

CHEMISTRY

BOOKS - A N EXCEL PUBLICATION

ALDEHYDE, KETONES AND CARBOXYLIC ACIDS

Question Bank

- 1. Write the structures of the following compounds
- (i) α -Methoxypropionaldehyde



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2. Write the structures of the following compounds (ii) 3-Hydroxybutanal



3. Write the structures of the following compounds (iii) 2-Hydroxycyclopentane carbaldehyde



4. Write the structures of the following compounds (iv) 4-Oxopentanal

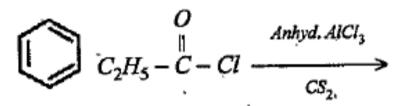


5. Write the structures of the following compounds (v) Di-sec-butyl ketone



6. Write the structures of the following compounds (vi) 4-Fluoroacetophenone

7. Write the structues of the product of the following reactons. (i)





8. Write the structues of te product of the following reactons. (ii)

`(C_6H_5CH_2)_2Cd+2Ch_3COCl gives

9. Write the structues of te product of the following reactons. (iii)



 $CH_3-C\equiv CH\stackrel{Hg^{2+}H_2SO_4}{\longrightarrow}$

10. Arrange the following compounds in the increasing order of their boiling points.

 $CH_3CHO, CH_3CH_2OH, CH_3OCH_3, CH_3CH_2CH_3$



11. Arrange the following compounds in increasing order of their reactivity in nucleophilic addition reactions. (i) Ethanal, propanal, propanone, butanone



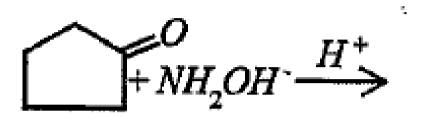
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12. Arrange the following compounds in increasing order of their reactivity in nucleophilic addition reactions. (ii) Benzaldehyde, P-Tolualdehyde, P-Nitrobenzaldehyde Acetophenone



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13. Predict the products of the following reactions.



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14. Predict the products of the following reactions.



15. Predict the products of the following reactions.

$$R - CH = CH - CHO + H_2N - C - NH - NH_2 \longrightarrow$$



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16. Predict the products of the following reactions.

$$C-CH_3 + CH_3CH_2NH_2 \xrightarrow{H^+}$$



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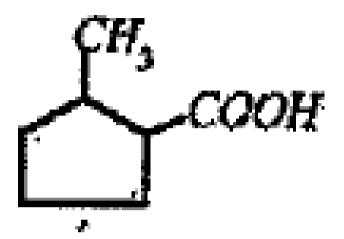
17. Give the IUPAC names of the following. (i) $PhCH_2CH_2COOH$



18. Give the IUPAC names of the following. (ii) $\left(CH_3\right)_2 = CHCOOH$

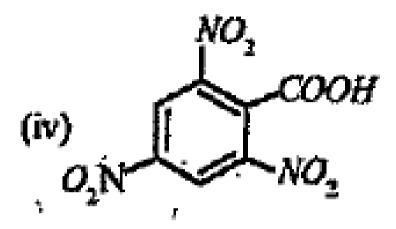


19. Give the IUPAC names of the following. (iii)





20. Give the IUPAC names of the following. (iii)



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21. Show how each of the following compounds can be converted to benzoic acid. (i) Ethylbenzene



22. Show how each of the following compounds can be converted to benzoic acid. (ii) Acetophenone



23. Show how each of the following compounds can be converted to benzoic acid. (iii)

Bromobenzene



24. Show how each of the following compounds can be converted to benzoic acid. (iv) Phenylethene (styrene)



25. Which acid of each pair shown here would you expect to be stronger? (i) CH_3COOH, FCH_2COOH



26. Which acid of each pair shown here would you expect to be stronger? (ii) $FCH_2COOH, ClCH_2COOH$



27. Which acid of each pair shown here would you expect to be stronger? (iii) $FCH_2CH_2CH_2COOH$ or CH_3CHFCH_2COOH



28. Which acid of each pair shown here would you expect to be stronger? (i) CH_3COOH, FCH_2COOH



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29. Aldehydes and ketones are organic compounds containing carbonyl group. Write a chemical reaction to distinguish between aldehydes and ketones.



