(iv)
$$CH_3CH=CH-CH_2Br$$

v
$$CH_2=CH-\overset{\circ}{C}-CH=CH_2$$

vi
$$CH_2=CH-rac{CH_2CH_3}{C}H-C=CH_2$$

vii.
$$(CH_3)_3C-CH=CH_2$$



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$$egin{aligned} {f 3.} \ CH_3 - CH - CH - CH_3 \ & ig|_{Cl} & ig|_{I} \ CH_3 - CH - (CH_2)_4 - CH - CH - CH_2CH_3 \ & ig|_{CH_3} & ig|_{Br} \end{aligned}$$



4.
$$CH_3 - C - CH_2OH$$

ii CH (3)-CH=CH-CHO`



- **5.** Give the IUPAC names for the following polyfunctional compounds:
- i. $CH_3CH_2O-CH_2-CHOH-CH_3$

ii.
$$CH_3- {C \atop |}{H-CH_2- {C \atop |}{C \atop |}{C \atop |}{CH_3}}$$

iii.
$$CH_2 = C - CH - CH_2CN$$

iv. $CH_3 - CH - COOH_2$

v. $HOOCCH_2 - CH - CH_2COOH_2$

vi. $CH_3 - CH - CH_2 - CH_2$
 $COOH_3 - CH_2 - CH_2$
 $COCH_3 - CH_2$
 CHO_1

vii. $HOOCCH_2 - CH_2 - CH_2$
 $COCH_3 - CH_2$
 CHO_1

viii. $CH_3 - CH_2 - CH_2$
 $COCH_3 - CH_2$
 CHO_1
 CHO_1



- **6.** a. Rewrite the following structual formula in bond line notation.
- (i) $CH_3CH_2CH_2CH_2CH_3$

iii.
$$(CH_3CH_2)NCH_3$$

ii. $(CH_3)_2CHCH_2CH(CH_3)NH_2$

iv. $CH_3CHBrCH_2CH_2CHO$



i.

ii.

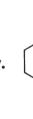
iii.

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7. Given the IUPAC names of the following compounds:

$$\begin{array}{c} \longrightarrow \text{CH} = \text{CH} - \text{CH} - \text{CH}_2\text{CH}_3 \\ \mid \text{CH}_3 \end{array}$$

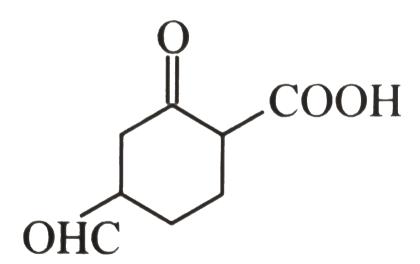
CH₂CHO



COOCH₃

vi.
$$HOCH_2 - C - CH = CHCH_2COOH$$

 $\begin{array}{c} \text{CH}_3 \\ \text{Vii.} \\ \hline \\ \text{CH}_3 \end{array}$





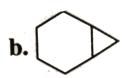
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8. Give the IUPAC names of the following compounds :

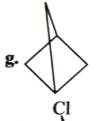


c.

e.



d. ()



i.



j. Br

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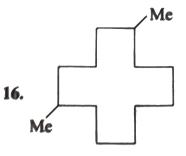
1.

Me-

Me

H

15. Me () Me



17. Me_____NH

35.

33.
$$Me$$
 Me
 Me
 Me
 Me
 Me
 Me
 Me

Me

 $\ddot{\Omega}$

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2. Draw Structure of Benzonitrile

- 3. Write the correct name of the following:
- 1. Pent-1-yn-5-ol



4. Write the structure of cetylethyldimethylammonium bromide, a compound with antiseptic property and also used as a cationic detergent.

Predict its solubility behavior in water and diethyl ether.

b. Write structure of cetyltrimethylammonium bromide, a popular cationic detergent used in hair conditioners



5. Write the structure (s) of the simplest alkane (s), with fewest number of C atoms, possessing $1^{\circ}, 2^{\circ}, 3^{\circ}$, and $4^{\circ}C$ atoms.



