

CHEMISTRY

BOOKS - UNITED BOOK HOUSE

HINDU SCHOOL QUESTION PAPER

Exercise

1. Calculate the equivalent wt. of $Kmno_4$ in

Acid medium. [K = 39 Mn = 55]



2. The percentages of oxygen in two oxides of a metal are 22.2 and 30 respectively. The formula of the second oxide is M_2O_3 . Deduce the formula of first.



3. Calculate the ratio of the molecules of two gaseous substances having same volume and same mass each.

4. An electron is present in 4 f orbital. Write its possible quantum Nos. (n, 1, m. s).



5. The wave number of a specturm in Lyman series in hydrogen atom is $82260cm^{-1}$. Show that this specturm is created by shifting the electron from second to first orbit. $[R=109670cm^{-1}]$

6. The Lanthanides contraction is due to



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7. The actual order of Ionic radius is

A.
$$Li^+ > Na^+ > Mg^{+2} < Be^{+2}$$

B.
$$Mg^{\,+\,2}>Be^{\,+\,2}>Li^{\,+}>Na^{\,+}$$

C.
$$Li^{+1} > Be^{2+} > Na^{+} > Mg^{+2}$$

D.
$$Na^+>Li^+>Mg^{+2}>Be^{+2}$$

Answer:



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8. Which of the followings expresses same No. of electrons?

A.
$$C_2^{2-}$$
 , O_2^2 , CO , NO

B.
$$NO^+, C_2^{2-}, CN^-, N_2$$

C.
$$CN^-, N_2, O_2^{2-}, CO_2^{2-}$$

D. $N_2, O_2^{2-}, NO^+, CO^-$

Answer:



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9. " $\mathbb{C}I_4$ does not have any hydrolysis lent $SiCl_4$ suffers hydrolysis— Why?



10. Write Hund's rule regarding atomic sutructure. Cite one example.



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11. What is dipole moment?



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12. "Nitrogen can create NCI_3 but phosphorous can create both PCI_3 and PCl_5

" — why?



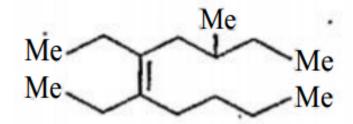
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13. Write only the five postulates of the VSEPR theory.



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14. Write the names of the following two organic compound in I.U.P.A.C. system





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15. Write the names of the following two organic compound in I.U.P.A.C. system $CH_3CH_2CH_2CH_2CH_2COOCH_3$



16. Give figure of molecular orbital energy diagram of N_2 molecule.