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30. Aldehydes and ketones can be subjected to Clemmensen reduction and Wolf-Kishner reduction. Name the reagents used in both cases.



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31. How will you make the following conversions? Ethanoic acid to ethanol.



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32. How will you make the following conversions? Propanoic acid to 2-chloropropanoic acid.



33. How will you make the following conversions? Toluene to benzoic acid.



34. Following are a group of compounds showing acidic behaviour:

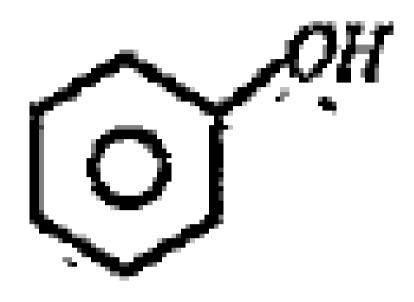
$$HCOOH$$
, О , CH_3COOH , CH_3-CH_2-COOH

Give the IUPAC names of these compounds.



35. Following are a group of compounds showing acidic behaviour:does not contain carboxylic

group, still it is acidic. Why?



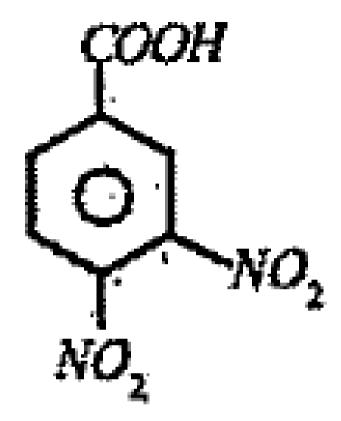


36. Following are a group of compounds showing acidic behaviour: Phenols are less acidic than carboxylic acids. Why?



37. Following are a group of compounds showing acidic behaviour: Formic acid is stronger than acetic acid. Why?



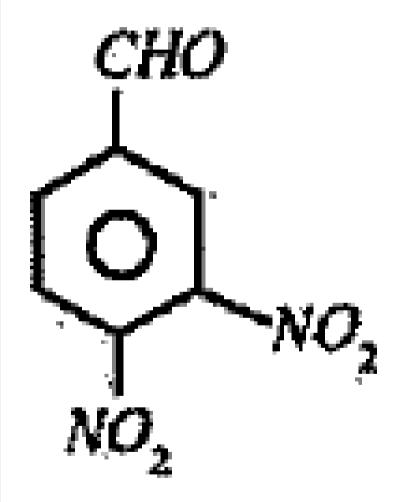


38.

is an aromatic acid. What is it IUPAC name?

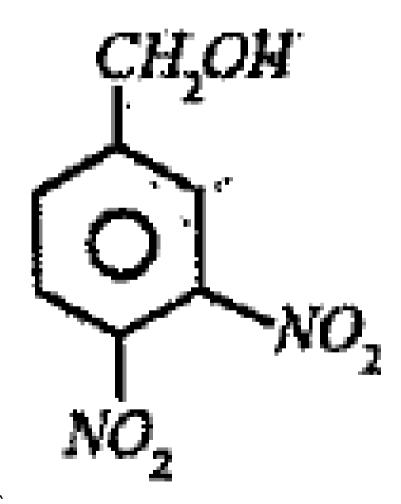


39. Explain the conversion of the 3,4-dinitrobenzoic acid to the following.

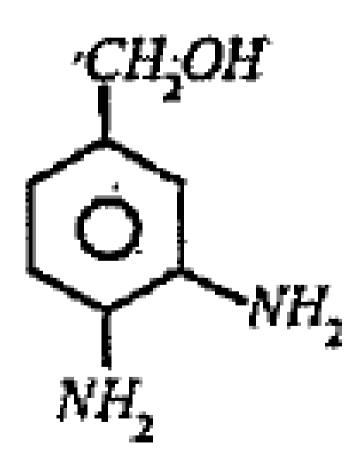




40. Explain the conversion of the above acid to the following.



41. write the conversion of the 3,4-dinirtobenzoic acid to the following.