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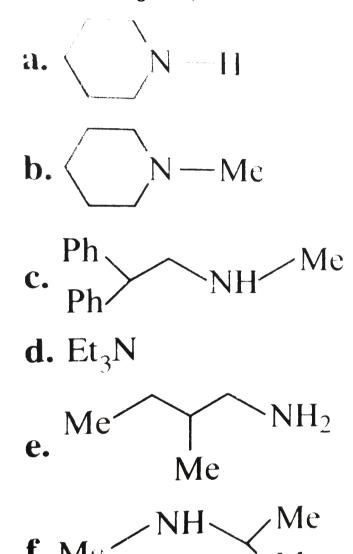
Exercises

- 1. write the structural formula for each of the following
- a. A 3° amine with the formula C_3H_9N .
- b. Three ethers with the formula $C_4H_{10}O$.
- c. A 3° alcohol with the formula C_4H_8O .



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2. Indicate the following as $1^{\circ}\,,\,2^{\circ}$, and 3° amine:





3. Indicate the following as 1° , 2° , and 3°

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4. Write the structural formula for seven compounds with the formula C_3H_6O and identify the funcitonal groups.



5. There are seven isomeric compounds with the formula $C_4H_{10}O$ Write their structures and identify their functional groups

6. There are four alky1 chlorides with the formula C_4H_9CI . Writre their structures and identify then as 1° , 2° , and 3° alky1 chlorides



7. There are four amides with the formula C_3H_7NO .

Write their structures.

Identify the amide which has lower melting point and boiling point than the other three.



8. Write the IUPAC name of the compound (A) which is a 2-methy1 branched alkane having a molecular mass of 254. This

compound is a sex-attractant and is isolated from female tiger moths.



9. Write the IUPAC name of the compound (A) which is a 2-methyl alkane with molecular mass of 72.

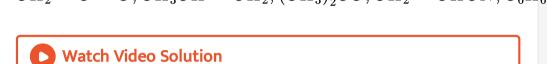


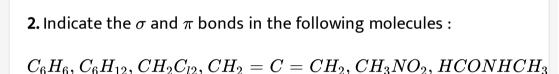
10. Write the IUPAC name of the compound (A) in which the molar ratios of C, H, and O of equal having a molecular mass of 58



Concept Application Type

1. What are hybridisation states of each carbon atom in the following compounds ? $CH_2=C=O, CH_3CH=CH_2, (CH_3)_2CO, CH_2=CHCN, C_6H_6$





bond line formulas for : Isopropyl alcohol, 2,3-





3. Write

4. Give the IUPAC names of the following compounds

f. Cl₂CHCH₂OH



a.

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5. Which of the following represents the correct I.U.P.A.C. name for the compounds concerned?

2,2-dimethylpentane or 2-dimethylpentane



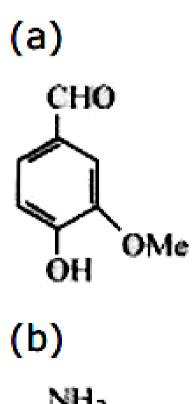
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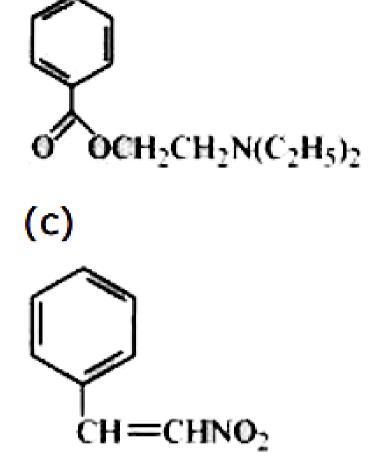
6. Draw formulas for the first five members of each homologous series beginning with the following compounds. (a)H-COOH (b) CH_3COCH_3 (c) $H-CH=CH_2$



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7. Identify the functional groups in the following compounds







Linked Comprehesion Type

1. The analgesic drug ibuprofen (A) is chiral and exists in (+) and (-) froms. One enantiomer is physiologically active, while the other is inactive. The other is inactive. The structure of ibuprofen is given below.

