

India's Number 1 Education App

CHEMISTRY

BOOKS - OMEGA PUBLICATION

ORGANIC CHEMISTRY SOME BASIC PRINCIPLES AND TECHNIQUES

Questions

1. Give the difference between alicyclic and aromatic compounds.



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2. What are hybridisation states of each carbon atom in the following

compound: C6H6



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3. Identify the o and it bonds in the following molecules. $i)C_6H_6 \qquad ii)C_6H12 \qquad iii)CH_2Cl_2 \qquad iv)CH_2=C=CH_2 \qquad v)CH_3.$

$$(v) C_6 H_6 \qquad (v) C_6 H_1 2 \qquad (v) C_{12} C_2 \qquad (v) C_{12} = C = C_{12} \qquad (v) C_{13}$$

$$(v) C_6 H_6 \qquad (v) C_{13} C_2 \qquad (v) C_{12} C_2 \qquad (v) C_{13} C_2 \qquad (v) C_$$

5. What is homologous series? Give its important characteristics.





4. Define functional group?







6. Write IUPAC names of the following organic compounds. $CH_3-CH=CH-CH_2-OH$



7. Write IUPAC names of the following organic compounds.

$$CH-egin{pmatrix} CH_3 & | & \ CH-egin{pmatrix} CH_2 & -CHO \ | & \ CH_3 \end{pmatrix}$$



- **8.** Define the term- Creepers and give one example?
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9. Define the term- climbers with one example?



10. Give the IUPAC name of the following organic compounds:

$$CH_3 - CH_3 - CH_3 - CH_3 - CH_3$$

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11. What are the main functions of a leaf?
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12. What are unisexual flowers? Give one example?
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13. What are bisexual flowers? Give one example?
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14. Name a plant which eat insects?
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15. What do you mean by complete and incomplete flower?



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16. Write the IUPAC name of the following organic compounds:

$$H_3C - C = C - C \equiv C - CH_3$$



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17. Name the pollen producing part of the flower? Explain with the help of a well labelled diagram?



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18. Which part of the flower is known as the female reproductive part? Explain with the help of well labelled diagram?



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19. Which part of the skull bones are movable?
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20. Why our elbow cannot move in backward direction?
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21. Write IUPAC name of the following:
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22. Give one example of ball and socket joint?
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23. Write IUPAC name of the following:

$$CH_3 - CH = CH - COOH$$



24. Write the IUPAC name of the following organic compounds.

$$CH_3-CH_2-igcup_{CHO}^{CH_3}-CH_2-CH_3$$

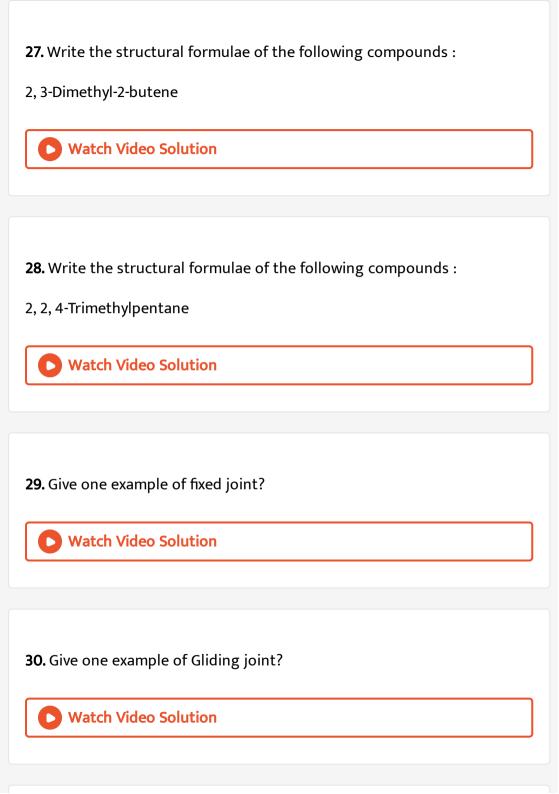


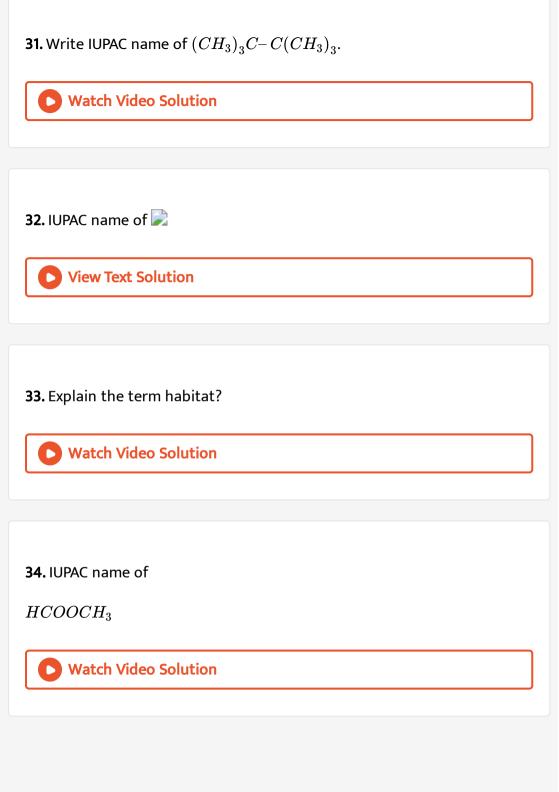
25. Give one example of pivotal joint?



26. Give one example of hinge joint?







35. IUPAC name of

$$(C_2H_5)_3C - C(CH_3)_3$$



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36. Name three things which are non-living but once were a part of living things?



37. Name the following compounds according to IUPAC-system.

$$CH_{3}-CH_{2}-CH_{2}-CH_{2}-CH_{-} egin{pmatrix} CH_{3} & & & & \\ & & & & & \\ CH_{2} & & CH_{3} & & \\ & & & & \\ CH_{2} & & & \\ & & & & \\ CH_{2} & & & \\ & & & & \\ CH_{3} & & & \end{pmatrix}$$



38. Write IUPAC name of the following compounds

$$C_6H_5 - CH = CH - CH_2CI$$



39. Name the following compounds according to IUPAC-system.

$$HC \equiv C - CH = CH = CH_2$$



40. Write the IUPAC name of the following organic compounds :

$$CH_3-CH-CH_2-CH_2-CH_2-CH_2-Cl$$



 CH_3

41. Write the IUPAC name of the following organic compounds:

$$CH_3-O-C\atop C\atop CH_3$$



42. Write the IUPAC name of the following organic compounds:

$$CH_3- egin{array}{cc} C \ H-COOH \end{array}$$

 CH_3



43. Write the IUPAC name of the following organic compounds:

$$CH_3 - CH_3 - CH_3 - CH_3 - CH_3$$



44. Give the IUPAC name of the following organic compounds:

$$CH_3-CH-CH=C-egin{pmatrix} ec{ec{ec{ec{ec{ec{ec{ec{S}}}}}}}_{NH_2} - CH-CH=C-egin{pmatrix} ec{ec{ec{ec{ec{ec{ec{S}}}}}}_{2H_5} - H$$



