



BIOLOGY

BOOKS - MAXIMUM PUBLICATION

BIOMOLECULES



1. Many elements are found in living organisms

either free or in the form of compounds. One

of the following is not found in living

organisms.

A. Silicon

B. Magnesium

C. Iron

D. Sodium

Answer:

2. Aminoacids, as the name suggests, have both an amino group and a carboxyl group in their structure. In addition, all naturally occurring aminoacids (those which are found in proteins) are called L-aminoacids.From this, can you guess from which compound can the simplest aminoacid be made?

A. Formic acid

B. Methane

C. Phenol

D. Glycine

Answer:

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3. When we homogenise any tissue in an acid the acid soluble pool represents

A. Cytoplasm

B. Cell membrane

C. Nucleus

D. Mitochondria

Answer:

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4. Many organic substances are negatively charged e.g., acetic acid, while others are positively charged e.g., ammonium ion. An aminoacid under certain conditions would have both positive and negative charges simultaneously in the same molecule. Such a

form of aminoacid is called

- A. Positively charged form
- B. Negatively charged form
- C. Neutral form
- D. Zwitterionic form

Answer:



5. Sugars are technically called carbohydrates, referring to the fact that their formulae are only multiple of $C(H_2O)$. Hexoses therefore have six carbons, twelve hydrogens and six oxygen atoms. Glucose is a hexose.Choose from among the following another hexose.

A. Fructose

B. Erythrose

C. Ribulose

D. Ribose

Answer:

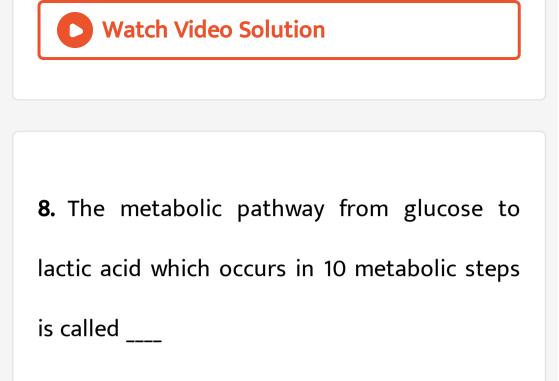


6. Fill the gap :- Inhibition of succinic dehydrogenase by Malonate is an example for

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7. Name the phospholipid found in cell

membrane.



9. Energy currency in living systems is

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10. Find the odd one out. Rubber, Amino Acids,

Drugs, Pigments



11. Arrange the following Carbohydrates in the order of increasing complexity of chemical structure. Glucose, Oligosaccharides, Lactose, Starch.



12. One full turn of the DNA double helix would

involve.

- A. 8 base pairs
- B. 10 base pairs
- C. 20 base pairs
- D. 12 base pairs

Answer:

13. Match the column A with B and C.

A - Monomer	B - chemical bond	C - Polymer
Glucose	phosphor diester	Nucleic acid
Amino acids	Glycoside	Carbohydrate
Nucleotide	Peptide	Protein



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14. Add suitable word in the gap. a)A protein molecule is a polymers of __ b)Nucleic acids are polymers of c)DNA has instead of uracil.d)Single stranded structure is found

A. Aminoacids

B. Nucleotides

C. Thymine

D. RNA

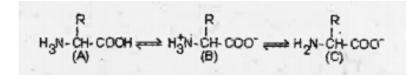
Answer:

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15. The structure of amino acids in solutions of

different PH are given above. Identify the





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16. $CO_2 + H_2O ightarrow H_2CO_3$ Name the enzyme

catalyse this reaction

17. Observe the table and fill the blanks from

the brackets (Collagen, Cholesterol, chitin,

	Protein	Carbohydrates	Fats
	Trypsin		
Lecithin)		Glycogen	

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18. Building blocks of proteins are amino acids and that of polysaccharides are monosaccharides. Many polysaccharides are called homopolymers but all proteins are known as heteropolymers. a) Do you agree

with the above statement? Give reason.



19. Building blocks of proteins are amino acids and that of polysaccharides are monosaccharides. Many polysaccharides are called homopolymers but all proteins are known as heteropolymers. b) Give one example for a homopolymer.

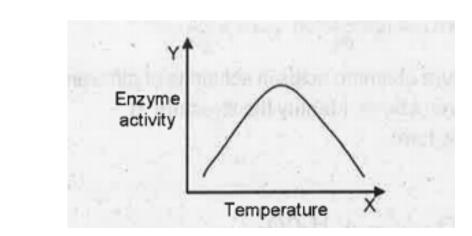


20. Classify the following as polypeptide and

polysac charide.(Insulin, Glycogen, Chitin,

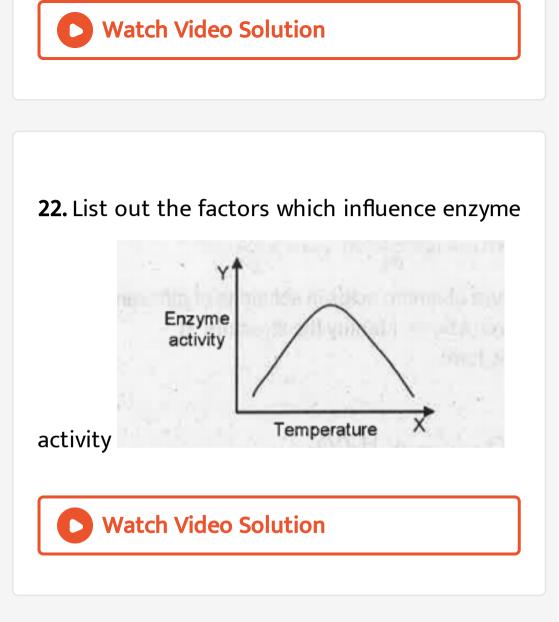
Paper)

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21.

Observe the graph and comment.



23. Write the general names of the following

a) Enzymes catalyzing the linking together of

two compounds



24. Write the general names of the following

b) Enzymes that catalyze removed of groups

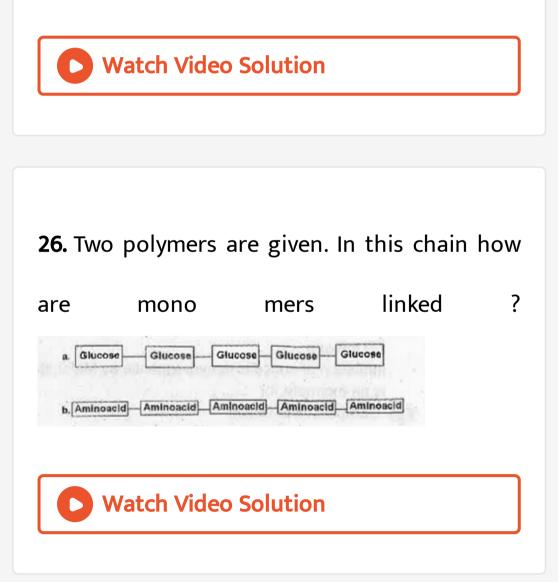
from substrates by mechanism other than

hydrolysis leaving double bonds.



