



CHEMISTRY

BOOKS - MS CHOUHAN

ALKYL HALIDES (ELIMINATION REACTION)



1. Which of the following alkyl halide gives only one product (excluding stereoisomer) when undergo E_2 reaction ?

 $(E_2$ = elimination bi - molecular)





Answer: D

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In the above reaction, maximum Saytzeffproduct will obtained when:

In the above reaction Hoffmann product is major when X is :

- A. A) $X=\,-\,I$, -I
- B. B) X = -Cl, -Cl
- C. C) $X= \ -Br$, -Br
- D. D) X = -F, -F

Answer: A

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when (P) undergoes Hoffmann exhaustive methylation (twice) then the product obtained will be :





Answer: A





Product (A) & (B) of the above reaction is

A. A=P,B=P

B. A=Q,B=Q

C. A=P,B=Q

D. A=Q,B=P

Answer: C



Β.

A.



C.



D.

Answer: B

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6. Which of these dehydrates most easily?

A.
$$CH_3 - CH_2 - CH_2 = OH$$





$$\mathsf{D}.\,CH_3 - \overset{CH_3}{\overset{|}{\underset{CH_3}{C}}} - CH_2 - OH$$

Answer: B



7.

Product,

Product of the reaction is :





A.



C.

D. No reaction

Answer: D







A. A)



B. B)



C. C)



D. D)

Answer: B





Answer: C



10. In which of the following reaction Saytzeffalkene is major product?

A. A)
$$CH_3 - CH_2 - \overset{CH_3}{\overset{l}{\underset{CH_3}{CH_3}} - N\overset{\oplus}{M}e_3 \xrightarrow{HO^-}{\overset{}{\underset{CH_3}{\Delta}}}$$

B. B) $CH_3 - CH_2 - CH_2CH - CH_3 \xrightarrow{EtO^-}{\overset{}{\underset{F}{\Delta}}}$
C. C) $CH_3 - CH_2 - \overset{l}{\underset{Br}{CH_3}} - CH_3 \xrightarrow{t-BuOK}{\overset{}{\underset{Br}{\Delta}}}$
D. D) $CH_3 - CH_2 - CH_2 - \overset{l}{\underset{Br}{CH_3}} - CH_3 \xrightarrow{CH_3OK}{\overset{}{\underset{CH_3}{\Delta}}}$

Answer: D

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reaction is



Answer: C



12. The conversion of 2, 3-dibromobutane to 2-butene with Zn is

A. A) Redox reaction

B. B) α - Elimination

C. C) β - Elimination

D. D) Both α - elimination and redox reaciton

Answer: C

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13. 1, 3-Dibromopropane reacts with metallic zinc to form :

A. propene

B. propane

C. cyclopropane

D. 3-bromopropane

Answer: D





obtained in above reactions (1),(2) & (3) is :

A. A = B but C is different

B. A = C, but B is different

C. B = C, but A is different

D. A = B = C all product are identical

Answer: D





Product obtained in above reactions (1),(2) & (3)

A. A = B, C is different

B. A = C, B is different

C. B = C, A is different

D. A = B = C is same

Answer: D



Answer: C





A.





C.



18. Which one of the following compound will be least susceptible to elimination of hydrogen bromide?

A.
$$Br-CH_2-CH_2-NO_2$$

- $\mathsf{B}.\,Br-CH_2-CH_2-CH_3$
- $\mathsf{C.}\,Br-CH_2-CH_2-CN$
- $\mathsf{D.}\,Br-CH_2-CH_2-CO_2Et$

Answer: B

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19. Two alkenes, X(91% yield) and Y(9% yield) are formed when the following compound is heated.



The structures of X and Y, respectively are :



Answer: C



20. In the dehydrohalogenation of2-bromobutane, which conformation

leads to the formation ofcis-2-butene ?













Answer: A





Product (B) of given reaction is :





22. What produict will be formed from Hoffmann exhaustive methylation

of following compound ?

