

## **CHEMISTRY**

## BOOKS - VIKRAM PUBLICATION ( ANDHRA PUBLICATION)

**MARCH - 2016 (ANDHRA PRADESH)** 

Section A

1. What are DO, COD and BOD?



**2.** Name two adverse effects caused by acid rains.



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**3.** Give the hybridization of carbon in diamond and fullerene.



**4.** State and explain the Hess's law of constant heat summation.



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5. State Graham's law of diffusion.



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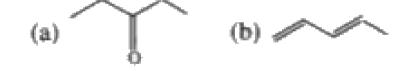
**6.** The empirical formula of a compound is  $CH_2O$ . Its molecular weight is 90. Calculate

the molecular formula of the compound. **Watch Video Solution** 7. Lechatelier's principle is **Watch Video Solution** 8. Define Entropy. **Watch Video Solution** 

**9.**  $SiF_6^{2-}$  is known while  $SiCl_6^{-2}$  is not explain.



**10.** Write the systematic names of the following





## **Section B**

**1.** Write the postulates of kinetic molecular theory of gases .



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2. Explain the structure of diborane.



**3.** What is Hard water? How can you remove hardness of water by Calgon method?



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**4.** Explain different types of hydrogen bonds with examples.



**5.** What is conjugate Acid -Base pair ? Write the conjugate acid and conjugate base of  $NH_3 \ {
m and} \ H_2O$ 



**6.** Compare diploment of  $NH_3$  molecule with that of  $NF_3$  molecule .



**7.** Discuss the various reactions that occur in the Solvay process.



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**8.** Balance the following equation in acid medium by Ion-electron method :

$$Fe^{+2}_{
m (aq)} + Cr_2O^{2-}_{7{
m (aq)}} 
ightarrow Fe^{3+}_{
m (aq}j + Cr^{+3}_{
m (\it{aq})}$$



**1.** What are quantum numbers? Explain the significance of various types of quantum numbers.



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**2.** Define  $IE_1$  and  $IE_2$ . Why is  $IE_2 > IE_1$  for a given atom? Discuss the factors than effect IE of an element.



**3.** Complete the following reactions and name the products A,B and C.

