



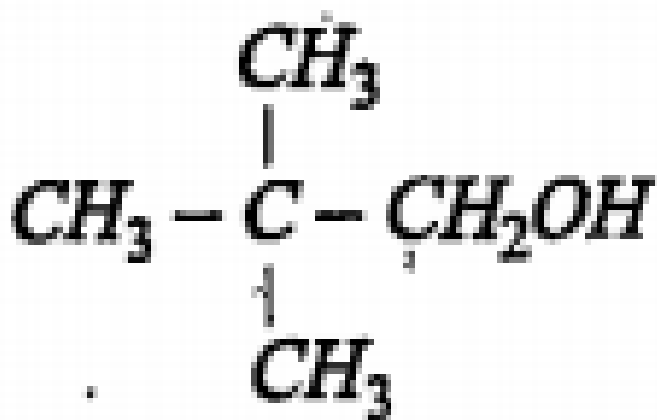
CHEMISTRY

BOOKS - A N EXCEL PUBLICATION

ALCOHOLS, PHENOLS AND ETHERS

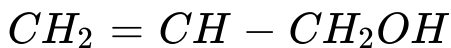
Question Bank

1. Classify the following as 1°, 2° and 3° alcohols



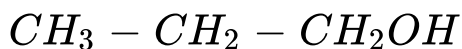
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2. Classify the following as 1°, 2° and 3° alcohols



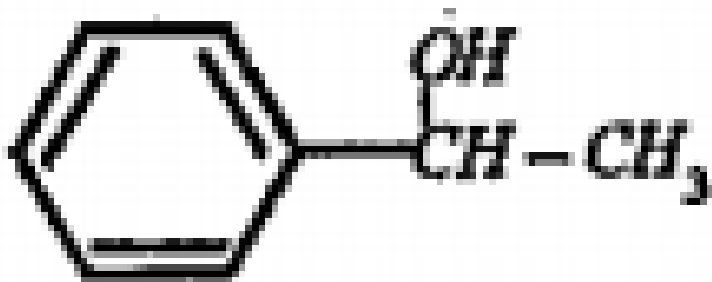
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3. Classify the following as 1° , 2° and 3° alcohols



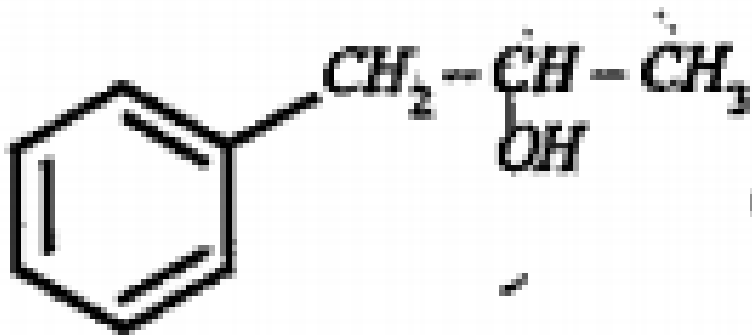
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4. Classify the following as 1° , 2° and 3° alcohols



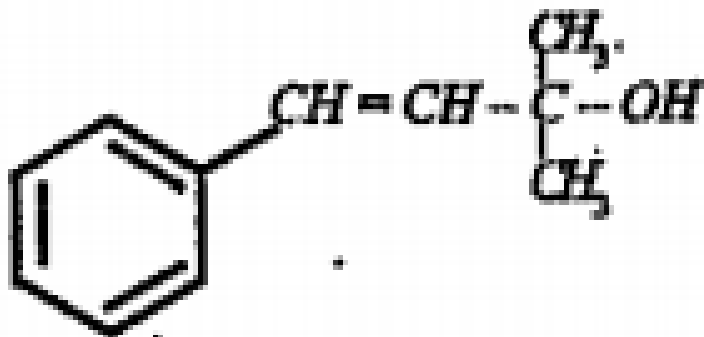
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5. Classify the following as 1° , 2° and 3° alcohols



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6. Classify the following as 1° , 2° and 3° alcohols

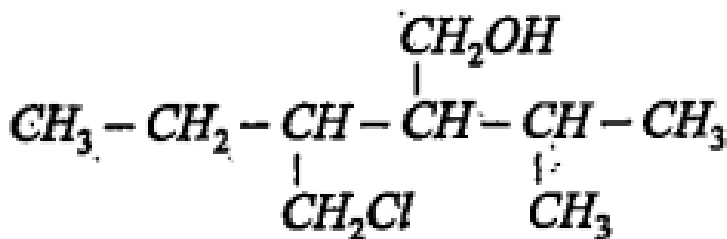


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7. Identify allylic alcohols in the above examples

