

India's Number 1 Education App

CHEMISTRY

BOOKS - NTA MOCK TESTS

HALOALKANES AND HALOARENES TEST

Multiple Choice Questions

1. An $S_N 2$ reaction at an asymmetric carbon of

a compound always gives -

A. An enantiomer of the substrate

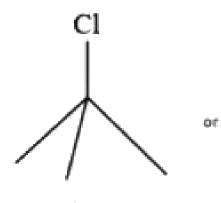
- B. A product with opposite optical rotation
- C. A mixture of diastereomers
- D. A single steroisomer

Answer: D

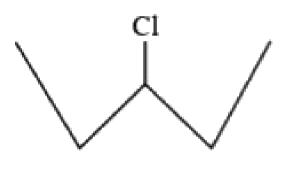


2. Which one of the following two substances

undergoes $S_N 1$ reaction faster ?









A. (i)

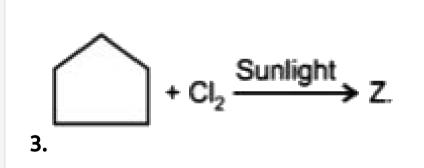
B. (ii)

C. (iii)

D. Both have equal rate for reaction.

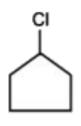
Answer: A





Major Product (Z) will be if Cl_2 is taken in

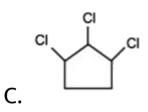
large excess.

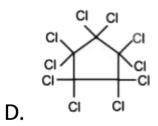


A.

Β.



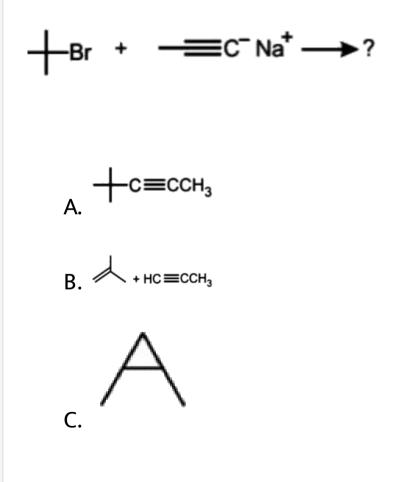




Answer: D



4. What do you expect to be the main product when tertbutyl bromide reacts in the following manner ?

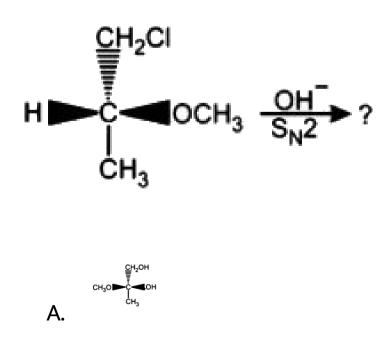


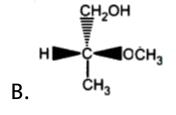
D.

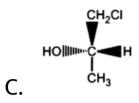
Answer: B

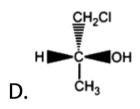


5. What is the major product of the reaction ?





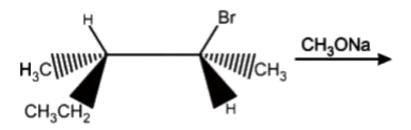




Answer: B



6. Choose the correct statement about the major product formed in E_2 reaction ?



A. The major product will be optically active

B. The major product will be trans-3-

Methyl-2pentene

C. The major product will be cis-3-Methyl-2-

pentene

D. The major product will be 3-Methyl-1-

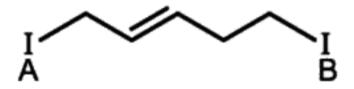
pentene

Answer: B

View Text Solution

7. Which one of the two lodine atoms will be

more reactive in the $S_N 1$ and $S_N 2$ reaction ?



