



# PHYSICS

## AAKASH INSTITUTE ENGLISH

### Mock Test 38: PHYSICS

#### Example

1. The work function of material (in eV), whose threshold frequency is  $6 \times 10^{14}$  Hz would be

A. 5.48 eV

B. 3.48 eV

C. 2.48 eV

D. 8.48 eV

**Answer: C**



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2. The energy ( $E$ ) and momentum ( $p$ ) of photon are respectively, (symbols have their usual meaning)

A.  $E = hc, p = \frac{v}{c}$

B.  $E = hv, p = h\frac{v}{c}$

C.  $E = \frac{c}{v}, p = \frac{h}{c}$

D.

**Answer: B**



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**3.** The de-Broglie wavelength of an electron accelerated by and electric potential of V volts is given by:

$$\text{A. } \lambda = \frac{12.27}{\sqrt{\quad}} vcm$$

$$\text{B. } \lambda = \frac{12.27}{\sqrt{\quad}} vmm$$

$$\text{C. } \lambda = \frac{12.27}{\sqrt{\quad}} vcm$$

$$\text{D. } \lambda = \frac{12.27}{\sqrt{\quad}} vA$$

**Answer: D**



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**4.** On increasing the temperature the work function of metal will

A. Increase

B. Decrease

C. Remains same

D. First increase and then decrease

**Answer: B**



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5. Find the ratio of de Broglie wavelength of a proton and an  $\alpha$ -particle which have been

accelerated through same potential difference.

A. 1 : 2

B. 1 : 1

C.  $1 : 2\sqrt{2}$

D. 1 : 4

**Answer: C**



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6. In the Davisson-Germer experiment, if the scattering angle is  $60^\circ$ , then the glancing angle is equal to

A.  $30^\circ$

B.  $45^\circ$

C.  $60^\circ$

D.  $90^\circ$

**Answer: C**



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7. According to Einstein's photoelectric equation (symbols have their usual meaning)

A.  $h\nu - \phi = (KE)_{\max}$

B.  $h\nu > (KE)_{\max}$

C.  $\phi = h\nu + \frac{(KE)_{\max}}{2}$

D.  $\phi = (KE)_{\max} - \frac{h\nu}{2}$

**Answer: A**



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8. In photoelectric effect experiment, the slope of the graph of the stopping potential versus frequency gives the value of :

A.  $h$

B.  $\frac{h}{e}$

C.  $he$

D.  $\frac{h^2}{e}$

**Answer: B**



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9. Explain photoelectric effect Explain the effect of increase of intensity of incident radiations on photoelectrons emitted by a photo-tube.

A. increases

B. decreases

C. remains constant

D. first decrease and then increase

constant

**Answer: A**



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10. Find the maximum velocity of an electron emitted by light of wavelength  $\lambda$  incident on the surface of a metal of work function  $\phi$ .

$$\text{A. } v = \left[ 2m_e \frac{\lambda}{hc - \lambda\phi} \right]^{1/2}$$

$$\text{B. } v = \left[ 2h \frac{c}{m_e \lambda - \phi} \right]^{1/2}$$

$$\text{C. } v = \left[ m_e \frac{\lambda}{hc + \lambda\phi} \right]^{1/2}$$

$$\text{D. } v = \left[ 2 \frac{hc - \lambda\phi}{m_e \lambda} \right]^{1/2}$$

**Answer: D**



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11. The De-Broglie wave is

A. a mechanical wave

B. An electromagnetic wave

C. neither mechanical wave nor  
electromagnetic wave

D. both (1) & (2)

**Answer: C**



12. The work function of two metals metal A and metal B are 6.5 eV and 4.5 eV respectively. If the threshold wavelength of metal A is 2500 A, the threshold wavelength on metal B will be approximately equal to

A. 3611A

B. 3500 A

C. 6332 A

D. 4321 A

**Answer: A**



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**13.** The wave like character is significant and measurable for

A. macroscopic objects only

B. sub-atomic particles

C. both macroscopic objects and sub-atomic particles

D. heavy and more energetic particles

**Answer: B**



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**14.** If the material particle is moving with velocity  $v$  and the velocity of light is  $c$ , then mass of particle is taken as ( $m_0$  is rest mass)

A.  $m = \frac{m_0}{c}$

B.  $m = \frac{m_0}{\sqrt{\left(1 - \frac{v^2}{c^2}\right)}}$

$$C. m = m_0 \sqrt{\left(1 + \frac{v^2}{c^2}\right)}$$

$$D. m = \frac{m_0}{\sqrt{\left(1 + \frac{v^2}{c^2}\right)}}$$

**Answer: B**



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**15.** The material having minimum work function is

A. copper

B. silver



C. gold

D. cesium

**Answer: D**



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**16.** The ionisation energy of electron (E) is second orbit of hydrogen atom is

