

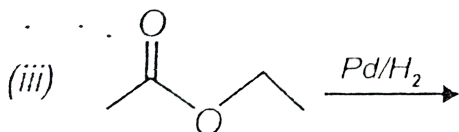
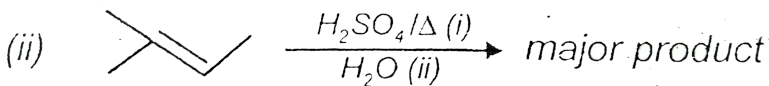
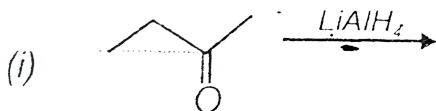
## CHEMISTRY

### AAKASH INSTITUTE ENGLISH

### ALCOHOLS, PHENOLS AND ETHERS

#### Example

1. Give IUPAC name of the expected product in the following reactions



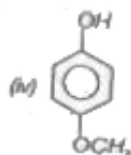
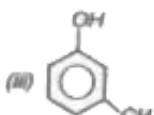
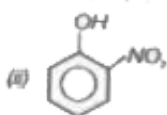
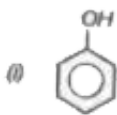
2. Calculate the frequency and wave number of radiation with wavelength 480 nm.

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3. Arrange the following in decreasing order of acidic nature of

(a) (i)  $F_3CCH_2OH$  , (ii)  $(CH_3)_3CCH_2OH$  , (iii)  $FCH_2CH_2C_2OH$

(b)



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4. a : When  $3^\circ$  alkyl halide is used in Williamson 's synthesis what will be the major product and why ?

b : When  $3^\circ$  alkoxide is used in Williamson's synthesis what will be the major product and why ?

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5. Find the pH of a 0.002N acetic acid solution if it is 2.3% ionised at a given dilution.

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## Try Yourself

1. The pH value of decinormal solution of  $\text{NH}_4\text{OH}$ , which is 20% ionised, is:

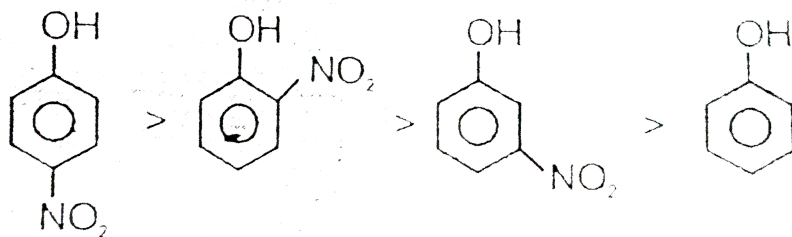
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2. The pH of  $10^{-3}\text{M } \text{NH}_4\text{OH}$  ( $K_b = 10^{-5}$ ) is:

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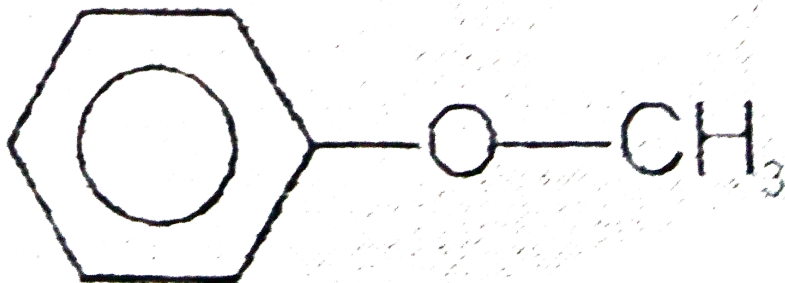
3. (a) Why acidic nature of alcohol and phenol increase with electron withdrawing substituent

(b) Explain the order of acidic nature of



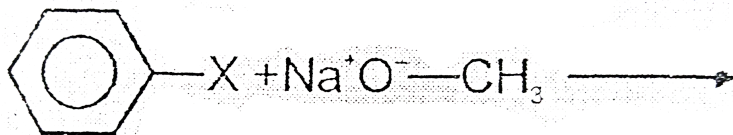
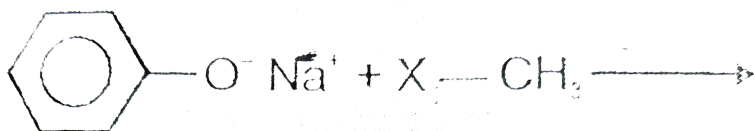
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4. For the preparation of anisole



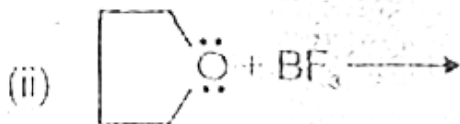
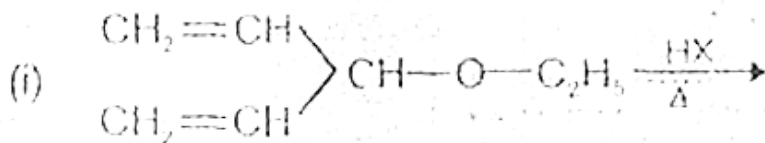
which one is

preferable reaction and why?



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5. Predict final products in the reactions



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## Exercise

1. Ketones can be converted to tertiary alcohols by

A. Reduction

B. Oxidation

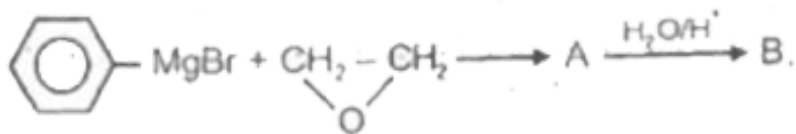
C. Reaction with Grignard reagent

D. All of these

Answer: C

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2. In the sequence of reactions, . The product B is



- A. Benzyl alcohol
- B. 2-phenyl ethanol
- C. 1-phenyl ethanol
- D. Quinol

Answer: B

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3. Iso - butylene when subjected to hydroboration oxidation reaction yields

- A. Sec-butyl alcohol
- B. Tert-butyl alcohol
- C. Iso-butyl alcohol
- D. n-butyl alcohol

**Answer: C**

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4. Identify the final product  $CH_2 = CH_2 \xrightarrow{Cl_2 / H_2O} X \xrightarrow{-aq. KOH} Y$

- A. 2-chloroethanol
- B. 2-chloromethanol



C. 1-chloroethanol

D. Ethylene glycol

**Answer: D**

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5. 3 - methyl - 1 - butene on oxymercuration - demercuration yields ...

... As the major product

A. 3-methyl-2-butanol

B. 2-methyl-3-butanol

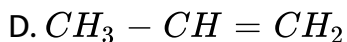
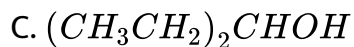
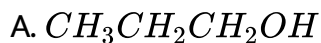
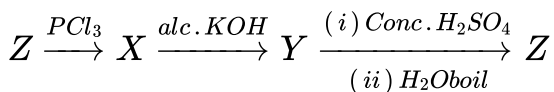
C. 3-methyl-1-butanol

D. 2-methyl-1-butanol

**Answer: A**

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6. What is Z in the following sequence of reactions?



**Answer: B**



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7. Which of the following is the strongest base ?

A. tert-butoxide

B. Ethoxide

C. iso-propoxide

D. Methoxide

**Answer: A**



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8. An alcohol on vigorous oxidation is found to give ethanoic acid and propanoic acid . The alcohol may be

A. 1-pentanol

B. 2-pentanol

C. 1-butanol

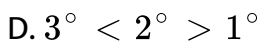
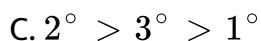
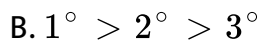
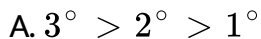
D. 2-butanol

**Answer: B**



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9. The order of reactivity of alcohols with sodium metal is

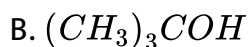
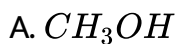


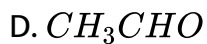
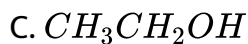
Answer: B



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10. Which one of the following compounds would not be easily oxidised by  $K_2Cr_2O_7$  in  $dilH_2SO_4$  ?

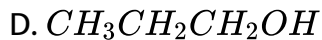
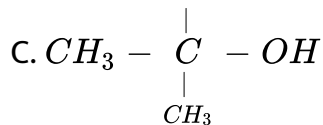
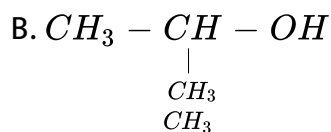
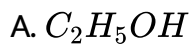




Answer: B

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11. Which of the following alcohols can be most easily dehydrated?



Answer: C

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12. When wine is exposed to air it becomes sour due to

A. Bactena

B. Oxidation of  $C_2H_5OH$  to  $CH_3COOH$

C. Virus

D. Formic acid formation

**Answer: B**



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13. Starch is converted to maltose by

A. Zymase

B. Maltase

C. Diastase

D. Invertase

Answer: C

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14. Rectified spirit is a mixture of

A. 95 %  $C_2H_5OH$  and 5 %  $H_2O$

B. 94 %  $C_2H_5OH$  and 6 %  $H_2O$

C. 95.6 %  $C_2H_5OH$  and 4.4 %  $H_2O$

D. 94.47 %  $C_2H_5OH$  and 5.53 %  $H_2O$

Answer: C

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15. Methanol and ethanol can be distinguished by

- A. Lucas test
- B. Iodoform test
- C. Victor Meyer's test
- D. All of these

**Answer: B**



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16. In Reimer Tiemann reaction dichlorocarbene acts as

- A. Nucleophile
- B. Electrophile
- C. Free radical



D. All of these

**Answer: B**

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**17. Carbolic acid is**

A. Phenol

B. Phenyl benzoate

C. Phenyl acetate

D. Salot

**Answer: A**

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18. Which of the following is a trihydric phenol ?

A. Resorcinol

B. p - cresol

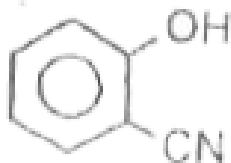
C. phloroglucinol

D. Catechol

Answer: C

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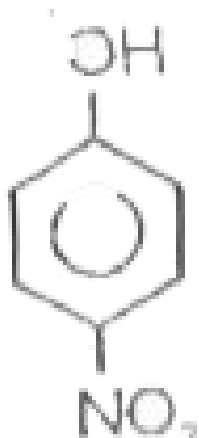
19. Which of the following is the strongest acid ?



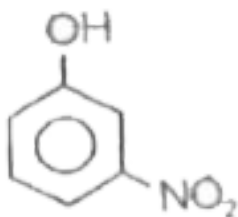
A.



B.



C.



D.

