

India's Number 1 Education App

#### **CHEMISTRY**

#### **BOOKS - ARIHANT PUBLICATION**

#### **ARYL DIAZONIUM SALTS**

#### Questions For Practice Multiple Choice Type Questions 1 Mark

**1.** Which of the following will be most stable diazonium salt  $RN_2^+X^-$  ?

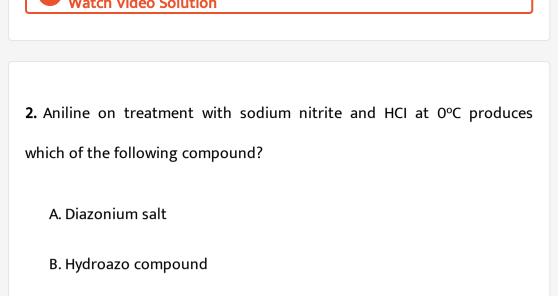
A. 
$$CH_3N_2^{\,+}\,X$$

B. 
$$C_6 H_5 N_2^{\,+} X^{\,-}$$

C. 
$$CH_3CH_2N_2^{\,+}X^{\,-}$$

D. 
$$C_6H_5CH_2N^+X^-$$

#### Answer: b



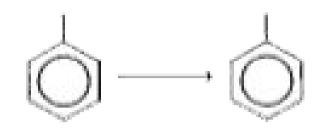
C. Phenol and  $N_2$ 

D. Nitroaniline

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Answer: a

3. Consider the following reaction,



The above

reaction is called

- A. Carbylamine reaction
- B. Gattermann synthesis
- C. Sandmeyers reaction
- D. Balz-Schiemann reaction

#### Answer: b



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- 4. Identify the product in the following reaction
- $3, 4, 5 \text{tribromoaniline} \xrightarrow{(i) \text{ Diazotisation}} \xrightarrow{(ii) H_3 PO_2}$

A. 3, 4, 5-tribromobenzene

B. 1, 2, 3-tribrompbenzene

C. 2, 4, 6.-tribromobenzene

D. 3, 4, 5-tribromonitrobenzene

### Answer: b



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5. Which of the following reaction will not occur?

A. 
$$C_6H_5N_2^+ + CuBr \stackrel{HBr}{\longrightarrow} C_6H_5 - Br$$

B.

$$C_6H_5N_2^{+} + C_6H_5OH 
ightarrow C_6H_5 - N = (\#\#ARH_{\mathbb{C}}\ \_CHSE_ODI_CH_2)$$

C. 
$$C_6H_5N_2^+ \xrightarrow[HOH]{H_3PO_2} C_6H_5H$$

(##ARH CC CHSE ODI CHE XII C22 E01 005 O04.png" width="30%">

D.  $C (6)H (5)N (2)^{+} + I^{-} to C (6)H (5) - N$ 

# Answer: b View Text Solution Questions For Practice Very Short Answer Type Questions 1 Mark 1. Write the structure of benzene diazonium chloride. Watch Video Solution 2. The conversion of primary aromatic amines into diazonium salts is known as **Watch Video Solution**

**3.** Diazonium salts of aromatic amines are more stable than those of aliphatic amines. Explain.



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**4.** Why is benzene diazonium chloride not stored and used immediately after its preparation?



