



## **CHEMISTRY**

# **BOOKS - OMEGA PUBLICATION**

# SAMPLE QUESTION PAPER -IV (PUNJAB)



**1.** Mass of 1 molecule .of  $NH_3$  is\_\_\_\_\_

A. 
$$5.6 imes10^{-23}$$
 g

 $\mathsf{B}.\,1.4\times10^{-23}\,\mathsf{g}$ 

C. 
$$5.6 imes10^{-23}$$
 g

D.  $7 imes 10^{23}$  g

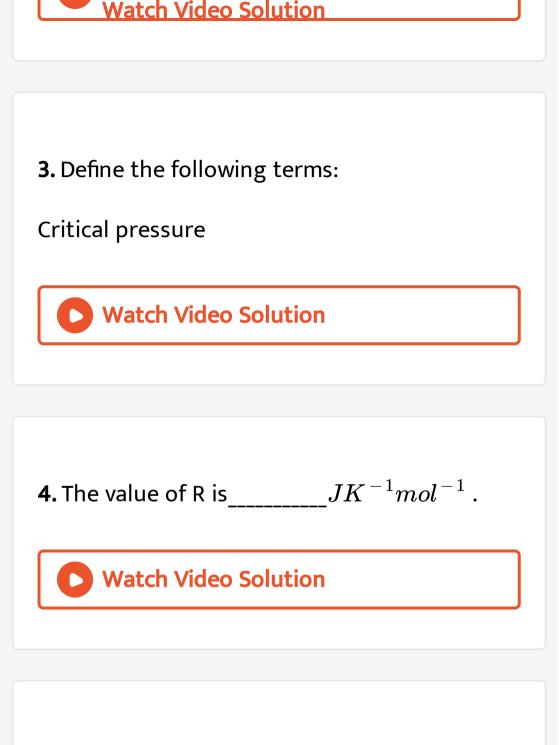
#### **Answer:**

Watch Video Solution

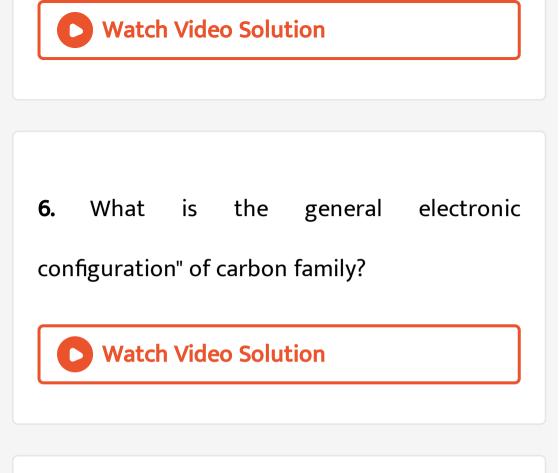
2. What is periodicity? What is the cause of

periodicity?





**5.** What is the structure of  $H_2 0_2$  ?



7. Draw the structure of in-organic benzene.

8. What is the principle of chromatographic separation ? Name the different types of chromatography commonly used.



#### 9. What is the diffrence between covalent.

raclius and vanderwall's radius?



10. Explain why carbon differs from rest of the

family members.

Watch Video Solution

11. Define solubility product. How does it differ

from ionic product ?

12. What is the difference between e.m.f. and

potential diffrence?

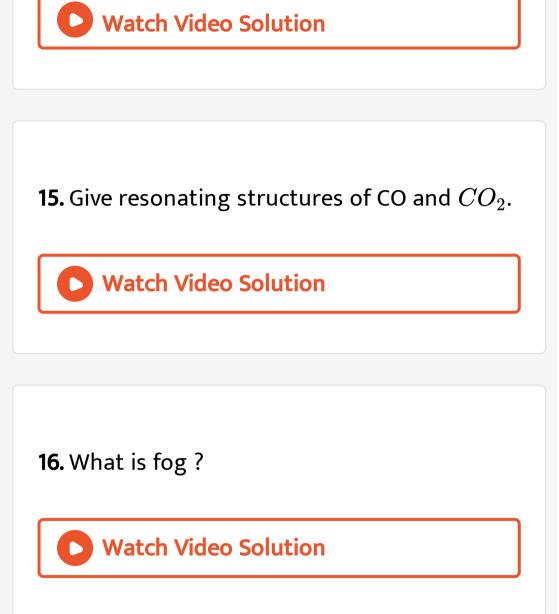
Watch Video Solution

**13.** Give general electronic configuration of p

and d-block elements.