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42. Aldehydes resemble ketones in many respects. Give the reason for their resemblance.



43. Aldehydes resemble ketones in many respects. Give a reaction in which aldehydes resemble ketones.



44. Write a test to distinguish between aldehydes and ketones.



45. Aldehydes resemble ketones in many respects. What is Cannizzaro reaction?



46. Which named reaction is used to reduce CH_3COCl to CH_3CHO ?

47. Aldehydes and ketones undergo reactions due to the presence of α -hydrogen atom. Write the name of the reaction of aldehyde which takes place only because of the presence of α -hydrogen atom.



48. How will you bring about the above reaction?



49. $CH_2ClCOOH$ is a stronger acid than CH_3COOH . Why?



50. How will you convert CH_3COOH to $CH_2ClCOOH$?



51. Complete the following. Write down the structures of A, B and C. (i)

$$CH_3 - CH_2 - CHO \xrightarrow{KMnO_4} A$$



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52. Completethefollowing.WritedownthestructureofB.(i)

 $CH_3-CH_2-CO-CH_3 \stackrel{Z\,\in\, camal\,gam\,/\,HCl}{-\!\!\!-\!\!\!-\!\!\!-\!\!\!-\!\!\!-\!\!\!-\!\!\!-\!\!\!-\!\!\!-} B$



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 $CH_3-CH_2-CH_2-COOH \xrightarrow{Bro\, {
m min}\, e\,/\, RedP} C$



54. Write down the IUPAC names of A, B and C.



55. Explain the following reactions. Cannizzaro reaction



56. Explain the following:

Esterification



57. Suggest a method of preparation of benzaldehyde from toluene.



58. Aldehydes and ketones differ in their chemical reactions. How do they with the following.?

Tollens's reagent



59. Aldehydes and ketones differ in their chemical reactions. How do they with the following.? Alcohol



60. How will you convert propanoic acid into the following componds? Ethane



61. How will you convert propanoic acid into the following compond? Butane



62. Among formaldehyde, acetaldehyde, benzaldehyde and formic acid, which compounds undergo Cannzzaro reaction? Give reason.



63. Explain the following :

Esterification



64. Thionyl chloride is preferred to as the reagent to prepare acid chlorides. Why?



65. Write the chemical reaction to effect the transformation of sodium acetate to ethane.

$$CH_3-CH_2-O-CH_2-CH_3$$
 ?
a) $CH_3-O-CH_2-CH_2-CH_3$

b)
$$CH_3-CH_2-CH_2-CH_2-OH$$
c) $CH_3-CH_2-CO-CH_3$

 $\mathsf{d})CH_3 - CH(OH) - CH_2 - CH_3$



67. Write the IUPAC names of the $HOOC-CH_2-COOH$ compound.



68. Aldol condensation reaction is a special reaction of aldehydes. (i) What is aldol condensation reactions?



69. Aldol condensation reaction is a special reaction of aldehydes. (ii) Write the structural formula of aldol formed from ethanal.



70. Write simple chemical tests and observations used to distinguish between the following compounds. (i) Propanal and propanone



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71. Write simple chemical tests and observations used to distinguish between the following compounds. (ii) Phenol and benzoic acid



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72. Write the names of reagents used to bring about the following transformations. (i) $C_6H_5COCl
ightarrow C_6H_5CHO$



73. Write the names of reagents used to bring about the following transformations. (ii) $CH_3COOH
ightarrow CICH_2 - COOH$



74. Methanal (HCHO) is an aldehyde having no α hydrogen atom. What are the products formed when methanal is treated with strong KOH solution?