The principal functional group in (A) is:

A. Phenyl

B.-COOH group

C. Isopropyl

D. Both (a) and (b)

Answer: B

2. The analgesic drug ibuprofen (A) is chiral and exists in (+) and (-) froms. One enantiomer is physiologically active, while the other is inactive. The other is inactive. The structure of ibuprofen is given below.

The IUPAC name of (A) is:

- A. 3-(p-Isobutyl phenyl) propanoic acid
- B. 2-(p-Isobutyl phenyl) propanoic acid
- C. 3-(p-sec-Butyl phenyl) propanoic acid
- D. 2-(p-sec-Butyl phenyl) propanoic acid

Answer: B



3. The analgesic drug ibuprofen (A) is chiral and exists in (+) and (-) froms. One enantiomer is physiologically active, while the other is inactive. The other is inactive. The structure of ibuprofen is given below.

The number of π - bonds in (A) is :

A. 2

B. 3

C. 4

D. 5

Answer: C

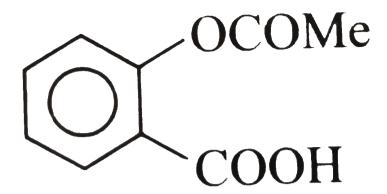


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4. The analgesic drug ibuprofen (A) is chiral and exists in (+) and (-) froms. One enantiomer is physiologically active, while the other is inactive. The other is inactive. The structure of ibuprofen is given below.

The number of σ -bonds in (A) is:

A. 30				
B. 31				
C. 32				
D. 33				
Answer: D				
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5. Aspirin is widely used as an analgesic drug. It is optically inactive.				
The structure of asprim is:				



The principal functional group in(A) is :

A. Pheny1

B. - COOH

C. Ester

D. All

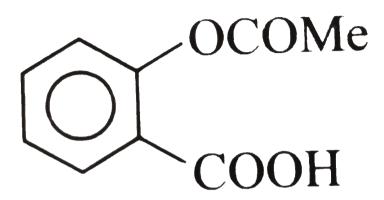
Answer: B



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6. Aspirin is widely used as an analgesic drug. It is optically inactive.

The structure of aspirin is:



Which of the following is not the correct name for aspirin?

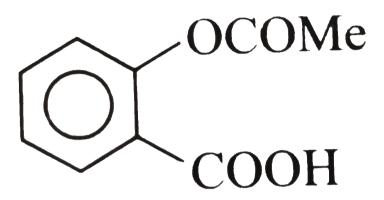
- A. 2-Acetyl salicylic acid
- B. 2-Acetoxy benzoic acid
- C. 2-Acetoxy salicylic acid
- D. None

Answer: C



7. Aspirin is widely used as an analgesic drug. It is optically inactive.

The structure of aspirin is:



The number of π - bonds in (A) is:

A. 3

B. 4

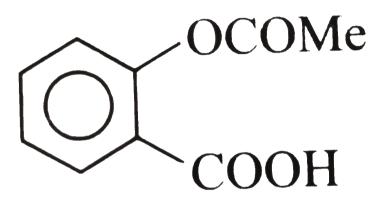
C. 5

D. 6

Answer: C

8. Aspirin is widely used as an analgesic drug. It is optically inactive.

The structure of aspirin is:



The number is σ - bonds in (A) is:

- A. 19
- B. 20
- C. 21
- D. 22



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9. Crixivan, a drug produced by Merck and Co., is widely used in the fight at against AIDS (acquied immune dificiency syndrome). The sturcture of cirxivan is given below:

How many 2° alcohol groups are present in the above compound?

A. Zero

B. 1

C. 2



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10. Crixivan, a drug produced by Merck and Co., is widely used in the fight at against AIDS (acquied immune dificiency syndrome). The sturcture of cirxivan is given below:

How may amide groups are present in the compound?

A. Zero

B. 1

C. 2



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11. Crixivan, a drug produced by Merck and Co., is widely used in the fight at against AIDS (acquied immune dificiency syndrome). The sturcture of cirxivan is given below:

How many 3° amine groups are present in the compound?