45. Give the IUPAC name of the following organic compounds:

$$CH_2 - CH_2 - CH_2 - CH_2 \ \mid \ \mid \ \mid \ OH \ OH$$



46. Give the IUPAC name of the following organic compounds:



47. Give the IUPAC name of the following organic compounds :

$$CH_3 - CH_2 - COOH$$



48. Give the IUPAC name of the following organic compounds :

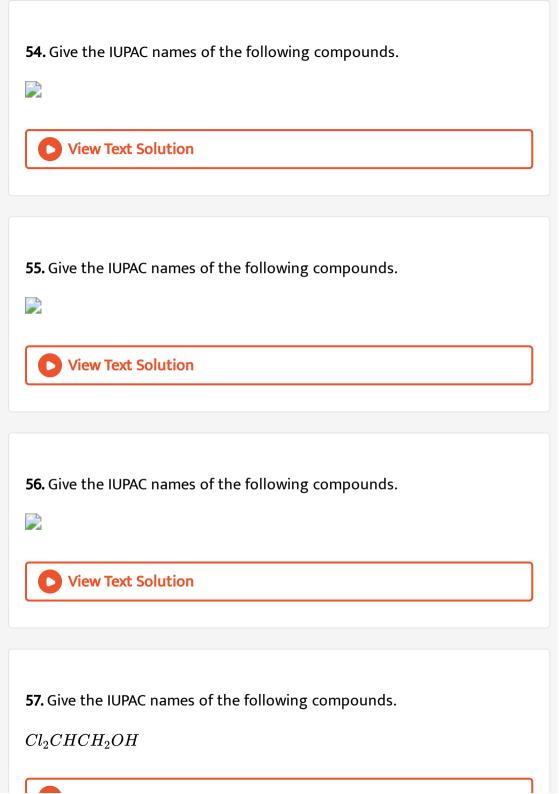
$$CH_3 - CH_2 - COOH$$



49. Give the IUPAC name of the following organic compounds :



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51. Assign IUPAC name of the following compounds:
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52. Give the IUPAC names of the following compounds.
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53. Give the IUPAC names of the following compounds.
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58. Give the IUPAC names of the following compounds.

$$CH_3 - \overset{CH_3}{C} = CH - CHO$$



59. Explain the followings with an example:

Position isomerism.



60. Explain the followings with an example :

Functional isomerism



61. Explain the followings with an example :

Chain isomerism



62. Explain the followings with an example :

Which structural isomerism is exhibited by the following pairs ?

