



PHYSICS

BOOKS - R G PUBLICATION

DUAL NATURE OF RADIATION AND MATTER

Exercise

1. Find the dimension of Planck's constant.



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2. Find the dimension of Planck's constant.



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3. What is the dimension of work function?



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4. Give the dimension of Planck's Constant



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5. Write down Einstein's photo electric equation and explain each of its terms.



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6. What is photoelectric effect? Why is photo electric current proportional to the intensity of incident radiation?



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7. Write down Einstein's photo electric equation and explain each of its terms.



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8. Work function of Caesium is 2.14 eV.

Find its threshold frequency.

$$h = 6.63 \times 10^{-34} \text{ Js}, 1\text{eV} = 1.6 \times 10^{-19} \text{ J}$$



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9. Find the wavelength of an electron accelerated through a potential difference of 1 volt.



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10. What is radiation pressure.



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11. A monochromatic source of light operating 200W emits 4×10^{20} photons/sec. Find the wavelength of the light. Given $h = 6.63 \times 10^{-34} \text{ J} \cdot \text{s}$, $C = 3 \times 10^8 \text{ m/s}$.



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12. What is the de-Broglie wavelength of an electron in the Bohr's orbit of radius 0.51 \AA in hydrogen atom?



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13. Derive an expression for the radius of the first orbit of the electron of the hydrogen atom.



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14. Obtain the binding energies of the nuclei ${}_{26}^{56}\text{Fe}$ and ${}_{83}^{209}\text{Bi}$ in units of MeV from the given data.

$$m_H = 1.007824a\mu$$

$$m_n = 1.008665a\mu$$

$$m({}_{26}^{56}\text{Fe}) = 55.934939\text{amu}$$

$$m({}_{83}^{209}\text{Bi}) = 208.980388\text{amu}$$

1 amu = 931.5 MeV

Which nucleus has greater binding energy per nucleon?



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15. A photo electric surface has work function 2eV. What is the maximum kinetic energy of the emitted photoelectrons ejected by light by wavelength 3000\AA

Given

$$h = 6.6 \times 10^{-34} \text{ J} \cdot \text{S}, e = 1.6 \times 10^{-19} \text{ C}.$$

$$m_e = 9.1 \times 10^{-31} \text{ kg}, C = 3 \times 10^8 \text{ m/sec}$$



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16. There are mainly three ways to eject an electron from the surface of metals. What are those?



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17. If kinetic energy of a free electron is increased by two times then by how many times will the De-Broglie wavelength be changed?



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18. What is the de Broglie wavelength associated with an electron of mass $9.11 \times 10^{-31} \text{ kg}$ moving with a speed $5.4 \times 10^6 \text{ m/s}$. Given $h = 6.63 \times 10^{-34} \text{ Js}$.





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19. What is the de Broglie wavelength associated with

a ball of mass 150g travelling at 30m/s ?

Given $h = 6.63 \times 10^{-34} \text{Js}$.



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20. What determines the intensity of light in the photon picture of light?



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21. Fill in the blanks

X rays is discovered by _____ in _____.



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22. Fill in the blanks

Einstein was awarded the Nobel Prize in __ for his contribution to __ and ____



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23. Fill in the blanks

Photons are electrically neutral and not deflected by _____ and _____ fields.



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24. Fill in the blanks

In _____ De Broglie was awarded the Nobel Prize in Physics for his discovery of _____.



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25. Write down Einstein's photo electric equation and explain each of its terms.



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26. What is cut-off or stopping potential?



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27. What is de Broglie wave length? Give the de Broglie equation.



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28. What is the charge on cathode ray particle?



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29. Express Joule in eV.



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30. Which oil should be used in Milikan's method.



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31. An electron is accelerated through a potential difference 1000V. What is its kinetic energy in MeV.



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32. The ratio of specific charges of an electron to that of a hydrogen ion is ?



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33. Can we use water drop in milkan oil drop method, why?



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34. Write any 3 properties of cathode rays?



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35. An electron of charge ' e ' at rest is accelerated in a uniform electric field E . It covers a distance x . What is the kinetic energy gained by the electron?



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36. What is specific charge? State its S.I. unit.



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37. Define work function of a metal.



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38. What is threshold frequency?



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39. What is rest mass of photon?



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40. In a photo electric effect kinetic energy of photo electron depend upon which factor?



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41. What is dark current?



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42. Is there any difference between light wave and matter wave?



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43. Define work function of a metal.



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44. Write a short note on Hertz's observation photo electric effect.



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45. Draw the variation curve of photo current with collector plate potential for different frequencies but same intensity of incident radiation.



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46. What is threshold frequency?



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47. Write a short note on particle nature of light.



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48. Calculate the de Broglie wave length of a ball of mass 0.16kg moving with a speed of $18\frac{\text{m}}{\text{s}}$.



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49. Write the Huygenberg's uncertainty principle.



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50. Find out the maximum frequency and minimum wavelength of X ray produced by 30 KV electrons.



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51. What is a photon? Write its properties.



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52. State laws of photoelectric emission.

Establish Einesteins photoelectric relation.



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53. How much charges is carried by 1 kg of photo electron.



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54. State laws of photoelectric emission.

Establish Einsteins photoelectric relation.



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55. The working function for a certain metal is

4.2ev. Will these metal give photoelectric

emision for incident radiation of wave length

330 nm.



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56. What is the important proof of de-Broglie wave.



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57. Prove that $\lambda = \frac{h}{\sqrt{2mE}}$.



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58. Explain one physical process for release of electron from surface of a metal.



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59. A cathod ray is operated 2500 V. What is the speed of electron?



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60. How is the mass of an electron determined?



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61. An electron of energy 150 eV describes a circle in a magnetic field of 1 T . Calculate the radius of the circle.



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62. Draw the variation curve of photo current with collector plate potential for different frequencies but same intensity of incident radiation.



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63. The minimum energy required for the electron emission from the metal surface can be supplied to the free electrons by one of some physical processes. What are they.



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64. Give the experimental study of photo electric effect. Draw the variation curve photo electric current with intensity of light.



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65. Establish Einestein's photoelectric equation.



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66. What is wave nature of matter? What is the hypothesis forwarded by Louis Victor de Broglie in 1924?



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67. A ball of mass 0.12 kg is moving with a speed 20 m/s. Calculate the de Broglie wavelength. (Planck constant $h=6.62 \times 10^{-34}$ J.s)



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68. Describe Milikan's oil drop experiment to determine the charge of an electron.



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69. What are Cathod rays? How they are produced .



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70. Discuss J.J. Thomoson's method for the determination of specific chage of electron.



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71. Give the experimental study of photo electric effect. Draw the variation curve photo electric current with intensity of light.



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72. What is photoelectric effect? Why is photo electric current proportional to the intensity of incident radiation?



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73. State laws of photoelectric emission.

Establish Einesteins photoelectric relation.



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74. Describe photoelectric cell and mention some of its important applications.



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75. Explain why wave theory can't explain photoelectric effect.



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76. Describe Davisson and Germer experiment to establish the wave nature of electron. Draw labelled diagram of the apparatus used.



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77. Calculate the number of photon emitted per second by a 10 watt Na lamp. Assume 90% of consumed energy converted into light. Wave length of Na lamp 530 nm.



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78. Calculate the number of photons emitted per second by a 10 W sodium vapour lamp. Assume that 60% of the consumed energy is

converted into light. Wavelength of sodium light = 590 nm.



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79. Calculate the

de-Broglie wave length of the electrons accelerated through a potential difference 56V.



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