A. Zero

B. 1

C. 2



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12. Crixivan, a drug produced by Merck and Co., is widely used in the fight at against AIDS (acquied immune dificiency syndrome). The sturcture of cirxivan is given below:

How many 2° amine groups are present in the compound ?

A. Zero

B. 1

C. 2

Answer: A



Multiple Corrcct Answers Types

- **1.** Which of the following statements is/are wrong?
 - A. $C_n H_{2n}$ is the general formula of alkanes.
 - B. In homologous series, all member have the same physical properties.
 - ${\sf C.}\ IUPAC$ means International Union of Physics and Chemistry.
 - D. Butane contains two $1\,^{\circ}\,C$ atoms and two $2\,^{\circ}\,C$ atoms.

Answer: A::B::C



- **2.** Which of the following statements is/are correct?
 - A. Homologous series can be represented by a general formula.
 - B. The chemical properties of an organic compound depend on the functional group.
 - C. Groups obtained by the removal of one ${\cal H}$ atom from the alkane are called alkyl group.
 - D. Alkynes consist of one double-bond in their molecules.

Answer: A::B::C



- **3.** Which of the following statements is/are wrong?
 - A. Acetic acid is the systematic name of vinegar.

B. $Me-\stackrel{|}{C}-OH$ is an unsaturated compound.

C. Prefixes like n-,iso, sec-,tert,neo-,etc., are used in IUPAC system.

D. The systematic names of acids are formed by dropping-e of the name of parent alkane and adding -oic acid.

Answer: A::B::C



4. Which of the following statements is/are correct?

A.
$$R-C-O-C-R$$
 is an unsaturated compound.

B. Neohydrocarbons contain a $3^{\circ} C$ atom.

C. The IUPAC name of isopropyl alcohol is propan-2-ol.

D. The IUPAC name of (CH_3CN) ie ethanenitrile.

Answer: C::D



5. Which of the following statements is/are correct?

mixture with air.

A. Methane was named as fire damp as it forms explosive

B. Primary suffixes are added to root word to show saturation or unsaturation in a ${\cal C}$ atom.

C. The IUPAC name of valeric acid is pentanoic acid.

D. The common name of hexanoic acid is caproic acid.

Answer: A::B::C::D



- **6.** Which of the following statements is/are correct?
 - A. The IUPAC name of amyl alcohol is pentanol.
 - B. The IUPAC name of isoamyl alcohol is 3-methyl butanol.
 - C. Wood spirit is methanol.
 - D. Methyl alcohol is also called carbinol

Answer: A::B::C::D

names.



- **7.** Which of the following statements is/are correct?
 - A. The trivial name of organic compounds are called common

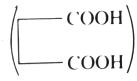
- B. The systematic name of organic compounds are obtained from the IUPAC system.
- C. The systematic names of alkanes are based on the numebe of ${\it C}$ atoms in the longest continuous chain of ${\it C}$ atoms.
- D. The maximum number of functional groups must be included in the ${\cal C}$ atom chain selected even if it does not satisfy the longest chain rule.

Answer: A::B::C::D



- **8.** Which of the following statements is/are correct?
 - A. The common name of $(HOOC-CH_2-COOH)$ is malonic acid.

B. The common name of



is succinic acid.

C. The IUPAC name of $(CH_2=CH-OCOCH_3)$ is viny1 acetate.

D. The IUPAC name of acrylonitrile is Prop-2-ene-nitrile.

Answer: A::B::D



- **9.** Which of the following statements is/are correct?
 - A. The common name of benzene-1,2-diol is catechol.
 - B. The common name of benzene-1,3-diol is resorinol.
 - C. The common name of benzene-1,4-diol is quinol.

D. The common name of benzene -1,4-diol is hydroquinone

Answer: A::B::C::D



10. Which of the following statements is/are correct?

- A. The common name of benzene-1,2,3-triol is pyrogallol.
- B. The common name of benzene-1,2,4-triol is hydroxquinol.
- C. The common name of benzene-1,3,5-triol is phoroglucinol.
- D. The common name of $(CH_2=CH-Ph)$ is styrene.