i)
$$CH_3-\overset{O}{C}-OH$$
 ii) $H-\overset{O}{C}-O-CH_3$

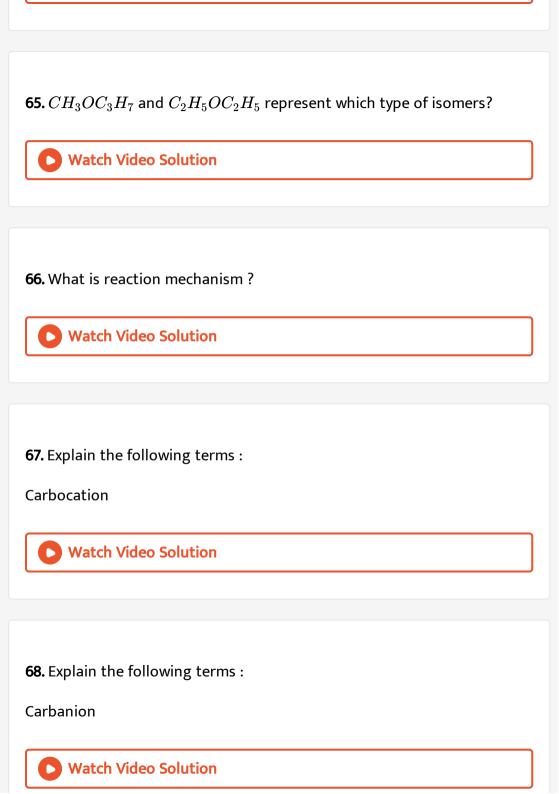


63. Write a note on geometrical isomerism.



64. What is metamerism? Give an example.

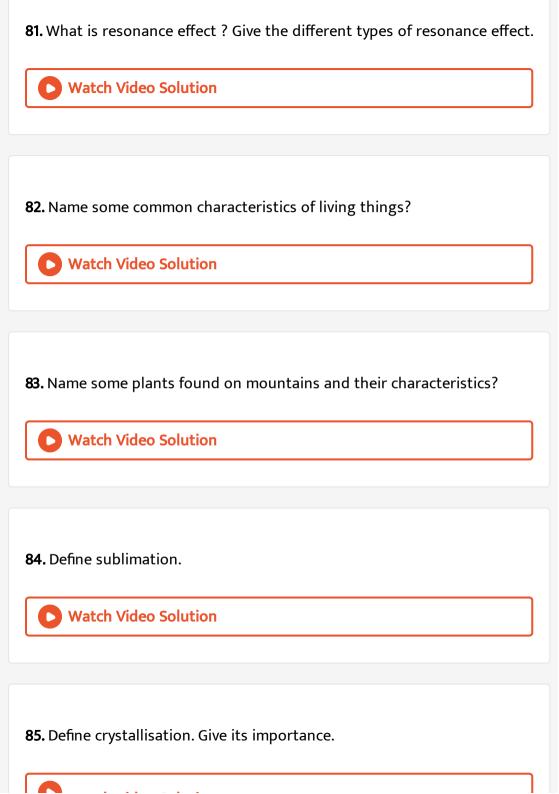




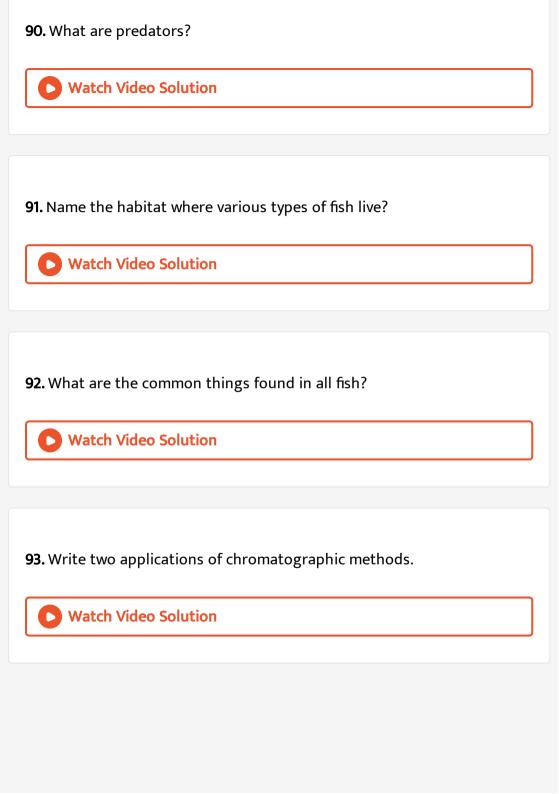
69. Explain the following terms: Free radical. **Watch Video Solution** 70. Explain the following term: Carbenes **Watch Video Solution** 71. Give the different types of carbocations and their order of stability. **Watch Video Solution** 72. Why tirtary carbocations are more stable than, secondary and primary carbocations? Explain.

