

## **CHEMISTRY**

## **BOOKS - VGS PUBLICATION-BRILLIANT**

## **MODEL PAPER 4**

Section A

**1.** Name two adverse effects caused by acid rains.



2. Define the terms 'Receptor' and 'Sink'.



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3. What is surface tension? Write its units.



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**4.** 4 gram NaOH dissolved in 250 ml of the solution. Calculate Molarity.

**5.** Write the conjugate acid and conjugate base of each of the following :

- a)  $HSO_4^-$
- b)  $H_2O$



**6.** Give an account of the biological importance of  $Na^{\,+}$  and  $K^{\,+}$  ions.



7. What is Plaster of Paris? Write its uses.



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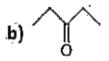
8. Poisonous gas



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9. How is water gas prepared?

**10.** Write IUPAC names of the following compounds.





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**Section B** 

**1.** Explain different types of hydrogen bonds with examples.



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**2.** What is Hybridization ? Explain the structure of  $CH_4$  on the basis of Hybridization.



**3.**  $360cm^3$  of  $CH_4$  gas diffused through a porous membrane in 15 minutes. Under similar conditions,  $120cm^3$  of another gas diffused in 10 minutes. Find the molar mass of the gas.



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**4.** Balance the following Relox reaction by ionelectron method an acidie medium.

$$egin{array}{cccc} Cr_2 & O_7^{2\,-} + SO_2 
ightarrow Cr^{3\,+} + SO_4^{2\,-} \ (aq) & (g) & (aq) & (aq) \end{array}$$



**5.** State and explain the Hess's law of constant heat summation.



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6. Discuss the application of LE Chatellier's principle for the industrial synthesis of Ammonia and sulphur trioxide.



7. Explain the terms hard water and soft water.

Write a note on the

(ii) Calgon method for the removal of hardness of water.



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**8.** Explain Borax bead test with suitable example.



## **Section C**

**1.** What are the postulates of Bohr's model of hydrogen atom? Discuss the importance of this model to explain various series of line spectra in hydrogen atom.



**2.** What is a periodic property? How the following properties vary in a group and in a

period? Explain

(b) EN.



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**3.** Describe two methods of preparation of Benzene. Explain Halogenation and Alkylation reactions of Benzene.