25. Find out the differences between DNA and

RNA and fill the table given below.



27. The following graph shows the relationship between subtrate concentration and rate of enzymatic reaction. What is the effect of subtrate concentration on rate of reaction ? And identify A.

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28. The following graph shows the relationship between subtrate concentration and rate of enzymatic reaction. What is the effect of

subtrate concentration on rate of reaction ?

And identify A.

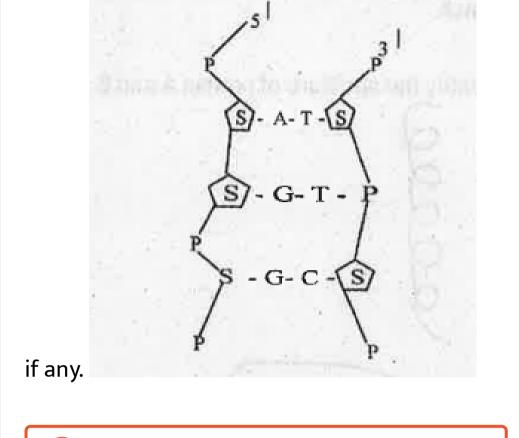


29. Identify the class of enzyme which catalyse

the following reaction

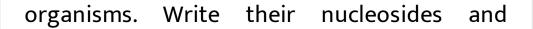
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30. A portion of DNA double helix is given below. Copy the diagram and make correction



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31. Adenine Guanine, thymine, Cytosine, Uracil, are the nitrogen bases present in the living



nucleotides.



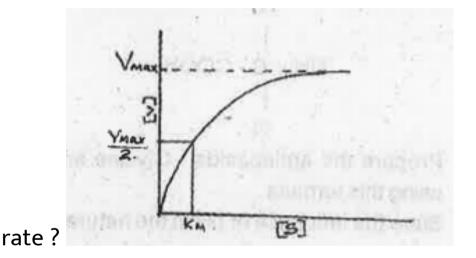
32. Prepare a graph showing the action of

temperature on enzyme activity

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33. What is optimum temperature ?

34. If we add more substrate after reaching Vmax What will be the effect in the reaction

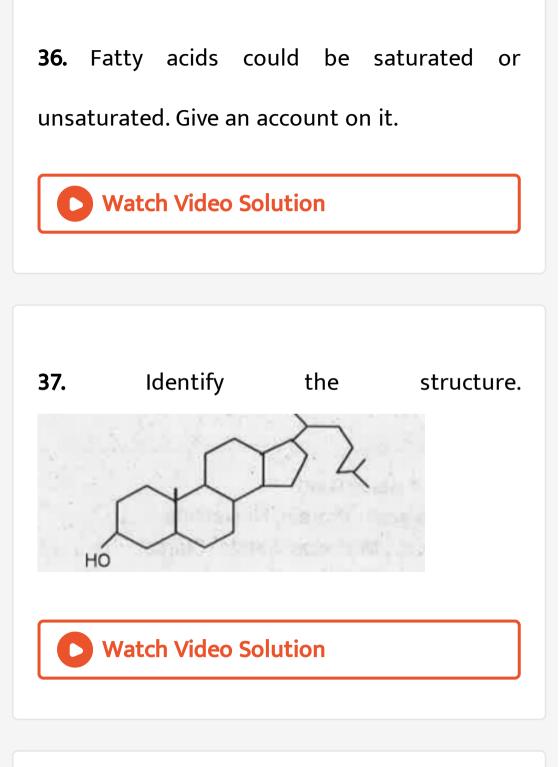




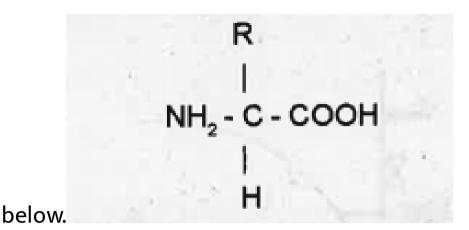
35. The table shows the percentage composition of different elements of a living matter and a non living matter represented as

X and Y. Identify X and Y.

Element	% composition of		
Element	X	Y	
Hydrogen	0.14	0.5	
Carbon	0.03	18.5	
Oxygen	46.6	65.0	
Calcium	36	1.5	
Şilicon	27.7	Negligible	



38. General Formula of amino acid is given

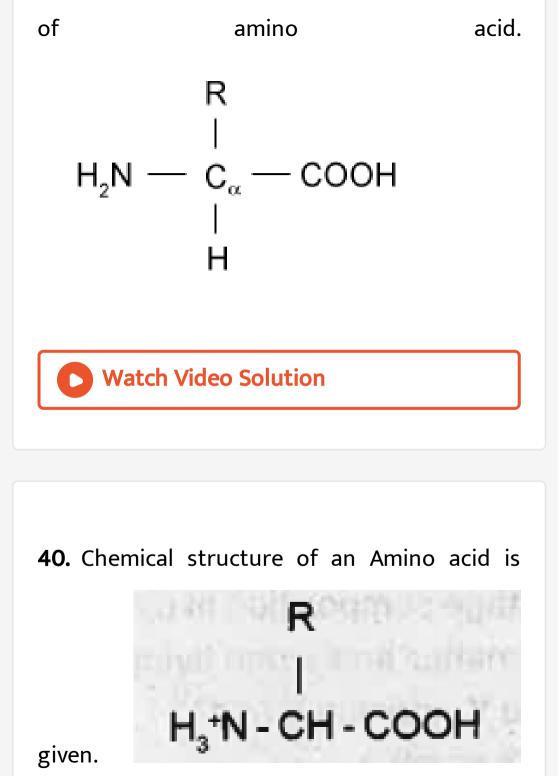


Prepare the aminoacids - Glycine and Serine

using this formula

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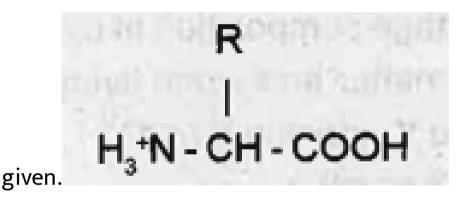
39. General Formula of amino acid is given below. State the influence of pH in the nature



Prepare its Zwitter lonic form.



41. Chemical structure of an Amino acid is



What happens to the Zwitter lon if it is placed

acidic medium?

42. Write the missing word. Amino acid Protein, Nucleotide:..... Watch Video Solution 43. Write the missing word.Glucose Monosaccharide, Starch:.... Watch Video Solution

44. Write the missing word. Adenine : Thymine,

Guanine:....

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45. Write the missing word.Animals: Glycogen.