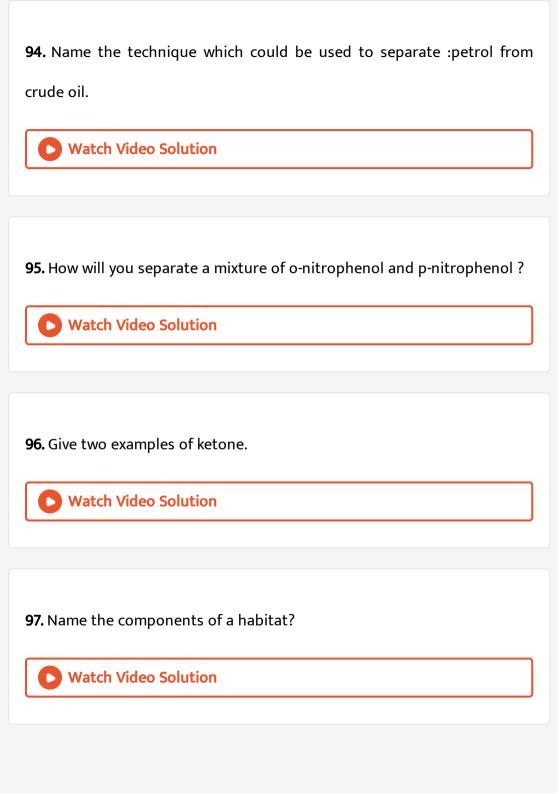
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73. Give the structure of carbocation.
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74. Give the different types of carbanions, also their order of stability.
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75. Why primary carbanions are more stable than the secondary and
tertiary carbanions ?
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76. Explain the stability order of free radicals.
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77. Differentiate between electrophiles and nucleophiles.
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78. Which of the two : $O_2NCH_2CH_2O^-$ or $CH_3CH_2O^-$ is expected
to be more stable and why?
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79. What is resonance ? Explain with the help of one example.
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80. What do you mean by resonance energy?
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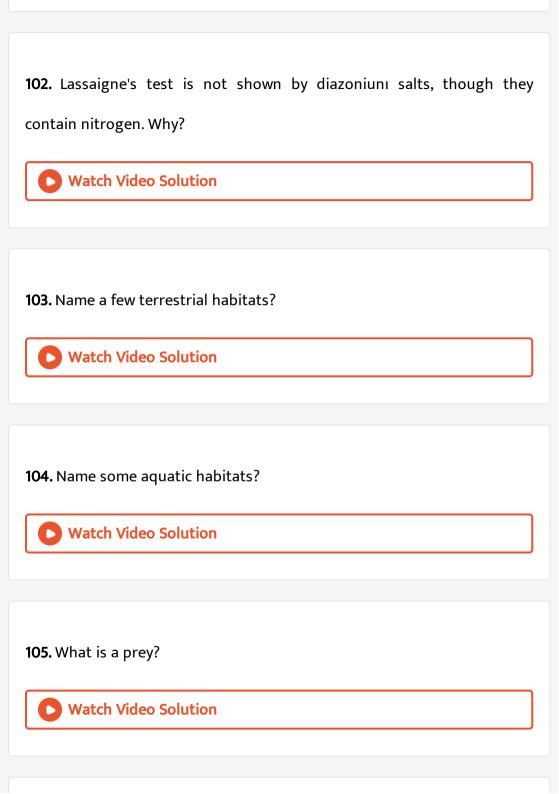


86. Discuss various steps involved in crystallisation.
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87. Name few habitats?
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88. Name two living organisms that live in deserts?
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89. Name few plants that are found in ponds?
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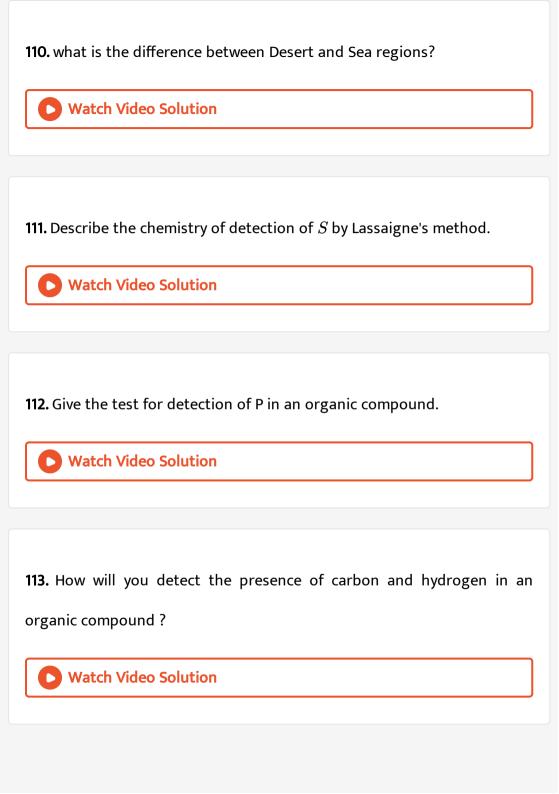