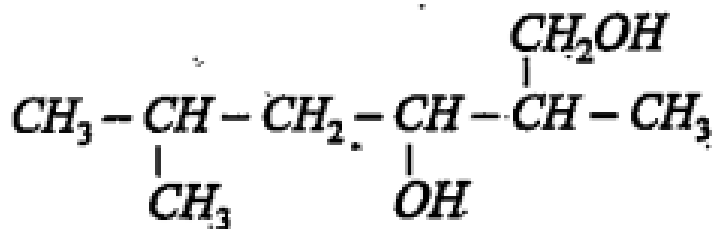
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8. Name the following compounds according to IUPAC system



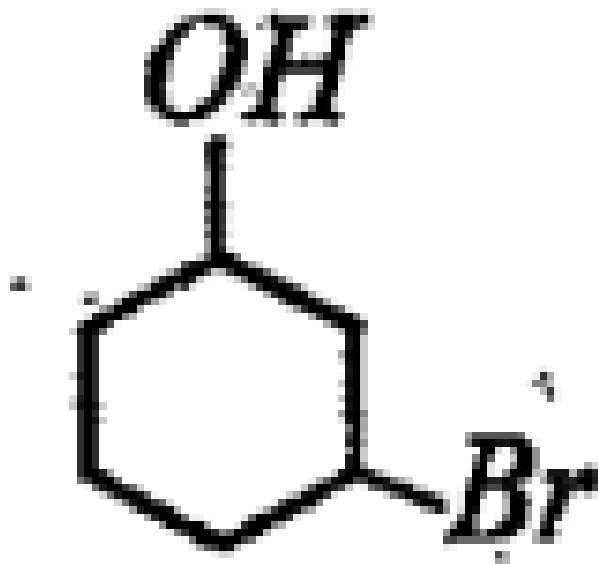
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9. Name the following compounds according to IUPAC system



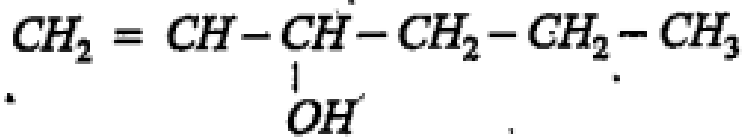
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10. Name the following compounds according to IUPAC system



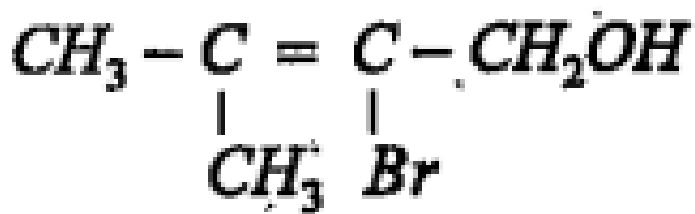
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11. Name the following compounds according to IUPAC system



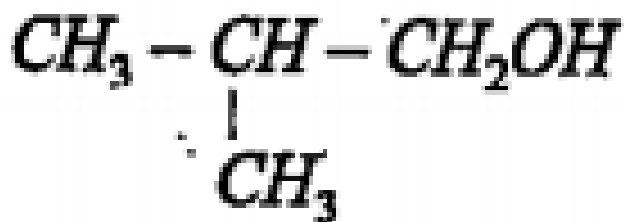
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12. Name the following compounds according to IUPAC system



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13. Show how the following alcohols are prepared by the reaction of a suitable Grignard reagent on methanal.

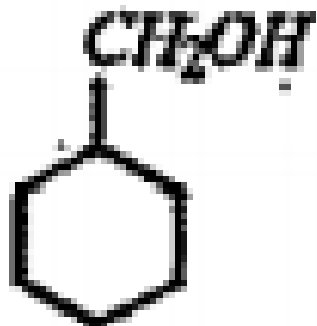


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14. Show how the following alcohols are prepared by the reaction of a suitable Grignard reagent on

methanal.