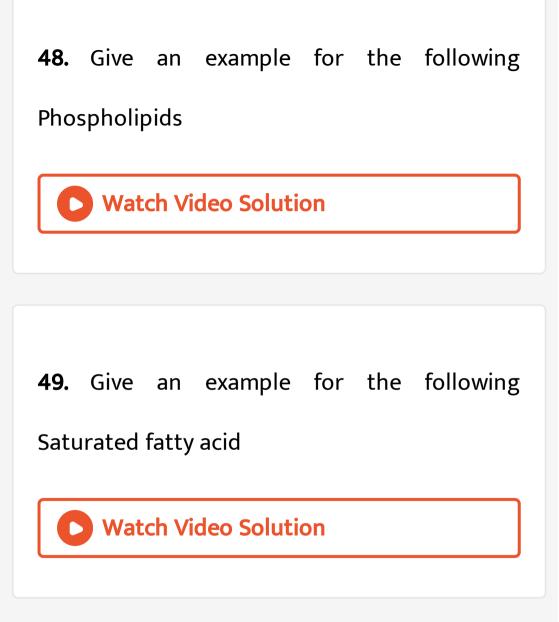
Plants:.....



46. Write an example for a protein with quatemary structure.

47. Proteins carry out many functions in living organisms. List them. Sorme protein functions

as enzymes



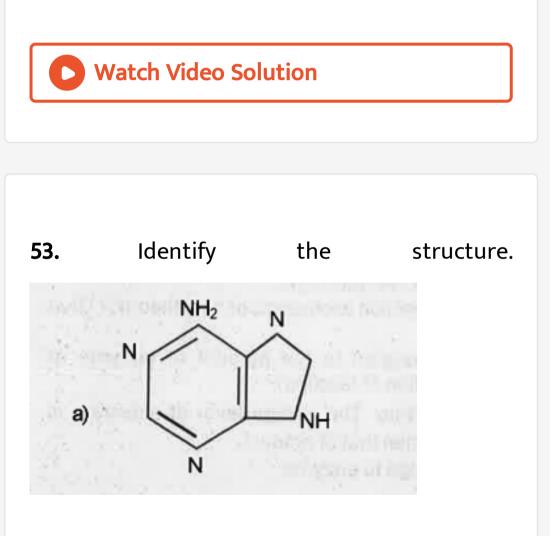
50. Give an example for the following Nucleoside
Watch Video Solution
51. Give an example for the following

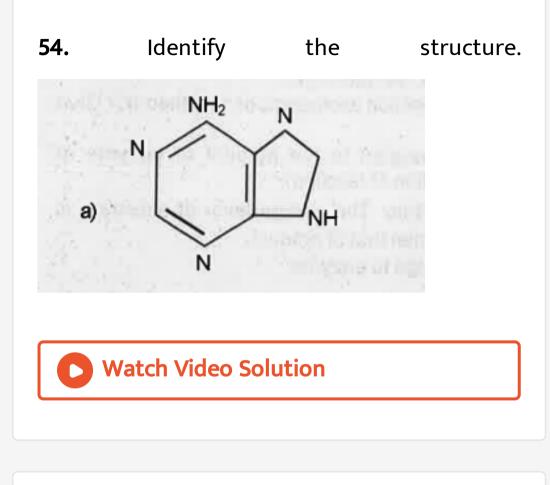
Nucleotide



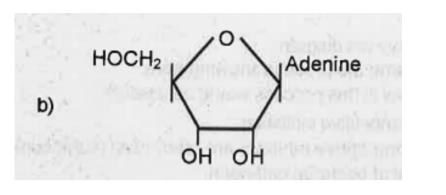
52. B-DNA is the most common DNA. Write the

salient features of this DNA.



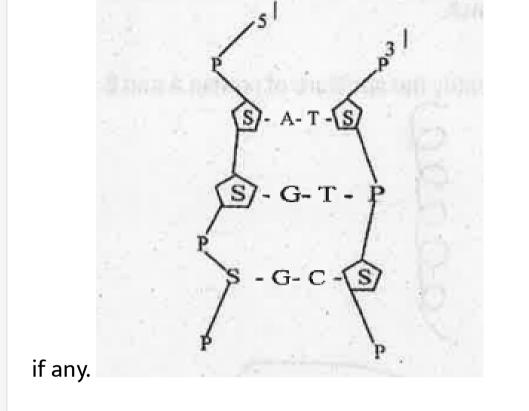


55. The given structure showing the molecule give ready energy for biological activities.



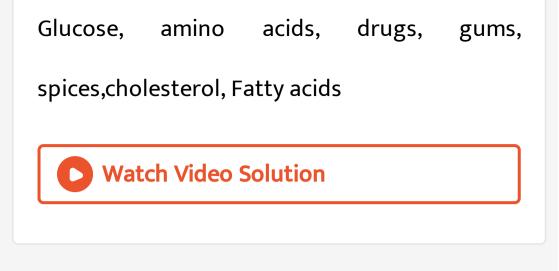
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56. A portion of DNA double helix is given below. Copy the diagram and make correction

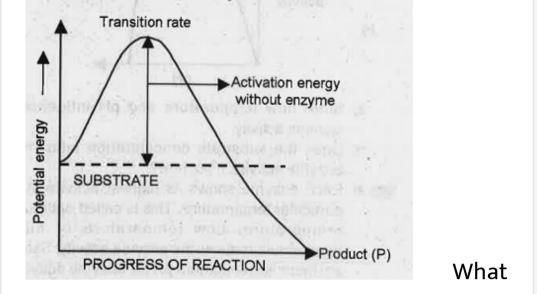


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57. Classify the given terms into two columns and give appropriate headings. rubber,



58. Progress of a chemical reaction and potential energy changes associated with it is plotted as curve.



happens to the activation energy of the substrate, when enzyme is added to the

reaction system?

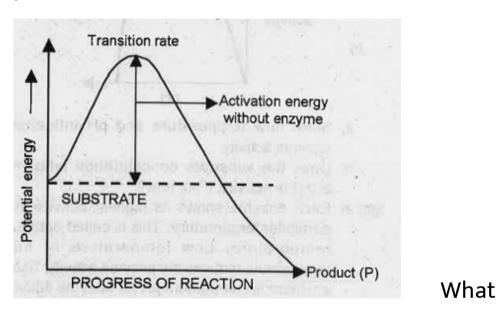


59. Progress of a chemical reaction and potential energy changes associated with it is

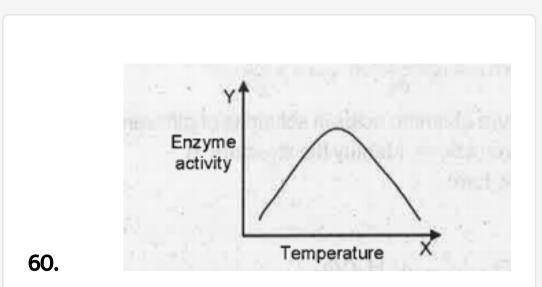
plotted



curve.