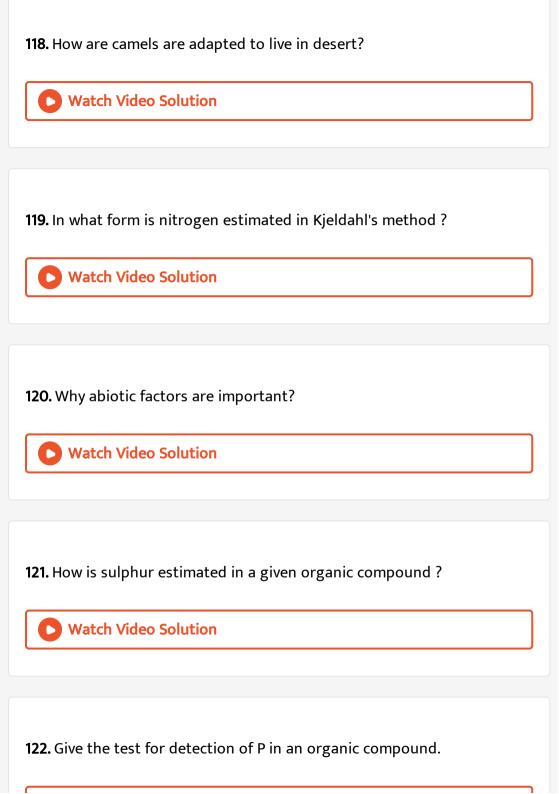
98. What do you mean by the germination of seed? Watch Video Solution 99. How will you detect the presence of carbon and hydrogen in an organic compound? **Watch Video Solution 100.** Will CCl_4 give white precipitate of AqCl on heating with silver nitrate? Give reason for your answer... **Watch Video Solution** 101. Why is an organic compound fused with sodium for testing halogens, nitrogen, sulphur and phosphorus? **Watch Video Solution**



106. Give the chemistry of Lassaigne's test for detection of N in an organic compound. **Watch Video Solution** 107. What do you mean by breathing? **Watch Video Solution** 108. Name two animals which have no gills? **Watch Video Solution** 109. Define the term-Respiration? **Watch Video Solution**



114. How oxygen is estimated in a given organic compound? Watch Video Solution 115. Discuss Dumas method for estimation of nitrogen in an organic compound. **Watch Video Solution** 116. Explain the features of a fish which help them to live them in water? **Watch Video Solution** 117. Discuss Dumas method for estimation of nitrogen in an organic compound. **Watch Video Solution**



Multiple Choice Questions

1. The I.U.P.A.C name of

$$CH_3-CH=CH-\overset{O}{\overset{|}{C}}-O-CH_2-CH_3$$

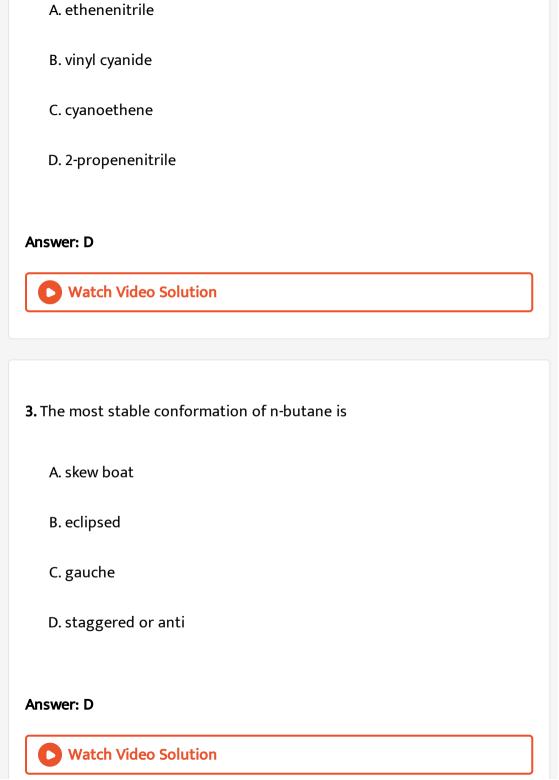
- A. ethylbut 1- enoate
- B. ethylbut -2-enoate
- C. ethyl prop-2-enoate
- D. none of above

Answer: B



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2. The I.U.P.A.C. name of $CH_2=CH_2-CN$ is



4. Which process is suitable for the purification of aniline?
A. Simple distillation
B. Steam distillation
C. Fractional distillation
D. Fractional crystallisation,
Answer: B
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5. Anthracene is purified by
A. filtration
B. crystallisation
C. distillation

D. sublimation
Answer: D
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6. Which of the following compounds does not give Lassaigne's test for nitrogen ?