+2 Concise Chemistry



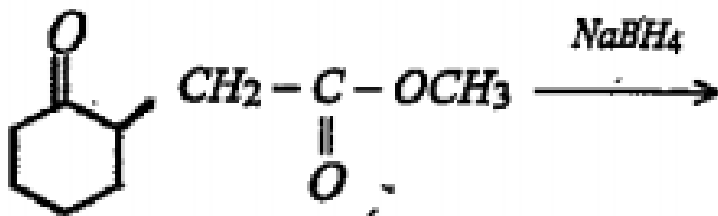
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15. Write the structures of the products of the following reactions



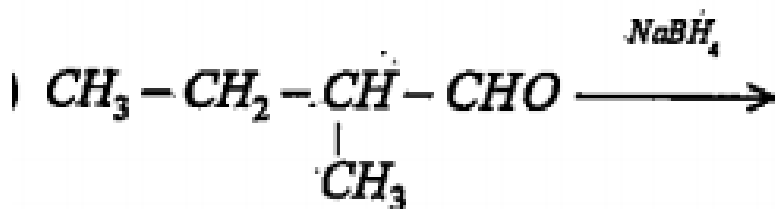
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16. Write the structures of the products of the following reactions



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17. Write the structures of the products of the following reactions



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18. Give the structures of the products you would expect when each of the following alcohol reacts with HBr
a. butan-1-ol
b. 2-methylbutan-2-ol

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19. Give the structures of the products you would expect when each of the following alcohol reacts with HBr
a. butan-1-ol b. 2-methylbutan-2-ol

 [Watch Video Solution](#)

20. Give the structures of the products you would expect when each of the following alcohol reacts with SOCl_2
a. butan-1-ol b. 2-methylbutan-2-ol

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21. Predict the major products of acid catalysed dehydration of
1-Methylcyclohexanol

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22. Predict the major products of acid catalysed dehydration of
butan-1-ol

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23. Ortho and para nitrophenols are more acidic than phenol. Draw the resonance structures of the corresponding phenoxide ions

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24. Write the equation involved in the following reactions

Reimer-Tiemann reaction

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25. Write the equation involved in the following reactions

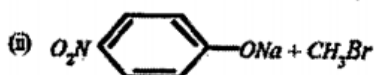
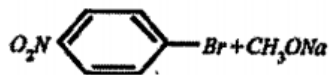
Kolbe's reaction

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26. Write the reaction of Williamson's synthesis of 2-ethoxy-3-methylpentane starting from ethanol and 3-methylpentan-2-ol

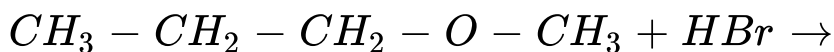
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27. Which of the following is an appropriate set of reactants for the preparation of 1-methoxy-4-nitrobenzene? Why?



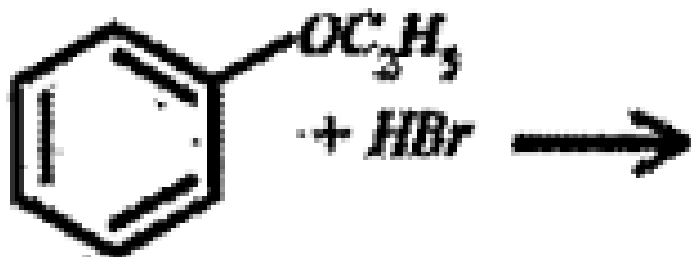
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28. Predict the products of the following reactions



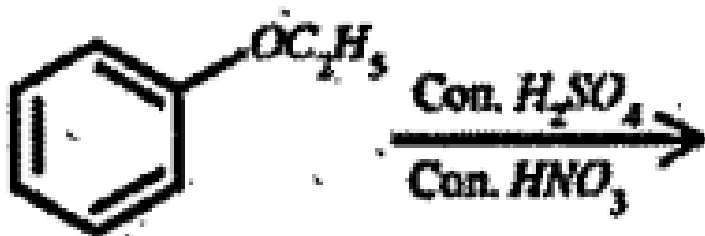
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29. Predict the products of the following reactions



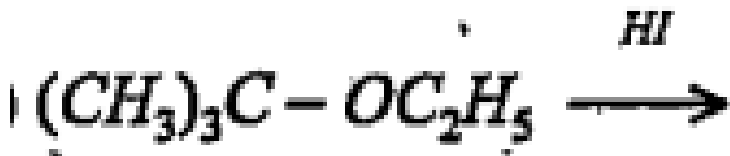
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30. Predict the products of the following reactions



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31. Predict the products of the following reactions



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32. Phenols are more acidic than alcohols.

Name the product obtained when phenol is treated with chloroform in the presence of NaOH.



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33. Phenols are more acidic than alcohols.

Name the above reaction.



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34. Phenols are more acidic than alcohols.

