

CHEMISTRY

BOOKS - UNITED BOOK HOUSE

MODEL QUESTION PAPER 17

Exercise

1. The possible m values for '2p' orbital are

$$A. +2, 0, -2$$

$$B. -2, 0, +1$$

$$C. +1, 0, -1$$

$$D. +1, 0, -3$$

Answer:



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2. Which one has highest bond angle—

A. PCI_3

 $\mathsf{B}.\,BH_3$

 $\mathsf{C}.\,CH_4$

D. CO_2

Answer:



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3. In which Compound the Central atom is sp_2 hybridised—

A. N_2O

B. CO

 $\mathsf{C}.\,CO_2$

D. SO_2 .

Answer:



- **4.** Spreadibility increases for the liquid when—
 - A. Viscosity increases
 - B. Surface tension decreases
 - C. Surface tension increases

D. Viscosity decreases

Answer:



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5. For Adiabetic Expansion of an ideal gas—

A. Temperature decreases

B. w = 0

C. q = 0

D. H = 0.

Answer:



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- 6. Internal energy does not include
 - A. Atomic energy
 - B. Rotational energy
 - C. Combustion energy
 - D. Vibrational energy

Answer:

7. If the $t\frac{1}{2}$ for any reaction be inversely proportion at to the initial Concentration of the reactant then the order of reaction is—

A. Zero

B. one

C. Two

D. or more.

Answer:

8. Which one does not react with NaOH

A. Na_2O

B. SiO_2

C. BeO

D. P_2O_5 .

Answer:



9. Gypsum is

- A. $MgCl_2$
- B. $CaSO_4$
- C. MgO
- D. $MgCO_3$

Answer:



10. The General name of

$$CH_3-CH_2-C(CH_3)_2-CH_3$$
 is

- A. Neohexane
- B. Isopentane
- C. 2, 2 dimethyl butane
- D. Isopentane.

Answer:



11. Whic one show ge	ometric is	omerism
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- A. 2 -Butene
- B. 2- Butyne
- C. 2- Butanol
- D. I Butene.

Answer:



12.
$$CnH_{2n}+H_2\stackrel{Ni/300^{\circ}c}{\longrightarrow}CnH_{2n}+2$$
 this is

known as

- A. Sabettier & sandaren
- B. Friedel Craft
- C. Kolbe
- D. Wurtz-reaction.

Answer:



13. Which one is not meta - orienting

A.
$$-SO_3H$$

$$B.-NO_2$$

$$\mathsf{C.}-NH_2$$

$$D.-CN$$

Answer:



14. Which gas evolyes in maximum in Volcanic erruption

- A. CO_2
- B. SO_2
- $\mathsf{C}.\,H_2S$
- D. NO_2

Answer:



15. 1 Amu =how much gm?



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16. The reagent which consumed whole in a reaction is called?



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17. Name a 'd' block element which is not transitional.



18. In which block inner transition metal is.included?



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19. What type of property entorpy is?



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20. Name one anti aromatic Compound.



21. Show that $\operatorname{E=}A/v$ where E, A, V used are in usual meaning?



22. Establish the relation between .vapour density and moleculer weight of gas.



23. Prove that in 'n'th shell $2n^2$ number of electron, can be accommodated.



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24. Give the difference between orbit and orbital.



25. Write with Condition & balanced equation the preparation of CO from oxalic acid.



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26. Illustrate with example metamerism.



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27. Illusttrate with example tautomerism.



28. Write full form of TLV. How Taj Mahal is effected by atmosphere?



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29. If the uncertainty of position of a particle of mass 1×10^{-4} gm be 1.56×16^{-3} cm then find the uncertainty of its velocity?



30. Write the electronic configurration of $Cr_{3\,+}$ ion and find no of unpaired electron in it.



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31. Define Ionisation potential. How it changes form left,to right in 2nd period?



32. Define electorn affinity. State its changes down a group.



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33. Illustrate with example Sizweek maximum valency theory.



34. Give an example of Compression of octate.

Which one is more soluble in alcohol Kl or KCI.



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35. In which Condition real gas behalves ideally? Why real gas diviates from ideal behaviour.



36. What do you mean by surface tension? State dimension of surface tension. Why the viscosity of a liquid decreases with increase of temperature.



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37. Write twodifferences between reversible and irreversible process. What is open system?



38. Calculate, $\Delta U, W$ and Q when an ideal gas is compressed at 5 atm pressure from 25 litre to 5 litre at a constant temp of $27^{\circ}\,C$.



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39. What is oxidaiton number? Explain that oxidation and reduction Occurs simultaneously.



40. Why Li can't be stored in Kerosene like Na? Give an example of super oxide.



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41. Calculate the hardness of water in ppm when $2.56gmMgCO_3$ dissolved in 1 litre water.



42. Write the structural formula of the following organic Compound:-2, 3- dimethyl -1, 3- pentadiene.



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43. Write the structural formula of the following organic Compound:-Butane -1, 2, 3 - tri Carbaldehyde.



44. Write the structural formula of the following organic Compound:-3- Chloro- 3, 4-dimethylpentanal.



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45. Explain why atomic radius of Ga is less than that of Al? $AlCI_3$ forms .dimer but BCl_3 does not - Explain? What is corrundum?



46. Give an example of poly nuclear aromatice compound. Write one merit and onedimesti of wurtz reaction. Shall methane can be synthesie using kolbe electrolyris process?



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47. How would you convert:Ethene Acetylene.



48. How would you convert:Acetylene ightharpoonup Methane.



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49. What reagent is used for synthesis of alkane by Corey-house synthen's.

