

## **CHEMISTRY**

## BOOKS - PATHFINDER CHEMISTRY (BENGALI ENGLISH)

## SOME BASIC CONCEPTS OF CHEMISTRY

**Question Bank** 

1. A 2.0 g sample containing  $Na_2CO_3$  and  $NaHCO_3$  loses 0.148 g when heated at  $300^{\circ}C$ , the temperature at which  $NaHCO_3$  decomposes to  $Na_2CO_3$ ,  $CO_2$  and  $H_2O$ . What is the percentage of  $Na_2CO_3$  in the mixture?



**2.** If 4g NaOH dissolves in 36 g of  $H_2O$ , calculate the mole fraction of each component

in the solution. Also determine the molarity of solution ( specific gravity of solution is  $1gml^{-1}$ ).



**View Text Solution** 

3. What is limiting reagent?



**View Text Solution**