What is the product obtained when phenol is treated with conc. HNO_3 ?

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35. Phenols are more acidic than alcohols.

Write the structure and IUPAC name of the above product.

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36. Phenols are more acidic than alcohols.

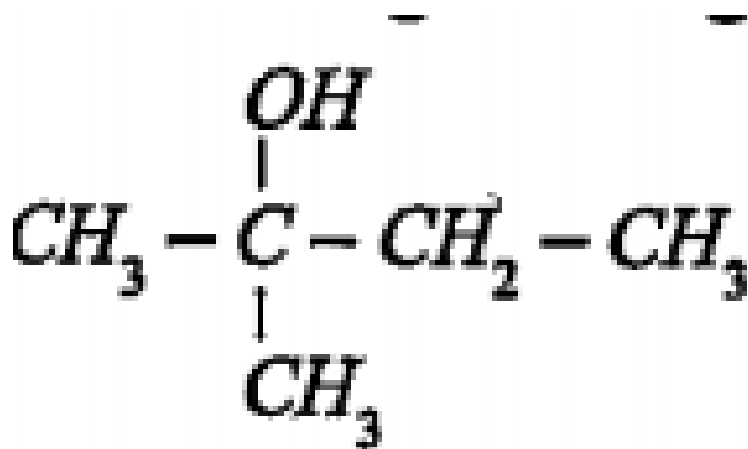
Ethanol and propane have comparable molecular masses but boiling points differ widely. Which of them has higher boiling point? Substantiate your answer.



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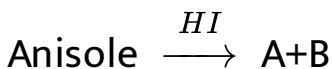
37. A compound A reacts with thionyl chloride to give compound B. B reacts with magnesium in ether medium to form a Grignard reagent which is treated with acetone and the product on hydrolysis gives Identify A and B. Write down the chemical

equations for the reactions involved.



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38. Ethers are generally non-reactive compound. One of the important reactions of ethers is the action of HI.



Identify A and B.

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39. Mixture of conc. HCl and anhydrous $ZnCl_2$ is an important reagent which helps to distinguish between 1° , 2° and 3° alcohols.

Explain how the above reagent helps to distinguish above three types of alcohols.



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40. Mixture of conc. HCl and anhydrous $ZnCl_2$ is an important reagent which helps to distinguish between 1° , 2° and 3° alcohols.

Give one example each for 1° , 2° and 3° alcohols.



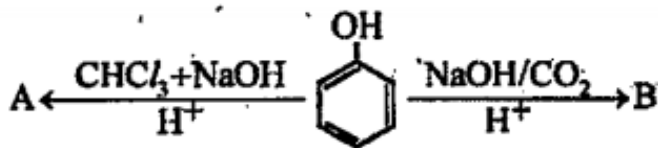
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41. Mixture of conc. HCl and anhydrous $ZnCl_2$ is an important reagent which helps to distinguish between 1° , 2° and 3° alcohols.

Explain how the above reagent helps to distinguish above three types of alcohols.

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42. Write the name or structure of the compounds A and B in the following reactions.





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43. Vapours of an alcohol 'C' on passing over heated copper produce compound 'D'. 'D' on reaction with CH_3MgCl following by hydrolysis produces 2-methyl butan-2-ol. Write the name or structure of compounds 'C' and 'D'.



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44. Methanol and ethanol are two commercially important alcohols.

Write one method of preparation of methanol and ethanol.

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45. Methanol and ethanol are two commercially important alcohols.

Name the products obtained when ethanol is treated with CrO_3 in anhydrous medium

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46. Methanol and ethanol are two commercially important alcohols.

The boiling point of ethanol is higher than that of methoxy methane. Give reason.

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47. Write the IUPAC names of all the possible isomers with molecular formula C_3H_8O

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48. Phenol is usually manufactured from cumene. Write the structure of cumene.

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49. Primary, secondary and tertiary alcohols can be distinguished by Lucas test.

What is Lucas reagent?

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50. Primary, secondary and tertiary alcohols can be distinguished by Lucas test.

Write the observations for primary, secondary and tertiary alcohols in Lucas test.

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51. How will you prepare the following compounds using Grignard reagent?

Primary alcohol

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52. How will you prepare the following compounds using Grignard reagent?

Secondary alcohol

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53. Primary, secondary and tertiary alcohols can be distinguished by Lucas test.

Write the observations for primary, secondary and tertiary alcohols in Lucas test.

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54. Write the correct pair of reactants for the preparation of t-butyl ethyl ether by Williamson synthesis.