Answer: A::B::C::D



11. The compounds in which C uses its sp^3 - hybrid orbitals for bond formation are:

 $\mathsf{A.}\,HCOOH$

B. $(H_2N)_2CO$

 $\mathsf{C}.\,(CH_3)_3COH$

D. CH_3CHO

Answer: C::D



Single Correct Answer Type

1. The decreasing order of priority for the following functional group is:

 $I.-COOH II.-SO_3H$

III. -COOR IV. -COC1

A. (IV) > (III) > (II) > (I)

 $\mathsf{B.}\left(I\right)>\left(II\right)>\left(III\right)>\left(IV\right)$

 $\mathsf{C.}\left(II\right)>\left(I\right)>\left(III\right)>\left(IV\right)$

 $\mathsf{D}.\left(IV\right)>\left(III\right)>\left(I\right)>\left(II\right)$

Answer: B



2. The decreasing order of prority for the following functional group is

I. $C \equiv N$ II. $-CONH_2$

III. \triangleright IVgt -CHO

A. (II) > (I) > (IV) > (III)

B.
$$(III) > (IV) > (I) > (II)$$

$$\mathsf{C.}\left(I\right)>\left(II\right)>\left(IV\right)>\left(III\right)$$

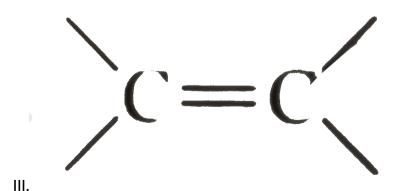
Answer: A



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3. The decreasig order of priority for the following functional groups is:

I.
$$-OH$$
 II gt $-C \equiv C$ $-$



 ${\sf IV.}-NH_2$

$$\mathsf{A.}\left(IV\right)>\left(I\right)>\left(II\right)>\left(III\right)$$

$$\mathsf{B.}\left(IV\right)>\left(I\right)>\left(III\right)>\left(II\right)$$

$$\mathsf{C.}\left(I\right)>\left(II\right)>\left(IV\right)>\left(III\right)$$

$$\mathsf{D}.\left(I\right)>\left(IV\right)>\left(III\right)>\left(II\right)$$

Answer: C



4. The number of 1° , 2° , and $3^{\circ}H$ atoms in 2,5,6-trimethy1 octane, respectively,is

A. 16,5,3

B. 15,5,3

C. 16,6,3

D. 15,5,2

Answer: B



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- **5.** The number of $1^{\circ}, 2^{\circ}$, and $3^{\circ}H$ atoms in 3-ethy1-5-methy1 heptane, respectively , is:
 - A. 12,8,1
 - B. 14,4,2
 - C. 12,6,2
 - D. 12,8,2

Answer: D



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6. The number of σ - and π -bond in hexan-2,4-diol, respectively,is:

- A. 18,2
- B. 17,2
- C. 17,1
- D. 18,1

Answer: B



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7. The number of σ - and π -bond in 5-oxohexanoic acid, respectively, is:

- A. 18,2
- B. 18,1
- C. 17,2
- D. 17,1



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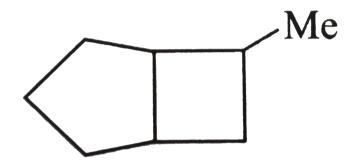
8. Which of the following is correctly named?

A. a.
$$O_2N$$
 O_2

D.

В.

9. The systematic naming of the following cycloalkane is



- A. 6-Methy1 bicyclo [3.2.0] heptane
- B. 7-Methy1 bicyclo [3.2.0] heptane
- C. 2-Methy1 bicyclo [3.2.0] heptane
- D. 3-Methy1 bicyclo [3.2.0] heptane

Answer: A



10. In which of the following reactions, the principal group loses its preferences?

c. I, II, III

d. I, II

II. Me

COOH

$$\Delta$$
 $-CO_2$

Me

Me

Me

COOH

 Δ
 $-CO_2$

Me

COOH

COOH

COOH

COOH

COOH

COOH

b. I, II

A. I

B. I, II

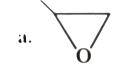
C. I, II, III

D. I,II

Answer: B



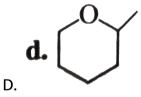
11. Which of the following is oxetane?