**Answer: C**



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20. Electrophilic substitution reaction in phenol take place at :

A. Ortho and para positions

B. Meta - position

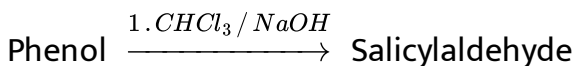
C. Ortho - position

D. Para - position

**Answer: A**

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21. The following reaction is known as :



A. Gattermann aldehyde synthesis

B. Sandemeyer's reaction

C. Kolbe's reaction

## D. Reimer - Tiemann reaction

**Answer: D**

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22. What amount of bromine will be required to convert 2g of phenol into 2, 4, 6 – tribromphenol

A. 4. 0

B. 6. 0

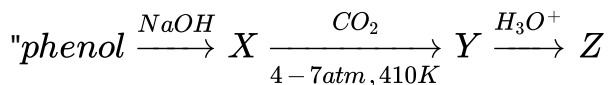
C. 10. 22

D. 20. 44

**Answer: C**

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23. Identify the product Z in the following sequence of reactions



- A. Aspirin
- B. Salicylaldehyde
- C. Benzoic acid
- D. Salicylic acid

Answer: D



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24. Phenol can be distinguished from ethanol by the following reagents except

- A. NaOH
- B.  $\text{FeCl}_3$

C.  $Br_2 / H_2O$

D. Na

**Answer: D**



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**25.** The most suitable method of separation of 1:1 mixture of ortho and para-nitrophenols is :

sublimation

chromatography

crystallisation

steam distillation

A. Sublimation

B. Chromatography

C. Crystallisation

D. Steam distillation

**Answer: D**

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26. Neutral  $FeCl_3$  gives purple colour with

A. Only phenol

B. p-cresol

C. 2,4,6-tribromophenol

D. All of these

**Answer: D**

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27. Aspirin is obtained by the reaction of salicylic acid with

A. Acetic anhydride

B. Acetaldehyde

C. Acetyl chloride

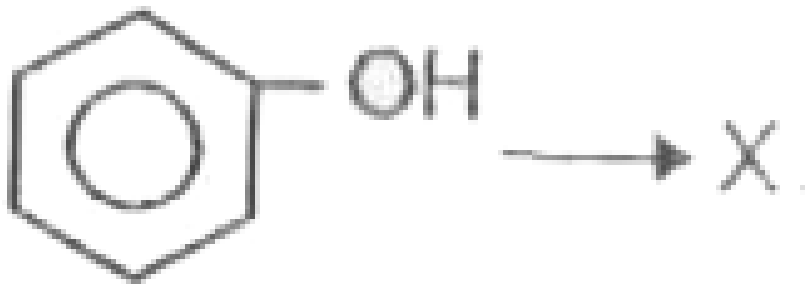
D. Methanol

**Answer: A**



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28. Phenol on treatment with zinc powder gives X . In the above reaction X will be



- A. Benzaldehyde
- B. Benzene
- C. Anisole
- D. Phenyl acetate

**Answer: B**

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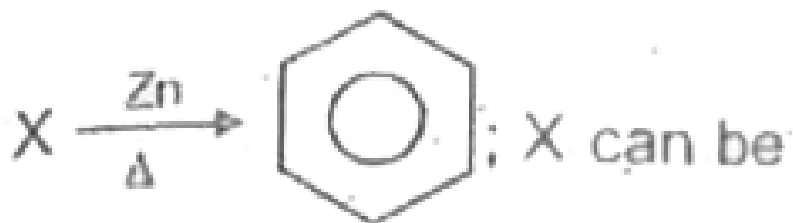
29. The ionization constant of a phenol is higher than that of ethanol because

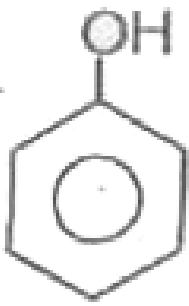
- A. Phenoxide ion is a stronger base than ethoxide ion
- B. Phenoxide ion is stabilized through delocalisation
- C. Phenoxide ion is less stable than ethoxide ion
- D. Phenoxide ion is bulkier than ethoxide ion

**Answer: B**

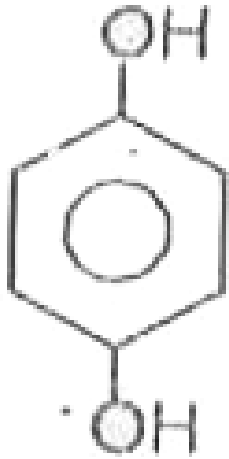
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30. Can be

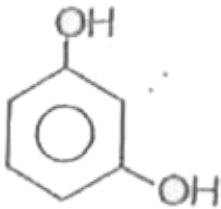




A.



B.



C.

D. All of these

**Answer: D**



31. The number of metamers possible for  $C_4H_{10}O$  is

A. 2

B. 3

C. 4

D. 1

**Answer: B**

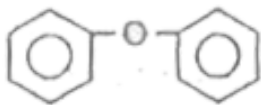
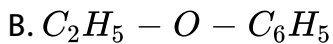


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32. Which of the following has maximum bond angle around oxygen

?

A.  $CH_3 - O - CH_3$



C.

D. Same in all

**Answer: C**

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33.  $OCH_3$  group is

A. Stronger + R group than - OH

B. Weaker + R group than - OH

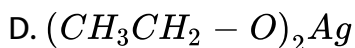
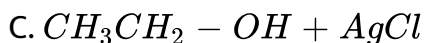
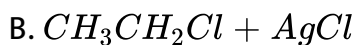
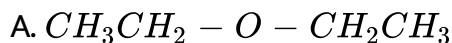
C. Stronger + I group than - OH

D. Inert group

**Answer: B**

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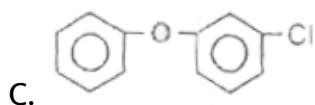
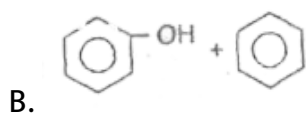
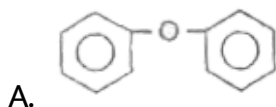
34.  $CH_3CH_2Cl + Ag_2O \xrightarrow{\Delta}$  Product . Product formed in the reaction is



**Answer: A**

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35. Product of the reaction is



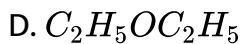
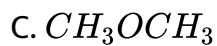
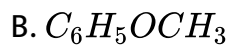
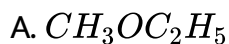
D. No reaction

Answer: D

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36. The ether that undergoes electrophilic substitution reaction is





**Answer: B**

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37. When ether is exposed to air for some time, an explosive substance produced is :

A. Peroxide

B. TNT

C. Oxide

D. Superoxide

**Answer: A**

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**38.** An ether is more volatile than alcohol having the same molecular formula. This is due to

- A. Inter - molecular hydrogen bonding in alcohols
- B. Dipolar character of ethers
- C. Alcohols having resonance structure
- D. Inter - molecular hydrogen bonding in ether

**Answer: A**

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**39.** Anisole with conc.  $HNO_3$  and conc.  $H_2SO_4$  gives

- A. Phenol
- B. nitrobenzene
- C. o-and p-nitroanisole
- D. o-nitroanisole

**Answer: C**

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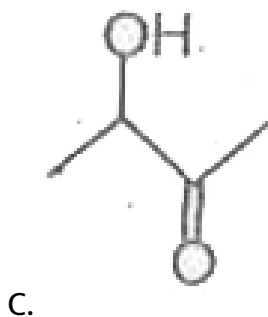
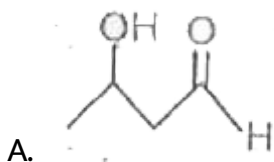
**40. Williamson's synthesis involves**

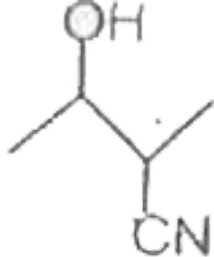
- A.  $S_N1$  mechanism
- B.  $S_N2$  mechanism
- C. Nucleophilic addition
- D. Electrophilic addition

**Answer: B**

## Assignment Section A Objective Type Questions

1. Which among the following is 1 alcohol ?



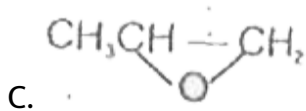
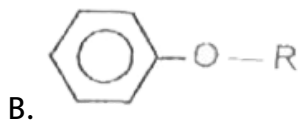


D.

**Answer: B**

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2. Which one is ether ?

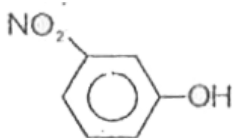


D. all of these

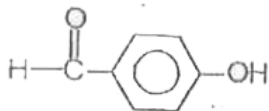
Answer: D

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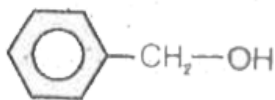
3. Which one is phenol ?



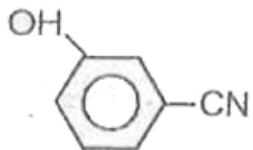
A.



B.



C.



D.

Answer: A

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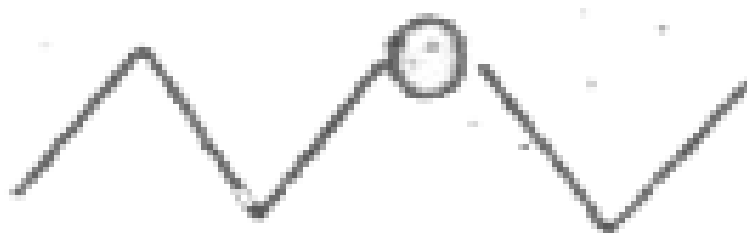
4. IUPAC name of  $\text{HO} - \text{CH}_2\text{CH}_2 - \text{OH}$  is

- A. Ethylene glycol
- B. Ethane -1,2-diol
- C. Ethyl -1,2-diol
- D. Ethylene diol

**Answer: B**

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5. IUPAC name of is



A. Ethyl propyl ether

B. Propyl elhoxide

C. Ethoxy propane

D. Propoxy ethane

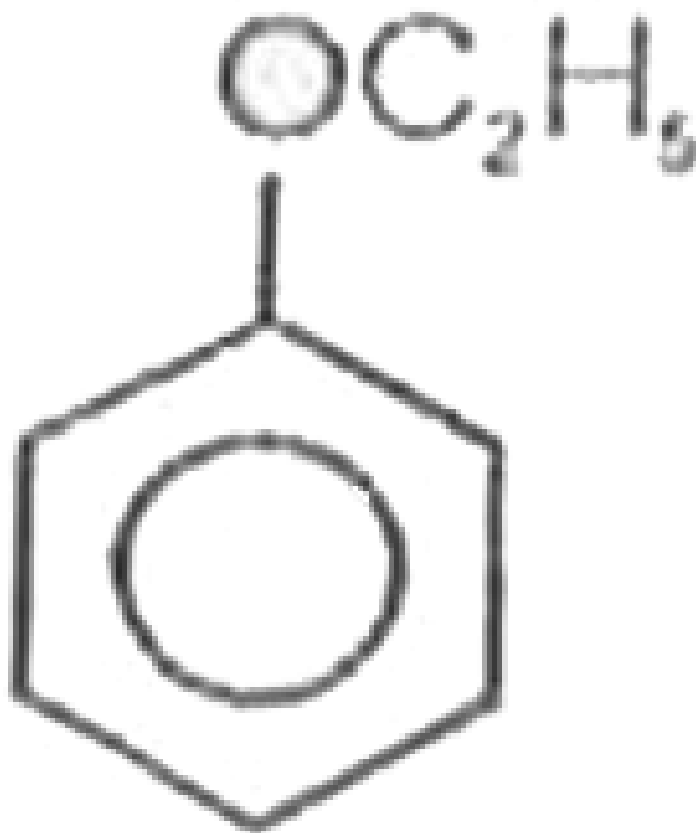
**Answer: C**



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6. IUPAC name of is



- A. Benzyl ethoxide
- B. Ethoxy benzyl
- C. Benzene ethoxide

D. Ethoxy benzene

**Answer: D**

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7. Which among the following show tautomerism ?

A. Alcohols

B. Phenol

C. Ethers

D. Anisole

**Answer: B**

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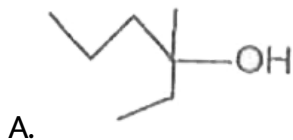
8. Alcohols and ethers are

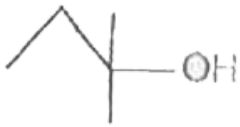
- A. Position isomers
- B. Functional isomers
- C. Chain isomers
- D. Metamers

Answer: B

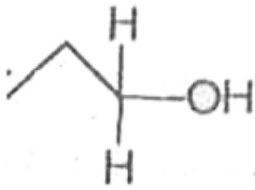
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9. Which of the following be optically active ?

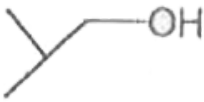




B.



C.



D.

**Answer: A**



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**10.** How many minimum number of carbons are needed for an optically active ether ?

A. 2

B. 3

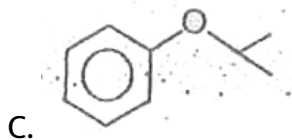
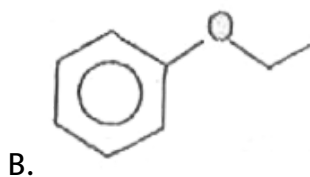
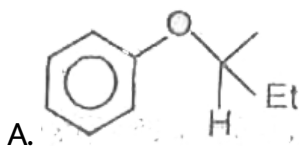
C. 4

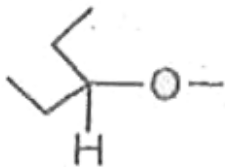
D. 5

**Answer: B**

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**11. Which one is optically active aromatic ether ?**





D.