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55. Write the name or formula of the following:

A simple ether

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56. Write the name or formula of the following:

A mixed ether

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57. Write the name or formula of the following:

A dihydric alcohol

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58. Write the name or formula of the following:

A trihydric alcohol

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59. Phenol on treatment with Br_2 in CS_2 at low temperature gives two isomeric monobromo phenols 'X' and 'Y'. But phenol on treatment with bromine water gives a white precipitate 'Z'.

Identify the products 'X','Y' and 'Z'.

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60. Alcohols are compounds with general formula R-OH

Alcohols are soluble in water. What is the reason?

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61. Alcohols are compounds with general formula R-OH

Explain a method for manufacture of Ethanol.

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62. Alcohols are compounds with general formula R-OH

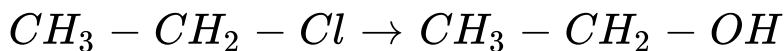
How will you convert phenol to benzene?

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63. Write a test to distinguish between phenol and alcohol

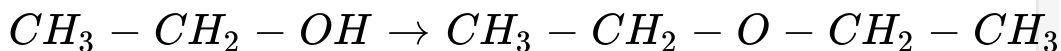
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64. Write suitable reagent used for the following conversions:



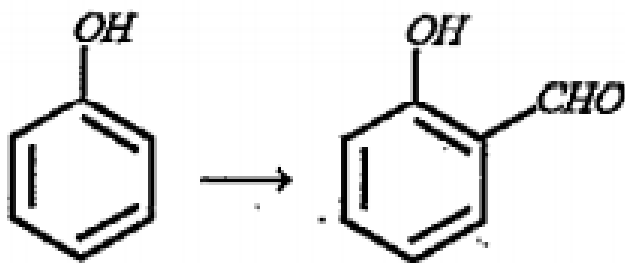
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65. Write suitable reagent used for the following conversion:



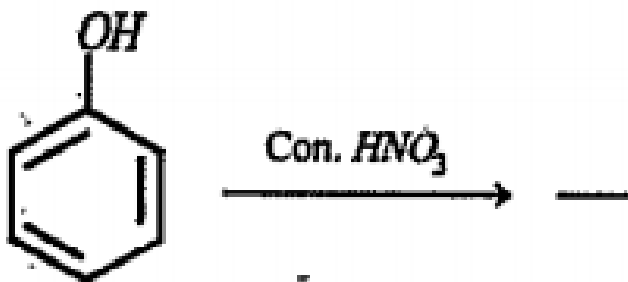
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66. Write suitable reagent or reagents used for the following conversions:



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67. Complete the following:





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68. Explain the following :

Esterification



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69. Write the correct pair of reactants for the preparation of t-butyl ethyl ether by Williamson synthesis.



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70. Phenols are more acidic than alcohols.

What is the product obtained when phenol is treated with conc. HNO_3 ?

- A. o-Nitrophenol
- B. p-Nitrophenol
- C. 2,4,6-Trinitrophenol
- D. a mixture of o-nitrophenol

Answer: C



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71. Methanol and ethanol are two commercially important alcohols. Write one method each for the preparation of methanol and ethanol.

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72. Arrange the following compounds in the order of increasing boiling point: Ethanol, propan-1-ol, Butan-1-ol, Butan-2-ol.

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73. In the lab students were asked to carry out the reaction between phenol and cons. HNO_3 . But one student, 'A' carried out the reaction between phenol and dil. HNO_3 . Do you think that the student 'A' got the same result as others. Substantiate with suitable explanations. [Also write the chemical equations wherever necessary].

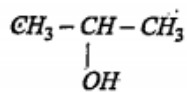


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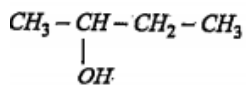
74. Identify the product. $HCHO \xrightarrow[H_2O]{CH_3MgX} \text{----}$

A. CH_3OH

B. CH_3CH_2OH



C.

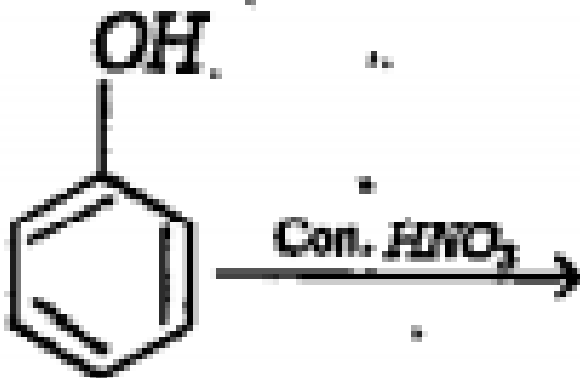


D.

