

## **CHEMISTRY**

## BOOKS - VIKRAM PUBLICATION ( ANDHRA PUBLICATION)

**MARCH - 2016 (TELANGANA)** 

Section A

1. Define TLV.



2. Why is carbon monoxide poisonous?



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3. Write the preparation of plaster of Paris.



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4. State and explain law of mass action.



5. What is disproportionation reaction?



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**6.** Why is  $KO_2$  paramagnetic?



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**7.** Calculate the ratio of kinetic energies of 3g of hydrogen and 4g of oxygen at an given

temperature.

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**8.** How is Ozone layer depleted in the atmophere and what are the harmful effects caused by Ozone layer depletion ?



**9.** Write any two uses of zeolites .



**10.** Write IUPAC names of the following compounds:

(a) 
$$CH_3-CH_2-CH_2-CH=CH_2$$
 (b)  $ightharpoons$ 



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Section B

1. Derive ideal gas equation.



**2.** Explain hybridisation of phosphorous in the formation of  $PCl_5$ 



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**3.** Balance the following Relox reaction by ionelectron method an acidie medium.

$$egin{array}{cccc} Cr_2 & O_7^{2-} + SO_2 
ightarrow Cr^{3+} + SO_4^{2-} \ (aq) & (g) & (aq) & (aq) \end{array}$$



**4.** Write two oxidizing and two reducing properties of  $H_2O_2$ .



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**5.** Explain borax bead test with a suitable example



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**6.** State the third law of thermodynamics.

**7.** Explain different types of hydrogen bonds with examples.



- **8.** Explain the term Hydrolysis of salts with examples. Discuss the pH of the following types of salt solutions.
- (i) Salts of weak acid and strong base.
- (ii) Salts of strong acid and weak base.



## Section C

**1.** What are the postulates of Bohr's model of hydrogen atom?



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**2.** Define  $IE_1$  and  $IE_2$ . Why is  $IE_2>IE_1$  for a given atom? Discuss the factors than effect IE

of an element.



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**3.** How does acetylene react with the following reagents? Give the corresponding equations and name the product formed in the reactions?

Water