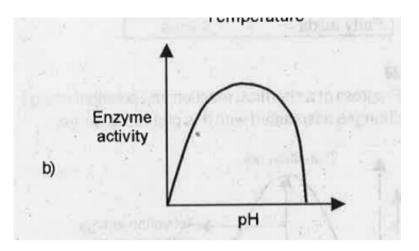
happens to the activation energy of the substrate, when enzyme is added to the reaction system?



Observe the graph and comment.



61. Observe the given graph a and b.



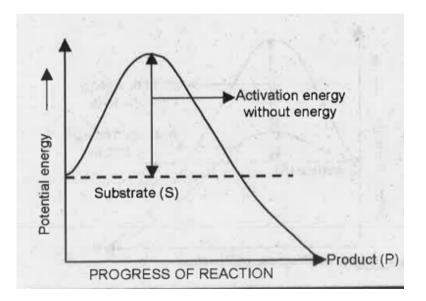
Does

the substrate concentration influence enzyme

activity. If so, how?



62. The graph shows the relation betweenpotential energy and progress of reaction.Observe the graph and answer the questions.What happen to the amount of enzyme atcompletion of reaction?





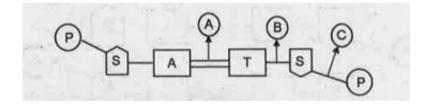
63. Fill in the blank columns with the correct

terms/sentences. 尾

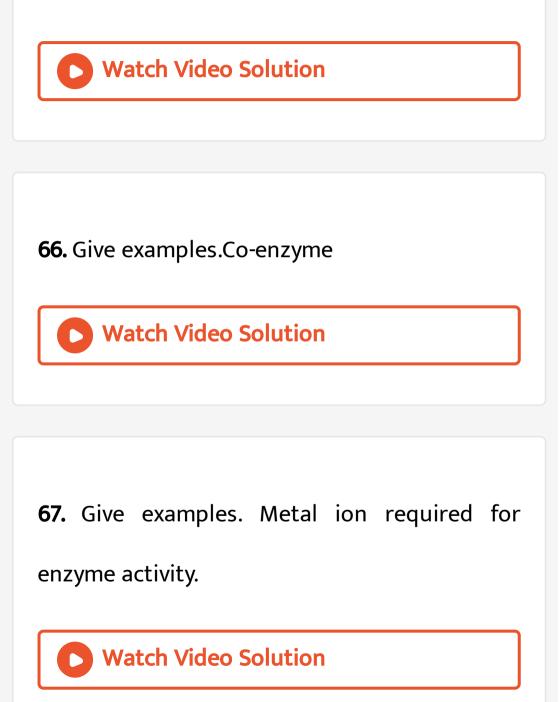
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64. Structure of a nucleotide pair in DNA is

showing below. Identify the bonds A, B and C.



65. Give examples. Prosthetic group



68. Give one word The nucleic acids that behave like enzymes

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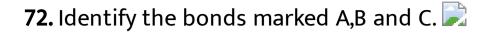
69. Give one word The organic compound

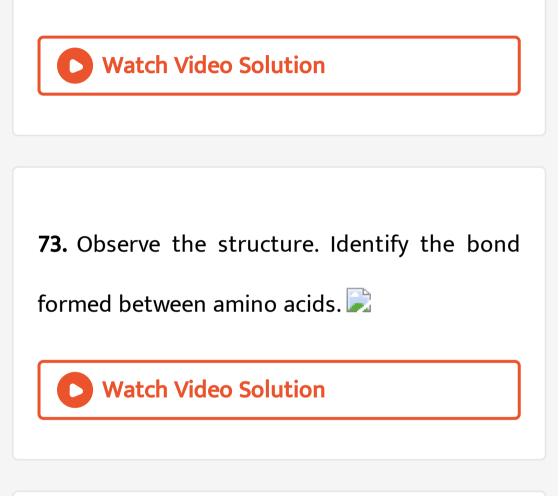
tightly bound to the apo enzyme.

70. Give one word The non protein organic compound that are not tightly bpund to the apoenzyme.



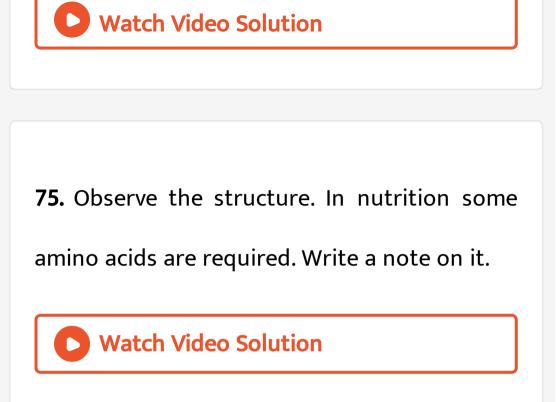
71. Give one word The protein portion of the enzyme.





74. Observe the structure. How many types of

amino acids are present in animal body?

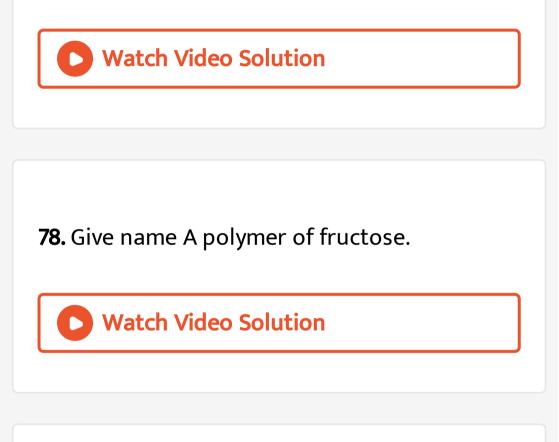


76. Give name The most abundant protein in animal.



77. Give name The most abundant protein in

the whole of the biosphere



79. Give name - Paper made from plant pulp

is.....





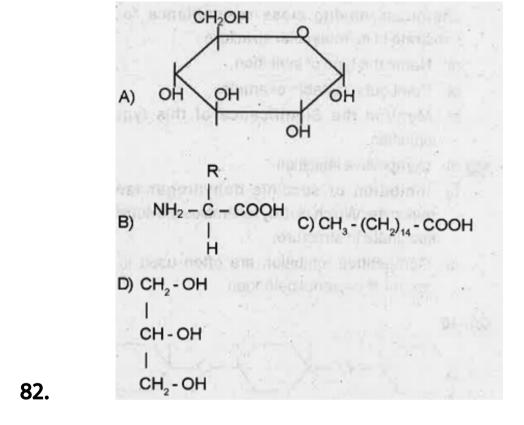
80. Give name Complex polysaccharide present

in the exoskeleton of Arthropods



81. Give name - Polysaccharide with helical

structure



Identify the structures A, B, C and D.