 $Me_2CHCH_2NHCH_2CH_2Me \xrightarrow{(i) CH_3 - 1(excess)}_{(ii) Ag_2O} Product$

A.
$$Me - CH = CH_2$$

B. $H_2C = CH_2$
C. $CH_3 - C = CH_2$
 $CH_3 - CH_3 - CH_2$
 $CH_3 - CH_3 - CH_3 - CH_3$

Answer: A









Β.



C.



D.

Answer: B



24.

, Products

obtained are :

A. Racenic

B. Diastereomers

C. G.I

D. Positional isomers

Answer: A



product of the reaction is :



A.



B.



C.



D.

Answer: B





Major

26.

product of the reaction is :











Answer: D



28. An halide $C_5H_{11}Br$ on treatment with alc. KOH give 2-pentene only. The halide will be :

Br

A. A)
$$CH_3 - CH_2 - CH_3$$

B. B) $CH_3 - CH_3 - CH_2 - CH_2 - CH_3$
 B_r
C. C) $CH_3 - CH_2 - CH_2 - CH_2 - CH_3$
 B_r
D. D) $CH_3 - CH_3 - CH_2 - CH_3$
 B_r

Answer: C

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29. End product CD) in the given sequence is:





(3)
$$H_3C - CH - CH_3 + CH_3O^-$$

 $|_{Br}^{|}$

(4) $H_3C-CH-CH_3+CH_3S^-$



A. A) 2,4,6

B. B) 1,3,5

C. C) 2,3,5

D. D) 2,4,5

Answer: C

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31.

Br Br $CH_3 - \overset{|}{C} - \overset{|}{C} - CH_3 \xrightarrow{xNaNH_2} (yCH_3I)
ightarrow CH_3 - C \equiv C - C \equiv - CH_3$ Br Br x and y mole consumed.

x and y mole consumed .

Value of x + y =

A. 5

B. 6

C. 7
Answer: D



32. The following bimolecular elimination reaction (E_2) is carried out with different halogen leaving groups. The per cent yield of the two products (2-hexene and I-hexene) for each leaving group is listed below.



Which of the following statement is (are) true concerning this series of E_2 reactions?

A. Based on the pK_s 's of the conjugate acid , I^- is the best leaving

group and F^{-} is the pooreset leaving group

B. When $I^{-B}r^{-\text{and}}Cl^{-}$ are used as leaving groups, Zaitsev's rule is

followed.

C. F^{-} is the stronger base (and therefore the poorest leaving group)

and the transition statefor reactionwithfluoride as

theleavinggrouphas theleastdoublebondcharacter

D. a,b,c are true

Answer: D

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33.

Product (A) & (B) respectively are :



Answer: A



$$(3)\Delta$$

Product in above reaction is :





Β.





Answer: A

D.

35. Major product obtained in the reaction of I-phenyl-2-bromobutane with NaOMe is

A. A) (E)-l-phenylbut-l-ene

B. B) (E)-1-phenylbut-2-ene

C. C) I-phenyl-2-ethoxybutane

D. D) (Z)-1-phenylbut-2-ene

Answer: A

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36. Which of the following alkyl halides give most complex mixture of alkene in an E_2 reaction ?

A. A)
$$CH_3-CH_2-CH_2-CH_2-Br$$

B. B)
$$CH_3-CH_2-CH_2-CH_2-CH_3$$

C. C)
$$CH_3-CH_2-CH-CH_2-CH_3$$

 Br
D. D) $CH_3-\overset{CH_3}{\overset{C}{D}}_{Br}-CH_2-CH_3$
 Br

Answer: B

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Product (A) is : 37.







38.

Sum of number of α - hydrogen present is compound A + B is :

A. A) 18

B. B) 19

C. C) 20

D. D) 21

Answer: C







j

Identify A :



A.





Answer: C

















Answer: C



 H_3PO_4

Δ

41.

Major product obtained by dehydration of given alcohol is :





B. B)



C. C)



Answer: C











C.