A



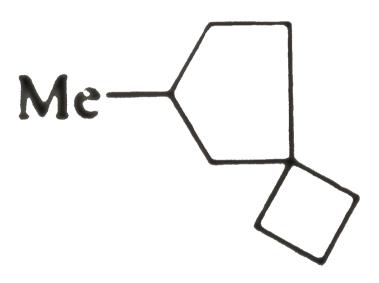
c. ______



Answer: B



12. The systematic nomencalutre of the following spiro-compound is:



- A. 2-Methy1 spiro [3.4] octane
- B. 3-Methy1 spiro [3.4] octane
- C. 6-Methy1 spiro [3.4] octane
- D. 7-Methy1 spiro [3.4] octane

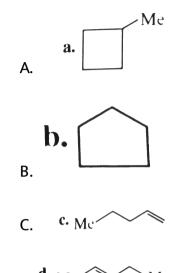
13. An alkane (A) having a molecular mass of 72 produces one monochlorination product. Compound (A) is:

Answer: C

C.



14. A compound (A) with molecular formula C_5H_{10} gives one monochlorination product. Compound (A) is:



Answer: B



15. Which of the following the following in a 3° amine?

A. Propan-2-amine

B. N- Methy1 ehtanamine

C. Ally1 amine

D. N, N-Diethy1 butan-1-amine

Answer: D



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A. t-Buty1 carbinol

B. 2-Methy1 propan-2-ol

16. Which of the following as a 3° alcohol?

C. 2-Methy1 butan-1-ol

D. isoamy1 alcohol

Answer: B



17. Which of the following is zerone?

A. MeOH

D. EtOH

Answer: A



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18. Which of the following is pyrogallol?

A.

Answer: A

D.

В.



- **19.** In 3-chloro cyclohexanol, the primary prefix is:
 - A. 3-Chloro
 - B. Cycle

C. an((e)
--------	-----

$$\mathsf{D.}-ol$$

Answer: B



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20. In 2-Chloro-3-methy1 hexanoic acid, the primary suffix is:

A. 2-Chloro-

B.-3-Methy1

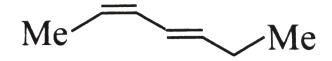
C. an (e)

D. oic acid

Answer: C



21. The correct name of the compound (I) is:



- A. $(E-2),\,(E-4),\,$ Hepta-2,4-diene
- B. $(Z-2),\,(Z-4)$, Hepta-2,4-diene
- C. (E-2), (Z-4), Hepta-2,4-diene
- D. (Z-2), (E-4), Hepta-2,4-diene

Answer: D



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22. Which of the following is not a cumulated diene?

A. Hexa-1,2-diene

B. Hexa-2,3-diene

- C. Penta-2,3-diene
- D. Penta-1,3-diene

Answer: D



- **23.** Which of the following has only 1° and $2^{\circ}C$ atoms?
 - A. 2-Methy1 butane
 - B. Butane
 - C. 2,2-Dimethy1 butane
 - D. 2,2,3,3-Tetramethy1 pentance

Answer: C



24. The IUPAC name of viny1 acetylene is:

- A. Pent-1-en-4-yne
- B. Pent-4-yn-1-ene
- C. But-1-en-3-yne
- D. But-1-yn-3-ene

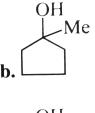
Answer: C

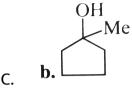


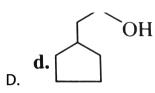
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25. Which of the following structures represents cyclopenty1 methy1 carbino1?

$$\mathbf{a.} \qquad \mathbf{CH}_3$$







Answer: C

В.



26. The IUPAC name of acrolein is:

A. But-2-enal

B. Prop-2-enal

C. But-3-enal

D. 2-Methy1 prop-2-enal

Answer: B



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27. The IUPAC name of the following compound is

$$\left(\begin{array}{c} NC \\ CN \end{array}\right)$$

- A. Propane-1,2,3-tricarbonitrile
- B. 3-Cyanopentane1,5-dinitrile
- C. Pentane-1,3,5-trinitrile
- D. All

Answer: A



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28. Which of the following is not the name of CH_3NC ?