**Answer: A**

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12.  $3^\circ$  alkyl halides form alcohols preferably via

A.  $S_N2$

B.  $S_N1$

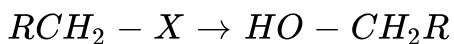
C. Transition state

D.  $S_Ni$

**Answer: B**

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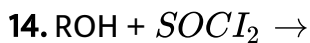
13. Which one is preferable reagent for given reaction ?



- A. ( $H_2O + KOH$ )
- B. ( $OH + KOH$ )
- C. ( $ROH + KOH$ ) /  $\Delta$
- D. ( $H_2O + KOH$ ) /  $\Delta$

**Answer: A**

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The final product is

- A. Alkyl chloride
- B. Alkyl sulphate

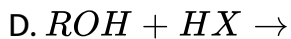
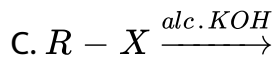
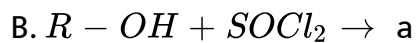
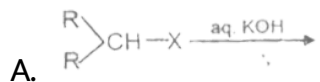
C. Alkene

D. Ether

**Answer: A**

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15.  $S_N1$  is observed in

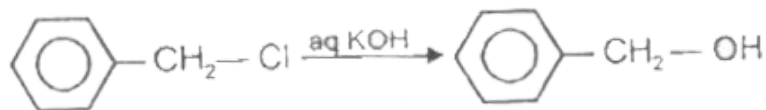


**Answer: A**

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16. Reaction happens via

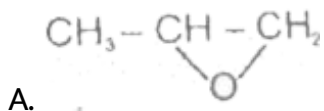


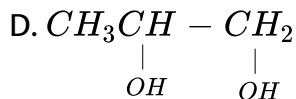
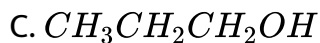
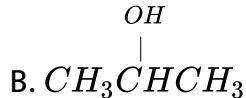
- A.  $S_N1$
- B.  $S_N2$
- C.  $S_Ni$
- D.  $ArS_N1$

Answer: A

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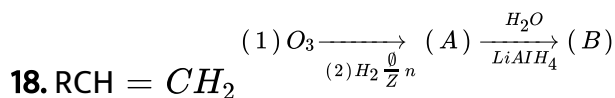
17.  $\text{CH}_3\text{CH}=\text{CH}_2 \xrightarrow{\text{H}/\text{H}_2\text{O}}$  major product is



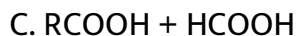
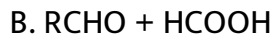
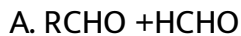


**Answer: B**

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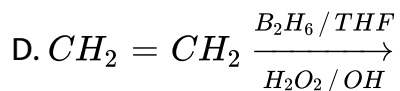
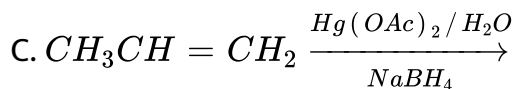
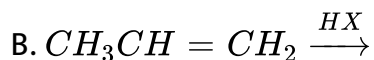
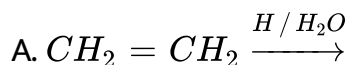
Product (B) is



Answer: D

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19. Reaction involvin anti addition is



Answer: B

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20. Grignard reagent is most suitable for preparation of which of the following alcohol with carbonyl compound ?

A. 1° alcohols

B. 2° alcohols

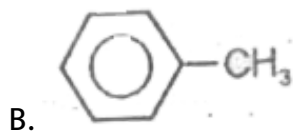
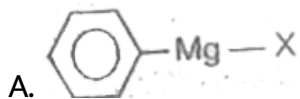
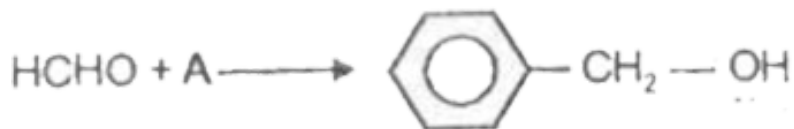
C. 3° alcohols

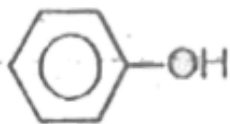
D. All of these

**Answer: D**

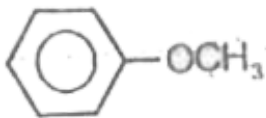
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21. A is





C.

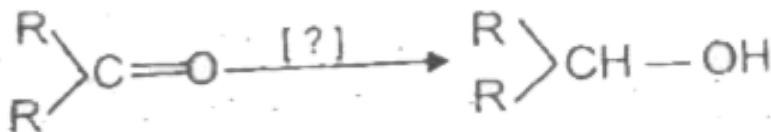


D.

**Answer: A**

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22. Here reagent is



A.  $\text{LiAlH}_4$

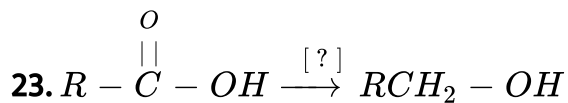
B.  $\text{NaBH}_4$

C.  $\text{Ni} / \text{H}_2$

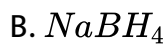
D. All of these

Answer: D

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Here reagent is



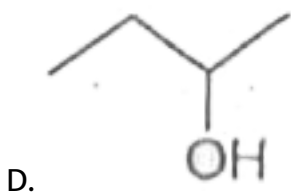
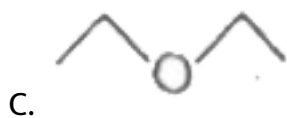
C. Both (1) & (2)

D. Red P /HI

Answer: A

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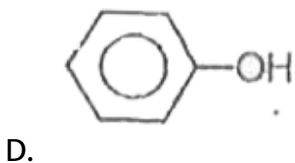
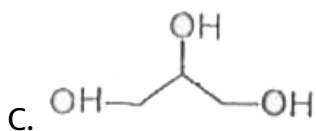
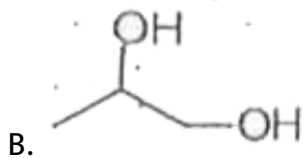
24. Boiling point will be least for



Answer: B

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25. Which of the most viscous ?



**Answer: C**

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26. Lowest boiling point is for

A. Butanol

B. Pentanol

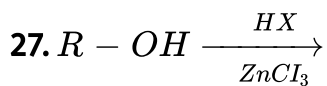


C. 2-methyl propane - 2 - ol

D. 2-methyl butane -2-ol

**Answer: C**

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A. R - X

B. Alkene

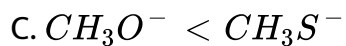
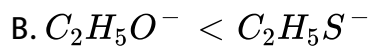
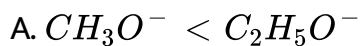
C. Both (1) & (2)

D. No product

**Answer: A**

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28. Order of nucleophilicity is



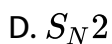
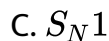
D. All of these

Answer: D



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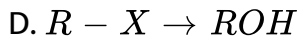
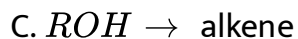
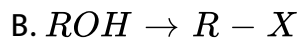
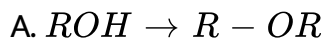
29.  $1^\circ$  alcohols preferably undergo dehydration via



**Answer: B**

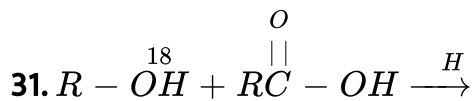
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**30.** Which of the inter- molecular dehydration ?

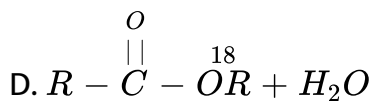
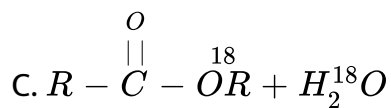
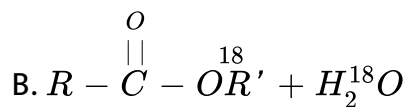
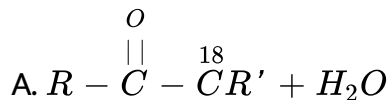


**Answer: A**

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Products are



Answer: A

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32. Lucas test is used to distinguish

A. Phenols

B. Ethers

C. Alcohols

D. Alkyl halides

**Answer: C**

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**33.** Which of the following can give immediate turbidity on treatment with Lucas Reagent?

A.  $3^\circ$  alcohols

B.  $2^\circ$  alcohols

C.  $1^\circ$  alcohols

D. Phenol

**Answer: A**

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34. Phenols can be distinguished from alcohols by

A.  $FeCl_3$  (neutral)

B. Fehling solution

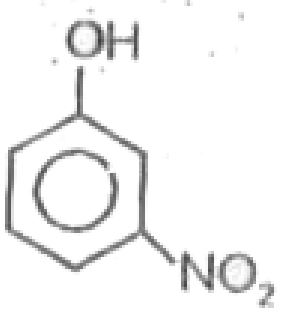
C. Tollen's reagent

D. 2,4-DNP

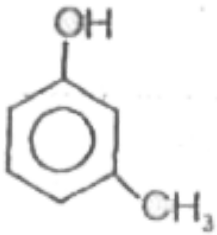
**Answer: A**

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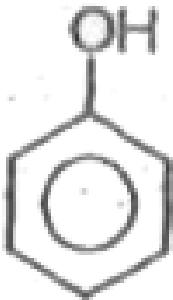
35. Most acidic among the following is



A.



B.



C.

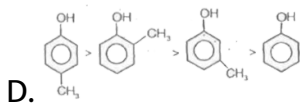
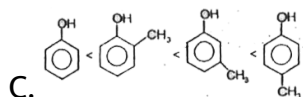
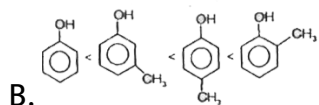
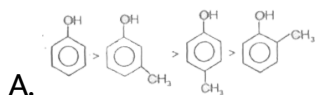


D.

Answer: D

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36. Correct acidic order is

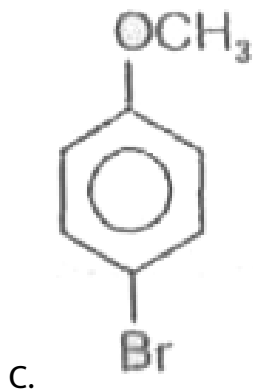
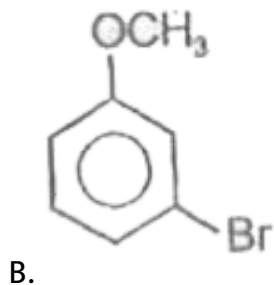
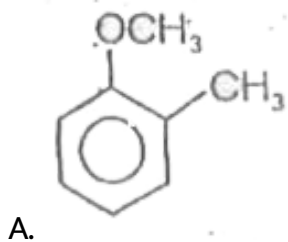
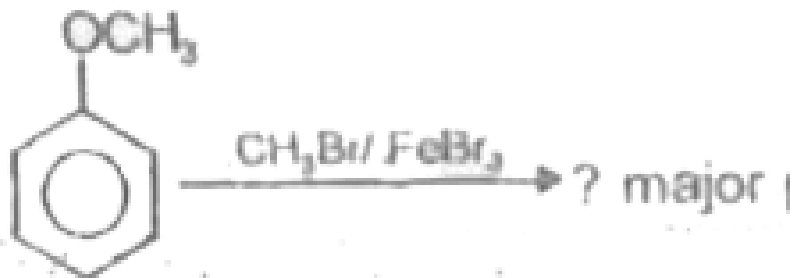


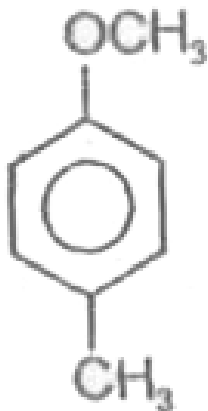
Answer: A

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37. Major product is

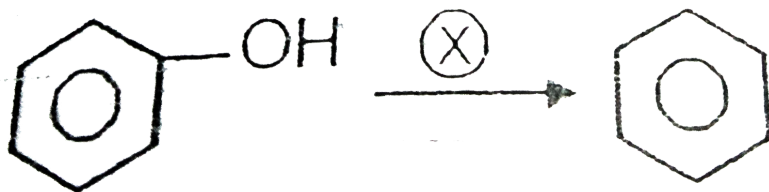




D.