A. -13.6 eV

B. -3.4 eV

C. +3.4 eV

D. zero

**Answer: C**



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**17.** The minimum energy needed to separate the constituent of nucleons system is known as  
as

A. ionisation energy

B. binding energy

C. excitation energy

D. kinetic energy

**Answer: B**



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**18.** calculate the longest wavelength in Balmer series and the series limit . (Given

$$R = 1.097 \times 10^7 m^{-1} )$$

A. 6000 A

B. 6563 A

C. 7200 A

D. 7383 A

**Answer: B**



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**19.** In alpha-particle scattering experiment, the expression of distance of closest approach ( $r_0$ ) is (symbols have their usual meanings)

A.  $\frac{Ze^2 \sin^2\left(\frac{\theta}{2}\right)}{4\pi\epsilon_0 E}$

B.  $\frac{Ze^2 \cot\left(\frac{\theta}{2}\right)}{4\pi\epsilon_0 E}$

C.  $\frac{Ze^2 \cos\left(\frac{\theta}{2}\right)}{4\pi\epsilon_0 E}$

D.  $\frac{Z^2 e^2 E}{\sin^4\left(\frac{\theta}{2}\right) 4\pi\epsilon_0}$

**Answer: B**



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20. The total energy (E) of the electron in an orbit of radius  $r$  in hydrogen atom is

A.  $\frac{e^2}{8\pi\epsilon_0 r}$

B.  $-\frac{e^2}{4\pi\epsilon_0 r}$

C.  $-\frac{e^2}{8\pi\epsilon_0 r}$

D. zero

**Answer: C**



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21. According to Bohr's theory the possible value of angular momentum of an electron orbiting in hydrogen atom is

A.  $\frac{4h}{3\pi}$

B.  $\frac{h}{\pi}$

C.  $\frac{4h}{5\pi}$

D. both (1) & (3)

**Answer: B**



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22. For hydrogen atom, energy of  $n$ th level is given by

A.  $E_n = - \left( \frac{13.6}{n} \right) eV$

B.  $E_n = + \left( \frac{13.6}{n} \right) eV$

C.  $E_n = \left( \frac{13.6}{n^2} \right) eV$

D.  $E_n = - \left( \frac{13.6}{n^2} \right) eV$

**Answer: D**



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23. Calculate the value of Rydberg's constant.

A.  $1.097 \times 10^7 m^{-1}$

B.  $8.85 \times 10^{-12} m^{-1}$

C.  $1.6 \times 10^{-19} m^{-1}$

D.  $6.63 \times 10^{-34} m^{-1}$

**Answer: A**



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24. The ratio of the velocity of electron in 3rd and 5th orbit of hydrogen atom is

A. 5 : 3

B. 1 : 3

C. 5 : 1

D. 1 : 1

**Answer: A**



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25. According to Rutherford, which force was responsible for centripetal acceleration of an electron revolving around the nucleus?

- A. gravitational force
- B. weak nuclear force
- C. strong nuclear force
- D. electrostatic force

**Answer: D**



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26. In x-ray spectra, the  $K_{\gamma}$  lines in k series corresponds to

- A. The vacancy created in K-shell being filled by an electron from L-shell
- B. The vacancy created in L-shell being filled by an electron from M-shell
- C. The vacancy created in L-shell being filled by an electron from K-shell
- D. The vacancy created in K-shell being filled by an electron from N-shell

**Answer: D**



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**27.** Total energy of an electron in an atom is negative. What does it signify?

A. electron is bound to nucleus

B. the force between electron and nucleus  
is attractive or repulsive

C. the centripetal force provided for revolution is provided by electrostatic force of attraction

D. both (1) & (2)

**Answer: A**



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**28.** In alpha-particle scattering experiment, the expression of distance of closest approach ( $r_0$ ) is (symbols have their usual meanings)

A.  $\frac{1}{4\pi\epsilon_0 E} \times \frac{Ze^2}{E}$

B.  $\frac{1}{4\pi\epsilon_0} \times \frac{2Ze^2}{E}$

C.  $\frac{1}{4\pi\epsilon_0} \times (Z^2 e^2)$

D.  $\frac{1}{4\pi\epsilon_0} \times \frac{Z^2 e^4}{E}$

**Answer: B**



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**29.** In Lyman series the wavelength  $\lambda$  of emitted radiation is given by (R is rydberg constant)

A.  $\frac{1}{\lambda} = R \left( \frac{1}{2^2} - \frac{1}{n^2} \right), n = 3, 4, 5, \dots$

B.  $\frac{1}{\lambda} = R \left( \frac{1}{3^2} - \frac{1}{n^2} \right), n = 4, 5, 6, \dots$

C.  $\frac{1}{\lambda} = R \left( \frac{1}{1^2} - \frac{1}{n^2} \right), n = 2, 3, \dots$

D.  $\frac{1}{\lambda} = R \left( \frac{1}{4^2} - \frac{1}{n^2} \right), n = 5, 6, 7, \dots$

**Answer: C**



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