**5.** Complete the following reactions.

(i) 
$$C_6H_5N_2Cl+H_3PO_2+H_2O
ightarrow$$

(ii) 
$$C_6H_6N_2Cl \xrightarrow[(ii) NaNO_2/Cu]{(ii) NaNO_2/Cu}$$



**6.** What is Sandmeyer's reaction?



**7.** Complete the following reaction:

$$C_6H_5N_2Cl+C_2H_5OH
ightarrow$$



8. Benzene diazonium chloride reacts with potassium iodide to form\_\_\_\_\_



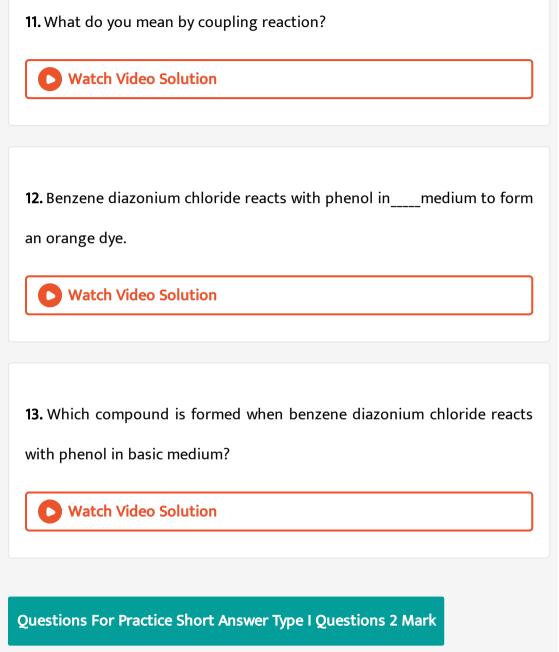
**9.** Complete the following reaction.

$$C_6 H_5 N_2^{\,+} C l^{\,-} \xrightarrow[ ext{Room temperature}]{H_2 O}$$



**10.** Write the chemical reaction occuring in the preparation of fluorobenzene from aniline.





1. What is diazo reaction? Give equation.



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2. How is phenol prepared from diazonium salt?



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3. Suggest a route by which the following conversion can be accomplished.

 $C_6H_5\overset{+}{N_2}\overset{-}{C}l
ightarrow C_6H_5CONH_2$ 



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**4.** Give a chemical test to distinguish between aniline and ethyl amine.



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Questions For Practice Short Answer Type Ii Questions 3 Mark

<b>1.</b> Accomplish	the following o	conversions. (	i) Benzene to	m-bromopheno
	0	•	• •	



- 2. Accomplish the following conversions.
- (i) Aniline to benzyl alcohol
  - Watch Video Solution

- 3. What is coupling reaction? Give an example.
  - Watch Video Solution

4. Give the structures of A, B and C in the following reactions.

 $C_6H_5N_2^+Cl^- \stackrel{CuCN}{\longrightarrow} A \stackrel{(H_2O)\ /H^+}{\longrightarrow} B \stackrel{NH_3}{\longrightarrow} C$ 



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5. Do the conversion: 3 - Methylaniline to - 3- Nitrotoluene.



**Questions For Practice Long Answer Type Questions 7 Marks** 

1. Give one methods of preparation of benzene diazonium chloride. Write its synthetic applications for preparation of (a) chlorobenzene (b) phenol (c) benzene (d) iodobenzene.



2. Starting from diazonium salt how can (i) benzene (ii) iodobenzene and (iii) benzoic acid be prepared?



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- 3. How is benzene diazonium chloride prepared from aniline? How are the following compounds prepared from it?
- (i) Benzene (ii) Phenol (iii) Iodobenzene (iv) Phenylcyanide.



4. What is Diazo reaction? Write the structure of the product of this reaction. Using this reaction how the following compounds are prepared?

(i)Benzene (ii) Fluorobenzene



5. A hydrocarbon  $A(C_4H_8)$  on reaction with HCl gives a compound 'B'  $(C_4H_9Cl)$  which on reaction with 1 mole of  $NH_3$  gives compound ' $C'(C_4H_{11}N)$ '. On reacting with NaNO2 and HCl followed by the treatment with water, compound 'C yields an optically active alcohol, 'D'. Ozonolysis of 'A gives 2 moles of acetaldehyde. Identify compounds 'A to 'D'. Explain the reactions involved.