Answer: D

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38.

The reagent (X) required for above conversion is

A.  $\text{LiAlH}_4$

B. Zn



C.

D.  $NaBH_4$

**Answer: B**



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**39.** The electrophile involved in the Reimer-Tiemann reaction is

A.  $CHCl_3$

B.  $:CH_2$

C.  $:CCl_2$

D.  $CO_2$

**Answer: C**

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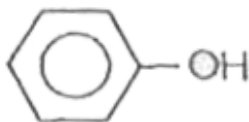
40. In Reimer - Tiemann reaction the major product is

- A. Ortho isomer due to intra molecular H-bonding
- B. Meta isomer
- C. Para isomer due to symmetry
- D. None of these

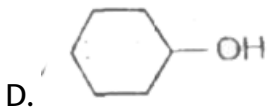
**Answer: A**

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41. Molecule which does not oxidise



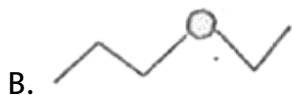
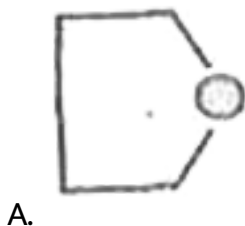
A.

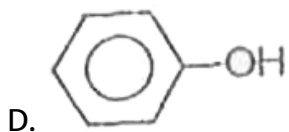
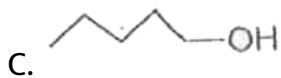


Answer: A

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42. Which one of the following is best lewis base ?

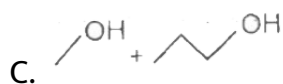
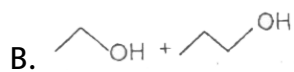
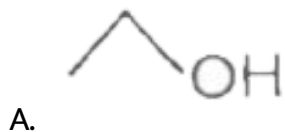
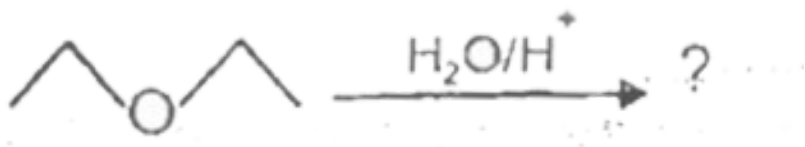




Answer: B

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43. Product / (S) will be :

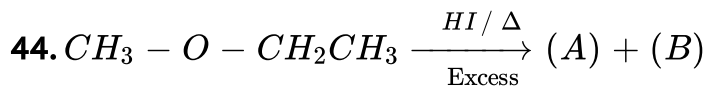


D.

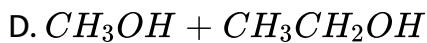
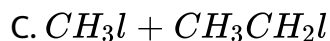
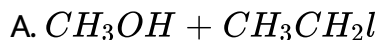


**Answer: A**

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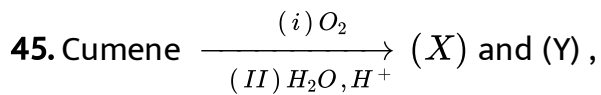


Product (A) and (B) are



**Answer: C**

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(X) and (Y) respectively are :

- A. Toluene, propene
- B. Toluene, propylchloride
- C. Phenol, acetone
- D. Phenol, acetaldehyde

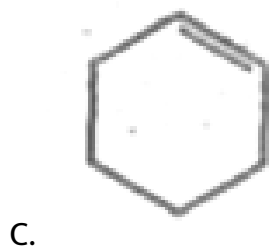
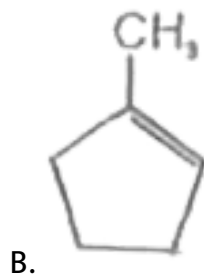
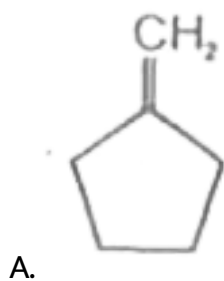
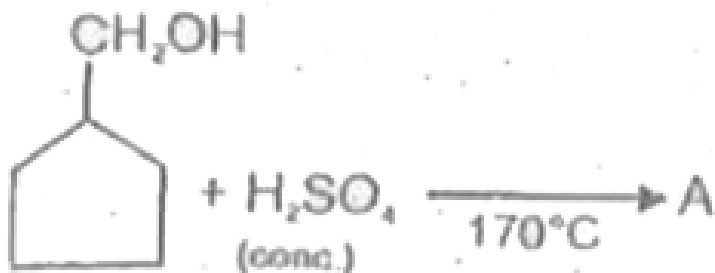
**Answer: C**

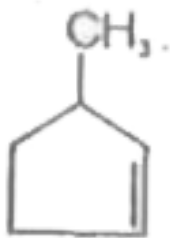
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**Assignment Section B Objective Type Questions**



1. What is the major product A?



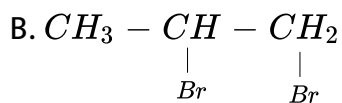
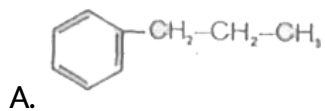
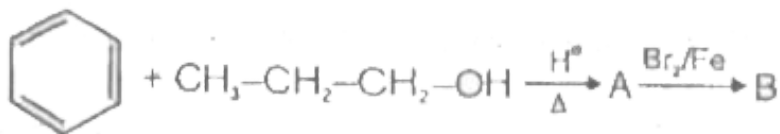


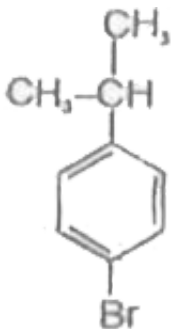
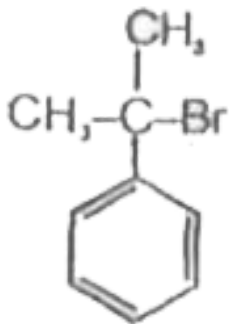
D.

Answer: C

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2. What is the major product B ?

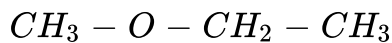




**Answer: D**

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3. Which of the following is the correct increasing order of boiling point of following compounds ?



A.  $II < I < III$

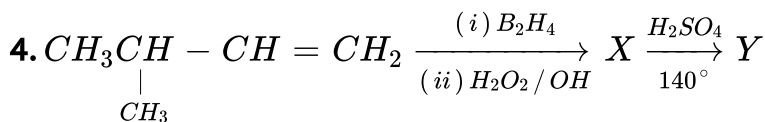
B.  $III < II < I$

C.  $I < II < III$

D.  $II < III < I$

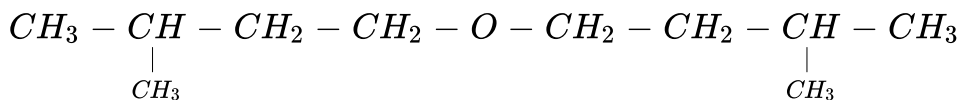
Answer: B

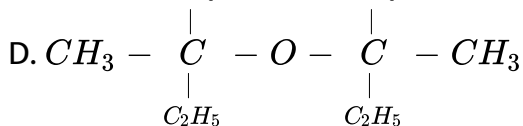
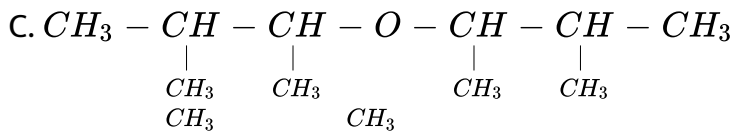
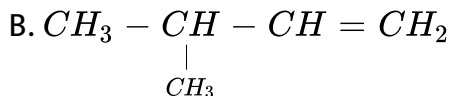
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What is Y?

A.

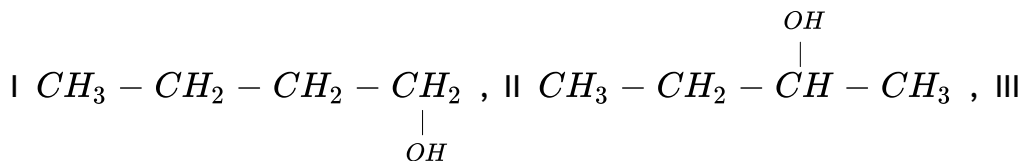


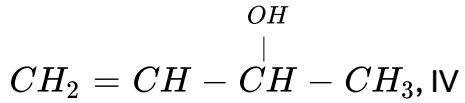


Answer: A

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5. Which of the following is the correct ease of dehydration ?





IV

A.  $I > III > II > IV$

B.  $IV > III > II > I$

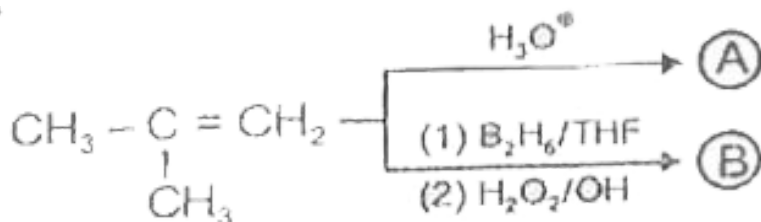
C.  $IV > II > III > I$

D.  $III > IV > II > I$

Answer: B

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6. Product(A) and (B) can be distinguished by

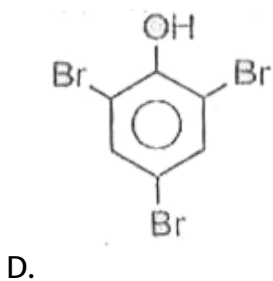
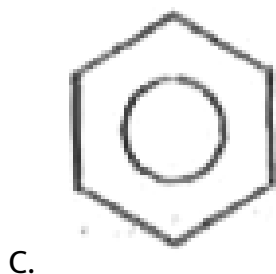
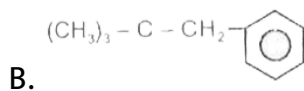
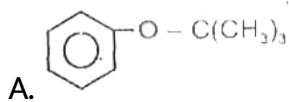
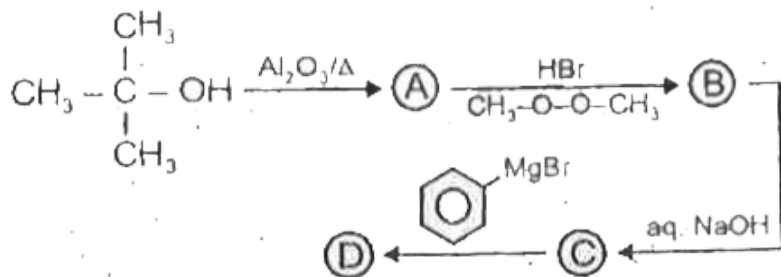


- A. Sodium metal
- B. Neutral  $\text{FeCl}_3$
- C. Lucase reagent
- D. Esterification reaction

Answer: C

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7. The end product (D) of the reaction is

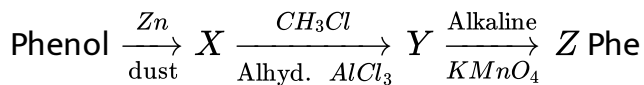




Answer: C

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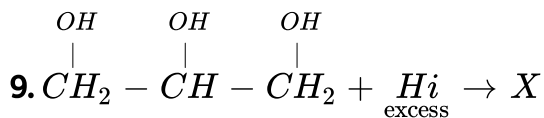
8. What is Z in the following sequence of reactions?



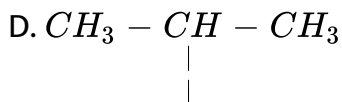
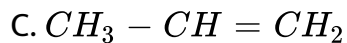
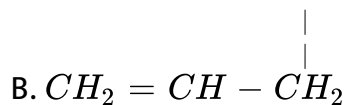
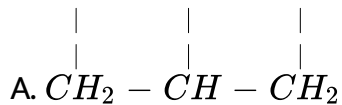
- A. Benzene
- B. Toluene
- C. Benzaldehyde
- D. Benzoic acid

Answer: D

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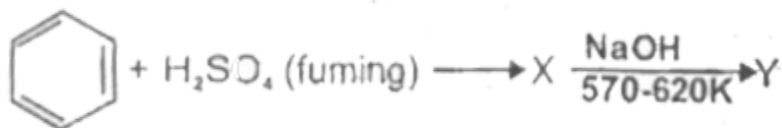
What is X ?