A. A will be faster in $S_N 1$ reaction but

slower in $S_N 2$

B. A will be faster both in $S_N 1$ and $S_N 2$

reaction

C. A and B will be equally reactive.

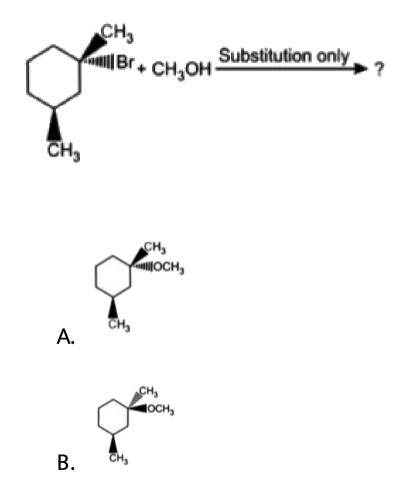
D. B will be faster in both $S_N 1$ and $S_N 2$

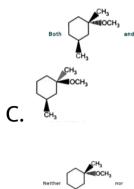
reaction

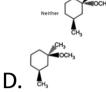
Answer: B

8. What is the substitution product of the

following reaction ?





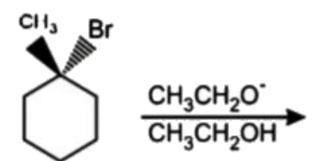


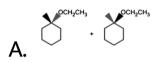
Answer: C

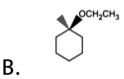


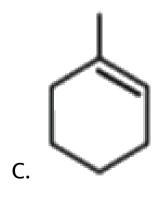
9. What is the major product of the reaction

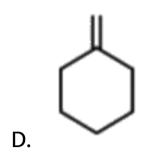
shown?











Answer: C

View Text Solution

10. $CH_3Br \xrightarrow{KCN} A \xrightarrow[LiAlH_4]{4[H]} CH_3CH_2NH_2$

IUPAC name of A is:

A. Methyl cyanide

- B. Methyl isonitrile
- C. Acetonitrile
- D. Ethane nitrile

Answer: D

View Text Solution

11. Which of the following would be the best (most reactive) nucleophile in the polar medium ?

A. $I^{\,-}$

B. Br^{-}

C. Cl^{-}

D. $F^{\,-}$

Answer: A



12. The isomer of hexane that gives the minimum and maximum number of monochloro derivatives are respectively

A. 3-methylpentane		and	2,	3-
dimethylbutane				
B. 2, 3-dimethylbutane and n-hexane				
C. 2,	2-dimethylbu	tane	and	2-
methylpentane				
D. 2, 3-dimethylbutane and 2-methypentane				

Answer: D

13. Of the five isomeric hexanes, the isomer which can give only two monochlorinated compounds is (excluding stereoisomers)

A. 2-methylpentane

B. 2,2-dimethylbutane

C. 2, 3-dimethylbutane

D. n-hexane

Answer: C

14. Which of the following alkane on mixing with chlorine and irradiating the mixture with ultraviolet light, form only one monochloroalkane.

A. Propane

B. Pentane

C. Iso-pentane

D. Neo-pentane

View Text Solution

Answer: D

15. In the reaction sequence,

$$C_2H_5Cl+KCN \xrightarrow{C_2H_5OH} X \xrightarrow{H_3O^\oplus} Y$$

What is the molecular formula of Y?

A. $C_3H_6O_2$

 $\mathsf{B.}\, C_3H_5N$

 $\mathsf{C.}\, C_2 H_4 O_2$

D. C_2H_6O

Answer: A





16. Which halide does not get hydrolysed by sodium hydroxide ?

A. Vinyl chloride

B. Methyl Chloride

C. Ethyl chloride

D. Isopropyl chloride

Answer: A

17. Identify X and Y in the following sequence: $C_2H_5Br \xrightarrow{X} \text{product} \xrightarrow{Y} C_3H_7NH_2$

A. X = KCN, $Y = LiAlH_4$

B. X = KCN, $Y = H_3 O^+$

 $\mathsf{C}.\, X=CH_3NH_2, Y=HNO_3$

D. None of these

Answer: A

18. In the following sequence of reactions:

 $C_2H_5Br \xrightarrow{AgCN} X \xrightarrow{ ext{Reduction}} Y$, Y is:

A. n-propyl amine

B. Isopropylamine

C. Ethylamine

D. ethylmethyl amine

Answer: D

19. Which of the following applies in the reaction,

$$CH_3CHBrCH_2CH_3 \xrightarrow{\text{Alc. KOH}}$$

(i) $CH_3CH = CHCH_3$ (major product)

(ii) $CH_2 = CHCH_2CH_3$ (minor product)