A. Urea

B. Hydrazine

C. Phenyldrazine

D. Azobenzene

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Answer: B

7. In Kjeldahl method, the presence of nitrogen is estimated as
A. N_2
B. NH_3
$C.\ NO_2$
D. N_2O_3
Answer: B
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8. Molecular mass of volatile substance may be determined by
A. Kjeldahl's method
A. Kjeldahl's method B. Duma's method
B. Duma's method

Answer: C



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9. Convey the following information in the form of a balanced chemical equation: On adding an aqueous solution of sodium hydroxide to an aqueous solution of copper sulphate, copper hydroxide is precipitated and sodium sulphate remains in the solution.



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10. The most stable ion is

A.
$$CH_3CH_2^{\,+}$$

B.
$$(CH_3)_2 CH^+$$

C.
$$(CH_3)_3C^+$$

D.
$$C_6C_5CH_2^+$$

Answer: C



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- 11. The functional group for aldehydes is
 - A. OH
 - B.-CHO
 - C.-COOH
 - D. > CO

Answer: C



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- 12. The functional group O belongs to the family of
 - A. ketones

B. aldehydes
C. ethers
D. alcohols
Answer: B
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13. The word root 'oct' is for
A. C_6
B. C_7
$C.C_8$
D. C_9
Answer: C
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14. Steam distillation is applied for the separation of those compounds which are

A. steam volatile and soluble in water..

B. steam volatile and decompose in water

C. steam volatile and insoluble in water

D. capable of chemical reaction in water

Answer: D



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15. The process of differential extraction is based upon

A. different solubilities

B. different molecular masses

C. different boiling points

D. different chemical properties

Answer: A



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16. A substance which decomposes below its boiling point can be best purified by

- A. steam distillation
- B. simple distillation
- C. fractional distillation
- D. distillation under reduced pressure

Answer: D



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17. In Duma's method, the gas which is collected in nitrometer is

A. N_2
B. NO
C. NH_3
D. H_2
Answer: A
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10 The first energy is some and a new and is
18. The first organic compound prepared is
A. ammonia
B. methane
C. urea
D. ethane
Answer: C
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19. General formula C_nH_{2n+1} , OH is of

A. alcohol

B. aldehyde

C. ketone

D. carboxylic acid.

Answer: A



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20. Which of the following compound will give blood red colour while doing the Lassaigne's test for N?

A.
$$\left(NH_{2}
ight)_{2}C=O$$

B. $H_2N(C_6H_4)SO_4H$

 $\mathsf{C.}\ C_6H_5SO_3H$

D. $CHCl_3$

Answer: B



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- 21. The violet colour in the Lassaigne's test of sulphur is due to
 - A. $FeCl_3$
 - B. $Na_2igl[Fe(CN)_6igr]$
 - $\mathsf{C.}\,Na_{4}ig[Fe(CN)_{5}NOSig]$
 - D. $Fe_4igl[Fe(CN)_6igr]_3$

Answer: C



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22. In ethene $(CH_2=CH_2)$ each C-atom undergoes

- A. sp- hybridization
- B. sp^2 hybridization
- C. sp^3 -hybridization
- D. sp^2 and sp^3 -hybridization

Answer: B



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23. Which of the following compound contains aldehydic functional group?

- A. CH_3COOH
- B. CH_3CHO
- C. CH_3CONH_2
- D. None of these,

Answer: B

24. The IUPAC name of the compound

$$CH_2 = CH - CH - CH_3$$
 is $_{CH_3}^{ert}$

A. 1, 1 dimethyl-2 propene

B. 3-methyl but-1-ene

C. 2-Vinyl propene

D. 1-Isopropyl ethene

Answer: B

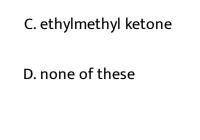


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25. The common name of 2-butanone is

A. acetone

B. butyraldehyde



Answer: D



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26. The IUPAC name of the compound. $CH_3-CH=CH-C\equiv CH$

A. Pent-2-en-4-yne

B. Pent-1-en-4-yne

C. Pent-3-en-1-yne

D. Pent-2-en-5-yne

Answer: A



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