6. An aromatic compound 'A' of molecular formula  $C_7H_7ON$  undergoes a series of reactions as shown below. Write the structures of A, B, C, D and E in the following reactions.

$$(C_7H_7ON)A \xrightarrow{N_2 + KOH} C_6H_6NH_2$$

$$N_6NO_2 + HCI \longrightarrow B \xrightarrow{E} CH_4CH_7OH \longrightarrow C$$



Odisha Bureau S Textbook Solutions Multiple Choice Type Questions

1. The diazonium salts are the reaction products of the reaction of nitrous acid with

- A. primary aliphatic amines B. primary aromatic amines C. secondary aliphatic amines D. secondary aromatic amines **Answer: B Watch Video Solution** 2. Preparation of a diazquium salt from a primary aromatic amine is known as
- - A. coupling reaction
  - **B.** Sandmeyer reaction
  - C. diazotisation
  - D. Corey-House synthesis

Answer: C

3. Which of the following reagents is used to prepare benzene diazonium chloride from aniline?

A. 
$$NaNO_2 + HCl$$

B. 
$$LiAlH_4$$

$$\mathsf{C.}\,NH_2NH_2+KOH$$

D. NaOH

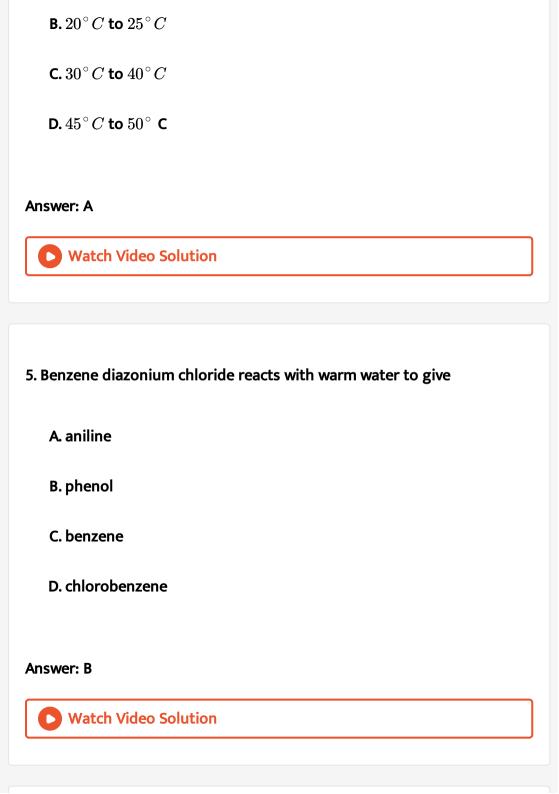
**Answer: A** 



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4. Which of the following are optimum temperature conditions for making benzene diazonium chloride from aniline?

A.  $0^{\circ}C$  to  $10^{\circ}$  C



A. conc.HBr
B. $Br_2/FeBr_3$
C. CuBr
D. nitrous acid then CuBr
Answer: D
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7. Chlorobenzene can be prepared by treatment of aniline with
A. cuprous chloride
B. chlorine in the presence of UV light
C. nitrous acid followed by treatment with CuCl
D. chlorine in the presence of $FeCl_3$

6. Bromobenzene can be prepared by treating aniline with

#### Answer: C



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- 8. Iodobenzene can be prepared by
  - A. treating chlorobenzene with  $l_2$  using  $FeCl_3$  catalyst
  - B. treating phenol with  $I_2$  in NaOH solution
  - C. treating benzene diazonium chloride with KI
  - D. treating benzene with  $CH_3I$  using  $AlCI_3$  catalyst

#### **Answer: C**



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- 9. Benzene diazonium chloride reacts with KI to form
  - A. Benzene diazonium iodide

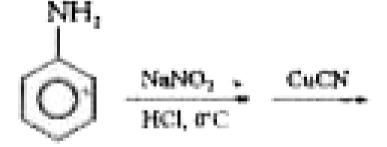
- B. m-diiodobenzene
- C. Iodobenzene
- D. o-,m and p-diiodobenzene