83. Name the molecule formed by the esterification of one 'D' molecule and three 'C'molecule.



84. Match the column A and B. 戻



85. The activity of the enzyme is inhibited by certain chemicals having close resemblance to the substrate in its molecular structure. Name

the type of inhibition



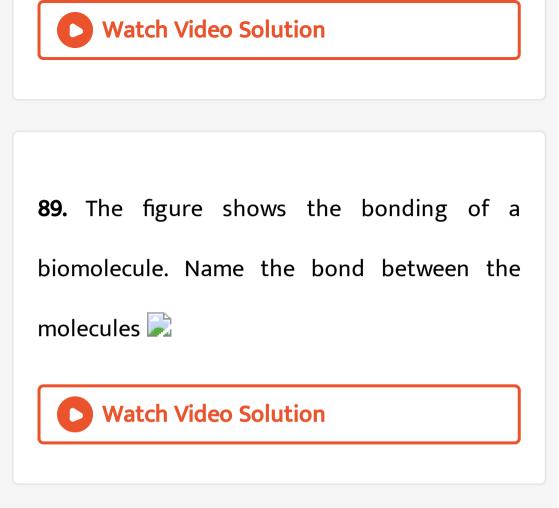
86. The activity of the enzyme is inhibited by certain chemicals having close resemblance to the substrate in its molecular structure. Point out a suitable example



87. The activity of the enzyme is inhibited by certain chemicals having close resemblance to the substrate in its molecular structure. Mention the Significance of this type of Inhibition.

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88. The figure shows the bonding of a biomolecule. Identify the monomer



90. The figure shows the bonding of a biomolecule. Name the macromolecule formed





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91. Name the macromolecule

formed

BaseNucleosideNucleotideAdenineAdenosineAdenylic acidGuanine....a....Guanylic acid....b...CytidineCytidilic acidThymine....c.........d.....

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92. Give one word The protein portion of the

enzyme.



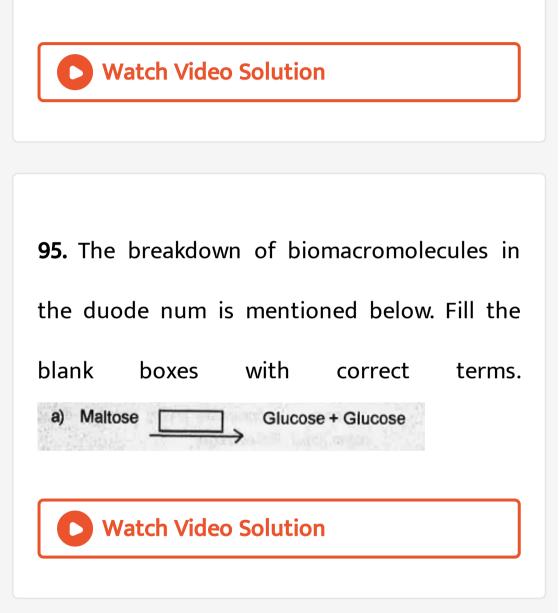
93. Non protein constituents called cofactors are bound to the enzyme to make the enzyme catalytic activity. What happens to the catalytic activity when the cofactor is removed from the enzyme ?

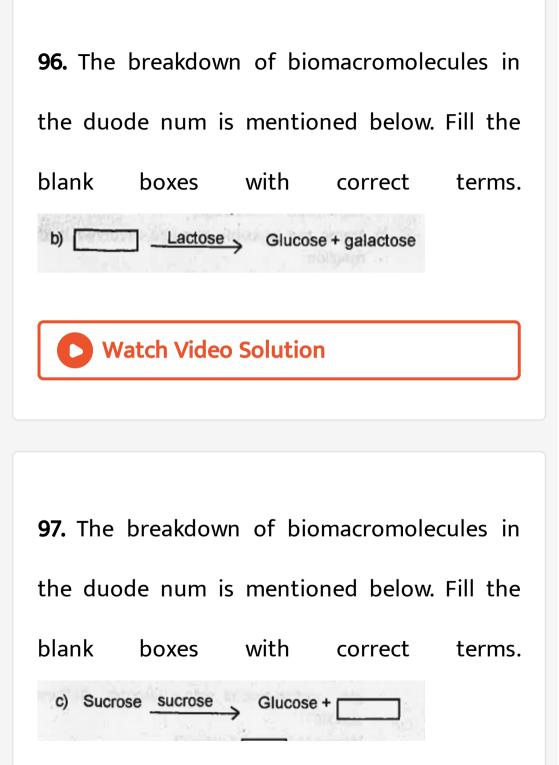
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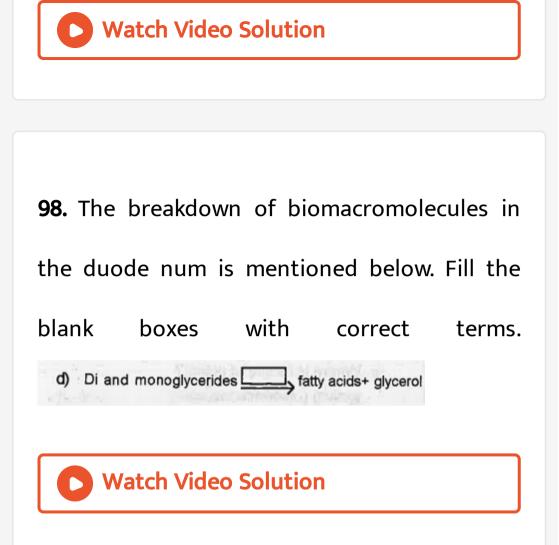
94. Non protein constituents called cofactors are bound to the enzyme to make the enzyme

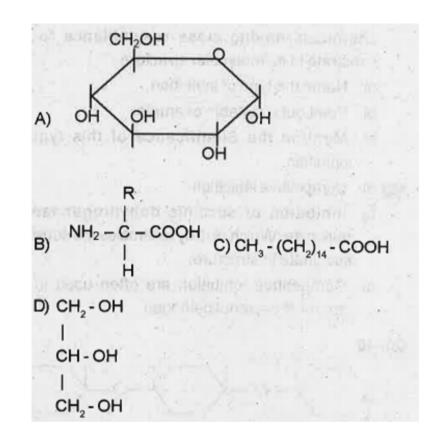
catalytic activity. Mention any two kinds of

cofactors with examples









99.

Identify the structures A, B, C and D.

100. Name the molecule formed by the esterification of one 'D' molecule and three 'C'molecule.



101. Carbohydrate: Sugars :: Proteins :

 102. Prepare an equation for a chemical

 reaction
 using
 the
 following

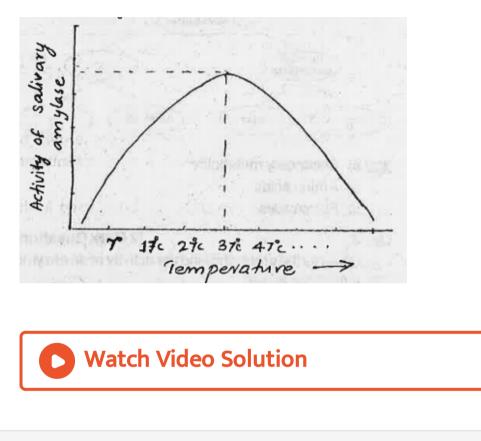
 components.Carbonic
 anhydrase,
 carbonic

 acid, water, carbondioxide.