Answer: B



43.
$$Ph - CH - CH_2 - CH_2 \xrightarrow[Br]{Zn-cu}{A} Product$$

Product of the above reaction is :

A.
$$Ph - CH = CH - CH_2 - Br$$



C.
$$PH-CHBr-CH=CH_2$$

$$\mathsf{D}.\,Ph-C=C-CH_3$$

Answer: B







Β.



C.

Answer: B



Answer: C





is:







Answer: A

C.

D.







A.





D. $H_2C = CH_2$

Answer: D

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product of the reactiono is :





Β.



Answer: D



49.

Stereochemistry of the product is :

A. Meso compound

B. Racemic mixture

C. Diastereomer

D. Optically pure enantiomers

Answer: B

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50.

Which of the following reactantis used to obtain bove compound (A). CAssume that ErO^{-} is used in all the reaction)





Β.





D.

Answer: A



is:





Β.





Answer: B





54.

Relation between (A) and (B) is :



C. Enantiomer

D. Chain isomer

Answer: B





The product is :



- B. B) $HC = C (CH_2)_3 Ona$
- C. C) $NaC = C (CH_2)_3ONa$

D. D) $H - C = C - (CH_2)_3 OH$

Answer: C



56. Which best describes the product of the following reaction?



A. Absolute configuration has been inverted

B. Absolute configuration has been retained

C. Racemization (loss of absolute configuration) has occurred

D. Loss ofchirality has occurred Cthe product is achiral)

Answer: D

57. What is the major product of the following reaction?











D.

Answer: D

58. What will be the major product of each of the two reaction shown

below?

1. $CH_3CH_2CHCH_3 \xrightarrow{heat} CH_3CH_2CHCH_3 \xrightarrow{heat} CH_3CH=CHCH_3+CH_3CH_2CH=CH_2$ 2. $CH_3CH_2CHCH_3 + CH_3CH_2ONa \xrightarrow{heat} X Y$ Br

A. A) 1-X,2-X

B. B) 1 - Y, 2 - X

C. C) 1 - X, 2 - Y

D. D) 1 - Y, 2 - Y

Answer: B



+ $CH_{3}I$ (excess) \longrightarrow product;

59.

product is :

A. A) a primary amine

B. B) a tertiary amine

C. C) a secondary amine

D. D) a quaternary ammonium salt

Answer: D

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(A) on heating isomerizes to (B). What is the structure of (B) ?

The









C.



D.

Answer: B



product (A) is :





62. Which of the following carbocation will undergo rearrngement ?





Β.

C.
$$CH_3 - \mathop{C}_{CH_3} H - \mathop{C}_{C}^{\oplus} = O$$

D. $CH_3 - NH - \mathop{C}_{CH_3}^{\oplus} H - \mathop{C}_{H_3} H - CH_3$

Answer: B

63. In which of the following reaction resonance stabilized product will

form ?



D. All of these

Answer: D

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64. In which offollowing reaction rearrangement take place with change

in carbon skeleton

A.
$$CH_3 - egin{pmatrix} CH_3 \ dots \ CH_3 - egin{pmatrix} CH_3 \ dots \ CH_2 \ dots \ CH_3 \ CH_3 \ \end{pmatrix}$$
B.
$$CH_3 - CH_2CH_2^{\oplus}$$

$$\mathsf{C}.\,CH_3 - \mathop{C}_{|}_{CH_3} H - CH_2 - \mathop{C}\limits^\oplus_{CH_2} H_2$$

D.
$$CH_3 - CH^{\oplus} - CH_3$$

Answer: A

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65. Consider the following reaction:



Which response contains all the correct statement about this process?

- (1) Dehydration
- (2) E_2 mechanism
- (3) Carbon skeleton migration
- (4) Most stable alkene will form
- (5) Single -step reaction

A. 1,3

B. 1,2,3

C. 1,2,5

D. 1,3,4

Answer: D



66.