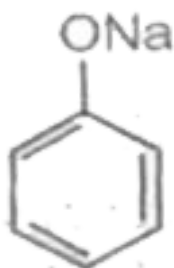
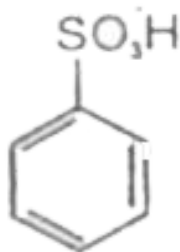
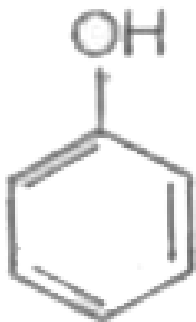


Answer: D

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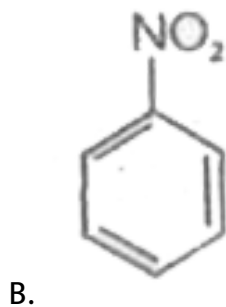
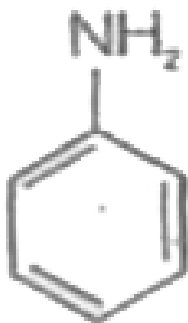
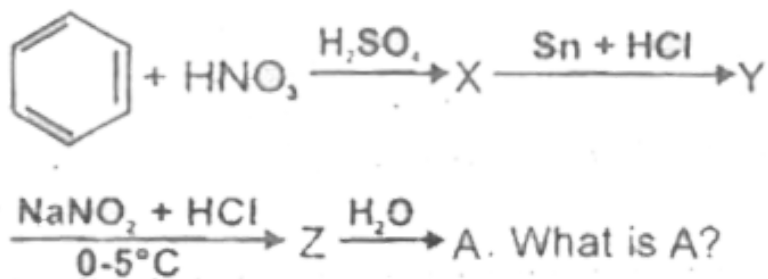
10. What Y ?



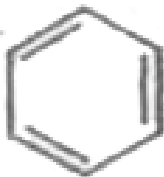


Answer: C

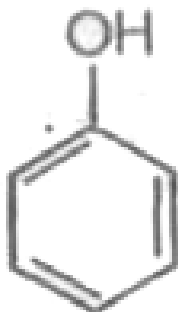
11. What is A?



C.



D.

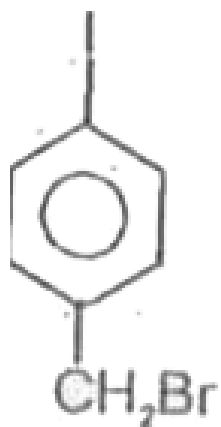
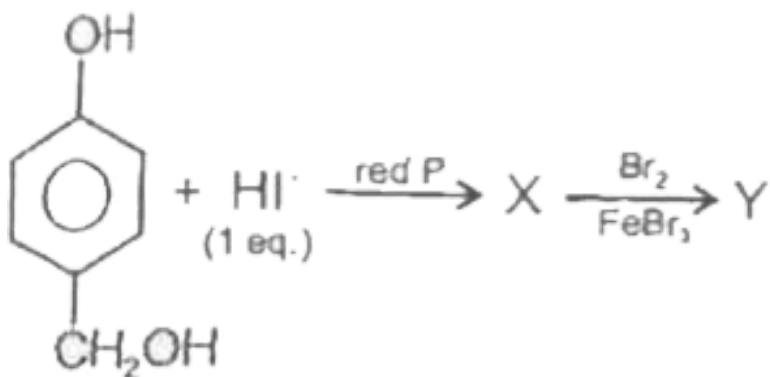


**Answer: D**

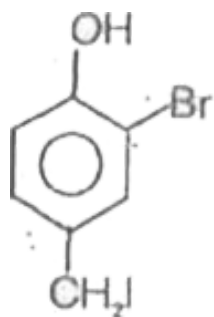


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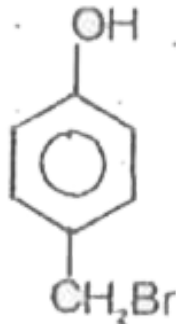
12. What is Y?



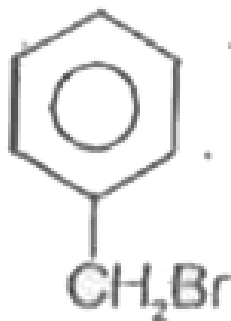
A.



B.



C.



D.

**Answer: B**

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**13.** Which of the following will not give positive test with neutral  $FeCl_3$  ?

A. Nitrophenol

B. Phenol

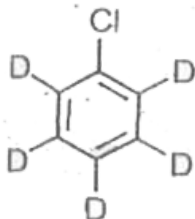
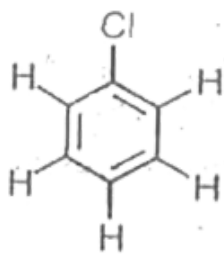
C. Allyl alcohol

D. o-cresol

**Answer: C**

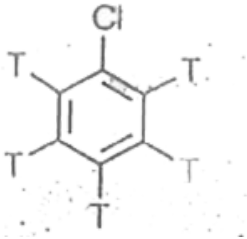
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**14.** In Dow's process haloarene is converted to phenol with fused NaOH . The most reactive compound is



B.





C.

D. All are equally reactive

**Answer: A**

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**15.** Among the following four compounds

(a) Phenol

(b) methyl phenol

(c) metanitrophenol

(d) paranitrophenol

the acidity order is –

A.  $a > c > a > b$

B.  $c > d > a > b$

C.  $a > d > c > b$

D.  $b > a > c > d$

**Answer: A**

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**16.** The reaction of tertiary butyl bromide with sodium methoxide gives

A. Sodium -t - butoxide

B. t-butyl methyl ether

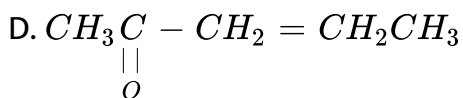
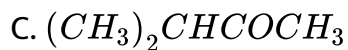
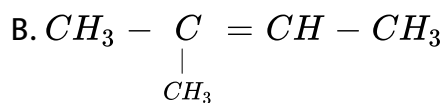
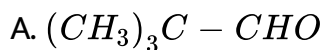
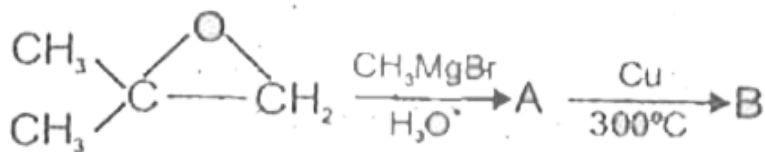
C. Isobutane

D. Isobutylene

**Answer: D**

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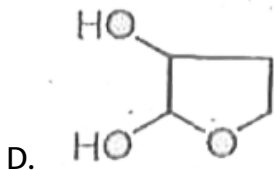
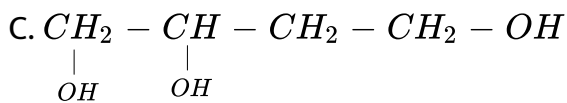
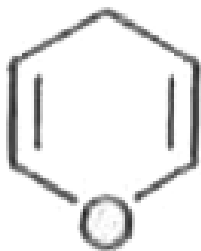
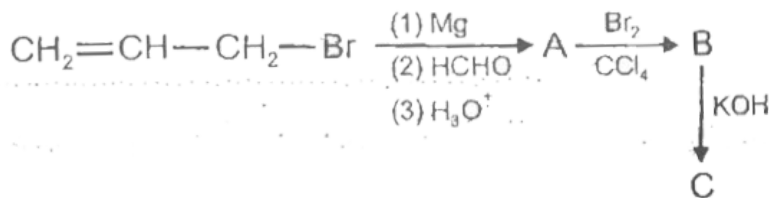
17. B is



Answer: B

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18. Product (C) is



Answer: C



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19. Ethyl chloride is converted into diethyl ether by

- A. Perkin's reaction
- B. Grignard reaction
- C. Wurtz synthesis
- D. Williamson's synthesis

**Answer: D**



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20. Ethylene oxide when treated with Grignard reagent yields

- A. Primary alcohol
- B. Secondary alcohol

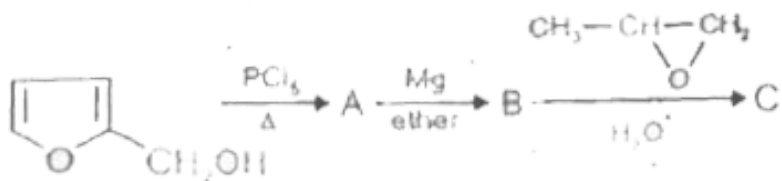
C. Tertiary alcohol

D. Cyclopropyl alcohol

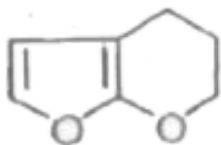
**Answer: A**

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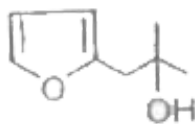
21. Product (C) is

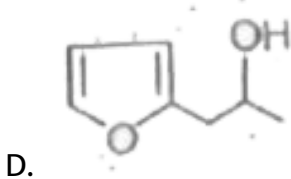
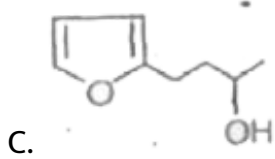


A.



B.





Answer: C

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22. Product © is



A. Alkyl iodide

B. Vinyl chloride

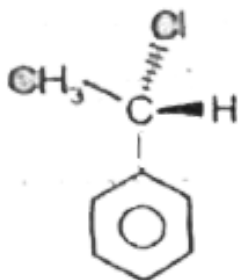
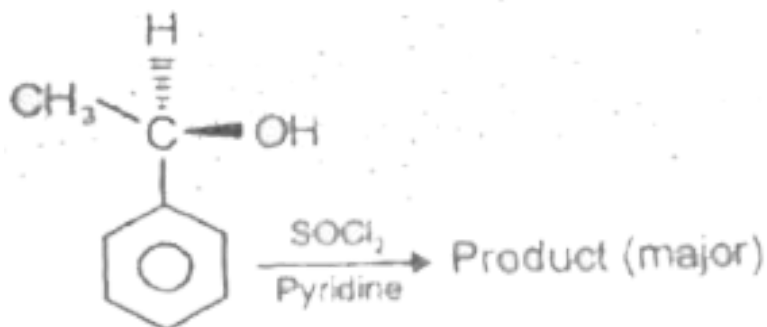
C. Vinyl iodide

D. Allyl chloride

Answer: D

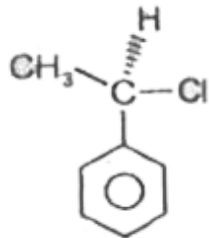
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23. The product is



A.





B.

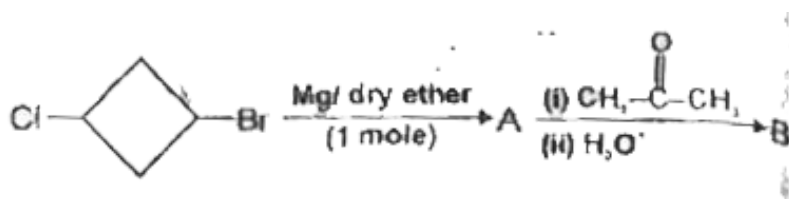
C. Mixture of (1) & (2)

D. No reaction

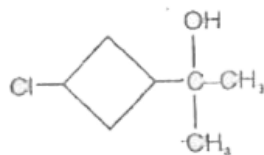
**Answer: A**

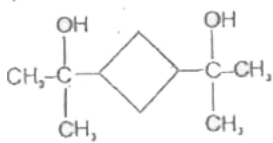
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24. What is B ?

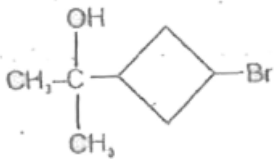


A.





B.



C.