103. Analyse the graph showing the activity of salivary amylase Which is the optimum temperature for salivary amylase obtained

from

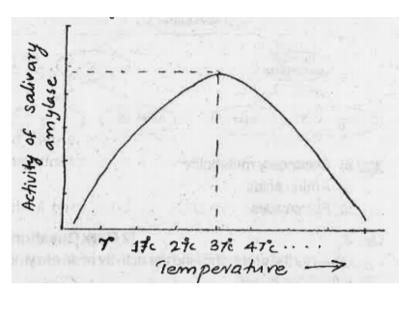
graph.



104. Analyse the graph showing the activity of salivary amylase Why the activity declines

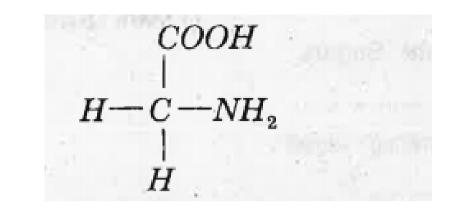
below

the



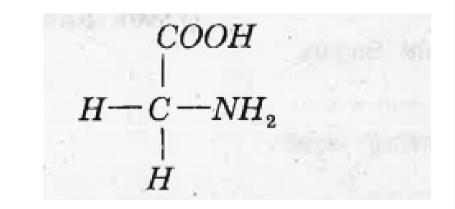
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value?



105.

Identify this compound.



Name the bond produced when another biomolecule of the same category combines with this.

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106.

COOH $C - - NH_{2}$ H^- Hlf a 107.

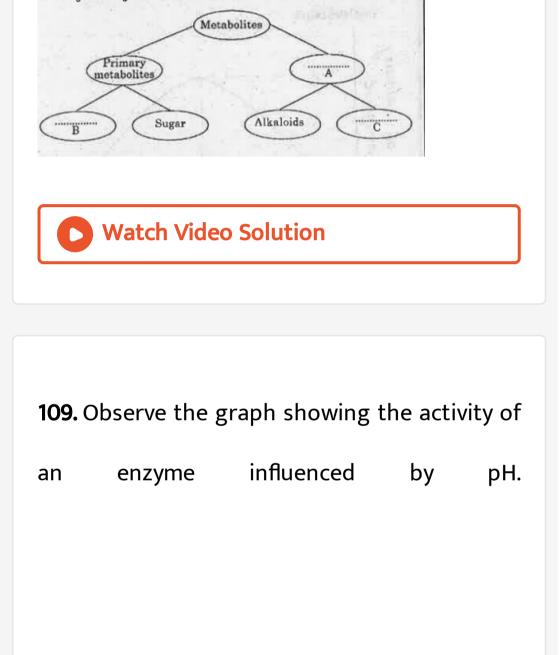
number of such molecules are bonded to-

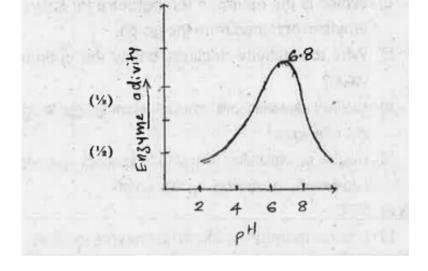
gether, what will be the resultant molecule?



108. Observe the following representation and

fill up the blanks appropriately





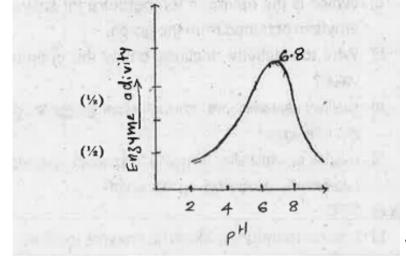
Name

the possible enzyme involved in this reaction.



110. Observe the graph showing the activity of

an enzyme influenced by pH.



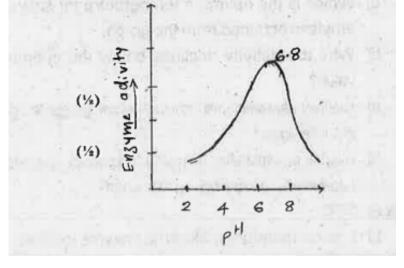
Where is

its site of action?



111. Observe the graph showing the activity of

an enzyme influenced by pH.



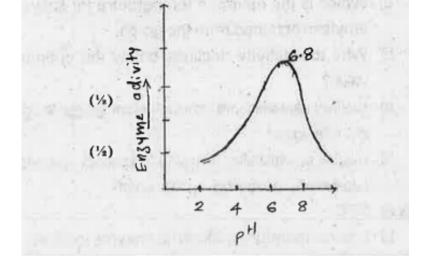
Mention

any other factor which affects this en zyme activity that results in a similar pattern of graph

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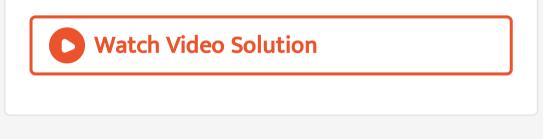
112. Observe the graph showing the activity of

an enzyme influenced by pH.

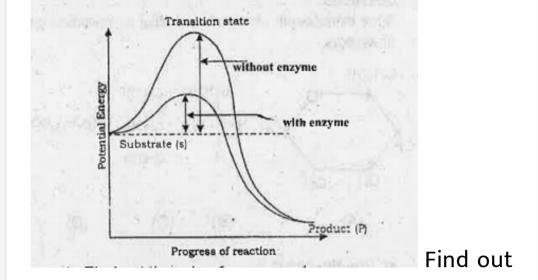


Name

another similar enzyme acting on the same substrate.



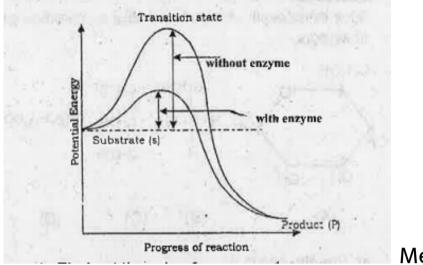
113. Observe the graph ad answer the following:



the role of enzyme,



114. Observe the graph ad answer the following:



Mention

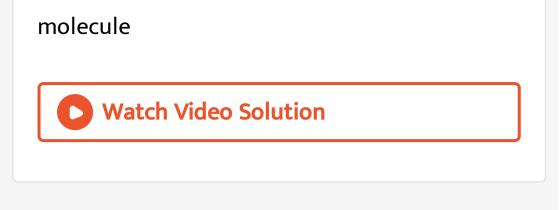
any two factors that influence the activity of

an enzyme and state their influence.