 $\stackrel{Hbr}{\longrightarrow}(X)\stackrel{Nal}{\underset{ ext{acetone}}{\longrightarrow}}(Y)$ (Major), Product (Y) is :

A. cis-2-butene

B. trans - 2- butene

C. 1-butene

D. Iso-butene

Answer: B

$$\overset{Br}{\stackrel{|}{\stackrel{}}}_{C}H_{2}-CH=CH-\overset{Br}{\stackrel{|}{\stackrel{}}}_{C}H_{2}\overset{Zn\,(\,dust\,)}{\longrightarrow}(A)$$

Above reaction is an example of 1,4-elimination. Predict the product.

A.
$$CH_3 - CH = C = CH_2$$

$$\mathsf{B}.\,CH_3-C=C-CH_3$$

C.
$$CH_3 - CH_2 - C = CH$$

$$\mathsf{D}.\,H_2C=CH-CH=CH_2$$

Answer: D



68.

Major

product of the reaction is :





Β.



C.

D. None of these

Answer: B





(B) rects with H_2SO_4 products obtained are

A. p,q,r,s

B. p,q

C. p,q,r

D. p,q,s

Answer: B

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70. Which of the following compoudn gives even number of Hoffmann's exhaustive methylation







Answer: A::B





D. 8



B.





Answer: B





which of the following is product "A" in the above reaction?



Answer: A



74. Which of the following is true regarding increasing reactivity in the E_2

reaction with
$$Na^+C_2H_5O^-$$
 ?



A. A is more reactive than B

B. B is more reactive thanA

C. A is more reactive than C

D. B is less reactive than C

Answer: A::C::D





Compare of the Reaction .

A. $r_1>r_2$ B. $r_1=r_2$ C. $r_2>r_1$ D. $r_1=rac{r_2}{2}$

Answer: A



Which statement is not true for above reaction ?

A. It is dehydration reaction.

B. The mechanism followed by reaction is E_1

C. The reaction is stereo selective.

D. The major product is Z-alkene.

Answer: D

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Level 2

1. Comprehension

 E_2 raction ightarrow Elimination bimolecular

In the general mechanism of the E_2 reaction a strong base abstract a

proton on a carbon atom adjacent to the one of the leaving group. As the abstracts a proton , a double bond forms and the leaving group leaves.



Identify the rate of reaction of given compounds in E_2 reaction:



A. a > b > c > d

 $\mathsf{B}.\, a > c > b > d$

 $\mathsf{C}.\, b > a > c > d$

 $\mathsf{D}.\,b>d>a>c$

Answer:



2. Comprehension

 E_2 raction ightarrow Elimination bimolecular

In the general mechanism of the E_2 reaction a strong base abstract a proton on a carbon atom adjacent to the one of the leaving group. As the abstracts a proton , a double bond forms and the leaving group leaves.



In given pairs, which compound is more reactive toward E_2 reaction



$(Q)CH_3- egin{array}{ccc} & H-CH_3 & \ & \ & \ & \ & \ & \ & \ & \ & \ & $	$CH_3- egin{array}{cc} C & H-CD_3 \ ert \end{array}$
Br (III)	Br (IV)
()	()



(S)
$$Ph - CH_3 - CH_2Br$$
 $Ph - CH_3 - CH_3$
 $BrVIII$

A. P-II,Q-III,R-VI,S-VII

B. P-II,Q-III,R-VI,S-VI

C. P-I,Q-III,R-VI,S-VII

D. P-I,Q-II,R-V,S-VIII

Answer:



3. Comprehension

 E_2 raction ightarrow Elimination bimolecular

In the general mechanism of the E_2 reaction a strong base abstract a proton on a carbon atom adjacent to the one of the leaving group. As the abstracts a proton, a double bond forms and the leaving group leaves.