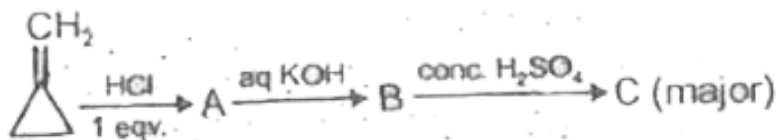


D.

Answer: A

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25. Product (C) is





A.



B.



C.



D.

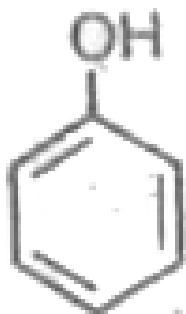
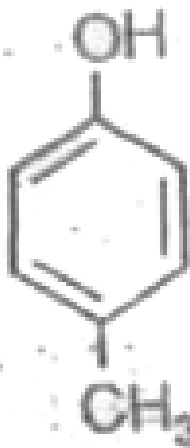
Answer: B



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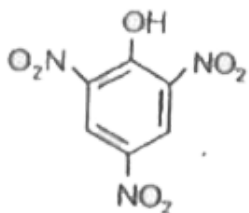
## Assignment Section C Previous Years Questions

1. Which one is the most acidic compound?





C.



D.

**Answer: D**

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2. The heating of phenyl-methyl ethers with *HI* produces

A. Ethyl chlorides

B. Iodoenzyme

C. Phenol

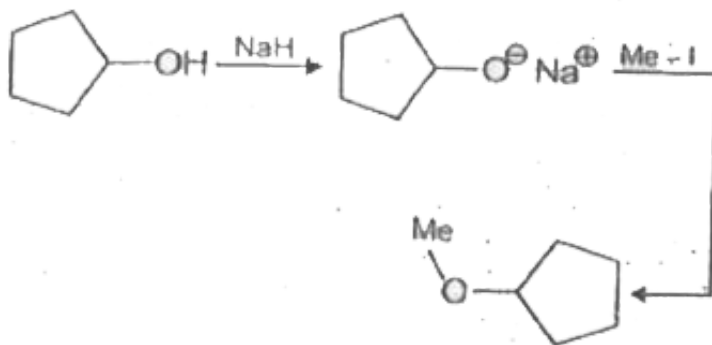
D. Benzene

**Answer: D**

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**3.** The reaction

can be classified as



A. Williamson alcohol synthesis reaction

B. Williamson ether synthesis reaction

C. Alcohol formation reaction

D. Dehydration reaction

**Answer: B**

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4. Which of the following reagents would distinguish cis-cyclopentane-1,2-diol from the trans-isomer?

A. Aluminium isopropoxide

B. Acetone

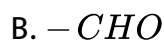
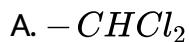
C. Ozone

D.  $MnO_2$

**Answer: B**

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5. Reaction of phenol with chloroform in presence of dilute sodium hydroxide finally introduces which one of the following functional group ?



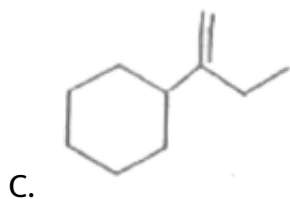
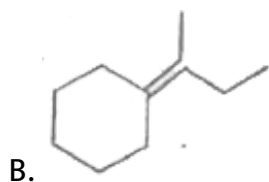
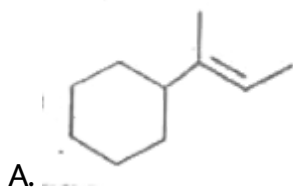
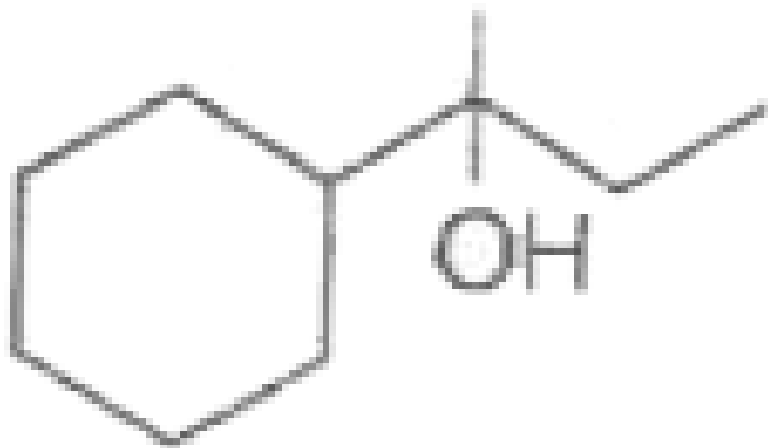
**Answer: B**



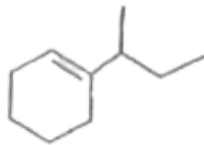
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6. Which of the following is not the product of dehydration of



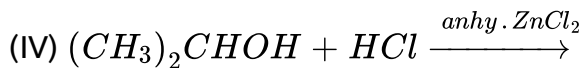
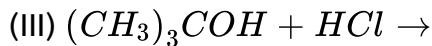
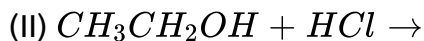
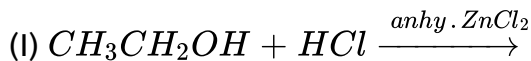
D.



**Answer: D**

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7. Which of the following reaction(s) can be used for the preparation of alkyl halides?



A. (IV) only

B. (III) and (IV) only

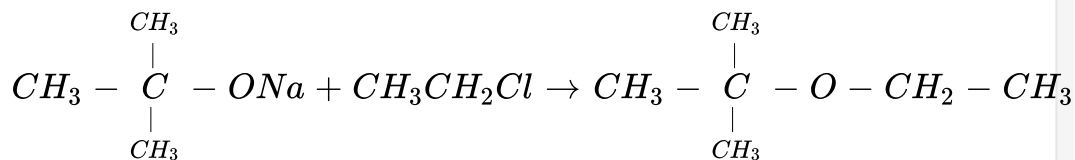
C. (I), (III) and (IV) only

D. (I), and (II) only

**Answer: C**

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8. The reaction



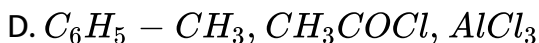
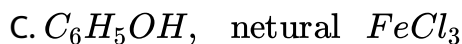
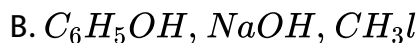
is called

- A. Gatterman -Koch reaction
- B. Williamson - synthesis
- C. Williamson continuous etherification process
- D. Etard reaction

**Answer: B**

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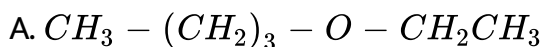
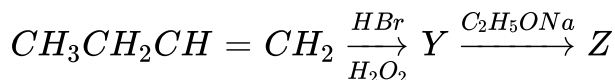
9. Among the following sets of reactants which one produces anisole?

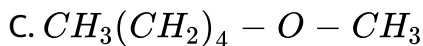
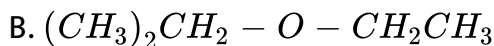


Answer: B

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10. Identify Z in the sequence of reactions :

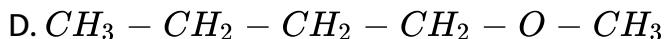
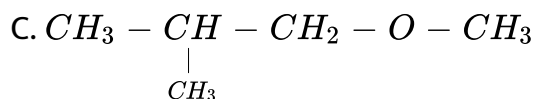
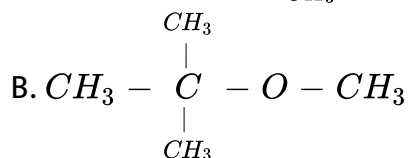
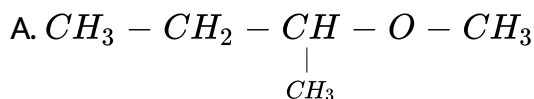




Answer: A

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11. Among the following ethers, which one will produce methyl alcohol on treatment with hot concentrated  $HI$ ?

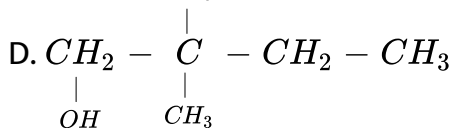
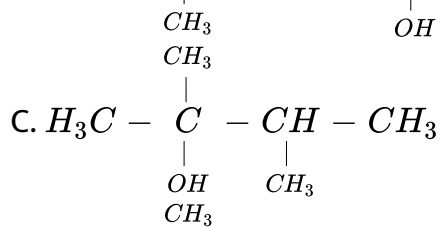
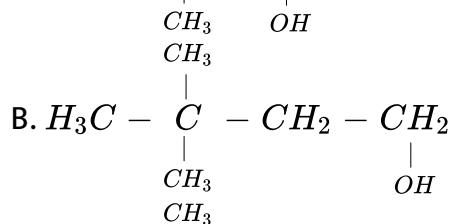
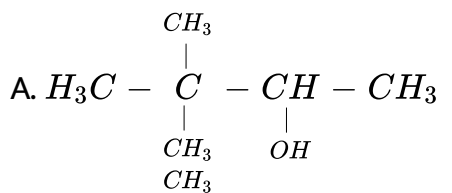
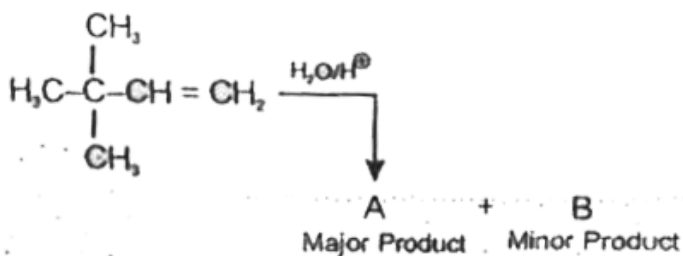


Answer: B

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12. In the following reaction

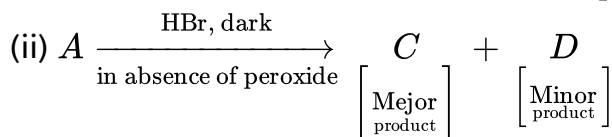
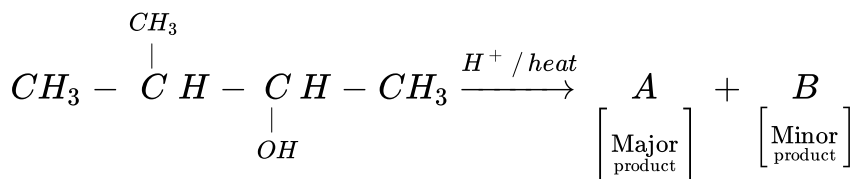
The major product is



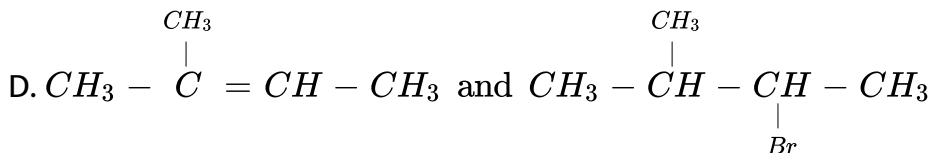
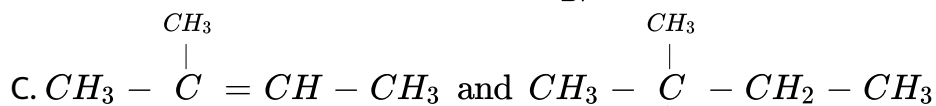
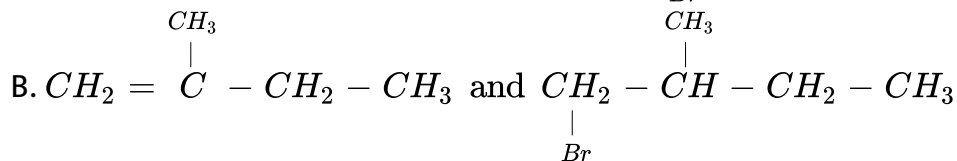
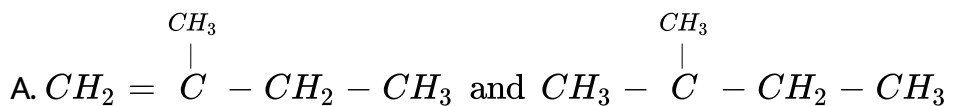
Answer: C

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13. In the following reactions . ltb rgt



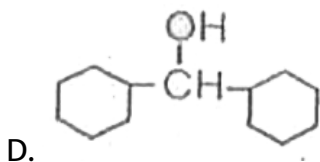
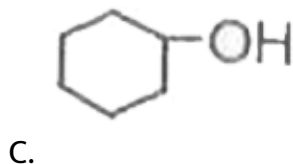
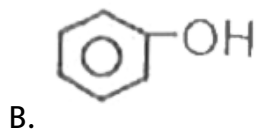
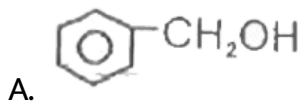
The major products A and C are repectively



Answer: C

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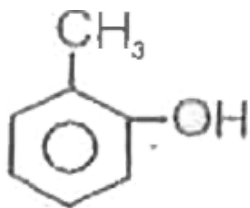
14. Which one of the following compounds has the most acidic nature ?



