

## **CHEMISTRY**

# **BOOKS - UNITED BOOK HOUSE**

# **BETHUNE COLLEGIATE SCHOOL QUESTION PAPER**

# Exercise

1. Number of p electrons in bromine atom are:

A. 12

B. 15

C. 7

D. 17

#### Answer:



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2	Which	of the	folloAving	having	mavimum	alactron	affinity?
Z.	VVIIICII	or the	TOHOAVING	Havillg	IIIaxiiiiuiii	electron	allillitys

A.  $F^{\,-}$ 

B.O

 $\mathsf{C}.\,O^-$ 

D.  $Na^+$ 

#### **Answer:**



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3. For which compound both covalent and ionic bonds are present?

A.  $CCl_4$ 

 $\operatorname{B.}\mathit{CaCI}_2$ 

 $\mathsf{C}.\,NH_4Cl$ 

D.	$H_2C$

### **Answer:**



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- **4.** When the hydride of basic metal, reacts with water, it gives .
  - A. acid solution
  - B. basic solution
  - C. neutral solution
  - D. hybride ion

## Answer:



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**5.** The strong base is:

A. LiOH
B. NaOH
C. KOH
D. CsOH
<b>A</b>
Answer:
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6. How many- helium atoms are present in 52u of Helium?  Watch Video Solution
<b>7.</b> Arrange the following in order of increasing dipole,moment $NH_3,NF_3,CBr_4.$
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9. Write down the formula of hydrolith.
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equation in ion electronic

 $K_2Cr_2O_7 + FeSO_4 + H_2SO_4 
ightarrow K_2SO_4 + Cr_2(SO_4)_3 + Fe_2(SO_4)_3 + H_2SO_4 + Cr_2(SO_4)_3 + H_2SO_4 + H_2SO_5 + H_2SO_5$ 

method,

**8.** Determine the oxidation of Fe in  $K_4\big[Fe(CN)_6\big]$ .



Balance

10.

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**11.** Draw the structure of  $H_2O_2$  molecule.

the





**13.** Since oxygen and sulphur are belong in same group, but  $H_2O$  is liquid but  $H_2S$  is gas, describe.



14. Between Berilium and Boron which have higher ionisation enthalpy?



**15.** In 0.09 gm of `H\_2O, how many hydrogen and oxygen atoms are present?



**16.** State Hund's multiplicity Rule and explain it. Give electronic configuration of sulphide ion.



17. Eelctroncgativity of fluorine is greater than chlorine, but electron affinity is higher in chlorine that fluorine— explain. Why 2nd ionisation potential is always greater than 1st ionisation potential?



**18.** Bond angel of  $H_2O$  is greater than  $H_2S$ — Explain.Explain, ammonia is more basic than phosphin.



**19.** How will you prepare watergas from coke, give equation. Write two names of the compound which cause parmenent hardness of water.



**20.** From a mixture of acetylene, ethelene, methene gas, how will' you separate these three gas mixture?



21. Why methene does not prepared from 'Wurtz Reaction'?Convert :

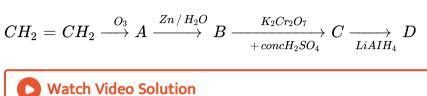
$$CH_3-CH_3 o CH_2=CH_2.$$



22. Why methene does not prepared from 'Wurtz Reaction'?Convert :

$$CH_3-CH_2-OH o CH_2=CH_2.$$

23. Identify A, B, C, D.
$$\frac{Z_n}{H_0}Q = \frac{K_0 C_0}{R_0}Q$$



# **24.** Identify E, F. $CH_3-CH_2-CH=CH_2 \stackrel{HBr}{\longrightarrow} E \stackrel{KOH/alcohol}{\longrightarrow} F$



**25.** Why nitrobenzene is used as a solvent in Friedel Craft Reaction?



**26.** Determine the volume strength of the 1.5  $(N)H_2O_2$  solution.



**27.** Hardness, which is related' with Mg-ions, does not completely removed by only boiling— explain.



28. Why sugar is soluble in water?