115. Name the chemical bonds formed between

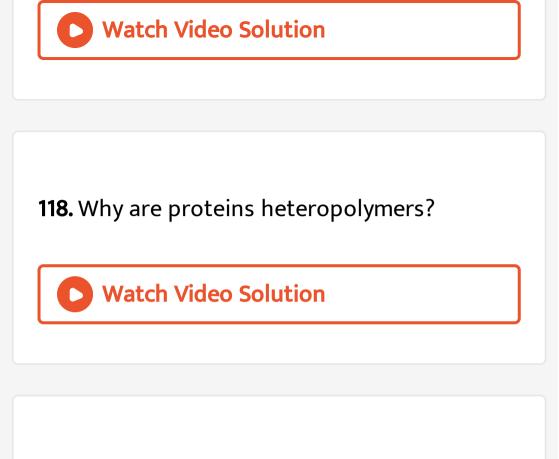
the following : Amino acids in a protein



116. Name the chemical bonds formed between the following : Sugar and phosphate in nucleic acids.

117. Distinguish between cofactor and

coenzyme with an example for each.

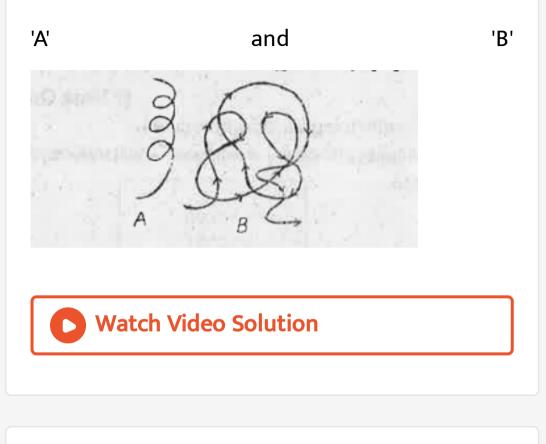


119. Identify the protein from the given list of

biomacromolecules and write its function.

(Cellulose, Starch, Antibody, Inulin)

120. Identify the type of protein structures of

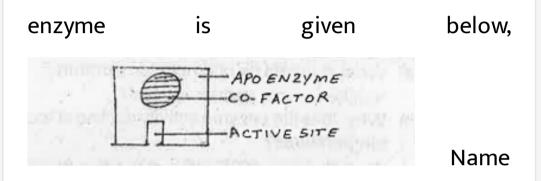


121. Symbolic representation of a functional enzyme is given below, Write one difference between a co-factor and an Apoenzyme





122. Symbolic representation of a functional



the different types of co-factors.



123. Symbolic representation of a functional

enzyme

is

given

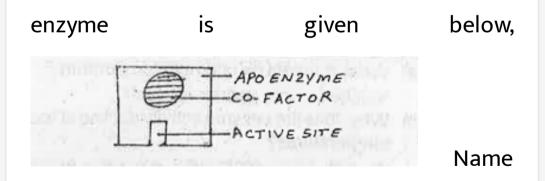
below,

APO ENZYME CO-FACTOR ACTIVE SITE What is

the -factor for the enzyme, Carboxypeptidase?

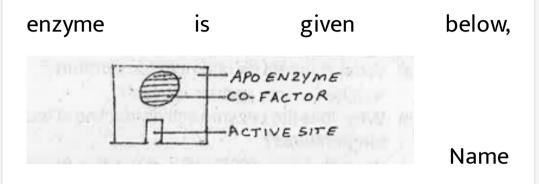
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124. Symbolic representation of a functional



the different types of co-factors.

125. Symbolic representation of a functional



the different types of co-factors.

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126. Symbolic representation of a functional

enzyme

is

given

below,

APO ENZYME ACTOR VE SITE What is

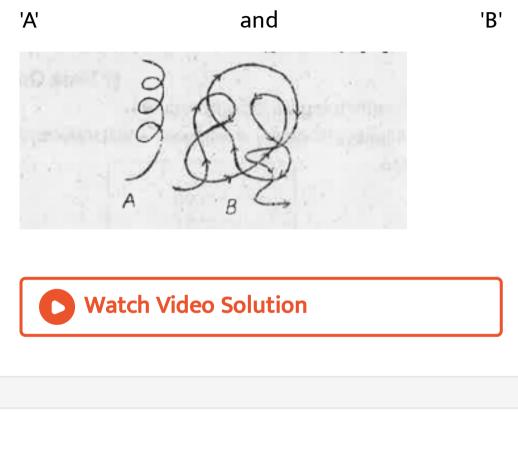
the -factor for the enzyme, Carboxypeptidase?

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127. Identify the protein stuctures. (A) and (b)

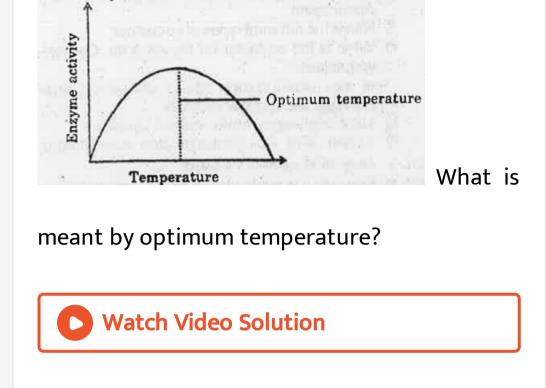
from the following figure. 属

128. Identify the type of protein structures of



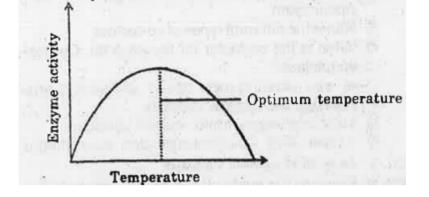
129. Analyze the graph showing the activity of

an enzyme, influenced by temperature.



130. Analyze the graph showing the activity of

an enzyme, influenced by temperature.



Why

does the enzyme activity decline at too high

temperatures?

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131. Complete the following sequence with appropriate words.Amino acids:..(a)... bond: Proteins.. (b).. Glycosidic bond: Polysaccharides

132. Metabolites are organic compounds constantly uti lized in various metabolic activities in the cells. What are the two types of metabolites in cells?

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133. Metabolites are organic compounds constantly uti lized in various metabolic

activities in the cells. Given an example for

each type of metabolites.