#### **Answer: C**



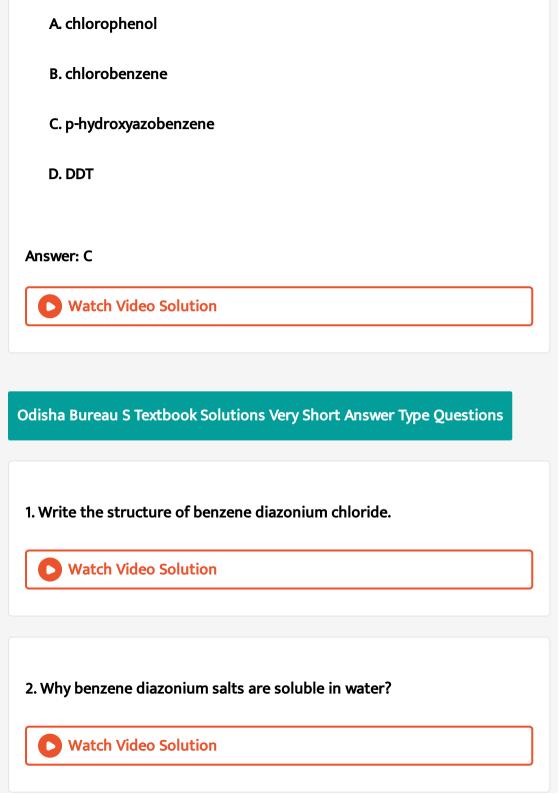
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### 10. What is the major product of the following reaction?



- A. Benzonitrile
- B. Benzoic acid
- C. Nitrobenzene

D. Benzene diazonium chloride		
Answer: A		
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11. Benzene diazonium chloride reacts with hypophosphorus acid to form		
A. phenol		
B. benzaldehyde		
C. aniline		
D. benzene		
Answer: D		
Watch Video Solution		
12. Benzene diazonium chloride reacts with phenol to form		



3. What should be the nature of medium (neutral/acidic/basic) for diazotisation of arylamines.



- 4. Give only the equations for the synthesis of the following from benzene diazonium chloride.
- (i)Chlorobenzene
- (ii) Fluorobenzene
- (iii) p-hydroxyazobenzene



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Odisha Bureau S Textbook Solutions Short Answer Type I Questions

1. Suggest reason why excess mineral acid is used in diazo reaction.



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- 2. Complete the following equations.
- (i)  $C_6H_5N_2Cl+H_2O\stackrel{\Delta}{\longrightarrow}$
- (ii)  $C_6H_5N_2Cl \stackrel{CuCl}{\underset{HCl}{\longrightarrow}}$
- (iii)  $C_6H_5OH \xrightarrow{ArN_2^+Cl^-} OH^-$

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- 3. Give the chemical equation for the following.
- (i) Diazotisation reaction
- (ii) Coupling reaction
- (iii) Sandmeyer reaction



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- 4. How will you convert?
- (i) Aniline to nitrobenzene.
- (ii) Aniline to iodobenzene

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5. How you will obtain p-hydroxyazobenzene from aniline?



6. What is coupling reaction? Give an example.



7. How will you carry out the conversion of aniline to m-bromonitrobenzene?



8. Give the structure of A,B and C in the following reactions.

(i) 
$$C_6H_5N_2Cl \stackrel{CuCN}{\longrightarrow} A \stackrel{(H_2O)\ /\ H^+}{\longrightarrow} B \stackrel{NH_3}{\longrightarrow} C$$

(ii)  $C_6H_5NO_2 \stackrel{Fe/HCl}{\longrightarrow} A \stackrel{HNO_2}{\underset{273-278K}{\longrightarrow}} B \stackrel{C_6H_5OH}{\longrightarrow} C$ 



# Odisha Bureau S Textbook Solutions Short Answer Type Ii Questions

1. Diazonium salts of aromatic amines are more stable than those of aliphatic amines. Explain.



# Odisha Bureau S Textbook Solutions Long Answer Type Questions

1. Discuss in detail the preparation of benzene diazonium chloride from aniline. Why it is not separated in solid state and used as soon as produces in solution? How can you convert benzene diazonium chloride to fluoro and iodobenzene.