- A. (a) Methyl isocyanide
- B. (b) Aceto isonitrile
- C. (c) Methyl carbyl amine
- D. (d) Acetronitrile

Answer: D



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29. The IUPAC name of Ph-CN is :

- A. (a) Phenyl cyanide
- B. (b) Benzonitrile
- C. (c) Benzene nitrile
- D. (d) All

Answer: C



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30. Give the IUPAC name of :

- A. 2,2-Dimethy1-3-propy1-4-isopropy1 heptane
- B. 4-Isoproy11-5-t-buty1 octane
- C. 4-t-Buty1-5-isoprpy1 octane
- D. 2-Methy1-3-propy1-4-isopropy1 heptane

Answer: C



- **31.** Which of the following statements is wrong for homologous series?
 - A. (a) All members have a general formula.
 - B. (b) All members have the same functional group.
 - C. (c) All members have the same chemical properties.
 - D. (d) All members have the same physical properties.

Answer: D



- **32.** The alkane which has only $1^{\circ}H$ atoms is:
 - A. Neopentane
 - B. Isopentane
 - C. Pentane
 - D. 2,2-Dimethy1 butane

Answer: A



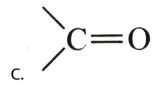
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33. Which group is always taken as a substituent in the IUPAC

system of nomenclauture?

$$A.-NO_2$$

$${\rm B.}-C\equiv N$$



 $\mathsf{D.}-NH_2$

Answer: A



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34. IUPAC name of ${(CH_3)}_3C-CH=CH_2$ is

A. 2,2-Dimethy1 but-3-ene

B. 2,2-Dimethy1 pent-4-ene

C. 3,3-Dimethy1 but-1-ene

D. Hex-1-ene

0

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35. Which of the following compounds is 2,2,3-trimethylhexane?

Answer: D



36. The bond between carbon atom (1) and carbon atoms (2) in compound , $N\equiv {C\over (1)}-{C\over (2)}H=CH_2$ involves the overlapping between

- A. sp^2 and sp^2
- B. sp^3 and sp
- C. sp and sp^2
- D. sp and sp

Answer: C



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37. If two compounds have the same empirical formula but different molecular formula, they must have

A. Different percentage composition

B. Different molecular weights

C. Same velocity

D. Same vapour density

Answer: B



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38. The number of σ -and π -bond in 1-butene-3-yne is:

A. 5σ and 5π

B. 7σ and 3π

C. 8σ and 2π

D. 6σ and 4π

Answer: B



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39. Which of the following compounds has isopropyl group?

- A. 2,2,3,3-Trimethy1 pentae
- B. 2,2-Dimethy1 pentane
- C. 2,2,3-Trimethy1 pentane
- D. 2-Methy1 pentane

Answer: D



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40. The correct IUPAC name of the compound is:

$$CH_3CH_2-C=CH-CH-CH_2-CH_3$$

 $CH_3CH_2-CH-CH_2-CH_2-CH_3$
 $CH_3CH_2-CH-CH_2-CH_2-CH_3$

A. 5,6-Dimethy1-3-methy1 dec-4-ene

B. 5,6-Dimethy1-8-methy1 dec-6-ene

C. 6-Buty1-5-ethy1-3-methy1 oct-4-ene

D. 2,4,5-Triethy1-3-nonene

Answer: A



41. The hybridisation of C atoms in (C-C) single-bond of

$$H-C\equiv C-CH=CH_2$$
 is :

A.
$$H_2C=CH-C\equiv CH$$

$$\mathsf{B}.\,HC\equiv C-CH_2-C\equiv CH$$

$$\mathsf{C.}\,H_2C=C=C=CH_2$$

D. $sp-sp^2$

Answer: C



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42. Which of the following represents the given sequence of hybridisation of carbon atoms from left to right sp^2 , sp^2 , sp, sp?