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75. How are the following conversions archieved? (i) Benzoyl chloride $[C_6H_5COCl]$ to benzaldehyde (C_6H_5-CHO)





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76. How are the following conversions archieved? (ii) Acetic acid (CH_3COOH) to chloro acetic acid $(CH_2Cl-COOH)$



77. How are the following conversions archieved?(iii) Benzene to Benzaldehyde



78. Aldehydes, Ketones and Acids contain angle C=O group. Name the product obtained by the reaction

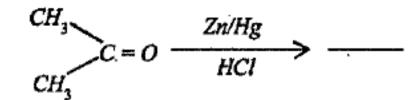
between Acetic acid and Ethanol.



79. Write a test to distinguish between aldehydes and ketones.

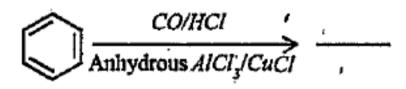


80. Two chemical reactions are given below: (1) Identify the products of each reaction.





81. Two chemical reactions are given below: (2) Give the name of each reaction.





82. Explain aldol condensation taking CH_3-CHO as example.



83. Write the named reactions involved in the following conversions: (i)

$$CH_3-CO_Cl \xrightarrow{H_2/Pd-BaSO_4} CH_3-CHO$$



84. Write the named reactions involved in the following conversions: (ii)

$$2HCHO \stackrel{NaOH}{\longrightarrow} HCOONa + CH_3 - OH$$



85. How are the following conversions achieved? (i)

$$CH_3 - CN \rightarrow CH_3 - COOH$$



86. How are the following conversions achieved?(ii)

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87. Aldehydes, Ketones and Carboxylic acids are Carbonyl compounds. Aldehydes differ from Ketones in their oxidation reaction. Illustrate with one example.



88. How will you prepare benzaldehyde by Gatterman-Koch reaction?



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89. Write the reactions of carboxylic acid with the following reagents. (Write the chemical equations)

(i) Thionyl chlorides `(SOCl 2)



90. Write the names of reagents used to bring about the following transformations. (ii) $CH_3COOH \rightarrow CICH_2 - COOH$



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91. Write the reactions of carboxylic acid with the following reagents. (Write the chemical equations) (iii) Lithium Aluminium hydride $(LiAlH_4)$ ether.



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92. Write a test to distinguish between aldehydes and ketones.



93. How will you prepare benzaldehyde by Etard's reaction?



94. How will you bring about the following conversions? (Write the chemical equations) (i)

Ethonal to ethene



95. How will you bring about the following conversions? (Write the chemical equations) (ii) Benzamide



96. How will you bring about the following conversions? (Write the chemical equations) (iii) benzoic acid to Benzaldehyde

97. Aldehydes, Ketones are the compounds having >C=O group. Choose the IUPAC name of the compound $CH_3CH=CH=CHO$

A. Propen-1-al

B. But-2-en-1-al

C. Butanal

D. But-2-en-2-al

Answer: A



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98. Complete the reaction: $2HCHO + ConKOH \xrightarrow{\Delta}$



99. Complete the reaction:
$$CH_3CHO \xrightarrow{dil.NaOH}$$



 $CH_3CHO + H_2N - NH_2 \rightarrow$

101. Complete the reaction: $C_6H_5COCH_3 \xrightarrow[HCl]{Zn/Hg}$



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102. Aldehydes, Ketones and acids contain >C=O group. Choose the IUPAC name of the compound $(CH_3)_2CHCOOH$

A. Butanoic acid

B. Ethanoic acid

C. 2-methyl propanoic acid

D. Propanoic acid

Answer: A



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103. Complete the reaction:

$$CH_{3}CH_{2}COOH \xrightarrow{\mathit{LiAlH_4/Ether}}$$



$$CH_{3}CH_{2}COOH + SOCl_{2}
ightarrow$$

Complete the reaction:

the

reaction:

reaction:

104.

105.