Product (A) and (B) are :

A. A = cis, B = cis

B. A = trans, B= cis

C. A= trans, B = trans

D. A = cis, B= trans

Answer:

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4. Comprehension

 E_2 raction ightarrow Elimination bimolecular

In the general mechanism of the E_2 reaction a strong base abstract a proton on a carbon atom adjacent to the one of the leaving group. As the abstracts a proton , a double bond forms and the leaving group leaves.



A. A = P,B=P

B. A=Q,B= Q

C. A = Q,B= P

D. A=Q,B=P

Answer:

5. Comprehension

 E_2 raction ightarrow Elimination bimolecular

In the general mechanism of the E_2 reaction a strong base abstract a proton on a carbon atom adjacent to the one of the leaving group. As the abstracts a proton , a double bond forms and the leaving group leaves.



Which of the following compound is inert toward E_2 reaction.



$${f C.} \, CH_3 = egin{array}{c} CH_3 & & & \ ert \ CH_3 & - \ CH_2 - Br \ ert \ CH_3 & \ ert \ ert \ Br & \ ert \ H_3 & \ e$$

Answer:

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6. Match the column :

	Column (I)		Column (II)
	<i>E</i> ₂ reaction (elimination bimolecular)	No. of possible products (including stereoisomeris	
(a)	$\xrightarrow{\text{Br}}_{\Delta}$ alc. KOH	(p)	0
(b)	$\xrightarrow{\text{alc. KOH}}$ $\xrightarrow{\text{alc. KOH}}$	(q)	1 00 F
(c)	β Br $\xrightarrow{\text{alc. KOH}} \Delta$	(r)	2
(d)	$Br \xrightarrow{alc. KOH} \Delta$	(s)	3

	Column (I)		Column (II)		
	Reaction		Product		
(a)	$ \begin{array}{c} & & \\ & & $	(p)	$\mathrm{H_2C}=\mathrm{CH}-\mathrm{CH_2}-\mathrm{CH}=\mathrm{CH_2}$		
(b)	$\overbrace{\substack{N\\H\\H}}^{HEM} \xrightarrow{HEM}$	(q)	$H_2C = CH - CH_2 - CH_2 - CH_2 = CH_2$		
(c)	$\overbrace{\substack{N \\ H}}^{\text{HEM}} \xrightarrow{\text{HEM}}$	(r)	$\begin{array}{c} \mathrm{CH}_{3} \\ \mathrm{H}_{2}\mathrm{C} = \mathrm{CH} - \mathrm{CH}_{2} - \overset{\mathrm{CH}_{3}}{-\mathrm{C}} = \mathrm{CH}_{2} \end{array}$		
(d)	$() \qquad $	(s)	$\begin{array}{c} \operatorname{CH}_3 \\ \downarrow \\ \operatorname{H}_2 \operatorname{C} = \operatorname{CH} - \operatorname{CH} - \operatorname{CH} = \operatorname{CH}_2 \end{array}$		

HEM = Hoffmann exhaustive methylation followed by elimination .

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8. Match the column :

Column (I)	Column (II)
(a) $(H^{\oplus}) \to (H^{\oplus})$	(p) Product are Diastereomers
(b) $\xrightarrow[Br]{alc, KOH} \xrightarrow{Alc, KOH}$	(q) Carbocation is intermediate
(c) $\bigvee_{OH} \xrightarrow{H^{\oplus}}$	(r) 2nd order reaction
(d) $\underset{Br}{\overset{alc. KOH}{\frown}}$	(s) Ist order reaction

5.37	Column (I)		Column (II)	
(a)	$\overbrace{Cl}^{ACI} \xrightarrow{alc. KOH}$	(p)	Optically active product	
(b)	$\bigcup_{CI}^{ACI} \xrightarrow{alc. KOH}$	(q)	Optically inactive product	
(c)	Ci ↓ aq. KOH	(r)	2nd order reaction	
(d)	$\overbrace{CH_{3}}^{CI} \xrightarrow{aq. KOH}$	(s)	unimolecular reaction	