14. Define green chemistry.



17. Give-one reaction in each case to show that

 $H_2O_2$  is

an reducing agent



#### 18. Give-one reaction in each case to show that

 $H_2O_2$  is

an reducing agent

19. Give-one reaction in each case to show that

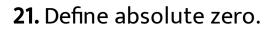
 $H_2O_2$  is

an reducing agent

Watch Video Solution

20. Calculate volume occupied by two moles of gas at  $27^{\circ}C$  and under a pressure of 1

atmosphere.





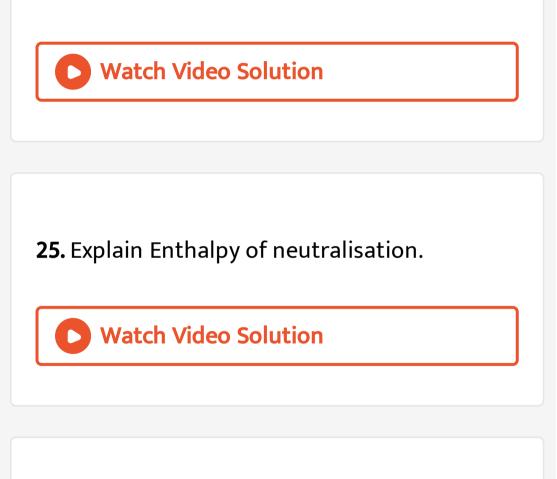
#### 22. Distinguish between inductive and

electromeric effect.

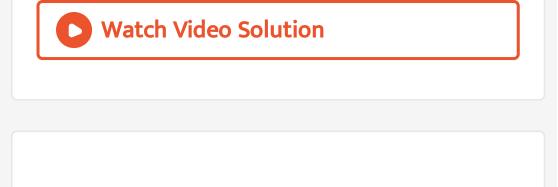
Watch Video Solution

23. Explain the following

Enthalpy of fusion



**26.** How ionic product of water change with temperature ?



#### 27. Classify the following into Lewis acids or

Lewis bases giving reasons

 $H_2O, BF_3, NH_3, SiF_4, Ag^+, CI^-, CO_2.$ 

### Watch Video Solution

#### 28. Give the IUPAC name of the following

 $(C_2H_5)_3C - OH$ 



#### 29. Give the IUPAC name of the following

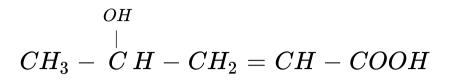
 $CH_2 = CH - C = CH$ 

Watch Video Solution

#### 30. Give the IUPAC name of the following

# $CH_3 - CH - CH_2 - COOH \ ert_{Cl}$

31. Give the IUPAC name of the following



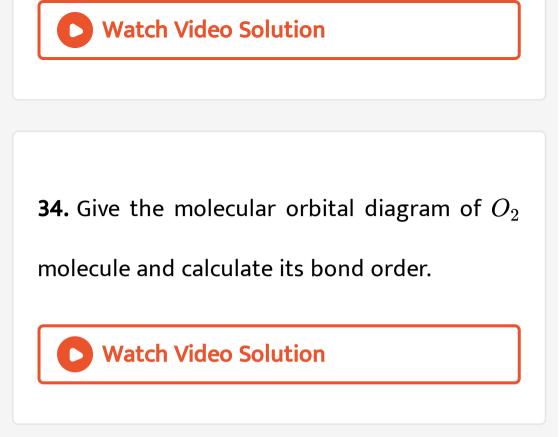
#### Watch Video Solution

#### 32. Group 2 elements are denser and harder

than group 1 elements. Why?

Watch Video Solution

33. How is caustic soda manufactured ?



35. What is the difference between atomic and

molecular orbital

**36.** What is hybridisation on the basis of

hybridisation explain the geometry of  $PF_5$  .



**37.** What is the difference between orbit and orbital.



38. What.is'H - spectrum? How it can be

explained by Bohr's Theory.

Watch Video Solution



#### configuration of copper and chromium.



**40.** What is the wavelength of light emitted when the electron in a hydrogen atom undergoes trnsition from an energy level with n = 4 to an energy level with n = 2?

Watch Video Solution

41. Give the mechanism of sulpho'nation and

nitration of benzene.



42. Why alkenes are more reactive than

alkanes?



**43.** Write a short notes on following:

Markownikov's rule

**44.** Write a short notes on following:

Wurtz reaction

Watch Video Solution

**45.** Write a short notes on following:

Ozonolysis of alkenes