A.
$$H_2 = CH - C = CH$$

B.
$$HC \equiv C - CH_2 - C \equiv CH$$

C.
$$H_2C = C = CH_2$$

D d. H₂C CH₂

Answer: A



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43. The IUPAC name of C_6H_5COC1 is :

A. Benzoy chloride

B. Benzene chloro ketone

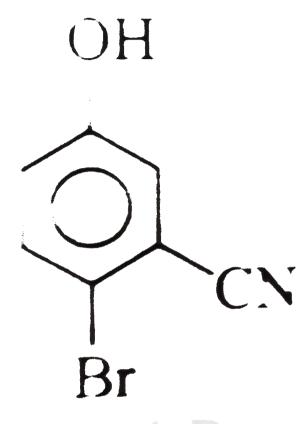
C. Benzene carbony1 chloride

D. Chloropheny1 ketone

Answer: A



44. The IUPAC name of the following compound is



- A. 4-Bromo-3-cyanophenol
- B. 2-Bromo-5-hydroxy benzo nitrile
- C. 2-Cyano-4-hydroxy bromo benzene
- D. 6-Bromo-3-hydroxy benzonitrile

Answer: B



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Assertion Reasoning Type

1. Assertion (A): Pentane and 2-methy1 pentane are homolo-gues.

Reason (R): Pentane is a straight-chain alkane, while 2-methy1 pentane is a branched-chain alkane.

- A. If both (A) and (R) are correct and (R) is the correct explanation for (A).
- B. If both (A) and (R) are correct and (R) is not the correct explanation.
- C. If (A) is correct and (R) is incorrect.
- D. If both (A) and (R) are incorrect.

Answer: B



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2. Assertion (A) : All the C atoms of but-2-ene lie in one plane

Reason (R): Double-bond C atoms are sp^2 -hypbridised.

A. If both (A) and (R) are correct and (R) is the correct

explanation for (A).

- B. If both (A) and (R) are correct and (R) is not the correct explanation.
- C. If (A) is correct and (R) is incorrect.
- D. If both (A) and (R) are incorrect.

Answer: A



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3. Assertion (A): The IUPAC name of the citric acid is 2-hydroxy-propane-1,2,3-tricarboxylic acid

Reason (R): When an unbranched C atom directly linked to more than two like-functional groups, then it is named as a dervative of the parent alkane which does not include the C of the functional groups.

- A. If both (A) and (R) are correct and (R) is the correct explanation for (A).
- B. If both (A) and (R) are correct and (R) is not the correct explanation.

C. If (A) is correct and (R) is incorrect.

D. If both (A) and (R) are incorrect.

Answer: A



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4. Assertion (A): Rochelle's salt is used as a complexing agent in Tollens reagent.

Reason (R). Sodium potassium salt of tartaric acid is known as Rochelle's salt. The IUPAC name of Rochelle's salt is sodium potassium-2,3-dihydroxy butance-1,4- diote.

A. If both (A) and (R) are correct and (R) is the correct

B. If both (A) and (R) are correct and (R) is not the correct

explanation.

explanation for (A).

C. If (A) is correct and (R) is incorrect.

D. If both (A) and (R) are incorrect.

Answer: D

1,3-diene.



5. Assertion (A): The IUPAC name of isoprene is 2-methy1 buta-

Reason (R): Isoprene unit is a monomer of natural rubber

A. If both (A) and (R) are correct and (R) is the correct explanation for (A).

B. If both (A) and (R) are correct and (R) is not the correct explanation.

C. If (A) is correct and (R) is incorrect.

D. If both (A) and (R) are incorrect.

Answer: B



Fill In The Blanks Type

- **1.** The compounds having both $sp-\$ and sp^2 -hybrideised C atmos is (propane, propdiene.
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2ring is the most strained. (Cyclopropane, Cyclobutane, Cyclopentane)
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2. The terminal Coste main buttons is buttons in
3. The terminal C atom in butane ishybridised.
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4. A vic diol has two hydroxy1 group on C atoms
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5. The $IUPAC$ name of succinic acid is
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Analytical And Descriptive Type

1. Give the IUPAC name of the following compound.

$$CH_2 = CH - CH(CH_3)_2$$



2. Give the IUPAC name of the following compound.



3. Give the IUPAC name of the following compound.