	Column (I)		Column (II)
	E ₂ reactions (elimination bimolecular)	Number of products (including stereoisomeris	
(a)	$CH_3 - CH_2 - CH_2 - CH_2 - Br \xrightarrow{alc. KOH}$	(p)	1 10
(b)	$ \begin{array}{c} \operatorname{CH}_3 \operatorname{-CH} \operatorname{-CH}_2 \operatorname{-CH}_3 \xrightarrow{ \text{alc. KOH}} \\ \downarrow \\ \operatorname{Br} \end{array} $	(q)	2
(c)	$\begin{array}{c} {}^{CH_3}\\ {}^{I}\\ {}^{CH_3-C}-{}^{CH_2-CH_3} \xrightarrow{ \text{alc. KOH}} \\ {}^{I}\\ {}^{Br}\end{array}$	(r)	3
(d)	Ph-CH ₂ -CH-CH ₂ -CH ₃ I Br	(s)	104 8

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11. Match the column :

Column (I)			Column (II)	
(a)	$\overbrace{OH}^{H^{+}} (A)$	(p)	<i>E</i> ₁	
(b)	$\overbrace{Cl}^{\text{NaNH}_2}$	(q)	E2	
(c)	$CH_3 - C - CH_2 - CH - CH_3 \xrightarrow{EtONa}{\Delta}$ Br	(r)	Ei (elimination intramolecular)	
(d)	() () () () () () () () () ()	(s)	E _{1CB}	







x,y,z are moles used.

sum of [x + y + z = 1]



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Sum of α hydrogen is (A + B + C =)



19. Match the column (I) and (II)

Column (I) Column (I Reaction Type of Reac			Column (II)
		Type of Reaction	
(a)	R -2 -chlorobutane $\xrightarrow{\text{KSH}}_{\text{acctone}}$	(p)	S _{N1}
(b)	$\begin{array}{c} R \ - \ 2- \ chlorobutane & \underbrace{ErO^{-} \overset{ \Phi}{Na}}_{BrOH} \end{array} \end{array}$	(q)	S _{N²}
(c)	2 - bromo- 2- methyl propane $\xrightarrow{H_2O}$	(r)	E1
(d)	2- butanol $\xrightarrow{H_2SO_4}_{\Delta}$	(s)	E2

20. Match the column (I) and (II)

	Column (l)		Column (II)
	Reaction		Type of Reaction
(a)	CIaq.KOH ,	(p)	S _{N1}
(b)	Cl 	(q)	S _{N²}
(c)		(r)	E1
(d)	$\rightarrow \qquad \qquad$	(s)	E2

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21. Select whether the following reaent combination will result in elimination of substitution reactions leading of the major product.

	Reaction	Substitution	Elimination
(a)	$\begin{array}{c} CH_{3} \\ CH_{3} - \overset{ }{\underset{H}{C}} - Cl \xrightarrow{\kappa^{\oplus} \widetilde{O}C(CH_{3})_{3}} \\ H \\ H \end{array} \xrightarrow{H} $		
(b)	$\begin{array}{c} CH_{3} \\ CH_{3} - \overset{ }{C} - OH \xrightarrow{H_{2}SO_{4}} \\ \stackrel{ }{\underset{CH_{3}}{}} \end{array} \rightarrow$		







23. For each of the following amines (A through D), exhaustive methylation (treatment with excess methyl iodide), followed by Hoffmann elimination (heating with AgOH), repeated as necessaly, removes the nitrogen atom in the form oftrimethylamine. Indicate the numberof repetitive Hoffmann eliminations required to remove the nitrogen by a

number (1 to 4) in the designated answer sheet.



24.	Complete	the	following	reaction
	$\longrightarrow x \text{ is total number of } eliminations) to remove a line KOH \rightarrow y is total number of x$	of HEM (Hoffman Ex e nitrogen from given f possible E ₂ produce	haustive Methylation and compound. t (including stereoisomer)	
Sum of x	+y= ?			
	/atch Video Solutior	1		