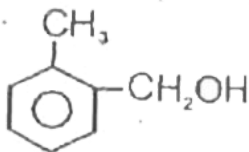
Answer: B



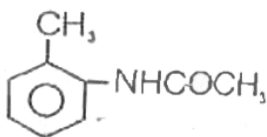
15. Which one of the following is most reactive towards electrophilic reagent?



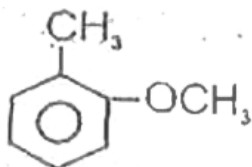
A.



B.



C.



D.

Answer: A

16. Among the following four compounds

(a) Phenol

(b) methyl phenol

(c) metanitrophenol

(d) paranitrophenol

the acidity order is –

A.  $d > c > a > b$

B.  $c > d > a > b$

C.  $a > d > c > b$

D.  $b > a > c > d$

Answer: A

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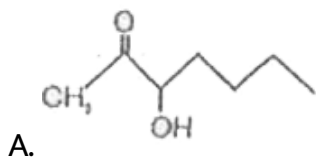
17. When glycerol is treated with excess of HI, it produces -

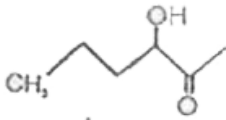
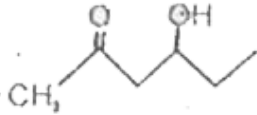
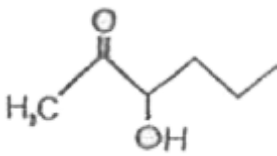
- A. 2 - iodopropane
- B. Allyl iodide
- C. Propene
- D. Glycerol triiodide

**Answer: A**

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18. Which one of the following compounds will be most readily dehydrated?

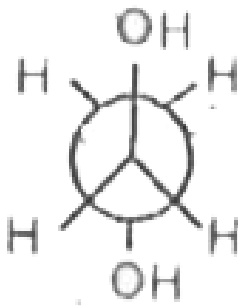




**Answer: C**

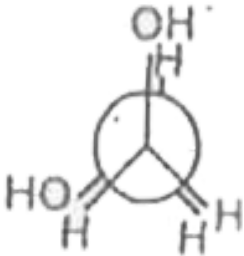
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19. Which of the following conformers for ethylene glycol is most stable ?

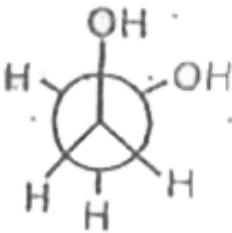


A.

B. 



C.



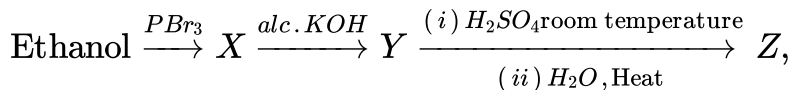
D.

**Answer: D**

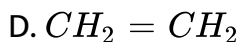
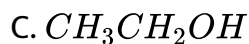
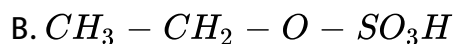


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20. Consider the following reaction



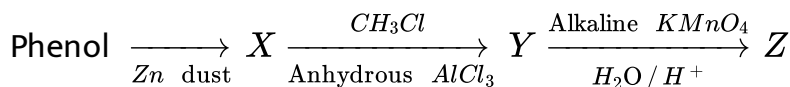
The product Z is :



Answer: C

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21. Consider the following reaction :



The product Z is

A. Benzaldehyde

B. Benzoic acid

C. Benzene

D. Toluene

**Answer: B**

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22.  $H_2COH \cdot CH_2OH$  on heating with periodic acid gives :

A.  $2HCOOH$

B.  $\begin{array}{c} CHO \\ | \\ CHO \end{array}$

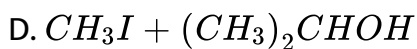
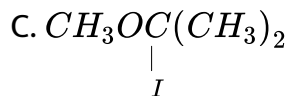
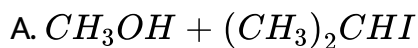
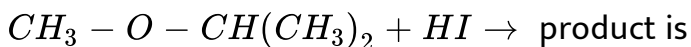
C.  $\begin{array}{c} H \\ \diagup \\ 2 \quad C = O \\ \diagdown \\ H \end{array}$

D.  $2CO_2$

Answer: C

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23. The major organic product in the reaction



Answer: D

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24. Ethylene oxide when treated with Grignard reagent yields

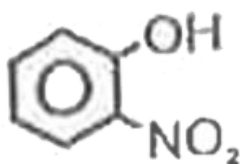
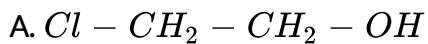


- A. Secondary alcohol
- B. Tertiary alcohol
- C. Cyclopropyl alcohol
- D. Primary alcohol

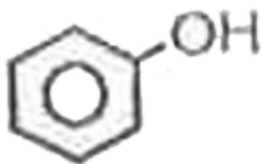
**Answer: D**

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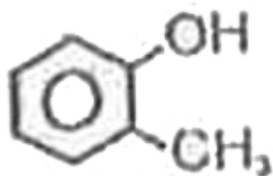
**25.** Which one of the following compounds is most acidic



B.



C.



D.

**Answer: B**

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## Assignment Section C Questions Asked Prior To Medical Ent Exams 2005

1. When 3, 3-dimethyl 2-butanol is heated with  $H_2SO_4$ , the major product obtained is

A. 2,3-dimethyl 2-butene

B. cis and trans isomers of 2,3-dimethyl 2-butene

C. 2,3-dimethyl 1-butene

D. 3,3-dimethyl 1-butene

**Answer: A**

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2. The correct order of reactivity of hydrogen halides with ethyl alcohol is

A.  $HCl > HBr > HI > HF$

B.  $HI > HBr > HCl > HF$

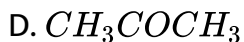
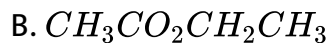
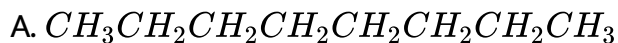
C.  $HF > HCl > HBr > HI$

D.  $HF > HBr > HI > HCl$

Answer: B

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3. More acidic than ethanol is



Answer: C

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4. Which reagent converts propene to 1-propanol:

A.  $H_2O$ ,  $H_2SO_4$

B.  $B_2H_6$ ,  $H_2O_2$ ,  $OH^-$

C.  $Hg(Oac)_2$ ,  $NaBH_4 / H_2O$

D. Aq. KOH

**Answer: B**

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5. n-propyl alcohol and isopropyl alcohol can be chemically distinguished by which reagent : -

A.  $PCl_5$

B. Reduction

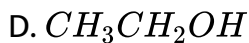
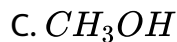
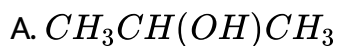
C. Oxidation with potassium dichromate

D. Ozonolysis

Answer: C

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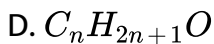
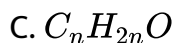
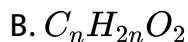
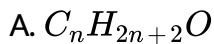
6. Which of the following will not form a yellow precipitate on heating with an alkaline solution of iodine :-



Answer: C

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7. The general molecular formula, which represents the homologous series of alkanols is



**Answer: A**



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8. On heating glycerol with conc.  $H_2SO_4$  a compound is obtained which has a bad odour. The compound is :

A. Acrolein

B. Formic acid

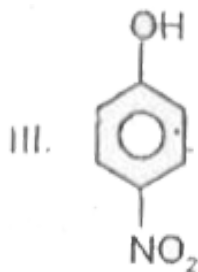
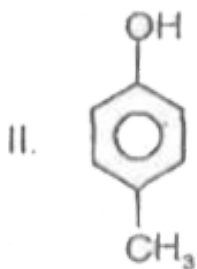
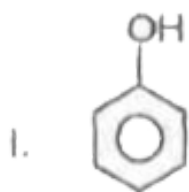
C. Allyl alcohol

D. Glycerol sulphate

**Answer: A**

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9. The correct acidic order of the following is



A.  $I > II > III$

B.  $III > I > II$

C.  $II > III > I$

D.  $I > III > II$



**Answer: B**

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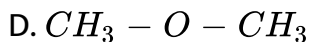
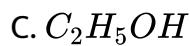
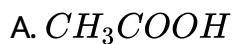
10. When phenol is treated with  $CHCl_3$  and  $NaOH$ , the product formed is

- A. Benzaldehyde
- B. Salicylaldehyde
- C. Salicylic acid
- D. Benzoic acid

**Answer: B**

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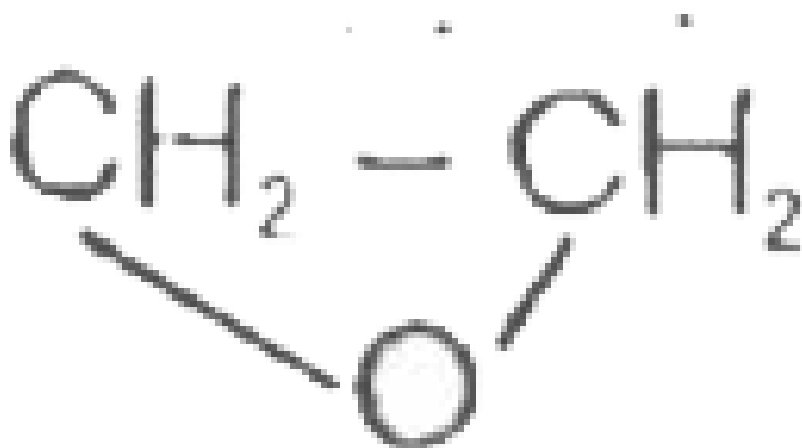
11. The compound which does not react with sodium is

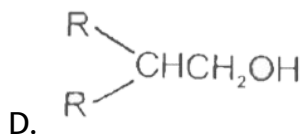
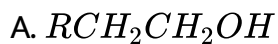


Answer: D

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12. Reaction of with  $RMgX$  leads to formation of

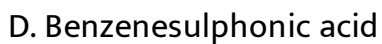




**Answer: A**

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**13.** Which of the following will not be soluble in sodium hydrogen carbonate?



Answer: C

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### Assignment Section D Assertion Reason Type Question

1. Assertion : Boiling point of p-nitrophenol is greater than that of o-nitrophenol.

Reason : There is intramolecular hydrogen bonding in p-nitrophenol and intermolecular hydrogen bonding in o-nitrophenol.

- A. If both Assertion & Reason are true and the reason is the correct explanation of the assertion, then mark (1)
- B. If both Assertion & Reason are true but the reason is not the correct explanation of the assertion, then mark (2)

C. If Assertion is true statement but Reason is false , then mark

(3)

D. If both Assertion and Reason are false statements , then mark

(4)

**Answer: D**

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2. A : When  $C_2H_5 - O - CH_3$  is reacted with one mole of HI then  $C_2H_5OH$  &  $CH_3I$  is formed .

R : It is  $S_N1$  reaction

A. If both Assertion & Reason are true and the reason is the correct explanation of the assertion, then mark (1)

B. If both Assertion & Reason are true but the reason is not the correct explanation of the assertion, then mark (2)

C. If Assertion is true statement but Reason is false , then mark

(3)

D. If both Assertion and Reason are false statements , then mark

(4)

**Answer: C**



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3. A : When 3,3-dimethyl butan - 2 - ol is heated in presence of concentrated  $H_2SO_4$  then 2, 3-dimethyl but -2-ene is formed as major product .