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3. How will you obtain?
(i) lodobenzene from nitrobenzene.
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4. How will you synthesis the following?
(i) Azobenzene from benzene
(ii) Benzene from azobenzene
(iii) Azobenzene from nitrobenzene
(iv) Nitrobenzene from azobenzene
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2. Describe the synthetic applications of aryl diazonium salts.

5. Give the chemical equation for the following. (i) Diazotisation reaction (ii) Coupling reaction (iii) Sandmeyer reaction **Watch Video Solution** 6. Starting from diazonium salt how can (i) benzene (ii) iodobenzene and (iii) benzoic acid be prepared? **Watch Video Solution Chapter Practice Multiple Choice Type Questions 1 Mark** 1. The diazonium salts have the general formula ..... Select an appropriate statement in order to complete the above statement.

A.  $RN_2^{\,+}X^{\,-}$  , where, R stands for an aryl group and  $X^{\,-}$  ion may be

$$Cl^{-}\,,Br^{-}\,,HSO_{4}^{-}\,,BF_{4}^{-}$$
 etc.

B.  $R_2 N^+ X^-\,$  where , R stands for an aryl group and  $X^-\,$  ion may be

$$Cl^-, Br^-, HSO_4^-, BF_4^-$$
 etc.

C.  $RN_2^{\,+}X^{\,-}$  , where, R stands for an alkyl group and  $X^{\,-}$  ion may be

$$Cl^{-}, Br^{-}, HSO_{4}^{-}, BF_{4}^{-}$$
 etc.

D. None of the above

#### Answer: A



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2. When aqueous solution of benzene diazonium chloride is boiled, the product formed is:

A. 
$$C_6H_5CH_2OH$$

$$\mathbf{B.}\,C_6H_6+N_2$$

C. $C_6H_5COOH$
D. $C_6H_5OH$
Answer: D
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3. Benzene diazonium chloride on reaction with phenol in weakly basic
medium gives:
A. diphenyl ether
B. p-hydroxy azobenzene
C. chlorobenzene
D. benzene
Answer: B
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# **Chapter Practice Very Short Answer Type Questions 1 Mark** 1. A diazo group is...... **Watch Video Solution** 2. Why diazonium group get easily substituted by other groups? **Watch Video Solution** 3. Why are benzene diazonium salts more stable than alkyl diazonium salts? Watch Video Solution

Chapter Practice Short Answer Type I Questions 1 Mark

Watch Video Solution 2. Usually, benzene diazonium salts are unstable but benzene diazonium fluoroborate is stable, why? **Watch Video Solution** 3. Aniline is diazotised in cold and then treated with dimethylaniline to give a coloured product. What is the structure of this product? **Watch Video Solution** 4. Starting from toluene, outline a synthesis of each of the following compounds using diazonium salts as intermediates.

1. Draw the resonating structures of benzene diazonium salt.

(i) p-fluorotoluene (ii) p-cresol Watch Video Solution Chapter Practice Short Answer Type Ii Questions 2 Mark 1. (i) What are coupling reactions? How does the electrophilic substitution reaction take place during this reaction? (ii) List some important applications of diazonium salts. **Watch Video Solution** Chapter Practice Short Answer Type Ii Questions 3 Mark 1. Accomplish the following conversions: (i) Benzene to m-bromophenol (ii) Aniline to 2,4,6-tribromofluorobenzene.

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## **Chapter Practice Long Answer Type Questions 7 Marks**

1. What is coupling reaction? Give an example.



2. How benzene diazonium chloride is converted to (a) benzene (b) anisole (c) phenol?