A. Markownikoff's rule

B. Saytzeff's rule

C. Kharasch effect

D. Hofmann's rule

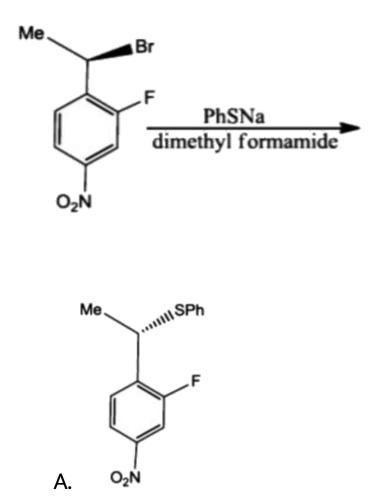
Answer: B

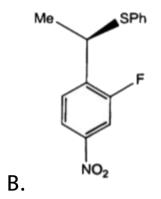


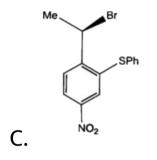


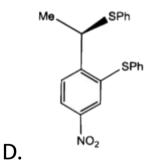
20. The major product of the following

reaction is:









Answer: A



21. On warming with silver powder, chloroform is converted into:

A. Acetylene

B. Hexachloroethane

C. 1,1, 2, 2-tetrachloroethane

D. Ethene

Answer: A

22. When Ethyl chloride and alcoholic KOH are

heated, the compound obtained is:

A. C_2H_4

 $\mathsf{B.}\, C_2 H_2$

 $\mathsf{C.}\, C_6 H_6$

D. C_2H_6

Answer: A

23. Which of the following will displace the halogen from the solution of the halide ?

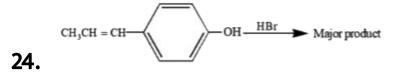
A. Br_2 added to Nal

B. Br_2 added to NaCl

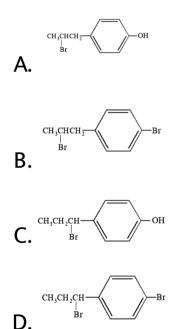
C. Cl_2 added to KCI

D. Cl_2 added to NaF

Answer: A



The major product in the above reaction is:



Answer: C



25. The organic chloro compound, which shows most readily $S_N 2$ reaction, is:

A. CH_3Cl

 $\mathsf{B.} \left(C_2 H_5 \right)_2 CHCl$

 $\mathsf{C}.\,(CH_3)_3\mathbb{C}l$

D. $(CH_3)_2 CHCl$

Answer: A

26. $CH_3CH_2COOH \xrightarrow{Cl_2} A \xrightarrow{\operatorname{alc} \operatorname{KOH}} B$

What is B?

A. CH_3CH_2COCl

B. CH_3CH_2CHO

C. $ClCH_2CH_2COOH$

 $\mathsf{D}. \, CH_2 = CHCOOH$

Answer: D

27. The $S_N 2$ reaction involves back-side attack and therefore results in a "Walden Inversion." For which one of the substrates shown would you be able to demonstrate that such backside attack with "Walden Inversion" has in fact occurred ?

- A. 1-bromopropane
- B. 2-bromobutane
- C. 3-bromopentane
- D. Methyl bromide

Answer: B



28. 2, 2-dimethyl-1-chloropropane undergo nucleophilic substitution reaction to produce an optically inactive product which of the following is correct regarding the product ? (i) Formation of product takes place through SN^1 reaction.

(ii) The product is optically inactive due to internal compensation.

(iii) The product is optically inactive due to

external compensation.

(iv) The carbocation is not an intermediate in

the given reaction.

Choose the correct option.

A. ony (i)

B. (i) and (ii)

C. (i), (ii) and (iii)

D. All of these

Answer: A



29. In the nucleophilic substitution reactions ($S_N 2$ or $S_N 1$) the reactivity of alkyl halides follows the sequence.

A. R-I > R-Br > R-Cl > R-FB. R-Cl > R-F > R-Br > R-IC. R-F > R-Cl > R-Br > R-ID. R-I > R-F > R-Cl > R-Br > R-I

Answer: A





30. The order of reactivities of methyl halides in the formation of Grignard reagent is,

A. $CH_3I > CH_3Br > CH_3Cl$

B. $CH_3Cl > CH_3Br > CH_3I$

C. $CH_3Br > CH_3Cl > CH_3I$

D. $CH_3Br > CH_3I > CH_3Cl$

Answer: A