R : In this reaction , carbocation is formed as an intermediate

A. Both Assertion & Reason are true and the reason is the correct explanation of the assertion

B. Both Assertion & Reason are true but the reason is not the correct explanation of the assertion

C. Assertion is true statement but Reason is false

D. Both Assertion and Reason are false statements

**Answer: B**



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4. A : In esterification reaction ,  $\text{HCOOH}$  is the most reactive acid among carboxylic acid .

R : Alcohol acts as nucleophile

A. If both Assertion & Reason are true and the reason is the correct explanation of the assertion, then mark (1)

B. If both Assertion & Reason are true but the reason is not the correct explanation of the assertion, then mark (2)

C. If Assertion is true statement but Reason is false , then mark

(3)

D. If both Assertion and Reason are false statements , then mark

(4)

**Answer: B**

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5. A : Ethers can't be distilled upto dryness due to fear of explosion .

R : Due to the formation of peroxide , it is explosive

A. If both Assertion & Reason are true and the reason is the correct explanation of the assertion, then mark (1)

B. If both Assertion & Reason are true but the reason is not the correct explanation of the assertion, then mark (2)



C. If Assertion is true statement but Reason is false , then mark

(3)

D. If both Assertion and Reason are false statements , then mark

(4)

**Answer: C**



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**6. Phenol is less acidic than..... .**

A. If both Assertion & Reason are true and the reason is the correct explanation of the assertion, then mark (1)

B. If both Assertion & Reason are true but the reason is not the correct explanation of the assertion, then mark (2)

C. If Assertion is true statement but Reason is false , then mark

(3)

D. If both Assertion and Reason are false statements , then mark

(4)

**Answer: A**

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7.A :  $CH_3 - \overset{\overset{O}{||}}{C} - COOH$  gives haloform reaction .

R : It is more acidic than acetic acid .

A. If both Assertion & Reason are true and the reason is the correct explanation of the assertion, then mark (1)

B. If both Assertion & Reason are true but the reason is not the correct explanation of the assertion, then mark (2)

C. If Assertion is true statement but Reason is false , then mark

(3)

D. If both Assertion and Reason are false statements , then mark

(4)

**Answer: B**

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**8. A :** Diphenyl ether is prepared by Williamson synthesis .

**R :** This reaction generally proceed by  $S_N1$  mechanism .

A. If both Assertion & Reason are true and the reason is the correct explanation of the assertion, then mark (1)

B. If both Assertion & Reason are true but the reason is not the correct explanation of the assertion, then mark (2)

C. If Assertion is true statement but Reason is false , then mark

(3)

D. If both Assertion and Reason are false statements , then mark

(4)

**Answer: D**

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**9. A :** Grignard's reagent is prepared in the presence of ether .

**R :** Grignard's reagent is soluble and stable in ether.

A. If both Assertion & Reason are true and the reason is the correct explanation of the assertion, then mark (1)

B. If both Assertion & Reason are true but the reason is not the correct explanation of the assertion, then mark (2)

C. If Assertion is true statement but Reason is false , then mark

(3)

D. If both Assertion and Reason are false statements , then mark

(4)

**Answer: A**



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10. A :  $CH_3 - \overset{\overset{CH_3}{|}}{C} - CH = CH_2$  on hydroboration oxidation gives

$CH_3 - \overset{\overset{CH_3}{|}}{C} - \underset{\underset{OH}{|}}{CH} - CH_3$  as major product .

R : It involves the formation of carbocation so undergoes rearrangement .

- A. If both Assertion & Reason are true and the reason is the correct explanation of the assertion, then mark (1)
- B. If both Assertion & Reason are true but the reason is not the correct explanation of the assertion, then mark (2)
- C. If Assertion is true statement but Reason is false , then mark (3)
- D. If both Assertion and Reason are false statements , then mark (4)

**Answer: D**



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**11. A :** Two moles of Grignard reagent is consumed in the formation of tertiary alcohol from ester following by hydrolysis .

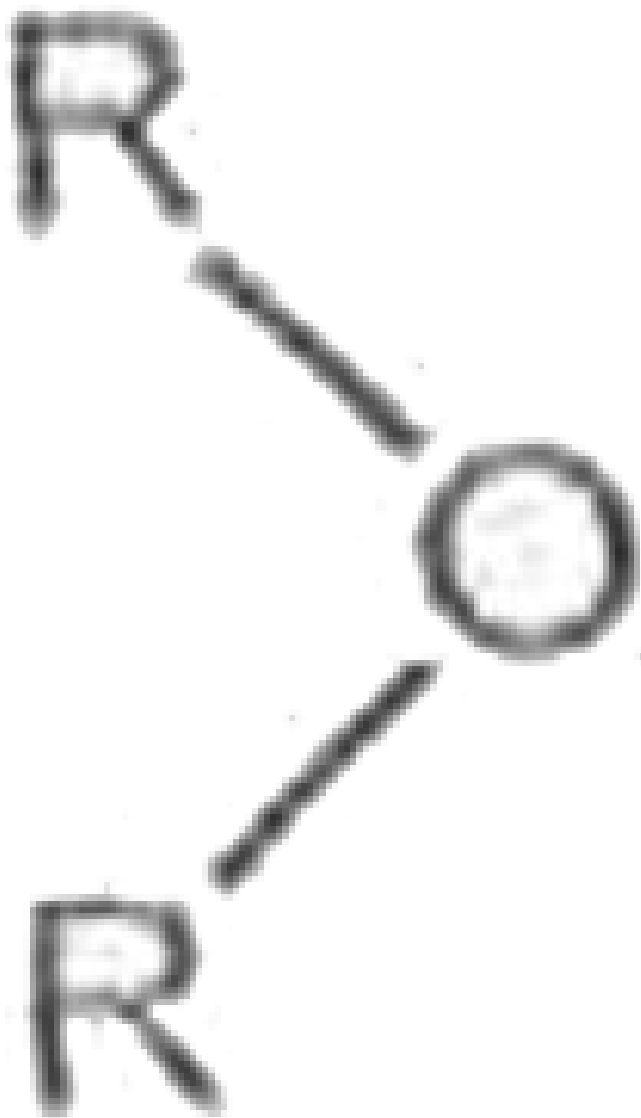
R : One mole of Grignard reagent convert ester into Ketone and second mole of Grignard reagent adds to Ketone .

- A. Both Assertion & Reason are true and the reason is the correct explanation of the assertion
- B. Both Assertion & Reason are true but the reason is not the correct explanation of the assertion
- C. Assertion is true statement but Reason is false
- D. Both Assertion and Reason are false statements

**Answer: A**



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12.

A : bond angle in ether is slightly greater than normal tetrahedral



angle ( $109.5^\circ$ )

R : the hybridisation of oxygen atom in ether is  $sp^3$

A. If both Assertion & Reason are true and the reason is the correct explanation of the assertion, then mark (1)

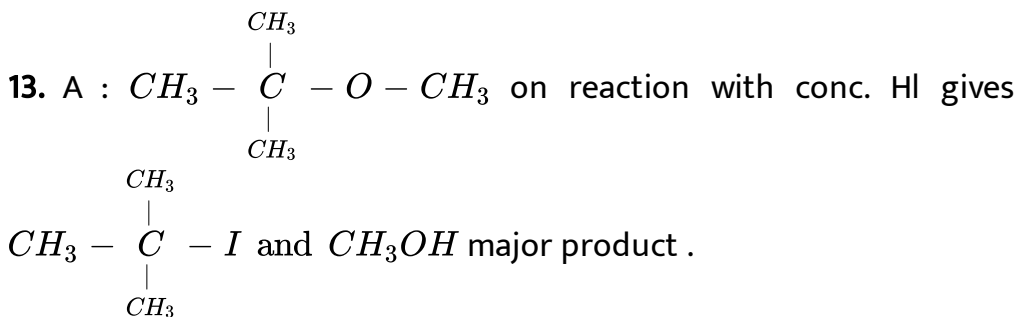
B. If both Assertion & Reason are true but the reason is not the correct explanation of the assertion, then mark (2)

C. If Assertion is true statement but Reason is false , then mark (3)

D. If both Assertion and Reason are false statements , then mark (4)

**Answer: B**

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R : This reaction proceed by  $S_N1$  mechanism .

- A. If both Assertion & Reason are true and the reason is the correct explanation of the assertion, then mark (1)
- B. If both Assertion & Reason are true but the reason is not the correct explanation of the assertion, then mark (2)
- C. If Assertion is true statement but Reason is false , then mark (3)
- D. If both Assertion and Reason are false statements , then mark (4)

**Answer: A**

**14.** A : Ortho - cresol is weaker acidic than meta-cresol .

R : It is due to ortho effect .

A. If both Assertion & Reason are true and the reason is the correct explanation of the assertion, then mark (1)

B. If both Assertion & Reason are true but the reason is not the correct explanation of the assertion, then mark (2)

C. If Assertion is true statement but Reason is false , then mark (3)

D. If both Assertion and Reason are false statements , then mark (4)

**Answer: C**

15. A : Among all ortho halophenol , fluorophenol is least acidic .

R : Ortho - fluorophenol forms intramolecular H - bond .

A. If both Assertion & Reason are true and the reason is the correct explanation of the assertion, then mark (1)

B. If both Assertion & Reason are true but the reason is not the correct explanation of the assertion, then mark (2)

C. If Assertion is true statement but Reason is false , then mark (3)

D. If both Assertion and Reason are false statements , then mark (4)

**Answer: A**

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**16. A :** In esterification reaction alcohol act as nucleophile .

**R :** In this reaction O - H bond of alcohol is broken .

A. If both Assertion & Reason are true and the reason is the correct explanation of the assertion, then mark (1)

B. If both Assertion & Reason are true but the reason is not the correct explanation of the assertion, then mark (2)

C. If Assertion is true statement but Reason is false , then mark (3)

D. If both Assertion and Reason are false statements , then mark (4)

**Answer: A**



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**17. A :** Phenol is manufactured by Dow 's process.

**R :** It involves the formation of benzyne intermediate .

A. If both Assertion & Reason are true and the reason is the correct explanation of the assertion, then mark (1)

B. If both Assertion & Reason are true but the reason is not the correct explanation of the assertion, then mark (2)

C. If Assertion is true statement but Reason is false , then mark (3)

D. If both Assertion and Reason are false statements , then mark (4)

**Answer: B**



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18. A : Primary alcohol is prepared by the reaction of primary amine with  $HNO_2$  .

R : Dimethyl amine is a primary amine but does not form methyl alcohol with  $HNO_2$  .

A. If both Assertion & Reason are true and the reason is the correct explanation of the assertion, then mark (1)

B. If both Assertion & Reason are true but the reason is not the correct explanation of the assertion, then mark (2)

C. If Assertion is true statement but Reason is false , then mark (3)

D. If both Assertion and Reason are false statements , then mark (4)

**Answer: C**



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19. A : The reactivity order of alcohols is  $1^\circ > 2^\circ > 3^\circ$  for the reaction in which O-H bond is broken .

R : The reactivity order of alcohol is  $3^\circ > 2^\circ > 1^\circ$  for the reaction in which C - O bond is broken .

A. If both Assertion & Reason are true and the reason is the correct explanation of the assertion, then mark (1)

B. If both Assertion & Reason are true but the reason is not the correct explanation of the assertion, then mark (2)

C. If Assertion is true statement but Reason is false , then mark (3)

D. If both Assertion and Reason are false statements , then mark (4)

**Answer: B**



20. A : The dehydration of ethyl alcohol in presence of  $Al_2O_3$  at 633 K gives ethene.

R : The reaction proceed through the formation of carbocation intermediate.

A. If both Assertion & Reason are true and the reason is the correct explanation of the assertion, then mark (1)

B. If both Assertion & Reason are true but the reason is not the correct explanation of the assertion, then mark (2)

C. If Assertion is true statement but Reason is false , then mark (3)

D. If both Assertion and Reason are false statements , then mark (4)

**Answer: C**



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