 $CH_3CH_2COOH \xrightarrow{Br_2 \, / \, RedP}$

Complete

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106. Complete the
$$CH_3CH_2COOH + CH_3OH^{+}$$

107. The product obtained when benzene is treated with carbon monoxide and hydrogen chloride in presence of anhydrous $AlCl_3$ is

- A. Chlorobenzene
- B. Phenol
- C. Benzaldehyde
- D. Benzoic acid

Answer: A



108. How will you carry out the following conversion?

 $CH_3CH_2COOH o CH_3CH_2CH_2OH$



109. How will you carry out the following conversions? (iv)

 $CH_3COOH
ightarrow CH_3CO - O - CO - CH_3$

(acetic anhydride)



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110. Explain the following:

Esterification



111. Explain the Tollen's test.



112. Explain the HVZ reaction.



113. Explain the Decarboxylation of Carboxylic acid.



114. Which among the following reduces Tollen's reagent?

A. Methanal

B. Propanone

C. Benzophenone

D. Acetophenone

Answer: A



115. Since both aldehydes and ketones possess carbonyl functional group, they undergo similar chemical reactions. (i) Explain the structure of carbonyl group.



116. Since both aldehydes and ketones possess carbonyl functional group, they undergo similar

chemical reactions. (ii) Explain Aldol condensation with an example.



117. Which among the following does not give red precipitate with Fehling's solution?

- A. Ethanal
- B. Propanal
- C. Butanal
- D. Benzaldehyde

Answer: D



118. How will you bring about the following conversions? (i) Toluene into Benzaldehyde



119. How will you bring about the following conversions? (ii) Benzoic acid to Benzamide



120. Explain Cannizaro reaction with an example.



121. Acid chlorides can be use to prepare aldehydes and ketones. How is acetyl chloride converted to acetaldehyde?



122. Acid chlorides can be use to prepare aldehydes. What is the above reaction known as?

123. Acid chlorides can be use to prepare aldehydes and ketones. Can you prepare acetone from acetyl chloride? Explain.



124. Identify the product Y in each of the following.

(i)
$$CH_3CHO + HCN o X \xrightarrow{hydrolysis} Y$$



125. Identify the product Y in each of the following.

$$CH_3CN+CH_3MgBr o X\stackrel{H_2O}{\longrightarrow} Y$$



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126. Identify the product Y in each of the following.

(iii)
$$CH_3CH_2MgBr + CO_2 \stackrel{ether}{\longrightarrow} X \stackrel{hydrolysis}{\longrightarrow} Y$$



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127. Identify the product Y in each of the following.

(iii)
$$CH_3COOH \xrightarrow{(i)\,NH_3} X \xrightarrow{heat} Y$$



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128. Aldehydes and ketones undergo addition reaction followed by elimination of water with certain reagents. Give the structure and names of two such reagents.



129. Aldehydes and ketones undergo addition reaction followed by elimination of water with certain reagents. Write equation for the reaction of any one of them with propanone.

130. Aldehydes and ketones undergo addition reaction followed by elimination of water with certain reagents. What happens when benzaldehyde is heated with zinc amalgam and conc. HCI?



131. Give reasons for the following observations. Aldehydes and ketones undergo nucleophilic

addition reactions.



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132. Give reasons for the following observations.



 CH_3COOH is weaker than HCOOH

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133. Complete the following table by writing the name of the reagent, organic products and name

of the reaction wherever required.

SL. No.	Reactant	Reagents	Organic product	Name of reaction
1.	C ₆ H ₅ CONH ₂		C ₆ H ₅ NH ₂	
2.	C ₆ H ₅ NH ₂	C ₆ H ₅ COCl/NaOH		
3.	ČH ₃ CH ₂ NH ₂		CH ₃ CH ₂ NC	
4.	C ₆ H ₅ N ₂ Cl	$C_6H_5NH_2$		
5.	$C_6H_5N_2CI$	Cu/HBr		********



134. Carboxylic acids can be prepared using Grignard reagents. Explain the preparation acid using a suitable Grignard reagent.



135. Carboxylic acids can be prepared using Grignard reagents. Why is benzoic acid more acidic than ethanoic acid?



136. Carboxylic acids can be prepared using Grignard reagents. Why is carboxylic acid stronger acid than phenol?



137. Which among the following give aldol condensation? HCHO



138. Which among the following give aldol condensation? $\mathbb{C}l_3-CHO$



139. Which among the following give aldol condensation? CH_3-CHO