Answer: C

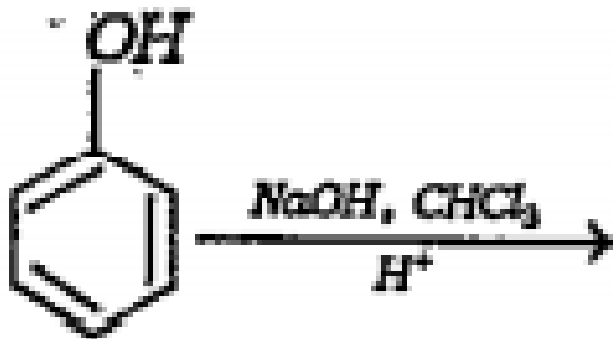
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75. Complete the following:



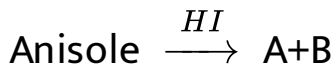
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76. Complete the following:



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77. Ethers are generally non-reactive compound. One of the important reactions of ethers is the action of HI.



Identify A and B.

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78. Grignard a method can be used to synthesise different type of alcohols

Suggest a method to prepare 2-Methyl-2-propanol using Grignard reagent. Write equation for the reaction.

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79. Grignard a method can be used to synthesise different type of alcohols

What happens when this alcohol is passed over copper heated to 573K?



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80. Alcohols can be prepared by treating alkyl halide with aqueous NaOH at room temperature and atmospheric pressure.

Can we prepare phenol from halobenzene and aqueous NaOH at the same conditions? Justify.



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81. Alcohols can be prepared by treating alkyl halide with aqueous NaOH at room temperature and atmospheric pressure.

You are given aniline, NaNO_2 and dilute HCl. Suggest a method to prepare phenol using these reagents.

Write equations for the reactions.

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82. Phenol is known as carbolic acid and it is acidic.

Why is phenol acidic?

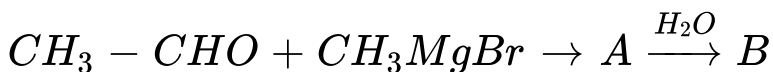
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83. Phenol is known as carboic acid and it is acidic.

Write equation for the reaction of phenol with nitrating mixture. Write the name of the product.

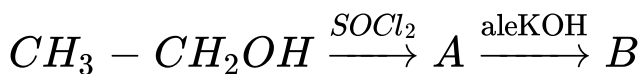
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84. Identify the organic product 'B' in each of the following sequence of reactions and complete the equations.



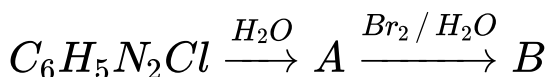
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85. Identify the organic product 'B' in each of the following sequence of reactions and complete the equations.



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86. Identify the organic product 'B' in each of the following sequence of reactions and complete the equations.



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87. In our country ethanol is mainly manufactured by suger industries the mother liquor left behind after the crystallisation of suger from suger cane juice.

How do they manufacture ethanol?

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88. In our country ethanol is mainly manufactured by suger industries the mother liquor left behind after the crystallisation of suger from suger cane juice.

How is wine prepared from grape juice?

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89. In our country ethanol is mainly manufactured by sugar industries the mother liquor left behind after the crystallisation of sugar from sugar cane juice.

What is denatured spirit?

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90. A student wants to prepare 2-propanol using methyl magnesium iodide (Grignard reagent). Which other organic compound he has to use to get the product?

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91. Which carbonyl compound will give tert-butyl alcohol when treated with methyl magnesium iodide?

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92. Explain the reaction of tert-butyl alcohol with Lucas reagent.

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93. Williamson's synthesis is a convenient method used to prepare both symmetrical and unsymmetrical ethers.

What are the starting materials required to prepare anisole by Williamson's synthesis?

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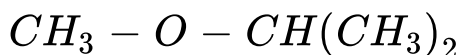
94. Williamson's synthesis is a convenient method used to prepare both symmetrical and unsymmetrical ethers.

Write equation for the reaction of anisole with hydriodic acid.

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95. Williamson's synthesis is a convenient method used to prepare both symmetrical and unsymmetrical ethers.

Write the IUPAC names of (1) anisole (2)



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96. Ethers show metamerism. Explain the term metamerism taking a suitable example.

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