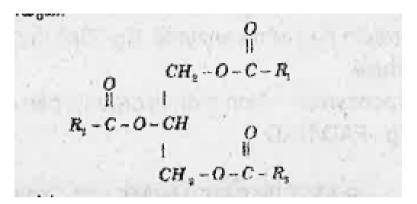


134. Enzymes are biocatalysts which regulate various biochemical reactions. Illustrate the following reactions. E +S ES rarr E + P

135. Identify the given biomolecule that comes

under



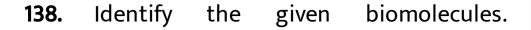


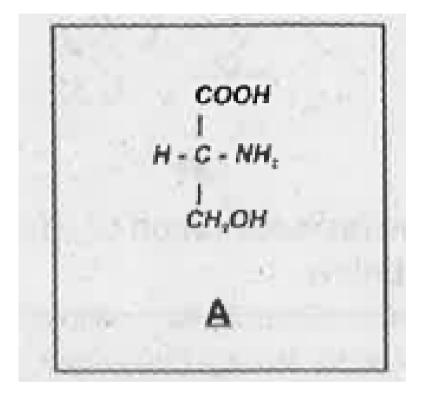


136. Name the biomacromolecule (polymer) in

which the peptide bond is present.

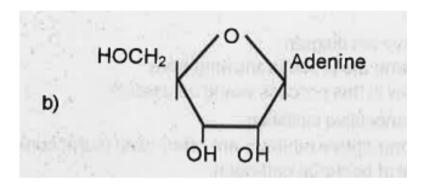
137. Name the chemical bonds formed between the following : Sugar and phosphate in nucleic acids.



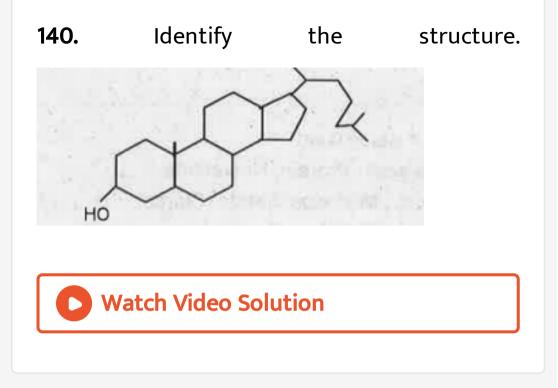




139. The given structure showing the molecule give ready energy for biological activities.Identify the structure.







141. Select the wrongly matched pair from the following:

Collagen substance കൊളാജൻ സബ്സ്റ്റൻസ് Insulin ഇൻസുലിൻ Antibody ആന്റിബോഡി Trypsin ട്രിപ്സിൻ Intercellular ground ഇന്റർസെല്ലുലാർ ത്രൗണ് Hormone ഹോർമോൺ Sensory reception സെൻസറി റിസപ്ഷൻ Enzyme

രാസാഗ്നി



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142. Proteins have primary structure. If you are

given a method to know which amino acid is at

either of the two termini (ends) of a protein,

can you connect this information to purity or

homogeneity of a protein?



143. Find out and make a list of proteins used

as therapeutic agents. Find other applications

of proteins (e.g., Cosmetics etc)



144. Explain the composition of triglyceride.



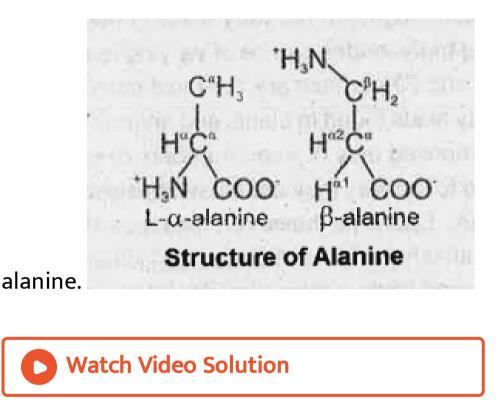
145. Can you describe What happens when milk is converted into curd or yoghurt, from your understanding of proteins?

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146. Can you attempt building models of biomolecules using commercially available atomic models (Ball and Stick models)?

147. Attempt titrating an amino acid against a weak base and discover the number of dissociating (ionizable) functional groups in the amino acid.

148. Draw the structure of the amino acid,



149. What are gums made of? Is Fevicol different?



150. Explain the basic structure of a nucleotide.



151. Enzyme often have additional parts in their structures that are made up of molecules other than proteins. When this additional chemical part is an organic molecule, it is called

A. cofactor

B. coenzyme

C. substrate

D. Both (a) and (b)

Answer: A

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152. Nucleotides are formed by

A. sugar and phosphate

B. purine, pyrimidine and phosphate

C. purine, pyrimidine, sugar and phosphate

D. pyrimidine, sugar and phosphate

Answer: B

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153. The effectiveness of an enzyme is affected

least by -

A. Temperature

B. Concentration of the substrate

C. Original activation energy of the system

D. Concentration of the enzyme

Answer: C

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154. Which is an organic compound found in most cells?

A. Glucose

B. Water

C. Sodium chloride

D. Oxygen

Answer: A

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155. Enzymes that catalyse inter-conversion of

optical, geometrical or positional isomers are

A. ligases

B. lyases

C. hydrolases

D. isomerases

Answer: D

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156. Which one of the following is wrongly matched?

A. Fungi - Chitn

B. Phospholipid - Plasma membrane

C. Enzyme- Lipopolysaccharide

D. ATP-Nucleotide derivative

Answer: C

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157. An organic substance bound to an enzyme

and essential for its activity is called

A. coenzyme

B. holoenzyme

C. apoenzyme

D. isoenzyme

Answer: A

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158. The 'Repeating unit of glycogen is

A. fructose

B. mannose

C. glucose

D. galactose

Answer: C



159. Feedback inhibition of enzymes is affected

by which

A. Enzyme

B. Substrate

C. End products

D. Intermediate end products

Answer: C

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160. The enzyme, which combines with non-

protein part

A. coenzyme

B. holoenzyme

C. apoenzyme

D. prosthetic group

Answer: B



161. Benedict's reagent test is conducted to

confirm the presence of

A. polysaccharides like starch

B. lipids

C. reducing sugars

D. proteins

Answer: C



162. Which disaccharide has different linkage?

A. Maltose

B. Starch

C. Sucrose

D. Lactose

Answer: C

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163. Select the wrong statement

A. The building blocks of lipids are amino

acids

B. Majority of enzymes contain a non-

protein part called the prosthetic group

C. The thylakoids are arranged one above

the other like a stack of coins forming a

granum

D. Crossing over occurs at pachytene stage

ofmeiosis-l

Answer: A

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164. Which of the following is a disaccharide?

A. Glucose

- **B.** Fructose
- C. Sucrose
- D. Galactose

Answer: C



165. Lactose is composed of following components.

- A. glucose and fructose
- B. glucose and glucose
- C. glucose, fructose and galactose
- D. glucose and galactose

Answer: D

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166. The simple polyhydroxy ketone molecule

containing 3-7 carbons is a

A. disaccharide

- B. monosaccharide
- C. polysaccharide
- D. dipeptide

Answer: B



167. How many of the twenty two amino acids

are essential for children?

A. 6

B. 8

C. 10

D. 7

Answer: C

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168. Starch and cellulose are compounds of many units of

- A. glycerol
- B. amino acids
- C. simple sugars
- D. fatty acids

Answer: C

